Electronic Supplementary Information

Ferrocenecarboxylate-functionalized titanium-oxo-cluster: ferrocene wheel

as sensitizer for photocurrent response

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	C1	C2	C3
Empirical formula	$C_{84}H_{96}O_{24}Ti_6Fe_6$	$C_{84}H_{84}O_{24}Ti_6$	C ₉₀ H ₉₆ O ₂₄ Ti ₆
Formula weight	2112.10	1764.91	1849.06
Crystal system	monoclinic	triclinic	triclinic
Space group	$P2_{1}/n$	<i>P</i> -1	<i>P</i> -1
<i>a</i> (Å)	16.8249(7)	13.3081(10)	13.4769(14)
<i>b</i> (Å)	13.8268(4)	13.8105(12)	13.9496(15)
<i>c</i> (Å)	19.4260(7)	14.5437(11)	14.1303(13)
α (°)	90	63.018(8)	65.816(10)
β (°)	90.638(4)	66.477(7)	83.373(9)
γ (°)	90	62.871(8)	62.506(11)
$V(Å^3)$	4518.9(3)	2056.8(4)	2140.7(4)
Ζ	2	1	1
$ ho_{ m calcd} ({ m g}{ m cm}^{-3})$	1.552	1.425	1.434
$\mu ({ m mm^{-1}})$	1.501	0.630	0.609
<i>F</i> (000)	2160	912	960
$T(\mathbf{K})$	295(2)	295(2)	295(2)
Measured refls.	19561	14685	16003
Independent refls.	8852	7251	8402
R _{int}	0.0391	0.0717	0.0248
GOF	1.022	1.036	1.034
$R_1 [I > 2\sigma(I)]^{[a]}$	0.0501	0.0805	0.0442
$wR_2 [I > 2\sigma(I)]^{[b]}$	0.1070	0.1310	0.1129

Table S1. Crystal data and structure refinements summary for C1, C2 and C3.

 ${}^{a}R_{1} = \sum \left\| F_{o} \right\| - \left\| F_{c} \right\| / \sum \left\| F_{o} \right\| \cdot {}^{b}wR_{2} = \left[\sum w(F_{o}^{2} - F_{c}^{2})^{2} / \sum w(F_{o}^{2})^{2} \right]^{1/2}.$

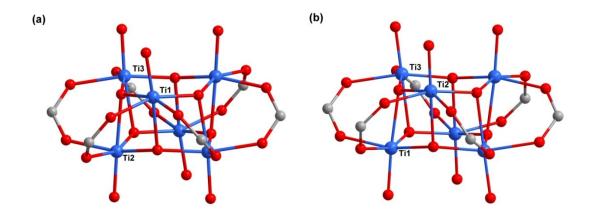


Fig. S1 (a) Coordination mode of $\{Ti_6\}$ in C2. (b) Coordination mode of $\{Ti_6\}$ in C3.

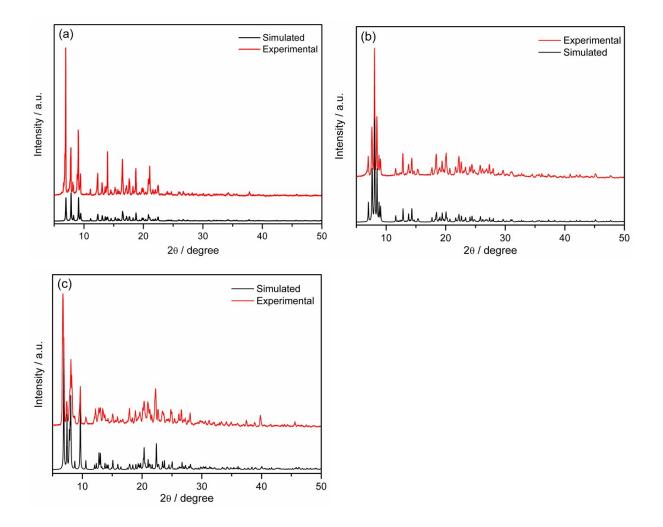


Fig. S2 XRD patterns of (a) C1, (b) C2 and (c) C3.

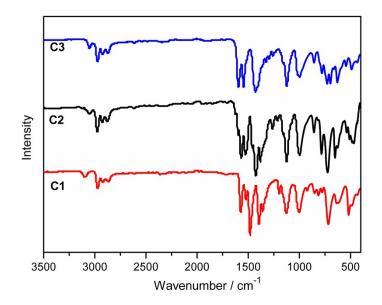


Fig. S3 FT-IR spectra of C1, C2 and C3.

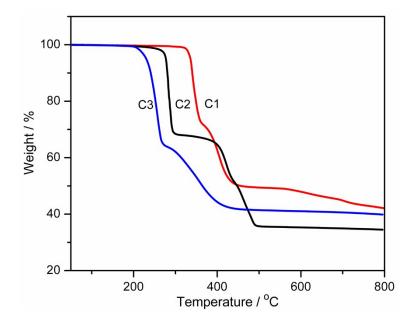


Fig. S4 The TGA curves of C1, C2 and C3.

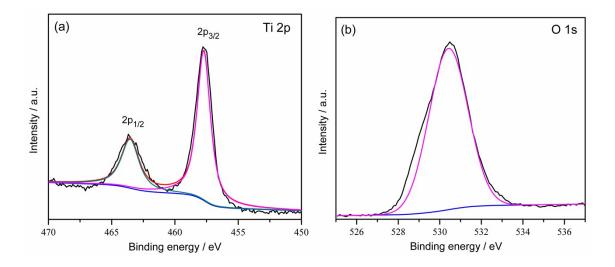


Fig. S5 XPS spectra of (a) Ti 2p peaks and (b) O 1s peaks for C2.

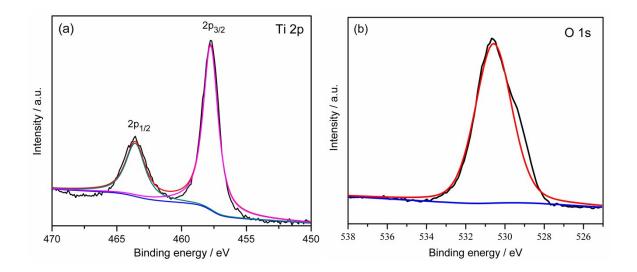


Fig. S6 XPS spectra of (a) Ti 2p peaks and (b) O 1s peaks for C3.

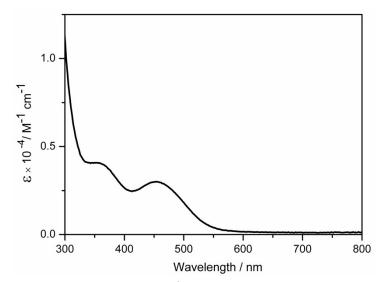


Fig. S7 UV-vis spectra of 1×10^{-4} M solution of C1 in dichloromethane.

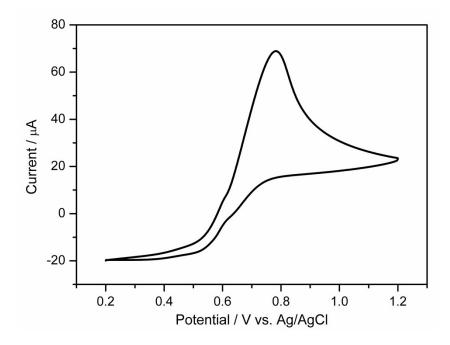
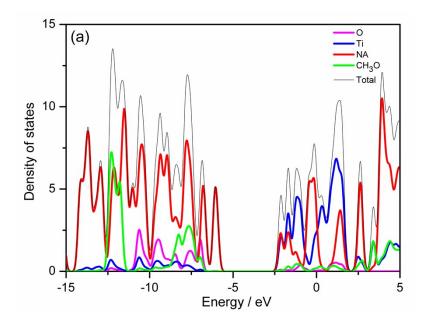


Fig. S8 CV curves of ferrocenecarboxylic acid in dichloromethane solution containing 0.1 M $n-Bu_4NPF_6$ as the supporting electrolyte, scan rate = 100 mV s⁻¹.



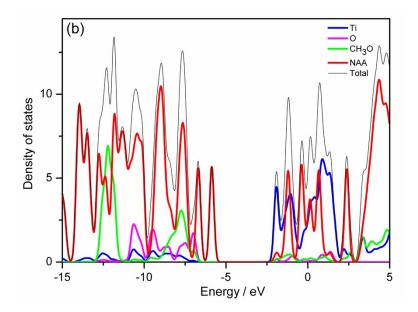


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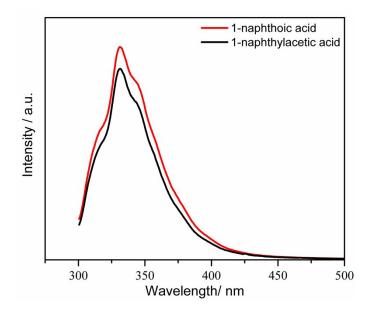


Figure S10 Room-temperature emission spectra of 1-naphthoic acid and 1-naphthylacetic acid in dichloromethane solutions $(1 \times 10^{-6} \text{ M})$ with 280 nm excitation.