

Electronic Supplementary Material (ESI) for Dalton Trans  
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## Supporting Information for:

# Hydrogen bonds assisted homochiral lattice packing between inorganic helices built from heterometallic units

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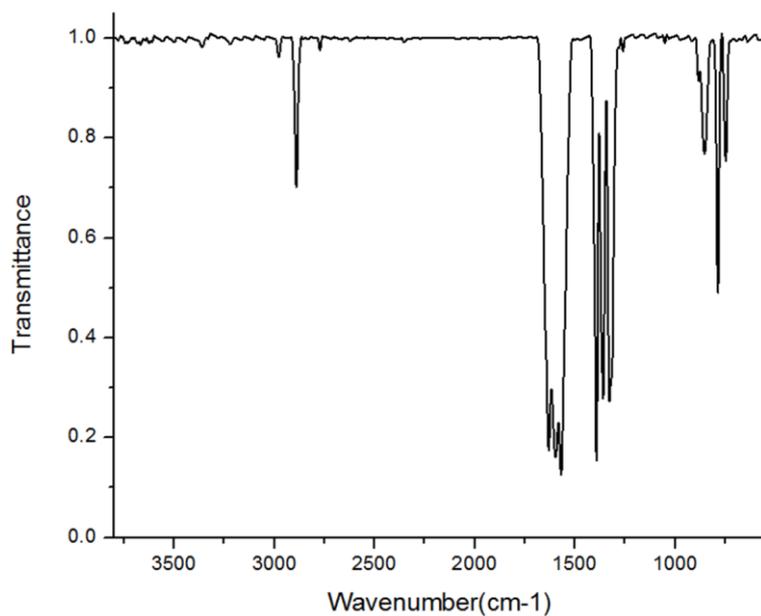
## Experimental Section

**Materials and Instrumentation.** We collected the fourier transform infrared spectroscopy (FTIR) data on a Perkin-Elmer Spectrum 100 FT-IR Spectrometer. Thermogravimetric analyses (TGA) was performed on a Mettler Toledo TGA/SDTA 851e analyzer in N<sub>2</sub> with a heating rate of 10 °C min<sup>-1</sup> from 20 to 800 °C. Powder X-ray diffraction (PXRD) data analysis was collected on a Rigaku Mini Flex II diffractometer using CuK<sub>α</sub> radiation ( $\lambda = 1.54056 \text{ \AA}$ ) in the 2 $\theta$  range of 5–50° with a scanning rate of 5° min<sup>-1</sup>. The UV-Visible diffuse reflection data was recorded at room temperature using a powder sample with BaSO<sub>4</sub> as a standard (100% reflectance) on a Perkin-Elmer Lambda-950 UV spectrophotometer and scanned at 200-800 nm. The absorption data are calculated from the Kubelka-Munk function,  $(F(R) = (1-R)^2/2R)$ , where R representing the reflectance.

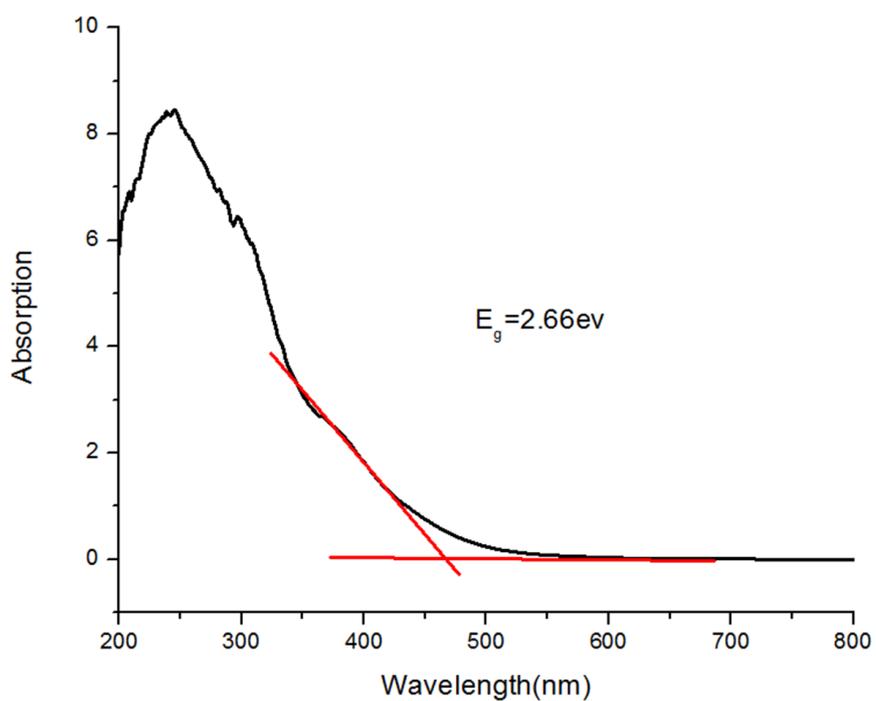
### Chemicals and Materials

All the reagents and solvents employed are purchased commercially and used as received without further treatment. Ti(O<sup>*i*</sup>Pr)<sub>4</sub> was purchased from Adamas, while isopropanol and formic acid were bought from Sino pharm Chemical Reagent Beijing.

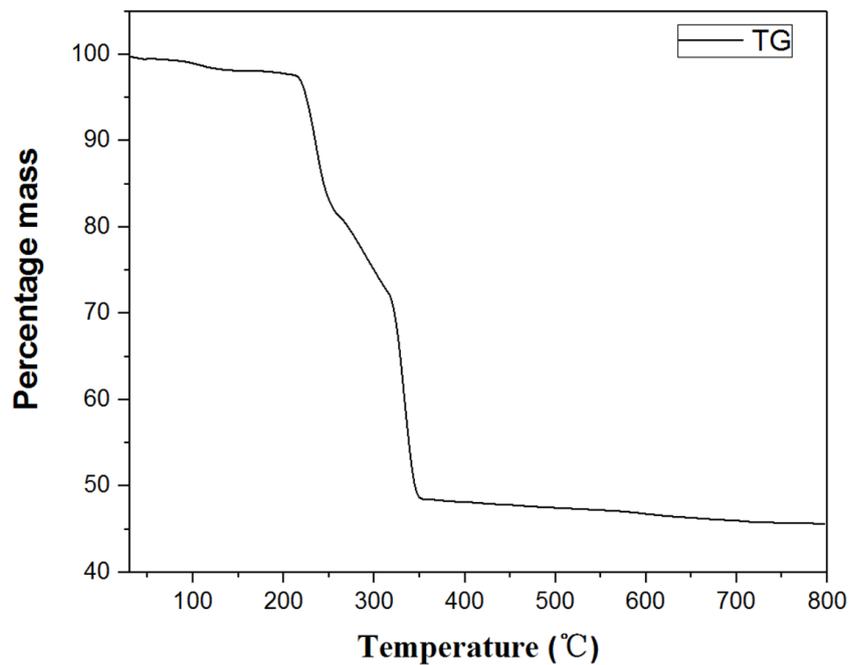
**General Methods for X-ray Crystallography.** Crystallographic data of **1P** and **1M** were collected on a Mercury single crystal diffractometer with graphite-monochromatic Mo K $\alpha$  radiation ( $\lambda = 0.71073 \text{ \AA}$ ). The structures were solved with direct methods using OLEX<sup>2</sup> and refined with the full-matrix least-squares technique based on F<sup>2</sup> using the SHELXL-2014. Non-hydrogen atoms were refined by anisotropic thermal parameters. All of the hydrogen atoms were theoretical hydrogenation and were modified using isotropic thermal parameters and cross-type models. Non-hydrogen atoms were refined anisotropically, and all hydrogen atoms bond C were generated geometrically.



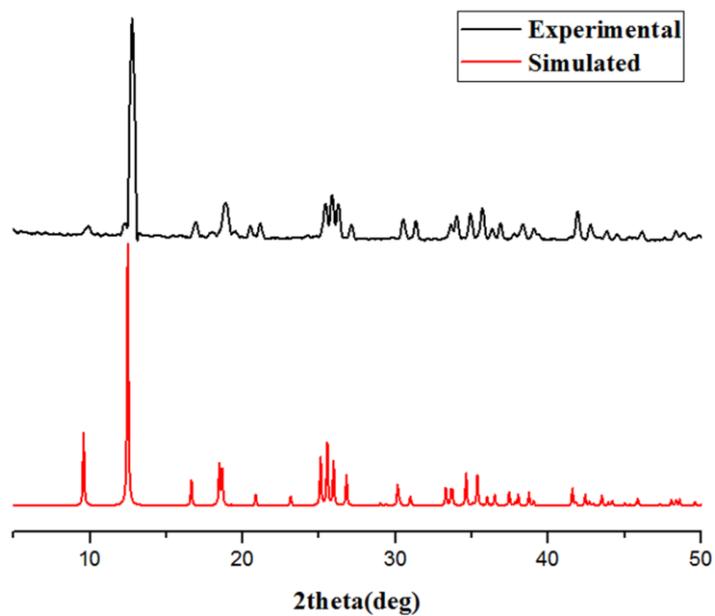
**Figure S1** Fourier transform infrared spectroscopy (FT-IR) of the bulk samples of **1P** and **1M**.



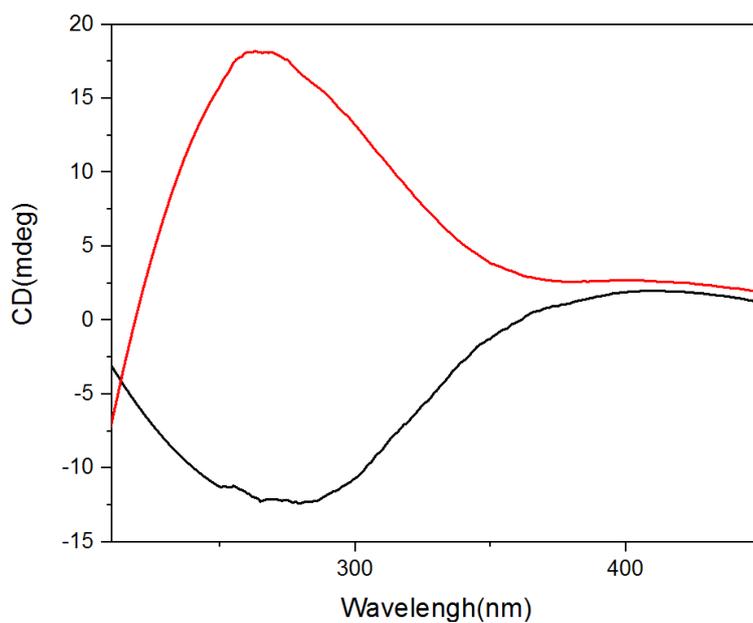
**Figure S2** Solid-state UV-vis spectra of the bulk samples of **1P** and **1M**.



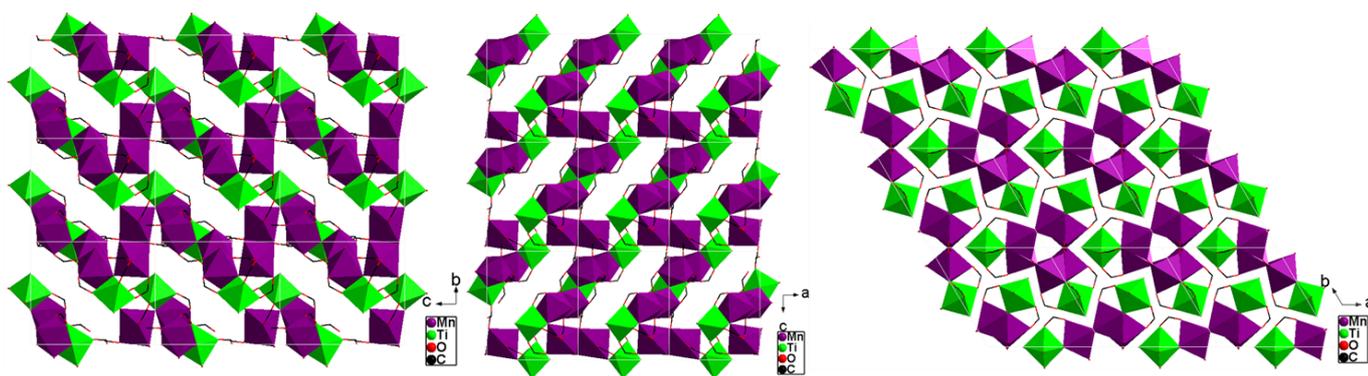
**Figure S3** TGA of the bulk samples of **1P** and **1M**.



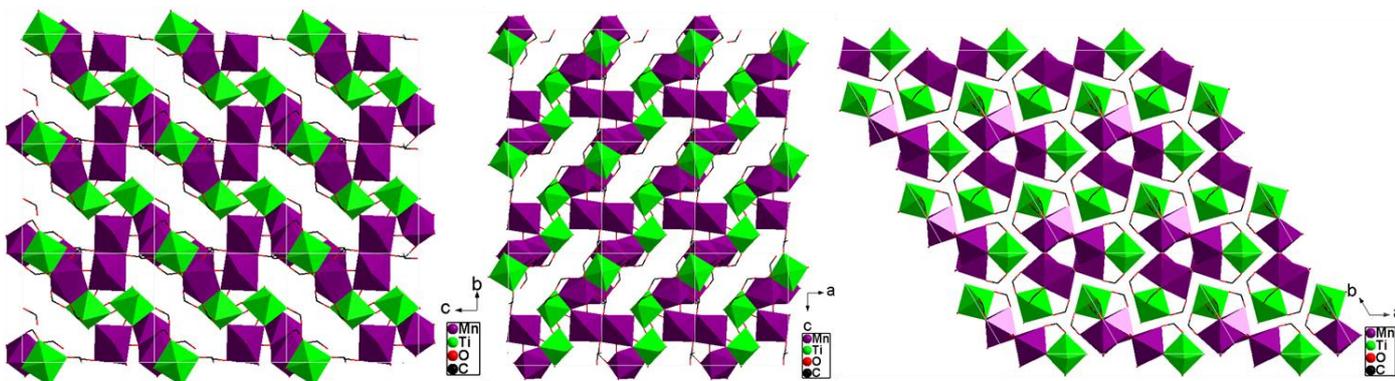
**Figure S4** Comparative powder X-ray diffraction (PXRD) patterns of the bulk samples of **1P** and **1M**.



**Figure S5** Solid-state circular dichroism (CD) patterns of the bulk samples of **1P** and **1M**.



**Figure S6** Packing view of **1M** along a, b, and c-axes.



**Figure S7** Packing view of **1P** along a, b, and c-axes.