Bodipy Dyes with Thienylthia-Substituents – Synthesis, Redox and Fluorescent Properties

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1.NMR specta of the studied compounds



1,9-di(thien-2-ylthio)-5-phenyl-dipyrromethene 14



1,9-di(thien-2-ylthio)-5-(4-nitrophenyl)-dipyrromethene 15





AB16 IU23(7)C\











3,5-di(2,2'-dithien-5-ylthio)-8-phenyl-BODIPY 7







IU31(2)1

2. UV-Vis spectra



Fig. S1. Normalized absorption spectra of **5-8**, toluene.





Fig S2. Sample of **5** crystals at (a) ambient and (b) UV ($\lambda = 254$ nm) light.



Fig. S3. Solid luminescence measurements of crystals of **5**. a) Emission, excited at 390 nm, filter down to 450 nm; b) emission, excited at 670 nm, non-filtered; c) Fluorescence excitation, registration at 750 nm, filtered down to 450 nm.

4. CVA measurements



Fig. S4 CVA curves of the compounds **5-8**. CVA was done on glassy carbon electrode for 10^{-3} M solutions in DMF, 0.1M of NBu₄CLO₄ was used as background, potentials were reported vs. Ag/AgCl



Fig S5. Current increase in CVA experiments of 7 on the repetitive cycling a) glass carbon electrode b) Au electrode





Fig. S6 – MO energies of 5 and 7, according to B3LYP/6-31+g(d) calculations, corrected to CVA experiments conditions ($E_{DMF, eV} = -(4.73 + E_{vac, Hartree} \times 27.1)$, referenced vs Fe/Fe⁺ in DMF.



Fig. S7 – MO energies of **6** and **8**, according to B3LYP/6-31+g(d) calculations, corrected to CVA experiments conditions ($E_{DMF, eV} = -(4.73 + E_{vac, Hartree} \times 27.1)$, referenced *vs* Fe/Fe⁺ in DMF.

SCE vs vacuum 4.71 NHE vs vacuum 4.6 V