

Supporting Information for

Manual Assembly of Rare-earth Polyoxometalate Micro-crystals Film Showing Polarized Luminescence

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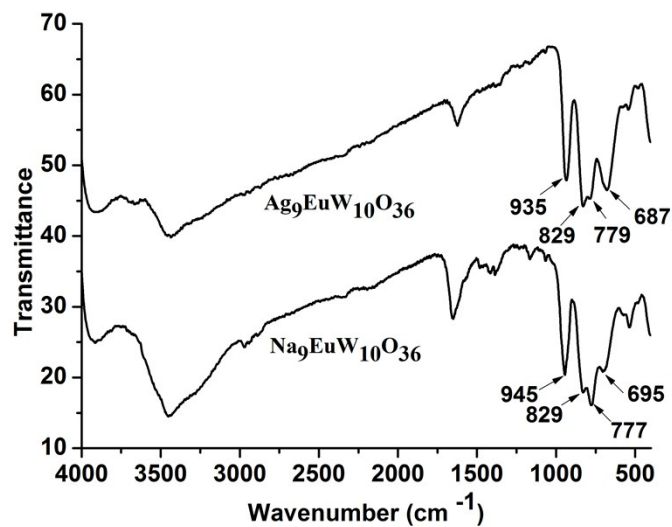


Figure S1. IR spectra of $\text{Ag}_9[\text{EuW}_{10}\text{O}_{36}]$ microplates and $\text{Na}_9[\text{EuW}_{10}\text{O}_{36}]$ powders.

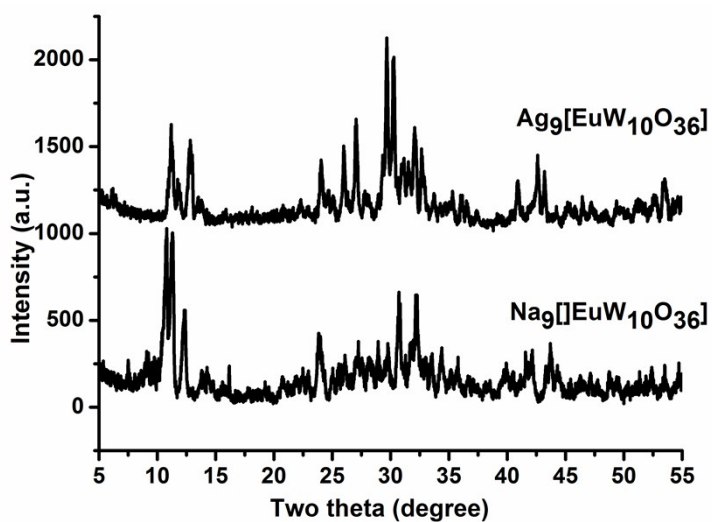
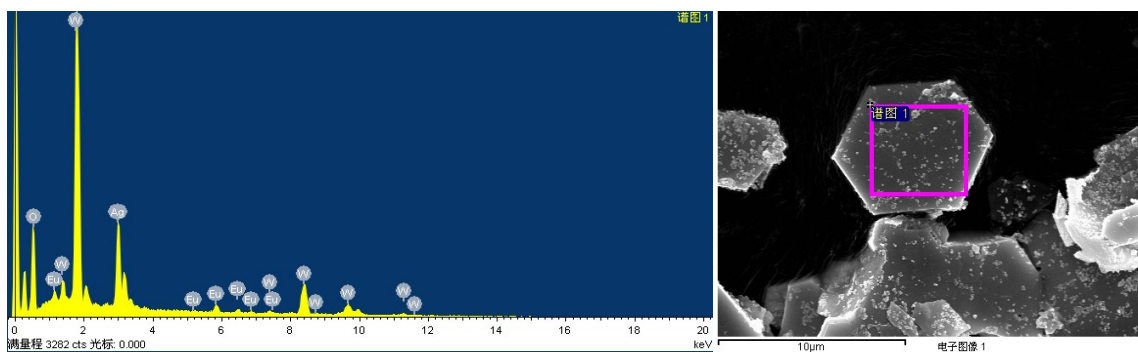


Figure S2. The XRD patterns of $\text{Na}_9[\text{EuW}_{10}\text{O}_{36}]$ and $\text{Ag}_9[\text{EuW}_{10}\text{O}_{36}]$ powders.



(a)

(b)

Figure S3. EDX analysis of the $\text{Ag}_9[\text{EuW}_{10}\text{O}_{36}]$ microplates (a) and the SEM image highlighting the area for EDX analysis (b).

Table S1. ICP analysis results of the atomic ratio of elements

element	weight percentage(%)	Atomic ratio
O	25.69	76.65
Ag	21.14	9.36
Eu	3.38	1.06
W	49.79	12.93

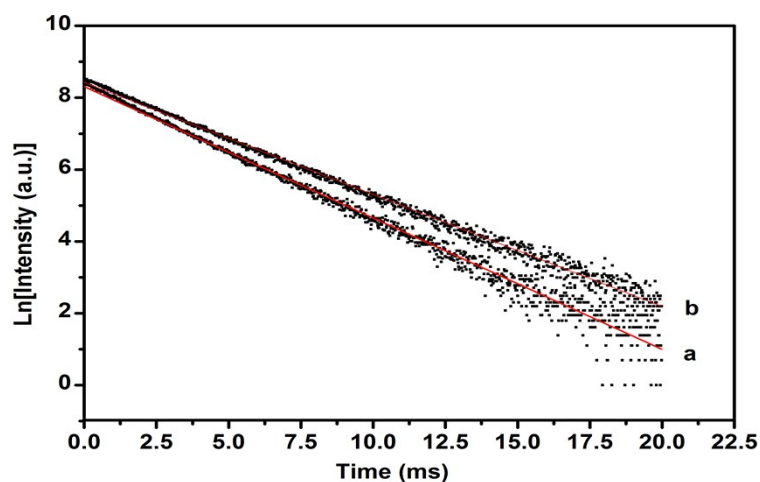


Figure S4. Emission decay curves of $^5\text{D}_0 \rightarrow ^7\text{F}_1$ emitting state under 270 nm excitation of the $\text{Ag}_9[\text{EuW}_{10}\text{O}_{36}]$ micro-crystals (a) and the manually assembled film (b).

Table S2. Summary of luminescence data of Ag₉[EuW₁₀O₃₆] micro-crystals and the film prepared by manual method

Sample	k _r (ms ⁻¹)	k _{nr} (ms ⁻¹)	k _{tot} (ms ⁻¹)	τ(ms)	η*
Ag₉[EuW₁₀O₃₆] micro-crystals	0.1930	0.1296	0.3226	3.10	59.83%
Film prepared by manual method	0.1810	0.1506	0.3436	2.91	52.67%

*the value of η is calculated by incorporating fluorescence lifetimes into the formula (1), (2) and (3),

$$k_r = k_{r(0 \rightarrow 1)} \frac{\sum_{j=0}^4 S_{(0 \rightarrow j)}}{S_{(0 \rightarrow 1)}} \quad (1)$$

where k_{r(0→1)} = 1.35×10⁻² s⁻¹ stands for radioactive transition, S stands for integral area of characteristic Eu(III) ion emission.

$$k_{tot} = \frac{1}{\tau} = k_r + k_{nr} \quad (2)$$

where τ stands for fluorescence life time k_{nr} stands for nonradioactive transition.

$$\eta = \frac{k_r}{k_r + k_{nr}} \quad (3)$$