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## Electronic Supplementary Information (ESI)

## Organic-free synthesis of zincoaluminosilicate zeolites from homogeneous gels prepared by a co-precipitation method

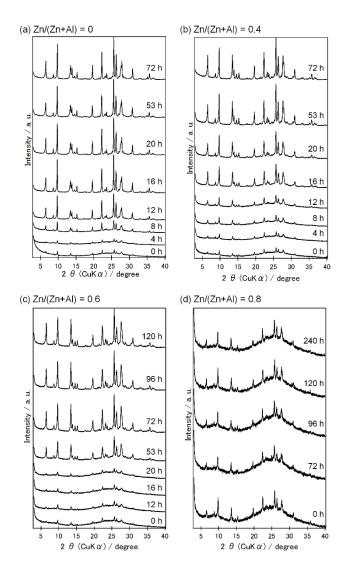
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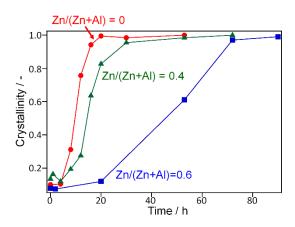
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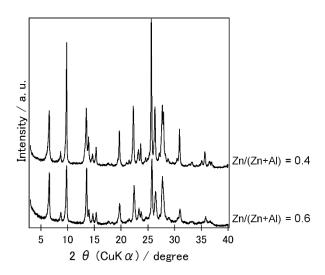
\*E-mail: okubo@chemsys.t.u-tokyo.ac.jp



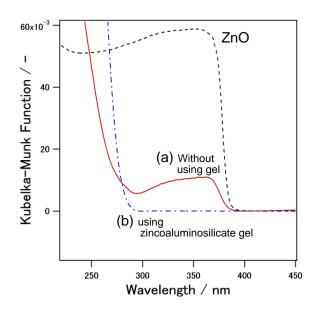
**Fig. S1** Powder XRD patterns of products synthesized for different periods of time from coprecipitated gels with Zn/(Zn+Al) of (a) 0, (b) 0.4, (c) 0.6, and (d) 0.8.



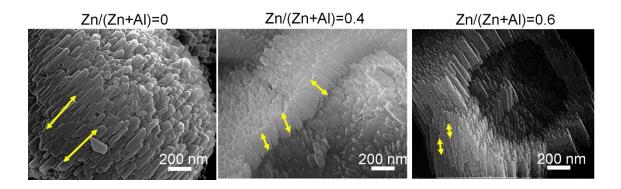
**Fig. S2** Crystallization curves of products synthesized from co-precipitated gels with Zn/(Zn+Al) of 0, 0.4, and 0.6. The crystallinity was calculated from the total areas of XRD peaks at around 19.7°, 22.4°, 25.8°, and 26.4°.



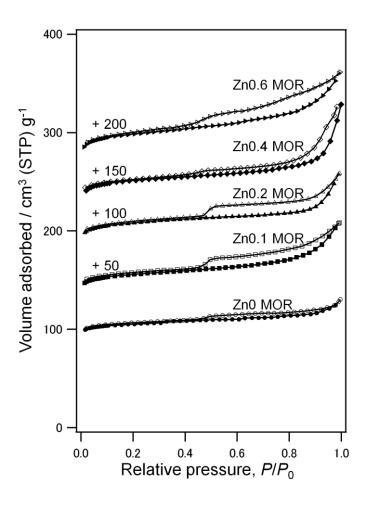
**Fig. S3** Powder XRD patterns of products synthesized using fumed silica, sodium aluminate, and zinc acetate at Zn/(Zn+AI) of 0.4 and 0.6.



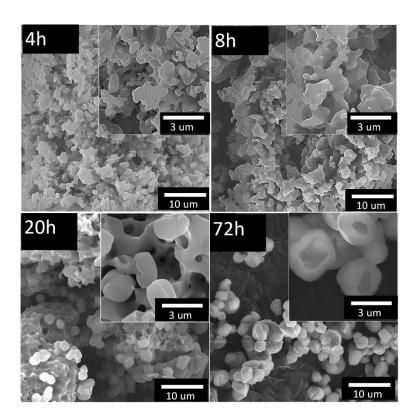
**Fig. S4** DR UV-vis spectra of products synthesized at Zn/(Zn+Al) of 0.6 using (a) conventional raw materials (fumed silica, sodium aluminate, and zinc acetate) and (b) co-precipitated zincoaluminosilicate gel.



**Fig. S5** Additional FE-SEM images of products synthesized from co-precipitated gels with Zn/(Zn+Al) of 0, 0.4, and 0.6.



**Fig. S6** Nitrogen adsorption—desorption isotherms of Zn0 MOR, Zn0.1 MOR, Zn0.2 MOR, Zn0.4 MOR, and Zn0.6 MOR. The isotherms of Zn0.4 MOR and Zn0.6 MOR were obtained after ion-exchange with  $NH_4NO_3$  and subsequent calcination.



**Fig. S7** FE-SEM images of products synthesized from co-precipitated gels with Zn/(Zn+Al) of 0.4 at 150  $^{\circ}$ C for 4, 8, 20, and 72 h.