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Supporting Information

**Organically-templated inorganic-organic hybrid Metal (Zn and Cd) Sulphite-Oxalate
with layered and three-dimensional structures**

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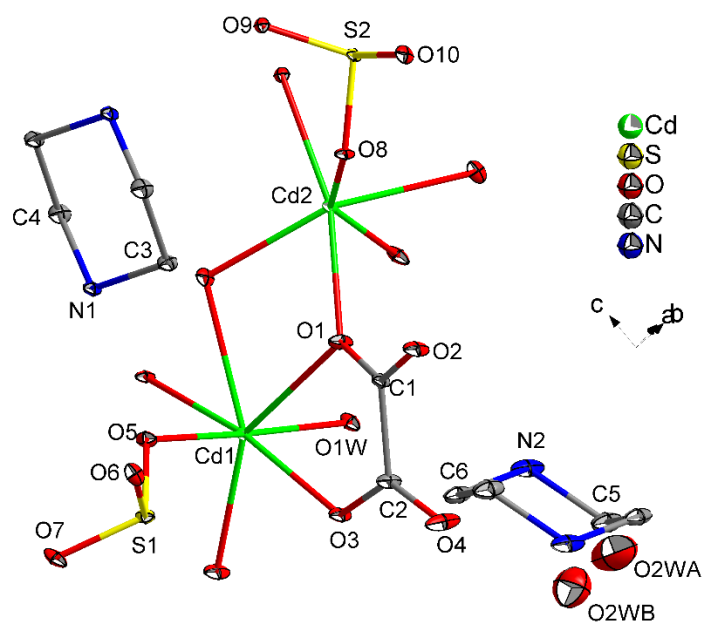


Fig. S1 Asymmetric unit of $[\text{C}_4\text{N}_2\text{H}_{12}][\text{Cd}_2(\text{SO}_3)_2(\text{C}_2\text{O}_4)(\text{H}_2\text{O})].(\text{H}_2\text{O})$ **3** with 15% ellipsoidal probability.

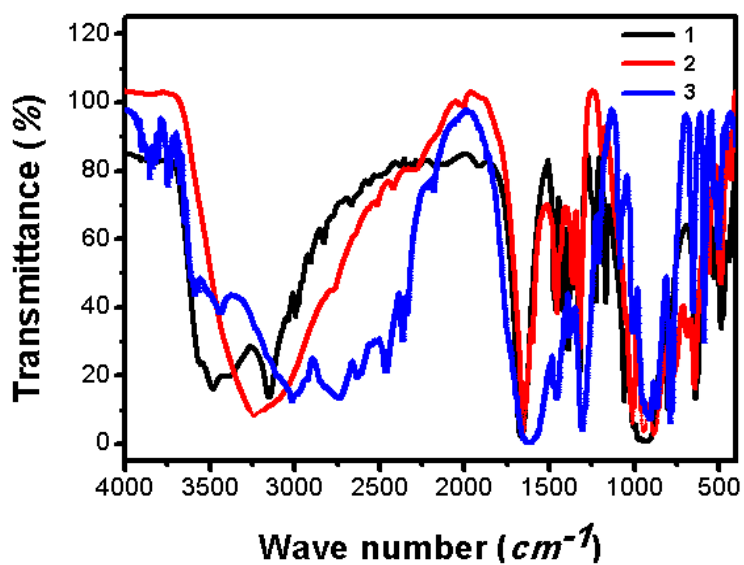


Fig. S2 FTIR spectra of $[\text{Zn}_2(\text{SO}_3)_2(\text{C}_2\text{O}_4)_{0.5}(\text{C}_6\text{N}_2\text{H}_{13})].(\text{H}_2\text{O})_2$, **1**; $[\text{C}_4\text{N}_2\text{H}_{12}]_{0.5}[\text{Zn}_2(\text{SO}_3)_2(\text{C}_2\text{O}_4)_{0.5}(\text{H}_2\text{O})_2]$, **2**; and $[\text{C}_4\text{N}_2\text{H}_{12}][\text{Cd}_2(\text{SO}_3)_2(\text{C}_2\text{O}_4)(\text{H}_2\text{O})].(\text{H}_2\text{O})$ **3**.

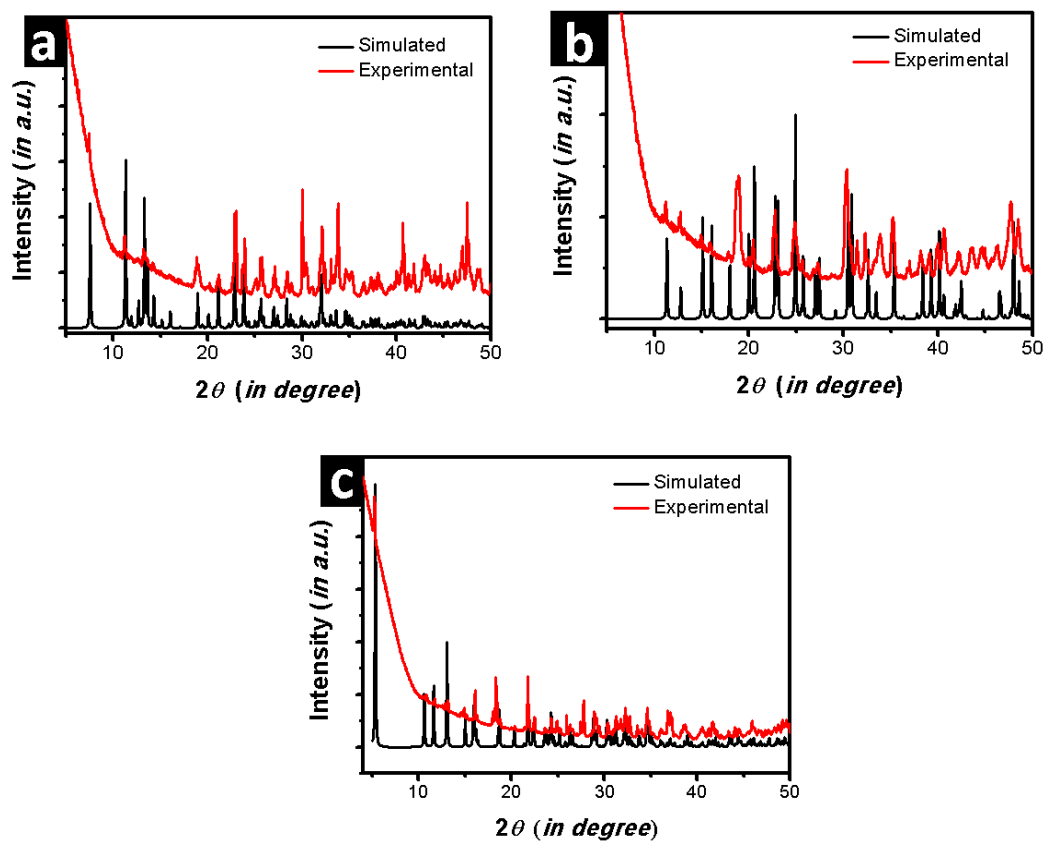


Fig. S3 PXRD pattern for (a) $[\text{Zn}_2(\text{SO}_3)_2(\text{C}_2\text{O}_4)_{0.5}(\text{C}_6\text{N}_2\text{H}_{13})] \cdot (\text{H}_2\text{O})_2$, **1**; and (b) $[\text{C}_4\text{N}_2\text{H}_{12}]_{0.5}[\text{Zn}_2(\text{SO}_3)_2(\text{C}_2\text{O}_4)_{0.5}(\text{H}_2\text{O})_2]$, **2**; and (c) $[\text{C}_4\text{N}_2\text{H}_{12}][\text{Cd}_2(\text{SO}_3)_2(\text{C}_2\text{O}_4)(\text{H}_2\text{O})] \cdot (\text{H}_2\text{O})$, **3**.

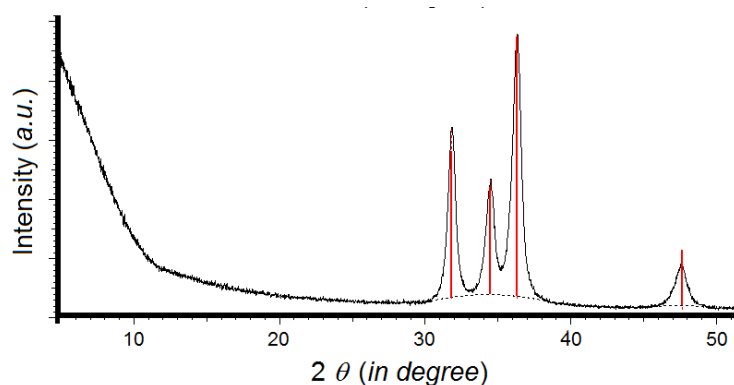


Fig. S4 Post calcination PXRD analysis of $[\text{Zn}_2(\text{SO}_3)_2(\text{C}_2\text{O}_4)_{0.5}(\text{C}_6\text{N}_2\text{H}_{13})] \cdot (\text{H}_2\text{O})_2$, **1** and $[\text{C}_4\text{N}_2\text{H}_{12}]_{0.5}[\text{Zn}_2(\text{SO}_3)_2(\text{C}_2\text{O}_4)_{0.5}(\text{H}_2\text{O})_2]$, **2** showing formation of ZnO (PDF-01-075-0576)

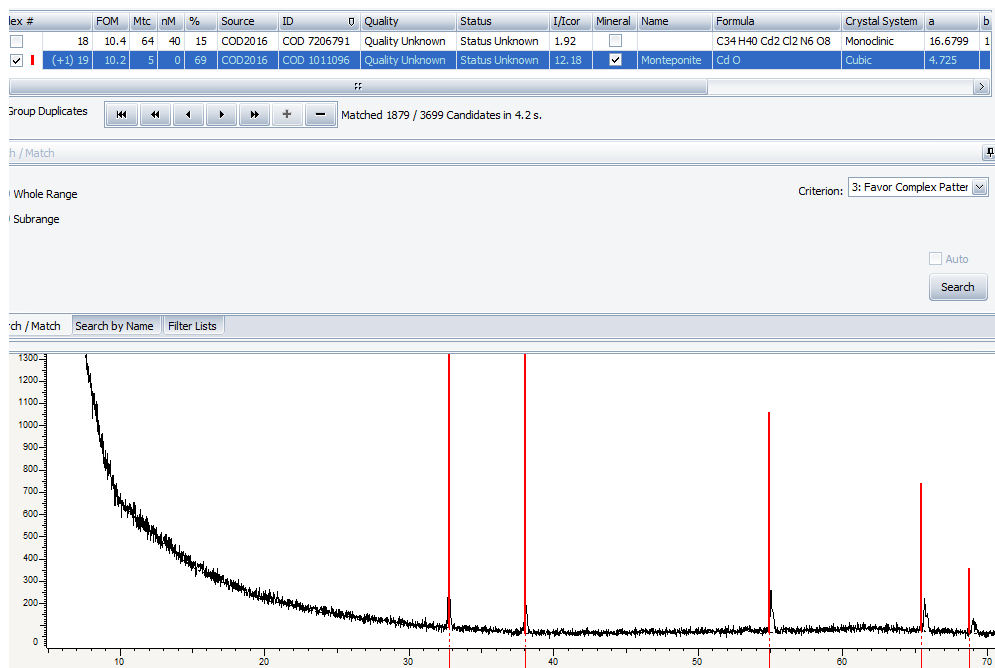


Fig. S5 Post calcination PXR analysis of $[\text{C}_4\text{N}_2\text{H}_{12}][\text{Cd}_2(\text{SO}_3)_2(\text{C}_2\text{O}_4)(\text{H}_2\text{O})] \cdot (\text{H}_2\text{O})$ **3** showing formation of CdO (COD-1011096)

Table S1. Hydrogen bonding table for complexes **1-3**.#

D-H...A	Symmetry of A	D-H	H...A	D-A	∠D-H...A
[Zn₂(SO₃)₂(C₂O₄)_{0.5}(C₆N₂H₁₃)]·(H₂O)₂, 1					
N2-H2...O6	-1+x,y,-1+z	0.91	2.18	2.908(2)	136
N2-H2...O7	-1+x,y,-1+z	0.91	2.32	2.917(2)	123
C2-H2B...O4		0.97	2.54	3.108(3)	117
C6-H6A...O2	x,1/2-y,-1/2+z	0.97	2.40	3.304(3)	155
C6-H6B...O2		0.97	2.57	3.201(3)	123
[C₄N₂H₁₂]_{0.5}[Zn₂(SO₃)₂(C₂O₄)_{0.5}(H₂O)₂], 2					
N1-H1A...O3	y,x,1-z	0.90	2.28	3.146(5)	161
O1W-H1W...O2	1/2+y,1/2-x,1/2-z	0.78(6)	2.00(5)	2.767(5)	172(6)
C2-H2A...O1W	1/2-x,-1/2+y,1/2-z	0.97	2.49	2.879(12)	104
C2-H2B...O3		0.97	2.24	3.023(12)	137
[C₄N₂H₁₂][Cd₂(SO₃)₂(C₂O₄)(H₂O)]·(H₂O), 3					
O2WA-H2WA...O1W	2-x,-y,1-z	0.85	2.15	2.997(13)	171
N1-H1A...O9	2-x,1-y,-z	0.89	2	2.812(4)	150
N1-H1B...O5		0.89	1.91	2.777(4)	164
O2WA-H2WB...O4		0.85	2.06	2.901(12)	168
N2-H2A...O3	-1+x,y,z	0.89	1.88	2.751(5)	166
N2-H2B...O2	-1+x,y,z	0.89	2.52	3.009(5)	116
N2-H2B...O4		0.89	1.86	2.689(5)	154
O1W-H1WA...O6	1+x,-1+y,z	0.83(5)	1.91(6)	2.728(4)	167(6)
O1W-H1WB...O7	x,-1+y,z	0.83(6)	1.88(6)	2.672(4)	161(6)
C3-H3A...O8	-1+x,y,z	0.97	2.52	3.429(5)	156
C3-H3B...O7	-1+x,y,z	0.97	2.56	3.334(5)	137
C4-H4A...O5	2-x,1-y,-z	0.97	2.55	3.372(6)	142
C4-H4B...O10		0.97	2.56	3.295(6)	133
C6-H6A...O4	1-x,1-y,1-z	0.97	2.45	3.370(6)	158

#Where 'D' is donor and 'A' is acceptor, the bond lengths are in (Å) and angles are in (°).

Table S2. Complete list of bond lengths [\AA] and bond angles [$^\circ$] for complex 1-3[#]

[Zn₂(SO₃)₂(C₂O₄)_{0.5}(C₆N₂H₁₃)]·(H₂O)₂, 1							
Zn1-O1	1.9766(14)	Zn1-O8	2.0528(14)	S2-O6	1.5267(14)	N2-C5	1.489(3)
Zn2-O2 ^a	1.9406(14)	Zn1-N1	2.1601(14)	O7-C1	1.241(2)	N2-C7	1.490(3)
Zn2-O3	1.9795(14)	S1-O1	1.5213(14)	O8-C1 ^e	1.258(2)	C1-C1 ^e	1.540(3)
Zn1-O4	1.9791(14)	S1-O2	1.5254(15)	N1-C2	1.472(3)	C3-C2	1.529(3)
Zn2-O5	1.9451(13)	S1-O3 ^b	1.5318(14)	N1-C4	1.480(2)	C5-C4	1.529(3)
Zn2-O6 ^a	1.9772(14)	S2-O4	1.5293(14)	N1-C6	1.476(2)	C6-C7	1.515(3)
Zn1-O7	2.1461(13)	S2-O5	1.5156(14)	N2-C3	1.492(3)		
O1-Zn1-O4	129.22(7)	O2 ^a -Zn2-O6 ^a	111.06(6)	S2-O4-Zn1	123.20(9)	C5-N2-C7	110.14(17)
O1-Zn1-O7	91.66(6)	O5-Zn2-O3	104.22(6)	S2-O5-Zn2	129.43(8)	C7-N2-C3	109.60(16)
O1-Zn1-O8	111.73(6)	O5-Zn2-O6 ^a	96.20(6)	S2-O6-Zn2 ^d	127.51(8)	O7-C1-O8 ^e	126.28(16)
O1-Zn1-N1	98.42(6)	O6 ^a -Zn2-O3	112.20(6)	S1-O2-Zn2 ^d	123.94(8)	O7-C1-C1 ^e	116.97(19)
O4-Zn1-O7	91.36(6)	O1-S1-O2	104.83(8)	C1-O7-Zn1	112.39(11)	O8 ^e -C1-C1 ^e	116.8(2)
O4-Zn1-O8	118.58(7)	O1-S1-O3 ^b	105.46(8)	C1 ^e -O8-Zn1	114.96(12)	N1-C2-C3	112.00(16)
O4-Zn1-N1	90.30(6)	O2-S1-O3 ^b	104.21(8)	C2-N1-C6	108.09(18)	N2-C3-C2	107.80(16)
O7-Zn1-N1	165.50(6)	O5-S2-O6	106.38(8)	C2-N1-C4	108.19(17)	N1-C4-C5	111.44(16)
O8-Zn1-O7	78.93(5)	O5-S2-O4	103.10(8)	C2-N1-Zn1	108.93(11)	N2-C5-C4	108.30(15)
O8-Zn1-N1	87.64(5)	O6-S2-O4	104.89(8)	C6-N1-Zn1	110.51(11)	N1-C6-C7	111.96(17)
O2 ^a -Zn2-O3	107.36(6)	S1-O1-Zn1	129.96(9)	C6-N1-C4	108.09(17)	N2-C7-C6	108.34(16)
O2 ^a -Zn2-O5	125.35(6)	S1 ^c -O3-Zn2	122.34(8)	C5-N2-C3	109.49(17)		
[C₄N₂H₁₂]_{0.5}[Zn₂(SO₃)₂(C₂O₄)_{0.5}(H₂O)₂], 2							
Zn1-O1	2.056(4)	Zn1-O1W	2.081(4)	C1-O3	1.252(5)	C2-C2 ⁿ	1.474(19)
Zn2-O2	1.978(3)	S1-O1	1.506(4)	C1-C1 ^l	1.518(15)	N1-C2	1.457(11)
Zn1-O3	2.128(4)	S1-O2	1.543(3)				
O1 ^l -Zn1-O1	163.3(2)	O1W-Zn1-O3 ^j	168.50(18)	O2 ^l -S1-O2	104.9(2)	C2 ^m -N1-C2	107.5(12)
O1-Zn1-O3	96.52(9)	O1W ^l -Zn1-O1W	99.9(3)	S1-O1-Zn1	122.6(3)	O3 ^k -C1-O3	126.5(7)
O1-Zn1-O1W	84.65(8)	O2 ^l -Zn2-O2	118.93(15)	S1-O2-Zn2	123.69(17)	O3-C1-C1 ^l	116.8(4)
O3-Zn1-O3 ^j	76.9(2)	O2 ^a -Zn2-O2	104.96(7)	C1-O3-Zn1	114.8(4)	N1-C2-C2 ⁿ	107.4(9)
O1W-Zn1-O3	91.63(17)	O1-S1-O2	104.41(16)				
[C₄N₂H₁₂][Cd₂(SO₃)₂(C₂O₄)(H₂O)]·(H₂O), 3							
Cd1-O1	2.481(3)	Cd2-O6 ^q	2.335(3)	S1-O7	1.534(3)	C2-C1	1.566(5)
Cd1-O2 ^o	2.336(3)	Cd2-O7 ^q	2.317(3)	S2-O8	1.545(3)	N1-C3	1.490(5)
Cd1-O3	2.292(3)	Cd2-O8	2.261(3)	S2-O9	1.525(3)	N1-C4	1.490(5)
Cd1-O5	2.255(3)	Cd2-O9 ^p	2.268(3)	S2-O10	1.533(3)	N2-C6	1.480(7)
Cd1-O8 ^o	2.383(3)	Cd2-O10 ^o	2.279(3)	O3-C2	1.263(4)	N2-C5	1.487(7)
Cd1-O10 ^o	2.481(3)	Cd2-S1 ^q	2.9742(10)	O1-C1	1.258(5)	C3-C4 ^u	1.505(6)
Cd1-O1W	2.313(3)	S1-O6	1.528(3)	O2-C1	1.239(5)	C5-C6 ^t	1.512(7)
Cd2-O1	2.284(3)	S1-O5	1.529(3)	O4-C2	1.240(5)		
O5-Cd1-O3	90.32(11)	O10 ^o -Cd2-O7 ^q	91.32(10)	S2-O10-Cd2 ^s	129.50(16)	N2-C6-C5 ^t	110.9(4)
O5-Cd1-O1W	175.84(10)	O1-Cd2-O7 ^q	90.51(11)	S2-O10-Cd1 ^s	97.98(13)	N2-C6-H6A	109.5
O3-Cd1-O1W	87.91(11)	O8-Cd2-O6 ^q	88.24(10)	Cd2 ^s -O10-Cd1 ^s	104.74(11)	C5 ^t -C6-H6A	109.5
O5-Cd1-O2 ^o	97.22(11)	O9 ^p -Cd2-O6 ^q	93.69(10)	S2-O8-Cd2	120.04(14)	N2-C6-H6B	109.5
O3-Cd1-O2 ^o	79.68(10)	O10 ^o -Cd2-O6 ^q	152.27(10)	S2-O8-Cd1 ^s	101.72(13)	C5 ^t -C6-H6B	109.5
O1W-Cd1-O2 ^o	86.16(11)	O1-Cd2-O6 ^q	101.86(11)	Cd2-O8-Cd1 ^s	123.96(11)	H6A-C6-H6B	108.1
O5-Cd1-O8 ^o	95.67(10)	O7 ^q -Cd2-O6 ^q	61.11(10)	S1-O7-Cd2 ^r	99.10(14)	H2WA-O2WA-H2WB	109.5
O3-Cd1-O8 ^o	167.05(9)	O8-Cd2-S1 ^q	118.53(7)	C1-O2-Cd1 ^s	137.7(3)	H2WC-O2WB-H2WD	111.2
O1W-Cd1-O8 ^o	86.82(10)	O9 ^p -Cd2-S1 ^q	96.94(7)	Cd1-O1W-H1WA	114(5)	C3-N1-C4	111.5(3)
O2 ^o -Cd1-O8 ^o	88.16(9)	O10 ^o -Cd2-S1 ^q	121.93(7)	Cd1-O1W-H1WB	110(5)	C3-N1-H1A	109.3
O5-Cd1-O10 ^o	84.11(10)	O1-Cd2-S1 ^q	98.39(8)	H1WA-O1W-H1WB	104(4)	C4-N1-H1A	109.3
O3-Cd1-O10 ^o	134.18(9)	O7 ^q -Cd2-S1 ^q	30.61(7)	O4-C2-O3	124.5(4)	C3-N1-H1B	109.3
O1W-Cd1-O10 ^o	94.43(10)	O6 ^q -Cd2-S1 ^q	30.55(7)	O4-C2-C1	118.7(3)	C4-N1-H1B	109.3
O2 ^o -Cd1-O10 ^o	146.13(9)	O9-S2-O10	105.51(16)	O3-C2-C1	116.8(3)	H1A-N1-H1B	108
O8 ^o -Cd1-O10 ^o	58.13(8)	O9-S2-O8	104.30(15)	C6-N2-C5	111.6(4)	N1-C3-C4 ^u	110.0(3)
O5-Cd1-O1	89.91(10)	O10-S2-O8	100.41(15)	C6-N2-H2A	109.3	N1-C3-H3A	109.7
O3-Cd1-O1	67.74(9)	O6-S1-O5	104.44(16)	C5-N2-H2A	109.3	C4 ^u -C3-H3A	109.7
O1W-Cd1-O1	85.94(10)	O6-S1-O7	101.14(16)	C6-N2-H2B	109.3	N1-C3-H3B	109.7
O2 ^o -Cd1-O1	146.70(10)	O5-S1-O7	104.67(17)	C5-N2-H2B	109.3	C4 ^u -C3-H3B	109.7
O8 ^o -Cd1-O1	123.58(9)	O6-S1-Cd2 ^r	50.95(11)	H2A-N2-H2B	108	H3A-C3-H3B	108.2
O10 ^o -Cd1-O1	66.81(9)	O5-S1-Cd2 ^r	110.72(11)	O2-C1-O1	128.3(3)	N1-C4-C3 ^u	110.7(3)
O8-Cd2-O9 ^p	89.78(9)	O7-S1-Cd2 ^r	50.29(11)	O2-C1-C2	117.6(3)	N1-C4-H4A	109.5
O8-Cd2-O10 ^o	118.27(10)	S1-O5-Cd1	120.98(16)	O1-C1-C2	114.0(3)	C3 ^u -C4-H4A	109.5

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O9 ^p -Cd2-O10 ^o	94.38(10)	S1-O6-Cd2 ^r	98.50(14)	N2-C5-C6 ^t	109.3(4)	N1-C4-H4B	109.5
O8-Cd2-O1	86.95(10)	C2-O3-Cd1	122.1(2)	N2-C5-H5A	109.8	C3 ^u -C4-H4B	109.5
O9 ^p -Cd2-O1	163.99(10)	S2-O9-Cd2 ^p	120.17(14)	C6 ^t -C5-H5A	109.8	H4A-C4-H4B	108.1
O10 ^o -Cd2-O1	73.54(10)	C1-O1-Cd2	139.5(2)	N2-C5-H5B	109.8		
O8-Cd2-O7 ^q	148.01(10)	C1-O1-Cd1	115.9(2)	C6 ^t -C5-H5B	109.8		
O9 ^p -Cd2-O7 ^q	100.43(10)	Cd2-O1-Cd1	104.59(11)	H5A-C5-H5B	108.3		

#Symmetry transformations used to generate equivalent atoms:

(a) $x, -y+1/2, z-1/2$; (b) $x, y, z+1$; (c) $x, y, z-1$; (d) $x, -y+1/2, z+1/2$; (e) $-x+1, -y, -z+2$; (f) $-x+1, -y, z$; (g) $-y+1/2, x-1/2, -z+1/2$; (h) $y+1/2, -x+1/2, -z+1/2$; (i) $y+1, x-1, z$; (j) $x, y, -z$; (k) $-x+2, -y, z$; (l) $-x+2, -y, -z$; (m) y, x, z ; (n) $-y+2, -x+2, -z+1$; (o) $x-1, y, z$; (p) $-x+1, -y+2, -z+2$; (q) $x, y+1, z$; (r) $x, y-1, z$; (s) $x+1, y, z$; (t) $-x+2, -y+1, -z+1$; (u) $-x+1, -y+1, -z+2$

Table S3. Atomic coordinates ($\times 10^4$) and equivalent isotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for **1-3** U(eq) is defined as one third of the trace of the orthogonalized U_{ij} tensor.

	x	y	z	U(eq)		x	y	z	U(eq)
[Zn₂(SO₃)₂(C₂O₄)_{0.5}(C₆N₂H₁₃)]·(H₂O)₂, 1									
Zn1	3448(1)	1022(1)	8731(1)	19(1)	O1W	8490(7)	-230(1)	8584(6)	142(2)
Zn2	5156(1)	2421(1)	5152(1)	20(1)	O2W	5724(7)	266(2)	6399(8)	165(2)
S1	2865(1)	1883(1)	11465(1)	20(1)	N1	1093(2)	1164(1)	6383(2)	18(1)
S2	6585(1)	1598(1)	8507(1)	17(1)	N2	-1603(2)	1342(1)	3549(2)	25(1)
O1	2727(2)	1294(1)	10632(2)	27(1)	C1	5786(2)	150(1)	10711(2)	19(1)
O2	3014(2)	2288(1)	10056(2)	26(1)	C2	1269(3)	948(1)	4750(3)	34(1)
O3	4623(2)	1902(1)	3052(2)	28(1)	C3	-352(2)	1037(1)	3024(3)	29(1)
O4	4713(2)	1431(1)	7525(2)	30(1)	C4	-373(2)	874(1)	6522(3)	31(1)
O5	6839(2)	2009(1)	7189(2)	23(1)	C5	-2017(2)	984(1)	4841(3)	28(1)
O6	6677(2)	1951(1)	10137(2)	25(1)	C6	707(3)	1778(1)	6126(3)	36(1)
O7	5762(2)	678(1)	10739(2)	25(1)	C7	-872(3)	1896(1)	4413(3)	33(1)
O8	3028(2)	165(1)	8265(2)	27(1)					
[C₄N₂H₁₂]_{0.5} [Zn₂(SO₃)₂(C₂O₄)_{0.5}(H₂O)₂], 2									
Zn1	8225(1)	-1776(1)	0	11(1)	O3	9287(3)	-713(3)	1484(5)	27(1)
Zn2	5000	0	2500	12(1)	O1W	7370(3)	-2630(3)	1786(5)	18(1)
S1	6834(1)	686(1)	0	12(1)	N1	9115(8)	9115(8)	5000	145(12)
O1	6737(4)	-668(4)	0	15(1)	C1	10000	0	851(8)	15(2)
O2	6086(3)	1087(3)	1373(3)	14(1)	C2	8854(10)	10352(10)	4558(13)	64(5)
[C₄N₂H₁₂][Cd₂(SO₃)₂(C₂O₄)(H₂O)]·(H₂O), 3									
Cd(1)	-1086(1)	8084(1)	7714(1)	10(1)	O(1W)	-2477(5)	10939(4)	7189(2)	17(1)
Cd(2)	2526(1)	10389(1)	8640(1)	11(1)	O(4)	3846(6)	6991(6)	5728(2)	37(1)
S(2)	7612(2)	9613(1)	9380(1)	10(1)	C(2)	2785(6)	7349(5)	6374(2)	16(1)
S(1)	1216(2)	3777(1)	7673(1)	14(1)	N(2)	8322(6)	6278(6)	5438(2)	30(1)
O(5)	525(5)	5322(3)	8217(2)	16(1)	C(1)	3994(6)	7839(5)	7079(2)	13(1)
O(6)	3555(5)	2666(4)	7954(2)	18(1)	C(5)	8718(8)	6596(8)	4565(3)	32(1)
O(3)	762(5)	7366(4)	6505(2)	18(1)	C(6)	8854(8)	4383(7)	5663(3)	29(1)
O(9)	7648(5)	8466(3)	10147(2)	13(1)	O(2WA)	3360(30)	9761(15)	4481(7)	91(6)
O(1)	2670(5)	8564(4)	7636(2)	19(1)	O(2WB)	1520(30)	9760(17)	4562(10)	82(7)
O(10)	9992(4)	8928(4)	8998(2)	15(1)	N(1)	3392(6)	4671(4)	9500(2)	15(1)

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O(8)	6247(4)	8955(3)	8797(2)	12(1)	C(3)	5532(7)	4907(6)	9149(2)	17(1)
O(7)	-228(5)	2618(4)	7980(2)	20(1)	C(4)	2648(7)	5656(6)	10240(3)	20(1)
O(2)	6072(5)	7531(4)	7016(2)	21(1)					

Table S4. Anisotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for complex **1-3**. The anisotropic displacement factor exponent takes the form: $-\pi^2 [h^2 a^{*2} U_{11} + \dots + 2 h k a^* b^* U_{12}]$

	U11	U22	U33	U23	U13	U12		U11	U22	U33	U23	U13	U12
[Zn₂(SO₃)₂(C₂O₄)_{0.5}(C₆N₂H₁₂)]·(H₂O)₂, 1													
Zn1	18(1)	16(1)	17(1)	1(1)	2(1)	0(1)	O1W	287(6)	52(2)	192(4)	-6(2)	204(5)	-18(2)
Zn2	21(1)	20(1)	15(1)	-1(1)	5(1)	0(1)	O2W	209(5)	101(3)	256(6)	-17(3)	167(5)	-7(3)
S1	21(1)	20(1)	18(1)	-3(1)	8(1)	-1(1)	N1	17(1)	17(1)	15(1)	0(1)	2(1)	0(1)
S2	17(1)	17(1)	14(1)	1(1)	4(1)	2(1)	N2	19(1)	31(1)	17(1)	1(1)	-1(1)	3(1)
O1	37(1)	21(1)	24(1)	-7(1)	13(1)	-5(1)	C1	17(1)	19(1)	17(1)	1(1)	2(1)	0(1)
O2	25(1)	28(1)	23(1)	4(1)	6(1)	-3(1)	C2	24(1)	53(2)	17(1)	-4(1)	3(1)	12(1)
O3	30(1)	29(1)	17(1)	-7(1)	2(1)	3(1)	C3	22(1)	42(1)	18(1)	-4(1)	5(1)	2(1)
O4	22(1)	42(1)	20(1)	5(1)	2(1)	-10(1)	C4	20(1)	40(1)	28(1)	13(1)	5(1)	-2(1)
O5	22(1)	28(1)	18(1)	6(1)	7(1)	0(1)	C5	19(1)	36(1)	24(1)	-1(1)	5(1)	-2(1)
O6	33(1)	25(1)	15(1)	-1(1)	7(1)	6(1)	C6	37(1)	17(1)	31(1)	0(1)	-8(1)	2(1)
O7	23(1)	16(1)	25(1)	1(1)	-2(1)	0(1)	C7	33(1)	22(1)	31(1)	5(1)	-1(1)	6(1)
O8	22(1)	18(1)	27(1)	1(1)	-5(1)	1(1)							
[C₄N₂H₁₂]_{0.5} [Zn₂(SO₃)₂(C₂O₄)_{0.5}(H₂O)₂], 2													
Zn1	11(1)	11(1)	10(1)	0	0	0(1)	O3	35(2)	35(2)	11(2)	1(1)	1(1)	-20(2)
Zn2	10(1)	10(1)	14(1)	0	0	0	O1W	17(1)	17(1)	20(2)	7(1)	7(1)	3(2)
S1	11(1)	11(1)	13(1)	0	0	-2(1)	N1	198(18)	198(18)	38(7)	0	0	-170(20)
O1	14(2)	12(2)	20(2)	0	0	1(2)	C1	17(2)	17(2)	9(3)	0	0	0(3)
O2	16(1)	13(1)	14(1)	-2(1)	2(1)	-3(1)	C2	37(6)	83(9)	73(10)	-70(7)	-37(6)	39(6)
[[C₄N₂H₁₂][Cd₂(SO₃)₂(C₂O₄)(H₂O)]·(H₂O)], 3													
Cd(1)	7(1)	13(1)	12(1)	-4(1)	1(1)	-3(1)	O(1W)	13(1)	15(1)	24(2)	-1(1)	-3(1)	-5(1)
Cd(2)	7(1)	12(1)	13(1)	-3(1)	0(1)	-4(1)	O(4)	13(2)	80(3)	24(2)	-28(2)	7(1)	-19(2)
S(2)	7(1)	12(1)	11(1)	-3(1)	1(1)	-4(1)	C(2)	6(2)	26(2)	18(2)	-12(2)	1(1)	-3(1)
S(1)	16(1)	11(1)	14(1)	-2(1)	-3(1)	-3(1)	N(2)	7(2)	58(3)	28(2)	-23(2)	3(1)	-10(2)
O(5)	18(1)	10(1)	17(1)	-3(1)	-3(1)	-2(1)	C(1)	10(2)	17(2)	15(2)	-4(1)	0(1)	-7(1)
O(6)	12(1)	14(1)	27(2)	-1(1)	2(1)	-3(1)	C(5)	20(2)	51(3)	29(2)	-13(2)	-2(2)	-11(2)
O(3)	9(1)	32(2)	17(1)	-9(1)	2(1)	-11(1)	C(6)	19(2)	52(3)	21(2)	-13(2)	0(2)	-16(2)
O(9)	13(1)	14(1)	11(1)	-2(1)	0(1)	-6(1)	O(2WA)	126(15)	73(7)	69(7)	-8(5)	-28(8)	-20(7)
O(1)	11(1)	28(2)	19(1)	-14(1)	5(1)	-7(1)	O(2WB)	99(17)	76(10)	72(10)	11(8)	-18(10)	-26(10)

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O(10)	8(1)	20(1)	18(1)	-5(1)	5(1)	-6(1)	N(1)	13(2)	18(2)	16(2)	1(1)	-5(1)	-8(1)
O(8)	6(1)	16(1)	13(1)	-5(1)	-2(1)	-3(1)	C(3)	11(2)	26(2)	16(2)	-4(2)	0(1)	-8(2)
O(7)	12(1)	17(1)	33(2)	-2(1)	-7(1)	-7(1)	C(4)	11(2)	26(2)	21(2)	-5(2)	1(2)	-5(2)
O(2)	8(1)	35(2)	22(2)	-12(1)	1(1)	-9(1)							