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Stepwise synthesis of the heterotrimetallic chains [MRu₂(dpa)₄X₂]^{0/1+} using group 7 to group 12 transition metal ions and [Ru₂(dpa)₄Cl]

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(5, 6, 7, 8 and [PdRu₂(dpa)₄Cl(OC(CH₃)₂)](PF₆)₂)

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Figure S1. ¹H NMR spectrum of $[CoRu_2(dpa)_4Cl_2]$ (1) in CD_2Cl_2 under air.

Crystallographic Section:



Figure S2. The bond lengths of Hdpa ligand¹ in comparison with its coordination counterparts in complexes **2** and **4**. The C and N atoms in three uncoordinated pyridyl groups in **2** are disordered, therefore, their distances are less accurate.

Compound	1	$2 \cdot CH_2Cl_2$	$3 \cdot \mathrm{CH}_2\mathrm{Cl}_2 \cdot \mathrm{CHCl}_3$	$4 \cdot 2 CH_2 Cl_2$	5.0.5CH ₂ Cl ₂ .acetone	6 ·1.5CH ₂ Cl ₂ ·0.5Et ₂ O
Formula	C40H32ClN12Ru2	$C_{41}H_{34}Cl_4CoN_{12}Ru_2$	C42H35Cl7CoF6N12PRu	$L_2 C_{44}H_{39}AgCl_5N_{12}O_2R$	$Ru_2 = C_{43.5}H_{39}Cl_4MnN_{12}O_5Ru_2$	C _{43.5} H ₄₀ Cl ₅ F ₆ FeN ₁₂ OPRu ₂
Formula weight	918.36	1097.67	1362.01	1255.13	1208.74	1327.08
Temperature	150(2) K	150(2) K	150(2) K	150(2) K	150(2) K	150(2) K
Crystal system	Monoclinic	Orthorhombic	Monoclinic	Monoclinic	Monoclinic	Monoclinic
Space group	$P2_1/n$	Pnn2	$P2_1/n$	C2/c	$P2_1/n$	$P2_1/n$
a / Å	9.7288(4)	13.0347(3)	11.5837(8)	19.1716(7)	11.3404(3)	11.5723(4)
b / Å	18.8437(7)	14.1300(3)	21.4250(18)	16.5217(6)	21.4694(5)	21.3821(7)
c / Å	20.4662(8)	11.2290(3)	20.3563(17)	15.6164(5)	19.9024(4)	20.4312(7)
α/ °	90	90	90	90	90	90
β/ °	99.8469(12)	90	92.830(3)	108.8009(11)	93.7271(12)	92.9440(11)
$\gamma/^{o}$	90	90	90	90	90	90
V / Å ³	3696.7(3)	2068.16(9)	5045.9(7)	4682.5(3)	4835.4(2)	5048.8(3)
Ζ	4	2	4	4	4	4
Reflection collected	21765	11811	21845	19416	24577	32579
Independent reflections	8396	4298	8810	5381	11009	11585
R1, wR2 [I>2sigma(I)]	0.0481, 0.1130	0.0310, 0.0586	0.0728, 0.1489	0.0282, 0.0753	0.0535, 0.1157	0.0608, 0.1471
R1, wR2 (all data)	0.0901, 0.1294	0.0549, 0.0642	0.1891, 0.1909	0.0345, 0.0783	0.0904, 0.1265	0.0740, 0.1547
Goodness-of-fit on F ²	1.031	0.962	1.016	1.076	1.120	1.106
Compound	7·2CH ₂ Cl ₂	$_2$ 8 ·2CH ₂ Cl ₂ ·().398CH ₃ CN	9·3CH ₂ Cl ₂	10 ·2.25CH ₂ Cl ₂	$11 \cdot 2 \operatorname{CH}_2\operatorname{Cl}_2$
Compound Formula	$\frac{7 \cdot 2 C H_2 C I_2}{C_{42} H_{36} C I_7 N_{12} O_4 I_2}$	$\begin{array}{c c} & & & \\ \hline & & \\ 2 & & & \\ Ru_2Zn & & & \\ C_{42.79}H_{37.19}Cd \end{array}$	0.398CH ₃ CN Cl ₇ N _{12.4} O ₄ Ru ₂ C ₄₃ H	$\begin{array}{c} 9 \cdot 3 CH_2 Cl_2 \\ \\ 38 Cl_8 PF_6 N_{12} PdRu_2 \end{array} C$	$\frac{10 \cdot 2.25 \text{CH}_2 \text{Cl}_2}{\text{C}_{42.25} \text{H}_{36.5} \text{Cl}_{6.5} \text{F}_6 \text{N}_{12} \text{PRhRu}_2}$	$\frac{11\cdot 2 \text{ CH}_2\text{Cl}_2}{\text{C}_{42}\text{H}_3\text{c}\text{Cl}_6\text{F}_6\text{IrN}_{12}\text{PRu}_2}$
Compound Formula Formula weight	7.2CH ₂ Cl C ₄₂ H ₃₆ Cl ₇ N ₁₂ O ₄ 1288.49	$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	0.398CH ₃ CN Cl ₇ N _{12.4} O ₄ Ru ₂ C ₄₃ H 1.80	9·3CH ₂ Cl ₂ ₃₈ Cl ₈ PF ₆ N ₁₂ PdRu ₂ C 1459.96	$\begin{array}{c} \textbf{10} \cdot 2.25 \text{CH}_2 \text{Cl}_2 \\ \hline C_{42.25} \text{H}_{36.5} \text{Cl}_{6.5} \text{F}_6 \text{N}_{12} \text{PRhRu}_2 \\ \hline 1392.78 \end{array}$	$\frac{11 \cdot 2 \text{ CH}_2\text{Cl}_2}{\text{C}_{42}\text{H}_{36}\text{Cl}_6\text{F}_6\text{IrN}_{12}\text{PRu}_2}$ 1460.84
Compound Formula Formula weight Temperature	7·2CH ₂ Cl ₂ C ₄₂ H ₃₆ Cl ₇ N ₁₂ O ₄ 1288.49 150(2) K	$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	D.398CH ₃ CN Cl ₇ N _{12.4} O ₄ Ru ₂ C ₄₃ H 1.80 2) K	9 ·3CH ₂ Cl ₂ ₃₈ Cl ₈ PF ₆ N ₁₂ PdRu ₂ C 1459.96 150(2) K	10·2.25CH ₂ Cl ₂ C _{42.25} H _{36.5} Cl _{6.5} F ₆ N ₁₂ PRhRu ₂ 1392.78 150(2) K	$\begin{array}{c} 11 \cdot 2 \text{ CH}_2\text{Cl}_2 \\ \hline C_{42}\text{H}_{36}\text{Cl}_6\text{F}_6\text{IrN}_{12}\text{PRu}_2 \\ 1460.84 \\ 150(2) \text{ K} \end{array}$
Compound Formula Formula weight Temperature Crystal system	7 · 2CH ₂ Cl ₂ C ₄₂ H ₃₆ Cl ₇ N ₁₂ O ₄ 1288.49 150(2) K Monoclinic	$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	$\begin{array}{c} \hline 0.398 CH_3 CN \\ \hline Cl_7 N_{12.4} O_4 Ru_2 \\ \hline Cu_3 H \\ \hline 1.80 \\ \hline 2) K \\ \hline celinic \\ \hline \end{array}$	9·3CH ₂ Cl ₂ 38Cl ₈ PF ₆ N ₁₂ PdRu ₂ C 1459.96 150(2) K Monoclinic	$\begin{array}{c} 10 \cdot 2.25 \text{CH}_2 \text{Cl}_2 \\ \hline \\ C_{42.25} \text{H}_{36.5} \text{Cl}_{6.5} \text{F}_6 \text{N}_{12} \text{PRhRu}_2 \\ 1392.78 \\ 150(2) \text{ K} \\ \text{Monoclinic} \end{array}$	$\begin{array}{c} 11 \cdot 2 \text{ CH}_2\text{Cl}_2 \\ \hline C_{42}\text{H}_{36}\text{Cl}_6\text{F}_6\text{IrN}_{12}\text{PRu}_2 \\ 1460.84 \\ 150(2) \text{ K} \\ \text{Monoclinic} \end{array}$
Compound Formula Formula weight Temperature Crystal system Space group	7·2CH ₂ Cl ₂ C ₄₂ H ₃₆ Cl ₇ N ₁₂ O ₄ 1288.49 150(2) K Monoclinic P2 ₁ /n	$\begin{array}{c ccccc} & & & & & & & \\ \hline 2 & & & & & & \\ \hline Ru_2Zn & & & & & \\ Ru_2Zn & & & & & \\ & & & & & \\ C_{42.79}H_{37.19}Cd \\ & & & & & \\ 135 \\ & & & & 150(0) \\ c & & & & & \\ \hline c & & & & & \\ P2 \end{array}$	$\begin{array}{c} \hline 0.398 CH_{3} CN \\ \hline Cl_{7} N_{12.4} O_{4} Ru_{2} \\ \hline Cl_{3} N_{12.4} O_{4} \\ \hline Cl_{3} N_{1$	$\begin{array}{c c} 9.3 \text{CH}_2 \text{Cl}_2 \\ \hline 38 \text{Cl}_8 \text{PF}_6 \text{N}_{12} \text{PdRu}_2 & \text{C} \\ 1459.96 \\ 150(2) \text{ K} \\ \text{Monoclinic} \\ \text{P2/n} \end{array}$	$\begin{array}{c} 10 \cdot 2.25 \text{CH}_2 \text{Cl}_2 \\ \hline \\ C_{42.25} \text{H}_{36.5} \text{Cl}_{6.5} \text{F}_6 \text{N}_{12} \text{PRhRu}_2 \\ 1392.78 \\ 150(2) \text{ K} \\ \text{Monoclinic} \\ \text{P2}_1/n \end{array}$	$\begin{array}{c} 11 \cdot 2 \text{ CH}_2\text{Cl}_2 \\ \hline C_{42}\text{H}_{36}\text{Cl}_6\text{F}_6\text{IrN}_{12}\text{PRu}_2 \\ 1460.84 \\ 150(2) \text{ K} \\ \text{Monoclinic} \\ P2_1/n \end{array}$
Compound Formula Formula weight Temperature Crystal system Space group a / Å	$\begin{array}{c} \hline 7.2 CH_2 CH_2 CH_2 CH_2 CH_2 CH_2 CH_2 CH_$	$\begin{array}{c ccccc} & & & & & & & \\ \hline & & & & & \\ Ru_2Zn & & & & & \\ C_{42.79}H_{37.19}Cd & & & \\ & & & & 135 \\ & & & & & 150(\\ c & & & & & \\ c & & & & & \\ c & & & &$	$\begin{array}{c} \hline 0.398 CH_3 CN \\ \hline Cl_7 N_{12,4} O_4 Ru_2 \\ \hline Cl_3 N_{12,4} O_4 Ru_$	$\begin{array}{c c} 9.3 \text{CH}_2 \text{Cl}_2 \\ \hline 38 \text{Cl}_8 \text{PF}_6 \text{N}_{12} \text{PdRu}_2 & \text{C} \\ 1459.96 \\ 150(2) \text{ K} \\ \text{Monoclinic} \\ \text{P2/n} \\ 12.3353(5) \end{array}$	$\begin{array}{c} \textbf{10} \cdot 2.25 \text{CH}_2 \text{Cl}_2 \\ \hline \textbf{C}_{42.25} \text{H}_{36.5} \text{Cl}_{6.5} \text{F}_6 \text{N}_{12} \text{PRhRu}_2 \\ 1392.78 \\ 150(2) \text{ K} \\ \text{Monoclinic} \\ \text{P2}_1/n \\ 11.4470(4) \end{array}$	$\begin{array}{c} \underline{11 \cdot 2 \ CH_2Cl_2} \\ \hline C_{42}H_{36}Cl_6F_6IrN_{12}PRu_2 \\ 1460.84 \\ 150(2) \ K \\ Monoclinic \\ P2_1/n \\ 11.4236(3) \end{array}$
Compound Formula Formula weight Temperature Crystal system Space group a / Å b / Å	$\begin{array}{c} \hline 7.2 C H_2 H_3 H_2 C H_2 H_3 H_2 H_2 H_2 H_2 H_2 H_2 H_2 H_2 H_2 H_2$	$\begin{array}{c ccccc} & & & & & & & \\ \hline & & & & & \\ Ru_2Zn & & & & \\ C_{42.79}H_{37.19}Cd & & \\ & & & & \\ 135 & & & \\ 150(\\ c & & & & \\ c & & & & \\ c & & & & \\ 0 & & & & \\ 11.57 \\ 0 & & & & \\ 11.57 \end{array}$	$\begin{array}{c} \hline 0.398 CH_3 CN \\ \hline Cl_7 N_{12.4} O_4 Ru_2 \\ \hline Cl_7 N_{12.4} O_4 Ru_2 \\ \hline Cl_3 N_{12.4} O_4 Ru_2 \\ \hline Cl_3 N_4 \\ \hline Cl_3 \\ \hline Cl_3 N_4 \\ \hline Cl_3 \\ \hline $	$\begin{array}{c c} 9.3 CH_2 Cl_2 \\ \hline 38 Cl_8 PF_6 N_{12} Pd Ru_2 & C \\ 1459.96 \\ 150(2) K \\ Monoclinic \\ P2/n \\ 12.3353(5) \\ 10.3177(4) \end{array}$	$\begin{array}{c} \textbf{10} \cdot 2.25 \text{CH}_2 \text{Cl}_2 \\ \hline \textbf{10} \cdot 2.25 \text{CH}_2 \text{Cl}_2 \\ 1392.78 \\ 150(2) \text{ K} \\ \text{Monoclinic} \\ \text{P2}_1/n \\ 11.4470(4) \\ 21.2891(8) \end{array}$	$\begin{array}{c} \underline{11 \cdot 2 \text{ CH}_2\text{Cl}_2} \\ \hline \\ \hline C_{42}\text{H}_{36}\text{Cl}_6\text{F}_6\text{IrN}_{12}\text{PRu}_2 \\ 1460.84 \\ 150(2) \text{ K} \\ \hline \\ \text{Monoclinic} \\ P2_1/n \\ 11.4236(3) \\ 21.3206(6) \end{array}$
Compound Formula Formula weight Temperature Crystal system Space group a / Å b / Å c / Å	$\begin{array}{c} \hline 7.2 CH_2 CH_2 CH_2 CH_2 CH_2 CH_2 CH_2 CH_$	$\begin{array}{c ccccc} & & & & & & & \\ \hline & & & & & \\ Ru_2Zn & & & & \\ C_{42.79}H_{37.19}Cd & & \\ & & & & \\ 135 & & & \\ 150(\\ c & & & & \\ Monc & & \\ P2 \\) & & & & \\ 11.57 \\) & & & & \\ 21.05 \\) & & & & \\ 20.48 \end{array}$	$\begin{array}{c} \hline 0.398 CH_3 CN \\ \hline Cl_7 N_{12.4} O_4 Ru_2 & C_{43} H \\ \hline 1.80 \\ 2) K \\ cclinic \\ 1/n \\ 28(1) \\ \hline 16(2) \\ 82(2) \end{array}$	$\begin{array}{c c} 9.3 \text{CH}_2 \text{Cl}_2 \\ \hline 38 \text{Cl}_8 \text{PF}_6 \text{N}_{12} \text{PdRu}_2 & \text{C} \\ \hline 1459.96 \\ 150(2) \text{ K} \\ \text{Monoclinic} \\ \text{P2/n} \\ 12.3353(5) \\ 10.3177(4) \\ 20.5839(8) \end{array}$	$\begin{array}{c} \textbf{10} \cdot 2.25 \text{CH}_2 \text{Cl}_2 \\ \hline \\ \textbf{C}_{42.25} \text{H}_{36.5} \text{Cl}_{6.5} \text{F}_6 \text{N}_{12} \text{PRhRu}_2 \\ 1392.78 \\ 150(2) \text{ K} \\ \text{Monoclinic} \\ \text{P2}_1/n \\ 11.4470(4) \\ 21.2891(8) \\ 20.5063(8) \end{array}$	$\begin{array}{c} \underline{11 \cdot 2 \ CH_2Cl_2} \\ \hline C_{42}H_{36}Cl_6F_6IrN_{12}PRu_2 \\ 1460.84 \\ 150(2) \ K \\ Monoclinic \\ P2_1/n \\ 11.4236(3) \\ 21.3206(6) \\ 20.5318(6) \end{array}$
Compound Formula weight Temperature Crystal system Space group a / Å b / Å c / Å α/°	$\begin{array}{c} \hline 7.2 CH_2 CH_2 CH_2 CH_2 CH_2 CH_2 CH_2 CH_$	$\begin{array}{c ccccc} & & & & & & & \\ \hline & & & & & \\ Ru_2Zn & & & & \\ C_{42.79}H_{37.19}Cd & & \\ & & & & \\ 135 & & & \\ 150(\\ c & & & & \\ Monc \\ & & & & \\ P2 \\) & & & & \\ 11.57 \\) & & & & \\ 21.05 \\) & & & & \\ 20.48 \\ & & & & \\ 9 \end{array}$	$\begin{array}{c} \hline 0.398 CH_3 CN \\ \hline Cl_7 N_{12.4} O_4 Ru_2 & C_{43} H \\ \hline 1.80 \\ 2) K \\ \hline clinic \\ 1/n \\ 28(1) \\ \hline 16(2) \\ 82(2) \\ 0 \end{array}$	$\begin{array}{c c} 9 \cdot 3CH_2Cl_2 \\ \hline 3_8Cl_8PF_6N_{12}PdRu_2 & C \\ 1459.96 \\ 150(2) K \\ Monoclinic \\ P2/n \\ 12.3353(5) \\ 10.3177(4) \\ 20.5839(8) \\ 90 \end{array}$	$\begin{array}{c} \textbf{10} \cdot 2.25 \text{CH}_2 \text{Cl}_2 \\ \hline \textbf{10} \cdot 2.25 \text{CH}_2 \text{Cl}_2 \\ \hline \textbf{1392.78} \\ 150(2) \text{ K} \\ \textbf{Monoclinic} \\ \textbf{P2}_1/n \\ 11.4470(4) \\ 21.2891(8) \\ 20.5063(8) \\ \textbf{90} \end{array}$	$\begin{array}{c} \underline{11 \cdot 2 \ CH_2Cl_2} \\ \hline C_{42}H_{36}Cl_6F_6IrN_{12}PRu_2 \\ 1460.84 \\ 150(2) \ K \\ Monoclinic \\ P2_1/n \\ 11.4236(3) \\ 21.3206(6) \\ 20.5318(6) \\ 90 \end{array}$
Compound Formula weight Temperature Crystal system Space group a / Å b / Å c / Å α/ ° β/ °	$\begin{array}{c} \hline 7.2 CH_2 CH_2 CH_2 CH_2 CH_2 CH_2 CH_2 CH_$	$\begin{array}{c ccccc} & & & & & & & \\ \hline & & & & & \\ Ru_2Zn & & & & \\ C_{42.79}H_{37.19}Cd & & \\ & & & & \\ 135 & & & \\ 150(\\ c & & & & \\ Monc \\ & & & \\ P2 \\) & & & & \\ 11.57 \\) & & & & \\ 21.05 \\) & & & & \\ 20.48 \\ & & & \\ 9 \\ 2) & & & 94.51 \end{array}$	$\begin{array}{c} \hline 0.398 CH_3 CN \\ \hline Cl_7 N_{12.4} O_4 Ru_2 & C_{43} H \\ \hline 1.80 \\ 2) K \\ \hline clinic \\ 1/n \\ 28(1) \\ 16(2) \\ 82(2) \\ 0 \\ 06(7) \end{array}$	$\begin{array}{c c} \textbf{9} \cdot 3CH_2Cl_2 \\ \hline \textbf{38}Cl_8PF_6N_{12}PdRu_2 & C \\ \hline 1459.96 & \\ 150(2) \text{ K} \\ \text{Monoclinic} \\ P2/n \\ 12.3353(5) \\ 10.3177(4) \\ 20.5839(8) \\ 90 \\ 92.4271(10) \end{array}$	$\begin{array}{c} 10 \cdot 2.25 \text{CH}_2 \text{Cl}_2 \\ \hline \\ $	$\begin{array}{c} \textbf{11} \cdot 2 \ CH_2Cl_2 \\ \hline C_{42}H_{36}Cl_6F_6IrN_{12}PRu_2 \\ 1460.84 \\ 150(2) \ K \\ Monoclinic \\ P2_1/n \\ 11.4236(3) \\ 21.3206(6) \\ 20.5318(6) \\ 90 \\ 92.9580(9) \end{array}$
Compound Formula Formula weight Temperature Crystal system Space group a / Å b / Å c / Å α/ ° β/ ° γ/ °	$\begin{array}{c} \hline 7.2 C H_2 H_{36} H_{$	$\begin{array}{c ccccc} & & & & & & & & \\ \hline & & & & & & \\ Ru_2Zn & & & & & \\ C_{42.79}H_{37.19}Cd & & & \\ & & & & & \\ 135 & & & & \\ 150(\\ c & & & & & \\ Monc & & & \\ c & & & & & \\ P2 \\ 0 & & & & & \\ 11.57 \\ 0 & & & & & \\ 11.57 \\ 0 & & & & & \\ 11.57 \\ 0 & & & & & \\ 21.05 \\ 0 & & & & & \\ 9 \\ 2) & & & & & \\ 9 \\ 2) & & & & & \\ 9 \\ 2) & & & & & \\ 9 \\ 2) & & & & & \\ 9 \\ 2) & & & & \\ 9 \\ 2) & & & & \\ 9 \\ 2) & & & & \\ 9 \\ 2) & & & & \\ 9 \\ 2) & & & & \\ 9 \\ 2) & & & & \\ 9 \\ 2) & & & & \\ 9 \\ 2) & & & & \\ 9 \\ 2) & & & & \\ 15 \\ 20 \\ 20 \\ 20 \\ 20 \\ 20 \\ 20 \\ 20 \\ 2$	$\begin{array}{c} \hline 0.398 CH_3 CN \\ \hline Cl_7 N_{12,4} O_4 Ru_2 & C_{43} H \\ \hline 1.80 \\ 2) K \\ \text{sclinic} \\ 1/n \\ 28(1) \\ \hline 16(2) \\ 82(2) \\ 0 \\ 06(7) \\ 0 \end{array}$	$\begin{array}{c c} 9 \cdot 3 CH_2 Cl_2 \\ \hline \\ 38 Cl_8 PF_6 N_{12} PdRu_2 & C \\ 1459.96 \\ 150(2) K \\ Monoclinic \\ P2/n \\ 12.3353(5) \\ 10.3177(4) \\ 20.5839(8) \\ 90 \\ 92.4271(10) \\ 90 \end{array}$	$\begin{array}{c} \textbf{10} \cdot 2.25 \text{CH}_2 \text{Cl}_2 \\ \hline \textbf{10} \cdot 2.25 \text{CH}_2 \text{Cl}_2 \\ \hline \textbf{1392.78} \\ 1592.78 \\ 150(2) \text{ K} \\ \text{Monoclinic} \\ \text{P2}_1/n \\ 11.4470(4) \\ 21.2891(8) \\ 20.5063(8) \\ 90 \\ 93.0985(12) \\ 90 \end{array}$	$\begin{array}{c} \textbf{11} \cdot 2 \ \text{CH}_2\text{Cl}_2 \\ \hline \textbf{C}_{42}\text{H}_{36}\text{Cl}_6\text{F}_6\text{IrN}_{12}\text{PRu}_2 \\ 1460.84 \\ 150(2) \ \text{K} \\ \text{Monoclinic} \\ P2_1/n \\ 11.4236(3) \\ 21.3206(6) \\ 20.5318(6) \\ 90 \\ 92.9580(9) \\ 90 \end{array}$
CompoundFormulaFormula weightTemperatureCrystal systemSpace group $a / Å$ $b / Å$ $c / Å$ $\alpha / °$ $\beta / °$ $\gamma / °$ $V / Å^3$	$\begin{array}{c} \hline 7.2 C H_2 H_{36} H_{3$	$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	$\begin{array}{c} \hline 0.398 CH_3 CN \\ \hline Cl_7 N_{12.4} O_4 Ru_2 & C_{43} H \\ \hline 1.80 \\ 2) K \\ \text{sclinic} \\ 1/n \\ 28(1) \\ \hline 16(2) \\ 82(2) \\ 0 \\ 06(7) \\ 0 \\ 00(8) \end{array}$	$\begin{array}{c c} 9.3 \text{CH}_2 \text{Cl}_2 \\ \hline 38 \text{Cl}_8 \text{PF}_6 \text{N}_{12} \text{PdRu}_2 & \text{C} \\ \hline 1459.96 \\ \hline 150(2) \text{ K} \\ \text{Monoclinic} \\ \text{P2/n} \\ \hline 12.3353(5) \\ \hline 10.3177(4) \\ 20.5839(8) \\ 90 \\ 92.4271(10) \\ 90 \\ 2617.40(18) \end{array}$	$\begin{array}{c} \textbf{10} \cdot 2.25 \text{CH}_2 \text{Cl}_2 \\ \hline \textbf{10} \cdot 2.25 \text{CH}_2 \text{Cl}_2 \\ \hline \textbf{1392.78} \\ 150(2) \text{ K} \\ \text{Monoclinic} \\ \text{P2}_1/n \\ 11.4470(4) \\ 21.2891(8) \\ 20.5063(8) \\ 90 \\ 93.0985(12) \\ 90 \\ 4990.0(3) \end{array}$	$\begin{array}{c} \textbf{11} \cdot 2 \ \text{CH}_2\text{Cl}_2 \\ \hline \textbf{C}_{42}\text{H}_{36}\text{Cl}_6\text{F}_6\text{IrN}_{12}\text{PRu}_2 \\ 1460.84 \\ 150(2) \ \text{K} \\ \text{Monoclinic} \\ P2_1/n \\ 11.4236(3) \\ 21.3206(6) \\ 20.5318(6) \\ 90 \\ 92.9580(9) \\ 90 \\ 4994.0(2) \end{array}$
CompoundFormulaFormula weightTemperatureCrystal systemSpace group $a / Å$ $b / Å$ $c / Å$ $\alpha / °$ $\beta / °$ $\gamma / °$ $V / Å^3$ Z	$\begin{array}{c} \hline 7.2 C H_2 H_3 (C H_2 H_3 (C H_2 H_3 (C H_2 H_2 H_2 H_2 H_2 H_2 H_2 H_2 H_2 H_2$	$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	$\begin{array}{c} \hline 0.398 CH_3 CN \\ \hline Cl_7 N_{12.4} O_4 Ru_2 & C_{43} H \\ \hline 1.80 \\ 2) K \\ cclinic \\ 1/n \\ 28(1) \\ \hline 16(2) \\ 82(2) \\ 0 \\ 06(7) \\ 0 \\ 00(8) \\ 4 \end{array}$	$\begin{array}{c c} 9 \cdot 3 C H_2 C I_2 \\ \hline 38 C I_8 P F_6 N_{12} P d R u_2 & C \\ 1459.96 \\ 150(2) K \\ Monoclinic \\ P2/n \\ 12.3353(5) \\ 10.3177(4) \\ 20.5839(8) \\ 90 \\ 92.4271(10) \\ 90 \\ 2617.40(18) \\ 2 \end{array}$	$\begin{array}{c} \textbf{10} \cdot 2.25 \text{CH}_2 \text{Cl}_2 \\ \hline \textbf{10} \cdot 2.25 \text{CH}_2 \text{Cl}_2 \\ \hline \textbf{1392.78} \\ 150(2) \text{ K} \\ \textbf{Monoclinic} \\ \textbf{P2}_1/n \\ 11.4470(4) \\ 21.2891(8) \\ 20.5063(8) \\ \textbf{90} \\ \textbf{93.0985(12)} \\ \textbf{90} \\ \textbf{4990.0(3)} \\ \textbf{4} \end{array}$	$\begin{array}{c} \underline{11 \cdot 2 \ CH_2Cl_2} \\ \hline C_{42}H_{36}Cl_6F_6IrN_{12}PRu_2 \\ 1460.84 \\ 150(2) \ K \\ Monoclinic \\ P2_1/n \\ 11.4236(3) \\ 21.3206(6) \\ 20.5318(6) \\ 90 \\ 92.9580(9) \\ 90 \\ 4994.0(2) \\ 4 \end{array}$
CompoundFormulaFormula weightTemperatureCrystal systemSpace group $a / Å$ $b / Å$ $c / Å$ $\alpha / °$ $\beta / °$ $\gamma / °$ $V / Å^3$ ZReflection collected	$\begin{array}{c} \hline 7.2 C H_2 H_3 6 C H_2 H_3 6 C H_2 H_2 H_2 H_2 H_2 H_2 H_2 H_2 H_2 H_2$	$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	$\begin{array}{c} \hline 0.398 CH_3 CN \\ \hline Cl_7 N_{12,4} O_4 Ru_2 & C_{43} H \\ \hline 1.80 \\ 2) K \\ clinic \\ \hline 1/n \\ 28(1) \\ \hline 16(2) \\ 82(2) \\ 0 \\ 06(7) \\ 0 \\ 00(8) \\ 4 \\ 517 \end{array}$	$\begin{array}{c c} 9 \cdot 3 C H_2 C I_2 \\ \hline \\ 38 C I_8 P F_6 N_{12} P d R u_2 & C \\ 1459.96 \\ 150(2) K \\ Monoclinic \\ P2/n \\ 12.3353(5) \\ 10.3177(4) \\ 20.5839(8) \\ 90 \\ 92.4271(10) \\ 90 \\ 2617.40(18) \\ 2 \\ 20401 \\ \end{array}$	$\begin{array}{c} \textbf{10} \cdot 2.25 \text{CH}_2 \text{Cl}_2 \\ \hline \textbf{10} \cdot 2.25 \text{CH}_2 \text{Cl}_2 \\ \hline \textbf{1392.78} \\ 150(2) \text{ K} \\ \text{Monoclinic} \\ \text{P2}_1/\text{n} \\ 11.4470(4) \\ 21.2891(8) \\ 20.5063(8) \\ 90 \\ 93.0985(12) \\ 90 \\ 4990.0(3) \\ 4 \\ 34072 \end{array}$	$\begin{array}{c} \textbf{11} \cdot 2 \ \text{CH}_2\text{Cl}_2 \\ \hline \textbf{C}_{42}\text{H}_{36}\text{Cl}_6\text{F}_6\text{IrN}_{12}\text{PRu}_2 \\ 1460.84 \\ 150(2) \ \text{K} \\ \text{Monoclinic} \\ P2_1/n \\ 11.4236(3) \\ 21.3206(6) \\ 20.5318(6) \\ 90 \\ 92.9580(9) \\ 90 \\ 4994.0(2) \\ 4 \\ 37341 \end{array}$
$\begin{tabular}{lllllllllllllllllllllllllllllllllll$	$\begin{array}{c} \hline 7.2 CH_2 CH_2 CH_2 CH_2 CH_2 CH_2 CH_2 CH_$	$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	$\begin{array}{c} \hline 0.398 CH_3 CN \\ \hline Cl_7 N_{12.4} O_4 Ru_2 & C_{43} H \\ \hline 1.80 \\ 2) K \\ \hline clinic \\ 1/n \\ 28(1) \\ \hline 16(2) \\ 82(2) \\ 0 \\ 06(7) \\ 0 \\ 00(8) \\ 4 \\ 517 \\ \hline 314 \end{array}$	$\begin{array}{c c} 9 \cdot 3 C H_2 C I_2 \\ \hline 38 C I_8 PF_6 N_{12} P d R u_2 & C \\ 1459.96 & \\ 150(2) \ K \\ Monoclinic \\ P2/n \\ 12.3353(5) \\ 10.3177(4) \\ 20.5839(8) \\ 90 \\ 92.4271(10) \\ 90 \\ 2617.40(18) \\ 2 \\ 20401 \\ 6006 \\ \end{array}$	$\begin{array}{c} \textbf{10} \cdot 2.25 \text{CH}_2 \text{Cl}_2 \\ \hline \textbf{2}_{42.25} \text{H}_{36.5} \text{Cl}_{6.5} \text{F}_6 \text{N}_{12} \text{PRhRu}_2 \\ 1392.78 \\ 150(2) \text{ K} \\ \text{Monoclinic} \\ \text{P2}_1/n \\ 11.4470(4) \\ 21.2891(8) \\ 20.5063(8) \\ 90 \\ 93.0985(12) \\ 90 \\ 4990.0(3) \\ 4 \\ 34072 \\ 11437 \end{array}$	$\begin{array}{c} \underline{11 \cdot 2 \ CH_2Cl_2} \\ \hline C_{42}H_{36}Cl_6F_6IrN_{12}PRu_2 \\ 1460.84 \\ 150(2) \ K \\ Monoclinic \\ P2_1/n \\ 11.4236(3) \\ 21.3206(6) \\ 20.5318(6) \\ 90 \\ 92.9580(9) \\ 90 \\ 4994.0(2) \\ 4 \\ 37341 \\ 11447 \end{array}$
CompoundFormulaFormula weightTemperatureCrystal systemSpace group $a / Å$ $b / Å$ $c / Å$ $\alpha / °$ $\beta / °$ $\gamma / °$ $V / Å^3$ ZReflection collectedIndependent reflectionsR1, wR2 [I>2sigma(I)]	$\begin{array}{c} \hline 7.2 CH_2 CH_2 CH_2 CH_2 CH_2 CH_2 CH_2 CH_$	$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	$\begin{array}{c} 0.398 CH_{3} CN \\ \hline Cl_{7} N_{12.4} O_{4} Ru_{2} \\ 2) K \\ clinic \\ 1/n \\ 28(1) \\ 16(2) \\ 82(2) \\ 0 \\ 06(7) \\ 0 \\ 00(8) \\ 4 \\ 517 \\ 314 \\ 0.1478 \end{array} $	$\begin{array}{c c} 9 \cdot 3 C H_2 C I_2 \\ \hline 38 C I_8 PF_6 N_{12} P d R u_2 & C \\ 1459.96 & \\ 150(2) \ K \\ Monoclinic \\ P2/n \\ 12.3353(5) \\ 10.3177(4) \\ 20.5839(8) \\ 90 \\ 92.4271(10) \\ 90 \\ 2617.40(18) \\ 2 \\ 20401 \\ 6006 \\ 0.0294, 0.0569 \end{array}$	$\begin{array}{c} \textbf{10} \cdot 2.25 \text{CH}_2 \text{Cl}_2 \\ \hline \textbf{10} \cdot 2.25 \text{CH}_2 \text{Cl}_2 \\ \hline \textbf{1392.78} \\ 150(2) \text{ K} \\ \text{Monoclinic} \\ P2_1/n \\ 11.4470(4) \\ 21.2891(8) \\ 20.5063(8) \\ 90 \\ 93.0985(12) \\ 90 \\ 4990.0(3) \\ 4 \\ 34072 \\ 11437 \\ 0.0429, 0.0971 \\ \end{array}$	$\begin{array}{c} \textbf{11} \cdot 2 \ \text{CH}_2\text{Cl}_2 \\ \hline \textbf{C}_{42}\text{H}_{36}\text{Cl}_6\text{F}_6\text{IrN}_{12}\text{PRu}_2 \\ 1460.84 \\ 150(2) \ \text{K} \\ \text{Monoclinic} \\ P2_1/n \\ 11.4236(3) \\ 21.3206(6) \\ 20.5318(6) \\ 90 \\ 92.9580(9) \\ 90 \\ 4994.0(2) \\ 4 \\ 37341 \\ 11447 \\ 0.0389, 0.0844 \end{array}$
$\begin{tabular}{lllllllllllllllllllllllllllllllllll$	$\begin{array}{c} \hline 7.2 C H_2 H_{36} C H_{36} H_{36$	$\begin{array}{c c} & 8 \cdot 2 \mathrm{CH}_2 \mathrm{Cl}_2 \cdot \mathrm{Cl}_2 $	$\begin{array}{c} \hline 0.398 CH_3 CN \\ \hline Cl_7 N_{12.4} O_4 Ru_2 \\ 1.80 \\ 2) K \\ clinic \\ 1/n \\ 28(1) \\ 16(2) \\ 82(2) \\ 0 \\ 06(7) \\ 0 \\ 00(8) \\ 4 \\ 517 \\ 314 \\ 0.1478 \\ 0.1478 \\ 0.1565 \end{array}$	$\begin{array}{c c} 9 \cdot 3 C H_2 C I_2 \\ \hline \\ 38 C I_8 P F_6 N_{12} P d R u_2 & C \\ 1459.96 \\ 150(2) K \\ Monoclinic \\ P2/n \\ 12.3353(5) \\ 10.3177(4) \\ 20.5839(8) \\ 90 \\ 92.4271(10) \\ 90 \\ 2617.40(18) \\ 2 \\ 20401 \\ 6006 \\ 0.0294, 0.0569 \\ 0.0351, 0.0592 \\ \end{array}$	$\begin{array}{c} \hline 10 \cdot 2.25 \text{CH}_2 \text{Cl}_2 \\ \hline 2_{42.25} \text{H}_{36.5} \text{Cl}_{6.5} \text{F}_6 \text{N}_{12} \text{PRhRu}_2 \\ 1392.78 \\ 150(2) \text{ K} \\ \hline \text{Monoclinic} \\ \text{P2}_1/n \\ 11.4470(4) \\ 21.2891(8) \\ 20.5063(8) \\ 90 \\ 93.0985(12) \\ 90 \\ 4990.0(3) \\ 4 \\ 34072 \\ 11437 \\ 0.0429, 0.0971 \\ 0.0493, 0.1000 \\ \end{array}$	$\begin{array}{r} \textbf{11} \cdot 2 \ \text{CH}_2\text{Cl}_2 \\ \hline \textbf{C}_{42}\text{H}_{36}\text{Cl}_6\text{F}_6\text{IrN}_{12}\text{PRu}_2 \\ 1460.84 \\ 150(2) \ \text{K} \\ \text{Monoclinic} \\ P2_1/n \\ 11.4236(3) \\ 21.3206(6) \\ 20.5318(6) \\ 90 \\ 92.9580(9) \\ 90 \\ 4994.0(2) \\ 4 \\ 37341 \\ 11447 \\ 0.0389, 0.0844 \\ 0.0459, 0.0871 \\ \end{array}$

 Table S1 : X-ray crystallographic data for compounds 1-11.



Figure S3. ORTEP view of the molecular structures of **5** (a), **6** (b), **7** (c), **8** (d) and $[PdRu_2(dpa)_4Cl(OC(CH_3)_2)](PF_6)_2$ (e). All atoms are drawn at the 50% probability level and hydrogen atoms are omitted for clarity. Crystal data for $[PdRu_2(dpa)_4Cl(OC(CH_3)_2)](PF_6)_2$: $C_{45}H_{45}ClF_{12}N_{12}O_{3.5}P_2PdRu_2$, Mw = 1443.86, Triclinic, space group Pī., a = 9.8959(3) Å, b = 17.3019(5) Å, c = 17.8514(5) Å, $\alpha = 98.5142(14)^\circ$, $\beta = 100.5271(15)^\circ$, $\gamma = 98.7123(14)^\circ$, V = 2921.39(15) Å³, Z = 2, T = 150(2) K, 34873 reflection collected, 16339 independent, R1 = 0.0984, wR2 = 0.2661[I>2sigma(I)], R1 = 0.1420, wR2 = 0.3137 for all data, crystal size 0.300 x 0.160 x 0.030 mm³. This structural refinement was not satisfactory due to the thin-plate shape crystal. Hence, it is less accurate in C–C bond distances. However, the distances of heavy atoms (Ru and Pd) are still reliable. (Ru1–Ru2 = 2.2584(13); Ru2…Pd1 = 2.4910(13)).



Figure S4. The crystal structures of $[RhRu_2(dpa)_4Cl_2](PF_6)$ (10) (left) and $[IrRu_2(dpa)_4Cl_2](PF_6)$ (11)(right). The solvent molecules and hydrogen atoms are omitted for clarity. Occupancy contributions: M1 = 0.53 Rh / 0.47 Ru and M2 = 0.47 Rh / 0.53 Ru for $[RhRu_2(dpa)_4Cl_2]^+$; M1 = 0.523 Ir / 0.477 Ru and M2 = 0.477 Ir / 0.523 Ru for $[IrRu_2(dpa)_4Cl_2]^+$

Spectroscopic Characterization:



Figure S5. (c) ¹H NMR spectrum of (3) in CD_2Cl_2



Figure S5. (f) ¹H NMR spectrum of (6) in CD₂Cl₂



Figure S5. (h) ¹H NMR spectrum of (8) in CD_2Cl_2



Figure S5. (i) ¹H NMR spectrum of (9) in CD_2Cl_2



Figure S5. (k) ¹H NMR spectrum of (11) in CD_2Cl_2



Figure S6. (a) 2D COSY NMR spectrum of (1) in CD₂Cl₂.

The "Ru" and "Co" marks represent the protons of pyridine coordinated to the correlated metal. All below figures use the same assignment. The peaks assignment of "Co" and "Fe" refers to the known dpa⁻ complexes. ^{2,3}



Figure S6. (b) 2D COSY NMR spectrum of (3) in CD₂Cl₂.



Figure S6. (c) 2D COSY NMR spectrum of (5) in CD₂Cl₂.



Figure S6. (d) 2D COSY NMR spectrum of (6) in CD₂Cl₂.



Figure S6. (e) 2D COSY NMR spectrum of (7) in CD_2Cl_2 .



Figure S6. (f) 2D COSY NMR spectrum of (8) in CD₂Cl₂.



Figure S6. (g) 2D COSY NMR spectrum of (9) in CD_2Cl_2 .



Figure S6. (h) 2D COSY NMR spectrum of (10) in CD_2Cl_2 . The assignment of protons is according to previous study on [NiCoRh(dpa)₄Cl](PF₆).³ Typically, the coupling constant of proton in *meta* position of pyridyl ring is higher than that in *ortho* position (in [NiCoRh(dpa)₄Cl](PF₆), coupling constants are 5.7 and ~8 Hz for the protons on the *ortho* and *meta* positions, respectively). The doublet peaks of H₁ in **10** (J = 5.7

Hz) could be therefore, assigned to an *ortho* position, as shown in the figure S6 (h). All protons can be then assigned according to their correlated peaks. The small red dots (lower intensity) are from long range coupling.



Figure S6. (i) 2D COSY NMR spectrum of (11) in CD₂Cl₂.

Table S2. Summary of ¹H NMR chemical shifts. (δ :ppm)

Compounds	H _{py} coordinated to 'M'	H_{py} coordinated to Ru_{outer}
$[\text{CoRu}_2(\text{dpa})_4\text{Cl}_2](1)$	126.04 (broad), 78.40, 57.24, 14.54	41.67, 1.87, -18.56, -49.06.
$[\text{CoRu}_2(\text{dpa})_4\text{Cl}_2](\text{PF}_6) (3)$	122.01 (broad), 47.72, 42.02, 1.63	36.27, -11.05, -27.80, -59.35
$[MnRu_{2}(dpa)_{4}Cl_{2}](ClO_{4}) (5)$	98.87 (broad), 33.58, 8.14, 1.41	24.37, 4.14, -15.82, -28.9
$[FeRu_2(dpa)_4Cl_2](PF_6) (6)$	123.43 (broad), 41.48, 30.72, 9.15	29.00, -4.95, -20.2, -42.42
$[ZnRu_2(dpa)_4Cl_2](ClO_4) (7)$	29.26, 15.59, -22.85, -36.6	35.84, -3.49, -30.08, -55.07
$[CdRu_2(dpa)_4Cl_2](ClO_4) (8)$	31.12, 15.85, -21.5, -34.48	35.23, -3.67, -30.35, -54.37
$[PdRu_2(dpa)_4Cl_2](PF_6) (9)$	24.11, 1.21, -16.36, , -38.31	40.14, -1.25, -20.75, -43.67
[RhRu ₂ (dpa) ₄ Cl ₂](PF ₆) (10)	25.26, 15.31, -26.43(d), -30.33	28.17, 11.28, -26.07(d), -33.5
[IrRu ₂ (dpa) ₄ Cl](PF ₆) (11)	26.04, 17.34, -23.22, -29.59	27.68, 11.68, -21.96, -29.59



Figure S7. The high-resolution mass spectra (MALDI-TOF) of complexes 9(a), 10(b) and 11(c). The red peaks are simulated results.

DFT Calculation:

Table 55. Experimental and calculated bold distances (7) and angles (7) for 10 (Rif) and 11 (fr)					
	10 ^{<i>a</i>}	10 (DFT)	11 ^a	11 (DFT)	
Ru-Ru	2.3254(4)	2.297	2.3312(4)	2.304	
Ru-M	2.3352(4)	2.401	2.3472(4)	2.408	
M-Cl	2.4869(10)	2.508	2.4782(11)	2.429	
Ru-Cl	2.4888(10)	2.423	2.4746(11)	2.512	
M-N _{outer}	2.046-2.175	2.030-2.146	2.053-2.141	2.034-2.139	
Ru-N _{outer}	2.044-2.151	2.026-2.233	2.057-2.163	2.022-2.242	
Ru-N _{inner}	1.965-2.096	1.961-2.126	1.967-2.101	1.960-2.134	
∠ M–Ru–Ru	167.48(2)	166.2	167.61(2)	165.9	

Table S3. Experimental and calculated bond distances (Å) and angles (°) for 10 (Rh) and 11 (Ir)

^{*a*} The data set is from higher occupancy of two disordered molecular.



Figure S8. Orbital scheme from localized quasi-restricted orbitals (doubly occupied) and quasi-restricted orbitals (singly occupied) for S = 1 configuration of [RhRu₂(dpa)₄Cl₂](PF₆) **10**. Hydrogen atoms are omitted for clarity.



Figure S9. Orbital scheme from localized quasi-restricted orbitals (doubly occupied) and quasi-restricted orbitals (singly occupied) for S = 1 configuration of $[IrRu_2(dpa)_4Cl_2]^+$ (11). Hydrogen atoms are omitted for clarity.

Rh	0.282386	-0.420638	-2.288418
Ru	0.071078	-0.001768	2.351399
Ru	0.115581	0.068292	0.056222
Cl	-0.156930	0.381042	4.733040
Cl	0.435423	-0.618803	-4.783466
Ν	0.884875	-1.852871	2.481624
Ν	0.234706	-2.045263	0.249009
Ν	-0.390719	-2.313257	-1.993169
Ν	-0.764277	2.033912	1.972295
Ν	0.047469	2.009918	-0.210365
Ν	0.938916	1.621596	-2.330736
Ν	2.019057	0.826047	2.295262
N	2.158439	0.068469	0.092646
Ν	2.208575	-1.041284	-1.965654
Ν	-1.847543	-0.818097	2.178198
Ν	-1.885219	-0.133815	-0.050937
Ν	-1.715700	0.245090	-2.352218
С	1.559142	-2.297489	3.570987
Н	1.514789	-1.638524	4.437449
С	2.236107	-3.502578	3.582904
Н	2.750824	-3.822778	4.486558
С	2.255428	-4.272397	2.409301
Н	2.813593	-5.207942	2.369190
С	1.580829	-3.822378	1.288348
н	1.614849	-4.376242	0.351968
С	0.867832	-2.604021	1.329780
Č	-0.318974	-2.835197	-0.727535
С	-0.905532	-4.093698	-0.468242
Ĥ	-0.895407	-4.472541	0.551881
С	-1.518018	-4.800870	-1.486733
Н	-1.977496	-5.767007	-1.277507
С	-1.564364	-4.252645	-2 776293
Н	-2.034848	-4.775665	-3.606327
С	-1.005431	-3.004601	-2.982353
Н	-1.003667	-2.508996	-3.953201
С	-1.505568	2.621982	2.939269
Н	-1.593839	2.050347	3.861149
С	-2.104547	3.865013	2.797979
Н	-2.685511	4.276160	3.621529
С	-1.958719	4.546628	1.586891
Н	-2.437142	5.512769	1.426287
С	-1.218158	3.960538	0.572659
Н	-1.134556	4.434131	-0.403235
С	-0.618128	2.705050	0.778692
С	0.707869	2.513953	-1.308923
С	1.227697	3.821159	-1.373177
Н	1.083239	4.488255	-0.526344
С	1.962896	4.219769	-2.478475
Н	2.376774	5.227466	-2.522174
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