Magnetic Behavior of Bimetallic Layered Phases

 $M'_{0.2}Mn_{0.8}PS_3$ (M' = Zn^{II}, Cu^{II}, Ni^{II}, Co^{II})

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Fig. S1. Arrangement of Mn(II) ions and vacancies in the potassium precursor $K_{0.4}Mn_{0.8}PS_3$ · H_2O , as determined by neutron diffraction (left) [^{1,2}]; and proposed arrangement for Mn(II) and secondary transition metal ions in the studied bimetallic phases (right)





Vacancy

Secondary transition metal ion

Fig. S2: Magnetic susceptibility plots $\chi_M(T)$ and $\chi_M^{-1}(T)$ of (a) $Zn_{0.2}Mn_{0.8}PS_3 \cdot 0.25H_2O$, (b) $Ni_{0.2}Mn_{0.8}PS_3 \cdot 0.25H_2O$ and (c) $Co_{0.2}Mn_{0.8}PS_3 \cdot 0.25H_2O$. The red lines are the least-square fit with Eq (2) as described in text. The blue line is the least square fit with Eq (4) as described in the text.





Fig. S3: FC and ZFC susceptibility curves of (a) $Zn_{0.2}Mn_{0.8}PS_3 \cdot 0.25H_2O$, (b) $Cu_{0.2}Mn_{0.8}PS_3 \cdot 0.25H_2O$, (c) $Ni_{0.2}Mn_{0.8}PS_3 \cdot 0.25H_2O$ and (d) $Co_{0.2}Mn_{0.8}PS_3 \cdot 0.25H_2O$.



Fig. S4: (a) to (c) Field dependence of magnetization; and (d) to (f) first derivative curves. Measurements made for $Zn_{0.2}Mn_{0.8}PS_3$, $Ni_{0.2}Mn_{0.8}PS_3$ and $Co_{0.2}Mn_{0.8}PS_3$ at different temperatures.



Fig. S5: EPR spectra of (a) $Zn_{0.2}Mn_{0.8}PS_3 \cdot 0.25H_2O$ and (b) $Ni_{0.2}Mn_{0.8}PS3 \cdot 0.25H_2O$.

Fig. S6. $I_{\text{DIN}}(T)$ of the studied phases.



Fig. S7: Comparisons between EPR intensities after double integration (I_{DIN}) with magnetic susceptibility data (χ_M), and EPR intensities ($I_{DIN}T$) with magnetic susceptibility ($\chi_M T$) temperature product data of pristine and bimetallic phases: (a) MnPS₃, (b) Zn_{0.2}Mn_{0.8}PS₃·0.25H₂O, (c) Cu_{0.2}Mn_{0.8}PS₃·0.25H₂O, (d) Ni_{0.2}Mn_{0.8}PS₃·0.25H₂O and (e) Co_{0.2}Mn_{0.8}PS₃·0.25H₂O.











Fig S8. First derivative of the $\chi T(T)$ plot for the pristine and bimetallic phases: (a) MnPS₃, (b) $Zn_{0.2}Mn_{0.8}PS_3 \cdot 0.25H_2O$, (c) $Cu_{0.2}Mn_{0.8}PS_3 \cdot 0.25H_2O$, (d) $Ni_{0.2}Mn_{0.8}PS_3 \cdot 0.25H_2O$ and (e) $Co_{0.2}Mn_{0.8}PS_3 \cdot 0.25H_2O$.









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