Supporting Information for

## Triaminoborane-Bridged Diphosphine Complexes with Ni and Pd: Coordination Chemistry, Structures, and Ligand-Centered Reactivity

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**Figure S1.** Molecular structure of ( ${}^{Ph}TBDPhos$ )PdCl<sub>2</sub> (**3**) with thermal ellipsoids at the 35% probability level. Hydrogen atoms were omitted from the figure.

|   | <sup>Ph</sup> TBDPhos     | <sup>iPr</sup> TBDPhos   | 1   | 2                           | 3                           | 4                           |
|---|---------------------------|--------------------------|---|-----------------------------|-----------------------------|-----------------------------|
| formula                                 | $C_{32}H_{36}BCl_4N_3P_2$ | $C_{18}H_{40}BN_3P_2 \\$ | C <sub>16</sub> H <sub>17</sub> B <sub>0.5</sub> Cl <sub>4</sub> N <sub>1.5</sub> Ni <sub>0.5</sub> P | $C_{18}H_{40}BCl_2N_3NiP_2$ | $C_{30}H_{32}BCl_2N_3P_2Pd$ | $C_{18}H_{40}BCl_2N_3P_2Pd$ |
| FW (g mol <sup>-1</sup> )               | 677.19                    | 371.28                   | 437.75  | 500.89                      | 684.63                      | 548.58                      |
| crystal system                          | Monoclinic                | Monoclinic               | Triclinic   | Monoclinic                  | Monoclinic                  | Monoclinic                  |
| space group                             | C2/c                      | $P2_1/c$                 | P-1   | C2/c                        | $P2_1/c$                    | $P2_1/n$                    |
| a (Å)                                   | 22.219(2)                 | 8.6254(9)                | 9.9895(10)  | 12.6274(13)                 | 12.9178(13)                 | 8.8840(9)                   |
| b (Å)                                   | 13.6886(14)               | 15.2128(15)              | 12.2230(12)   | 10.9057(11)                 | 13.3011(13)                 | 17.2077(17)                 |
| c (Å)                                   | 13.9571(14)               | 17.6220(18)              | 16.8174(17)   | 17.6347(18)                 | 18.1092(18)                 | 16.0964(16)                 |
| α (deg)                                 | 90                        | 90                       | 90.040(5)   | 90                          | 90                          | 90                          |
| β (deg)                                 | 127.323(5)                | 101.400(5)               | 103.664(5)  | 101.618(5)                  | 104.019(5)                  | 94.786(5)                   |
| γ (deg)                                 | 90                        | 90                       | 106.259(5)  | 90                          | 90                          | 90                          |
| volume (Å <sup>3</sup> )                | 3375.8(6)                 | 2266.7(4)                | 1910.6(3)   | 2378.7(4)                   | 3018.9(5)                   | 2452.1(4)                   |
| Z                                       | 8                         | 4                        | 4   | 4                           | 4                           | 4                           |
| $\rho_{calc}$ (g cm <sup>-3</sup> )     | 1.332                     | 1.088                    | 1.522   | 1.399                       | 1.506                       | 1.486                       |
| μ (mm <sup>-1</sup> )                   | 0.473                     | 0.197                    | 1.179   | 1.184                       | 0.923                       | 1.115                       |
| F(000)                                  | 1408                      | 816                      | 892   | 1064                        | 1392                        | 1136                        |
| $\theta$ range (deg)                    | 2.92/27.37                | 2.93/27.94               | 2.45/27.41  | 2.49/26.36                  | 2.92/29.04                  | 2.59/27.97                  |
| R(int)                                  | 0.0440                    | 0.0574                   | 0.0500  | 0.0309                      | 0.0294                      | 0.0351                      |
| data/restraints/parameters              | 4062/0/191                | 5424/0/225               | 8798/0/426  | 2444/0/128                  | 8050/0/352                  | 5882/0/252                  |
| GOF                                     | 1.050                     | 1.021                    | 1.026   | 1.035                       | 1.079                       | 1.023                       |
| $R_1 [I > 2\sigma(I)]^a$                | 0.0331                    | 0.0369                   | 0.0355  | 0.0239                      | 0.0233                      | 0.0245                      |
| wR <sub>2</sub> (all data) <sup>b</sup> | 0.0872                    | 0.0998                   | 0.0941  | 0.0578                      | 0.0558                      | 0.0577                      |
| Ext. Coeff                              | -                         | -                        | -   | -                           | -                           | -                           |
| Largest Peak/Hole (e·Å-3)               | 0.372/-0.372              | 0.399/-0.265             | 0.889/-0.943  | 0.320/-0.224                | 0.439/-0.422                | 0.470/-0.329                |
| Temp (K)                                | 190(2)                    | 190(2)                   | 190(2)  | 190(2)                      | 190(2)                      | 190(2)                      |

**Table S1**. Crystallographic data for <sup>Ph</sup>TBDPhos, <sup>iPr</sup>TBDPhos, (<sup>Ph</sup>TBDPhos)NiCl<sub>2</sub> (1), (<sup>iPr</sup>TBDPhos)NiCl<sub>2</sub> (2), (<sup>Ph</sup>TBDPhos)PdCl<sub>2</sub> (3), and (<sup>iPr</sup>TBDPhos)PdCl<sub>2</sub> (4).

 ${}^{a}\mathbf{R}_{1} = \sum |F_{o}| - |F_{c}|| / |\sum |F_{o}|$  for reflections with  $F_{o}^{2} > 2 \sigma(F_{o}^{2})$ .

 ${}^{b}wR_{2} = \left[\sum w(F_{o}^{2} - F_{c}^{2})^{2} / \sum (F_{o}^{2})^{2}\right]^{1/2}$  for all reflections.

|   | 5  | 7                | 8                 | 9                 | 10                                    |
|---|--|------------------|-------------------|-------------------|---------------------------------------|
| formula   | C <sub>30</sub> H <sub>35</sub> BClN <sub>3</sub> NiO <sub>2</sub> P | C31H36BCl2N3NiOP | C31H36BCl2N3OP2Pd | C33H38BCl2N3OP2Pd | $C_{62}H_{72}B_2Cl_6F_2N_6Ni_2O_2P_4$ |
| FW (g mol <sup>-1</sup> )                                 | 636.52   | 668.99           | 716.68            | 742.71            | 964.58                                |
| crystal system  | Monoclinic   | Monoclinic       | Monoclinic        | Monoclinic        | Monoclinic                            |
| space group   | $P2_1/n$   | $P2_1/c$         | $P2_1/c$          | $P2_1/c$          | $P2_1/n$                              |
| a (Å)   | 13.8071(14)  | 10.9172(11)      | 10.9956(11)       | 10.8193(11)       | 12.3721(12)                           |
| b (Å)   | 11.3545(11)  | 21.128(2)        | 21.129(2)         | 22.229(2)         | 32.375(3)                             |
| c (Å)   | 20.486(2)  | 14.1842(14)      | 14.3055(14)       | 14.5574(15)       | 16.7696(17)                           |
| a (deg)   | 90   | 90               | 90                | 90                | 90                                    |
| β (deg)   | 106.374(5)   | 110.222(5)       | 110.691(5)        | 110.787(5)        | 91.629(5)                             |
| γ (deg)   | 90   | 90               | 90                | 90                | 90                                    |
| volume (Å <sup>3</sup> )                                  | 3081.4(5)  | 3070.0(5)        | 3109.2(5)         | 3273.2(6)         | 6714.3(11)                            |
| Z   | 4  | 4                | 4                 | 4                 | 4                                     |
| $\rho_{calc}$ (g cm <sup>-3</sup> )                       | 1.372  | 1.447            | 1.531             | 1.507             | 1.431                                 |
| $\mu$ (mm <sup>-1</sup> )                                 | 0.852  | 0.941            | 0.902             | 0.860             | 0.948                                 |
| F(000)  | 1328   | 1392             | 1464              | 1520              | 2992                                  |
| $\theta$ range (deg)                                      | 1.00/25.35   | 2.46/24.33       | 2.24/29.03        | 2.21/28.33        | 2.25/27.81                            |
| R(int)  | 0.0718   | 0.0739           | 0.0466            | 0.0241            | 0.0471                                |
| data/restraints/parameters                                | 5637/3/373   | 6368/0/372       | 7766/0/375        | 8150/0/400        | 15831/1/783                           |
| GOF   | 0.881  | 0.912            | 1.039             | 1.056             | 0.855                                 |
| $\mathbf{R}_1 \left[ I > 2\sigma(I) \right]^{\mathbf{a}}$ | 0.0391   | 0.0408           | 0.0334            | 0.0212            | 0.0378                                |
| wR <sub>2</sub> (all data) <sup>b</sup>                   | 0.1320   | 0.1391           | 0.1263            | 0.0564            | 0.0985                                |
| Ext. Coeff  | -  | 0.0020(6)        | -                 | -                 | -                                     |
| Largest Peak/Hole (e·Å-3)                                 | 0.290/-0.340   | 0.564/-0.589     | 1.032/-0.973      | 0.425/-0.398      | 1.097/-1.203                          |
| Temp (K)  | 190(2)   | 190(2)           | 190(2)            | 170(2)            | 190(2)                                |

**Table S2**. Crystallographic data for  $\{[(^{Ph}TBDPhos-H_2O)Ni]_2(\mu-OH)_2\}Cl_2(5), (^{Ph}TBDPhos-MeOH)NiCl_2(7), (^{Ph}TBDPhos-MeOH)PdCl_2(8), (^{Ph}TBDPhos-C_3H_5OH)PdCl_2(9), and <math>\{[(^{Ph}TBDPhos-HF)Ni]_2(\mu-OH)_2\}Cl_2(10).$ 

 ${}^{a}\mathbf{R}_{1} = \sum |F_{o}| - |F_{c}|| / |\sum |F_{o}|$  for reflections with  $F_{o}^{2} > 2 \sigma(F_{o}^{2})$ .

 ${}^{b}wR_{2} = \left[\sum w(F_{o}^{2} - F_{c}^{2})^{2} / \sum (F_{o}^{2})^{2}\right]^{1/2}$  for all reflections.





Figure S2. <sup>1</sup>H NMR spectrum of TBD.



Figure S3. <sup>11</sup>B NMR spectrum of TBD.



**Figure S4.** <sup>1</sup>H NMR spectrum of <sup>Ph</sup>TBDPhos.



Figure S5. <sup>13</sup>C NMR spectrum of <sup>Ph</sup>TBDPhos.



Figure S6. <sup>11</sup>B NMR spectrum of <sup>Ph</sup>TBDPhos



Figure S7.  ${}^{31}P{}^{1}H$  NMR spectrum of  ${}^{Ph}TBDPhos$ .



**Figure S8.** <sup>1</sup>H NMR spectrum of <sup>iPr</sup>TBDPhos. The \* symbol indicates resonances assigned to residual Et<sub>2</sub>O and pentane solvent.



**Figure S9.** <sup>13</sup>C NMR spectrum of <sup>iPr</sup>TBDPhos.



Figure S10. <sup>11</sup>B NMR spectrum of <sup>iPr</sup>TBDPhos.



Figure S11.  ${}^{31}P{}^{1}H$  NMR spectrum of  ${}^{iPr}TBDPhos$ .



**Figure S12.** <sup>1</sup>H NMR spectrum of ( $^{Ph}TBDPhos$ )NiCl<sub>2</sub> (1). The \* symbol indicates resonances assigned to residual CH<sub>2</sub>Cl<sub>2</sub> and Et<sub>2</sub>O solvent.



**Figure S13.** <sup>13</sup>C NMR spectrum of (<sup>Ph</sup>TBDPhos)NiCl<sub>2</sub> (1). The \* symbol indicates a resonance assigned to residual CH<sub>2</sub>Cl<sub>2</sub> solvent.



Figure S14. <sup>11</sup>B NMR spectrum of ( $^{Ph}TBDPhos$ )NiCl<sub>2</sub> (1).



Figure S15.  ${}^{31}P{}^{1}H$  NMR spectrum of ( ${}^{Ph}TBDPhos$ )NiCl<sub>2</sub> (1).



**Figure S16.** <sup>1</sup>H NMR spectrum of ( $^{iPr}$ TBDPhos)NiCl<sub>2</sub> (**2**). The \* symbol indicates resonances assigned to residual CH<sub>2</sub>Cl<sub>2</sub> and Et<sub>2</sub>O solvent.



Figure S17. <sup>13</sup>C NMR spectrum of (<sup>iPr</sup>TBDPhos)NiCl<sub>2</sub> (2).



**Figure S18.** <sup>11</sup>B NMR spectrum of (<sup>iPr</sup>TBDPhos)NiCl<sub>2</sub> (**2**).



**Figure S19.** <sup>31</sup>P $\{^{1}H\}$  NMR spectrum of (<sup>iPr</sup>TBDPhos)NiCl<sub>2</sub> (**2**).



**Figure S20.** <sup>1</sup>H NMR spectrum of ( $^{Ph}TBDPhos$ )PdCl<sub>2</sub> (**3**). The \* symbol indicates resonances assigned to residual CH<sub>2</sub>Cl<sub>2</sub> and Et<sub>2</sub>O solvent.



**Figure S21.** <sup>13</sup>C NMR spectrum of (<sup>Ph</sup>TBDPhos)PdCl<sub>2</sub> (**3**). The \* symbol indicates a resonances assigned to residual CH<sub>2</sub>Cl<sub>2</sub>.



**Figure S22.** <sup>11</sup>B NMR spectrum of (<sup>Ph</sup>TBDPhos)PdCl<sub>2</sub> (**3**).



**Figure S23.**  ${}^{31}P{}^{1}H$  NMR spectrum of ( ${}^{Ph}TBDPhos$ )PdCl<sub>2</sub> (**3**).



**Figure S24.** <sup>1</sup>H NMR spectrum of (<sup>iPr</sup>TBDPhos)PdCl<sub>2</sub> (**4**).



Figure S25. <sup>13</sup>C NMR spectrum of (<sup>iPr</sup>TBDPhos)PdCl<sub>2</sub> (4).



**Figure S26.** <sup>11</sup>B NMR spectrum of (<sup>iPr</sup>TBDPhos)PdCl<sub>2</sub> (4).



**Figure S27.** <sup>31</sup>P $\{^{1}H\}$  NMR spectrum of (<sup>iPr</sup>TBDPhos)PdCl<sub>2</sub> (4).



**Figure S28.** <sup>1</sup>H NMR spectrum of  $\{[({}^{Ph}TBDPhos-H_2O)Ni]_2(\mu-OH)_2\}Cl_2(5)$ . The \* symbol indicates resonances assigned to added NEt<sub>3</sub>. The inset shows the upfield shift assigned to the bridging hydroxide.



**Figure S29.** <sup>13</sup>C NMR spectrum of {[( $^{Ph}TBDPhos-H_2O$ )Ni]<sub>2</sub>( $\mu$ -OH)<sub>2</sub>}Cl<sub>2</sub> (**5**). The \* symbol indicates resonances assigned to added NEt<sub>3</sub>, and the # symbol indicates resonances assigned to **1**.



**Figure S30.** <sup>11</sup>B NMR spectrum of  $\{[({}^{Ph}TBDPhos-H_2O)Ni]_2(\mu-OH)_2\}Cl_2$  (5) taken before overnight <sup>13</sup>C NMR data collection.



Figure S31. <sup>11</sup>B NMR spectrum of {[( $^{Ph}TBDPhos-H_2O$ )Ni]<sub>2</sub>( $\mu$ -OH)<sub>2</sub>}Cl<sub>2</sub> (5) taken after overnight <sup>13</sup>C NMR data collection. The # symbol indicates the broad resonance assigned to 1.



Figure S32.  ${}^{31}P{}^{1}H$  NMR spectrum of {[( ${}^{Ph}TBDPhos-H_2O$ )Ni]<sub>2</sub>( $\mu$ -OH)<sub>2</sub>}Cl<sub>2</sub> (5).



Figure S33. <sup>1</sup>H NMR spectrum of  $\{[({}^{Ph}TBDPhos-H_2O)Pd]_2(\mu-OH)_2\}Cl_2(6)$ . The \* symbol indicates resonances assigned to added NEt<sub>3</sub>.



Figure S34. <sup>13</sup>C NMR spectrum of  $\{[(^{Ph}TBDPhos-H_2O)Pd]_2(\mu-OH)_2\}Cl_2(6)$ . The \* symbol indicates resonances assigned to added NEt<sub>3</sub>.



Figure S35. <sup>11</sup>B NMR spectrum of  $\{[(^{Ph}TBDPhos-H_2O)Pd]_2(\mu-OH)_2\}Cl_2$  (6).



Figure S36.  ${}^{31}P{}^{1}H$  NMR spectrum of {[( ${}^{Ph}TBDPhos-H_2O$ )Pd]<sub>2</sub>( $\mu$ -OH)<sub>2</sub>}Cl<sub>2</sub> (6).



**Figure S37.** <sup>1</sup>H NMR spectrum of (<sup>Ph</sup>TBDPhos-MeOH)NiCl<sub>2</sub> (7). The \* symbol indicates resonances assigned to added NEt<sub>3</sub>.



Figure S38. <sup>13</sup>C NMR spectrum of (<sup>Ph</sup>TBDPhos-MeOH)NiCl<sub>2</sub> (7). The \* symbol indicates resonances assigned to added NEt<sub>3</sub>.



Figure S39. <sup>11</sup>B NMR spectrum of (<sup>Ph</sup>TBDPhos-MeOH)NiCl<sub>2</sub> (7).



Figure S40. <sup>31</sup>P{<sup>1</sup>H} NMR spectrum of (<sup>Ph</sup>TBDPhos-MeOH)NiCl<sub>2</sub> (7).



**Figure S41.** <sup>1</sup>H NMR spectrum of ( $^{Ph}TBDPhos-MeOH$ )PdCl<sub>2</sub> (8). The \* symbol indicates resonances assigned to H<sub>2</sub>O and CH<sub>2</sub>Cl<sub>2</sub>. The # symbol indicates resonances assigned to free MeOH. The ‡ symbol indicates resonances assigned to added NEt<sub>3</sub>.



**Figure S42.** <sup>13</sup>C NMR spectrum of ( $^{Ph}TBDPhos-MeOH$ )PdCl<sub>2</sub> (8). The \* symbol indicates resonances assigned to CH<sub>2</sub>Cl<sub>2</sub> and free MeOH. The # symbol indicates unassigned peak which had been growing during overnight data collection. The ‡ symbol indicates resonances assigned to added NEt<sub>3</sub>.



**Figure S43.** <sup>11</sup>B NMR spectrum of (<sup>Ph</sup>TBDPhos-MeOH)PdCl<sub>2</sub> (8).



**Figure S44.**  ${}^{31}P{}^{1}H$  NMR spectrum of ( ${}^{Ph}TBDPhos-MeOH$ )PdCl<sub>2</sub> (8).



Figure S45. <sup>1</sup>H NMR spectrum of (<sup>Ph</sup>TBDPhos-C<sub>3</sub>H<sub>5</sub>OH)PdCl<sub>2</sub> (9). The \* symbol indicates resonances assigned to added NEt<sub>3</sub>.



Figure S46. <sup>13</sup>C NMR spectrum of (<sup>Ph</sup>TBDPhos-C<sub>3</sub>H<sub>5</sub>OH)PdCl<sub>2</sub> (9). The \* symbol indicates resonances assigned to added NEt<sub>3</sub>.





Figure S48.  ${}^{31}P{}^{1}H$  NMR spectrum of ( ${}^{Ph}TBDPhos-C_{3}H_{5}OH$ )PdCl<sub>2</sub> (9).



Figure S49. <sup>1</sup>H NMR spectrum of  $\{[(^{Ph}TBDPhos-HF)Ni]_2(\mu-OH)_2\}Cl_2$  (10).



Figure S50. <sup>13</sup>C NMR spectrum of  $\{[(^{Ph}TBDPhos-HF)Ni]_2(\mu-OH)_2\}Cl_2$  (10).



Figure S51. <sup>11</sup>B NMR spectrum of  $\{[(^{Ph}TBDPhos-HF)Ni]_2(\mu-OH)_2\}Cl_2$  (10).



Figure S52. <sup>31</sup>P{<sup>1</sup>H} NMR spectrum of {[( $^{Ph}TBDPhos-HF$ )Ni]<sub>2</sub>( $\mu$ -OH)<sub>2</sub>}Cl<sub>2</sub> (10).

![](_page_55_Figure_0.jpeg)

Figure S53. <sup>19</sup>F NMR spectrum of  $\{[(^{Ph}TBDPhos-HF)Ni]_2(\mu-OH)_2\}Cl_2$  (10).