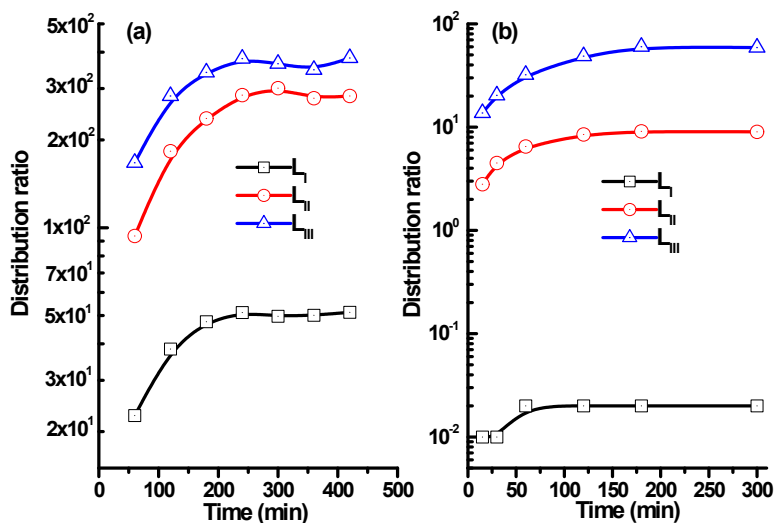


## **Unusual extraction of trivalent *f*-cations using diglycolamide dendrimers in a room temperature ionic liquid: extraction, spectroscopic and DFT studies**

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### **Electronic Supporting Information**

## 1. Kinetics of extraction studies



**Fig. S1** Kinetics for the extraction of Am<sup>3+</sup> at (a) 0.1 M HNO<sub>3</sub> and (b) 3 M HNO<sub>3</sub>.  
[L]: 0.1 mmol/L / [C<sub>4</sub>mim][Tf<sub>2</sub>N].

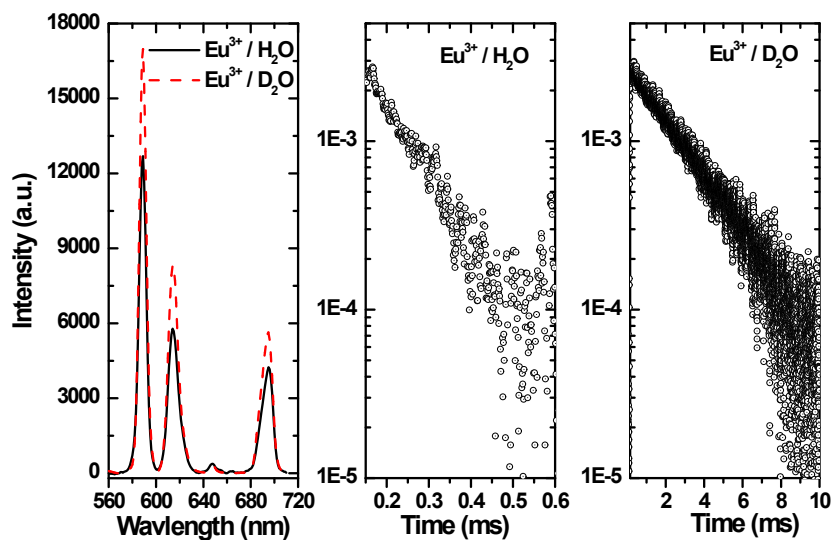
**Table S1** Extraction data and separation factor for Am and Eu by DGA-Dens L<sub>I-III</sub>.  
Ligand: 0.1 mmol/L in [C<sub>4</sub>mim][Tf<sub>2</sub>N]

[HNO <sub>3</sub> ], M	L <sub>I</sub>			L <sub>II</sub>			L <sub>III</sub>		
	<i>D</i> <sub>Am</sub>	<i>D</i> <sub>Eu</sub>	<i>D</i> <sub>Eu</sub> / <i>D</i> <sub>Am</sub>	<i>D</i> <sub>Am</sub>	<i>D</i> <sub>Eu</sub>	<i>D</i> <sub>Eu</sub> / <i>D</i> <sub>Am</sub>	<i>D</i> <sub>Am</sub>	<i>D</i> <sub>Eu</sub>	<i>D</i> <sub>Eu</sub> / <i>D</i> <sub>Am</sub>
0.1	52.2	110	2.11	261	455.5	1.75	344	580	1.69
1	1.58	5.62	3.56	41.7	96.5	2.31	92.4	210	2.27
3	0.02	0.11	5.50	9.48	23.35	2.46	52.0	98.7	1.90
4	0.01	0.04	4.00	5.43	14.3	2.63	41.1	73.8	1.80
6	0.01	0.03	3.00	2.83	5.13	1.81	22.3	44.3	1.99

**Table S2**  $SF$  values obtained for the extraction of  $\text{Am}^{3+}$  with respect to  $\text{UO}_2^{2+}$ ,  $\text{Cs}^+$  and  $\text{Sr}^{2+}$  ions using  $1.0 \times 10^{-4}$  M DGA-Dens  $\text{L}_{\text{I-III}}$  in  $[\text{C}_4\text{mim}][\text{Tf}_2\text{N}]$ . Aqueous phase: 3 M  $\text{HNO}_3$

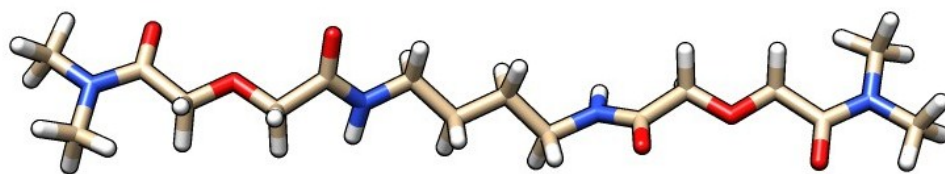
Ligand	$SF (D_{\text{Am}}/D_{\text{U}})$	$SF (D_{\text{Am}}/D_{\text{Cs}})$	$SF (D_{\text{Am}}/D_{\text{Sr}})$
$\text{L}_{\text{I}}$	1	190	1733
$\text{L}_{\text{II}}$	2	474	5200
$\text{L}_{\text{III}}$	0.2	95	520

## 2. Luminescence spectroscopy of $\text{Eu}^{3+}$ ion

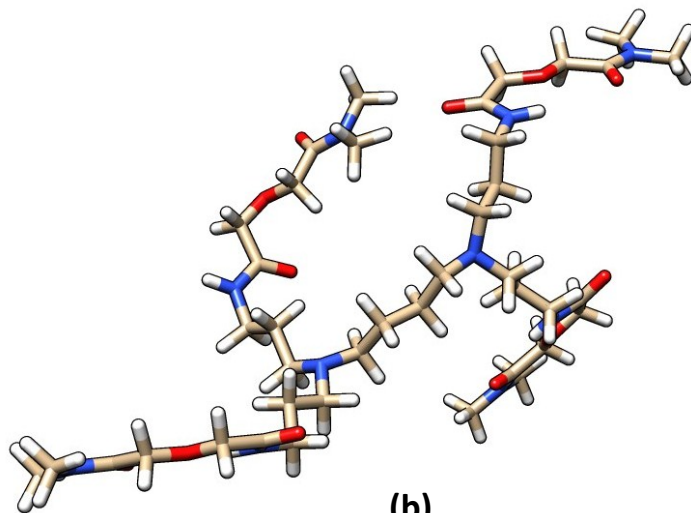


**Fig. S2** Emission spectra and fluorescence lifetime of  $\text{Eu}^{3+}$  in  $\text{H}_2\text{O}$  and  $\text{D}_2\text{O}$ .  
Excitation: 394 nm; Emission: 614 nm.

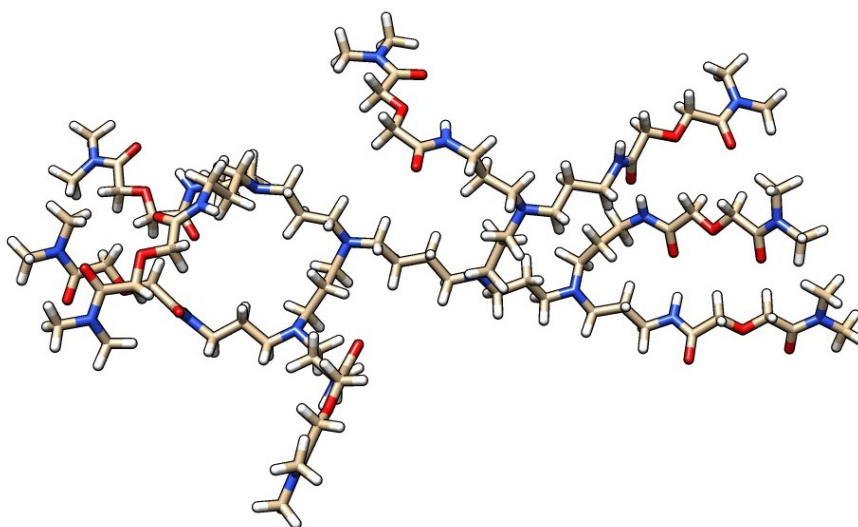
### 3. DFT Studies



(a)



(b)



(c)

**Fig. S3** Optimized structures of (a)  $L_I$ , (b)  $L_{II}$  and (c)  $L_{III}$  at the B3LYP/SVP level of theory.

**Table S3** Calculated charge and orbital population using NBO analysis in ionic liquid phase at the B3LYP/TZVP level of theory

System	charge	<i>s</i>	<i>p</i>	<i>d</i>	<i>f</i>
(L <sub>I</sub> ) <sub>2</sub> -Am <sup>3+</sup> ,a	2.045	4.14	11.98	10.69	6.13
(L <sub>I</sub> ) <sub>2</sub> -Am <sup>3+</sup> ,b	2.343	4.10	11.98	10.47	6.09
L <sub>II</sub> -[Am <sup>3+</sup> ] <sub>2</sub>	2.289	4.10	11.98	10.51	6.11
	2.298	4.09	11.99	10.49	6.11
L <sub>III</sub> -[Am <sup>3+</sup> ] <sub>4</sub>	2.357	4.09	11.98	10.44	6.11
	2.314	4.09	11.98	10.48	6.12
	2.336	4.09	11.98	10.47	6.10
	2.358	4.09	11.98	10.45	6.11

Note: a: Structure given in Fig. 7a; b: Structure given in Fig. 7b

**Table S4** Calculated quantum chemical descriptors in the gas phase at the B3LYP/TZVP level of theory

System	$\Delta E_{\text{LUMO-HOMO}}$ (eV)	$\eta$ (eV)	$\chi$ (eV)	$\Delta N$
Am <sup>3+</sup> - (H <sub>2</sub> O) <sub>9</sub>	3.79	1.89	16.74	
L <sub>I</sub>	6.76	3.38	3.26	1.27
L <sub>II</sub>	5.60	2.80	2.62	1.50
L <sub>III</sub>	4.64	2.32	3.03	1.62