Electronic Supporting Information

Cluster transformation of $[Cu_3(\mu_3-H)(\mu_3-BH_4)((PPh_2)_2NH)_3](BF_4)$ to $[Cu_3(\mu_3-H)(\mu_2,\mu_1-S_2CH)((PPh_2)_2NH)_3](BF_4)$ via reaction with CS₂. X-ray structural characterisation and reactivity of cationic clusters explored by multistage mass spectrometry and computational studies.

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Figure S1: ORTEP-3 representation of the cluster $[Cu_4(L^{Ph}-H+2S)_3](BF_4)$ (**3.BF**₄). Displacement ellipsoids are set at the 30% probability level.

Table S1: Crystal data and structure refinement for cluster 3.BF₄

| Identification code | LiJ170519 |
|--------------------------------------|--|
| Empirical formula | $C_{4.41}H_{4.09}B_{0.06}Cl_{0.45}Cu_{0.23}F_{0.23}N_{0.17}P_{0.35}S_{0.35}$ |
| Formula weight | 117.18 |
| Temperature/K | 130.00(10) |
| Crystal system | monoclinic |
| Space group | $P2_1/c$ |
| a/Å | 14.6628(5) |
| b/Å | 18.9702(14) |
| c/Å | 30.924(3) |
| a/° | 90 |
| β/° | 90 |
| γ/° | 90 |
| Volume/Å ³ | 8601.7(11) |
| Ζ | 69 |
| $\rho_{calc}g/cm^3$ | 1.561 |
| μ/mm^{-1} | 6.186 |
| F(000) | 4092.0 |
| Crystal size/mm ³ | 0.2 	imes 0.2 	imes 0.1 |
| Radiation | $CuK\alpha \ (\lambda = 1.54184)$ |
| 2Θ range for data collection/ | ^o 7.376 to 154.066 |
| Index ranges | $-17 \le h \le 6, -19 \le k \le 22, -38 \le l \le 24$ |

 Reflections collected
 26830

 Independent reflections
 11696 [$R_{int} = 0.0290$, $R_{sigma} = 0.0390$]

 Data/restraints/parameters
 11696/0/890

 Goodness-of-fit on F²
 1.087

 Final R indexes [I>=2 σ (I)]
 R₁ = 0.0587, wR₂ = 0.1536

 Final R indexes [all data]
 R₁ = 0.0675, wR₂ = 0.1613

 Largest diff. peak/hole / e Å⁻³ 1.32/-0.91

Table S2: Bond distances of X-ray crystallography for cluster 3.BF₄.

| Cu1 | Cu2 | 2.8735(11) | C21 | C22 | 1.3900 |
|------------|-----|------------|-----|-----|--------|
| Cu1 | Cu3 | 2.7763(12) | C21 | C20 | 1.3900 |
| Cul | Cu4 | 2.7331(13) | C22 | C23 | 1.3900 |
| Cul | S1 | 2.2506(15) | C23 | C24 | 1.3900 |
| Cul | S5 | 2.2673(16) | C24 | C19 | 1.3900 |
| Cul | S3 | 2.2633(14) | C19 | C20 | 1.3900 |
| Cu2 | Cu3 | 2.8317(10) | C68 | C69 | 1.3900 |
| Cu2 | Cu4 | 2.7840(11) | C68 | C67 | 1.3900 |
| Cu2 | S4 | 2.2491(16) | C69 | C70 | 1.3900 |
| Cu2 | S3 | 2.2665(17) | C70 | C71 | 1.3900 |
| Cu2 | S7 | 2.2656(15) | C71 | C72 | 1.3900 |
| Cu3 | Cu4 | 2.7958(11) | C72 | C67 | 1.3900 |
| Cu3 | S1 | 2.2624(16) | C38 | C39 | 1.3900 |
| Cu3 | S6 | 2.2597(16) | C38 | C37 | 1.3900 |
| Cu3 | S7 | 2.2546(14) | C39 | C40 | 1.3900 |
| Cu4 | S4 | 2.2679(13) | C40 | C41 | 1.3900 |
| Cu4 | S5 | 2.2830(15) | C41 | C42 | 1.3900 |
| Cu4 | S6 | 2.2663(14) | C42 | C37 | 1.3900 |
| S4 | P5 | 2.0485(19) | C46 | C47 | 1.3900 |
| P5 | N2 | 1.580(5) | C46 | C45 | 1.3900 |
| P5 | C67 | 1.822(3) | C47 | C48 | 1.3900 |
| P5 | C37 | 1.809(2) | C48 | C43 | 1.3900 |
| S 1 | P3 | 2.0475(18) | C43 | C44 | 1.3900 |
| P6 | S3 | 2.0548(18) | C44 | C45 | 1.3900 |
| P6 | N2 | 1.583(5) | C11 | C10 | 1.3900 |
| P6 | C19 | 1.801(3) | C11 | C12 | 1.3900 |
| P6 | C25 | 1.808(3) | C10 | C9 | 1.3900 |
| P2 | S6 | 2.0507(19) | C9 | C8 | 1.3900 |
| P2 | N1 | 1.599(4) | C8 | C7 | 1.3900 |
| P2 | C55 | 1.796(3) | C7 | C12 | 1.3900 |
| P2 | C49 | 1.815(3) | C29 | C28 | 1.3900 |
| S5 | P4 | 2.0481(16) | C29 | C30 | 1.3900 |
| S7 | P1 | 2.055(2) | C28 | C27 | 1.3900 |
| P1 | C31 | 1.792(3) | C27 | C26 | 1.3900 |
| P1 | N1 | 1.585(5) | C26 | C25 | 1.3900 |
| P1 | C43 | 1.813(3) | C25 | C30 | 1.3900 |
| P4 | N3 | 1.589(4) | C53 | C54 | 1.3900 |
| P4 | C61 | 1.802(3) | C53 | C52 | 1.3900 |

| P4 | C13 | 1.814(3) | C54 | C49 | 1.3900 |
|-----|-----|----------|------|-----|-----------|
| P3 | N3 | 1.591(5) | C49 | C50 | 1.3900 |
| P3 | C7 | 1.809(3) | C50 | C51 | 1.3900 |
| P3 | C1 | 1.804(3) | C51 | C52 | 1.3900 |
| C36 | C35 | 1.3900 | C6 | C5 | 1.3900 |
| C36 | C31 | 1.3900 | C6 | C1 | 1.3900 |
| C35 | C34 | 1.3900 | C5 | C4 | 1.3900 |
| C34 | C33 | 1.3900 | C4 | C3 | 1.3900 |
| C33 | C32 | 1.3900 | C3 | C2 | 1.3900 |
| C32 | C31 | 1.3900 | C2 | C1 | 1.3900 |
| C56 | C57 | 1.3900 | Cl6 | C76 | 1.760(8) |
| C56 | C55 | 1.3900 | F5 | B1 | 1.363(10) |
| C57 | C58 | 1.3900 | F2 | B1 | 1.30(3) |
| C58 | C59 | 1.3900 | F4 | B1 | 1.362(13) |
| C59 | C60 | 1.3900 | B1 | F7 | 1.377(19) |
| C60 | C55 | 1.3900 | B1 | F6 | 1.395(15) |
| C62 | C61 | 1.3900 | B1 | F3 | 1.318(19) |
| C62 | C63 | 1.3900 | B1 | F1 | 1.314(19) |
| C61 | C66 | 1.3900 | Cl9 | C77 | 1.734(16) |
| C66 | C65 | 1.3900 | Cl4 | C73 | 1.741(10) |
| C65 | C64 | 1.3900 | C110 | C77 | 1.66(3) |
| C64 | C63 | 1.3900 | Cl5 | C73 | 1.760(10) |
| C14 | C15 | 1.3900 | C77 | C18 | 1.788(14) |
| C14 | C13 | 1.3900 | Cl7 | C76 | 1.733(9) |
| C15 | C16 | 1.3900 | Cl5A | C74 | 1.87(3) |
| C16 | C17 | 1.3900 | C74 | Cl1 | 1.39(4) |
| C17 | C18 | 1.3900 | Cl1 | C75 | 1.83(4) |
| C18 | C13 | 1.3900 | Cl3 | C75 | 1.71(3) |

Crystal Data for C_{4.405797}H_{4.094203}B_{0.057971}Cl_{0.451594}Cu_{0.231884}F_{0.231884}N_{0.173913}P_{0.347826}S_{0.347826} (M=117.18 g/mol): monoclinic, space group P2₁/c (no. 14), a = 14.6628(5) Å, b = 18.9702(14) Å, c = 30.924(3) Å, β = 90°, V = 8601.7(11) Å³, Z = 69, T = 130.00(10) K, µ(CuK α) = 6.186 mm⁻¹, *Dcalc* = 1.561 g/cm³, 26830 reflections measured (7.376° ≤ 2 Θ ≤ 154.066°), 11696 unique (R_{int} = 0.0290, R_{sigma} = 0.0390) which were used in all calculations. The final R_1 was 0.0587 (I > 2 σ (I)) and wR_2 was 0.1613 (all data).

Table S3: Crystal data and structure refinement for cluster 2a.BF₄

| LiJCu3CS2_orange_20170407 |
|---|
| $C_{76.94}H_{71.84}BCl_{3.9}Cu_3F_4N_4P_6S_2$ |
| 1718.30 |
| 130.01(10) |
| triclinic |
| P-1 |
| 14.5606(5) |
| 15.3131(5) |
| 18.0023(8) |
| 79.962(3) |
| 84.904(3) |
| 84.146(3) |
| 3921.5(3) |
| |

| 2 |
|--|
| 1.455 |
| 4.268 |
| 1756.0 |
| 0.1 	imes 0.1 	imes 0.1 |
| $CuK\alpha \ (\lambda = 1.54184)$ |
| 7.058 to 154.182 |
| $-17 \le h \le 18, -19 \le k \le 16, -22 \le l \le 22$ |
| 31081 |
| 16180 [$R_{int} = 0.0377$, $R_{sigma} = 0.0550$] |
| 16180/0/821 |
| 1.089 |
| $R_1 = 0.0493, wR_2 = 0.1233$ |
| $R_1 = 0.0681, wR_2 = 0.1331$ |
| |

 Table S4: Bond distances of X-ray crystallography for cluster 2a.BF₄.

| Cu1 | Cu2 | 2.6530(7) | C60C55 | 1.3900 |
|------------|------------|------------|---------|----------|
| Cu1 | Cu3 | 2.7278(7) | C46C45 | 1.3900 |
| Cu1 | P6 | 2.2588(10) | C46C47 | 1.3900 |
| Cu1 | P1 | 2.2491(10) | C45C44 | 1.3900 |
| Cu1 | S 1 | 2.3568(12) | C44 C43 | 1.3900 |
| Cu2 | Cu3 | 2.5853(7) | C43 C48 | 1.3900 |
| Cu2 | P5 | 2.2542(9) | C48C47 | 1.3900 |
| Cu2 | P4 | 2.2480(10) | N4 C73 | 1.132(6) |
| Cu2 | S2 | 2.3985(12) | C69C70 | 1.3900 |
| Cu3 | P3 | 2.2687(10) | C69C68 | 1.3900 |
| Cu3 | P2 | 2.2638(10) | C70C71 | 1.3900 |
| Cu3 | S2 | 2.4620(12) | C71 C72 | 1.3900 |
| P6 | N1 | 1.693(3) | C72C67 | 1.3900 |
| P6 | C1 | 1.8322(18) | C67C68 | 1.3900 |
| P6 | C37 | 1.829(2) | C9 C10 | 1.3900 |
| P5 | N1 | 1.676(3) | C9 C8 | 1.3900 |
| P5 | C55 | 1.841(2) | C10C11 | 1.3900 |
| P5 | C61 | 1.824(2) | C11C12 | 1.3900 |
| P1 | N2 | 1.696(3) | C12C7 | 1.3900 |
| P1 | C67 | 1.832(2) | C7 C8 | 1.3900 |
| P1 | C7 | 1.829(2) | C24 C23 | 1.3900 |
| P4 | N3 | 1.688(3) | C24 C19 | 1.3900 |
| P4 | C43 | 1.824(2) | C23 C22 | 1.3900 |
| P4 | C49 | 1.837(2) | C22C21 | 1.3900 |
| P3 | N3 | 1.683(3) | C21 C20 | 1.3900 |
| P3 | C31 | 1.832(2) | C20C19 | 1.3900 |
| P3 | C25 | 1.832(2) | C40C41 | 1.3900 |
| P2 | N2 | 1.685(3) | C40C39 | 1.3900 |
| P2 | C13 | 1.829(2) | C41 C42 | 1.3900 |
| P2 | C19 | 1.835(2) | C42C37 | 1.3900 |
| S 1 | C75 | 1.637(5) | C37C38 | 1.3900 |
| S2 | C75 | 1.706(4) | C38C39 | 1.3900 |
| Cl2 | C76 | 1.762(7) | C64C63 | 1.3900 |

| 1.76(3) | C64C65 | 1.3900 |
|-----------|--|--|
| 1.741(15) | C63C62 | 1.3900 |
| 1.349(6) | C62C61 | 1.3900 |
| 1.3900 | C61C66 | 1.3900 |
| 1.3900 | C66C65 | 1.3900 |
| 1.3900 | C73C74 | 1.449(7) |
| 1.3900 | F97 B1 | 1.283(11) |
| 1.3900 | C52C51 | 1.3900 |
| 1.3900 | C52C53 | 1.3900 |
| 1.492(18) | C51C50 | 1.3900 |
| 1.3900 | C50C49 | 1.3900 |
| 1.3900 | C49C54 | 1.3900 |
| 1.3900 | C54C53 | 1.3900 |
| 1.3900 | F95 B1 | 1.489(9) |
| 1.3900 | C27C28 | 1.3900 |
| 1.3900 | C27C26 | 1.3900 |
| 1.3900 | C28C29 | 1.3900 |
| 1.3900 | C29C30 | 1.3900 |
| 1.3900 | C30C25 | 1.3900 |
| 1.3900 | C25C26 | 1.3900 |
| 1.3900 | C76 Cl1 | 1.663(18) |
| 1.3900 | C76 Cl3 | 1.751(6) |
| 1.3900 | B1 F97A | 1.626(15) |
| 1.3900 | B1 F95A | 1.405(10) |
| 1.3900 | B1 F99A | 1.243(9) |
| 1.3900 | C77 Cl6 | 1.739(8) |
| 1.3900 | C77 Cl4 | 1.726(8) |
| | 1.76(3) 1.741(15) 1.349(6) 1.3900 1.3900 1.3900 1.3900 1.3900 1.492(18) 1.3900 1.390 | $\begin{array}{cccccccccccccccccccccccccccccccccccc$ |

Crystal Data for C_{76.94}H_{71.845}BCl_{3.905}Cu₃F₄N₄P₆S₂ (M =1718.30 g/mol): triclinic, space group P-1 (no. 2), a = 14.5606(5) Å, b = 15.3131(5) Å, c = 18.0023(8) Å, a = 79.962(3)°, β = 84.904(3)°, γ = 84.146(3)°, V = 3921.5(3) Å³, Z = 2, T = 130.01(10) K, μ (CuK α) = 4.268 mm⁻¹, *Dcalc* = 1.455 g/cm³, 31081 reflections measured (7.058° ≤ 2 Θ ≤ 154.182°), 16180 unique (R_{int} = 0.0377, R_{sigma} = 0.0550) which were used in all calculations. The final R_1 was 0.0493 (I > 2 σ (I)) and wR_2 was 0.1331 (all data).

Table S5: Comparison of bond distances (in Å) of cluster $2a.BF_4$ with reported values in relatedtrinuclear copper clusters from previously reported X-ray crystallographic studies. $2a.BF_4$ (thiswork), $1a.BF_4^2$ ([Cu₃(μ_3 -H)(μ_3 -BH₄)((Ph₂P)₂NH)₃](BF₄)), Che²² ([Cu₃(μ_3 -H)(dcpm)₃](BF₄)₂),Norton²³ ([Cu₃(μ -H)₃(dppbz)]) and Hayton²¹ ([Cu₃(μ_3 -H)(OAc)₂(dppm)₃]).

| | 2a.BF ₄ | 1a.BF ₄ | Che ²² | Norton ²³ | Hayton ²¹ |
|-------|--------------------|--------------------|-------------------|----------------------|----------------------|
| Cu-Cu | 2.6530(7) | 2.6164(5) | 2.866(1) | 2.564(2) | 2.816(1) |
| | 2.7278(7) | 2.6706(5) | 2.906(1) | 2.555(1) | 2.8077(9) |
| | 2.5853(7) | 2.6785(5) | 2.865(1) | 2.619(1) | 3.114 |
| Cu-H | 1.73(4) | 1.72(4) | 1.62(4) | 1.70(3) | 1.58(4) |
| | 1.80(5) | 1.79(3) | 1.63(4) | 1.59(3) | 1.79(4) |
| | 1.74(4) | 1.77(4) | 1.74(5) | 1.62(3) | 1.72(3) |
| Cu-P | 2.2588(10) | 2.2624(7) | 2.279(2) | 2.277(1) | 2.279(1) |
| | 2.2491(10) | 2.2709(6) | 2.290(2) | 2.277(1) | 2.274(1) |
| | 2.2542(9) | 2.2815(9) | 2.279(2) | 2.292(1) | 2.267(1) |
| | 2.2480(10) | 2.2681(7) | 2.277(2) | 2.293(1) | 2.284(1) |

| 2.2687(10) | 2.2731(7) | 2.288(2) | 2.298(1) | 2.284(1) |
|------------|-----------|----------|----------|----------|
| 2.2638(10) | 2.2525(9) | 2.287(2) | 2.292(2) | 2.283(1) |

Table S6: Comparison of key metal-sulfur bond distance (in Å), S-C-S bond angle (in °) and ¹H NMR resonance signal of the ⁻S₂CH unit (in ppm) of cluster **2a.BF**₄ with reported values in related dithioformate containing metallic clusters. **2a.BF**₄ (this work), **Adams**^{24a} ([Os₃(μ -H)(μ ₂-S₂CH)(CO)₁₀]), **Böttcher**^{24b} ([Ru₂(CO)₄(μ ₂-S₂CH)(μ -P^tBu₂)(μ -dppm)]), **Bianchini1**^{5c} ((PPh₃)₂Cu(K_2 -S₂CSCH₂SCS₂)Cu(PPh₃)₂) and **Bianchini2**^{5c} (Cu(K^2 -S₂CH)(triphos)).

| | 2a.BF ₄ | Adams ^{24a} | Böttcher ^{24b} | Bianchini1 ^{5c} | Bianchini2 ^{5c} |
|--------------------------------|--------------------|----------------------|--------------------------------|---------------------------------|--------------------------|
| M-S | 2.3568(12) | 2.445(7) | 2.4692(1) | 2.411(5) | n/a |
| | 2.3985(12) | 2.454(7) | 2.4303(1) | 2.479(5) | |
| | 2.4620(12) | | | 2.448(5) | |
| | | | | 2.449(5) | |
| S-C-S | 128.7(3) | 132.0(16) | 129.4(3) | 122.2(9) | n/a |
| | | | | 124.4(9) | |
| | | | | 113.3(9) | |
| | | | | 120.9(10) | |
| | | | | 113.7(9) | |
| | | | | 125.3(10) | |
| | | | | 113.6(10) | |
| ⁻ S ₂ CH | 9.87 | n/a | 10.67 | n/a | 11.26 |



Figure S2: ¹H NMR spectra of cluster **2a.BF**₄. Deuterochloroform (CDCl₃) was used as the solvent for the measurement in a 500 MHz NMR spectrometer.



Figure S3: ¹³C{¹H} NMR spectra of cluster **2a.BF**₄. Deuterochloroform (CDCl₃) was used as the solvent for the measurement in a 500 MHz NMR spectrometer.



Figure S4: ³¹P{¹H} NMR spectra of cluster **2a.BF**₄. Deuterochloroform (CDCl₃) was used as the solvent for the measurement in a 500 MHz NMR spectrometer.



Figure S5: ¹¹B{¹H} NMR spectra of cluster **2a.BF**₄. Deuterochloroform (CDCl₃) was used as the solvent for the measurement in a 500 MHz NMR spectrometer.

| 161.40 ppm |
|------------|
| |
| |
| |

Figure S6: ¹⁹ $F{^1H}$ NMR spectra of cluster **2a.BF**₄. Deuterochloroform (CDCl₃) was used as the solvent for the measurement in a 500 MHz NMR spectrometer.

-150

. -155 . -160 -165

-170

-175

-180

-185

-190

-195

-200



-100

. -105 . -110 -115

-120

-125

. -130 . -135 -140

-145

Figure S7: UV/Vis absorption spectra of cluster $2a.BF_4$ dissolved in dichloromethane (CH₂Cl₂) at a concentration of 20 μ M.



Figure S8: ESI-MS of 2a.BF₄ in the positive ion mode, HRMS a) measured, b) stimulated.



Figure S9: ESI-MS of 3.BF₄ in the positive ion mode, HRMS a) measured, b) stimulated.



Figure S10: DFT calculated vibrational vectors of C-S and C-H stretches in the model system for cluster **2a.BF**₄, **2b**, whereby the phenyl rings of the dppa ligand have been replaced with methyl groups. Vectors are shown as red arrows. Vibrational frequencies are unscaled. Level of theory: M06/6-31+G(d)/SDD. Hydrogen atoms of the ligand have been removed for clarity.



Figure S11: DFT Calculated vibrational vectors of Cu-H bonds in the model system for cluster **2a.BF**₄, **2b**, whereby the phenyl rings of the dppa ligand have been replaced with methyl groups. Vectors are shown as red arrows. Vibrational frequencies are unscaled. Level of theory: M06/6-31+G(d)/SDD. Hydrogen atoms of the ligand have been removed for clarity.







TS5b-6b



7b



6b





Figure S12: DFT calculated structures illustrating key interactions and bond distances within each fragment. Hydrogens on the dmpa (L^{Me}) ligand have been removed for clarity.

Cartesian coordinates of DFT calculated structures associated with Figure 8

E(B1) = energy of optimized structure for basis set 1 (M06/6-31+G(d))E(ZPE) = zero-point energy of optimized structure for basis set 1 (M06/6-31+G(d))E(B2) = single point energy at basis set 2 (M06/def2-TZVP)

 $[Cu_3(\mu_3-H)(\mu_2,\mu_1-S_2CH)(L^{Me})_3]^+$ (2b)



Cu3S2C13P6N3H41 E(B1) = -4119.834563 Hartrees E(ZPE) = 0.536712 Hartrees E(B2) = -8449.680825 Hartrees

| Cu | 1.52364 | 0.16382 | -0.28047 |
|----|----------|----------|----------|
| Cu | -0.67261 | -1.35914 | -0.05122 |
| Cu | -0.94942 | 1.15251 | -0.02565 |
| Р | 2.91730 | -1.59362 | -0.75776 |
| Р | 0.62253 | -3.17326 | 0.38427 |
| Р | 2.39608 | 2.20977 | -0.84894 |
| Р | -2.83913 | -1.83637 | -0.65689 |
| Р | -3.16666 | 1.17006 | -0.63360 |
| Р | -0.10804 | 3.19689 | 0.50221 |
| S | 1.81548 | 0.25120 | 2.13188 |
| S | -1.20506 | -0.18051 | 2.12861 |
| Ν | 2.17294 | -3.08593 | -0.34474 |
| Ν | -3.73944 | -0.41037 | -0.97904 |
| Ν | 1.48550 | 3.44407 | -0.07735 |
| Н | 1.86778 | 4.39094 | -0.05364 |
| С | -4.28755 | 1.75329 | 0.69886 |
| С | 4.47778 | -1.64127 | 0.20852 |
| С | -0.03789 | 3.70144 | 2.26033 |
| С | 0.02430 | -4.78571 | -0.25193 |
| С | -3.24046 | -2.87989 | -2.10918 |
| С | 0.31772 | 0.02750 | 2.84300 |
| С | 2.46885 | 2.74590 | -2.60182 |
| С | 4.09714 | 2.60212 | -0.28939 |

| С | -1.00633 | 4.62165 | -0.22737 |
|---|----------|----------|----------|
| С | 3.55683 | -1.89737 | -2.44870 |
| С | 0.94471 | -3.59556 | 2.13863 |
| С | -3.82304 | -2.66956 | 0.64991 |
| С | -3.80103 | 2.12350 | -2.06441 |
| Н | 0.30945 | 0.01784 | 3.93961 |
| Н | -4.73662 | -0.51779 | -1.17751 |
| Н | 2.76171 | -3.92032 | -0.30611 |
| Н | -0.05699 | -0.01183 | -1.00197 |
| Н | 5.09500 | -0.76720 | -0.04051 |
| Η | 4.24963 | -1.60509 | 1.28140 |
| Н | 5.06244 | -2.54788 | -0.00409 |
| Н | 4.16189 | -2.81332 | -2.49875 |
| Н | 2.72117 | -1.97887 | -3.15285 |
| Н | 4.18444 | -1.04743 | -2.75210 |
| Н | 0.73899 | -5.59440 | -0.04300 |
| Н | -0.92908 | -5.03987 | 0.23123 |
| Н | -0.13405 | -4.72223 | -1.33485 |
| Н | 1.45025 | -2.75399 | 2.62989 |
| Η | -0.00950 | -3.77182 | 2.65436 |
| Η | 1.56944 | -4.49544 | 2.22516 |
| Η | -3.39451 | -3.65792 | 0.86632 |
| Η | -3.78568 | -2.07364 | 1.57090 |
| Η | -4.87178 | -2.80240 | 0.34850 |
| Η | -2.84768 | -3.89344 | -1.95360 |
| Η | -4.32667 | -2.95092 | -2.26226 |
| Η | -2.77902 | -2.46110 | -3.01028 |
| Η | -3.25086 | 1.84875 | -2.97111 |
| Η | -4.87200 | 1.93484 | -2.22465 |
| Η | -3.66523 | 3.19799 | -1.88430 |
| Η | -4.07078 | 2.80553 | 0.93117 |
| Η | -5.34586 | 1.66840 | 0.41405 |
| Η | -4.10947 | 1.16307 | 1.60708 |
| Н | 0.35051 | 4.72345 | 2.36869 |
| Н | -1.04836 | 3.65491 | 2.68988 |
| Н | 0.60705 | 3.00921 | 2.81510 |
| Н | -1.10707 | 4.48202 | -1.31098 |
| Η | -2.01015 | 4.69287 | 0.21354 |
| Η | -0.48479 | 5.57048 | -0.03672 |
| Н | 4.35965 | 3.65379 | -0.47333 |
| Η | 4.19482 | 2.38936 | 0.78250 |
| Н | 4.81035 | 1.97078 | -0.83706 |
| Н | 3.12387 | 2.07176 | -3.17056 |
| Н | 1.46166 | 2.68757 | -3.03360 |
| Н | 2.84550 | 3.77334 | -2.70069 |

 $[Cu_3(\mu_3-H)(\mu_1,K_2-S_2CH)(L^{Me})_2]^+$ (4b)



Cu3S2C9P4N2H28 E(B1) = -3222.413246 Hartrees E(ZPE) = 0.366765 Hartrees E(B2) = -7552.109705 Hartrees

| Cu | 0.71066 | 0.84074 | -0.73466 |
|----|----------|----------|----------|
| Cu | -1.66390 | -0.46066 | -0.65117 |
| Cu | 0.42444 | -1.64784 | -0.01869 |
| Р | -0.28608 | 2.68920 | 0.13845 |
| Р | -2.51917 | 0.87150 | 0.93380 |
| Р | 2.99760 | 0.66344 | -0.90753 |
| Р | 2.33279 | -1.46824 | 1.11372 |
| S | -3.51873 | -1.58637 | -1.58785 |
| S | -1.43959 | -2.92691 | 0.21753 |
| Ν | -1.52430 | 2.24350 | 1.23819 |
| Ν | 3.45853 | -0.44522 | 0.31861 |
| Н | 4.44373 | -0.50800 | 0.57927 |
| С | -1.07383 | 3.83484 | -1.05592 |
| С | 3.32440 | -2.96178 | 1.45121 |
| С | -2.70505 | 0.08513 | 2.56760 |
| С | -2.07264 | -2.41593 | -1.33122 |
| С | 4.10681 | 2.07947 | -0.56846 |
| С | 0.71251 | 3.83218 | 1.15872 |
| С | -4.18109 | 1.56513 | 0.63040 |
| С | 3.70858 | -0.01909 | -2.44861 |
| С | 2.12613 | -0.74262 | 2.78224 |
| Η | 1.61831 | 0.22725 | 2.69101 |
| Н | 1.49932 | -1.40235 | 3.39680 |
| Н | 3.09390 | -0.59934 | 3.28154 |
| Н | 4.78279 | -0.21835 | -2.33582 |
| Н | 3.19286 | -0.95403 | -2.69943 |
| Н | 3.56348 | 0.68653 | -3.27677 |
| Н | 0.09672 | 4.63236 | 1.59244 |
| Н | 1.20630 | 3.28219 | 1.96864 |
| Н | 1.48256 | 4.29634 | 0.52909 |
| Н | -0.31887 | 4.26956 | -1.72384 |
| Н | -1.79432 | 3.28110 | -1.67141 |
| Н | -1.59671 | 4.64839 | -0.53452 |
| Н | -1.90619 | 2.98303 | 1.83171 |

| Η | -1.47900 | -2.70181 | -2.20647 |
|---|----------|----------|----------|
| Η | 4.24451 | -2.71048 | 1.99718 |
| Η | 2.73411 | -3.65442 | 2.06335 |
| Η | 3.58361 | -3.46520 | 0.51366 |
| Η | -3.12354 | 0.77900 | 3.30907 |
| Η | -3.37170 | -0.78110 | 2.46212 |
| Η | -1.72725 | -0.27629 | 2.90841 |
| Η | 3.97543 | 2.84550 | -1.34314 |
| Η | 3.86285 | 2.51982 | 0.40577 |
| Η | 5.16216 | 1.77272 | -0.56722 |
| Η | -4.18383 | 2.14274 | -0.30188 |
| Η | -4.89360 | 0.73711 | 0.52240 |
| Η | -4.50779 | 2.20789 | 1.45935 |
| Η | -0.06292 | -0.53318 | -1.38501 |

$[Cu_3(\mu_2,\mu_2-S_2CH_2)(L^{Me})_2]^+$ (5b)



Cu3S2C9P4N2H28 E(B1) = -3222.466461 Hartrees E(ZPE) = 0.372398 Hartrees E(B2) = -7552.159374 Hartrees

| Cu | 0.23479 | 0.37398 | -1.28120 |
|----|----------|----------|----------|
| Cu | -1.56323 | -1.30993 | -0.14722 |
| Cu | 0.87185 | -1.37758 | 0.85706 |
| Р | -1.17066 | 2.08884 | -0.63902 |
| Р | -2.69360 | 0.13083 | 1.08891 |
| Р | 2.45282 | 0.78891 | -0.99918 |
| Р | 2.01738 | 0.22289 | 1.88242 |
| S | -0.85375 | -1.29575 | -2.36726 |
| S | -0.07727 | -3.12464 | -0.06764 |
| Ν | -2.04341 | 1.70654 | 0.80278 |
| Ν | 2.73814 | 1.17916 | 0.66134 |
| Н | 3.59532 | 1.68363 | 0.89659 |
| С | -2.45051 | 2.47071 | -1.89541 |
| С | 3.42583 | -0.32656 | 2.90651 |
| С | -2.63436 | 0.03241 | 2.91395 |
| С | 0.14197 | -2.76070 | -1.86631 |
| С | 3.05418 | 2.30846 | -1.81958 |

| С | -0.56316 | 3.76269 | -0.20832 |
|---|----------|----------|----------|
| С | -4.49069 | 0.32015 | 0.79108 |
| С | 3.76509 | -0.41530 | -1.43124 |
| С | 1.15245 | 1.40666 | 2.97041 |
| Н | 0.26010 | 1.79142 | 2.45847 |
| Н | 0.83713 | 0.90044 | 3.89250 |
| Н | 1.81278 | 2.24301 | 3.23633 |
| Η | 4.76397 | -0.01935 | -1.20140 |
| Н | 3.60996 | -1.34844 | -0.87301 |
| Н | 3.71351 | -0.65029 | -2.50260 |
| Н | -1.37533 | 4.41545 | 0.14230 |
| Н | 0.20950 | 3.69857 | 0.56785 |
| Н | -0.12179 | 4.22406 | -1.10116 |
| Н | -1.98017 | 2.90249 | -2.78880 |
| Н | -2.94980 | 1.54136 | -2.19955 |
| Н | -3.19806 | 3.17793 | -1.50924 |
| Н | -2.58357 | 2.48014 | 1.20120 |
| Н | -0.22985 | -3.63535 | -2.41227 |
| Н | 4.00043 | 0.53290 | 3.27890 |
| Н | 3.06298 | -0.90128 | 3.76729 |
| Н | 4.08415 | -0.97279 | 2.31350 |
| Н | -3.10586 | 0.90908 | 3.37980 |
| Н | -3.16166 | -0.87010 | 3.24756 |
| Н | -1.59158 | -0.03721 | 3.24560 |
| Н | 3.01099 | 2.17534 | -2.90747 |
| Н | 2.41780 | 3.15834 | -1.54669 |
| Н | 4.09364 | 2.53105 | -1.54046 |
| Н | -4.67840 | 0.50615 | -0.27281 |
| Н | -5.01038 | -0.60495 | 1.07105 |
| Н | -4.90621 | 1.14743 | 1.38362 |
| Н | 1.20089 | -2.62339 | -2.11172 |

$[Cu_3(\mu_3-S)(K_2-SCH_2)(L^{Me})_2]^+$ (6b)



Cu3S2C9P4N2H28 E(B1) = -3222.441839 Hartrees E(ZPE) = 0.371907 Hartrees E(B2) = -7552.135693 Hartrees

Cu 0.45481 -0.31002 -1.36934

| Cu | -1.50036 | -1.46339 | 0.34981 |
|--------|----------------------------------|----------|--------------------|
| Cu | 1.03039 | -1.33300 | 1.06829 |
| Р | -1.19102 | 1.50065 | -0.71236 |
| Р | -2.77738 | 0.03809 | 1.33125 |
| Р | 2.51509 | 0.67148 | -0.81127 |
| Р | 2.10532 | 0.26944 | 2.13152 |
| S | 0.53178 | 0.40772 | -3.58426 |
| S | 0.16943 | -2.50093 | -0.53932 |
| Ν | -2.05143 | 1.50137 | 0.80776 |
| Ν | 2.71940 | 1.19743 | 0.82858 |
| Н | 3.49985 | 1.82855 | 1.02212 |
| С | -2.52266 | 1.75511 | -1.95598 |
| С | 3.57104 | -0.09857 | 3.15412 |
| С | -2.69780 | 0.13858 | 3.15136 |
| С | -0.31178 | -0.99718 | -3.22105 |
| С | 3.08721 | 2.16779 | -1.69323 |
| С | -0.46449 | 3.18295 | -0.65382 |
| С | -4.56887 | 0.21399 | 1.01306 |
| С | 3.89198 | -0.48933 | -1.13788 |
| С | 1.17781 | 1.47483 | 3.14251 |
| Н | 0.28184 | 1.79790 | 2.59366 |
| Н | 0.86655 | 1.01186 | 4.08807 |
| Н | 1.79855 | 2.35239 | 3.36873 |
| Н | 4.86634 | -0.04583 | -0.89124 |
| H | 3.74868 | -1.41057 | -0.55797 |
| H | 3.87784 | -0.76128 | -2.20263 |
| H | -1.22074 | 3.95209 | -0.44128 |
| H | 0.32489 | 3.23338 | 0.10763 |
| H | -0.02561 | 3.40171 | -1.63654 |
| H | -2.07513 | 1.92847 | -2.94374 |
| H | -3.14667 | 0.85352 | -2.02141 |
| H | -3.16008 | 2.61140 | -1.693/1 |
| H H | -2.54088 | 2.36/42 | 1.05228 |
| П U | -1.39/03 | -0.98/30 | -5.0/991 |
| н Ц | 4.08903 | 0.62419 | J.44937 1 06185 |
| н Н | <i>J.27192</i> <i>A</i> 26190 | -0.03030 | 2 58833 |
| H | -3 12149 | 1 08543 | 3 51312 |
| Н | -3 26242 | -0.69176 | 3 59322 |
| Н | -1 65519 | 0.06044 | 3 48015 |
| Н | 3 16069 | 1 94127 | -2 76414 |
| Н | 2 37975 | 2 99382 | -1 55846 |
| Н | 4.07974 | 2.48065 | -1.33970 |
| Н | -4.76301 | 0.27111 | -0.06401 |
| Н | -5.10397 | -0.65684 | 1.41210 |
| Н | -4.96344 | 1.11794 | 1.49734 |
| Н | 0.12735 | -1.97964 | -3.40331 |

 $[Cu_{3}(\mu_{3}-S)(L^{Me})_{2}]^{+}(7b)$



Cu3SC8P4N2H26 E(B1) = -2785.050371 Hartrees E(ZPE) = 0.342803 Hartrees E(B2) = -7114.700105 Hartrees

| Cu | 0.00000 | 0.80140 | -0.55963 |
|----|----------|----------|----------|
| Cu | -1.45286 | -1.37442 | -0.68788 |
| Cu | 1.45286 | -1.37443 | -0.68787 |
| Р | -2.00826 | 1.69487 | 0.13668 |
| Р | -3.08072 | -1.06983 | 0.76131 |
| Р | 2.00827 | 1.69486 | 0.13667 |
| Р | 3.08072 | -1.06983 | 0.76132 |
| S | 0.00000 | -0.89212 | -2.19531 |
| Ν | -2.99110 | 0.61652 | 1.06716 |
| Ν | 2.99110 | 0.61652 | 1.06716 |
| Н | 3.79240 | 1.03962 | 1.54145 |
| С | -3.06482 | 2.19160 | -1.27645 |
| С | 4.79123 | -1.33840 | 0.17427 |
| С | -3.11667 | -1.80831 | 2.42997 |
| С | 2.08555 | 3.18282 | 1.20086 |
| С | -2.08554 | 3.18282 | 1.20088 |
| С | -4.79124 | -1.33839 | 0.17426 |
| С | 3.06483 | 2.19158 | -1.27646 |
| С | 3.11666 | -1.80831 | 2.42998 |
| Н | 2.16028 | -1.63644 | 2.93508 |
| Н | 3.28606 | -2.88948 | 2.35370 |
| Н | 3.92368 | -1.36726 | 3.03067 |
| Н | 4.05965 | 2.51530 | -0.94032 |
| Н | 3.16840 | 1.34443 | -1.96728 |
| Н | 2.58962 | 3.01384 | -1.82777 |
| Η | -3.12286 | 3.48525 | 1.40333 |
| Η | -1.57425 | 2.99442 | 2.15177 |
| Η | -1.58226 | 4.01422 | 0.69120 |
| Η | -2.58961 | 3.01385 | -1.82775 |
| Н | -3.16839 | 1.34445 | -1.96727 |
| Η | -4.05964 | 2.51532 | -0.94031 |
| Η | -3.79240 | 1.03962 | 1.54145 |
| Η | 5.52180 | -0.95035 | 0.89754 |
| Н | 4.97490 | -2.41078 | 0.03341 |
| Н | 4.93527 | -0.83537 | -0.78934 |
| Н | -3.92369 | -1.36726 | 3.03066 |

| Η | -3.28607 | -2.88948 | 2.35368 |
|---|----------|----------|----------|
| Н | -2.16029 | -1.63645 | 2.93507 |
| Н | 1.58227 | 4.01423 | 0.69118 |
| Н | 1.57426 | 2.99443 | 2.15175 |
| Н | 3.12287 | 3.48525 | 1.40332 |
| Н | -4.93527 | -0.83536 | -0.78934 |
| Н | -4.97491 | -2.41077 | 0.03340 |
| Η | -5.52180 | -0.95034 | 0.89753 |

$[Cu_3(\mu_2-H)(\mu_2,\mu_1-S_2CH)(L^{Me})_1]^+(8b)$



Cu3S2C5P2N1H15 E(B1) = -2325.015122 Hartrees E(ZPE) = 0.197702 Hartrees E(B2) = -6654.559511 Hartrees

| Cu | -0.28934 | 0.55978 | 0.32921 |
|----|----------|----------|----------|
| Cu | 0.79474 | -1.04980 | -1.07498 |
| Cu | -1.51496 | -1.72831 | -0.26371 |
| Р | -2.08044 | 1.91438 | 0.15246 |
| Р | -3.49874 | -0.75016 | -0.20220 |
| S | 0.41794 | -0.71070 | 2.15351 |
| S | 0.35006 | -2.98470 | 0.05382 |
| Ν | -3.39872 | 0.94261 | -0.38702 |
| Η | -4.28467 | 1.42521 | -0.55562 |
| С | -4.34327 | -1.05006 | 1.38861 |
| С | 0.68925 | -2.24822 | 1.59787 |
| С | -2.00722 | 3.21627 | -1.11944 |
| С | -2.74761 | 2.80595 | 1.60214 |
| С | -4.74113 | -1.21777 | -1.44555 |
| Η | -4.36778 | -1.01081 | -2.45391 |
| Η | -4.95740 | -2.28960 | -1.36237 |
| Η | -5.67605 | -0.66396 | -1.28306 |
| Η | -3.66769 | 3.34905 | 1.34635 |
| Η | -2.96006 | 2.09752 | 2.41195 |
| Η | -2.00765 | 3.52960 | 1.96699 |
| Η | 1.14816 | -2.94539 | 2.30822 |
| Η | -5.29510 | -0.50389 | 1.43708 |
| Η | -4.54118 | -2.12223 | 1.51587 |
| Η | -3.69782 | -0.72066 | 2.21338 |
| Н | -1.27617 | 3.97761 | -0.81991 |

| Η | -1.68485 | 2.78293 | -2.07265 |
|---|----------|---------|----------|
| Η | -2.98396 | 3.70214 | -1.24911 |
| Н | 0.65326 | 0.53182 | -1.16604 |

 $[Cu_3(\mu_2-H)(\mu_2,K_2-S_2CH)(L^{Me})_2]^+(TS4b-5b)$



Cu3S2C9P4N2H28

E(B1) = -3222.379869 Hartrees E(ZPE) = 0.365197 Hartrees E(B2) = -7552.074888 Hartrees Imaginary frequency: -576.75

| Cu | 0.30186 | 0.77023 | -0.84838 |
|----|----------|----------|----------|
| Cu | -1.53306 | -0.96491 | -0.16567 |
| Cu | 0.99679 | -1.54839 | 0.59622 |
| Р | -1.16894 | 2.42013 | -0.31414 |
| Р | -2.47850 | 0.30814 | 1.31762 |
| Р | 2.52942 | 0.76777 | -1.27925 |
| Р | 2.56513 | -0.28980 | 1.51167 |
| S | -2.67321 | -2.10095 | -1.89973 |
| S | -0.24209 | -3.07842 | -0.33052 |
| Ν | -2.02211 | 1.95179 | 1.10279 |
| Ν | 3.32498 | 0.55248 | 0.22836 |
| Н | 4.30695 | 0.81505 | 0.31821 |
| С | -2.43921 | 2.69042 | -1.60410 |
| С | 3.95406 | -1.15303 | 2.31773 |
| С | -2.14977 | -0.02012 | 3.07963 |
| С | -0.93984 | -2.08488 | -1.73867 |
| С | 3.32090 | 2.29120 | -1.90464 |
| С | -0.66665 | 4.12417 | 0.11922 |
| С | -4.29889 | 0.30489 | 1.21688 |
| С | 3.19438 | -0.52054 | -2.39080 |
| С | 2.11332 | 0.99235 | 2.73443 |
| Н | 1.33187 | 1.63817 | 2.31216 |
| Η | 1.72021 | 0.51851 | 3.64368 |
| Η | 2.98383 | 1.60639 | 3.00099 |
| Η | 4.29092 | -0.48292 | -2.43800 |
| Н | 2.87901 | -1.50902 | -2.03151 |
| Н | 2.78911 | -0.37970 | -3.40170 |
| | | | |

| Η | -1.53201 | 4.74526 | 0.38922 |
|---|----------|----------|----------|
| Н | 0.04095 | 4.10981 | 0.95627 |
| Н | -0.17377 | 4.58388 | -0.74652 |
| Н | -1.98212 | 3.13347 | -2.49866 |
| Н | -2.87594 | 1.72414 | -1.89144 |
| Н | -3.23365 | 3.35862 | -1.24392 |
| Н | -2.61370 | 2.66061 | 1.54173 |
| Н | -0.32010 | -2.12582 | -2.63964 |
| Η | 4.74434 | -0.44460 | 2.60264 |
| Η | 3.59662 | -1.65636 | 3.22409 |
| Η | 4.36747 | -1.91037 | 1.64271 |
| Η | -2.67533 | 0.69746 | 3.72377 |
| Η | -2.48538 | -1.03503 | 3.32712 |
| Η | -1.07130 | 0.04530 | 3.26668 |
| Η | 2.96868 | 2.49445 | -2.92335 |
| Н | 3.05486 | 3.14127 | -1.26560 |
| Η | 4.41469 | 2.18840 | -1.93532 |
| Η | -4.60904 | 0.58786 | 0.20359 |
| Η | -4.66533 | -0.71183 | 1.40737 |
| Η | -4.74765 | 0.99028 | 1.94911 |
| Н | -0.41835 | -0.73454 | -1.52708 |

$[Cu_3(\mu_2-S)(K_2-SCH_2)(L^{Me})_2]^+$ (TS5b-6b)



Cu3S2C9P4N2H28 E(B1) = -3222.429057 Hartrees E(ZPE) = 0.371263 Hartrees E(B2) = -7552.121792 Hartrees Imaginary frequency: -170.93

| Cu | 0.26038 | 0.26279 | -0.90734 |
|----|----------|----------|----------|
| Cu | -1.60045 | -1.04065 | 0.47429 |
| Cu | 1.05622 | -1.32837 | 1.00434 |
| Р | -1.51394 | 1.71061 | -0.88197 |
| Р | -2.97766 | 0.45051 | 1.34519 |
| Р | 2.44027 | 0.90614 | -0.56470 |
| Р | 2.45267 | -0.10660 | 2.21181 |
| S | 0.32950 | -0.20494 | -3.22689 |
| S | -0.18534 | -2.41136 | -0.39644 |

| Ν | -2.48799 | 1.88099 | 0.54464 |
|---|----------|----------|----------|
| Ν | 3.01970 | 1.02581 | 1.06410 |
| Н | 3.90462 | 1.50511 | 1.24150 |
| С | -2.72087 | 1.45320 | -2.23395 |
| С | 3.96115 | -0.86363 | 2.90653 |
| С | -2.89994 | 0.86807 | 3.11866 |
| С | 0.51481 | -1.78600 | -2.61356 |
| С | 2.83576 | 2.56450 | -1.22246 |
| С | -1.00889 | 3.44608 | -1.15826 |
| С | -4.77478 | 0.30089 | 1.05391 |
| С | 3.68224 | -0.13317 | -1.42007 |
| С | 1.82331 | 0.90968 | 3.59081 |
| Н | 0.98239 | 1.51458 | 3.23042 |
| Н | 1.47695 | 0.26852 | 4.41107 |
| Н | 2.60594 | 1.58077 | 3.96892 |
| Н | 4.69348 | 0.27951 | -1.30229 |
| Н | 3.66056 | -1.15100 | -1.00736 |
| Н | 3.43516 | -0.18463 | -2.48915 |
| Н | -1.87597 | 4.11090 | -1.27724 |
| Н | -0.39689 | 3.79703 | -0.31898 |
| Н | -0.40935 | 3.49215 | -2.07731 |
| Н | -2.21760 | 1.55047 | -3.20411 |
| Н | -3.12624 | 0.43425 | -2.18019 |
| Н | -3.54074 | 2.18221 | -2.16915 |
| Н | -3.10672 | 2.69452 | 0.59478 |
| Н | -0.16029 | -2.56093 | -2.97762 |
| Н | 4.63796 | -0.09673 | 3.30784 |
| Н | 3.69455 | -1.55103 | 3.71856 |
| Н | 4.48284 | -1.43349 | 2.12855 |
| Η | -3.50680 | 1.75691 | 3.33725 |
| Н | -3.27798 | 0.02830 | 3.71465 |
| Н | -1.86188 | 1.06440 | 3.40800 |
| Η | 2.56956 | 2.58453 | -2.28763 |
| Η | 2.25587 | 3.33214 | -0.69739 |
| Η | 3.90627 | 2.79437 | -1.12871 |
| Н | -4.97353 | 0.13211 | -0.01070 |
| Н | -5.17282 | -0.55303 | 1.61616 |
| Η | -5.30124 | 1.20867 | 1.37948 |
| Н | 1 51006 | -2 14776 | -2.35050 |

dmpa (L)



C4P2N1H13 E(B1) = -897.3386603 Hartrees

E(ZPE) = 0.166819 Hartrees E(B2) = -897.4793544 Hartrees

| Р | -1.51423 | 0.04189 | -0.56603 |
|---|----------|----------|----------|
| Р | 1.51423 | 0.04189 | -0.56603 |
| Ν | -0.00000 | 0.26967 | 0.23448 |
| С | 2.15873 | -1.46374 | 0.31156 |
| С | -2.15873 | -1.46374 | 0.31156 |
| С | 2.52577 | 1.29326 | 0.35173 |
| С | -2.52577 | 1.29326 | 0.35173 |
| Н | -2.21865 | 2.30455 | 0.06121 |
| Η | -3.58785 | 1.17068 | 0.10155 |
| Η | -2.41293 | 1.19017 | 1.44188 |
| Н | -0.00000 | 0.26137 | 1.25886 |
| Н | 3.20575 | -1.64766 | 0.03495 |
| Η | 2.10125 | -1.35554 | 1.40557 |
| Η | 1.57143 | -2.34220 | 0.01546 |
| Η | 2.21865 | 2.30455 | 0.06121 |
| Η | 2.41293 | 1.19017 | 1.44188 |
| Н | 3.58785 | 1.17068 | 0.10155 |
| Η | -3.20575 | -1.64766 | 0.03495 |
| Н | -1.57143 | -2.34220 | 0.01546 |
| Н | -2.10125 | -1.35554 | 1.40557 |

CH_2S



SCH2 E(B1) = -437.3707119 Hartrees E(ZPE) = 0.024638 Hartrees E(B2) = -437.4122528 Hartrees

| S | 0.00000 | 0.00000 | 0.58684 |
|---|----------|----------|----------|
| С | -0.00000 | -0.00000 | -1.02655 |
| Η | 0.00000 | 0.92435 | -1.61508 |
| Н | -0.00000 | -0.92435 | -1.61508 |