Supporting Information

for

cis-Tetrachlorido-bis(indazole)osmium(IV) and its osmium(III) analogues: paving the way towards the *cis*-isomer of the ruthenium anticancer drugs KP1019 and/or NKP1339

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Figure S1. Atom numbering scheme for NMR resonances assignment.



Figure S2. Part of the crystal structure of $[1] \cdot Me_2CO$ showing intermolecular hydrogen bonding interactions. Parameters of hydrogen bonding interactions N1–H···O1 [N1···O1 2.740(3) Å, N1–H···O1 163.1°], N10–H···Cl4ⁱ(–x + 2, –y, –z + 1) [N10···Cl4ⁱ 3.316(2) Å, N10–H···Cl4ⁱ 137.7°].



Figure S3. Part of the crystal structure of [2]·2DMSO showing intermolecular hydrogen bonding interactions. Parameters of hydrogen bonding interactions N1–H…O1 [N1…O1 2.705(4) Å, N1–H…O1 155.6°], N10–H…O2 [N10…O2 2.677(4) Å, N10–H…O2 148.1°].



Figure S4. ¹H NMR spectrum of [1] in Me₂SO-d₆. The resonance at 6.96 ppm is due to 1,1,2,2-tetrachloroethane.



Figure S5. In situ UV-vis-NIR spectroelectrochemistry for [1] in 0.2 M $nBuN_4[PF_6]$ in DMSO (scan rate 5 mV s⁻¹, Pt-microstructured honeycomb working electrode): (a) the corresponding *in situ* cyclic voltammogram; (b) UV-vis spectra detected simultaneously during the *in situ* reduction of [1] in the region of the first cathodic peak (from 1.0 V to 0.5 V vs NHE).



Figure S6. UV-vis spectra measured during electrochemical oxidation of $(nBu_4N)[1]$ to [1] at 1.0 V vs NHE in 0.2 M $nBuN_4[PF_6]$ in DMSO (thin layer cell with Pt-microstructured honeycomb working electrode, under argon).



Figure S7. Cyclic voltammetry of [1] in 0.2 M nBuN₄[PF₆] in (a) acetonitrile and (b) in DMSO at scan rate of 10 mV s⁻¹ using thin layer cell with the Pt-microstructured honeycomb working electrode (black trace – the first voltammetric scan in the region of the first reduction step, the second voltammetric scan going to the region of the second electron transfer, green and blue traces - the next two consecutive voltammetric scans going to the region of the second electron transfer.



Figure S8. (a) UV-vis spectra monitored during the second voltammetric scan (see the corresponding CV in Figure S5 and solid trace in inset) of [1] in 0.2 M $nBuN_4[PF_6]/DMSO$ solution going to the region of the second electron transfer as well as during the next consecutive voltammetric scan (see dashed trace in inset in a). (b) Selected UV-vis spectra taken at different potentials (see the corresponding thick selected lines in (a))



Figure S9. (a) UV-vis spectra monitored during the fourth consecutive voltammetric scan (see the corresponding blue trace in Figure S5b) of [1] in 0.2 M $nBuN_4[PF_6]/DMSO$ solution going to the region of the second electron transfer. (b) The proposed reaction scheme for cathodically induced metal dechlorination upon osmium(III) to osmium(II) reduction.



Figure S10. Q-band EPR spectrum of Na[1] in water/ethylene glycol 5:1 v/v recorded at 2 K. Under these conditions, the spin system is in "rapid passage" mode and appears as an absorption lineshape. Only the g_{max} feature is observable, and appears at $g_{obs} \approx 2.6$, in agreement with X-band measurements; the g_{mid} feature is just coming into field range, while the g_{min} feature is well beyond the maximum available magnetic field (~1.45 T; $g_{obs} \approx 1.72$). Two microwave powers are used, 20 dB (~2 mW; red trace) and 30 dB (~0.2 mW; blue trace). The substantial decrease in intensity at lower power (the receiver gain has been increased to compensate for this) indicates the fast electronic relaxation of this system.



Figure S11. X-band EPR spectrum of *n*Bu₄[1] powder sample recorded at 94 K.

	IC ₅₀ [µM]			
Complex	HT-29	H446	4T1	HEK293
[1]	75.7	56.7	126.6	72.6
Na[1]	179	20.5	75.0	118.8
[3]	70.4	49.4	>300	54.7
Cisplatin	58.6	7.7	28.7	43.2

Table S1. Results of Alamar Blue assay presented as IC_{50} (μM) values obtained after 48 h treatment.