

Electronic Supplementary Information
Abnormal site occupancy and high performance in warm WLEDs
of a novel red phosphor NaHF₂:Mn⁴⁺ synthesized at room
temperature

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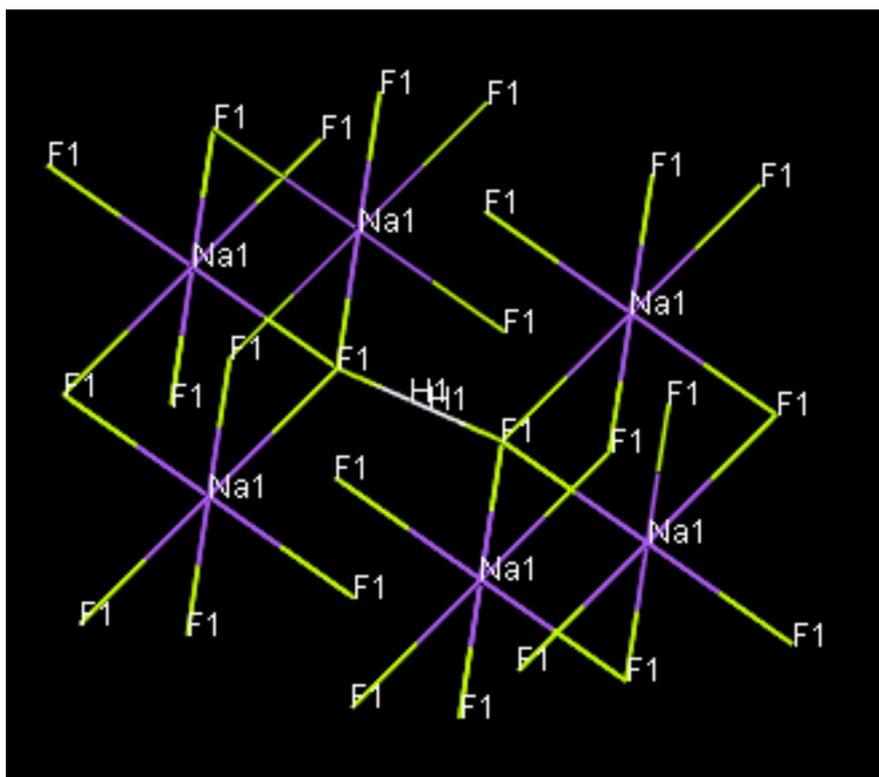


Figure S1. Structure diagram of solid NaHF₂ drawn by software Mercury 1.4.1

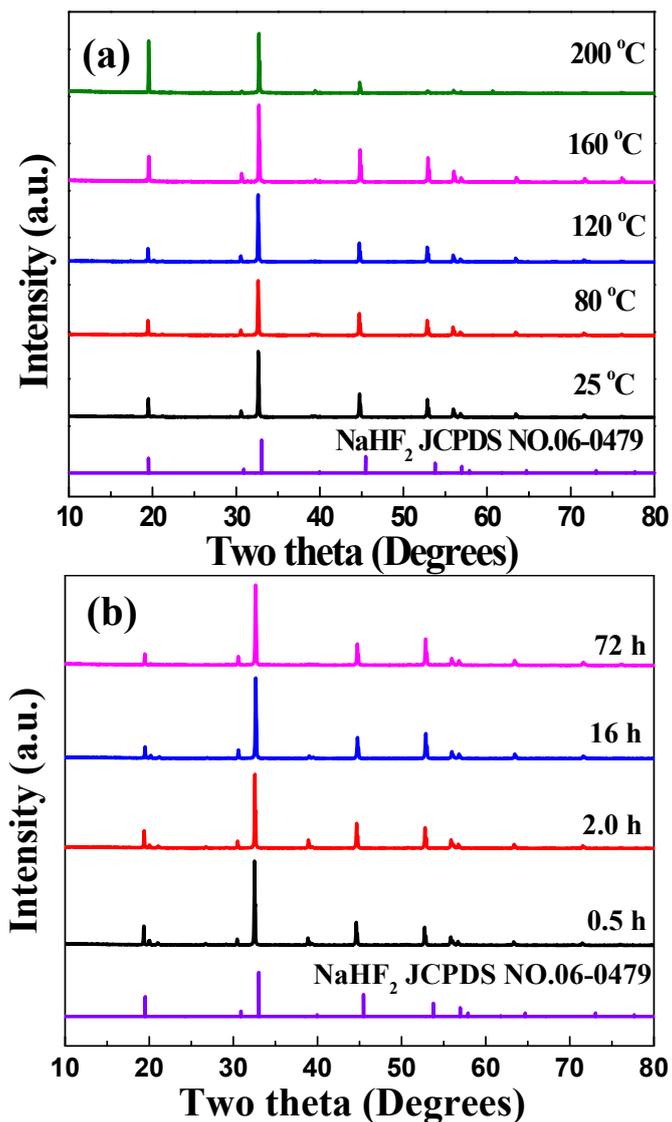


Figure S2. X-ray diffraction patterns of NaHF₂:Mn⁴⁺ samples obtained under different synthesis conditions compared with the standard JCDPS NO. 06-0479: (a) reaction temperature, (b) reaction time. (All synthetic parameters were kept constant except for the examined variables.)

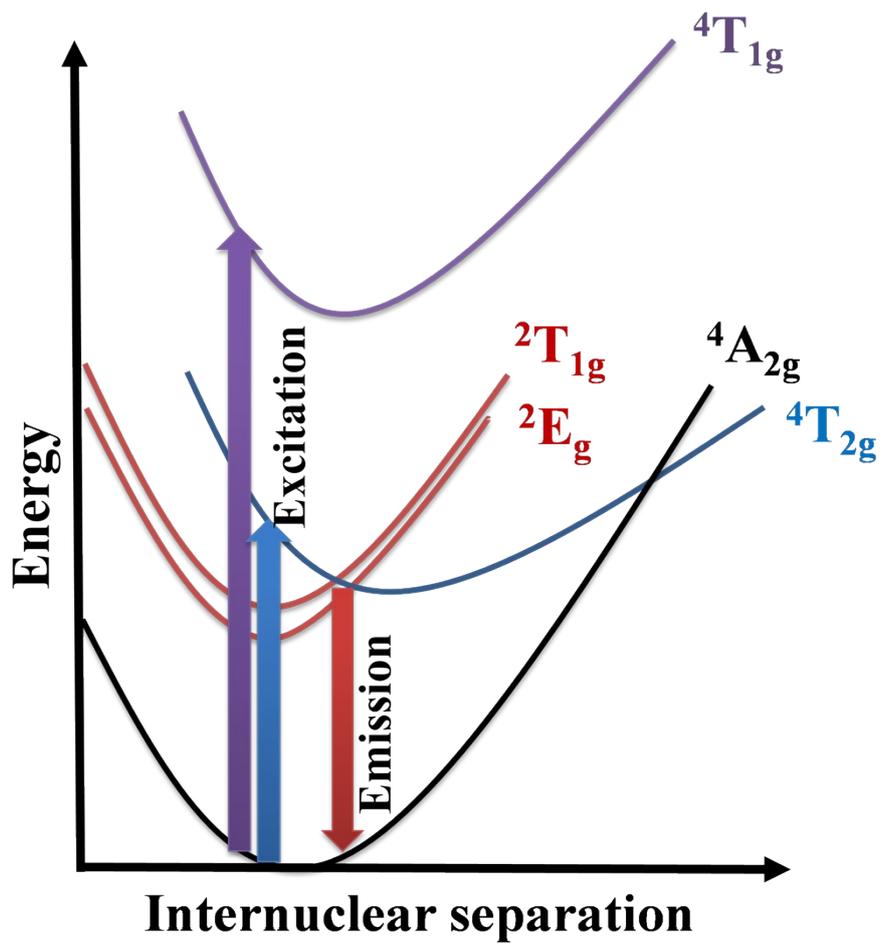


Figure S3. The configurational coordinates diagram of the energy states of Mn⁴⁺ located at the center of octahedron.

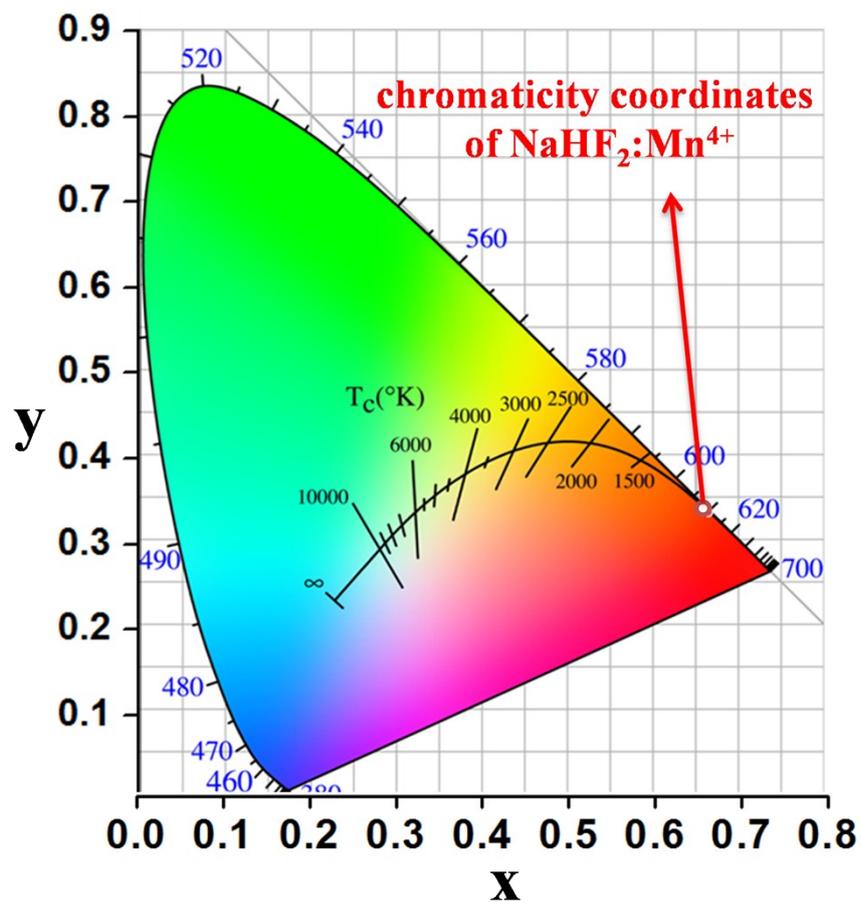


Figure S4. CIE chromaticity coordinates of the as-obtained red phosphor $\text{NaHF}_2:\text{Mn}^{4+}$.

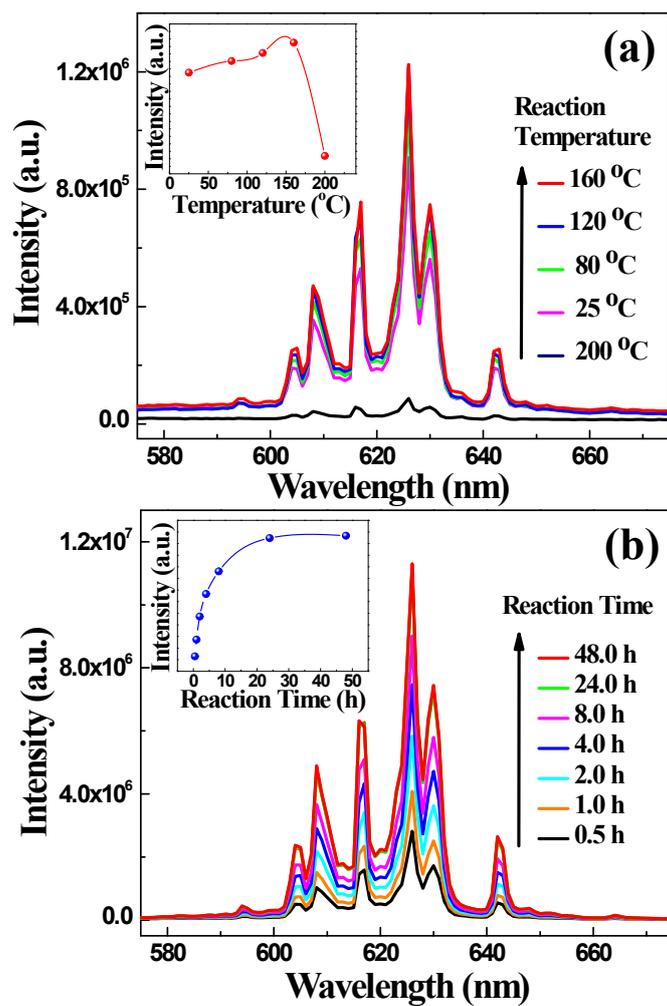


Figure S5. Dependence of emission spectra and integrated luminescence intensity of $\text{NaHF}_2:\text{Mn}^{4+}$ samples on reaction (a) temperature and (b) time. (All synthetic parameters were kept constant except for the examined variables.)

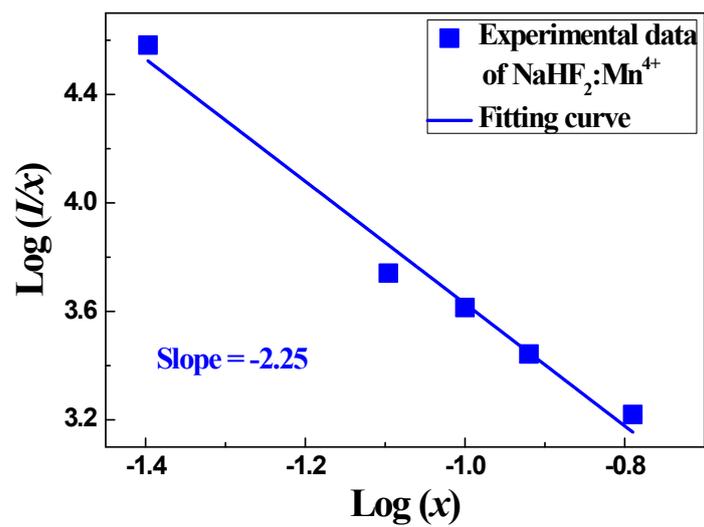


Figure S6. The relationship between $\log (x)$ versus $\log (I/x)$ in the phosphor $\text{NaHF}_2:\text{Mn}^{4+}$. (Note: x is the concentration of Mn^{4+} .)