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Enhanced thermoelectric performance in ternary spinel $\text{Cu}_4\text{Mn}_2\text{Te}_4$ via synergistic effect of tellurium-deficiency and chlorine-doping

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Experimental methods

Sample preparation and Property characterization

All of the samples were prepared by solid-state reactions. The weighing manipulation was carried out in an argon-filled glove-box (moisture and oxygen levels less than 0.1 ppm). An approximately total 4 g of reactants (Cu, 5N; Mn, 5N; Te, 5N; MnCl_2 , 3N, all purchased from Sinopharm Chemical Reagent Co., Ltd.) were mixed and loaded in carbon-coated silica tubes under a residual pressure of $\sim 10^{-3}$ Pa. The samples were placed into a temperature-controlled muffle furnace, slowly heated to 773 K in 50 h (to aid the volatilizations of raw materials), kept at this temperature for 1 day, and then heated to 1223 K in 30 h, subsequently, annealed at this temperature for 96h. After the furnace cooled to room temperature, the obtained ingots were taken out from the quartz and hand-milled into powders for follow-up hot pressed under 70 MPa at 973 K for 60 minutes in a graphite-steel mold with a 16 mm internal diameter. The obtained pellets were then cut into rectangular pieces of $\sim 10 \times 3 \times 2$ mm for electrical

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transport measurements, while coins of $\Phi \sim 10$ mm and thickness of ~ 2 mm were used for thermal diffusivity measurements. Seebeck coefficient and electrical resistivity were measured simultaneously in Ultravac ZEM-3 under a helium atmosphere. The thermal diffusivity (D) was measured through laser flash method with Netzsch LFA-457 system. The heat capacity (C_p) was indirectly derived using a representative sample (Pyroceram9606). The total thermal conductivity was calculated via $k = D \cdot C_p \cdot d$, where d was the measured density, which was determined using the dimensions and mass of the sample and then reconfirmed by measurements using Archimedes' principle on a home-built device. All the transport property measurements were performed in the temperature range of 300–773 K. Thermal gravimetric analysis (TGA) scan was measured on a NETZSCH STA 449C simultaneous analyzer under a constant flow of nitrogen gas. Differential scanning calorimetry (DSC) cycling curves were measured by the NETZSCH DTA 404PC with a heating/cooling rate of 5 K/min between 300 and 973 K under an atmosphere of N₂. The powder X-ray diffraction (XRD) patterns were taken at room temperature on a Rigaku DMAX 2500 powder X-ray diffractometer by using Cu K_α radiation.

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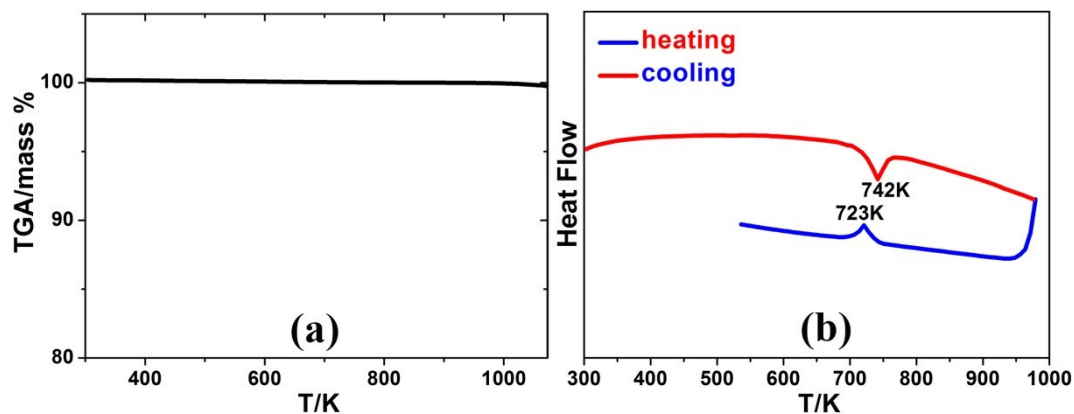


Figure S1. (a) TGA and (b) DSC diagrams of polycrystalline $\text{Cu}_4\text{Mn}_2\text{Te}_4$ sample.

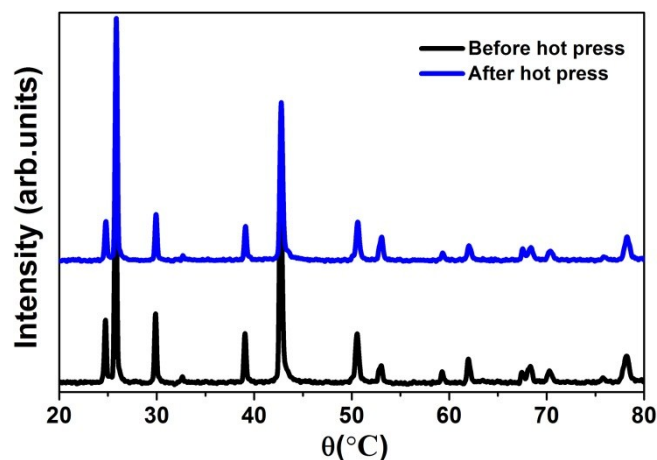


Figure S2. Powder XRD patterns of polycrystalline $\text{Cu}_4\text{Mn}_2\text{Te}_4$ sample before and after hot-pressing.

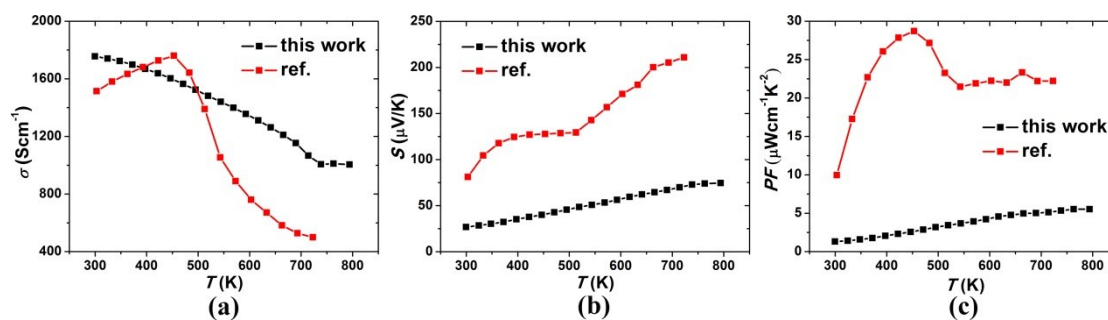


Figure S3. Temperature dependence of the electrical transport properties of polycrystalline $\text{Cu}_4\text{Mn}_2\text{Te}_4$ sample using hot-pressing: (a) electrical conductivity (σ), (b) Seebeck coefficient (S), and (c) power factor ($PF = S^2\sigma$). The data from ref.31 is also shown for comparison.

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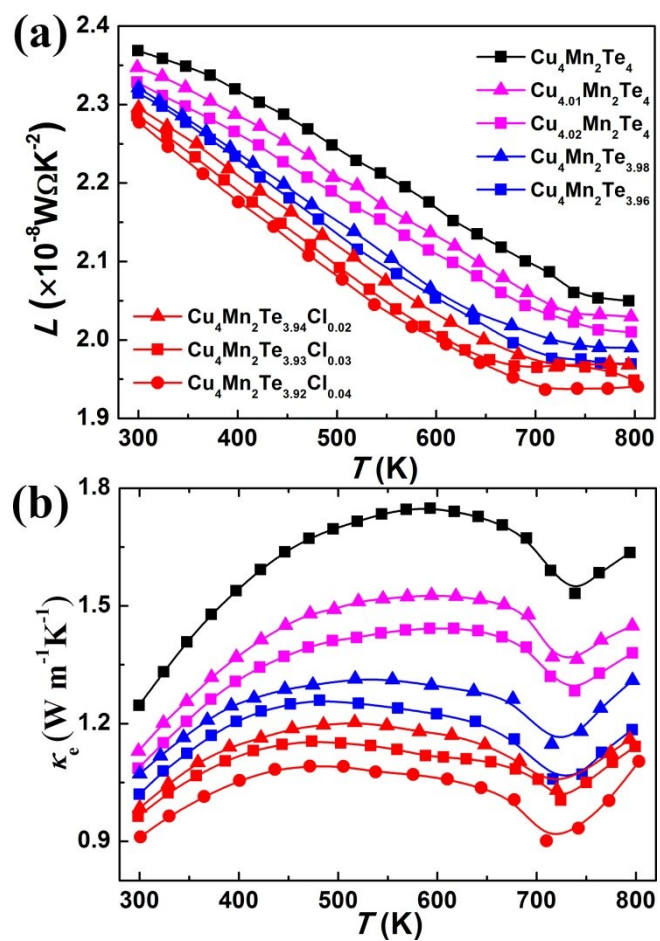


Figure S4. Temperature dependence of (a) Lorenz number (L) and electronic thermal conductivity (κ_e) for polycrystalline $\text{Cu}_{4+x}\text{Mn}_2\text{Te}_{4-\delta-y}\text{Cl}_y$ samples ($x = 0/0.01/0.02$, $\delta = 0$, $y = 0$; $x = 0$, $\delta = 0.02/0.04$, $y = 0$; $x = 0$, $\delta = 0.04$, $y = 0.02/0.03/0.04$).

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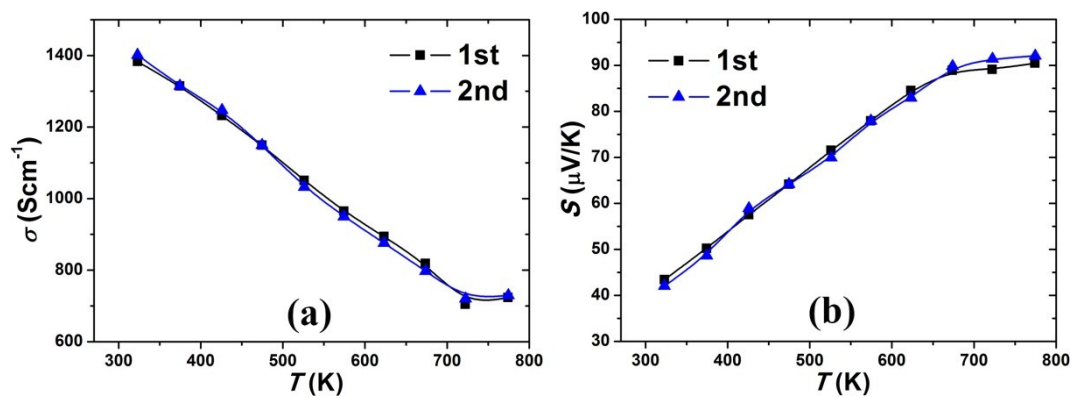


Figure S5. (a) Electrical conductivity (σ) and (b) Seebeck coefficient (S) as a function of the temperature for a hot-pressed polycrystalline sample of $\text{Cu}_4\text{Mn}_2\text{Te}_{3.93}\text{Cl}_{0.03}$ during two thermal cycles.