

Electronic Supporting information

[AlCl₃(BnMe₂-tacn)] – a new metal chelate scaffold for radiofluorination by Cl/F exchange

William Levason^a, Sajinder K. Luthra^b, Graeme McRobbie^b, Francesco M. Monzittu^a and Gillian Reid^{a*}

Infrared spectra were recorded as Nujol mulls between CsI plates using a Perkin-Elmer Spectrum100 spectrometer over the range 4000–200 cm⁻¹. ¹H NMR spectra were recorded using a Bruker AV300 spectrometer. ¹⁹F{¹H} and ²⁷Al NMR spectra used a Bruker AV400 spectrometer and are referenced (externally) to CCl₄ and aqueous [Al(H₂O)₆]³⁺ (pH 1), respectively. Microanalyses were undertaken by London Metropolitan University. AlCl₃ was used as received (Sigma). BzMe₂-tacnⁱ was prepared *via* the literature method.

[AlCl₃(BzMe₂-tacn)]: AlCl₃ (0.067 g, 0.50 mmol) was added to a solution of BnMe₂-tacn (0.13 g, 0.50 mmol) in CH₃CN (5 mL) at room temperature with stirring which leads to the rapid formation of a precipitate. After 30 mins. the solvent was removed by filtration. The white precipitate was washed with a small amount of CH₂Cl₂ solvent and dried *in vacuo*. Yield: 0.13 g, 66%. Required for C₁₅H₂₅AlCl₃N₃: C, 47.3; H, 6.6; N, 11.0. Found: C, 47.0; H, 6.6; N, 11.2. ¹H NMR (CD₂Cl₂, 298 K): δ 7.31 (m, [5H], ArH), 4.58 (s, [2H], Ar-CH₂), 3.54 (m, [2H], tacn-CH₂), 3.29 (m, [4H], tacn-CH₂), 2.92 (s, [6H], CH₃), 2.65 (m, [4H], tacn-CH₂), 2.28 (m, [2H], tacn-CH₂). IR (Nujol, ν/cm⁻¹): 398, 385 (Al–Cl).

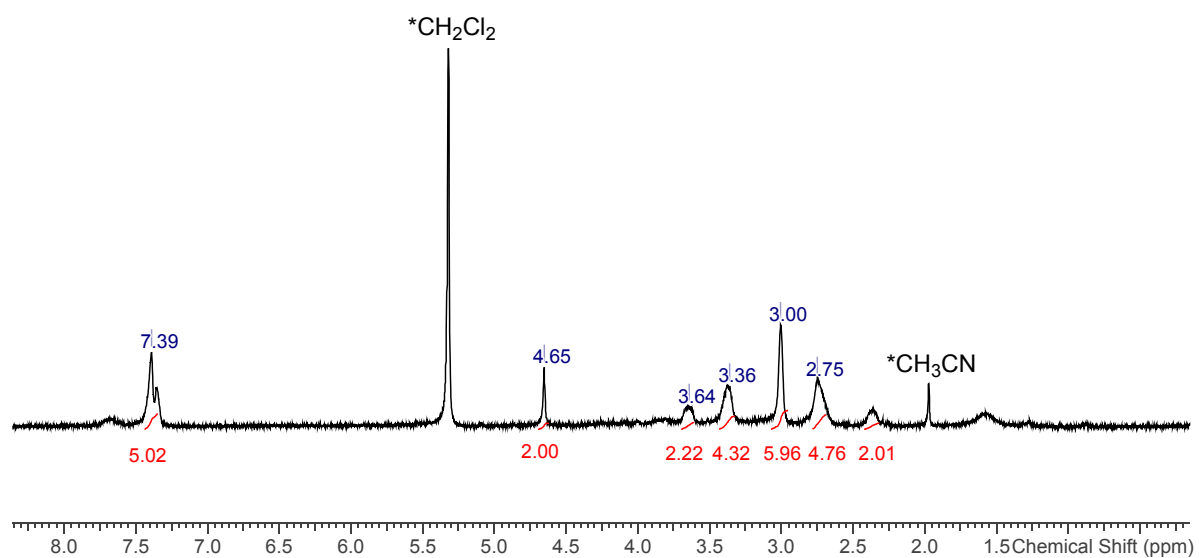


Figure S1. ^1H NMR spectrum of $[\text{AlCl}_3(\text{BnMe}_2\text{tacn})]$ in CD_2Cl_2 .

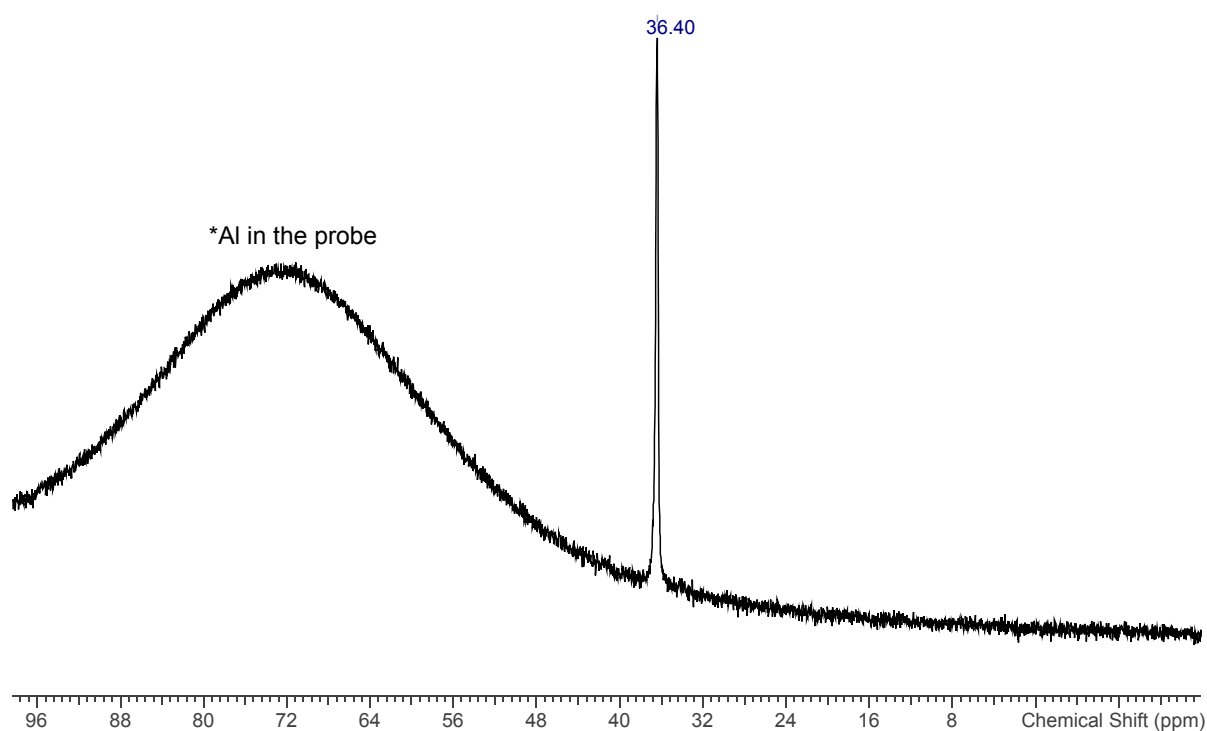


Figure S2. ^{27}Al NMR spectrum of $[\text{AlCl}_3(\text{BnMe}_2\text{tacn})]$ in CD_2Cl_2 .

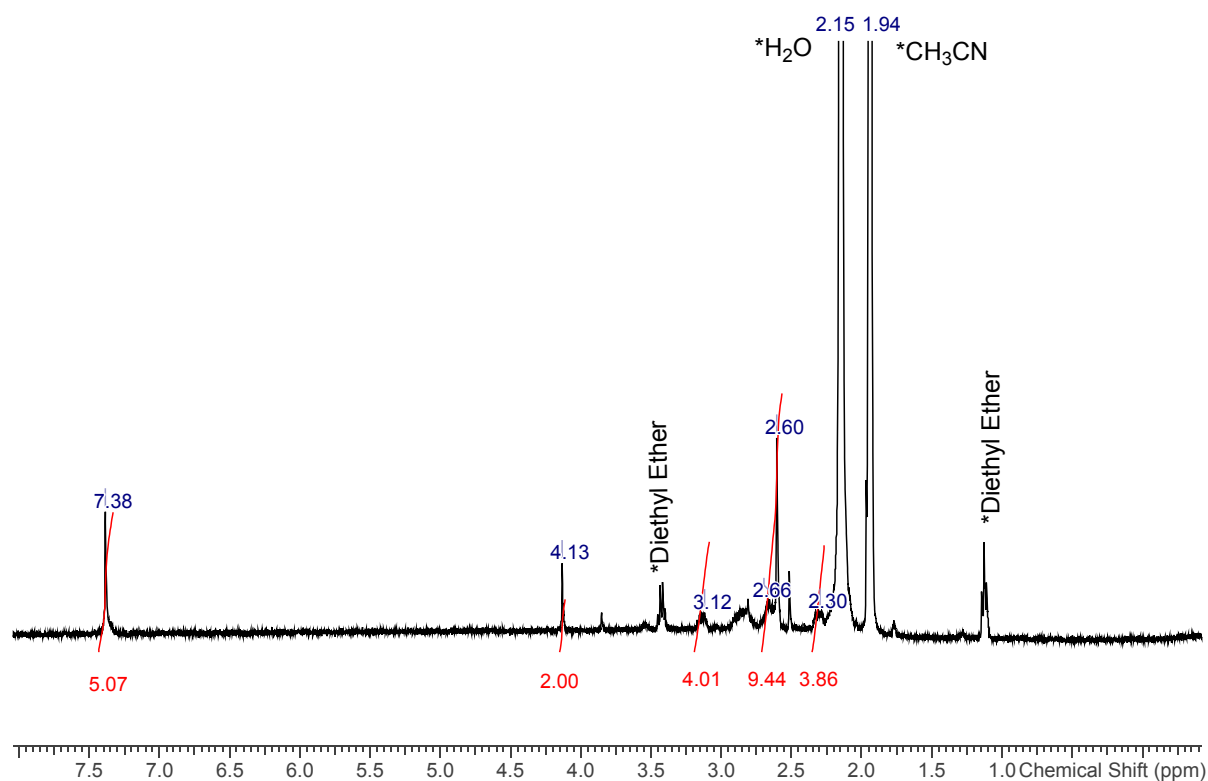


Figure S3. ¹H NMR spectrum of [AlF₃(BnMe₂tacn)] in CD₃CN.

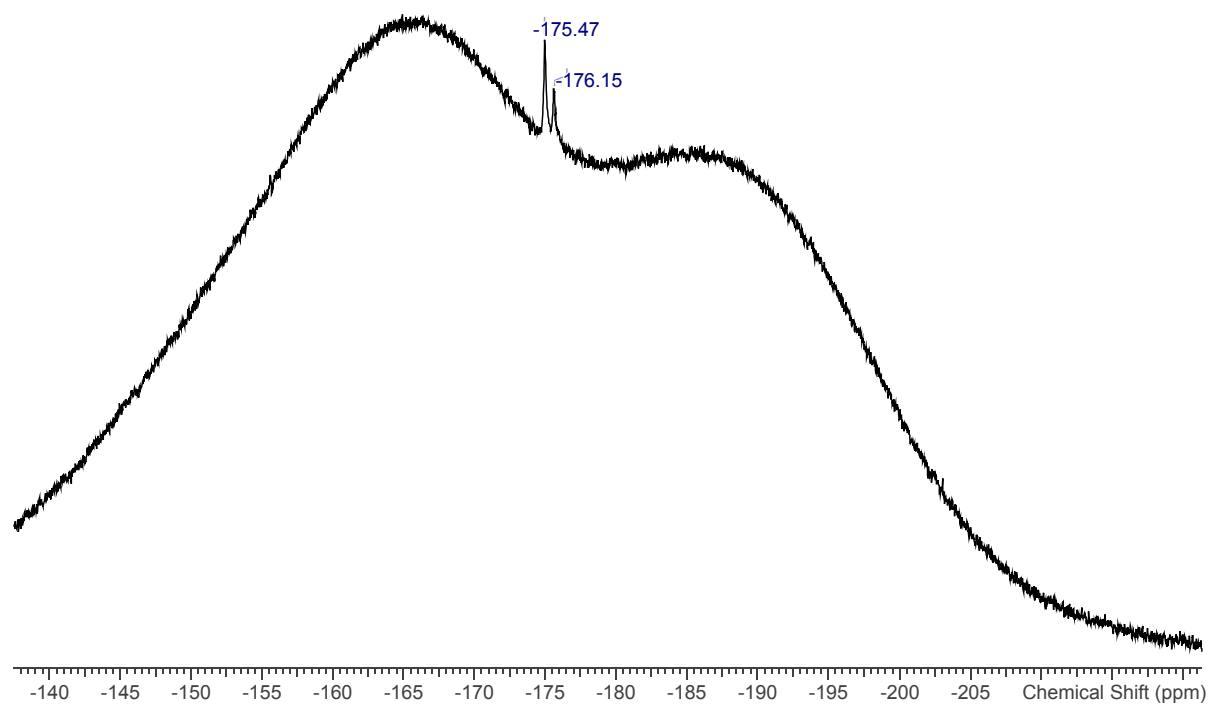


Figure S4. ¹⁹F{¹H} NMR spectrum of [AlF₃(BnMe₂tacn)] in CD₃CN (the broad background is due to Teflon in the probe).

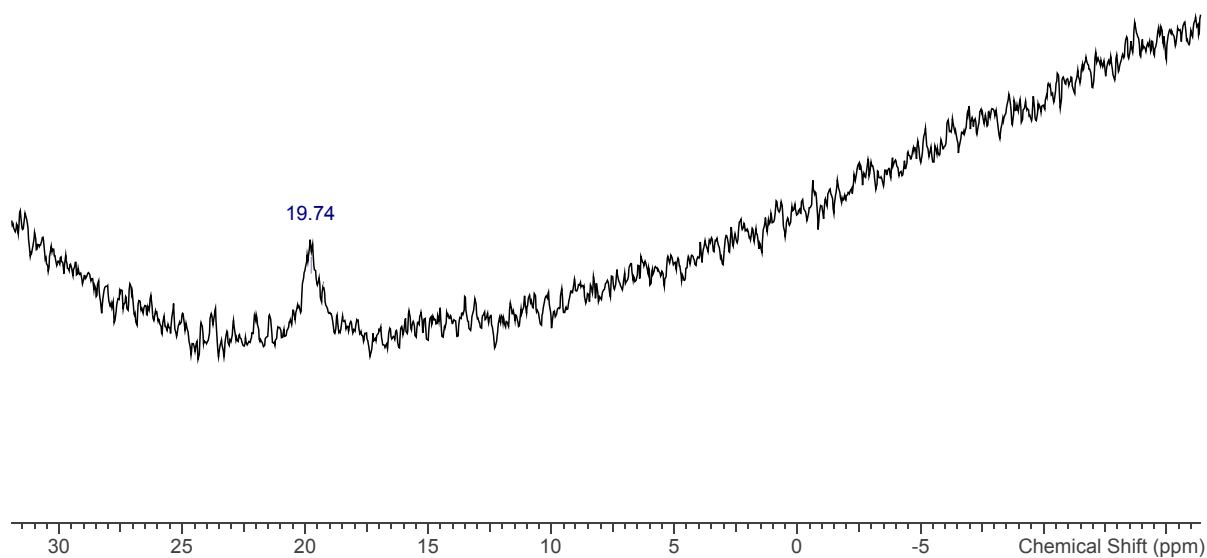


Figure S5. ^{27}Al NMR spectrum of $[\text{AlF}_3(\text{BnMe}_2\text{tacn})]$ in CD_3CN .

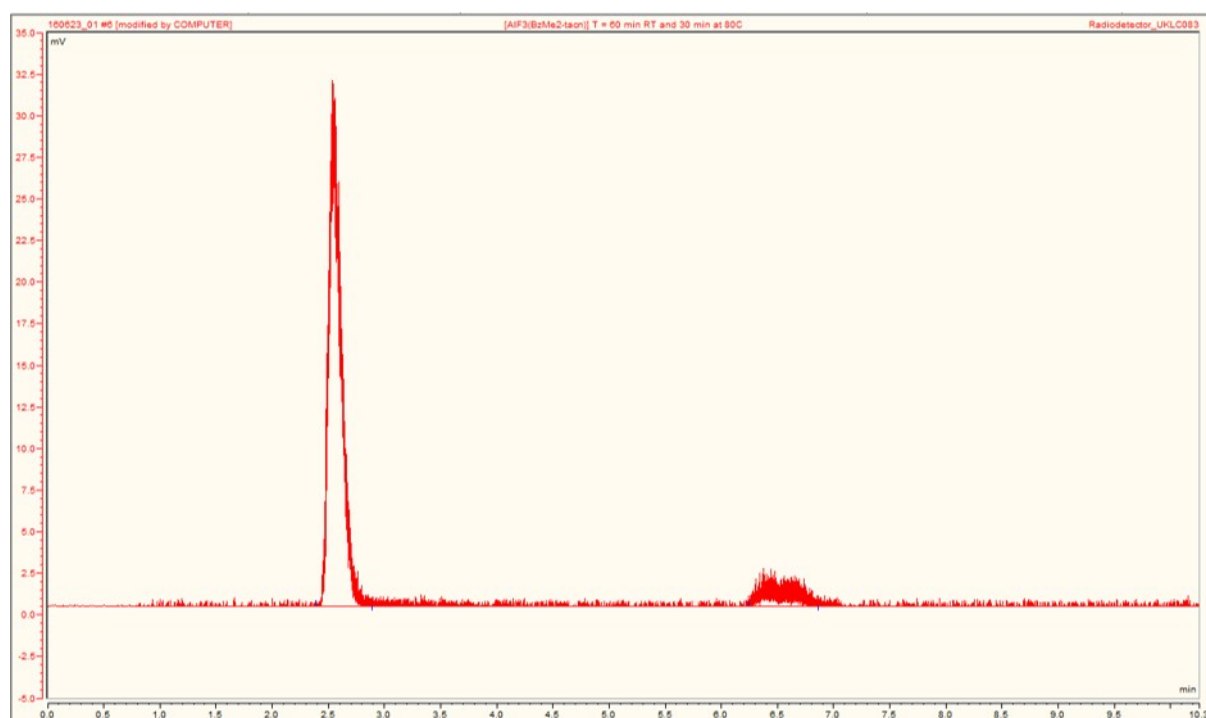


Figure S6. Analytical radio-HPLC chromatogram of the crude product from reaction of $[\text{AlCl}_3(\text{BnMe}_2\text{tacn})]$ (1 mg, 2.63 μmol) in MeCN (0.6 mL) with 2.99 eq of KF doped with 0.4 mL of aqueous $^{18}\text{F}^-$ at 80 $^\circ\text{C}$ for 30 mins. Peak 1: Rt = 2.54 min 91% (^{18}F). Peak 2: Rt = 6.68 min 9% ($[\text{Al}^{18}\text{F}^{19}\text{F}_2(\text{BnMe}_2\text{tacn})]$).

- i. M. J. Belousoff, M. B. Duriska, B. Graham, S. R. Batten, B. Moubaraki, K. S. Murray, L. Spiccia, *Inorg. Chem.*, 2006, **45**, 3746.