

Electronic supplementary information (ESI)

Syntheses and characterization of phosphonates and
diphosphonates of molybdenum,
 $A_4[(MoO_3)_5(O_3PR)_2] \cdot xH_2O$, $A_2[Mo_2O_5(O_3PR)_2]$ and
 $A_2[Mo_2O_5(O_3P-R-PO_3)]$ ($A = K, Rb, Cs, Tl, NH_4$).

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Table S1. Atomic coordinates ($\times 10^4$) and equivalent isotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for $\text{Rb}_4[(\text{MoO}_3)_5(\text{O}_3\text{PCH}_2\text{Ph})_2] \cdot 8\text{H}_2\text{O}(\mathbf{1})$ compound.

Atom	<i>x</i>	<i>y</i>	<i>z</i>	<i>U</i> (eq)
Rb(1)	7746(2)	1056(1)	4517(1)	63(1)
Rb(2)	2609(1)	1608(1)	6866(1)	36(1)
Mo(1)	276(1)	7629(1)	6711(1)	21(1)
Mo(2)	55(1)	9802(1)	6221(1)	21(1)
Mo(3)	0	1124(1)	7500	20(1)
P	8505(2)	9174(2)	7169(1)	18(1)
O(1)	0	7323(7)	7500	25(3)
O(2)	9141(6)	7203(6)	6180(4)	28(2)
O(3)	1207(6)	6853(6)	6677(4)	30(2)
O(4)	1664(5)	8344(5)	7465(4)	19(2)
O(5)	9262(6)	8934(5)	6770(4)	19(2)
O(6)	1607(6)	8589(5)	6218(4)	23(2)
O(7)	9469(6)	769(5)	6624(4)	24(2)
O(8)	1099(6)	337(6)	6072(4)	29(2)
O(9)	1047(6)	9954(5)	7361(4)	23(2)
O(10)	9077(7)	9816(6)	5498(4)	35(2)
O(11)	8992(6)	1838(6)	7512(4)	30(2)
O(12)	7602(9)	1718(8)	5785(6)	61(3)
O(13)	4080(9)	114(8)	6991(6)	68(4)
O(14)	6850(8)	1573(6)	3212(6)	56(3)
O(15)	6039(13)	9630(12)	4291(8)	97(5)
C(1)	6629(8)	8785(8)	6217(6)	27(3)
C(2)	6916(10)	8412(10)	5713(6)	34(3)
C(3)	6345(12)	7717(11)	5354(7)	47(4)
C(4)	5480(13)	7370(11)	5479(8)	53(4)
C(5)	5167(11)	7741(11)	5981(8)	50(4)
C(6)	5726(10)	8450(10)	6351(7)	41(3)
C(7)	7247(9)	9531(8)	6620(6)	23(2)

* U_{eq} is defined as one-third of the trace of the orthogonalized U_{ij} tensor.

Table S2. Atomic coordinates ($\times 10^4$) and equivalent isotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for $\text{Cs}_4[(\text{MoO}_3)_5(\text{O}_3\text{PCH}_2\text{Ph})_2] \cdot 3\text{H}_2\text{O}(2)$ compound.

Atom	<i>x</i>	<i>y</i>	<i>z</i>	<i>U</i> (eq)
Cs(1)	8797(2)	7294(1)	6098(1)	49(1)
Cs(2)	2933(1)	1277(1)	7905(1)	36(1)
Cs(3)	4867(1)	7229(1)	9183(1)	43(1)
Cs(4)	0	0	0	63(1)
Cs(5)	7230(1)	4994(1)	4070(1)	107(1)
Mo(1)	6394(1)	847(1)	8938(1)	19(1)
Mo(2)	8996(1)	572(1)	7534(1)	21(1)
Mo(3)	9194(1)	3390(1)	6518(1)	25(1)
Mo(4)	6391(1)	5431(1)	6970(1)	24(1)
Mo(5)	4402(1)	3957(1)	8243(1)	21(1)
P(1)	7801(2)	3196(2)	8602(1)	19(1)
P(2)	6175(2)	2500(2)	6717(1)	20(1)
O(1)	5005(5)	2290(4)	8798(3)	22(1)
O(2)	5664(5)	9862(5)	8541(4)	29(1)
O(3)	6414(6)	353(5)	35(3)	29(1)
O(4)	7407(5)	2269(4)	9307(3)	23(1)
O(5)	6918(5)	1524(4)	7431(3)	23(1)
O(6)	8174(5)	189(4)	8642(3)	22(1)
O(7)	9182(5)	1711(5)	6512(3)	27(1)
O(8)	553(5)	186(5)	7919(4)	35(1)
O(9)	8810(5)	2582(4)	7954(3)	23(1)
O(10)	8876(6)	9382(5)	7032(4)	38(1)
O(11)	746(6)	3313(6)	6799(4)	40(1)
O(12)	9227(6)	3827(6)	5414(4)	43(2)
O(13)	8280(5)	4855(5)	6918(4)	30(1)
O(14)	6075(7)	6078(6)	5917(4)	41(2)
O(15)	6447(6)	6618(5)	7499(4)	39(1)
O(16)	6893(5)	3532(5)	6426(3)	24(1)
O(17)	6575(5)	3983(4)	8113(3)	24(1)

Table S2continued

Atom	<i>x</i>	<i>y</i>	<i>z</i>	<i>U</i> (eq)
O(18)	4671(5)	5247(5)	7330(4)	29(1)
O(19)	4300(6)	4679(5)	9145(4)	34(1)
O(20)	2814(5)	4019(5)	8064(4)	32(1)
O(21)	4803(5)	2986(5)	7057(3)	24(1)
O(22)	591(11)	8795(12)	5203(6)	97(4)
O(23)	1610(30)	7180(30)	6580(20)	136(11)
O(24)	2419(17)	2009(16)	8062(14)	78(6)
O(25)	9548(19)	6950(18)	6467(14)	162(7)
C(1)	9477(7)	3671(6)	9738(5)	23(1)
C(2)	733(9)	3197(9)	9474(6)	38(2)
C(3)	1691(10)	2648(9)	78(8)	50(3)
C(4)	1406(11)	2576(9)	949(7)	52(3)
C(5)	1169(12)	3073(11)	1225(7)	55(3)
C(6)	9204(10)	3617(10)	622(6)	45(2)
C(7)	8446(8)	4258(6)	9091(5)	27(2)
C(8)	5072(8)	987(8)	5922(5)	31(2)
C(9)	5378(10)	9813(8)	6390(6)	42(2)
C(10)	4457(12)	9113(9)	6513(7)	54(3)
C(11)	3235(12)	9564(11)	6180(8)	57(3)
C(12)	2921(11)	726(12)	5721(8)	57(3)
C(13)	3847(10)	1426(9)	5581(6)	46(2)
C(14)	6052(9)	1774(8)	5787(5)	33(2)

* U_{eq} is defined as one-third of the trace of the orthogonalized U_{ij} tensor.

Table S3. Atomic coordinates ($\times 10^4$) and equivalent isotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for $\text{Cs}_4[(\text{MoO}_3)_5(\text{O}_3\text{PPh})_2] \cdot 6\text{H}_2\text{O}(\mathbf{3})$ compound.

Atom	<i>x</i>	<i>y</i>	<i>z</i>	<i>U</i> (eq)
Cs(1)	8918(1)	1195(1)	8576(1)	63(1)
Cs(2)	5622(1)	9517(1)	9673(1)	47(1)
Mo(1)	2386(1)	2386(1)	0	23(1)
Mo(2)	180(1)	1930(1)	717(1)	19(1)
Mo(3)	8449(1)	9270(1)	507(1)	18(1)
P	9259(2)	1491(2)	9758(1)	16(1)
O(1)	630(5)	1728(4)	9614(1)	20(1)
O(2)	9224(5)	1103(4)	194(1)	19(1)
O(3)	8574(5)	479(5)	9518(1)	22(1)
O(4)	2530(6)	3686(6)	9703(2)	39(2)
O(5)	1008(5)	2969(5)	329(1)	23(1)
O(6)	9050(5)	3031(6)	844(2)	33(1)
O(7)	9300(5)	387(5)	878(1)	23(1)
O(8)	8124(5)	7988(5)	802(2)	31(1)
O(9)	6986(5)	38(5)	474(2)	28(1)
O(10)	8486(5)	8486(5)	0	23(2)
O(11)	1292(6)	1993(6)	1085(2)	32(1)
O(12)	9823(9)	8025(9)	8384(2)	84(3)
O(13)	6535(9)	9237(9)	8566(3)	87(3)
O(14)	5692(13)	1443(12)	8953(3)	126(5)
C(1)	8347(7)	2924(7)	9710(2)	21(2)
C(2)	8685(9)	3875(7)	9441(2)	29(2)
C(3)	7880(10)	4912(7)	9381(2)	37(2)
C(4)	6752(10)	5003(9)	9597(3)	46(2)
C(5)	6424(10)	4081(10)	9859(3)	50(3)
C(6)	7230(8)	3029(7)	9923(2)	31(2)

* U_{eq} is defined as one-third of the trace of the orthogonalized U_{ij} tensor.

Table S4. Atomic coordinates ($\times 10^4$) and equivalent isotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for $(\text{NH}_4)_4[(\text{MoO}_3)_5(\text{O}_3\text{PPh})_2] \cdot 6\text{H}_2\text{O}$ (4) compound.

Atom	<i>x</i>	<i>y</i>	<i>z</i>	<i>U</i> (eq)
N(1)	2798(2)	994(2)	4553(2)	39(1)
N(2)	8498(2)	3518(2)	5314(2)	38(1)
Mo(1)	92(1)	4403(1)	3638(1)	17(1)
Mo(2)	459(1)	2265(1)	4181(1)	17(1)
Mo(3)	0	957(1)	2500	19(1)
P	8770(1)	2906(1)	2650(1)	15(1)
O(1)	0	4744(2)	2500	20(1)
O(2)	9428(1)	5158(1)	4076(1)	30(1)
O(3)	1172(1)	4785(1)	3931(1)	28(1)
O(4)	8755(1)	3731(1)	3178(1)	20(1)
O(5)	9204(1)	3139(1)	1844(1)	17(1)
O(6)	6(1)	3429(1)	4407(1)	21(1)
O(7)	775(1)	1343(1)	3432(1)	21(1)
O(8)	9819(1)	1675(1)	4824(1)	31(1)
O(9)	9270(1)	2141(1)	3117(1)	18(1)
O(10)	1502(1)	2339(1)	4716(1)	27(1)
O(11)	9261(1)	260(1)	2941(1)	31(1)
O(12)	7205(2)	4724(2)	3303(2)	41(1)
O(13)	8639(3)	9928(3)	4593(3)	105(2)
O(14)	6768(2)	3344(2)	4343(2)	75(1)
C(1)	7629(2)	2562(2)	2357(2)	19(1)
C(2)	7073(2)	3112(2)	1843(2)	28(1)
C(3)	6184(2)	2866(3)	1606(2)	39(1)
C(4)	5854(2)	2074(3)	1874(2)	44(1)
C(5)	6401(2)	1520(3)	2380(2)	43(1)
C(6)	7291(2)	1768(2)	2631(2)	29(1)

* U_{eq} is defined as one-third of the trace of the orthogonalized U_{ij} tensor.

Table S5. Atomic coordinates ($\times 10^4$) and equivalent isotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for $(\text{K})_{1.5}(\text{H}_3\text{O})_{0.5}[(\text{Mo}_2\text{O}_5)(\text{O}_3\text{PPh})_2](5)$ compound

Atom	<i>x</i>	<i>y</i>	<i>z</i>	<i>U</i> (eq)
Mo(1)	3803(1)	9129(1)	3732(1)	9(1)
Mo(2)	1776(1)	4123(1)	3740(1)	9(1)
Mo(3)	6781(1)	1505(1)	3731(1)	9(1)
Mo(4)	8795(1)	6507(1)	3733(1)	9(1)
P(1)	4066(3)	2227(3)	2824(2)	11(1)
P(2)	1931(3)	7217(3)	2832(2)	9(1)
P(3)	6923(3)	8866(3)	2817(2)	9(1)
P(4)	9074(3)	13859(3)	2828(2)	9(1)
K(1)	359(3)	10380(3)	3457(2)	30(1)
K(2)	5366(3)	5390(3)	3410(2)	32(1)
K(3)	3359(6)	11789(5)	5003(3)	26(3)
O(3W)	3359(6)	11789(5)	5003(3)	130(60)
K(4)	1653(6)	6779(6)	5000(3)	26(4)
O(4W)	1653(6)	6779(6)	5000(3)	160(80)
O(1)	3381(8)	3368(8)	3182(4)	13(2)
O(2)	223(8)	5248(8)	3992(4)	15(2)
O(3)	2457(8)	9569(9)	4286(4)	18(2)
O(4)	7582(9)	49(8)	3065(4)	16(2)
O(5)	8419(8)	2520(8)	3187(4)	14(2)
O(6)	8293(8)	5064(8)	3073(4)	14(2)
O(7)	5645(8)	12180(8)	2898(4)	12(2)
O(8)	7456(9)	7544(8)	3164(4)	18(2)
O(9)	335(8)	7184(8)	2906(4)	15(2)
O(10)	5318(8)	8856(8)	2896(4)	13(2)
O(11)	1112(9)	2612(8)	4128(4)	19(2)
O(12)	648(8)	3861(8)	2906(4)	14(2)
O(13)	2430(8)	8392(8)	3180(4)	17(2)
O(14)	4305(9)	7607(9)	4118(4)	21(2)
O(15)	2592(8)	5895(8)	3082(4)	16(2)

Table S5 continued

O(16)	9278(9)	7810(9)	4128(4)	22(2)
O(17)	3304(9)	925(8)	3060(4)	17(2)
O(18)	7885(9)	803(9)	4280(4)	17(2)
O(19)	2887(9)	4548(8)	4282(4)	15(2)
O(20)	5229(8)	263(8)	3986(4)	16(2)
O(21)	7441(9)	5826(9)	4274(4)	18(2)
O(22)	6134(9)	2816(9)	4120(4)	19(2)
C(1)	3911(13)	2655(12)	1950(6)	15(3)
C(2)	5070(20)	2990(30)	1510(9)	68(7)
C(3)	4950(20)	3390(30)	845(9)	74(7)
C(4)	3696(18)	3355(19)	601(8)	44(4)
C(5)	2520(20)	3057(16)	1021(8)	46(5)
C(6)	2609(17)	2662(17)	1708(8)	37(4)
C(7)	2517(14)	7631(13)	1972(6)	18(3)
C(8)	2665(19)	6599(19)	1565(9)	47(4)
C(9)	3090(20)	6950(20)	882(10)	65(6)
C(10)	3316(18)	8269(19)	616(9)	46(4)
C(11)	3117(18)	9250(20)	1009(9)	50(5)
C(12)	2741(15)	8973(15)	1691(8)	31(4)
C(13)	7493(12)	8892(11)	1942(6)	11(2)
C(14)	8867(17)	8901(18)	1697(8)	42(4)
C(15)	9310(20)	8860(20)	1029(9)	60(6)
C(16)	8344(18)	8885(17)	594(7)	38(4)
C(17)	6930(20)	8930(30)	830(9)	80(8)
C(18)	6500(20)	8910(30)	1494(10)	73(7)
C(19)	8881(13)	3866(12)	1969(6)	15(3)
C(20)	8920(20)	5073(19)	1561(8)	49(5)
C(21)	8860(20)	5030(20)	875(9)	60(6)
C(22)	8745(18)	3884(18)	618(9)	44(4)
C(23)	8671(18)	2700(20)	1009(9)	51(5)
C(24)	8729(16)	2659(18)	1701(8)	38(4)

* U_{eq} is defined as one-third of the trace of the orthogonalized U_{ij} tensor.

Table S6. Atomic coordinates ($\times 10^4$) and equivalent isotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for $\text{Rb}_2[(\text{Mo}_2\text{O}_5)(\text{O}_3\text{PC}_6\text{H}_4\text{PO}_3)](\mathbf{11})$ compound.

Atom	<i>x</i>	<i>y</i>	<i>z</i>	<i>U</i> (eq)
Rb(1)	1626(3)	1787(2)	5000	28(1)
Rb(2)	5000	0	2794(2)	37(1)
Mo	3560(1)	3784(1)	3306(1)	7(1)
P(1)	1451(4)	1661(3)	2091(2)	6(1)
O(1)	4098(10)	2397(10)	3875(6)	17(2)
O(2)	5000	5000	3652(8)	11(3)
O(3)	2307(11)	4362(11)	3991(6)	16(2)
O(4)	2174(11)	2870(9)	2510(6)	13(2)
O(5)	2065(10)	360(9)	2390(6)	11(2)
O(6)	4924(9)	3226(9)	2251(6)	9(2)
C(1)	1625(14)	1777(13)	923(8)	14(3)
C(2A)	1770(30)	2940(20)	448(17)	13(5)
C(3A)	1820(30)	580(30)	444(18)	15(5)
C(2B)	2870(40)	1740(50)	490(40)	13(6)
C(3B)	440(40)	2060(60)	450(30)	21(7)
C(2C)	2770(60)	2060(80)	410(60)	13(7)
C(3C)	400(50)	1490(70)	480(50)	15(7)

* U_{eq} is defined as one-third of the trace of the orthogonalized U_{ij} tensor.

Table S7. Atomic coordinates ($\times 10^4$) and equivalent isotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for $(\text{NH}_4)_2[(\text{Mo}_2\text{O}_5)(\text{O}_3\text{PC}_6\text{H}_4\text{PO}_3)]$ (**13**) compound.

Atom	<i>x</i>	<i>y</i>	<i>z</i>	<i>U</i> (eq)
N(1)	1622(6)	1778(6)	5000	30(1)
N(2)	5000	0	2806(4)	30(1)
Mo	3545(1)	3786(1)	3305(1)	11(1)
P(1)	1453(1)	1665(1)	2086(1)	10(1)
O(1)	4101(3)	2391(3)	3852(2)	24(1)
O(2)	5000	5000	3648(2)	16(1)
O(3)	2300(3)	4360(3)	4016(2)	22(1)
O(4)	2144(3)	2872(2)	2525(2)	19(1)
O(5)	2103(2)	371(2)	2400(2)	18(1)
O(6)	9891(3)	1726(3)	2234(2)	18(1)
C(1)	1673(4)	1776(4)	916(2)	19(1)
C(2A)	1830(14)	2908(9)	458(5)	41(2)
C(3A)	1624(14)	553(10)	450(6)	52(3)
C(2B)	2773(17)	1420(20)	450(10)	43(4)
C(3B)	652(18)	2552(18)	449(9)	43(3)
C(2C)	2710(20)	2280(30)	446(12)	28(4)
C(3C)	560(30)	1190(30)	455(13)	46(5)

* U_{eq} is defined as one-third of the trace of the orthogonalized U_{ij} tensor.

Table S8. Atomic coordinates ($\times 10^4$) and equivalent isotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for $\text{Rb}_2[(\text{Mo}_2\text{O}_5)(\text{O}_3\text{PC}_4\text{H}_8\text{PO}_3)](\mathbf{16})$ compound.

Atom	<i>x</i>	<i>y</i>	<i>z</i>	<i>U</i> (eq)
Rb(1)	1716(2)	3284(2)	0	41(1)
Rb(2)	5000	5000	1113(1)	29(1)
Mo	3673(1)	1327(1)	873(1)	18(1)
P	3454(3)	8454(3)	1494(2)	21(1)
O(1)	3090(9)	9906(13)	1398(3)	39(3)
O(2)	5000	0	714(4)	7(3)
O(3)	4275(10)	2624(9)	546(3)	32(2)
O(4)	2833(15)	2871(15)	1232(5)	17(3)
C(1)	3131(13)	8131(13)	2088(6)	19(4)
C(2)	1390(30)	1600(30)	2240(8)	22(6)

* U_{eq} is defined as one-third of the trace of the orthogonalized U_{ij} tensor.

Table S9. Atomic coordinates ($\times 10^4$) and equivalent isotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for $\text{Rb}_2[(\text{Mo}_2\text{O}_5)(\text{O}_3\text{PC}_3\text{H}_6\text{PO}_3)]$ (**19**) compound.

Atom	<i>x</i>	<i>y</i>	<i>z</i>	<i>U</i> (eq)
Rb(1)	4989(1)	1617(1)	8160(1)	21(1)
Rb(2)	2651(1)	24(1)	4593(1)	24(1)
Mo(1)	3163(1)	3574(1)	5898(1)	8(1)
Mo(2)	6901(1)	3543(1)	6543(1)	8(1)
P(1)	1832(1)	3540(1)	2783(1)	7(1)
P(2)	1777(1)	1386(1)	1137(1)	7(1)
O(1)	2035(3)	2970(4)	4222(3)	12(1)
O(2)	2326(3)	2109(4)	2387(3)	13(1)
O(3)	2522(3)	2111(3)	6861(4)	14(1)
O(4)	3923(3)	2376(4)	5351(4)	15(1)
O(5)	3455(3)	5004(3)	4710(3)	11(1)
O(6)	3827(3)	4177(4)	7352(4)	16(1)
O(7)	1947(3)	4845(3)	6298(4)	12(1)
O(8)	6118(3)	2352(4)	5754(4)	17(1)
O(9)	6286(3)	4089(4)	7826(4)	17(1)
O(10)	1903(3)	99(3)	7693(3)	11(1)
O(11)	2050(3)	3002(4)	4840(3)	13(1)
C(1)	668(4)	3052(6)	1983(6)	13(1)
C(2)	232(4)	3740(6)	2384(6)	17(1)
C(3)	529(4)	1730(6)	1259(6)	17(1)

* U_{eq} is defined as one-third of the trace of the orthogonalized U_{ij} tensor.

Table S10. Bond lengths (Å) for Rb₄[(MoO₃)₅(O₃PCH₂Ph)₂].8H₂O(**1**) compound.

Bond	Length	Bond	Length	Bond	Length
Rb(1)-O(6)	3.106(8)	Rb(2)-O(13)	2.906(12)	Mo(2)-O(9)	2.433(8)
Rb(1)-O(8)	3.067(9)	Rb(2)-O(9)	3.573(8)	Mo(2)-O(10)	1.702(9)
Rb(1)-O(10)	2.957(9)	BVS	1.059	BVS	5.924
Rb(1)-O(12)	2.978(12)	Mo(1)-O(1)	1.907(3)	Mo(3)-O(7) X2	1.890(8)
Rb(1)-O(12)	3.370(12)	Mo(1)-O(2)	1.713(8)	Mo(3)-O(9) X2	2.296(7)
Rb(1)-O(14)	2.822(13)	Mo(1)-O(3)	1.702(8)	Mo(3)-O(11) X 2	1.709(8)
Rb(1)-O(15)	3.023(17)	Mo(1)-O(4)	2.312(7)	BVS	6.208
BVS	0.913	Mo(1)-O(5)	2.378(7)	P-C(7)	1.808(12)
Rb(2)-O(1)	3.221(4)	Mo(1)-O(6)	1.908(8)	P-O(4)	1.517(8)
Rb(2)-O(2)	2.983(8)	BVS	6.042	P-O(5)	1.549(8)
Rb(2)-O(3)	3.093(9)	Mo(2)-O(5)	2.220(7)	P-O(9)	1.536(8)
Rb(2)-O(4)	2.969(8)	Mo(2)-O(6)	1.942(8)		
Rb(2)-O(8)	2.902(8)	Mo(2)-O(7)	1.955(8)		
Rb(2)-O(11)	2.852(8)	Mo(2)-O(8)	1.705(8)		

Table S11. Bond lengths (Å) for Cs₄[(MoO₃)₅(O₃PCH₂Ph)₂].3H₂O(2) compound.

Bond	Length	Bond	Length	Bond	Length
Cs(1)-O(10)	2.995(6)	Cs(4)-O(4)X2	3.389(5)	Mo(3)-O(9)	2.324(5)
Cs(1)-O(12)	3.252(7)	Cs(4)-O(3)X2	3.719(6)	Mo(3)-O(11)	1.704(6)
Cs(1)-O(13)	3.031(5)	BVS	0.914	Mo(3)-O(12)	1.707(6)
Cs(1)-O(14)	3.527(7)	Cs(5)-O(11)	3.308(6)	Mo(3)-O(13)	1.896(6)
Cs(1)-O(15)	3.351(7)	Cs(5)-O(12)	2.986(6)	Mo(3)-O(16)	2.414(5)
Cs(1)-O(22)	2.991(10)	Cs(5)-O(14)	3.346(7)	BVS	6.021
Cs(1)-O(23)	3.09(3)	Cs(5)-O(16)	3.804(5)	Mo(4)-O(13)	1.951(6)
Cs(1)-O(25)	0.97(2)	Cs(5)-O(18)	3.135(5)	Mo(4)-O(14)	1.702(6)
BVS	1.033	Cs(5)-O(21)	3.135(5)	Mo(4)-O(15)	1.703(6)
Cs(2)-O(1)	3.186(5)	Cs(5)-O(20)	3.325(6)	Mo(4)-O(16)	2.378(5)
Cs(2)-O(2)	3.093(6)	Cs(5)-O(23)	2.80(3)	Mo(4)-O(17)	2.208(5)
Cs(2)-O(3)	3.455(5)	Cs(5)-O(25)	3.725(19)	Mo(4)-O(18)	1.921(6)
Cs(2)-O(8)	3.062(6)	BVS	1.168	BVS	6.050
Cs(2)-O(11)	3.221(6)	Mo(1)-O(1)	1.922(5)	Mo(5)-O(1)	1.934(5)
Cs(2)-O(20)	3.148(6)	Mo(1)-O(2)	1.711(5)	Mo(5)-O(17)	2.304(5)
Cs(2)-O(21)	3.188(5)	Mo(1)-O(3)	1.706(5)	Mo(5)-O(18)	1.936(5)
BVS	0.908	Mo(1)-O(4)	2.302(5)	Mo(5)-O(19)	1.717(6)
Cs(3)-O(1)	3.279(5)	Mo(1)-O(5)	2.418(5)	Mo(5)-O(20)	1.703(6)
Cs(3)-O(2)	3.299(5)	Mo(1)-O(6)	1.911(5)	Mo(5)-O(21)	2.261(5)
Cs(3)-O(3)	3.159(6)	BVS	5.964	BVS	5.988
Cs(3)-O(4)	3.269(5)	Mo(2)-O(5)	2.212(5)	P(1)-O(4)	1.510(5)
Cs(3)-O(15)	3.097(7)	Mo(2)-O(6)	1.914(5)	P(1)-O(9)	1.540(6)
Cs(3)-O(19)	3.118(6)	Mo(2)-O(7)	1.917(5)	P(1)-O(17)	1.555(5)
Cs(3)-O(19)	3.137(6)	Mo(2)-O(8)	1.720(6)	P(1)-C(7)	1.805(7)
Cs(3)-O(24)	3.083(17)	Mo(2)-O(9)	2.440(5)	P(2)-O(5)	1.545(5)
BVS	1.041	Mo(2)-O(10)	1.700(6)	P(2)-O(16)	1.544(5)
Cs(4)-O(6)X2	2.895(5)	BVS	6.038	P(2)-O(21)	1.520(6)
Cs(4)-O(8)X2	3.229(6)	Mo(3)-O(7)	1.920(5)	P(2)-C(14)	1.797(8)

Table S12. Bond lengths (Å) for $A_4[(\text{MoO}_3)_5(\text{O}_3\text{PC}_6\text{H}_5)_2] \cdot 6\text{H}_2\text{O}$ ($A = \text{Cs}(\mathbf{3}), \text{NH}_4(\mathbf{4})$) compounds.

A	Bond	Length	Bond	Length	Bond	Length
Cs(3)	Cs(1)-O(3)	3.333(5)	Cs(2)-O(9)	3.140(5)	Mo(2)-O(11)	1.713(6)
	Cs(1)-O(4)	3.039(6)	Cs(2)-O(10)	3.366(3)	BVS	6.028
	Cs(1)-O(7)	3.121(5)	Cs(2)-O(12)	3.370(9)	Mo(3)-O(3)	2.239(5)
	Cs(1)-O(11)	3.302(6)	Cs(2)-O(14)	3.180(10)	Mo(3)-O(2)	2.335(5)
	Cs(1)-O(11)	3.412(6)	BVS	0.902	Mo(3)-O(7)	1.940(5)
	Cs(1)-O(12)	3.499(10)	Mo(1)-O(1) X2	2.361(5)	Mo(3)-O(8)	1.710(5)
	Cs(1)-O(13)	3.216(9)	Mo(1)-O(4) X2	1.701(6)	Mo(3)-O(9)	1.726(5)
	Cs(1)-O(14)	3.613(13)	Mo(1)-O(5) X2	1.925(5)	Mo(3)-O(10)	1.919(2)
	Cs(1)-O(14)	3.082(12)	BVS	5.982	BVS	5.940
	BVS	0.952	Mo(2)-O(1)	2.392(4)	P-C(1)	1.780(7)
	Cs(2)-O(3)	3.280(5)	Mo(2)-O(2)	2.223(5)	P-O(1)	1.531(5)
	Cs(2)-O(5)	3.174(5)	Mo(2)-O(5)	1.918(5)	P-O(2)	1.549(5)
	Cs(2)-O(6)	3.266(6)	Mo(2)-O(6)	1.702(5)	P-O(3)	1.515(5)
	Cs(2)-O(8)	3.291(6)	Mo(2)-O(7)	1.933(5)		
(NH₄)₄(4)	Mo(1)-O(1)	1.9044(7)	Mo(2)-O(6)	1.9287(17)	Mo(3)-O(11) X2	1.7165(19)
	Mo(1)-O(2)	1.699(18)	Mo(2)-O(7)	1.9310(17)	BVS	5.966
	Mo(1)-O(3)	1.7231(18)	Mo(2)-O(8)	1.7159(19)	P-O(4)	1.5122(18)
	Mo(1)-O(4)	2.2882(16)	Mo(2)-O(9)	2.3560(17)	P-O(5)	1.5439(17)
	Mo(1)-O(5)	2.3425(16)	Mo(2)-O(10)	1.7044(18)	P-O(9)	1.5336(17)
	Mo(1)-O(6)	1.9377(18)	BVS	6.022	P-C(1)	1.788(2)
	BVS	5.990	Mo(3)-O(7) X2	1.9018(17)		
	Mo(2)-O(5)	2.2105(16)	Mo(3)-O(9) X2	2.3573(17)		

Table S13. Bond lengths (Å) for (K)_{1.5}(H₃O)_{0.5}[(Mo₂O₅)(O₃PPh)₂](5) compound.

Bond	Length	Bond	Length	Bond	Length
K(1)-O(4)	2.871(9)	K(4)-O(2)	3.114(11)	Mo(3)-O(18)	1.705(8)
K(1)-O(5)	2.859(10)	K(4)-O(2)	3.126(10)	Mo(3)-O(20)	1.912(8)
K(1)-O(11)	2.883(10)	K(4)-O(3)	3.046(11)	Mo(3)-O(22)	1.693(9)
K(1)-O(13)	2.859(10)	K(4)-O(11)	3.114(11)	BVS	6.229
K(1)-O(16)	2.893(10)	K(4)-O(14)	2.996(11)	Mo(4)-O(2)	1.920(8)
K(1)-O(17)	2.866(9)	K(4)-O(16)	3.121(11)	Mo(4)-O(6)	2.174(8)
K(1)-O(18)	2.801(10)	K(4)-O(18)	3.037(11)	Mo(4)-O(8)	2.017(9)
BVS	1.137	K(4)-O(19)	2.982(10)	Mo(4)-O(9)	2.157(9)
K(2)-O(1)	2.874(9)	K(4)-O(21)	3.001(11)	Mo(4)-O(16)	1.697(9)
K(2)-O(6)	2.820(9)	K(4)-O(22)	2.971(11)	Mo(4)-O(21)	1.688(9)
K(2)-O(8)	2.894(10)	BVS	0.846	BVS	6.274
K(2)-O(14)	2.912(10)	Mo(1)-O(3)	1.703(9)	P(1)-C(1)	1.801(14)
K(2)-O(15)	2.819(9)	Mo(1)-O(10)	2.157(8)	P(2)-C(7)	1.779(14)
K(2)-O(19)	2.858(9)	Mo(1)-O(13)	2.011(8)	P(3)-C(13)	1.803(13)
K(2)-O(21)	2.853(10)	Mo(1)-O(14)	1.715(9)	P(4)-C(19)	1.777(13)
K(2)-O(22)	2.924(10)	Mo(1)-O(17)	2.205(9)	P(1)-O(1)	1.537(9)
BVS	1.097	Mo(1)-O(20)	1.922(8)	P(1)-O(7)	1.532(9)
K(3)-O(3)	2.977(11)	BVS	6.087	P(1)-O(17)	1.488(9)
K(3)-O(11)	2.987(10)	Mo(2)-O(1)	2.008(9)	P(2)-O(9)	1.512(9)
K(3)-O(14)	3.101(11)	Mo(2)-O(2)	1.917(9)	P(2)-O(13)	1.543(9)
K(3)-O(16)	2.972(11)	Mo(2)-O(11)	1.703(9)	P(2)-O(15)	1.515(9)
K(3)-O(18)	2.989(11)	Mo(2)-O(12)	2.151(8)	P(3)-O(4)	1.502(9)
K(3)-O(19)	3.022(10)	Mo(2)-O(15)	2.177(9)	P(3)-O(8)	1.530(9)
K(3)-O(20)	3.123(11)	Mo(2)-O(19)	1.703(9)	P(3)-O(10)	1.521(9)
K(3)-O(20)	3.124(11)	BVS	6.205	P(4)-O(5)	1.541(9)
K(3)-O(21)	3.012(10)	Mo(3)-O(4)	2.194(9)	P(4)-O(6)	1.528(9)
K(3)-O(22)	3.129(11)	Mo(3)-O(5)	2.012(8)	P(4)-O(12)	1.524(9)
BVS	0.864	Mo(3)-O(7)	2.149(8)		

Table S14. Bond lengths (Å) for $A_2[(Mo_2O_5)(O_3PC_6H_4PO_3)]$ ($A = Rb(11), NH_4(13)$) compounds.

A	Bond	Length	Bond	Length	Bond	Length
Rb(11)	Rb(1)-O(1)	3.008(10)	Rb(2)-O(4)X 2	3.026(10)	Mo-O(5)	2.188(9)
	Rb(1)-O(3)X 2	3.047(10)	Rb(2)-O(5)X 2	2.936(10)	Mo-O(6)	2.152(9)
	Rb(1)-O(3)X 2	3.066(11)	Rb(2)-O(6)X 2	3.326(9)	BVS	6.087
	Rb(1)-O(1)X 2	3.100(10)	BVS	1.250	P(1)-O(4)	1.535(10)
	Rb(1)-O(2)X 2	3.143(8)	Mo-O(1)	1.714(10)	P(1)-O(5)	1.499(9)
	BVS	0.996	Mo-O(2)	1.925(4)	P(1)-O(6)	1.506(10)
	Rb(2)-O(1)X 2	3.033(10)	Mo-O(3)	1.702(10)	P(1)-C(1)	1.789(13)
	Rb(2)-O(3)X 2	2.956(10)	Mo-O(4)	2.026(10)		
NH₄(13)	N(1)-H(1B)	0.90(2)	Mo-O(2)	1.9247(9)	P(1)-C(1)	1.793(4)
	N(1)-H(1C)	0.90(2)	Mo-O(3)	1.711(2)	P(1)-O(4)	1.532(3)
	N(1)-H(1A)	0.90(2)	Mo-O(4)	2.014(2)	P(1)-O(5)	1.514(3)
	N(2)-H(2C)	0.91(2)	Mo-O(5)	2.190(2)	P(1)-O(6)	1.519(2)
	N(2)-H(2D)	0.90(2)	Mo-O(6)	2.142(2)		
	Mo-O(1)	1.710(3)	BVS	6.098		

Table S15. Bond lengths (Å) for $Rb_2[(Mo_2O_5)(O_3PC_4H_8PO_3)](16)$ compound.

Bond	Length	Bond	Length	Bond	Length
Rb(1)-O(2) X 2	3.189(9)	Rb(2)-O(3) X 3	2.965(9)	Mo-O(4) X 2	2.031(15)
Rb(1)-O(3) X 4	3.028(10)	Rb(2)-O(4) X 3	3.006(14)	BVS	5.846
Rb(1)-O(3) X 4	3.061(9)	BVS	1.050	P-O(1) X 2	1.499(12)
BVS	1.132	Mo-O(1) X 2	2.170(9)	P-O(4) X 2	1.591(15)
Rb(2)-O(1) X 2	3.154(9)	Mo-O(3) X 2	1.707(9)	P-C(1)	1.819(17)

Table S16. Bond lengths (Å) for Rb₂[(Mo₂O₅)(O₃PC₃H₆PO₃)](**19**) compound.

Bond	Length	Bond	Length	Bond	Length
Rb(1)-O(4)	3.073(4)	Rb(2)-O(4)	2.933(4)	Mo(2)-O(9)	1.709(4)
Rb(1)-O(4)	2.955(4)	Rb(2)-O(6)	3.031(4)	Mo(2)-O(2)	1.984(3)
Rb(1)-O(5)	3.207(4)	Rb(2)-O(8)	2.926(4)	Mo(2)-O(5)	1.903(3)
Rb(1)-O(5)	3.234(4)	Rb(2)-O(9)	2.932(4)	Mo(2)-O(11)	2.197(4)
Rb(1)-O(6)	2.972(4)	Rb(2)-O(10)	3.375(4)	Mo(2)-O(10)	2.186(3)
Rb(1)-O(6)	3.020(4)	Rb(2)-O(11)	3.075(4)	BVS	6.145
Rb(1)-O(8)	3.010(4)	BVS	1.234	P(1)-O(1)	1.523(4)
Rb(1)-O(8)	3.108(4)	Mo(1)-O(1)	2.217(3)	P(1)-O(3)	1.547(4)
Rb(1)-O(9)	3.052(4)	Mo(1)-O(3)	1.989(4)	P(1)-O(10)	1.521(3)
Rb(1)-O(9)	3.117(4)	Mo(1)-O(4)	1.710(4)	P(1)-C(1)	1.779(6)
BVS	1.145	Mo(1)-O(5)	1.900(3)	P(2)-O(2)	1.540(3)
Rb(2)-O(1)	2.998(4)	Mo(1)-O(6)	1.714(3)	P(2)-O(7)	1.521(3)
Rb(2)-O(2)	3.005(3)	Mo(1)-O(7)	2.177(4)	P(2)-O(11)	1.511(4)
Rb(2)-O(3)	3.051(4)	BVS	6.123	P(2)-C(3)	1.793(6)
Rb(2)-O(7)	3.291(4)	Mo(2)-O(8)	1.710(4)		

Table S17. Unit cell parameters of compounds **6**, **10**, **12**, **14**, **15**, **17**, **18** and **20**

Sl. No	Composition	Space group	Volume	Unit Cell					
				<i>a</i> (Å)	<i>b</i> (Å)	<i>c</i> (Å)	α (°)	β (°)	γ (°)
1	Rb ₂ [(Mo ₂ O ₅)(O ₃ PPh) ₂](6)	<i>P</i> $\bar{1}$	2003.08	9.7497	10.0044	20.8846	82.004	83.241	89.805
2	K ₂ [(Mo ₂ O ₅)(O ₃ P(C ₆ H ₄)PO ₃)](10)	<i>Pbam</i>	1428.55	9.4106	10.0157	15.1564			
3	Cs ₂ [(Mo ₂ O ₅)(O ₃ P(C ₆ H ₄)PO ₃)](12)	<i>Pbam</i>	1540.28	9.9724	9.9694	15.5131			
4	Tl ₂ [(Mo ₂ O ₅)(O ₃ P(C ₆ H ₄)PO ₃)](14)	<i>P222</i>	1431.28	9.5700	9.8799	15.1378			
5	K ₂ [(Mo ₂ O ₅)(O ₃ P(CH ₂) ₄ PO ₃)](15)	<i>I4/mcm</i>	2777.15	9.7359	9.7359	29.2984			
6	Cs ₂ [(Mo ₂ O ₅)(O ₃ P(CH ₂) ₄ PO ₃)](17)	<i>I4/mcm</i>	2975.10	9.9150	9.9150	30.2792			
7	(NH ₄) ₂ [(Mo ₂ O ₅)(O ₃ P(CH ₂) ₄ PO ₃)](18)	<i>I4/mcm</i>	2854.96	9.8069	9.8069	29.6847			
8	Cs ₂ [(Mo ₂ O ₅)(O ₃ P(CH ₂) ₃ PO ₃)](20)	<i>P2₁/c</i>	1399.99	14.3051	9.9134	9.9511	90	97.223	90

Table S18. CHN analysis of compounds **1-20**

Sl. No	Composition	C %		H %		N %	
		Calc	Obs	Calc	Obs	Calc	Obs
1	Rb ₄ [(MoO ₃) ₅ (O ₃ PCH ₂ Ph) ₂].8H ₂ O(1)	10.88	10.65	1.96	1.17	-	-
2	Cs ₄ [(MoO ₃) ₅ (O ₃ PCH ₂ Ph) ₂].3H ₂ O(2)	10.22	10.15	1.23	0.88	-	-
3	Cs ₄ [(MoO ₃) ₅ (O ₃ PPh) ₂].6H ₂ O(3)	8.62	8.66	1.33	0.43	-	-
4	(NH ₄) ₄ [(MoO ₃) ₅ (O ₃ PPh) ₂].6H ₂ O(4)	11.89	14.96	3.16	1.89	4.62	4.72
5	K _{1.5} (H ₃ O) _{0.5} [(Mo ₂ O ₅)(O ₃ PPh) ₂](5)	22.09	22.45	1.78	1.51	-	-
6	Rb ₂ [(Mo ₂ O ₅)(O ₃ PPh) ₂](6)	19.09	19.43	1.34	0.83	-	-
7	Cs ₂ [(Mo ₂ O ₅)(O ₃ PPh) ₂](7)	16.61	16.53	1.19	0.84	-	-
8	(NH ₄) ₂ [(Mo ₂ O ₅)(O ₃ PPh) ₂](8)	23.24	23.49	2.93	1.86	4.52	6.07
9	Tl ₂ [(Mo ₂ O ₅)(O ₃ PPh) ₂](9)	14.52	14.10	1.02	1.51	-	-
10	K ₂ [(Mo ₂ O ₅)(O ₃ P(C ₆ H ₄)PO ₃)](10)	12.34	12.58	0.690	0.85	-	-
11	Rb ₂ [(Mo ₂ O ₅)(O ₃ P(C ₆ H ₄)PO ₃)](11)	10.64	10.73	0.596	0.37	-	-
12	Cs ₂ [(Mo ₂ O ₅)(O ₃ P(C ₆ H ₄)PO ₃)](12)	9.34	9.48	0.522	0.24	-	-
13	(NH ₄) ₂ [(Mo ₂ O ₅)(O ₃ P(C ₆ H ₄)PO ₃)](13)	13.29	13.37	2.23	1.13	5.17	5.31
14	Tl ₂ [(Mo ₂ O ₅)(O ₃ P(C ₆ H ₄)PO ₃)](14)	7.88	7.03	0.441	0.13	-	-
15	K ₂ [(Mo ₂ O ₅)(O ₃ P(CH ₂) ₄ PO ₃)](15)	8.52	9.07	1.43	0.94	-	-
16	Rb ₂ [(Mo ₂ O ₅)(O ₃ P(CH ₂) ₄ PO ₃)](16)	7.31	7.24	1.23	0.85	-	-
17	Cs ₂ [(Mo ₂ O ₅)(O ₃ P(CH ₂) ₄ PO ₃)](17)	6.39	6.74	1.07	0.24	-	-
18	(NH ₄) ₂ [(Mo ₂ O ₅)(O ₃ P(CH ₂) ₄ PO ₃)](18)	9.20	9.14	3.09	2.08	5.37	5.27
19	Rb ₂ [(Mo ₂ O ₅)(O ₃ P(CH ₂) ₃ PO ₃)](19)	5.61	5.59	0.94	0.57	-	-
20	Cs ₂ [(Mo ₂ O ₅)(O ₃ P(CH ₂) ₃ PO ₃)](20)	4.88	5.16	0.82	0.42	-	-

The carbon contents of all compounds except **4** agree with the calculated values. The hydrogen contents of all compounds except **4** reasonably agree with the calculated values.

Table S19. Compilation of molecular and non-molecular molybdoorganophosphonates

Molecular molybdoorganophosphonates		
Sl.No	Compound	Reference
1	[Cu(bpy)(MoO ₂)(H ₂ O)(O ₃ PCH ₂ PO ₃)]	42
2	[{Cu ₂ (bpyr)}(MoO ₂)(HO ₃ PCH ₂ PO ₃) ₂].2H ₂ O	65
3	[Cu(bpy)(Mo ₂ O ₅)(O ₃ PCH ₂ CH ₂ CH ₂ PO ₃)]	42
4	[Cu(bpy)(Mo ₂ O ₅)(H ₂ O)(O ₃ PCH ₂ PO ₃)]·H ₂ O	42
5	[Cu(o-phen)(Mo ₂ O ₅)(O ₃ PCH ₂ PO ₃)(H ₂ O)]	43
6	[Cu(terpy)(Mo ₂ O ₅)(O ₃ PCH ₂ PO ₃)]	42
7	[{Ni(tpyprz)(H ₂ O) ₂ }(Mo ₃ O ₈) ₂ (O ₃ PCH ₂ PO ₃) ₂]	39
8	[{Cu ₂ (tpypyzy)(H ₂ O)}Mo ₃ O ₈ (HO ₃ PCH ₂ PO ₃) ₂].8H ₂ O	37
9	[{Cu ₂ (tpypyzy)(H ₂ O)} ₂ (Mo ₃ O ₈) ₂ (O ₃ PCH ₂ PO ₃) ₃].16.9H ₂ O	37
10	[Co(H ₂ tpyprz) ₂][(Mo ₃ O ₈) ₃ (O ₃ PCH ₂ PO ₃) ₃].7H ₂ O·C ₅ H ₅ N·py	36
11	[{Co(1,10-phen) ₂ }] ₂ {(Mo ₄ O ₁₂)(O ₃ PCH ₂ CH ₂ PO ₃)}·(H ₂ O)]	53
12	[{Co(1,10-phen) ₂ }] ₂ {(Mo ₄ O ₁₂)(O ₃ PCH ₂ CH ₂ CH ₂ PO ₃)}·(1.5H ₂ O)]	53
13	[{Cu(bpy)} ₂ (Mo ₄ O ₁₂)(H ₂ O) ₂ (O ₃ PCH ₂ CH ₂ PO ₃)]·2H ₂ O	42
14	[{Cu(o-phen)} ₂ (Mo ₄ O ₁₂)(H ₂ O) ₂ (O ₃ PCH ₂ CH ₂ PO ₃)]·2H ₂ O	42
15	[C(NH ₂) ₃] ₄ ·(Mo ₅ O ₂₁ (PhP) ₂)	44
16	((CH ₃ CH ₂ CH ₂) ₃ NH) ₄ [Mo ₅ O ₁₅ (CH ₃ PO ₃) ₂]	64
17	(C ₄ H ₁₂ N ₂) ₂ [Mo ₅ O ₁₅ (C ₆ H ₅ O ₃ P) ₂].2H ₂ O	45
18	[{Cu ₂ (L7)(H ₂ O) ₂ }Mo ₅ O ₁₅ {O ₃ P(CH ₂) ₄ PO ₃ }]·2H ₂ O	25
19	(4,4'-H ₂ dpa) ₂ [Mo ₅ O ₁₅ (1,2-O ₃ PC ₈ H ₈ PO ₃)]·H ₂ O	35
20	(4,4'-H ₂ dpa) ₂ [Mo ₅ O ₁₅ (1,3-O ₃ PC ₈ H ₈ PO ₃)]	35
21	[(HOOCCH ₂ H ₉ NCH ₂ PO ₃) ₂ Mo ₅ O ₁₅] ⁴⁻	51
22	[{Co ₄ (tpyprz) ₃ }(Mo ₅ O ₁₅) ₂ (O ₃ PCH ₂ CH ₂ PO ₃) ₂].18H ₂ O	36
23	[{Co(H ₂ tpyprz)}Mo ₅ O ₁₅ {O ₃ PCH ₂ CH ₂ PO ₃ }]	36

Sl.No	Compound	Reference
24	$[\{ \text{Co}_2(\text{tpyprz})_2 \} \text{Mo}_5\text{O}_{15} \{ \text{O}_3\text{P}(\text{CH}_2)_3\text{PO}_3 \}] \cdot 5\text{H}_2\text{O}$	36
25	$[\{ \text{Co}_2(\text{tpyprz})(\text{H}_2\text{O})_3 \} \text{Mo}_5\text{O}_{15} \{ \text{O}_3\text{P}(\text{CH}_2)_3\text{PO}_3 \}] \cdot 7\text{H}_2\text{O}$	36
26	$[\{ \text{Co}_2(\text{tpyprz})(\text{H}_2\text{O})_3 \} \text{Mo}_5\text{O}_{15} \{ \text{O}_3\text{P}(\text{CH}_2)_3\text{PO}_3 \}] \cdot \text{H}_2\text{O}$	36
27	$[\text{H}_3\text{O}]_2[\{ \text{Co}_3(\text{tpyprz})_2(\text{H}_2\text{O})_3 \} (\text{Mo}_5\text{O}_{15})_2 \{ \text{O}_3\text{P}(\text{CH}_2)_3\text{PO}_3 \}_2] \cdot 21.4\text{H}_2\text{O}$	36
28	$[\{ \text{Co}_2(\text{tpyprz})(\text{H}_2\text{O})_2 \} \text{Mo}_5\text{O}_{15} \{ \text{O}_3\text{P}(\text{CH}_2)_4\text{PO}_3 \}] \cdot 2\text{H}_2\text{O}$	36
29	$[\{ \text{Co}_2(\text{tpyprz})(\text{H}_2\text{O})_3 \} \text{Mo}_5\text{O}_{15} \{ \text{O}_3\text{P}(\text{CH}_2)_5\text{PO}_3 \}] \cdot 2.5\text{H}_2\text{O}$	36
30	$[\{ \text{Co}_2(\text{tpyprz})(\text{H}_2\text{O})_3 \} \text{Mo}_5\text{O}_{15} \{ \text{O}_3\text{P}(\text{CH}_2)_9\text{PO}_3 \}] \cdot 2.5\text{H}_2\text{O}$	36
31	$[\text{Co}(\text{bpy})_3][\text{Mo}_5\text{O}_{14}(\text{OH})\{ \text{HO}_3\text{P}(\text{CH}_2)_3\text{PO}_3 \}] \cdot \text{H}_2\text{O}$	36
32	$[\text{Co}(\text{terpy})_2][\text{Mo}_5\text{O}_{14}(\text{OH})\{ \text{HO}_3\text{P}(\text{CH}_2)_3\text{PO}_3 \}] \cdot \text{H}_2\text{O}$	36
33	$[\{ \text{Co}_2(\text{phen})_3(\text{H}_2\text{O}) \} \text{Mo}_5\text{O}_{15} \{ \text{O}_3\text{P}(\text{CH}_2)_3\text{PO}_3 \}]$	36
34	$[\{ \text{Co}_3(\text{tpyprz})_2(\text{H}_2\text{O})_3 \} [\text{Mo}_5\text{O}_{15} \{ \text{O}_3\text{P}(\text{CH}_2)_3\text{PO}_3 \}]_2]$	36
35	$[\{ \text{Co}(1,10\text{-phen})_2(\text{H}_2\text{O}) \}_2 \{ (\text{Mo}_5\text{O}_{15})(4,4'\text{-dbp}) \} \cdot (5.5\text{H}_2\text{O})]$	70
36	$[\text{Ni}(\text{H}_2\text{O})(\text{bipy})]_2(\text{C}_6\text{H}_5\text{PO}_3)_2\text{Mo}_5\text{O}_{15}]_n$	68
37	$[\text{Co}(\text{H}_2\text{O})(\text{bipy})]_2(\text{C}_6\text{H}_5\text{PO}_3)_2\text{Mo}_5\text{O}_{15}]_n$	68
38	$[\{ \text{Ni}(1,10\text{-phen})_2(\text{H}_2\text{O}) \}_2 \{ (\text{Mo}_5\text{O}_{15})(4,4'\text{-dbp}) \} \cdot (5.75\text{H}_2\text{O})]$ (4,4'-dbp = $\text{O}_3\text{PCH}_2\text{C}_6\text{H}_4\text{C}_6\text{H}_4\text{CH}_2\text{PO}_3$)	70
39	$\{ [\text{Ni}(2,2'\text{-bpy})_2]_2[(\text{C}_6\text{H}_5\text{P})_2\text{Mo}_5\text{O}_{21}] \text{H}_2\text{O} \} \cdot 2\text{H}_2\text{O}$ (bpy = bipyridine)	70
40	$[\{ \text{Ni}_4(\text{tpypyz})_3 \} \{ \text{Mo}_5\text{O}_{15}(\text{O}_3\text{PCH}_2\text{CH}_2\text{PO}_3) \}_2]$	37
41	$[\{ \text{Ni}_2(\text{bpyr})(\text{H}_2\text{O})_4 \} \text{Mo}_5\text{O}_{15} \{ \text{O}_3\text{P}(\text{CH}_2)_3\text{PO}_3 \}] \cdot 9\text{H}_2\text{O}$	65
42	$[\{ \text{Ni}_2(\text{tpyprz})_2 \} \text{Mo}_5\text{O}_{15} \{ \text{O}_3\text{P}(\text{CH}_2)_4\text{PO}_3 \}] \cdot 6.5\text{H}_2\text{O}$	38
43	$[\{ \text{Ni}_2(\text{tpyprz})_2 \} \text{Mo}_5\text{O}_{15} \{ \text{O}_3\text{P}(\text{CH}_2)_5\text{PO}_3 \}] \cdot 3.5\text{H}_2\text{O}$	38
44	$[\{ \text{Ni}_4(\text{tpyprz})_3 \} \{ \text{Mo}_5\text{O}_{15}(\text{O}_3\text{PCH}_2\text{CH}_2\text{PO}_3) \}_2] \cdot 23\text{H}_2\text{O}$	38
45	$[\{ \text{Ni}_3(\text{tpyprz})_2(\text{H}_2\text{O})_2 \} (\text{Mo}_5\text{O}_{15})(\text{Mo}_2\text{O}_4\text{F}_2) \{ \text{O}_3\text{P}(\text{CH}_2)_3\text{PO}_3 \}_2] \cdot 8\text{H}_2\text{O}$	38
46	$[\{ \text{Ni}_2(\text{tpyprz})(\text{H}_2\text{O})_3 \} \text{Mo}_5\text{O}_{15} \{ \text{O}_3\text{P}(\text{CH}_2)_3\text{PO}_3 \}] \cdot \text{H}_2\text{O}$	38
47	$[\{ \text{Ni}_2(\text{tpyprz})(\text{H}_2\text{O})_2 \} \text{Mo}_5\text{O}_{15} \{ \text{O}_3\text{P}(\text{CH}_2)_4\text{PO}_3 \}] \cdot 2.25\text{H}_2\text{O}$	38
48	$[\{ \text{Cu}(\text{terpy}) \}_2(\text{Mo}_5\text{O}_{15})(\text{O}_3\text{PCH}_2\text{CH}_2\text{CH}_2\text{PO}_3)]$	42
49	$[\text{Ni}(\text{H}_2\text{O})_4(4,4'\text{-Hbpy})] \text{Mo}_5\text{O}_{15}(1,2\text{-O}_3\text{PC}_6\text{H}_8\text{PO}_3) \cdot 4\text{H}_2\text{O}$	35
50	$[\text{Ni}(\text{phen})_2(\text{H}_2\text{O})_2][\text{Ni}(\text{phen})_2\text{Mo}_5\text{O}_{15}\{ \text{HO}_2\text{CC}_6\text{H}_4\text{PO}_3 \}_2] \cdot 2\text{H}_2\text{O}$	33

Sl.No	Compound	Reference
51	$[\{\text{Ni}(\text{H}_2\text{O})(2,2'\text{-dpa})\}_2\text{Mo}_5\text{O}_{15}(1,4\text{-O}_3\text{PC}_8\text{H}_8\text{PO}_3)] \cdot 2\text{H}_2\text{O}$	35
52	$[\text{Ni}(2,2'\text{-bpy})_3]_2[\text{Mo}_5\text{O}_{21}(\text{C}_6\text{H}_5\text{P})_2] \cdot 2\text{H}_2\text{O}$	34
53	$\{[\text{Ni}(2,2'\text{-bpy})_2]_2[\text{Mo}_5\text{O}_{21}(\text{C}_6\text{H}_5\text{P})_{21}] \cdot \text{H}_2\text{O}\} \cdot 2\text{H}_2\text{O}$	34
54	$\{\text{Cu}_2(\text{L5})\}\text{Mo}_6\text{O}_{18}(\text{H}_2\text{O})\{\text{O}_3\text{P}(\text{CH}_2)_4\text{PO}_3\} \cdot 14\text{H}_2\text{O}$	25
55	$\{[\text{Cu}(2,2'\text{-bpy})_2]_2[\text{Mo}_5\text{O}_{21}(\text{C}_6\text{H}_5\text{P})_{21}]\} \cdot \text{H}_2\text{O}$	34
56	$[\{\text{Ni}(2,2'\text{-bpy})_3\}\{\text{Ni}(2,2'\text{-bpy})_2(\text{H}_2\text{O})\}\{(\text{Mo}_5\text{O}_{15})(4,4'\text{-dbp})\} \cdot (4.75\text{H}_2\text{O})]$	70
57	$[\text{Cu}(\text{terpy})_2](\text{H}_3\text{O})[\text{Mo}_5\text{O}_{15}(1,4\text{-O}_3\text{PC}_8\text{H}_8\text{PO}_3)] \cdot \text{H}_2\text{O}$	35
58	$[\{\text{Cu}(2,2'\text{-bpy})_2\}_2\text{Mo}_5\text{O}_{15}(1,4\text{-O}_3\text{PC}_8\text{H}_8\text{PO}_3)] \cdot 6\text{H}_2\text{O}$	35
59	$[\{\text{Cu}(\text{H}_2\text{O})_2(\text{o-phen})\}_2\text{Mo}_5\text{O}_{15}(1,4\text{-HO}_3\text{PC}_8\text{H}_8\text{PO}_3)] \cdot 4\text{H}_2\text{O}$	35
60	$[(\text{Cu}(\text{H}_2\text{O}))_2(\mu\text{-bipy})_2\text{Mo}_5\text{O}_{15}(\text{C}_6\text{H}_5\text{PO}_3)_2]_n$	46
61	$\{[\text{Cu}(2,2'\text{-bpy})_2]_2[\text{C}_6\text{H}_5\text{P})_2\text{Mo}_5\text{O}_{21}]\} \cdot \text{H}_2\text{O}$	34
62	$[\{\text{Cu}(\text{bpy})(\text{H}_2\text{O})\}_3\{\text{O}_2\text{CC}_6\text{H}_4\text{PO}_3\}_2\text{Mo}_5\text{O}_{15}] \cdot 4.5\text{H}_2\text{O}$	33
63	$[\{\text{Cu}_2(\text{bpyr})_2\}\text{Mo}_5\text{O}_{15}(\text{O}_3\text{PCH}_3)_2] \cdot 2.5\text{H}_2\text{O}$	65
64	$[\{\text{Cu}_2(\text{bpyr})(\text{H}_2\text{O})\}\text{Mo}_5\text{O}_{15}(\text{O}_3\text{PCH}_2\text{CH}_3)_2]$	65
65	$\{[\text{Cu}_2(\text{bpyr})]\text{Mo}_5\text{O}_{15}\{\text{O}_3\text{P}(\text{CH}_2)_n\text{PO}_3\}\} \cdot x\text{H}_2\text{O } n = 2-6$	65
66	$[\{\text{Cu}_2(\text{bpyr})(\text{H}_2\text{O})_4\}\text{Mo}_5\text{O}_{15}\{\text{O}_3\text{P}(\text{CH}_2)_3\text{PO}_3\}] \cdot 3\text{H}_2\text{O}$	65
67	$[\{\text{Cu}_2(\text{tpypyzy})(\text{H}_2\text{O})_3\}\text{Mo}_5\text{O}_{15}(\text{HPO}_4)(\text{O}_3\text{PCH}_2\text{CO}_2\text{H})] \cdot \text{H}_2\text{O}$	37
68	$[\{\text{Cu}_2(\text{tpypyzy})(\text{H}_2\text{O})\}\text{Mo}_5\text{O}_{15}(\text{O}_3\text{PC}_6\text{H}_5)_2] \cdot 2\text{H}_2\text{O}$	37
69	$[\{\text{Cu}_2(\text{tpypyzy})(\text{H}_2\text{O})_2\}\text{Mo}_5\text{O}_{15}\{\text{O}_3\text{P}(\text{CH}_2)_n\text{PO}_3\}] \cdot x\text{H}_2\text{O}$	37
70	$[\{\text{Co}(1,10\text{-phen})_2(\text{H}_2\text{O})_2\}\{\text{Co}(1,10\text{-phen})_2(\text{H}_2\text{O})\}\{(\text{Mo}_5\text{O}_{15})(\text{O}_3\text{PCH}_2\text{CH}_2\text{CH}_2\text{CH}_2\text{PO}_3)\} \cdot (6\text{H}_2\text{O})]$	53
71	$[\{\text{Cu}_2(\text{tpypyzy})(\text{H}_2\text{O})_2\}(\text{Mo}_5\text{O}_{15})(\text{HOPO}_3)_2] \cdot n\text{H}_2\text{O}$	40
72	$[\{\text{Cu}(\text{H}_2\text{O})_2(\text{o-phen})\}\{\text{Cu}(\text{o-phen})_2\}(\text{Mo}_5\text{O}_{15})(\text{O}_3\text{PCH}_2\text{CH}_2\text{CH}_2\text{PO}_3)] \cdot 2.5\text{H}_2\text{O}$	43
73	$[\{\text{Cu}(\text{bpy})_2\}\{\text{Cu}(\text{bpy})(\text{H}_2\text{O})\}(\text{Mo}_5\text{O}_{15})(\text{O}_3\text{PCH}_2\text{CH}_2\text{CH}_2\text{CH}_2\text{PO}_3)] \cdot \text{H}_2\text{O}$	42
74	$[\{\text{Cu}(\text{terpy})(\text{H}_2\text{O})\}_2(\text{Mo}_5\text{O}_{15})(\text{O}_3\text{PCH}_2\text{CH}_2\text{PO}_3)] \cdot 3\text{H}_2\text{O}$	42
75	$\{[\text{Cu}(2,20\text{-bpy})_2]_2[\text{Mo}_5\text{O}_{21}(\text{C}_6\text{H}_5\text{P})_{21}]\} \cdot \text{H}_2\text{O}$	43
76	$[(\text{Zn}(\text{bipy})(\mu\text{-OH}))_2(\text{Zn}(\text{bipy})_2\text{Mo}_5\text{O}_{15}(\text{C}_6\text{H}_5\text{PO}_3)_2)_2] \cdot 3\text{H}_2\text{O}$	42
77	$[(\text{Cu}(\text{bipy})_2)_2\text{Mo}_5\text{O}_{15}(\text{C}_6\text{H}_5\text{PO}_3)_2] \cdot 2\text{H}_2\text{O}$	42

Sl.No	Compound	Reference
78	$[(\text{Ni}(\text{H}_2\text{O})(\text{bipy}))_2 \text{Mo}_5\text{O}_{15}(\text{C}_6\text{H}_5\text{PO}_3)_2]_n$	68
79	$[(\text{Co}(\text{H}_2\text{O})(\text{bipy}))_2 \text{Mo}_5\text{O}_{15}(\text{C}_6\text{H}_5\text{PO}_3)_2]_n$	68
80	$[(\text{Cu}(\text{bipy})_2)_2 \text{Mo}_5\text{O}_{15}(\text{C}_6\text{H}_5\text{PO}_3)_2] \cdot 2\text{H}_2\text{O}$ (bipy = 2,2'-bipyridine)	68
81	$[(\text{Zn}(\text{bipy})(\mu\text{-OH}))_2(\text{Zn}(\text{bipy})_2(\text{C}_6\text{H}_5\text{PO}_3)_2\text{Mo}_5\text{O}_{15})_2] \cdot 3\text{H}_2\text{O}$ (bipy = 2,2'-bipyridine)	68
82	$[(\text{nC}_4\text{H}_9)_4\text{N}]_2[(\text{Mo}_5\text{O}_{15}\text{NH}_2\text{CH}_2\text{PO}_3)_2] \cdot (\text{CH}_3\text{NHCH}_3)_2 \cdot \text{H}_2\text{O}$	47
83	$[\text{H}_4\text{tphz}]_3[\{\text{Zn}_2(\text{tphz})(\text{H}_2\text{O})_4\}\{\text{Mo}_5\text{O}_{15}(\text{O}_3\text{PCH}_2\text{CH}_2\text{PO}_3)_4\}] \cdot 14\text{H}_2\text{O}$	39
84	$[\text{Ni}_x(4\text{-Hpt})_y(\text{H}_2\text{O})_z\{\text{Mo}_5\text{O}_{15}(\text{O}_3\text{PCH}_2\text{CH}_2\text{PO}_3)_3\}] \cdot n\text{hydrate}$ (x = 2, y = 2, z = 6)	32
85	$[\text{Co}_x(4\text{-Hpt})_y(\text{H}_2\text{O})_z\{\text{Mo}_5\text{O}_{15}(\text{O}_3\text{PCH}_2\text{CH}_2\text{PO}_3)_3\}] \cdot n\text{hydrate}$ (x = 2, y = 2, z = 6)	32
86	$[\text{Cu}_x(4\text{-Hpt})_y(\text{H}_2\text{O})_z\{\text{Mo}_5\text{O}_{15}(\text{O}_3\text{PCH}_2\text{CH}_2\text{PO}_3)_3\}] \cdot n\text{hydrate}$ (x = 4, y = 3, z = 1)	32
87	$[\text{NEt}_4]_2\text{Na}_3(\text{H}_3\text{O})_4\{\text{Na}[\text{Mo}_6\text{O}_{15}(\text{O}_3\text{PPh})(\text{HO}_3\text{PPh})_3]_2\} \cdot n\text{H}_2\text{O}$	31
88	$[\{\text{Co}_2(\text{tpyprz})(\text{H}_2\text{O})_2\}\text{Mo}_6\text{O}_{18}\{\text{O}_3\text{P}(\text{CH}_2)_5\text{PO}_3\}] \cdot 2\text{H}_2\text{O}$	36
89	$[\text{Ni}_2(\text{tpyprz})_2][\text{Mo}_7\text{O}_{21}(\text{O}_3\text{PCH}_2\text{PO}_3)] \cdot 3.5\text{H}_2\text{O}$	39
90	$\{[\text{Ni}(1,10\text{-phen})_3][\text{Ni}(1,10\text{-phen})_2(\text{Mo}_6\text{O}_{18})(\text{O}_3\text{P}(\text{CH}_2)_4\text{PO}_3)] \cdot 6\text{H}_2\text{O}\}_n$	55
92	$\{[\text{Ni}(1,10\text{-phen})_2(\text{H}_2\text{O})_2][\text{Ni}(1,10\text{-phen})_2(\text{H}_2\text{O})(\text{Mo}_5\text{O}_{15})(\text{O}_3\text{P}(\text{CH}_2)_4\text{PO}_3)] \cdot 6\text{H}_2\text{O}\}_n$	55
91	$(\text{Et}_3\text{NH})_2[\text{Mo}_2\text{O}_5\{\text{C}_6\text{H}_5\}_2] \cdot \text{CH}_3\text{CN}$	37
93	$(\text{NH}_4)_4[\text{Mo}_5\text{O}_{21}(\text{CH}_3\text{P})_2] \cdot 5\text{H}_2\text{O}$	57
94	$(\text{PhNMe}_3)_4[(\text{O}_3\text{POPO}_3)\text{Mo}_6\text{O}_{18}(\text{H}_2\text{O})_4]$	54
95	$(\text{ppzH}_2)_4[\text{Mo}_7\text{O}_{16}(\text{O}_3\text{PCH}_2\text{PO}_3)_3] \cdot 8\text{H}_2\text{O}$	30
96	$(\text{Et}_3\text{NH})_4[\text{Mo}_6\text{O}_{18}(\text{BuPO}_3)_2]$	64
97	$((\text{enH}_2)_4[\text{Mo}_7\text{O}_{16}(\text{O}_3\text{PCH}_2\text{PO}_3)_3] \cdot 7\text{H}_2\text{O})$	30
98	$\text{Na}_8[(\text{Mo}_3\text{O}_8)\{\text{O}_3\text{PC}(\text{O})(\text{CH}_2\text{-3-C}_5\text{NH}_5)\text{PO}_3\}]_4 \cdot 30\text{H}_2\text{O}$	28
99	$\text{Rb}_4\text{KNa}[\text{Mo}_5\text{O}_{15}(\text{O}_2\text{CCH}_2\text{PO}_3)_2] \cdot \text{H}_2\text{O}$	26
100	$(\text{NH}_4)_2[\text{Mo}_5\text{O}_{21}(\text{PCH}_2\text{C}_6\text{H}_4\text{NH}_3)_2]$	28
101	$\text{Cs}_4[(\text{O}_3\text{PCH}_2\text{PO}_3)\text{Mo}_6\text{O}_{18}(\text{H}_2\text{O})_4] \cdot 7\text{H}_2\text{O}$	27
102	$\text{Rb}_4[(\text{MoO}_3)_5(\text{O}_3\text{PCH}_2\text{Ph})_2] \cdot 3\text{H}_2\text{O}$	Present work
103	$\text{Cs}_4[(\text{MoO}_3)_5(\text{O}_3\text{PCH}_2\text{Ph})_2] \cdot 3\text{H}_2\text{O}$	Present work
104	$(\text{NH}_4)_4[(\text{MoO}_3)_5(\text{O}_3\text{PPh})_2] \cdot 6\text{H}_2\text{O}$	Present work
105	$\text{Cs}_4[(\text{MoO}_3)_5(\text{O}_3\text{PPh})_2] \cdot 6\text{H}_2\text{O}$	Present work

Non-molecular molybdoorganophosphonates		
Sl.No	Compound	Reference
106	$\text{Cs}_2(\text{MoO}_3)_3\text{PO}_3\text{CH}_3$	56
107	$\text{Rb}_2(\text{MoO}_3)_3\text{PO}_3\text{CH}_3$	56
108	$\text{Tl}_2(\text{MoO}_3)_3\text{PO}_3\text{CH}_3$	50
109	$\text{Rb}[\text{MoO}_2(\text{O}_3\text{PCH}_2\text{PO}_3\text{H})]$	59
110	$\text{Cs}[\text{MoO}_2(\text{O}_3\text{PCH}_2\text{PO}_3\text{H})]$	48
111	$\text{NH}_4[\text{MoO}_2(\text{O}_3\text{PCH}_2\text{PO}_3\text{H})]$	59
112	$\text{Tl}[\text{MoO}_2(\text{O}_3\text{PCH}_2\text{PO}_3\text{H})]$	59
113	$\text{NH}_4[\text{Mo}_2\text{O}_5(\text{O}_3\text{PCH}_2\text{CH}_2\text{PO}_3)]$	60
114	$\text{K}(\text{H}_3\text{O})[\text{Mo}_2\text{O}_5(\text{O}_3\text{PCH}_2\text{CH}_2\text{PO}_3)]$	60
115	$(\text{K})_{1.5}(\text{H}_3\text{O})_{0.5}[(\text{Mo}_2\text{O}_5)(\text{O}_3\text{PPh})_2]$	Present work
116	$\text{Rb}_2[(\text{Mo}_2\text{O}_5)(\text{O}_3\text{PC}_6\text{H}_4\text{PO}_3)]$	Present work
117	$(\text{NH}_4)_2[(\text{Mo}_2\text{O}_5)(\text{O}_3\text{PC}_6\text{H}_4\text{PO}_3)]$	Present work
118	$\text{Rb}_2[(\text{Mo}_2\text{O}_5)(\text{O}_3\text{PC}_4\text{H}_8\text{PO}_3)]$	Present work
119	$\text{Rb}_2[(\text{Mo}_2\text{O}_5)(\text{O}_3\text{PC}_3\text{H}_6\text{PO}_3)]$	Present work

Table S20. Red-ox potentials (V) for compounds **1-5**, **11**, **16** and **20**.

Compound	KNO ₃		KNO ₃		NaOAc-AcOH	
	Reduction I	Oxidation I	Reduction II	Oxidation II	Reduction II	Oxidation II
1	-0.39	-0.49	-1.04	-0.93	-0.55	-0.32
2	-0.44	-0.47	-0.77	-0.91	-0.58	-0.34
3	-	-	-0.90	0.40	-0.69	-0.36
4	-	-	-0.92	-0.33	-0.93	-0.52
5	-	-	-0.84	-0.33	-0.93	-0.29
11	-	-	-0.82	-	-0.80	-
16	-	-	-0.78	-	-0.73	-
20	-	-	-0.70	-	-0.73	-

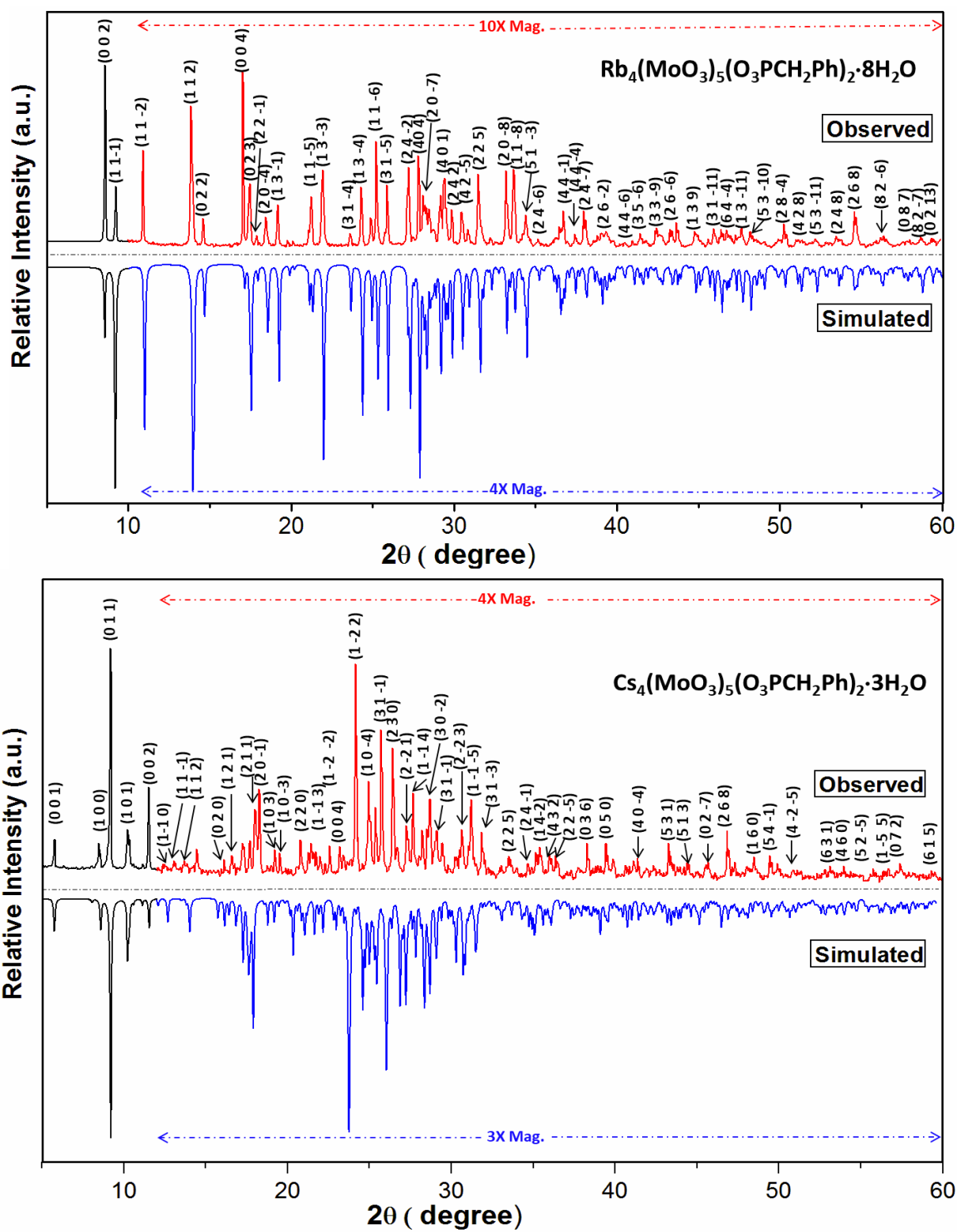


Figure S1. Simulated and observed powder XRD patterns of (top) $\text{Rb}_4[(\text{MoO}_3)_5(\text{O}_3\text{PCH}_2\text{Ph})_2] \cdot 8\text{H}_2\text{O}$ (**1**) and (bottom) $\text{Cs}_4[(\text{MoO}_3)_5(\text{O}_3\text{PCH}_2\text{Ph})_2] \cdot 3\text{H}_2\text{O}$ (**2**).

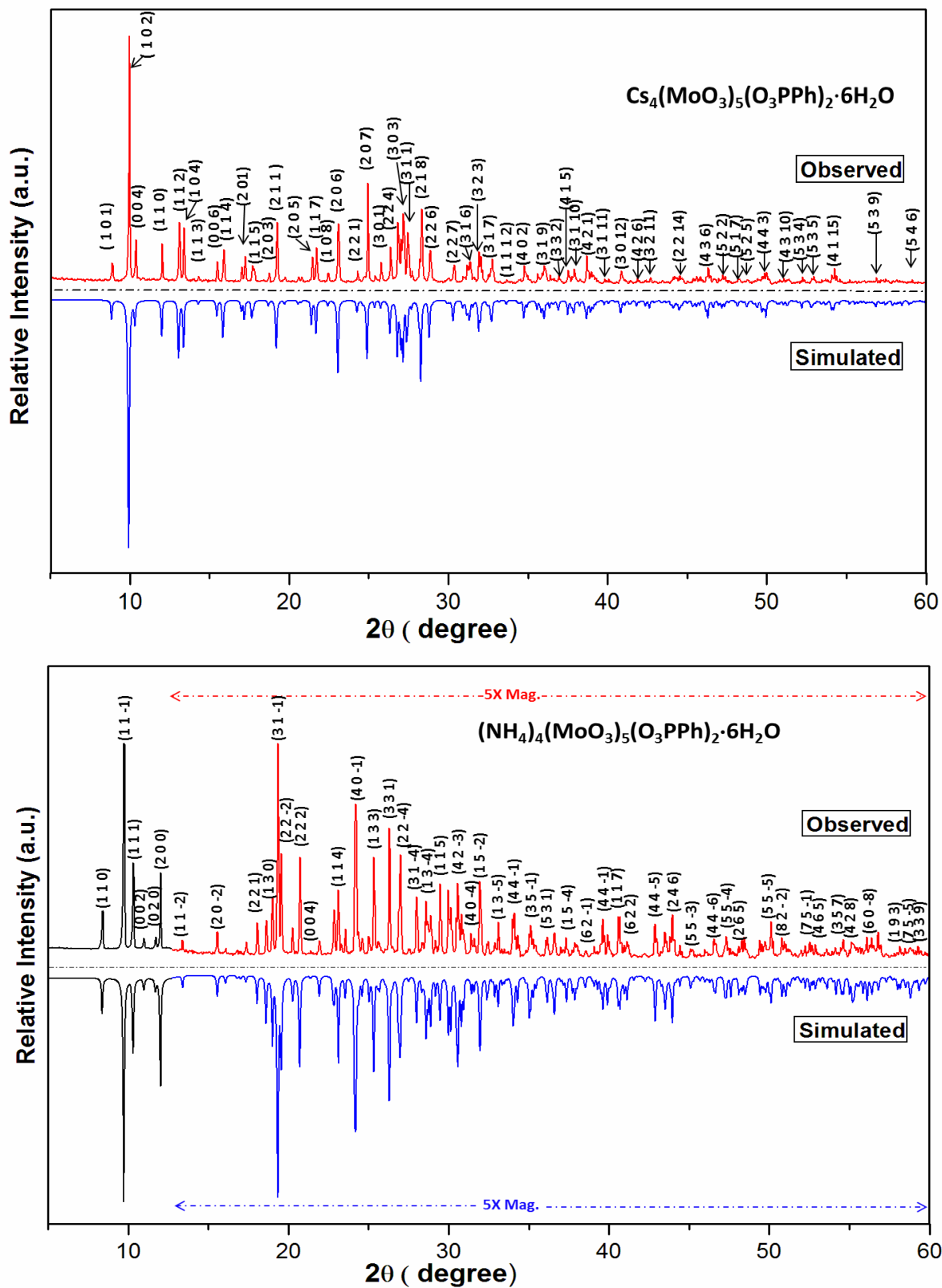


Figure S2. Simulated and observed powder XRD patterns of (top) $\text{Cs}_4(\text{MoO}_3)_5(\text{O}_3\text{PPh})_2 \cdot 6\text{H}_2\text{O}$ (3) and (bottom) $(\text{NH}_4)_4[(\text{MoO}_3)_5(\text{O}_3\text{PPh})_2] \cdot 6\text{H}_2\text{O}$ (4).

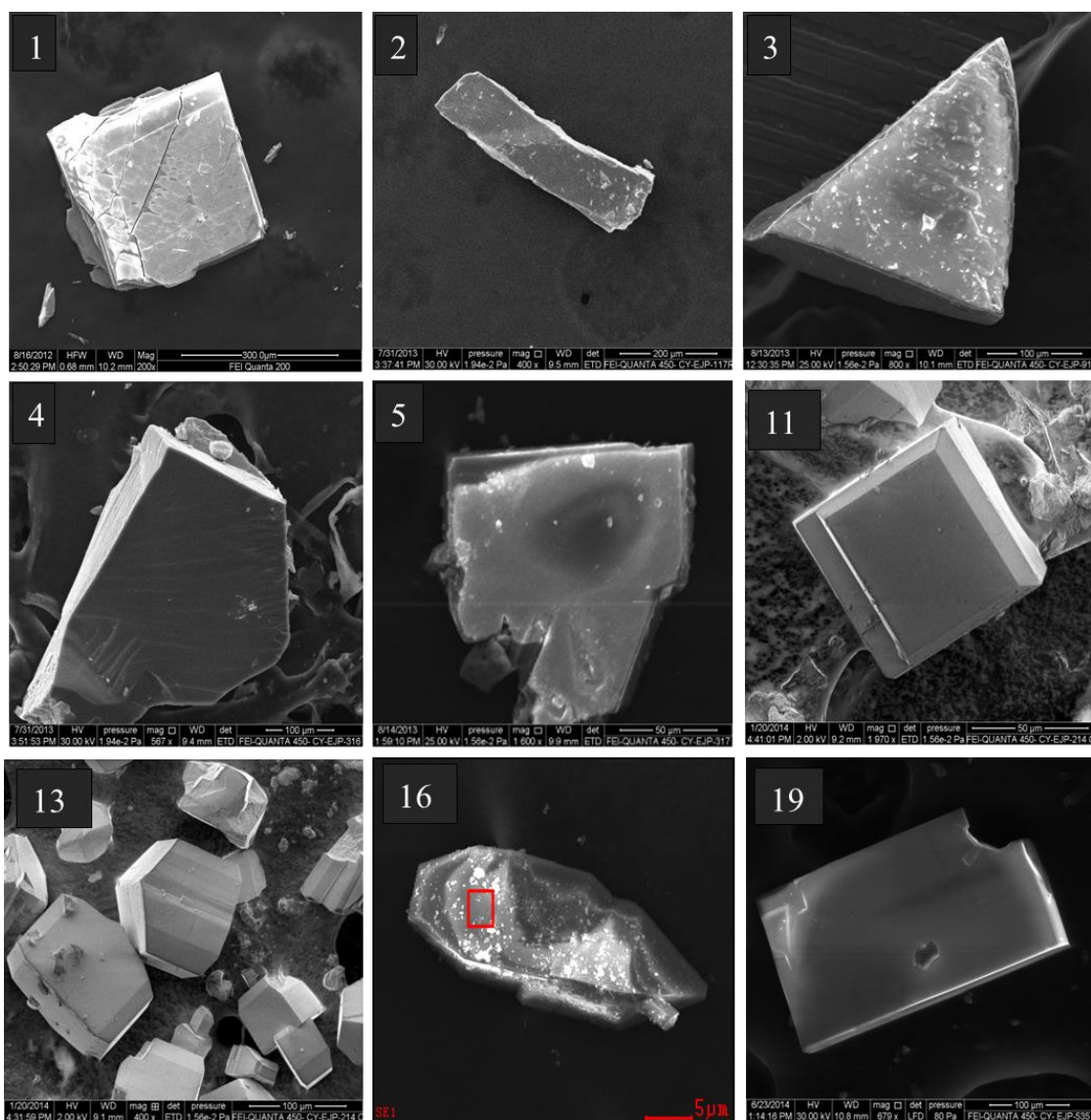


Figure S3. Scanning electron microscope (SEM) images of the crystals of nine compounds, 1- 4, 5, 11, 13, 16 and 19.

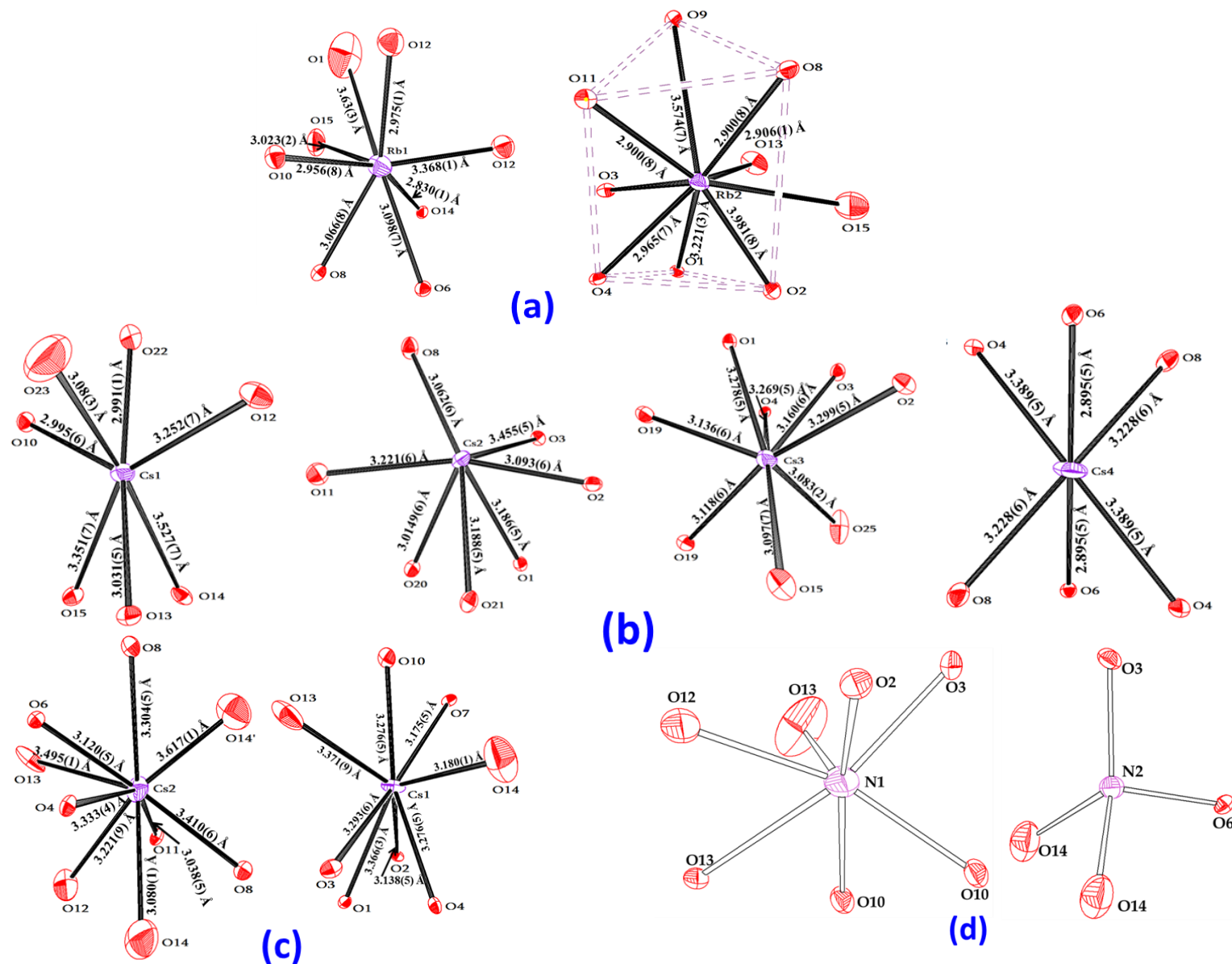


Figure S4. ORTEP diagram showing the atom labelling scheme and coordination of A^+ ions in (a) $Rb_4[(MoO_3)_5(O_3PCH_2Ph)_2] \cdot 8H_2O(1)$, (b) $Cs_4[(MoO_3)_5(O_3PCH_2Ph)_2] \cdot 3H_2O(2)$, (c) $Cs_4[(MoO_3)_5(O_3PPh)_2] \cdot 6H_2O(3)$ and (d) $(NH_4)_4[(MoO_3)_5(O_3PPh)_2] \cdot 6H_2O(4)$ compounds.

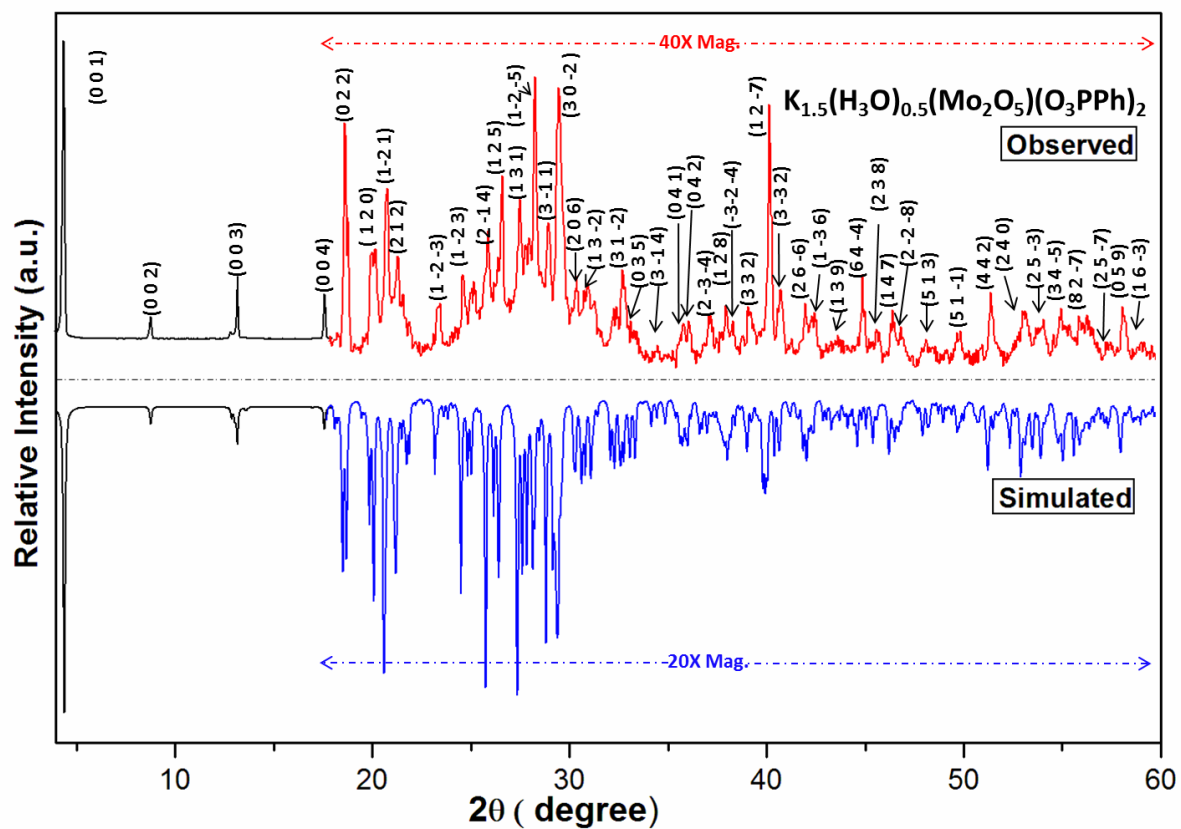


Figure S5. Simulated and observed powder XRD patterns of $K_{1.5}(H_3O)_{0.5}[(Mo_2O_5)(O_3PPh)_2](5)$

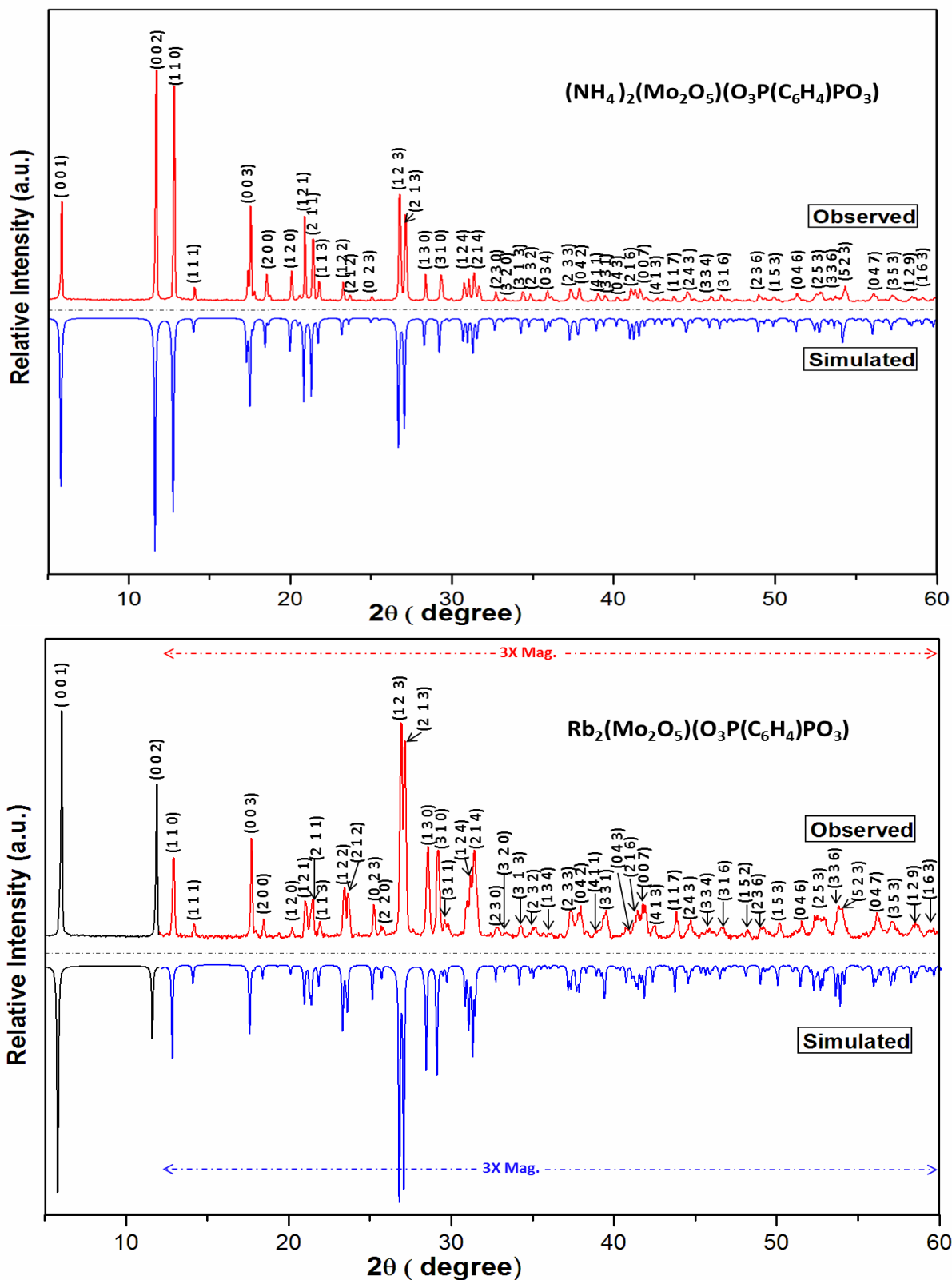


Figure S6. Simulated and observed powder XRD patterns of (top) $(\text{NH}_4)_2[(\text{Mo}_2\text{O}_5)(\text{O}_3\text{P}(\text{C}_6\text{H}_4)\text{PO}_3)]$ (**11**) and (bottom) $\text{Rb}_2[(\text{Mo}_2\text{O}_5)(\text{O}_3\text{P}(\text{C}_6\text{H}_4)\text{PO}_3)]$ (**13**).

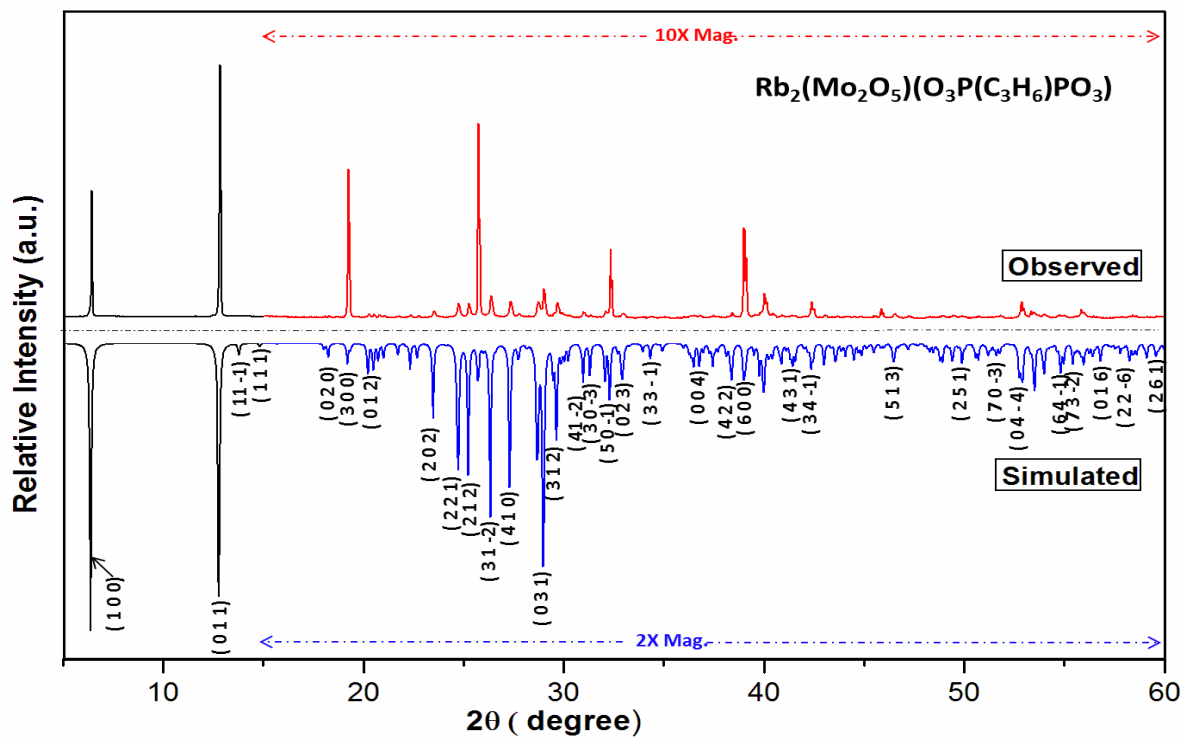
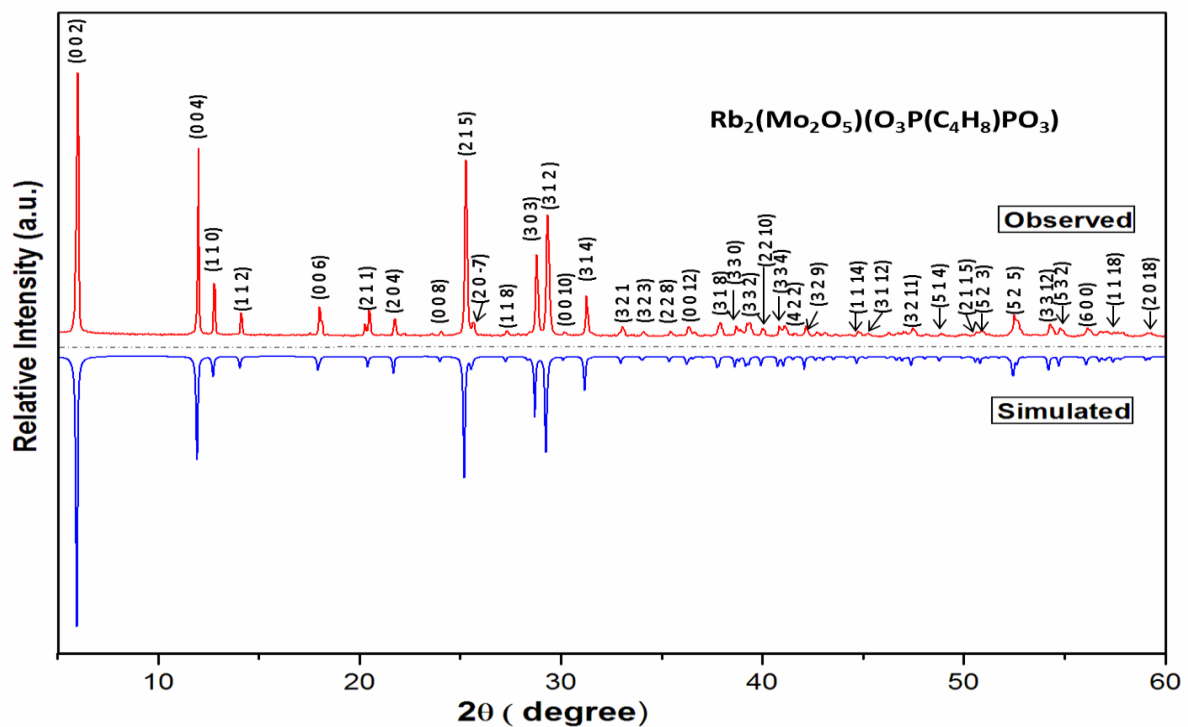


Figure S7. Simulated and observed powder XRD patterns of (top) $\text{Rb}_2[(\text{Mo}_2\text{O}_5)(\text{O}_3\text{P}(\text{C}_4\text{H}_8)\text{PO}_3)]$ (**16**) and (bottom) $\text{Rb}_2[(\text{Mo}_2\text{O}_5)(\text{O}_3\text{P}(\text{C}_3\text{H}_6)\text{PO}_3)]$ (**19**).

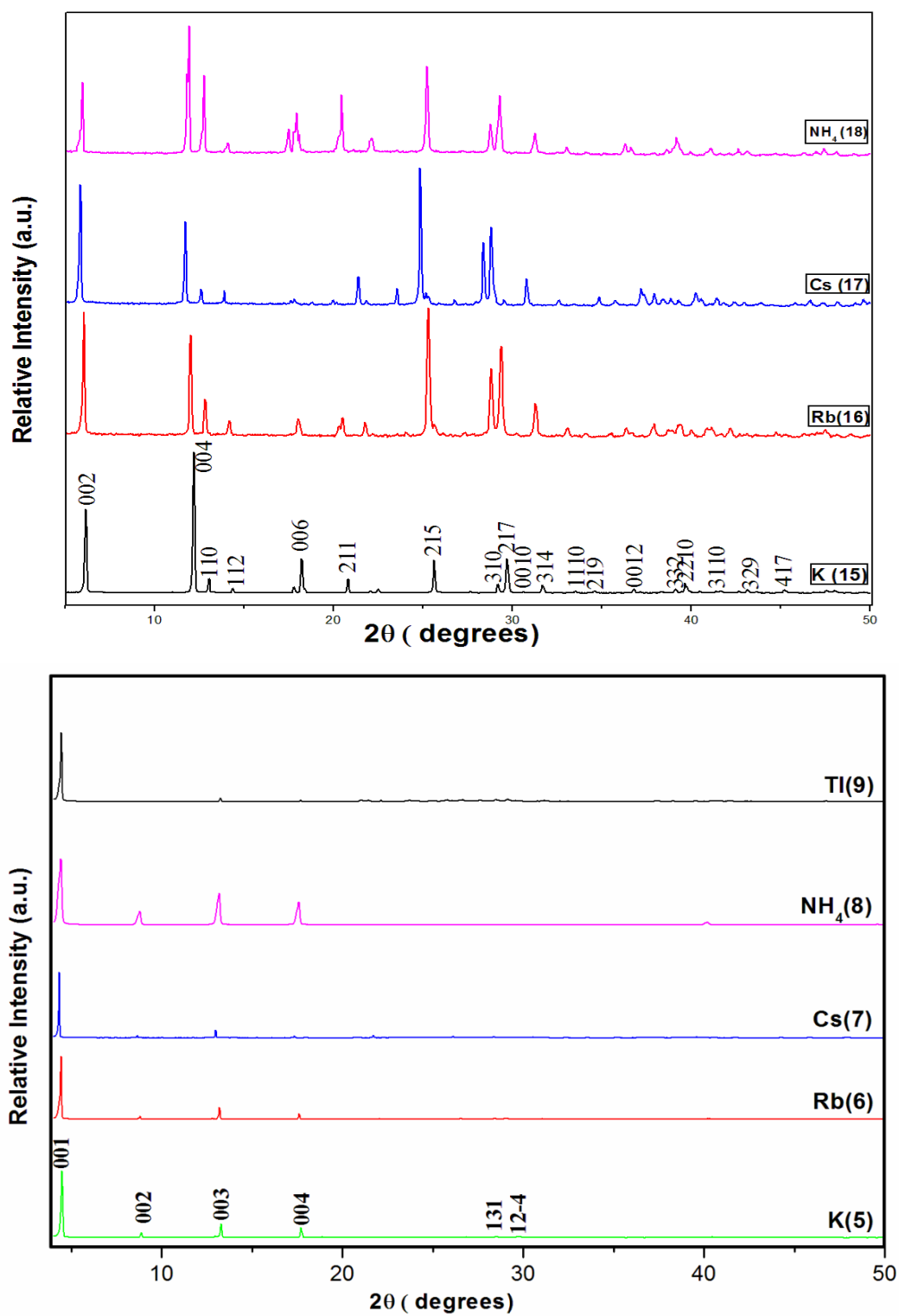


Figure S8. Powder XRD patterns of (top) butylenediphosphonates 15-18 and (bottom) phenylphosphonates 5-9

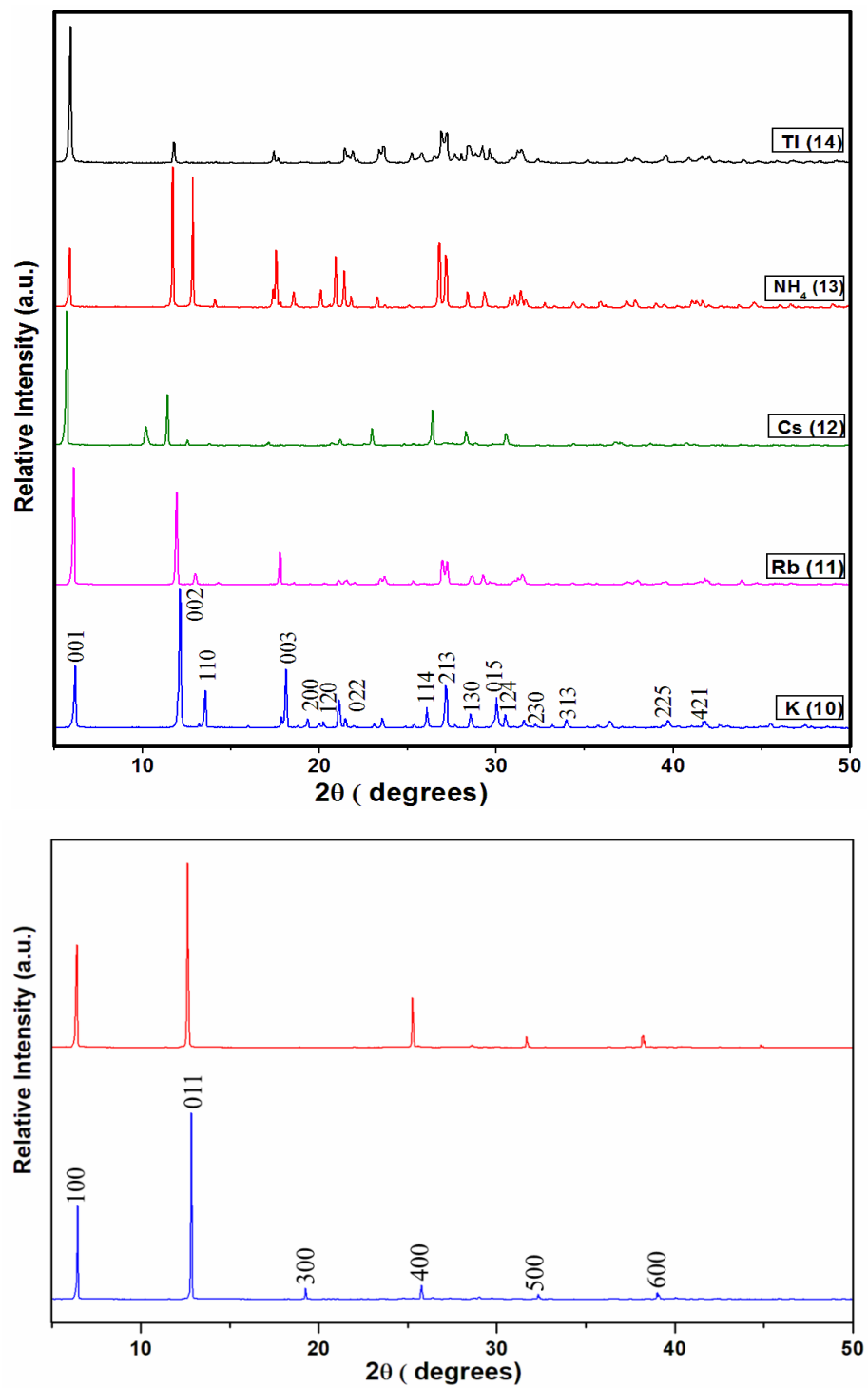


Figure S9. Powder XRD patterns of (top) phenylenediphosphonates **10-14** and (bottom) propylenediphosphonates **19** and **20**

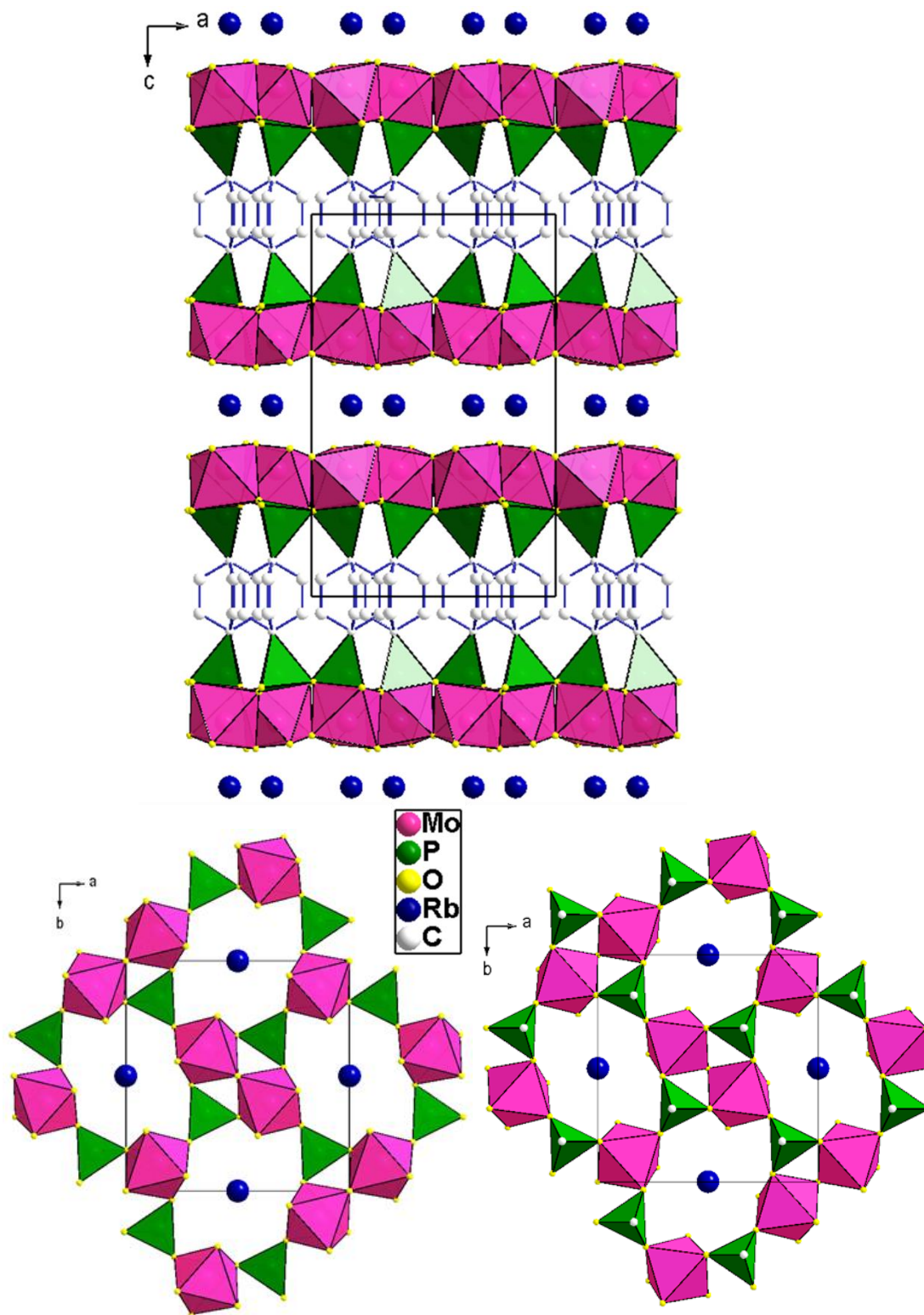


Figure S10. Polyhedral representation of (top) unit cell structure viewed along *a*-axis and (bottom) sublayers of $[\text{Mo}_2\text{O}_5(\text{O}_3\text{P}(\text{C}_6\text{H}_4)\text{PO}_3)]^{2-}$ anion viewed along *c*-axis of $\text{Rb}_2[\text{Mo}_2\text{O}_5(\text{O}_3\text{P}(\text{C}_6\text{H}_4)\text{PO}_3)]$ (13).

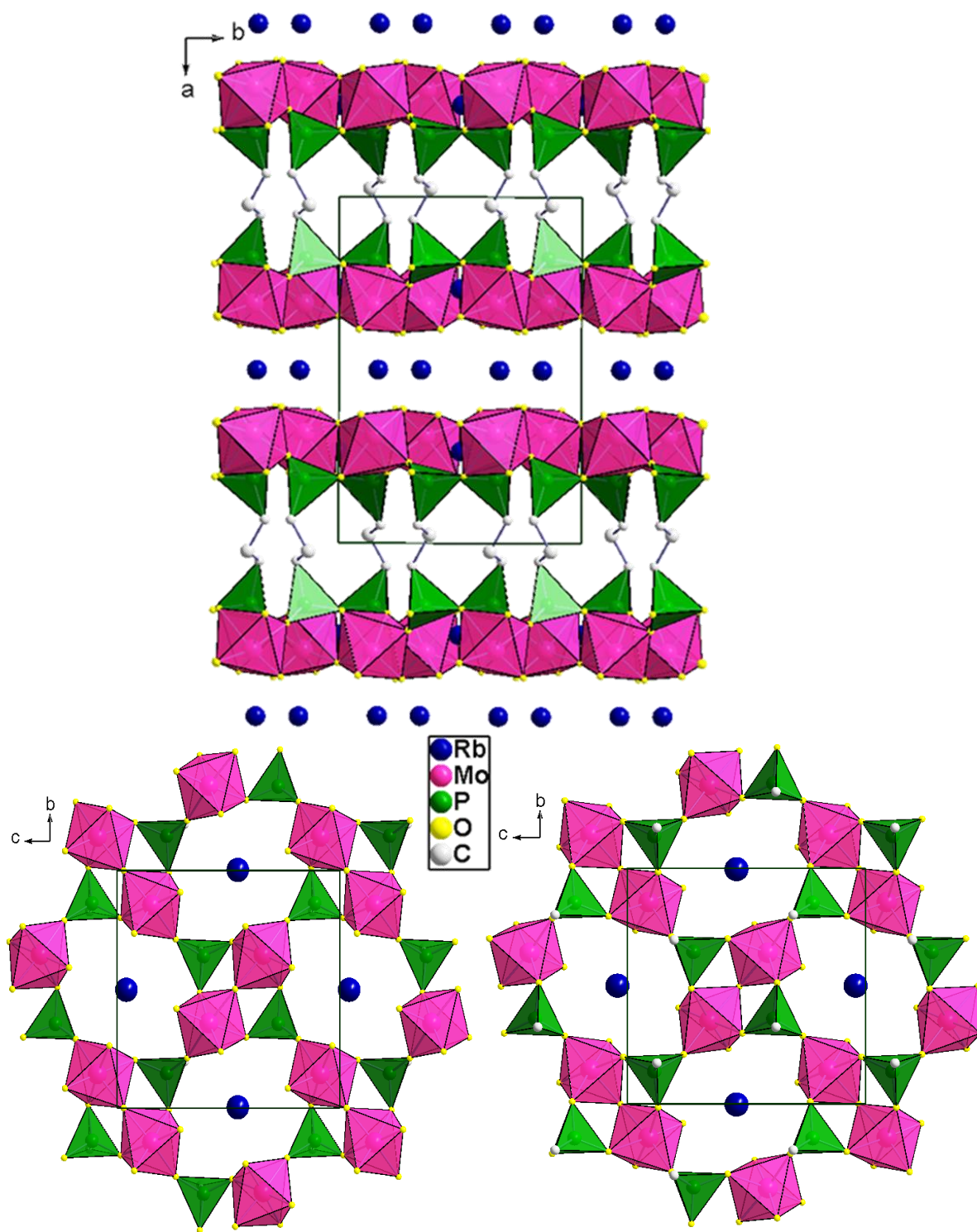


Figure S11. Polyhedral representation of (top) unit cell structure viewed along c -axis and (bottom) sublayers of $[\text{Mo}_2\text{O}_5(\text{O}_3\text{P}(\text{CH}_2)_3\text{PO}_3)]^{2-}$ anion viewed along a -axis, of $\text{Rb}_2[\text{Mo}_2\text{O}_5(\text{O}_3\text{P}(\text{CH}_2)_3\text{PO}_3)]$ (19).

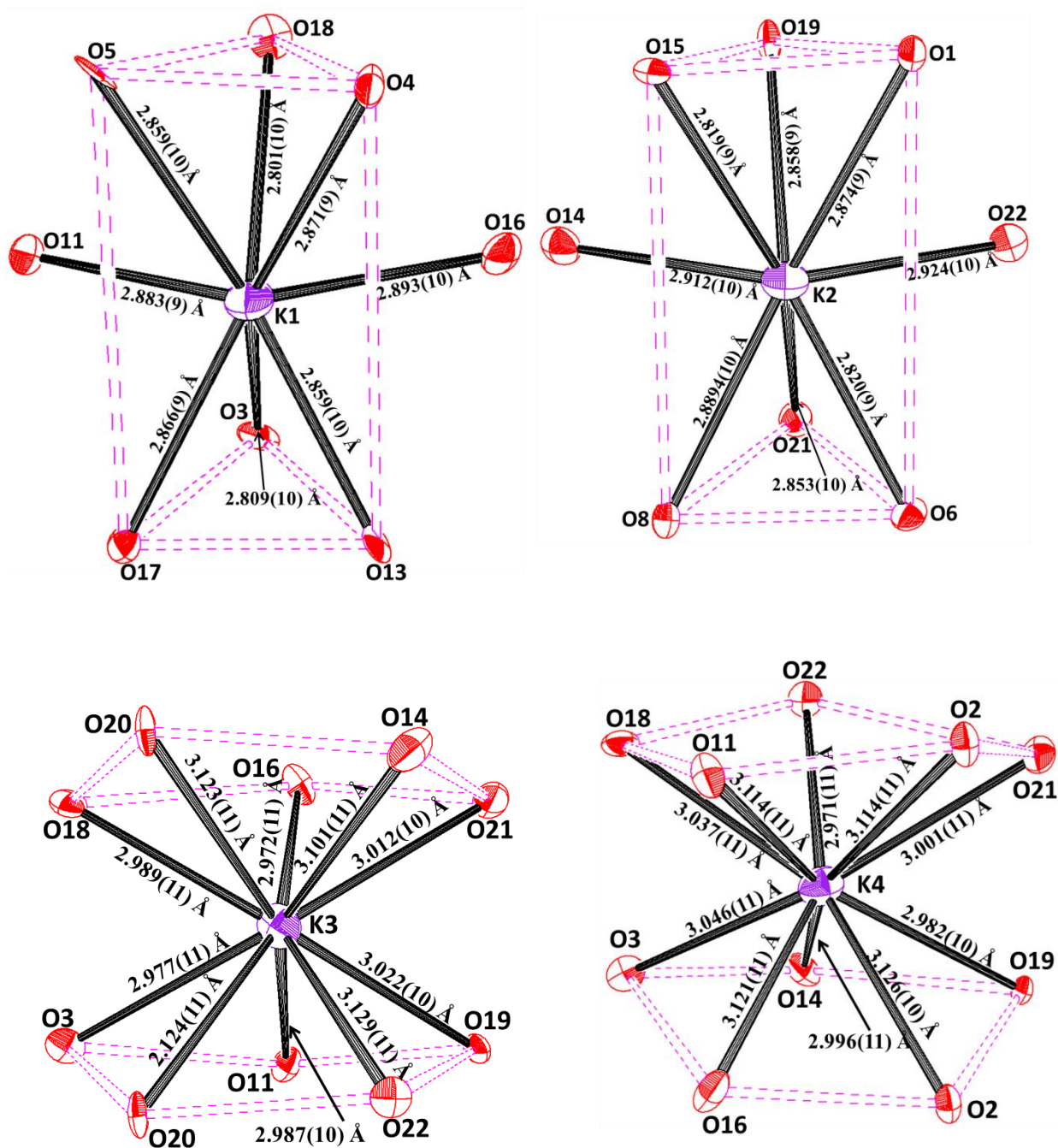


Figure S12. ORTEP diagram showing the atom labelling scheme and coordination of K^+ ions in $K_{1.5}(H_3O)_{0.5}[Mo_2O_5(O_3PPh)_2](5)$.

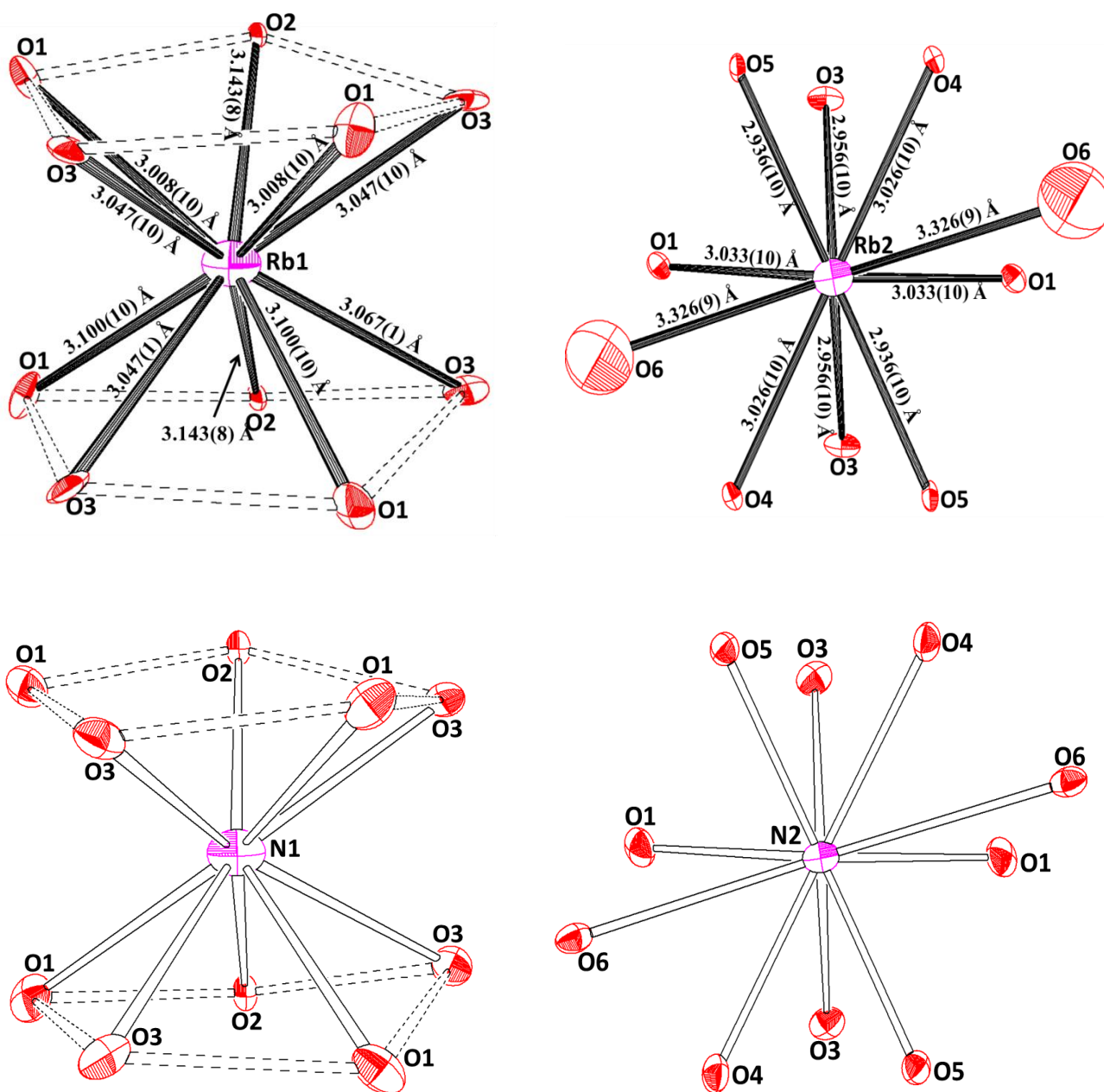


Figure S13. ORTEP diagram showing the atom labelling scheme and coordination of A^+ ions in (top) $Rb_2[Mo_2O_5(O_3PC_6H_4PO_3)]$ (**11**) and (bottom) $(NH_4)_2[Mo_2O_5(O_3PC_6H_4PO_3)]$ (**13**),

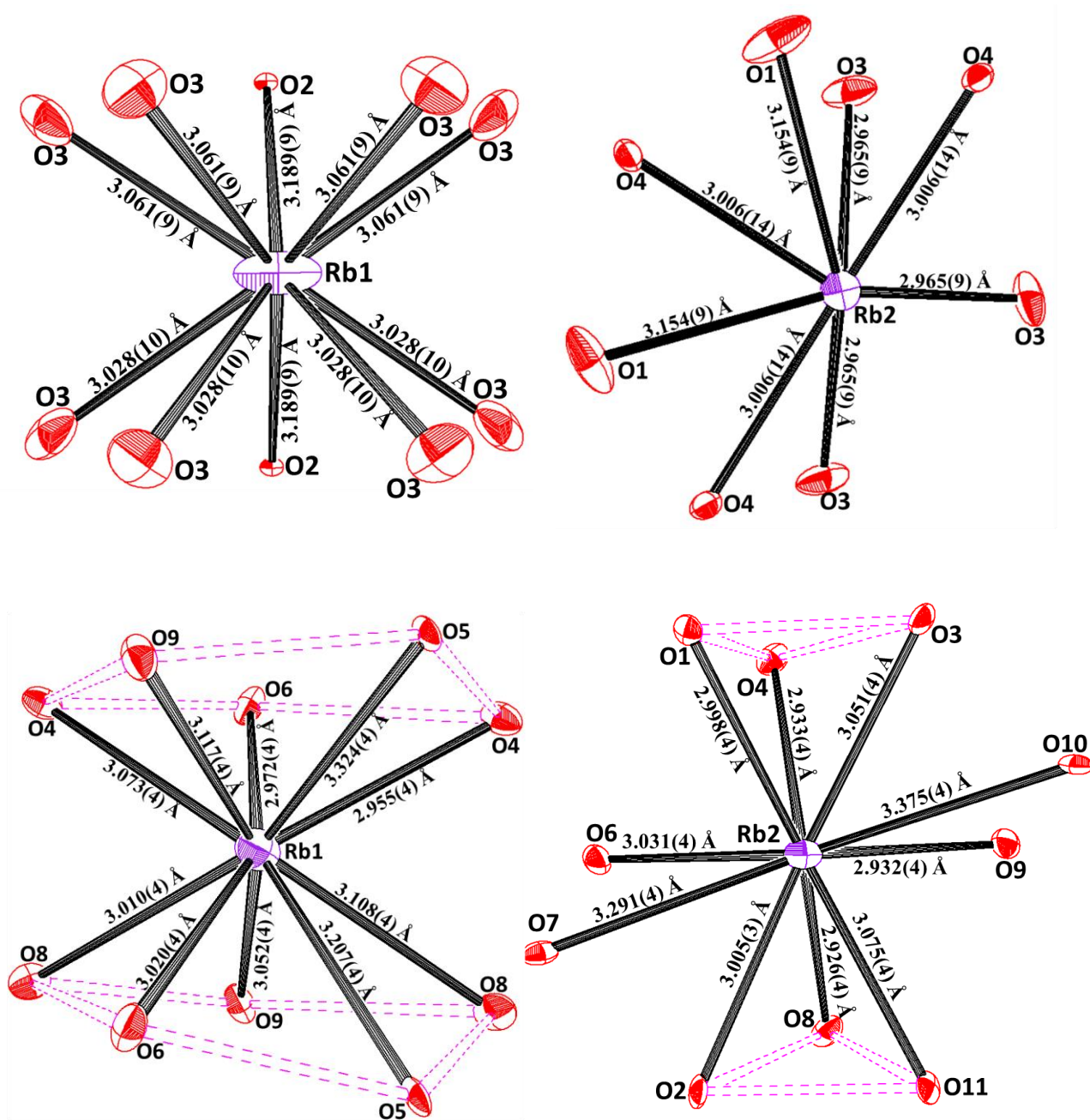


Figure S14 . ORTEP diagram showing the atom labelling scheme and coordination of A⁺ ions in (top) Rb₂[Mo₂O₅(O₃PC₄H₈PO₃)](**16**) and (bottom) Rb₂[Mo₂O₅(O₃PC₃H₆PO₃)](**19**) compounds.

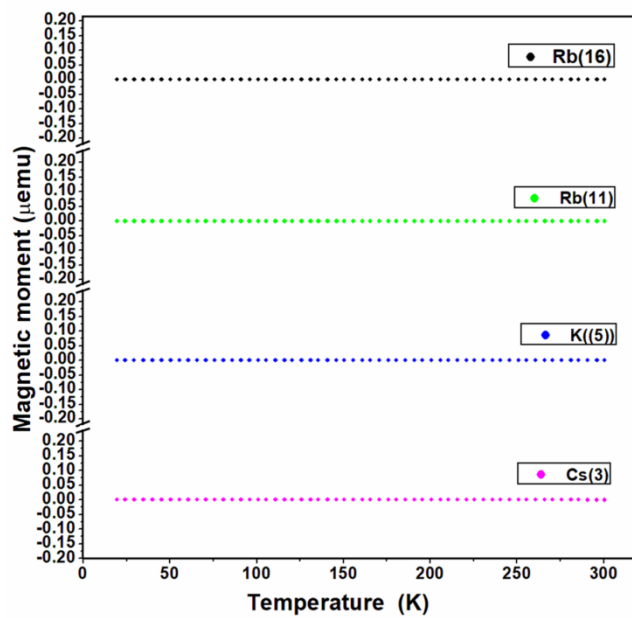


Figure S15. Magnetic moment versus temperature plots for compounds **3**, **5**, **11** and **16**.

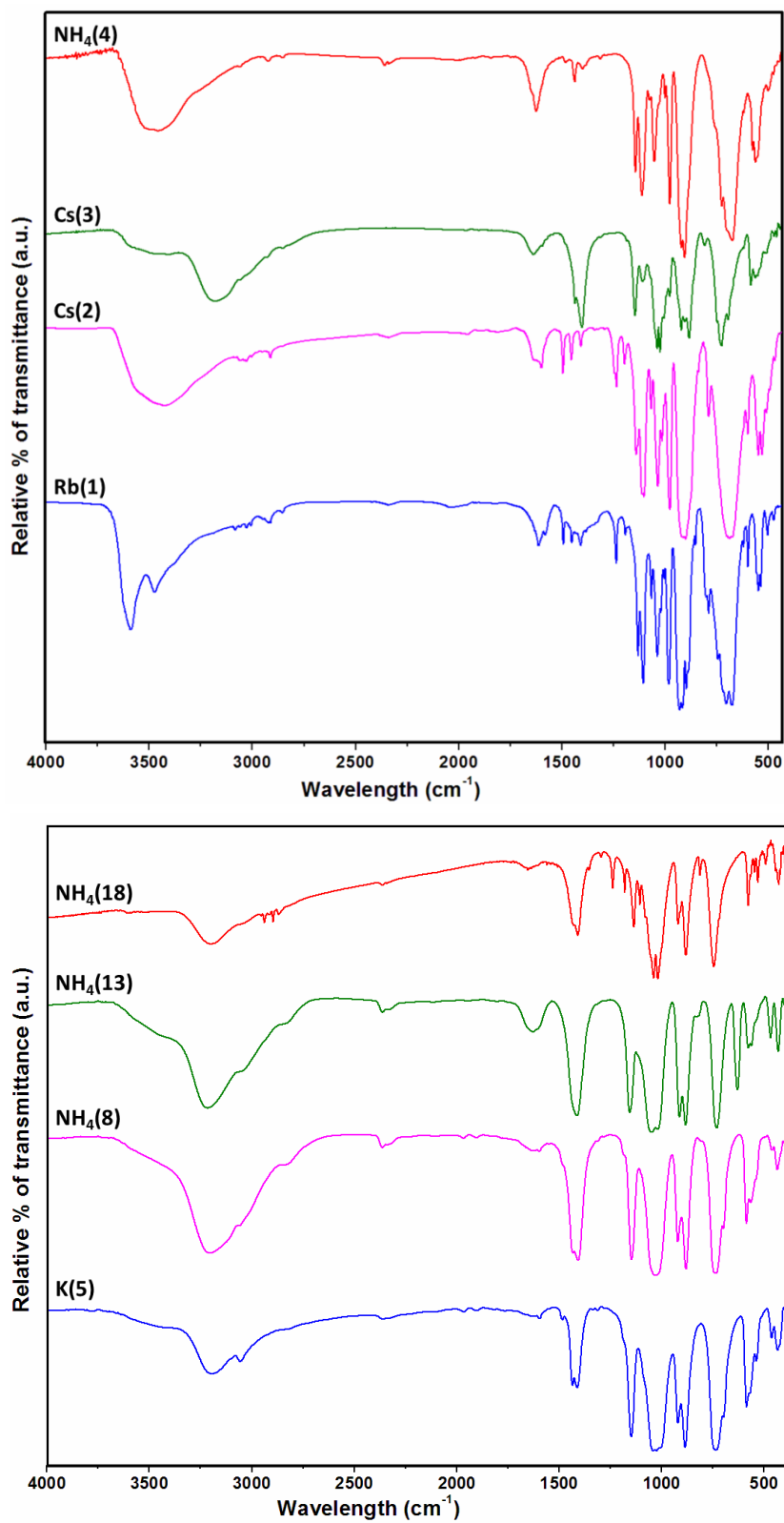


Figure S16. FT-IR spectra of compounds (top) 1 -4 and (bottom) 5, 8, 13, 18.

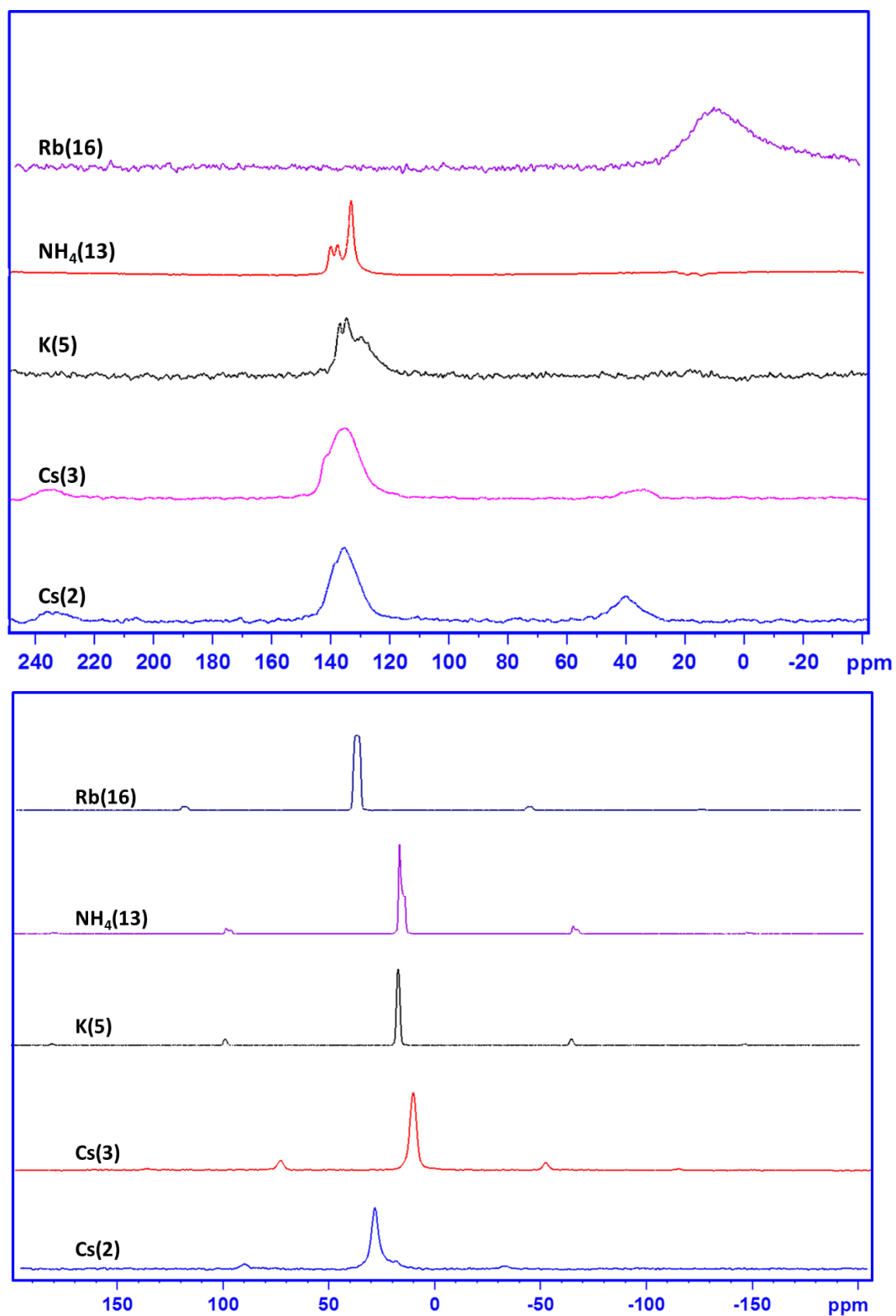


Figure S17. Solid state (top) ¹³C and (bottom) ³¹P nmr spectra of compounds 2, 3, 5, 13 and 16

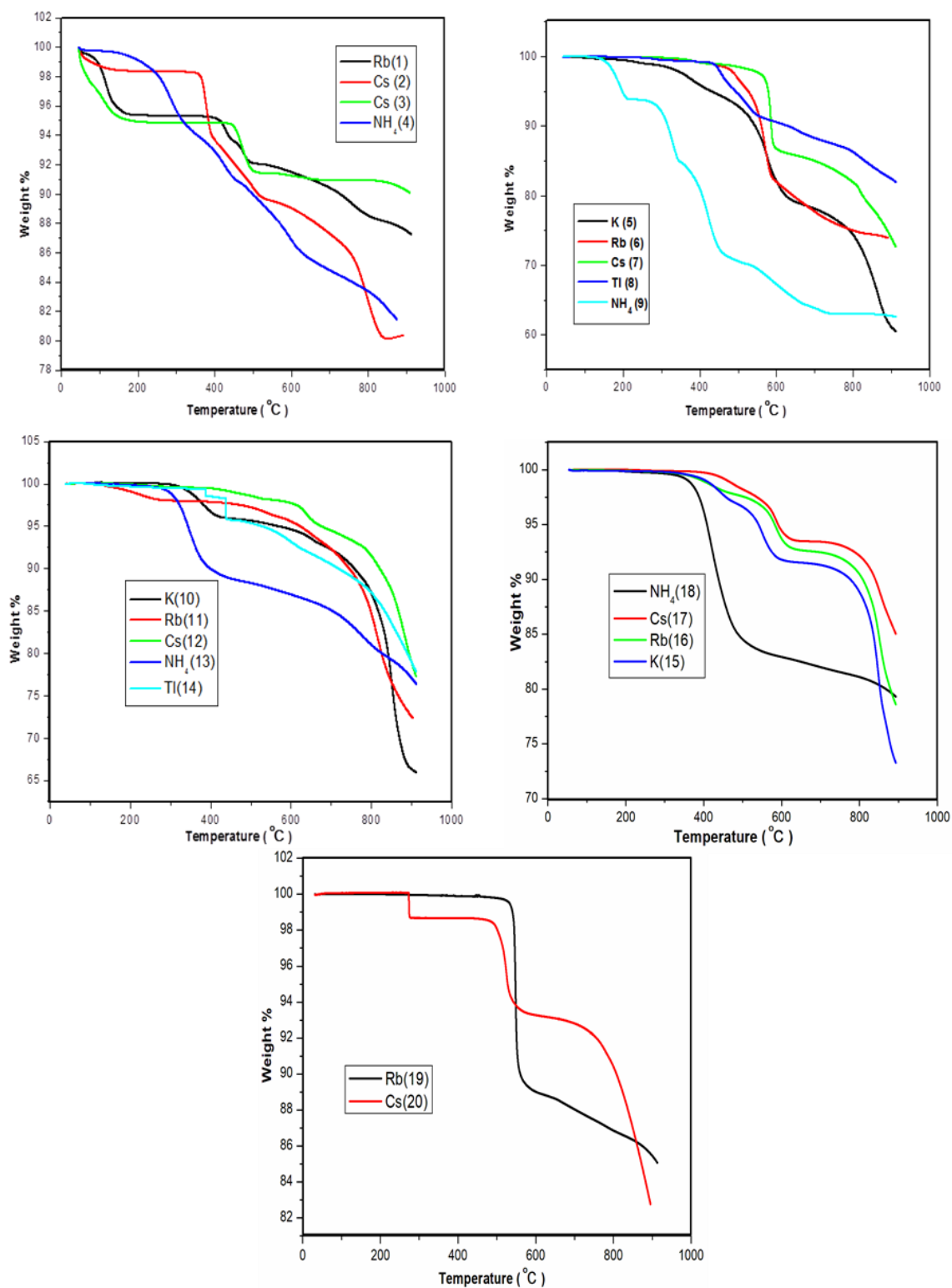


Figure S18. Thermogravimetric plots of compounds 1-20.

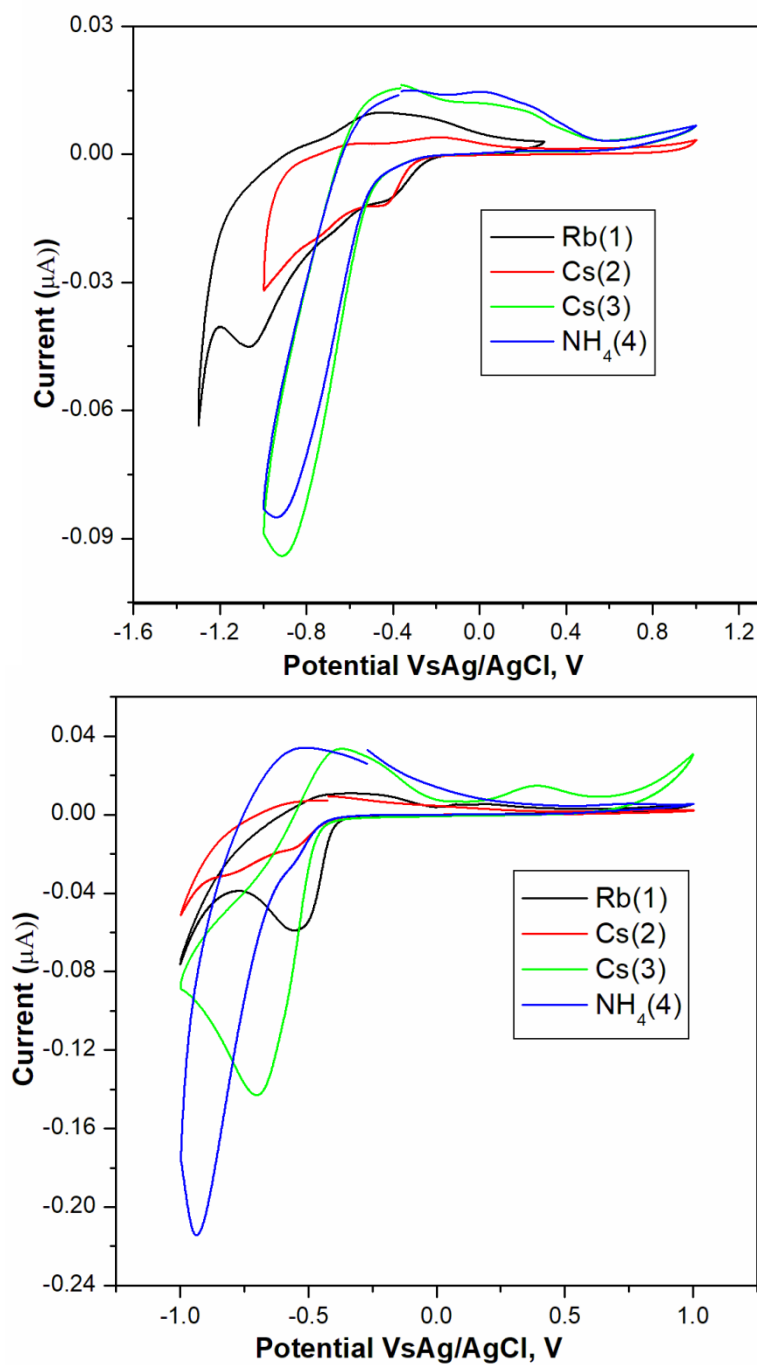


Figure S19. Cyclic voltammograms of molecular molybdoorganophosphonates **1-4**, in (top) 0.1N KNO₃ and (bottom) NaOAc-AcOH.

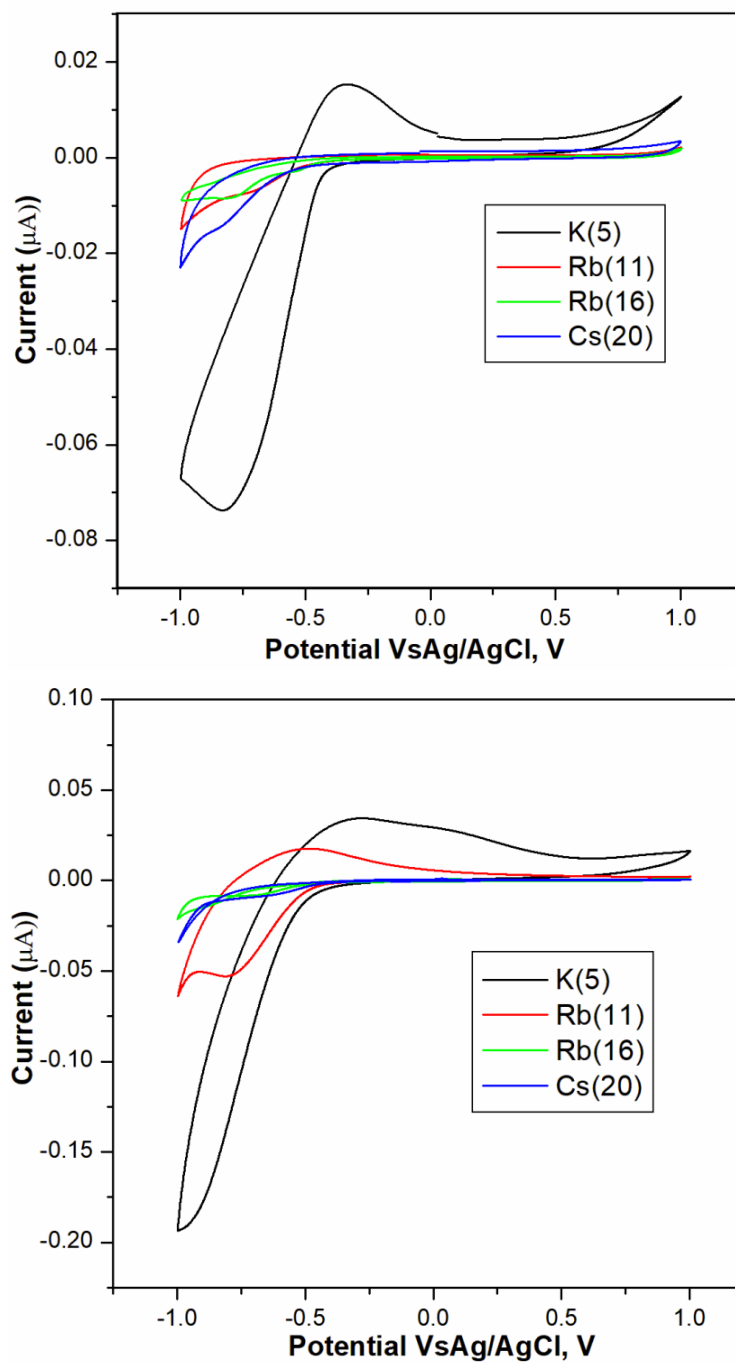


Figure S20. Cyclic voltammograms of non-molecular molybdoorganophosphonates **5**, **11**, **16** and **20**, in (top) 0.1N KNO_3 and (bottom) NaOAc-AcOH.