

Supplementary Information

Influence of Functional Groups on Ethylene Polymerization
Performance of Silsesquioxane-Supported Phillips-Type Catalyst

Ryuki Baba, Ashutosh Thakur, Patchanee Chammingkwan, Minoru Terano and

Toshiaki Taniike*

*Graduate School of Advanced Science and Technology, Japan Advanced Institute of
Science and Technology, 1-1 Asahidai, Nomi, Ishikawa 923-1292, Japan*

* Corresponding Author:

Tel: +81-761-51-1630; Fax: +81-761-51-1635; E-mail: taniike@jaist.ac.jp

Contents

- A. NMR spectra of modified POSS (**1a-1d**)
- B. NMR spectra of POSS-supported chromium catalysts (**2a-2d**)
- C. UV/vis spectra of POSS- and SiO₂-supported chromium catalysts (**2a-2d**, SiO₂-**b-d**)
- D. Assignment and analytical method in NMR for polyethylene

A. NMR spectra of modified POSS (**1a-1d**)

(^tBu)₇Si₇O₉(OH)₂(OSiMe₃) (**1a**)

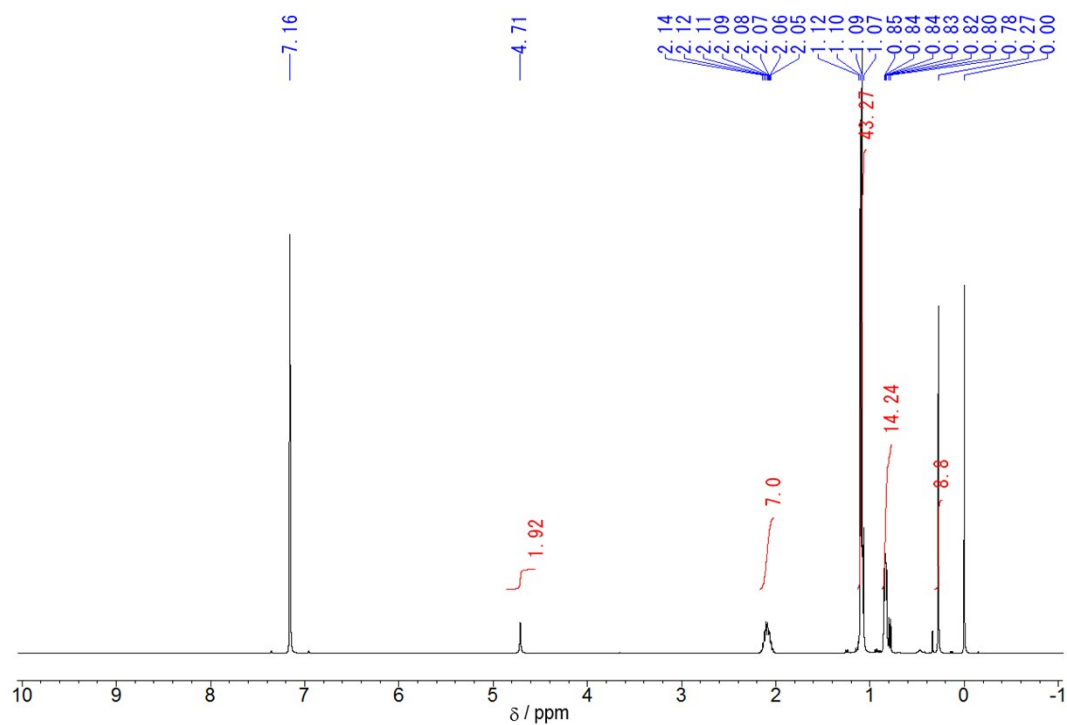


Figure S1.1. ¹H NMR spectrum of **1a** in benzene-*d*₆ at r.t. (400 MHz).

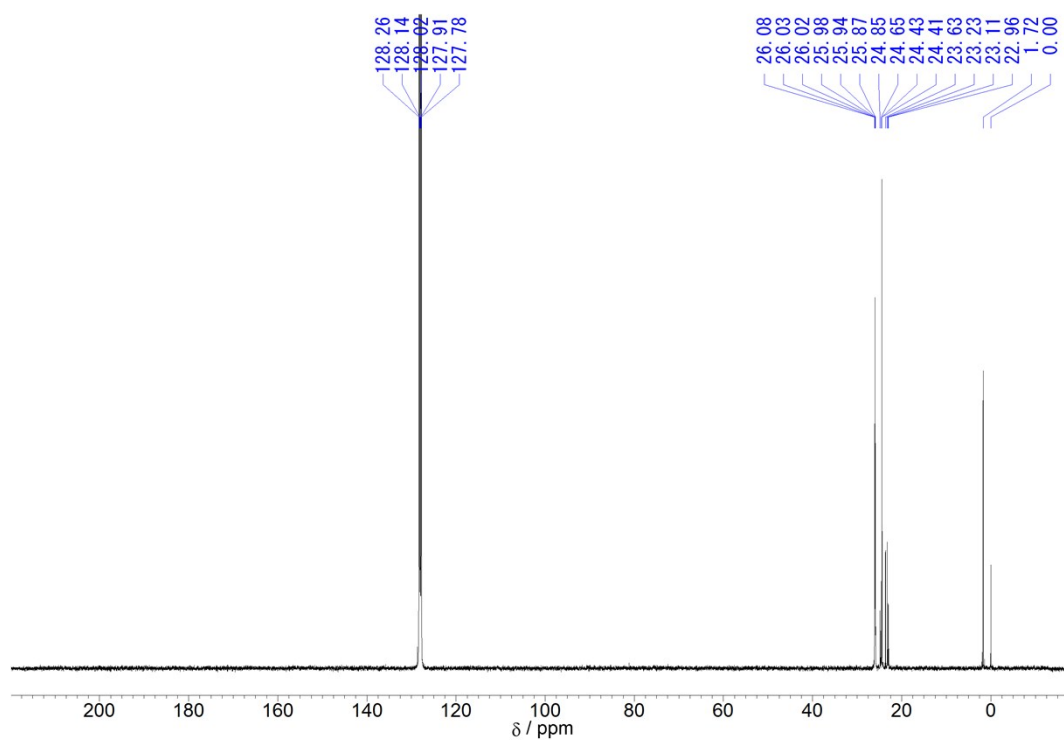


Figure S1.2. ¹³C{¹H} NMR spectrum of **1a** in benzene-*d*₆ at r.t. (100 MHz).

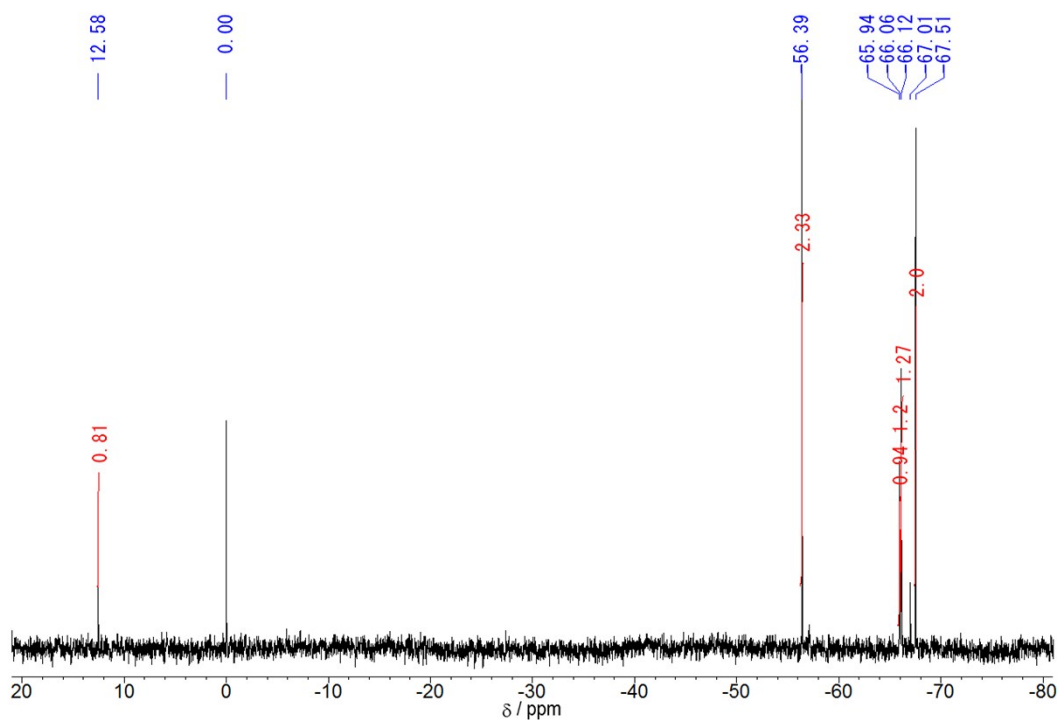


Figure S1.3. $^{29}\text{Si}\{^1\text{H}\}$ NMR spectrum of **1a** in benzene- d_6 at r.t. (79 MHz).

(*t*-Bu) $_7\text{Si}_7\text{O}_9(\text{OH})_2(\text{OSiMe}_2\text{Ph})$ (**1b**)

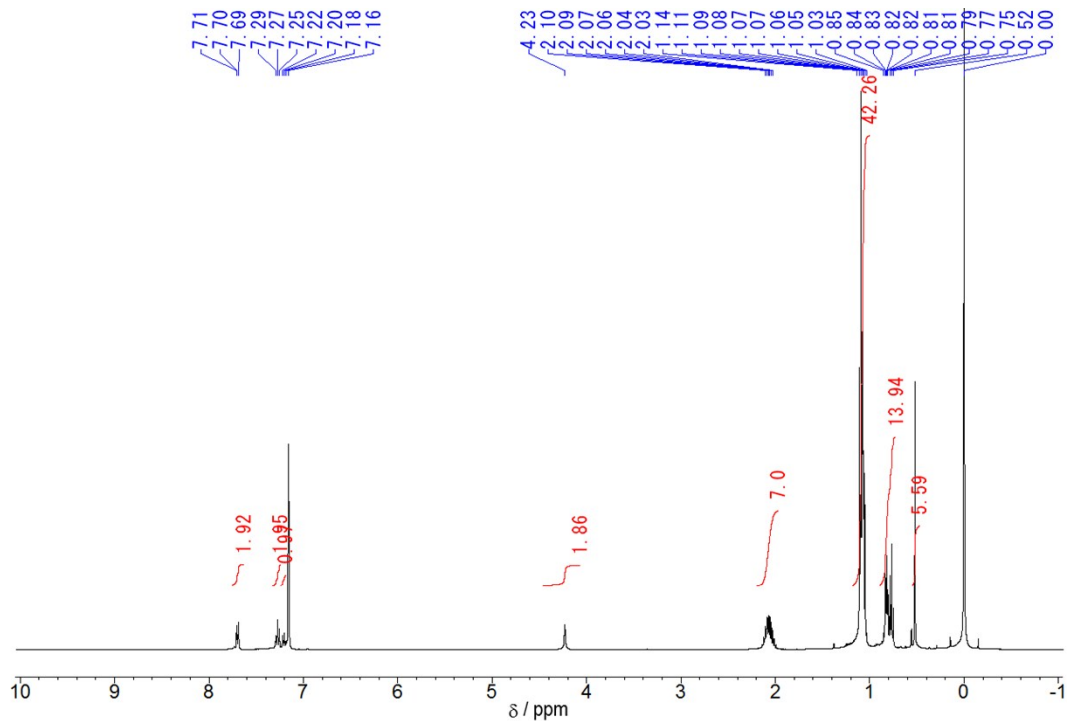


Figure S2.1. ^1H NMR spectrum of **1b** in benzene- d_6 at r.t. (400 MHz).

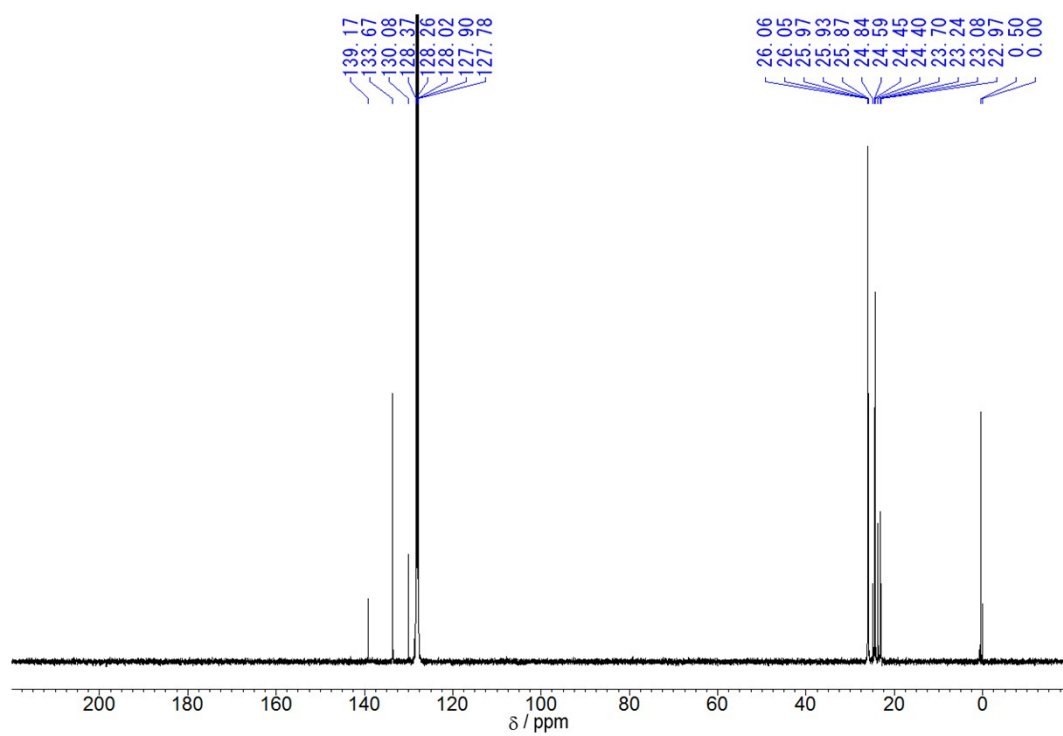


Figure S2.2. $^{13}\text{C}\{^1\text{H}\}$ NMR spectrum of **1b** in benzene- d_6 at r.t. (100 MHz).

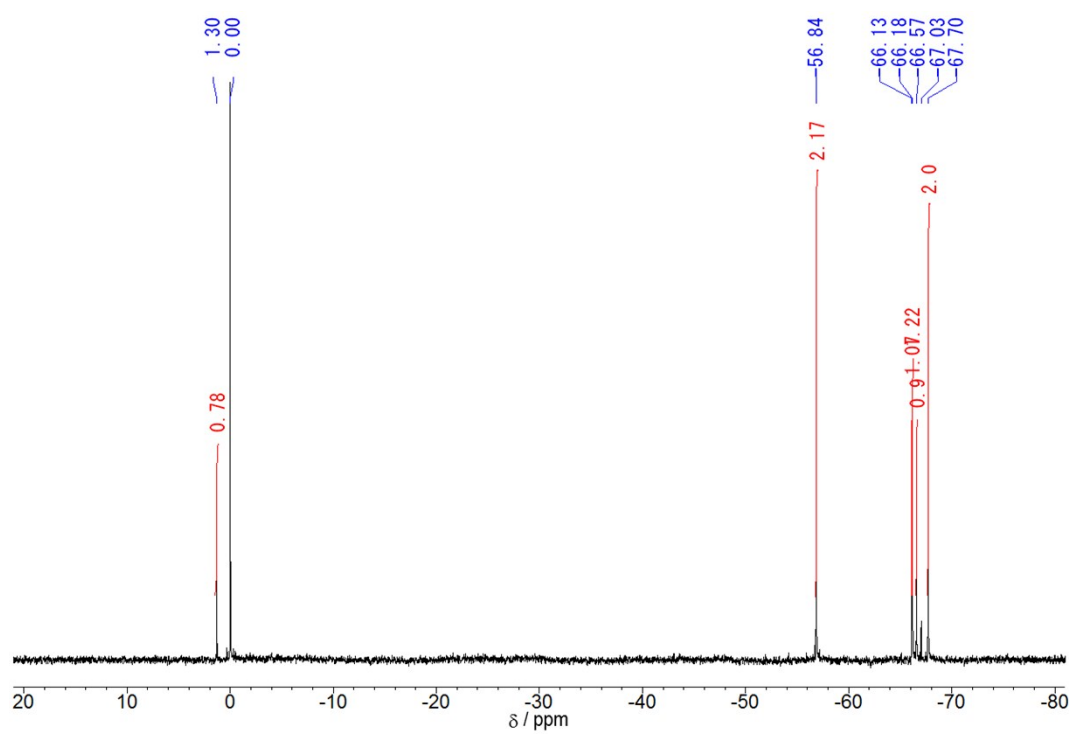


Figure S2.3. $^{29}\text{Si}\{^1\text{H}\}$ NMR spectrum of **1b** in benzene- d_6 at r.t. (79 MHz).

(^tBu)₇Si₇O₉(OH)₂[OSiMe₂C₆H₄(OMe-2)] (**1c**)

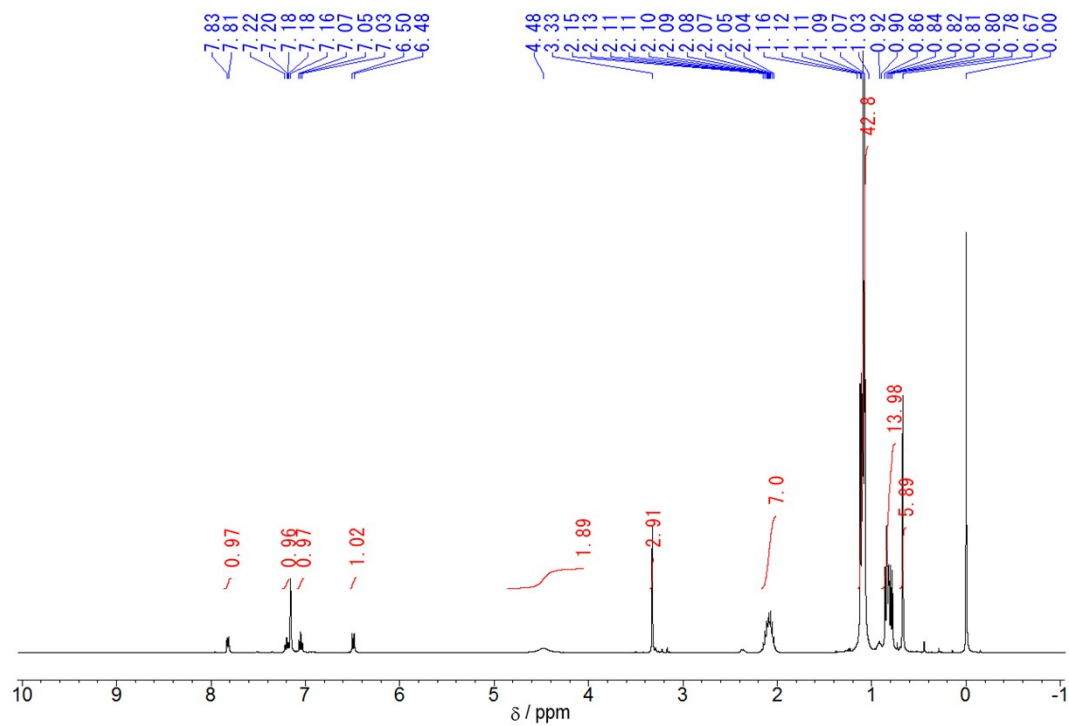


Figure S3.1. ¹H NMR spectrum of **1c** in benzene-*d*₆ at r.t. (400 MHz).

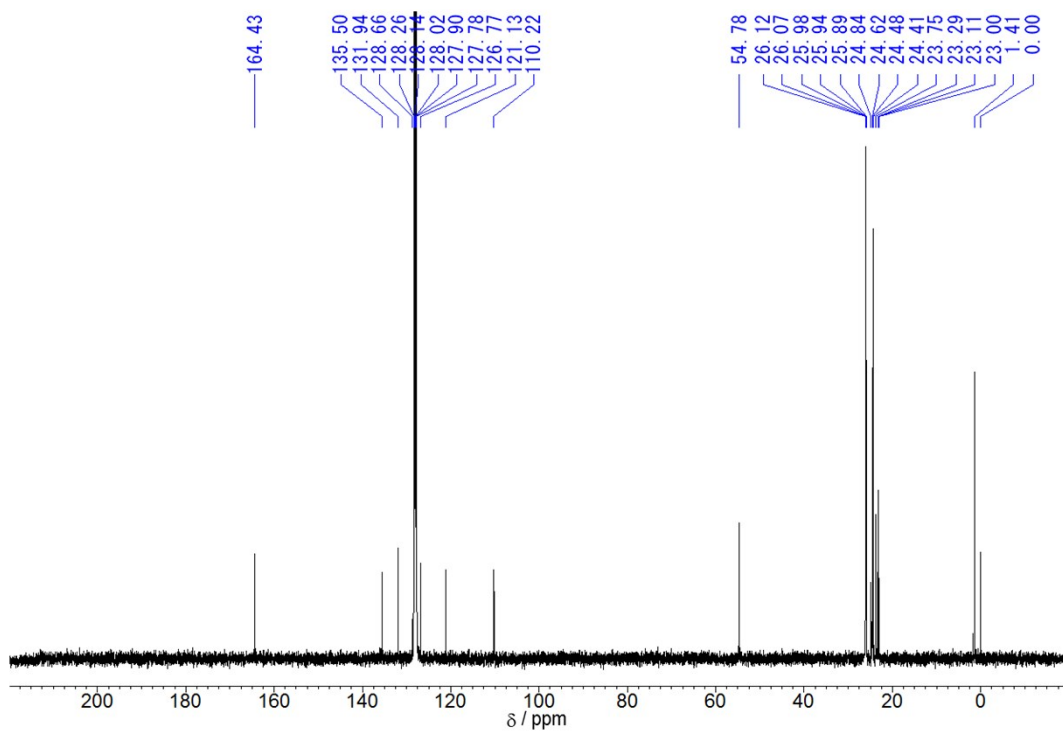


Figure S3.2. $^{13}\text{C}\{^1\text{H}\}$ NMR spectrum of **1c** in benzene- d_6 at r.t. (100 MHz).

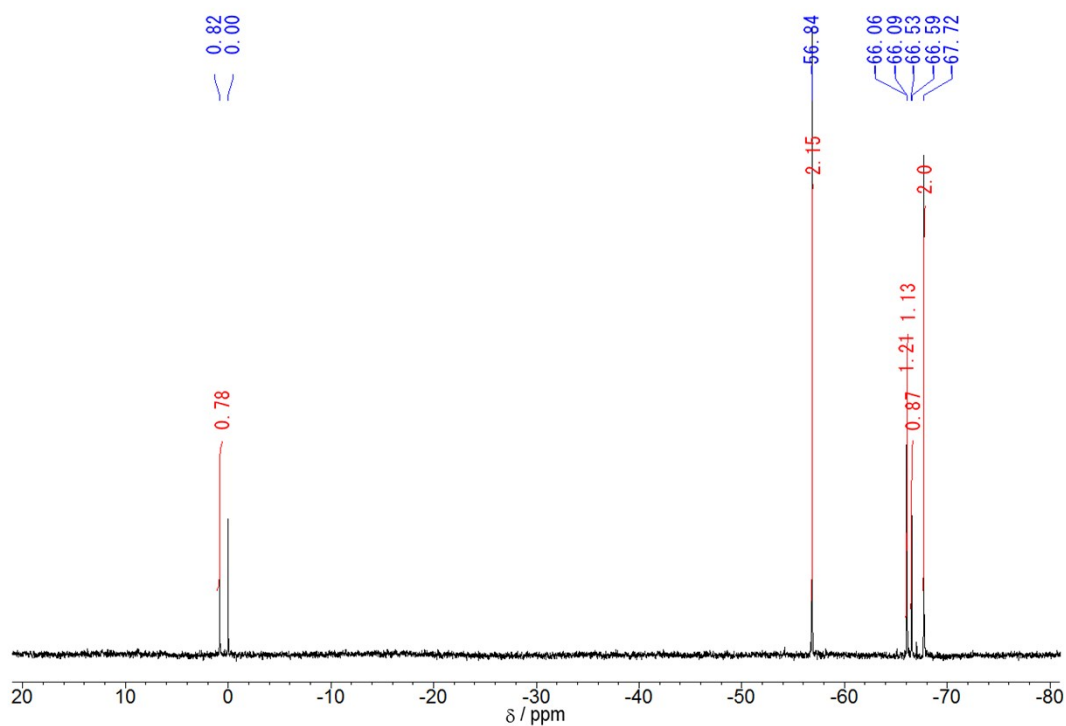


Figure S3.3. $^{29}\text{Si}\{^1\text{H}\}$ NMR spectrum of **1c** in benzene- d_6 at r.t. (79 MHz).

$(^t\text{Bu})_7\text{Si}_7\text{O}_9(\text{OH})_2[\text{OSiMe}_2\text{C}_6\text{H}_4(\text{PPh}_2-2)]$ (**1d**)

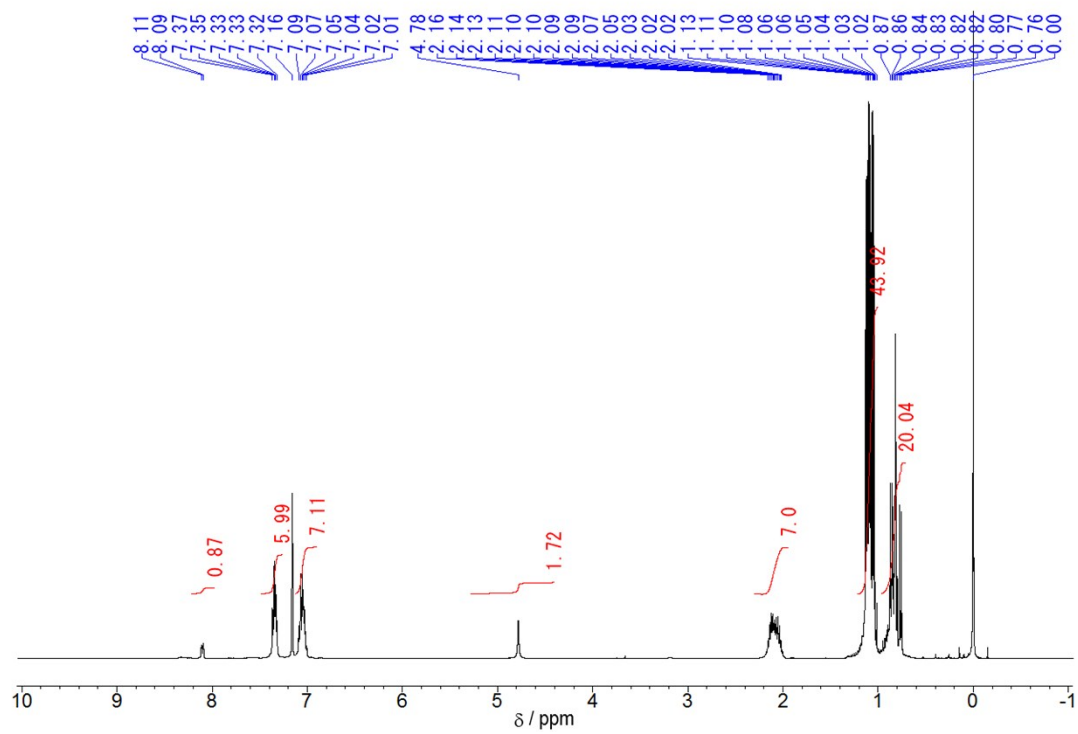


Figure S4.1. ^1H NMR spectrum of **1d** in benzene- d_6 at r.t. (400 MHz).

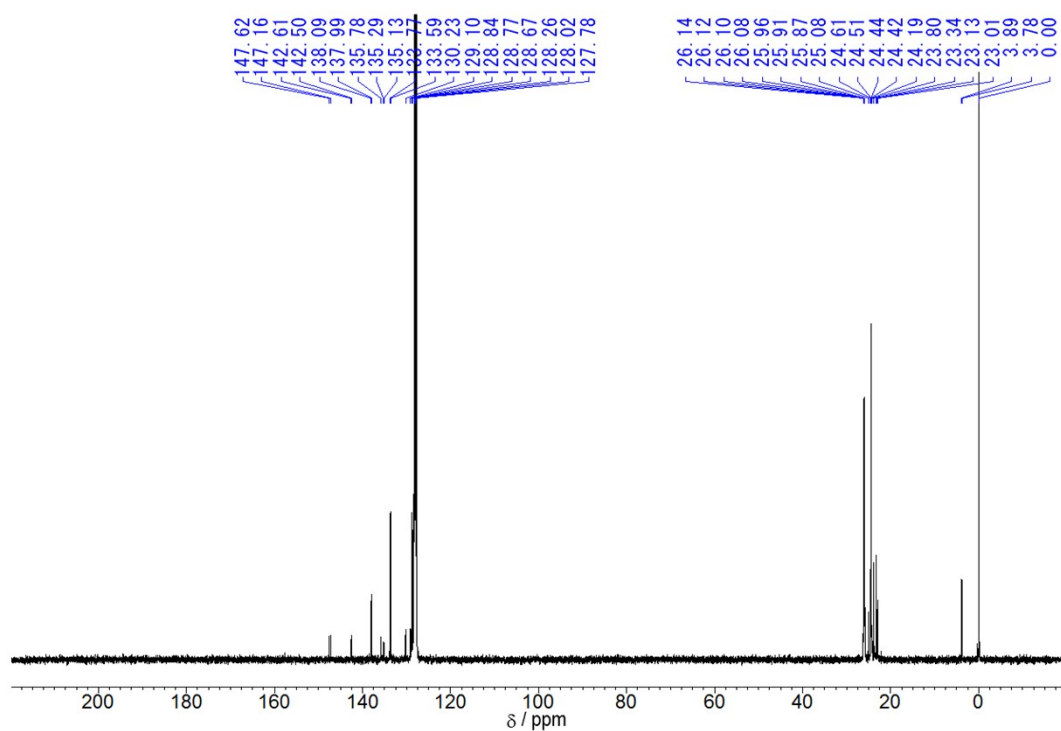


Figure S4.2. $^{13}\text{C}\{^1\text{H}\}$ NMR spectrum of **1d** in benzene- d_6 at r.t. (100 MHz).

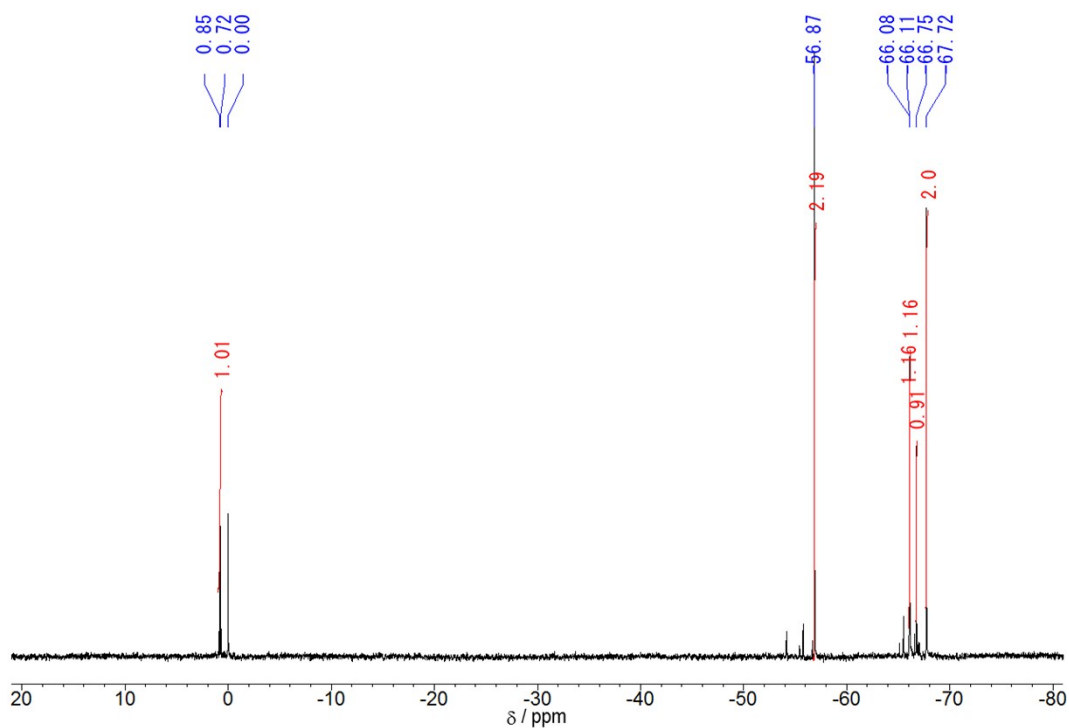


Figure S4.3. $^{29}\text{Si}\{^1\text{H}\}$ NMR spectrum of **1d** in benzene- d_6 at r.t. (79 MHz).

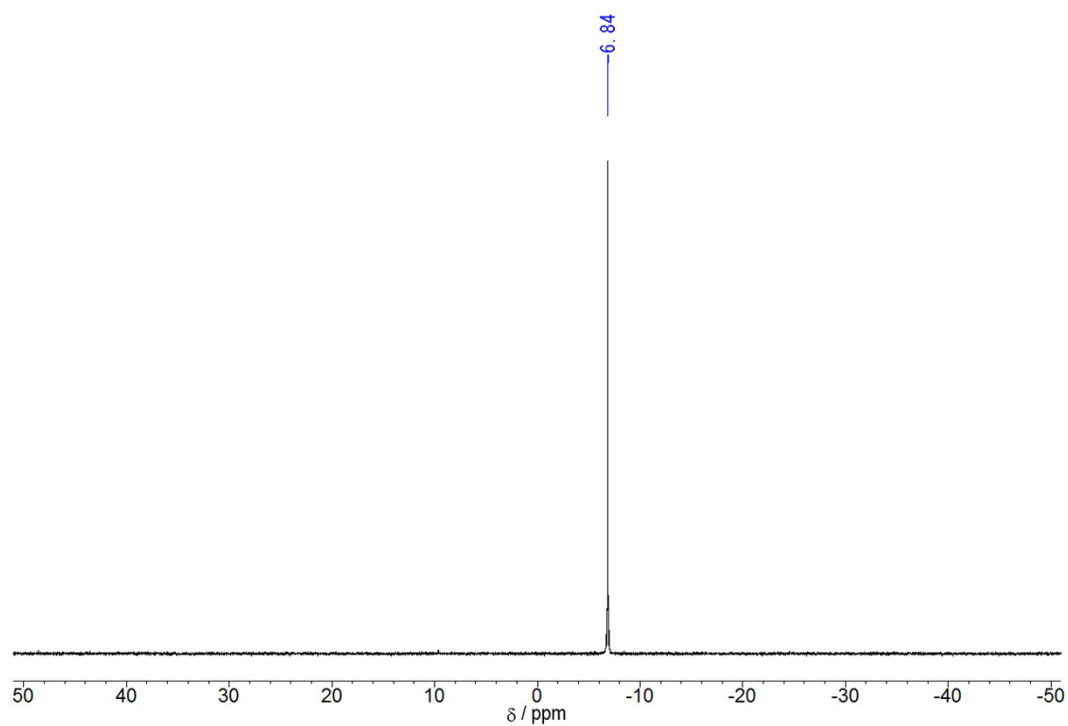


Figure S4.4. $^{31}\text{P}\{^1\text{H}\}$ NMR spectrum of **1d** in benzene- d_6 at r.t. (162 MHz).

B. NMR spectra of POSS supported chromium catalysts (**2a-2d**)

$[(^i\text{Bu})_7\text{Si}_7\text{O}_{11}(\text{OSiMe}_3)]\text{CrCH}(\text{SiMe}_3)_2$ (**2a**)

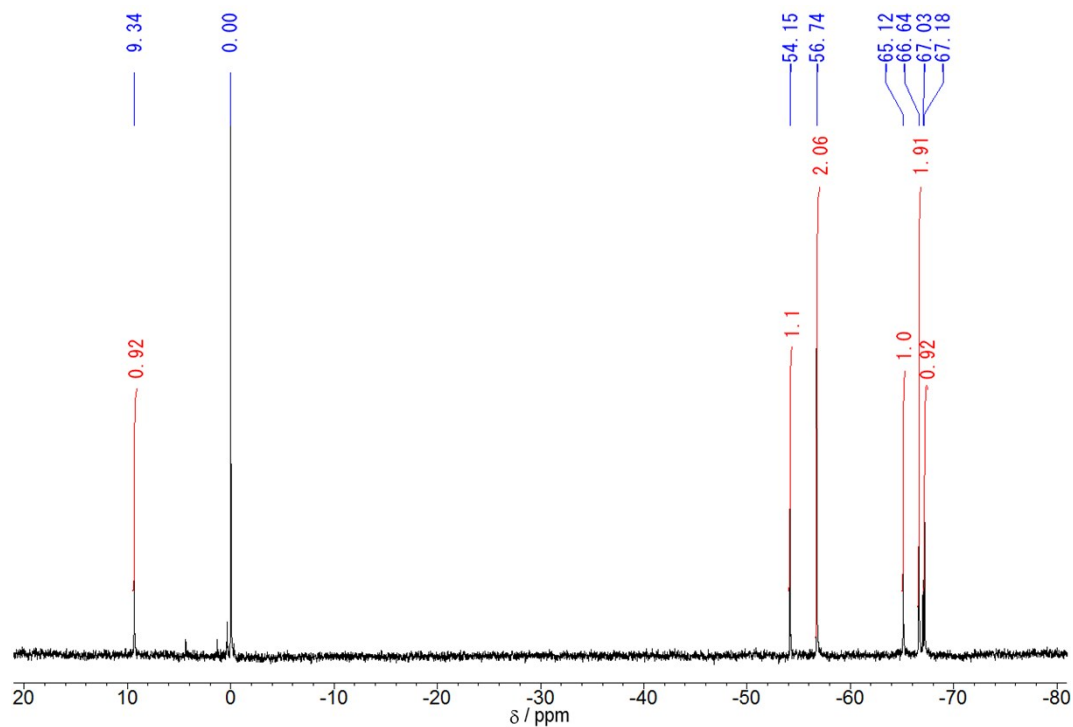


Figure S5.1. $^{29}\text{Si}\{^1\text{H}\}$ NMR spectrum of **2a** in benzene- d_6 at r.t. (79 MHz).

$(^i\text{Bu})_7\text{Si}_7\text{O}_{11}(\text{OSiMe}_2\text{Ph})]\text{CrCH}(\text{SiMe}_3)_2$ (**2b**)

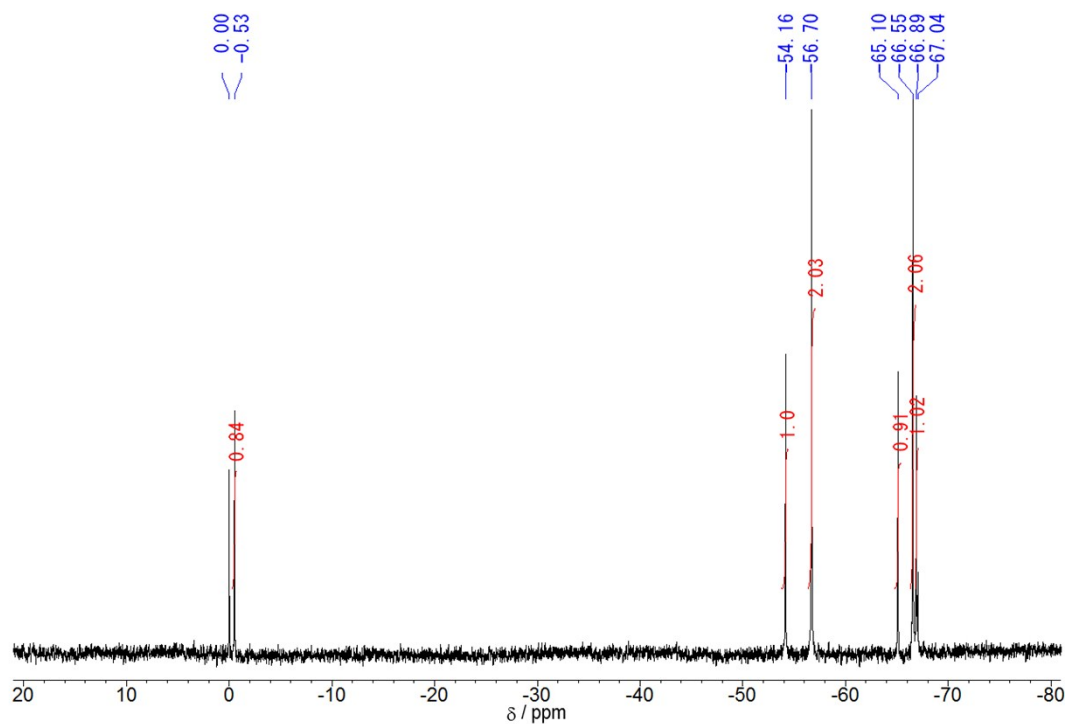


Figure S6. $^{29}\text{Si}\{^1\text{H}\}$ NMR spectrum of **2b** in benzene- d_6 at r.t. (79 MHz).

$\{(^i\text{Bu})_7\text{Si}_7\text{O}_{11}[\text{OSiMe}_2\text{C}_6\text{H}_4(\text{OMe-2})]\}_2\text{CrCH}(\text{SiMe}_3)_2$ (**2c**)

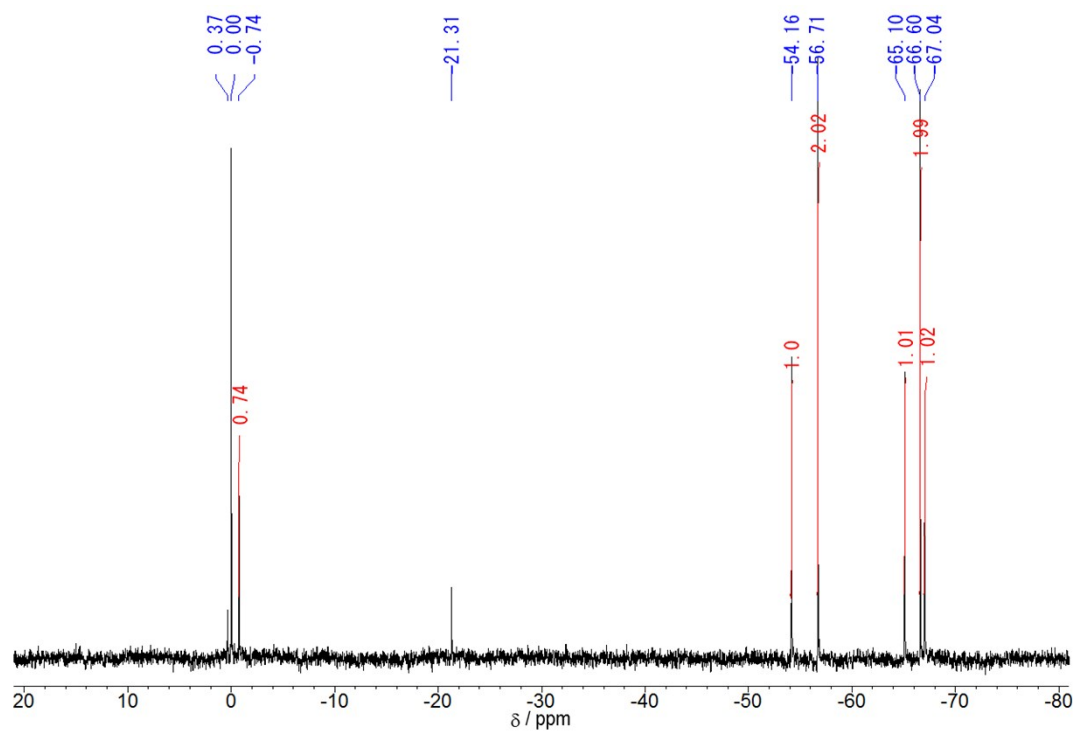


Figure S7. $^{29}\text{Si}\{^1\text{H}\}$ NMR spectrum of **2c** in benzene- d_6 at r.t. (79 MHz).

$\{(^i\text{Bu})_7\text{Si}_7\text{O}_{11}[\text{OSiMe}_2\text{C}_6\text{H}_4(\text{PPh}_2\text{-2})]\}_2\text{CrCH}(\text{SiMe}_3)_2$ (**2d**)

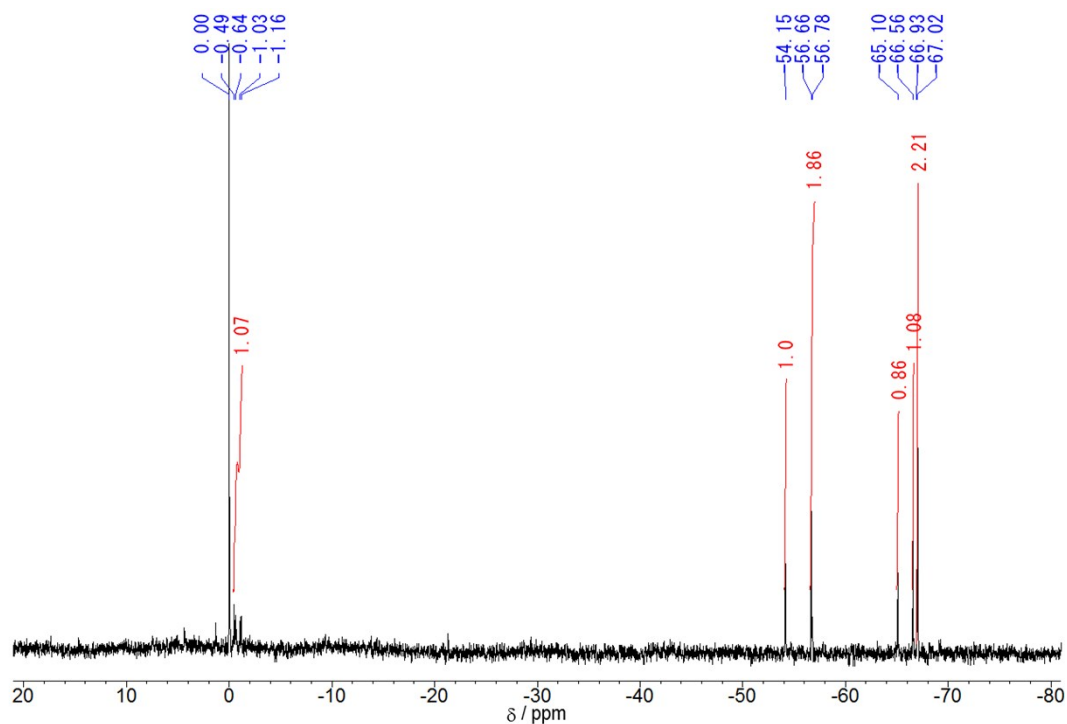


Figure S8.1. $^{29}\text{Si}\{^1\text{H}\}$ NMR spectrum of **2d** in benzene- d_6 at r.t. (79 MHz).

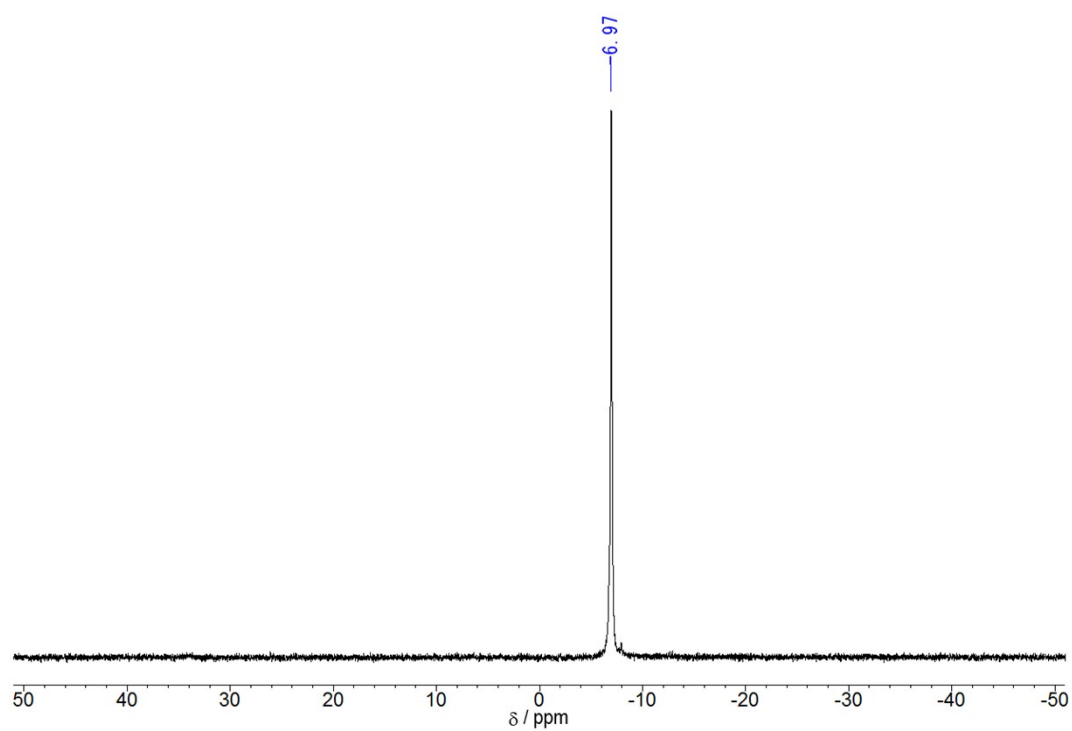


Figure S8.2. ^{31}P NMR spectrum of **2d** in benzene- d_6 at r.t. (162 MHz).

C. UV/vis spectra measurement

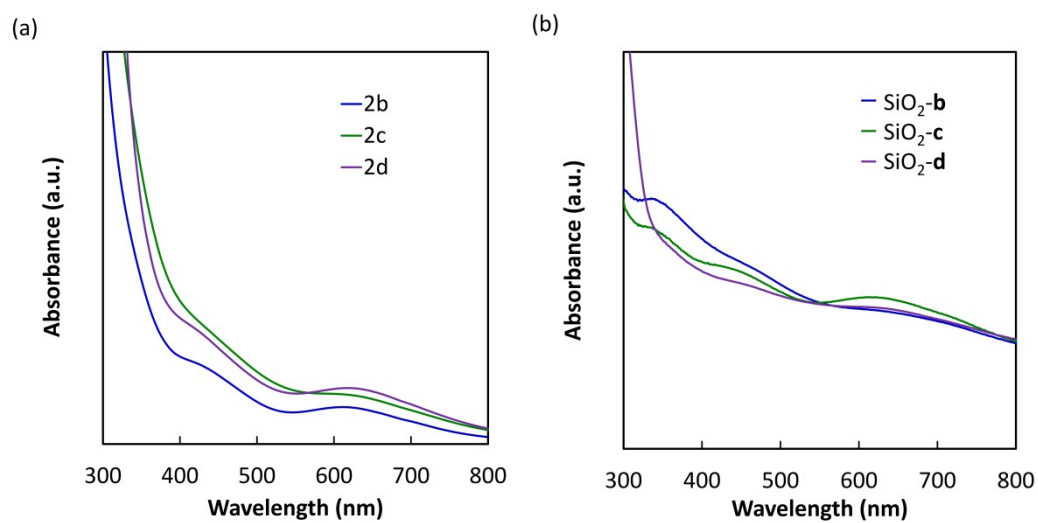


Figure S9. UV/vis (DRS) spectra of a) POSS-supported catalysts (**2a-2d**) and b) SiO₂-supported catalysts (SiO₂-**b-d**).

D. Assignment and analytical method in NMR for polyethylene

Figure S10 reports typical $^{13}\text{C}\{^1\text{H}\}$ and ^1H NMR spectra of the obtained PE (entries 3 and 5 in **Table 1**). The chemical shift was referenced to methyl carbon (1.98 ppm) and methyl proton (0.09 ppm) of hexamethyldisiloxane (HMDS) in $^{13}\text{C}\{^1\text{H}\}$ and ^1H NMR, respectively. The peak assignments have been done based on Refs. [1] and [2]. The fractions of methyl branches and the saturated ends were determined from $^{13}\text{C}\{^1\text{H}\}$ NMR based on the following equations.

$$\text{Eq. (1)} \quad \text{Fraction of methyl branch (/1000C)} = I_{B1}/I_{\text{totalC}} \times 1000$$

$$\text{Eq. (2)} \quad \text{Fraction of saturated end (/1000C)} = I_S/I_{\text{totalC}} \times 1000$$

$$\text{Eq. (3)} \quad I_{B1} = (I_{1B1} + I_{brB1} + I_{\alpha B1})/4$$

$$\text{Eq. (4)} \quad I_S = (I_{1S} + I_{2S} + I_{3S})/3$$

$$\text{Eq. (5)} \quad I_{\text{totalC}} = I_{\text{Main chain}}$$

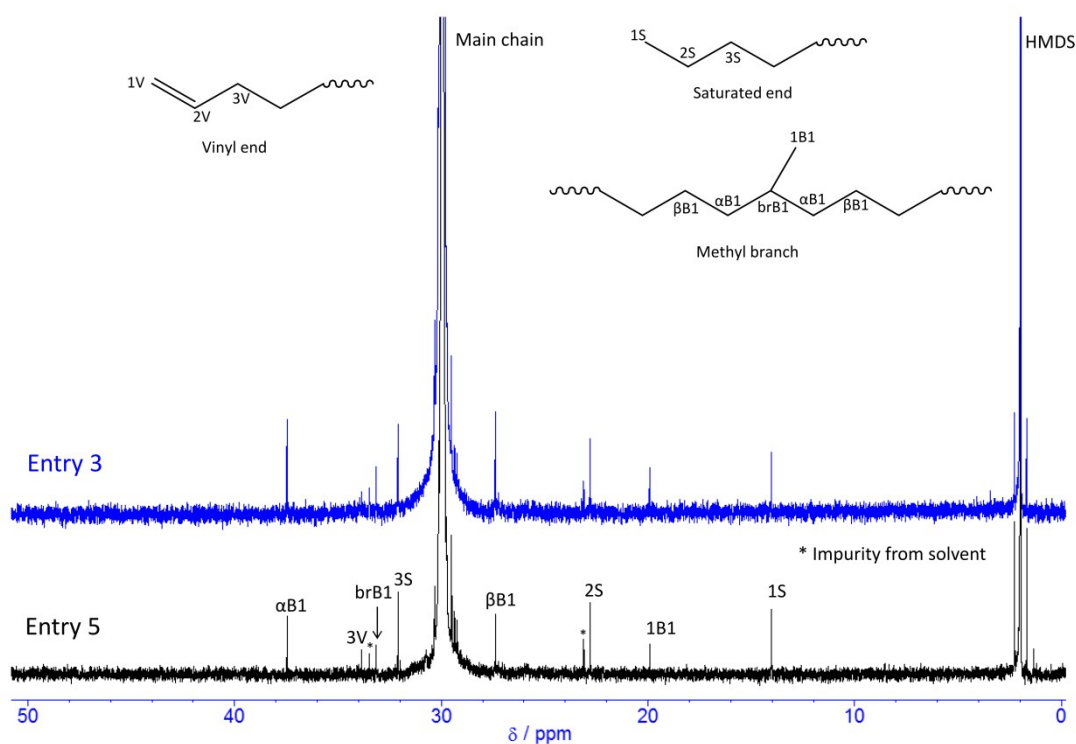


Figure S10.1. $^{13}\text{C}\{^1\text{H}\}$ NMR spectra of typical PE (100 MHz).

The fraction of the vinyl ends were determined from ^1H NMR based on the following equations.

Eq. (6) $\text{Fraction of vinyl end } (/1000\text{C}) = I_{Vi}/I_{total} \times 1000$

Eq. (7) $I_{Vi} = (I_{1V} + I_{2V})/3$

Eq. (8) $I_{total} = (I_{Main\ chain})/2$

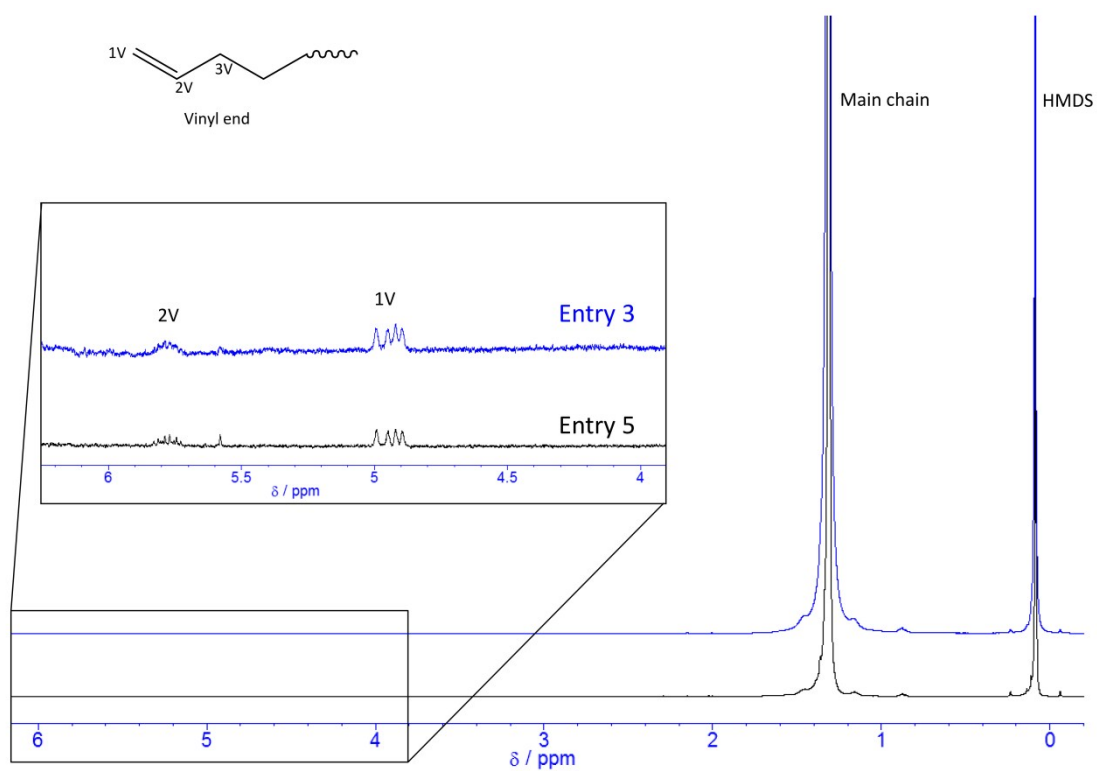


Figure S10.2. ^1H NMR spectra of typical PE (400 MHz).

References

- [1] B. Qu, X. Qu, Y. Xu, U. Jacobsson, B. Rånby, K. E. Russell and W. E. Baker, *Macromolecules*, 1997, **30**, 1408.
- [2] E. W. Hansen, R. Blom and O. M. Bade, *Polymer*, 1997, **38**, 4295.