

## Supporting Information

### Synthesis of amine-bridged bis(phenolate) rare-earth metal aryloxides and their catalytic performances for the ring-opening polymerization of $L$ -lactic acid $O$ -carboxyanhydride and $L$ -lactide

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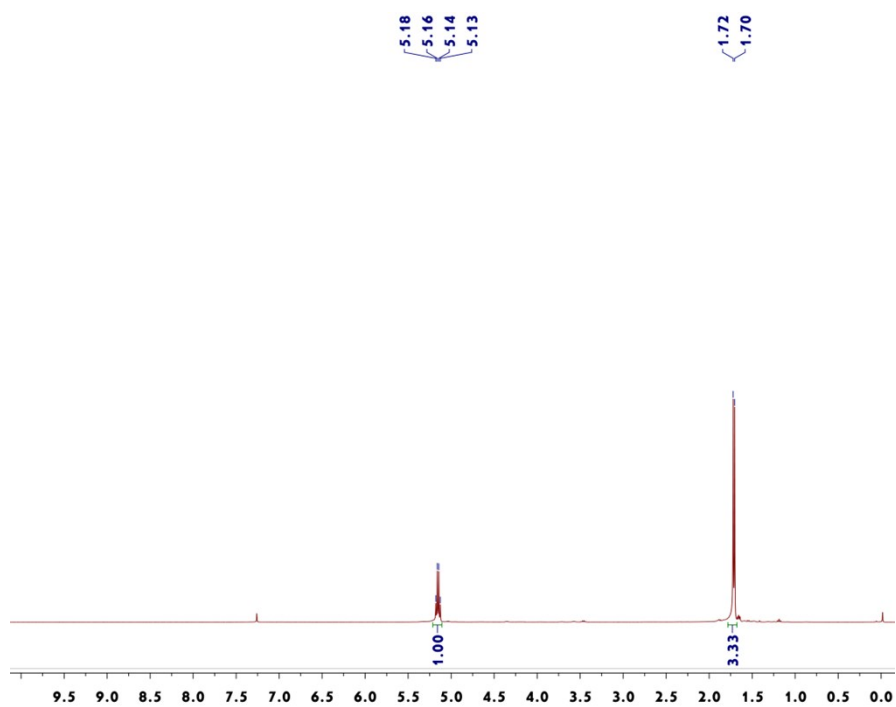
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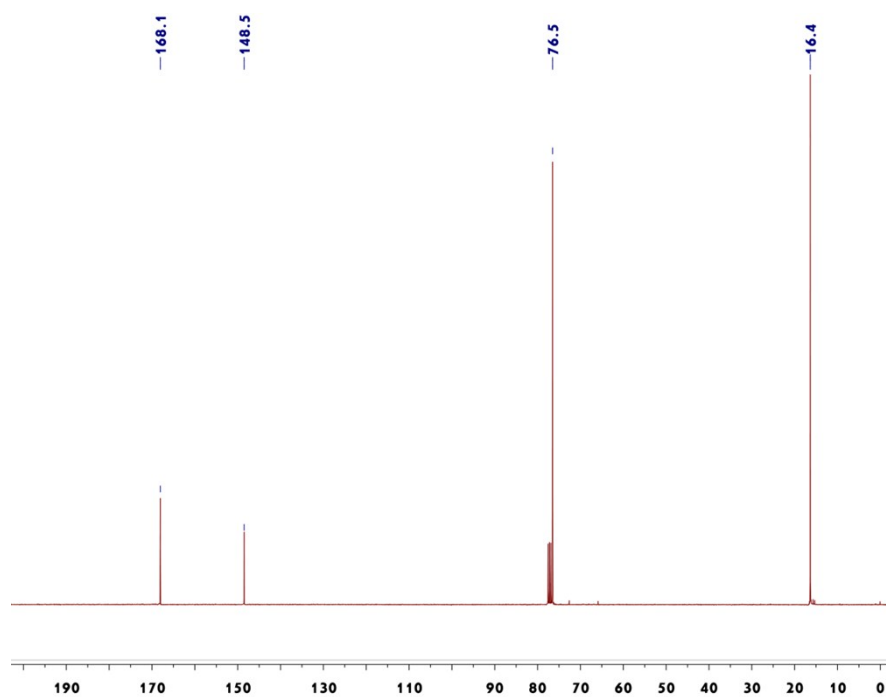
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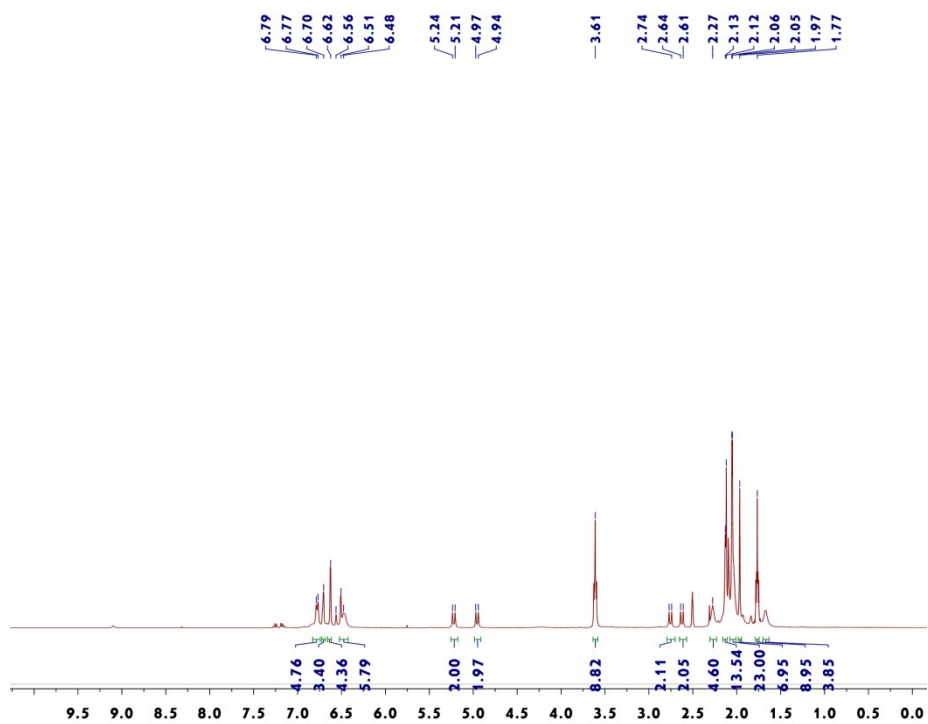
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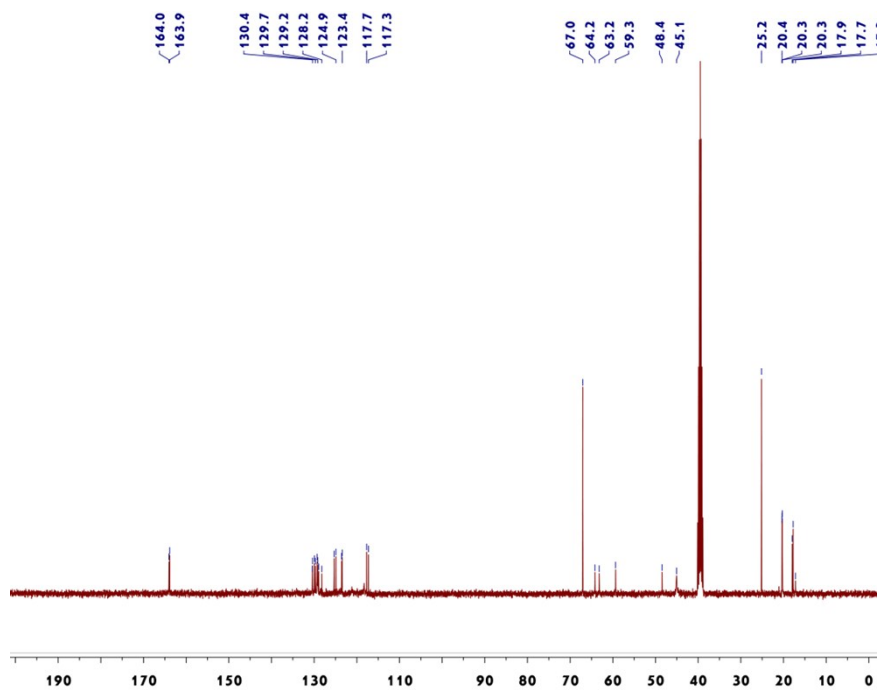
**Fig. S1**  $^1\text{H}$  NMR spectrum ( $\text{CDCl}_3$ , 25  $^\circ\text{C}$ ) of  $\text{L-lacOCA}$



**Fig. S2**  $^{13}\text{C}$  NMR spectrum ( $\text{CDCl}_3$ , 25  $^\circ\text{C}$ ) of  $\text{L-lacOCA}$



**Fig. S3**  $^1\text{H}$  NMR spectrum (DMSO, 25 °C) of complex **1**



**Fig. S4**  $^{13}\text{C}$  NMR spectrum (DMSO, 25 °C) of complex **1**

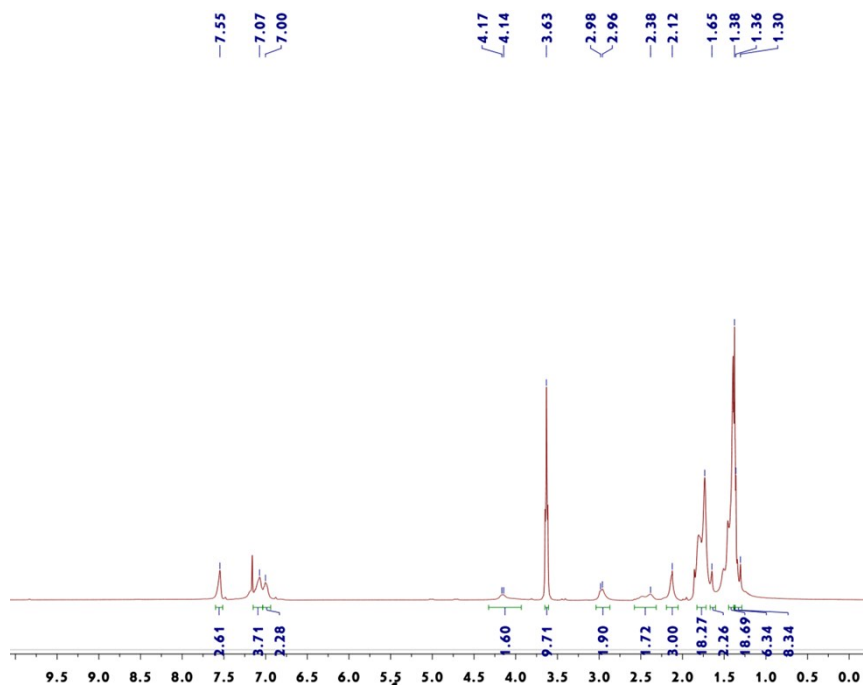


Fig. S5  $^1\text{H}$  NMR spectrum ( $\text{C}_6\text{D}_6$ , 25 °C) of complex 2

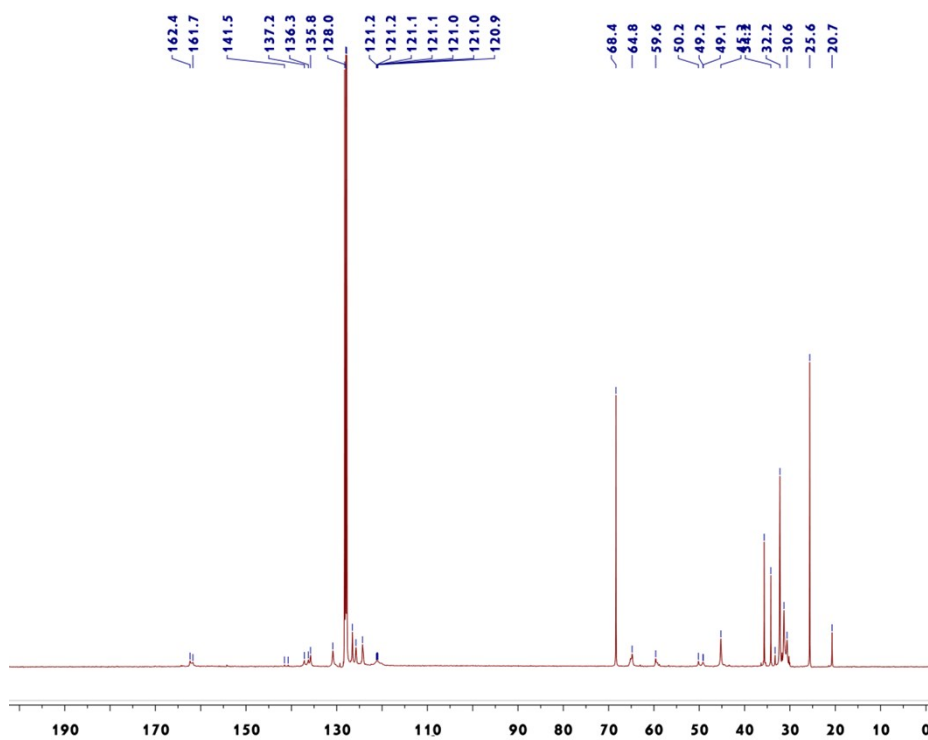


Fig. S6  $^{13}\text{C}$  NMR spectrum ( $\text{C}_6\text{D}_6$ , 25 °C) of complex 2

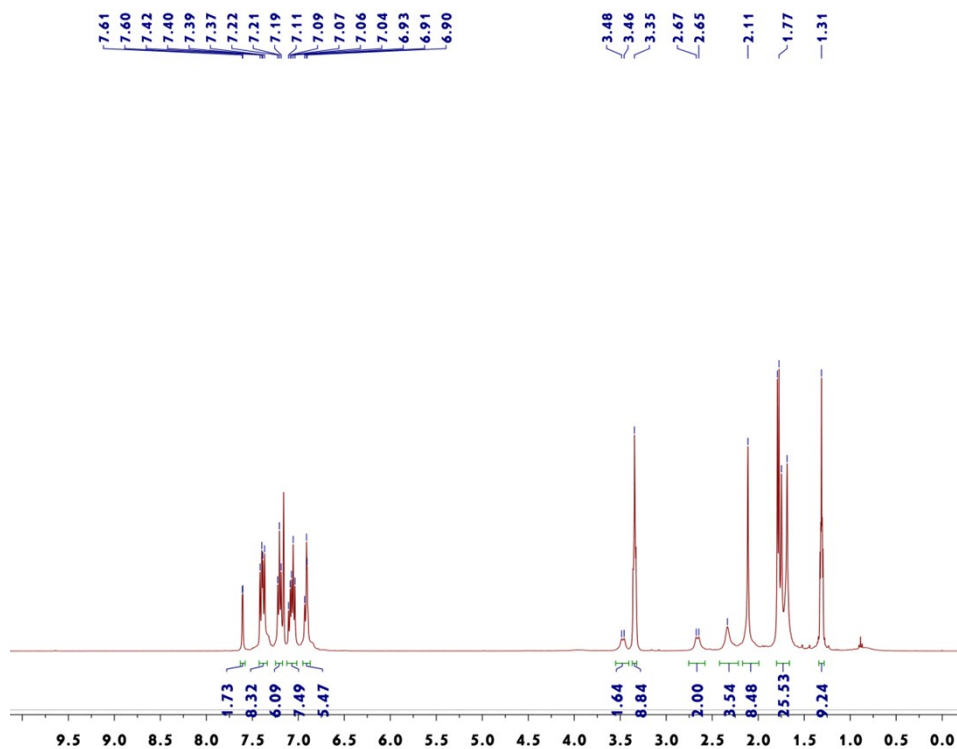


Fig. S7  $^1\text{H}$  NMR spectrum ( $\text{C}_6\text{D}_6$ , 25  $^\circ\text{C}$ ) of complex **3**

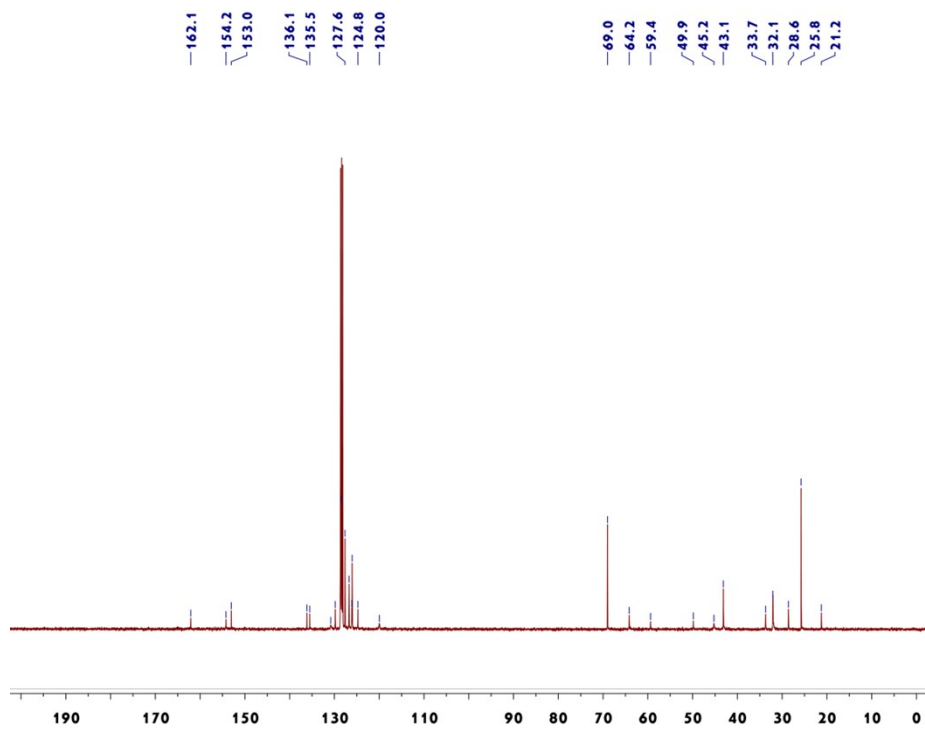
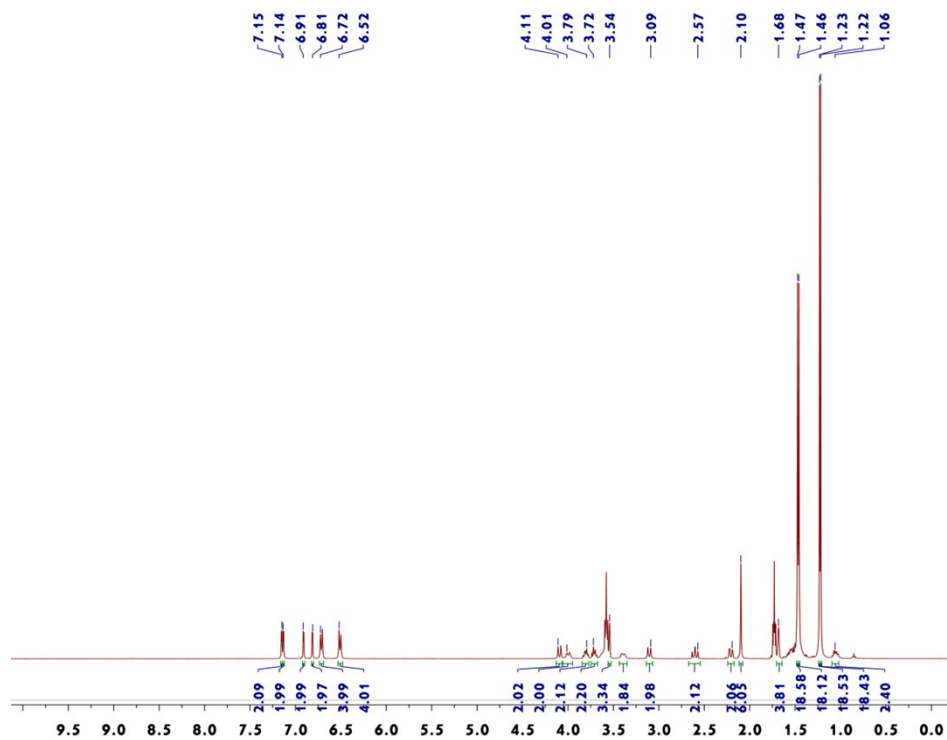
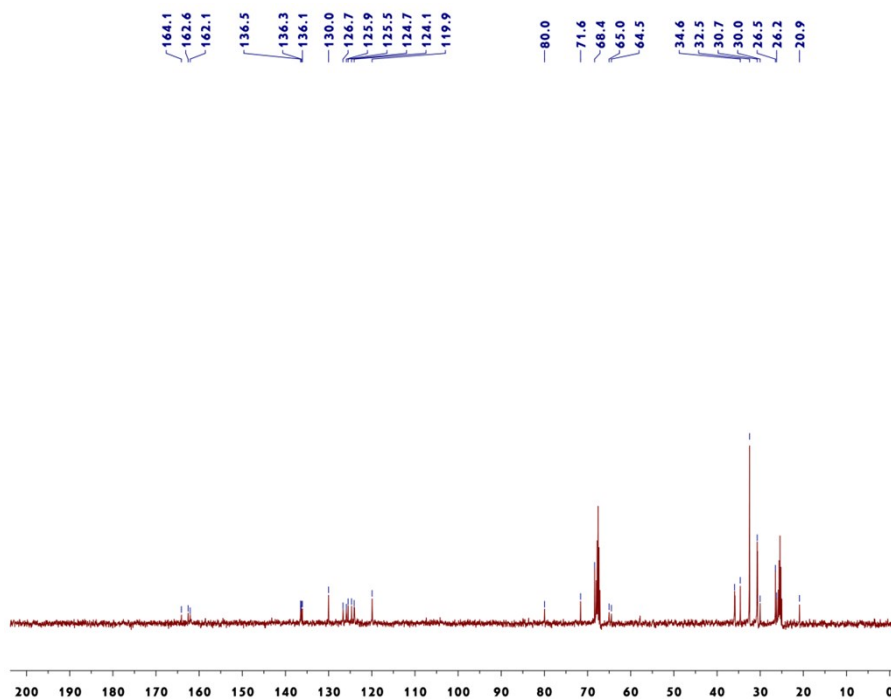


Fig. S8  $^{13}\text{C}$  NMR spectrum ( $\text{C}_6\text{D}_6$ , 25  $^\circ\text{C}$ ) of complex **3**



**Fig. S9**  $^1\text{H}$  NMR spectrum ( $d_8$ -THF, 25 °C) of complex **5**



**Fig. S10**  $^{13}\text{C}$  NMR spectrum ( $d_8$ -THF, 25 °C) of complex **5**

**Table S1.** Crystallographic data for complexes **1**, **2**, **3** and **5**

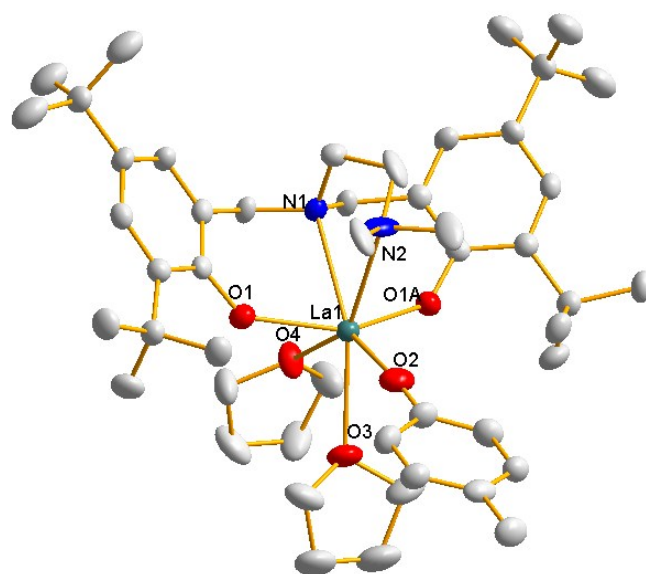
Compound	<b>1</b>	<b>2</b>	<b>3</b>	<b>5·3THF</b>
Formula	C <sub>74</sub> H <sub>106</sub> La <sub>2</sub> N <sub>4</sub> O <sub>10</sub>	C <sub>49</sub> H <sub>77</sub> N <sub>2</sub> O <sub>5</sub> L a	C <sub>69</sub> H <sub>85</sub> LaN <sub>2</sub> O <sub>5</sub>	C <sub>180</sub> H <sub>264</sub> N <sub>4</sub> O <sub>19</sub> Y <sub>4</sub>
fw	1489.45	913.03	1161.29	3143.58
<i>T</i> /K	293(2)	100(2)	293(2)	223(2)
Crystal system	triclinic	orthorhombic	monoclinic	triclinic
Crystal size/mm	0.60 x 0.40 x 0.20	0.35 × 0.30 × 0.20	0.75 x 0.6 x 0.4	0.80 x 0.70 x 0.30
Space group	P-1	Pnma	P 1	P-1
<i>a</i> /Å	12.4210(6)	21.8235(10)	15.1660(3)	15.5796(11)
<i>b</i> /Å	12.9473(5)	20.4867(8)	33.8032(8)	18.2982(14)
<i>c</i> /Å	13.3644(7)	10.6023(5)	24.5524(6)	18.7574(18)
<i>α</i> /deg	114.347(4)	90	90	75.610(15)
<i>β</i> /deg	109.435(5)	90	101.496(2)	70.403(14)
<i>γ</i> /deg	95.310(4)	90	90	66.585(13)
<i>V</i> /Å <sup>3</sup>	1779.31(16)	4740.2(4)	12334.6(5)	4581.5(9)
<i>Z</i>	1	4	8	1
<i>D</i> <sub>calcd</sub> /g cm <sup>-3</sup>	1.390	1.279	1.251	1.139
<i>μ</i> /mm <sup>-1</sup>	1.243	0.946	0.743	1.312
<i>F</i> (000)	772	1928	4880	1680
<i>θ</i> <sub>max</sub> /deg	26.37	27.51	26.37	26.37
Collected	17526	42156	74083	40070
Unique reflns	7289	5581	25165	18548
Obsd reflns [ <i>I</i> > 2.0σ( <i>I</i> )]	6495	4642	19002	11014
No. of variables	371	295	1471	951
GOF	1.052	1.072	1.060	1.071
<i>R</i>	0.0368	0.0266	0.0447	0.0804
w <i>R</i>	0.0875	0.0553	0.0936	0.2251
<i>R</i> <sub>int</sub>	0.0316	0.0508	0.0414	0.0675
Largest diff. peak, hole/e Å <sup>-3</sup>	1.573,- 1.712	0.875, -0.399	0.962,- 0.755	1.480, - 1.358



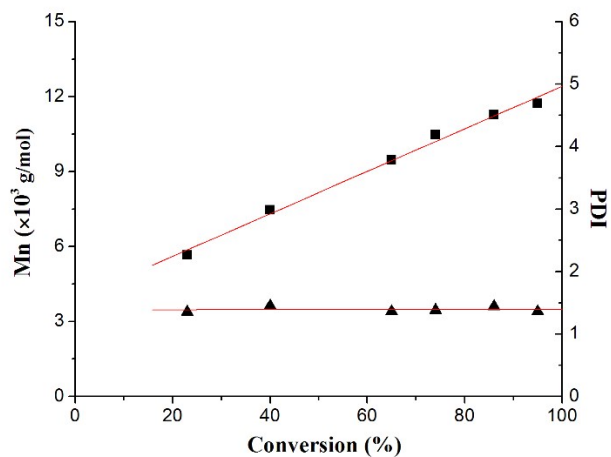
**Table S2.** Selected bond lengths (Å) and bond angles (deg) for complexes **1-3 and 5**.

<b>Bond lengths</b>	<b>1</b>	<b>Bond lengths</b>	<b>2</b>
Ln1-O1	2.465(2)	Ln1-O1	2.285(13)
Ln1-O1A	2.435(2)	Ln1-O1A	2.285(13)
Ln1-O2	2.313(2)	Ln1-O2	2.284(2)
Ln1-O3	2.289(3)	Ln1-O3	2.608(2)
Ln1-O4	2.682(3)	Ln1-O4	2.646(2)
Ln1-N1	2.848(3)	Ln1-N1	2.754(2)
Ln1-N2	2.730(3)	Ln1-N2	2.755(3)
<b>Bond angles</b>		<b>Bond angles</b>	
O2-Ln1-N1	71.20(8)	O1-Ln1-O2	106.74(4)
N2-Ln1-N1	63.97(9)	O2-Ln1-O1A	106.74(3)
O1A-Ln1-N1	106.15(8)	O1-Ln1-O1A	145.52(7)
O3-Ln1-N1	131.12(9)	O2-Ln1-N1	140.88(8)
O3-Ln1-O1	86.79(9)	O1-Ln1-N1	74.48(4)
O3-Ln1-O2	138.69(10)	O2-Ln1-N2	76.84(8)
O3-Ln1-O1A	105.72(9)	O1-Ln1-N2	89.85(4)
O2-Ln1-O1A	97.01(8)	N1-Ln1-N2	64.04(7)
O1A-Ln1-O1	67.67(8)		

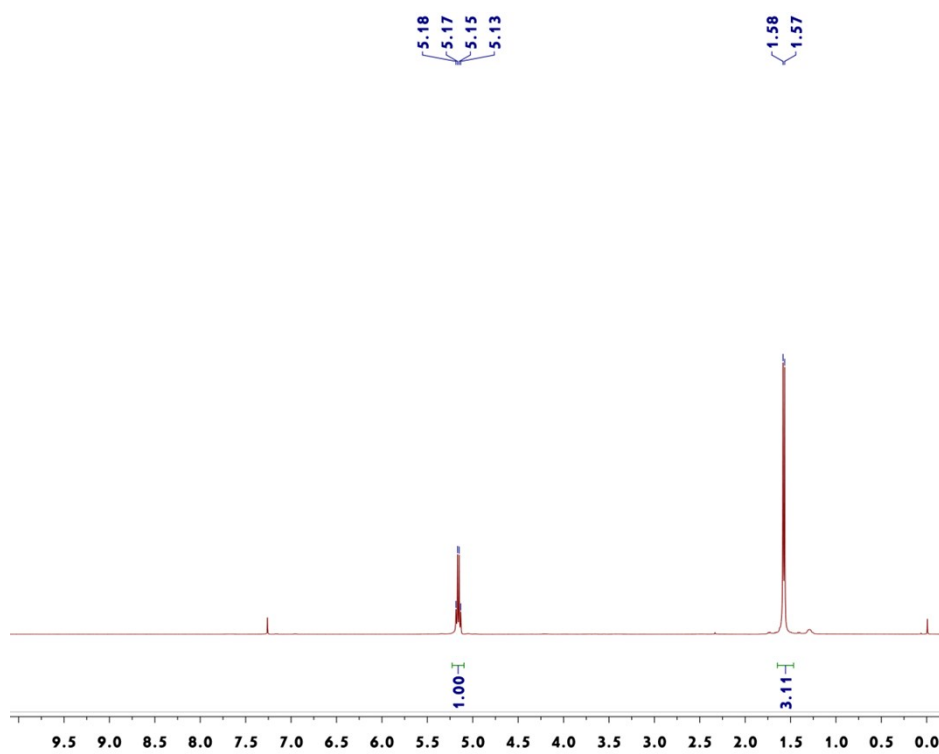
<b>Bond lengths</b>	<b>3</b>	<b>Bond lengths</b>	<b>5</b>
Ln1-O1	2.301(2)	Ln1-O1	2.136(4)
Ln1-O2	2.269(2)	Ln1-O2	2.128(4)
Ln1-O3	2.270(3)	Ln1-O3	2.339(4)
Ln1-O4	2.679(3)	Ln1-O3A	2.263(4)
Ln1-O5	2.619(3)	Ln1-O4	2.365(4)
Ln1-N1	2.783(2)	Ln1-N1	2.542(5)
Ln1-N2	2.809(3)		
<b>Bond angles</b>		<b>Bond angles</b>	
O2-Ln1-N1	74.26(7)	O2-Ln1-N1	76.66(15)
N1-Ln1-N2	63.46(8)	O3-Ln1-N1	127.13(14)
O3-Ln1-N1	139.95(9)	O3A-Ln1-N1	163.70(15)
O4-Ln1-N1	77.43(8)	O4-Ln1-N1	67.47(14)
O1-Ln1-O3	111.44(10)	O2-Ln1-O3	100.19(14)
O1-Ln1-O4	84.00(8)	O2-Ln1-O3A	104.36(14)
O3-Ln1-O4	141.24(9)	O3-Ln1-O3A	69.02(15)
O4-Ln1-O5	68.92(9)	O4-Ln1-O3A	119.38(14)



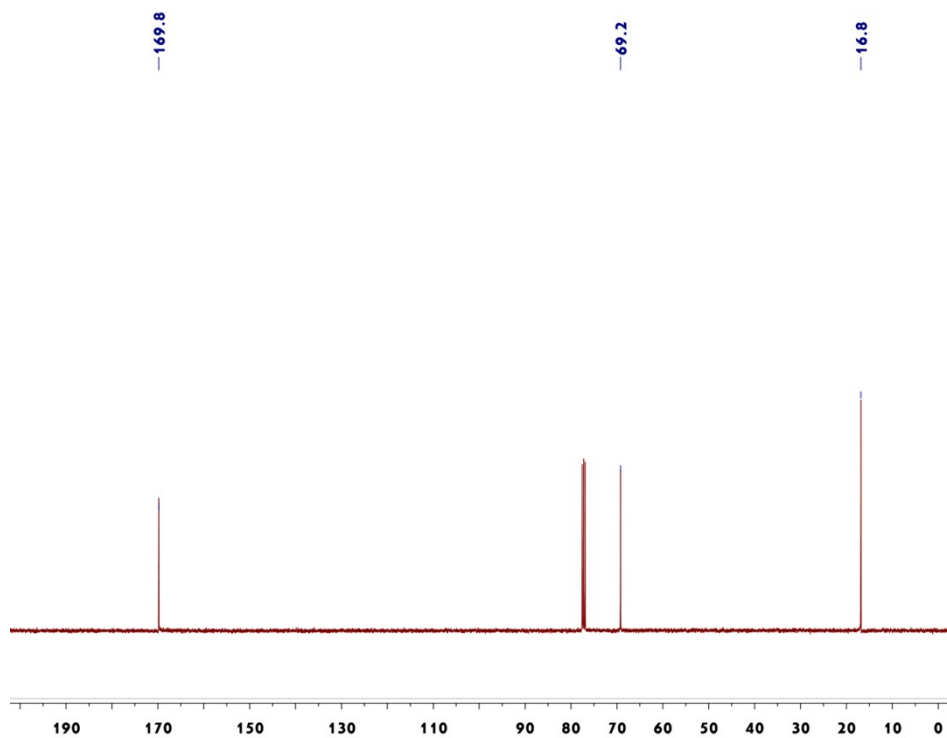
**Fig. S11** Solid state structure of complex **2** showing an atom numbering scheme. Thermal ellipsoids are drawn at 30% probability level, and hydrogen atoms are omitted for clarity.



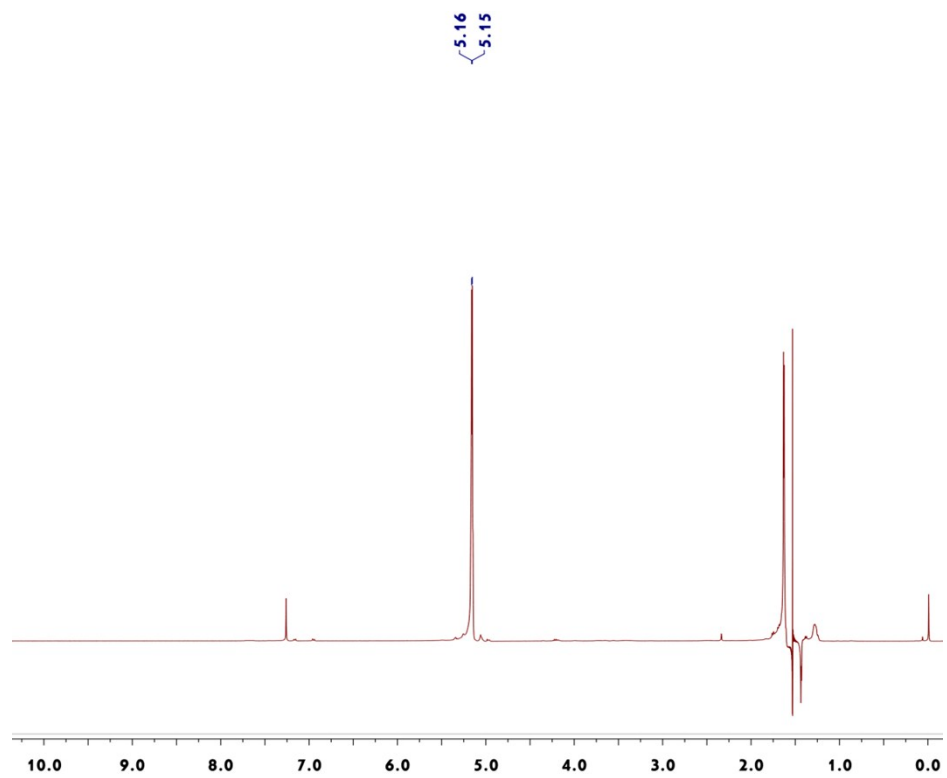
**Fig. S12** Plot of PLA Mn(■) and polydispersity (▲) as a function of  $L$ -lacOCA conversion initiated by complex **5**,  $[OCA]_0/[I]_0=200:1$ ,  $T=30\text{ }^\circ\text{C}$ .



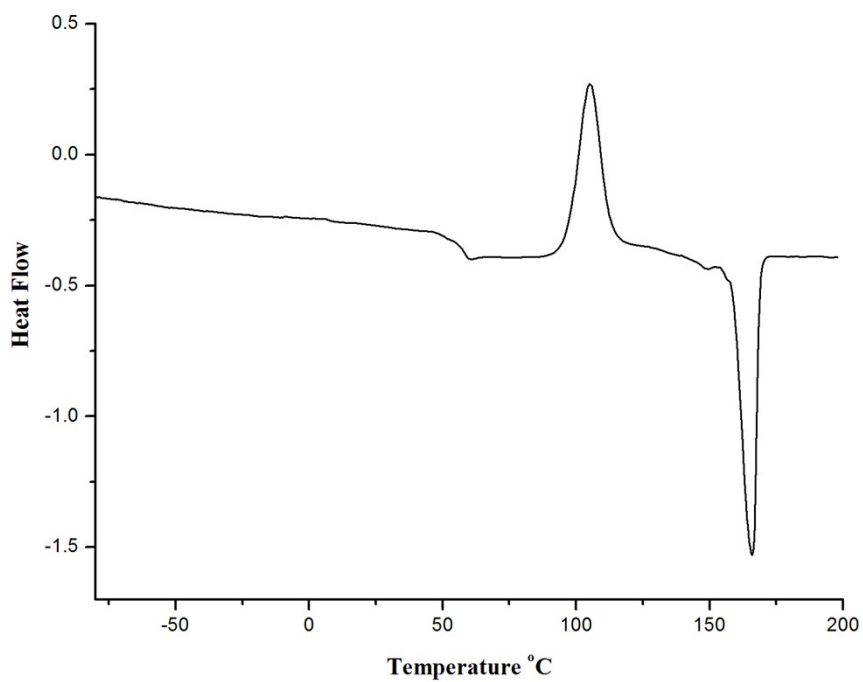
**Fig. S13**  $^1\text{H}$  NMR (normal) spectrum of PLLA obtained from  $L$ -lacOCA by complex **5**.



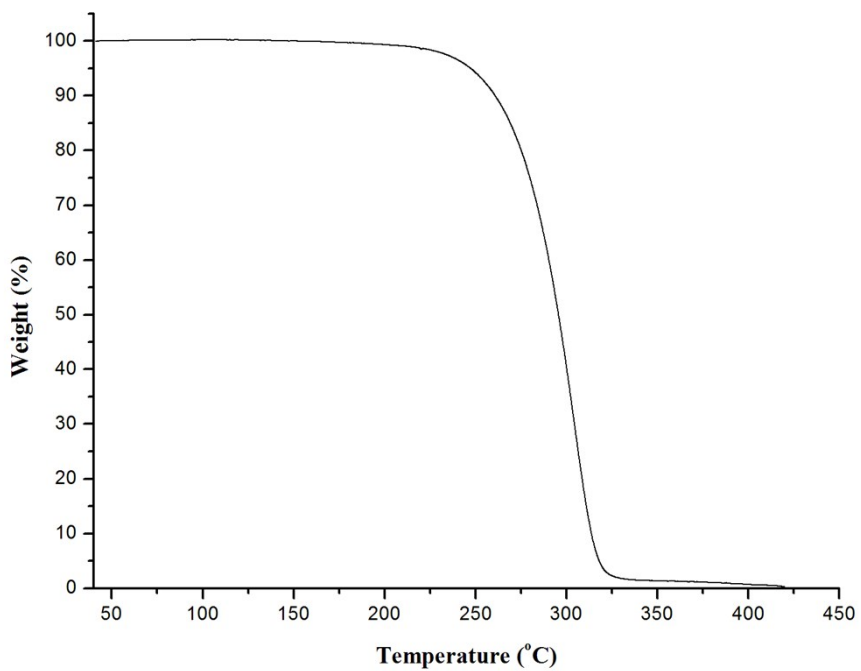
**Fig. S14**  $^{13}\text{C}$  NMR spectrum of PLLA obtained from  $L$ -lacOCA



**Fig. S15**  $^1\text{H}$  NMR (homodecoupled) spectrum of PLLA obtained from  $L$ -lacOCA



**Fig. S16** DSC thermogram of poly(lactic acid)



**Fig. S17** TGA thermogram of poly(lactic acid)