

## Supporting Information

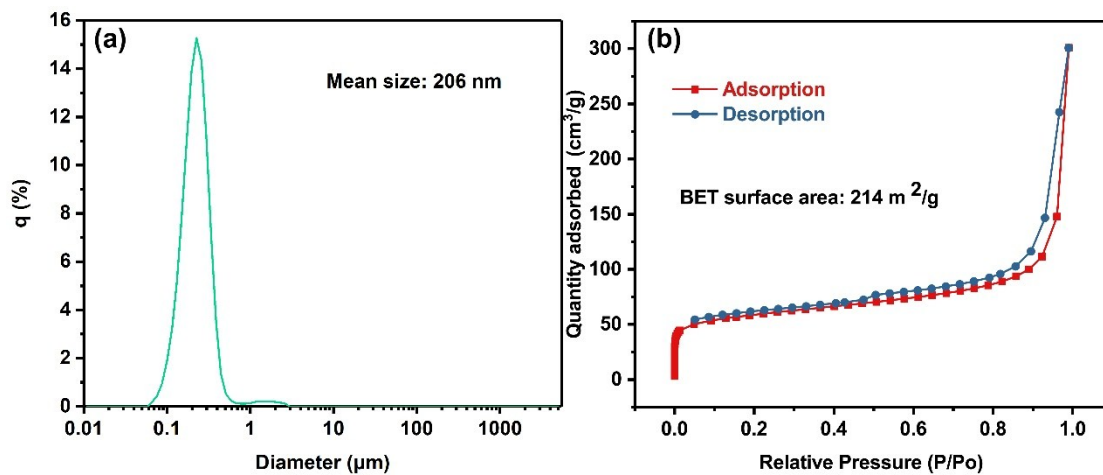
### **Green synthesis and evaluation of iron-based metal–organic framework MIL-88B for the efficient decontamination of arsenate from water**

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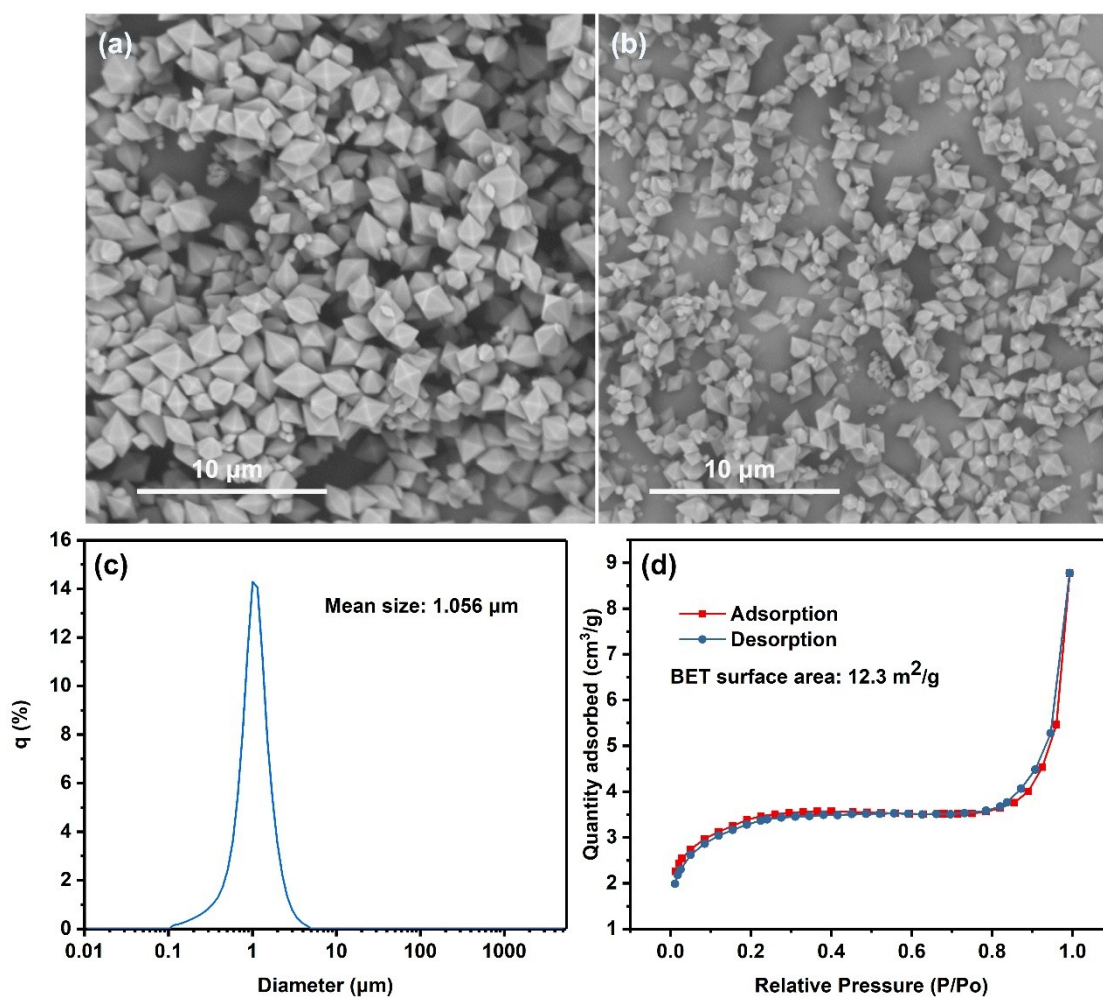
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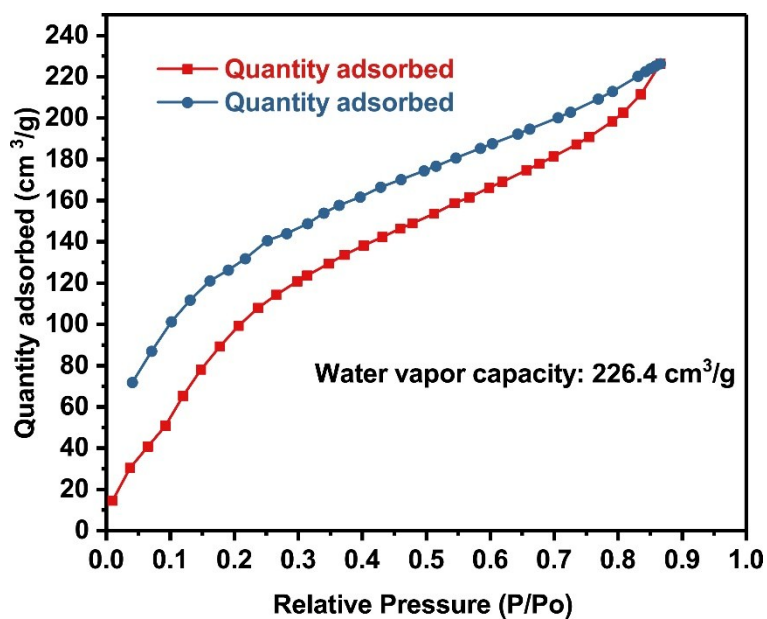
<sup>b</sup> Univ Lyon, Univ Lyon 1, INSA-Lyon, CNRS, MATEIS (UMR 5510), F-69621, Lyon, France.



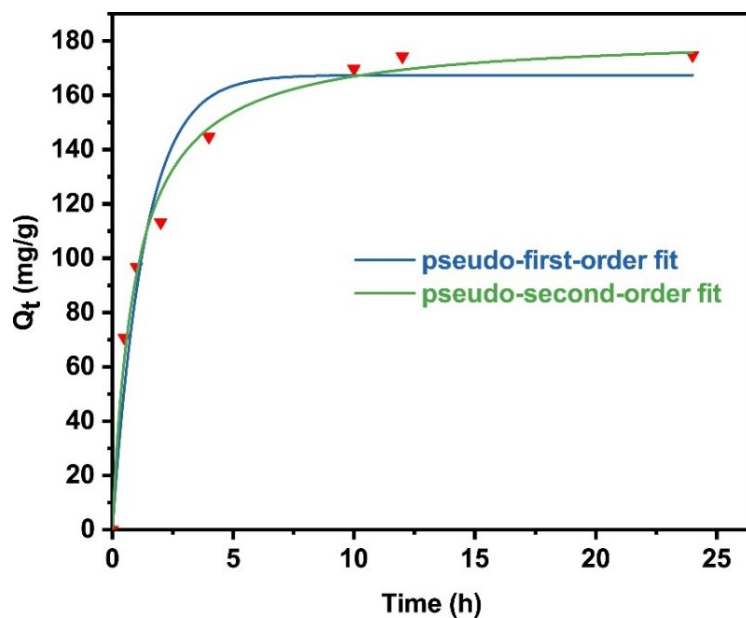
**Figure S1.** Size distribution in EtOH (a) and  $\text{N}_2$  adsorption-desorption isotherm (b) of the prepared MIL-88B(Fe)-E



**Figure S2.** SEM images (a) and (b), the size distribution (c) and  $\text{N}_2$  adsorption-desorption isotherm of MIL-88B(Fe)-D synthesized by conventional method.



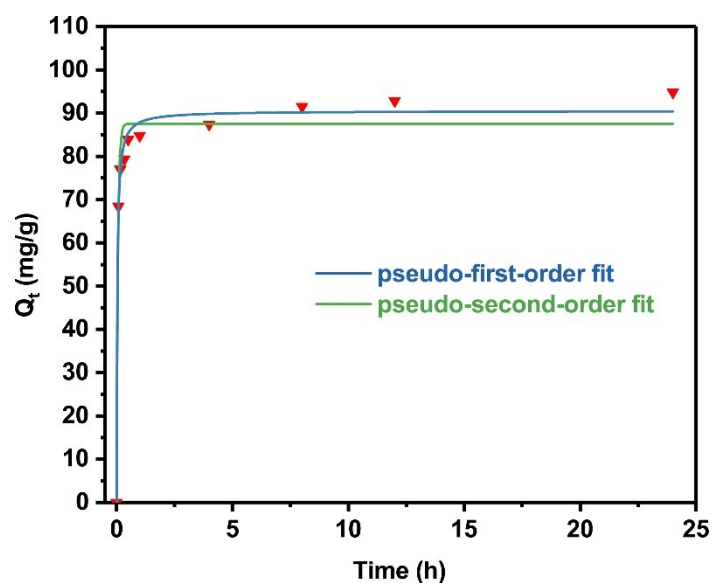
**Figure S3.** Water vapor adsorption isotherms of the MIL-88B(Fe)-E at 298 K.



**Figure S4.** Adsorption kinetics of As(V) on MIL-88B(Fe)-E fitted with the pseudo-first-order and pseudo-second-order model.

**Table S1.** Adsorption kinetics parameters of As(V) onto MIL-88B(Fe)-E.

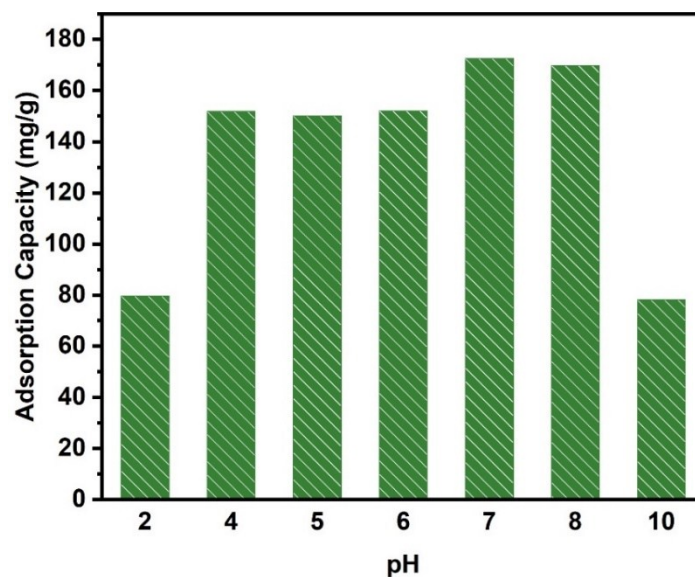
Adsorbent	Pseudo first order			Pseudo second order		
	$k_1$	$q_e$	$R^2$	$k_2$	$q_e$	$R^2$
MIL-88B(Fe)-E	0.750	167.46	0.962	0.006	182.72	0.992



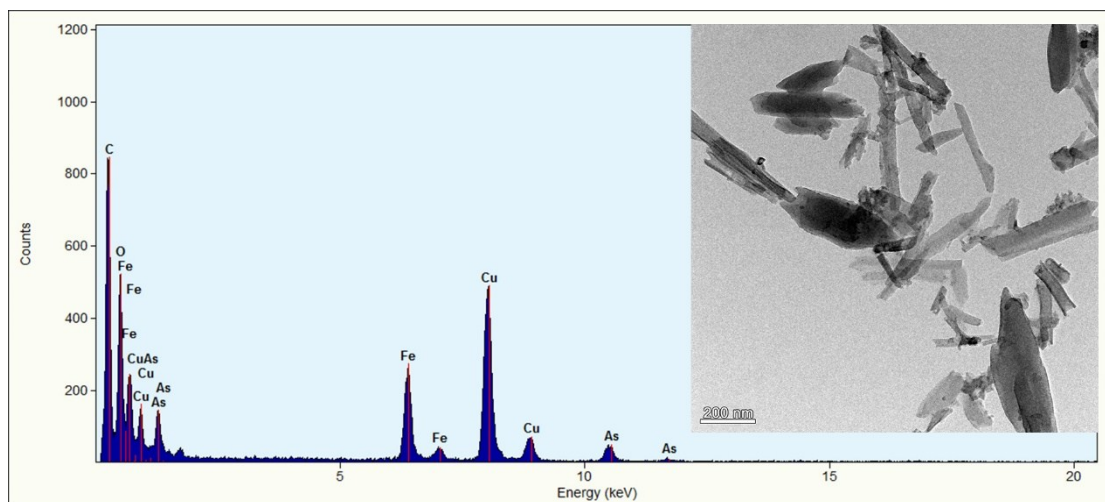
**Figure S5.** Adsorption kinetics of As(V) on MIL-88B(Fe)-D fitted with the pseudo-first-order and pseudo-second-order model.

**Table S2.** Adsorption kinetics parameters of As(V) onto MIL-88B(Fe)-D.

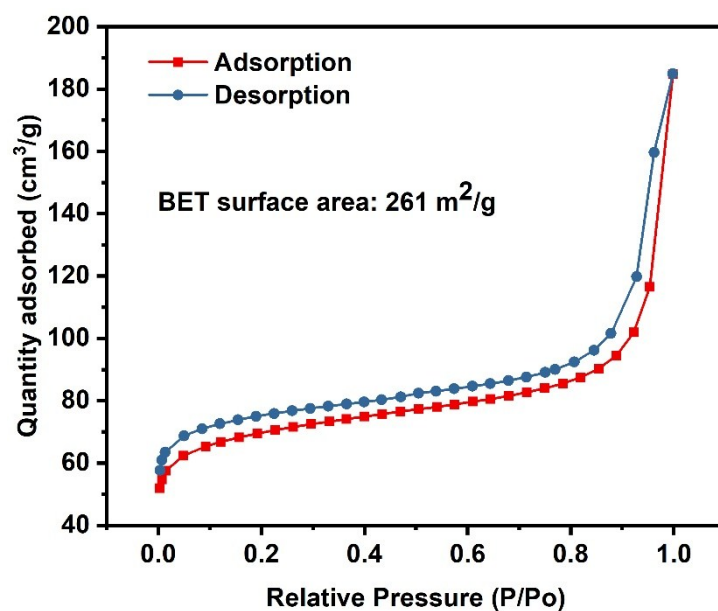
Adsorbent	Pseudo first order			Pseudo second order		
	$k_1$	$q_e$	$R^2$	$k_2$	$q_e$	$R^2$
MIL-88B(Fe)-E	16.54	87.51	0.969	0.366	90.50	0.991



**Figure S6.** the pH influence on adsorption behavior of As(V) on MIL-88B(Fe)-E.



**Figure S7.** TEM image and corresponding EDS spectra of MIL-88B(Fe)-E after the adsorption of As(V).



**Figure S8.** N<sub>2</sub> adsorption-desorption isotherm of MIL-88B(Fe)-E synthesized from 1 L autoclave at room temperature.