

## Pt<sup>IV</sup> azido complexes undergo copper-free click reactions with alkynes

Electronic Supplementary Information

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### Synthesis and Characterisation

#### **Mono-substituted *trans*-[Pt(C<sub>6</sub>H<sub>6</sub>N<sub>3</sub>O<sub>4</sub>)(N<sub>3</sub>)(py)<sub>2</sub>]**

Dimethyl acetylenedicarboxylate (**2**, 30.9  $\mu$ l, 0.25 mmol) in MeOH (5 ml) was added dropwise to *trans*-[Pt(N<sub>3</sub>)<sub>2</sub>(py)<sub>2</sub>] (100 mg, 0.23 mmol) in MeOH (50 ml). The solution was stirred at 35°C for 2 days and then reduced to dryness. The solution was resuspended in minimal MeCN, filtered (IM) and purified by HPLC under neutral conditions to give the monosubstituted complex as a pale yellow solid (43 mg, 0.07 mmol, 32 %).

<sup>1</sup>H NMR (400 MHz, *d*<sub>3</sub>-MeCN)  $\delta$ : 8.73 (d, 4H, <sup>3</sup>J<sub>HH</sub> = 7, <sup>3</sup>J<sub>HPt</sub> = 38, H<sub>o</sub>), 7.97 (dd, 2H, <sup>3</sup>J<sub>HH</sub> = 7, H<sub>p</sub>), 7.49 (dd, <sup>3</sup>J<sub>HH</sub> = 7, H<sub>m</sub>), 3.76 (s, OMe, 6H). <sup>195</sup>Pt NMR (86 MHz, *d*<sub>3</sub>-MeCN)  $\delta$ : -2219 (W  $\frac{1}{2}$  1150 Hz). **ESI-MS** (MeOH) (M = *trans*-[Pt(C<sub>6</sub>H<sub>6</sub>N<sub>3</sub>O<sub>4</sub>)(N<sub>3</sub>)(py)<sub>2</sub>]) *m/z*: 602.111 ([M + Na]<sup>+</sup>, calcd 602.084); 580.125 ([M + H]<sup>+</sup> calcd 580.102) **HRMS** (MeOH) *m/z*: 580.1020 ([M + H]<sup>+</sup>, C<sub>16</sub>H<sub>17</sub>N<sub>8</sub>O<sub>4</sub>Pt, calcd. 580.1015). **IR** v cm<sup>-1</sup>: 2043 (N<sub>3</sub>), 1728, 1452, 1289, 1172, 1090, 769, 693.

#### **Bis-substituted *trans*-[Pt(C<sub>6</sub>H<sub>6</sub>N<sub>3</sub>O<sub>4</sub>)<sub>2</sub>(py)<sub>2</sub>]**

Dimethyl acetylenedicarboxylate (**2**, 141  $\mu$ l, 1.14 mmol) and *trans*-[Pt(N<sub>3</sub>)<sub>2</sub>(py)<sub>2</sub>] (100 mg, 0.229 mmol) were stirred in MeOH (50 ml) at 35 °C for 3 d before the solvent was reduced in volume to 2 ml and placed in the fridge overnight. The solution was filtered and the precipitate washed with cold MeOH to give the title compound as a white solid (17.5 mg, 0.03 mmol, 11 %). NMR spectroscopic experiments were run on freshly made solutions since the complex was unstable in *d*<sub>6</sub>-acetone.

<sup>1</sup>H NMR (600 MHz, *d*<sub>6</sub>-acetone)  $\delta$ : 8.67 (d, 4H, <sup>3</sup>J<sub>HH</sub> = 6, H<sub>o</sub>), 8.03 (dd, 2H, <sup>3</sup>J<sub>HH</sub> = 7, H<sub>p</sub>), 7.55 (dd, <sup>3</sup>J<sub>HH</sub> = 7, H<sub>m</sub>), 3.90 (6H, OMe), 3.75 (6H, OMe). <sup>14</sup>N NMR (43 MHz, *d*<sub>6</sub>-acetone)  $\delta$ : 287.9 (sharp, N<sub>2</sub> gas), 216.1 (br, W<sub>1/2</sub> 909 Hz N<sub>triazole</sub>), 175.4 (br, W<sub>1/2</sub> 793 Hz, N<sub>py</sub>/N<sub>triazole</sub>). <sup>195</sup>Pt NMR (129 MHz, *d*<sub>6</sub>-acetone)  $\delta$ : -2331 (W<sub>1/2</sub> 1100 Hz).

**ESI-MS** (MeOH) *m/z*: (M = *trans*-[Pt(C<sub>6</sub>H<sub>6</sub>N<sub>3</sub>O<sub>4</sub>)<sub>2</sub>(py)<sub>2</sub>]): 1465.176 ([2M + Na]<sup>+</sup> calcd 1465.230); 1443.137 ([2M + H]<sup>+</sup> calcd 1443.248); 744.094 ([M + Na]<sup>+</sup> calcd 744.110);

722.045 ( $[M + H]^+$  calcd 722.128). **HRMS** (MeOH)  $m/z$ : 744.1095 ( $[M + Na]^+$ ,  $C_{22}H_{22}N_8O_8PtNa$ , calcd. 744.1101). **IR**  $\nu$   $cm^{-1}$ : 1726, 1454, 1209, 1172, 771, 693.

#### Rearrangement complex **2a''**

DMAD (**2**, 14  $\mu$ l, 0.117 mmol) in MeOH (2 ml) was added dropwise to **1** (50 mg, 0.106 mmol) in MeOH (3 ml). The solution was stirred at 35°C for 7d before being placed on ice. Product **2a''** was isolated as a white precipitate by filtration and rinsed with cold H<sub>2</sub>O, MeOH and diethylether (18 mg, 0.03 mmol, 28 %). Complex **2a''** could also be obtained by dissolving **2a'** in CDCl<sub>3</sub> and leaving for several days.

**<sup>1</sup>H NMR** (500 MHz, CDCl<sub>3</sub>)  $\delta$ : 9.01 (d,  $^3J_{HPt} = 27$ ,  $^3J_{HH} = 6.0$ , 4H, H<sub>o</sub>), 8.08 (t,  $^3J_{HH} = 6$ ,  $J_{HH} = 6$ , 2H, H<sub>p</sub>), 7.66 (dd,  $^3J_{HH} = 6$ ,  $^3J_{HH} = 6$ , 4H, H<sub>m</sub>), 3.88 (s, OCH<sub>3</sub>, 3H), 2.89 (br, 1H), 1.69 (br, 1H).

**<sup>195</sup>Pt NMR** (107 MHz, CDCl<sub>3</sub>)  $\delta$ : 764 (PWHH 680 ppm).

**<sup>14</sup>N NMR** (29 MHz, CDCl<sub>3</sub>)  $\delta$ : 288.6 (N<sub>2</sub> gas), 227.2 (N <sub>$\beta$</sub> ), 168.2 (N<sub>v</sub>/N<sub>py</sub>). N <sub>$\alpha$</sub>  not seen.

**<sup>13</sup>C NMR** (125 MHz, CDCl<sub>3</sub>)  $\delta$ : 164.3 (ester), 160.8 (ester), 149.4 C<sub>pyortho</sub>, 142.2 (C<sub>pypara</sub>), 139.1 ( $^3J_{Cpt} = 31.0$ , C<sub>triazole</sub>), 135.6 ( $^3J_{Cpt} = 18.0$ , C<sub>triazole</sub>), 127.1 ( $^3J_{Cpt} = 27.0$ , C<sub>pymeta</sub>), 52.4 (C<sub>alkyl</sub>).

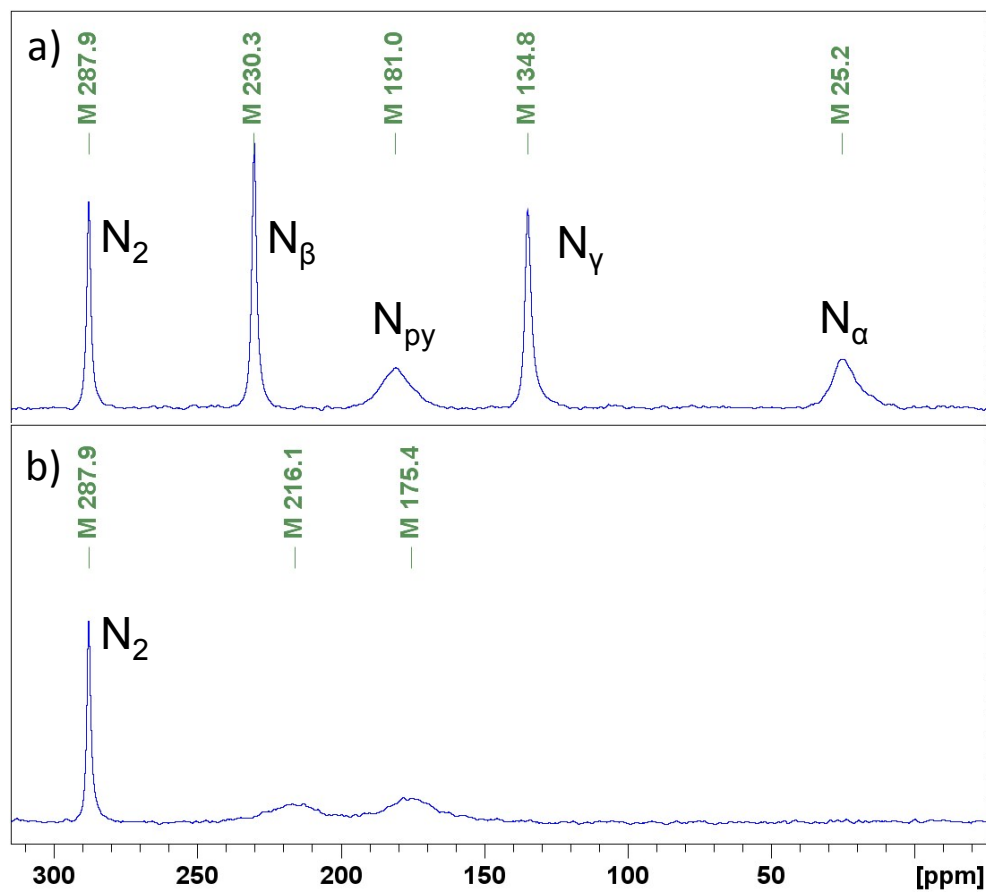
**IR** (solid)  $\nu$   $cm^{-1}$ : 3465, 2048 (asym N<sub>3</sub>), 1733, 1675, 1611, 1483, 1437, 1389, 1338, 1265, 1255, 1222, 1211, 1128, 1090, 1079, 836, 774, 691.

**ESI-MS** identical to **2a'**.

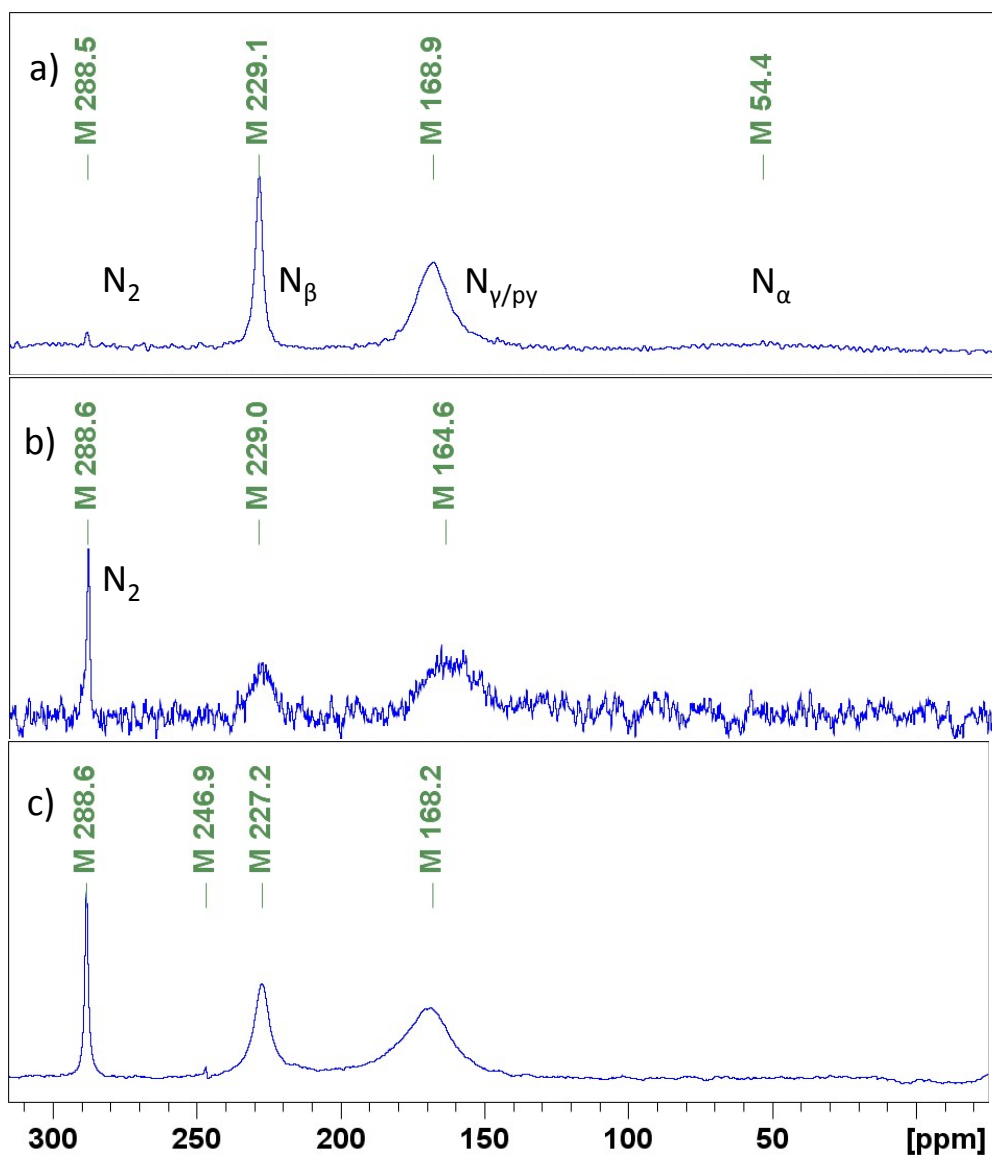
#### Rearrangement complex **2b''**

Complex **2b''** was obtained by dissolving **2b'** in CDCl<sub>3</sub> and leaving for several days.

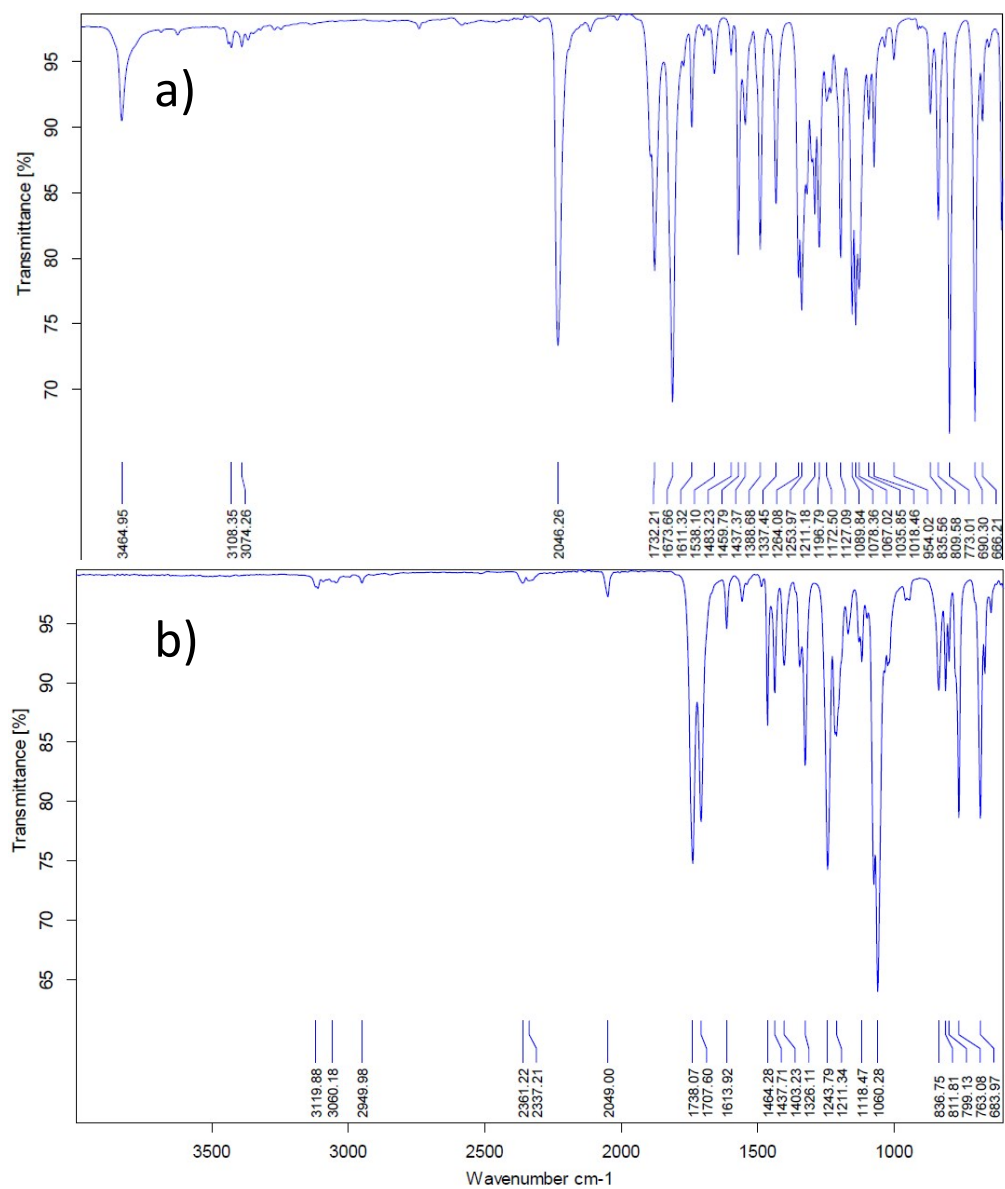
**<sup>1</sup>H NMR** (400 MHz, CDCl<sub>3</sub>)  $\delta$ : 8.94 (d,  $^3J_{HPt} = 27.0$ ,  $^3J_{HH} = 6$ , 2H, H<sub>o</sub>), 8.84 (dd,  $^3J_{HPt} = 27.0$ ,  $^3J_{HH} = 6$ , 2H, H<sub>o'</sub>), 8.10 (t,  $^3J_{HH} = 6$ , 1H, H<sub>p</sub>), 8.08 (t,  $^3J_{HH} = 6$ , 1H, H<sub>p'</sub>), 7.63 (m,  $^3J_{HH} = 6$ , 4H, H<sub>m</sub>), 3.91 (s, OCH<sub>3</sub>, 3H), 3.88 (s, OCH<sub>3</sub>, 3H).



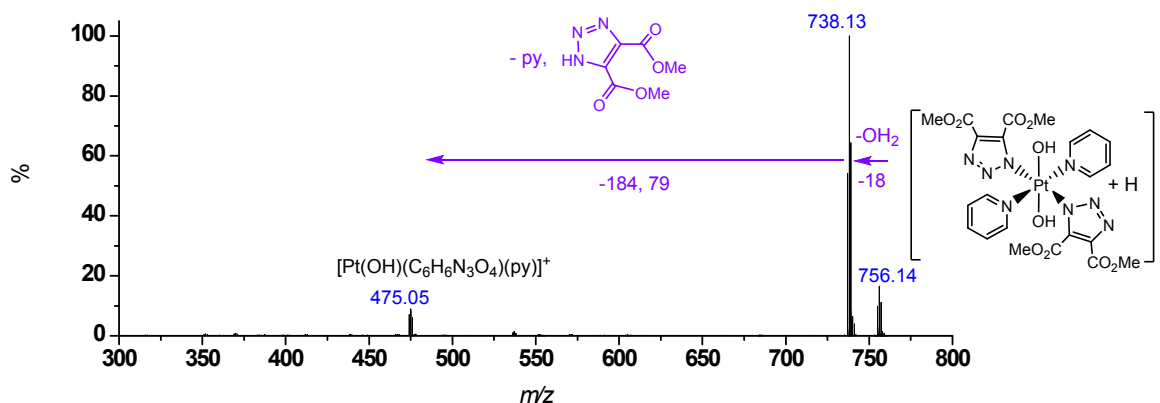
**Figure S1.**  $^{14}\text{N}$  NMR (43 MHz) ( $d_6$ -acetone) spectra of a)  $trans\text{-}[\text{Pt}(\text{N}_3)_2(\text{py})_2]$  starting material<sup>1</sup> and b) bis substituted  $trans\text{-}[\text{Pt}(\text{C}_6\text{H}_6\text{N}_3\text{O}_4)_2(\text{py})_2]$ .



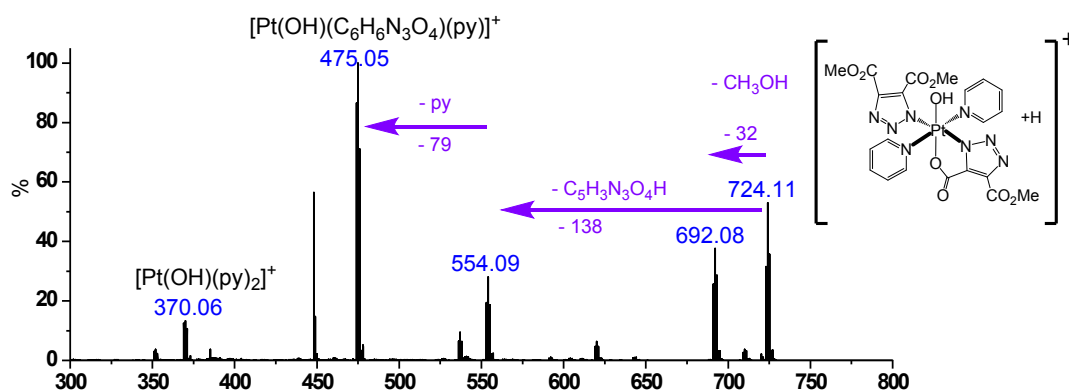
**Figure S2.**  $^{14}\text{N}$  NMR spectra of a) complex **1** ( $\text{D}_2\text{O}$ , 43 MHz) starting material<sup>1</sup> b) complex **2a'** ( $\text{D}_2\text{O}$ , 29 MHz) and c) complex **2a''** ( $\text{CDCl}_3$ , 28 MHz).



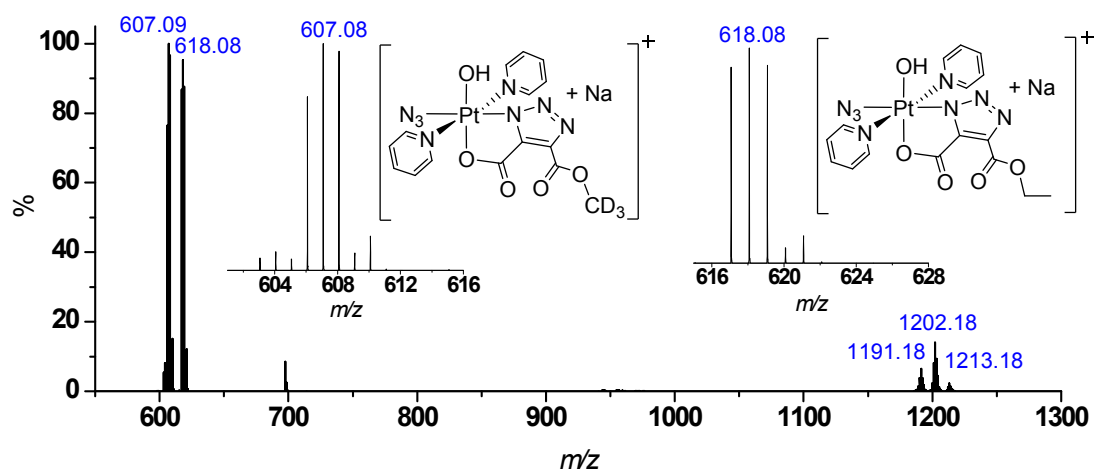
**Figure S3.** Solid-state IR spectra of a) *mono*-substituted  $2a'$  and b) *bis*-substituted  $2b'$ .



**Figure S4.** MS/MS of **2b**: *bis*-substituted DMAD complex *trans,trans,trans*- $[\text{Pt}(\text{C}_6\text{H}_6\text{N}_3\text{O}_4)_2(\text{OH})_2(\text{py})_2 + \text{H}]^+$ , 756.14  $m/z$  (model 756.13  $m/z$ ).

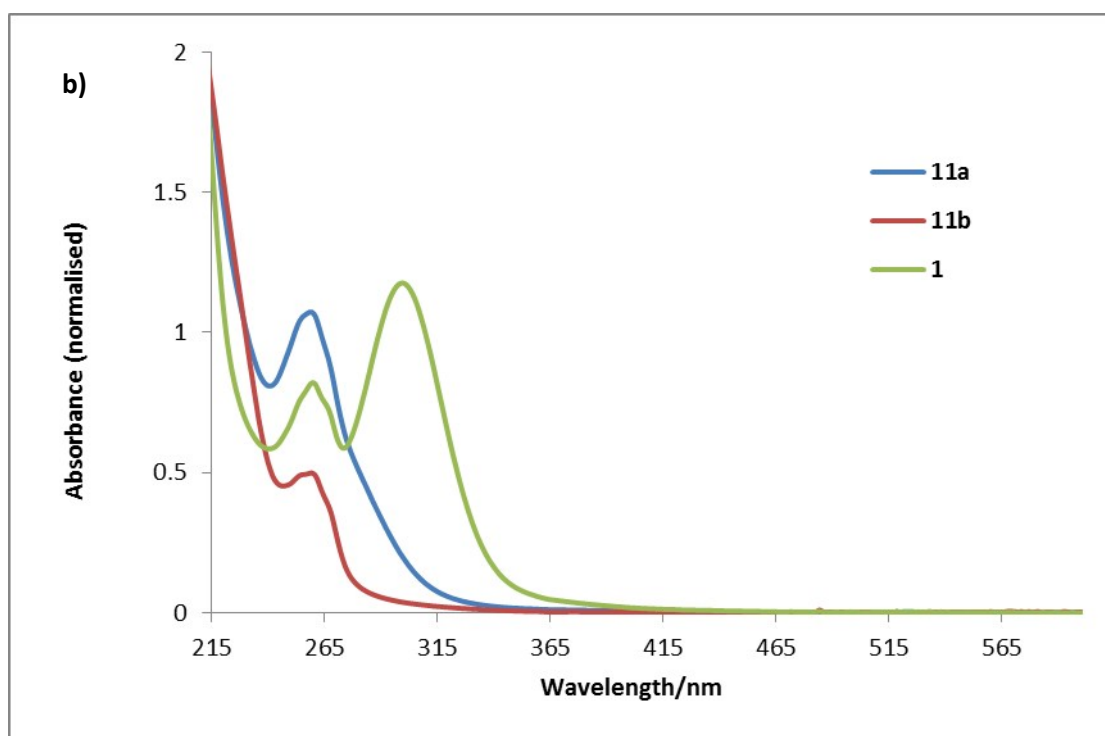
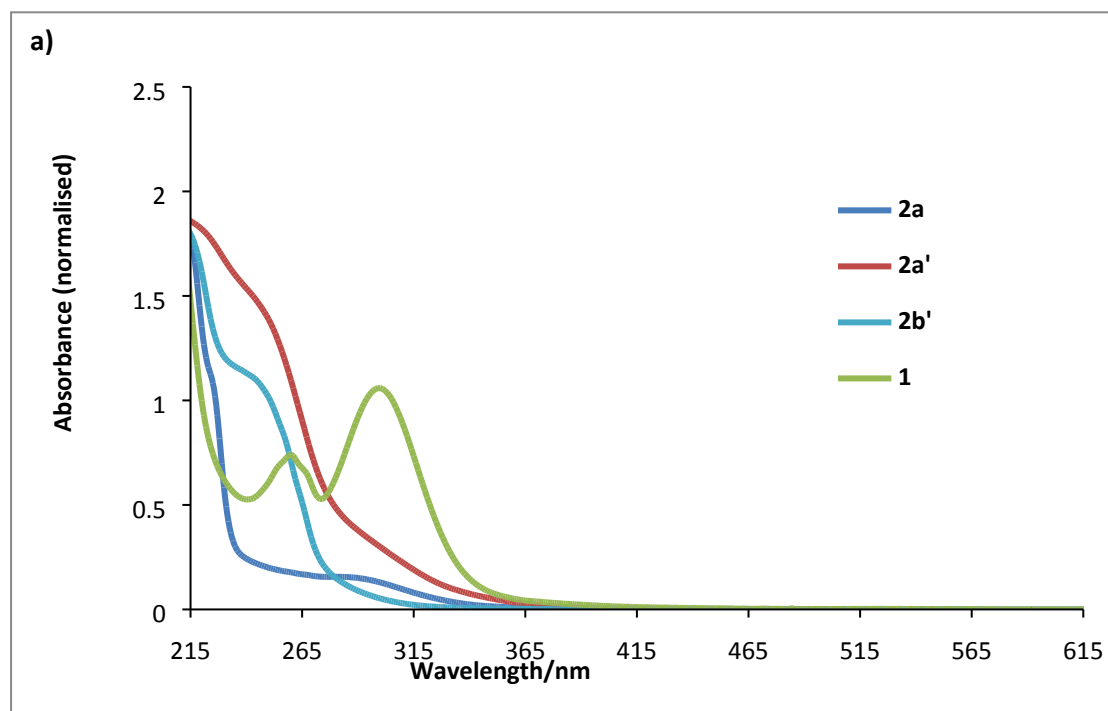


**Figure S5.** MS/MS of species intermediate between **2b** and **2b'**: *bis*-substituted, *mono*-axially tethered DMAD complex *trans,trans,trans*- $[\text{Pt}(\text{C}_6\text{H}_6\text{N}_3\text{O}_4)_2(\text{C}_5\text{H}_3\text{N}_3\text{O}_4)(\text{OH})(\text{py})_2 + \text{H}]^+$ , 724.11  $m/z$  (model 724.10  $m/z$ ).

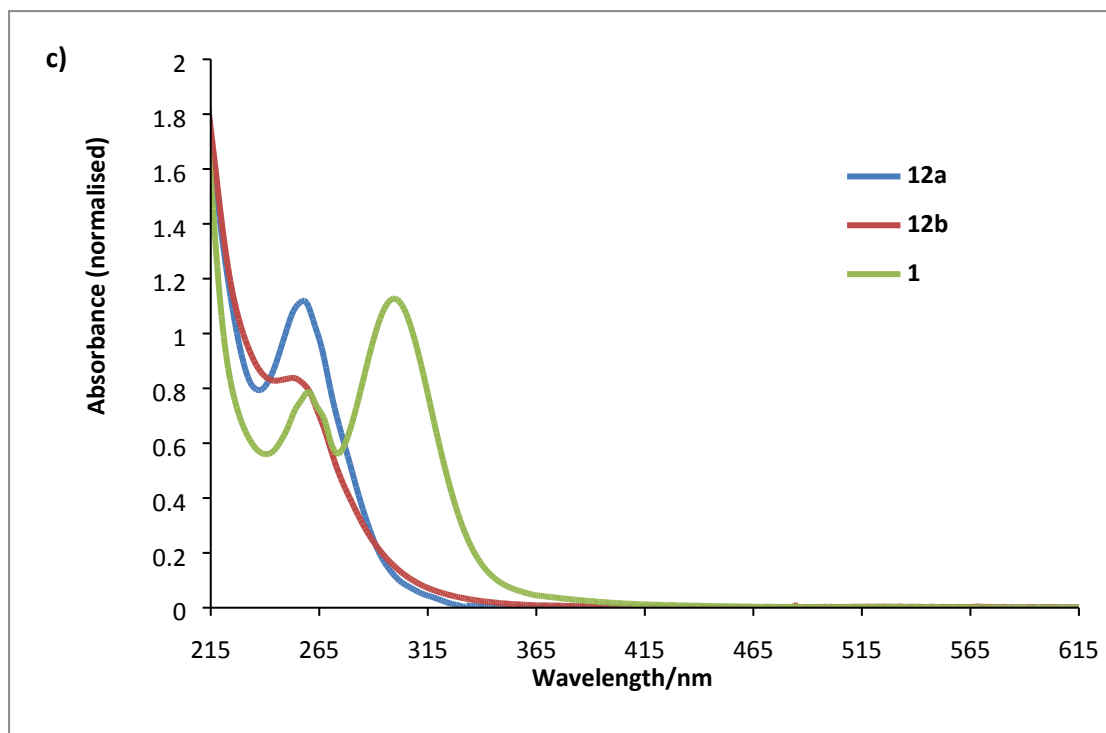


**Figure S6.** ESI-MS (MeCN) of crystalline DEACD product **3a'** from reaction in  $d_4$ -MeOH: a mixture of cyclometallated transesterified species  $[\text{Pt}(\text{py})_2(\text{N}_3)(\text{OH})(\text{N}_3\text{C}_5\text{O}_4\text{D}_3) + \text{Na}]^+$  at 607.09  $m/z$  (model 607.08  $m/z$ ) and cyclometallated product  $[\text{Pt}(\text{py})_2(\text{N}_3)(\text{OH})(\text{C}_6\text{H}_5\text{N}_3\text{O}_4) + \text{Na}]^+$  at 618.08  $m/z$  (model 618.08  $m/z$ ).

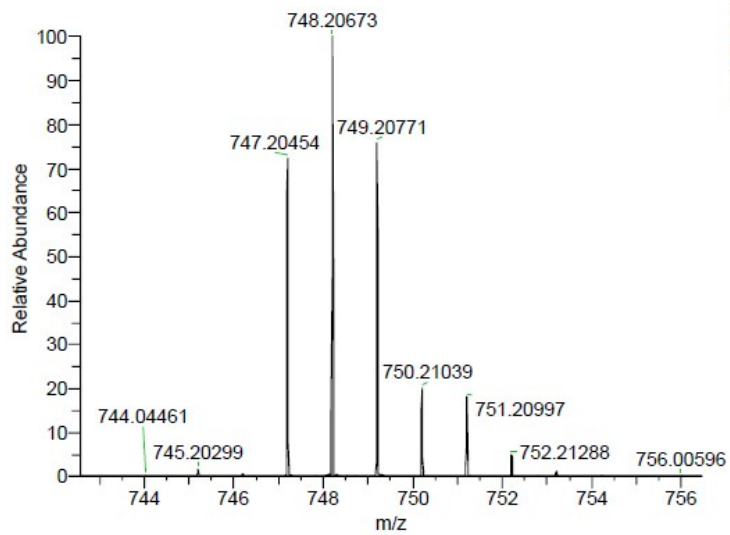
## UV-vis spectra



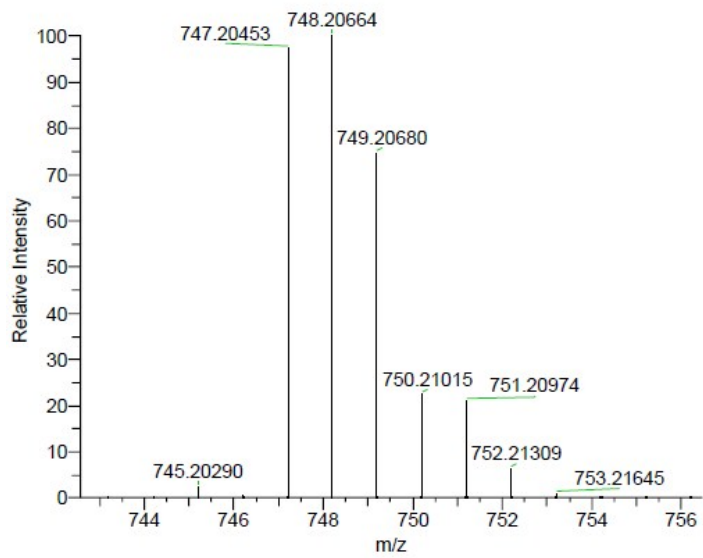




**Figure S7.** UV-Vis spectra of **1** and its derivatives in MeCN/H<sub>2</sub>O, with absorbance normalised at 211 nm for all species: a) DMAD (**2**) derivatives b) BCN (**11**) derivatives c) DBCO (**12**) derivatives

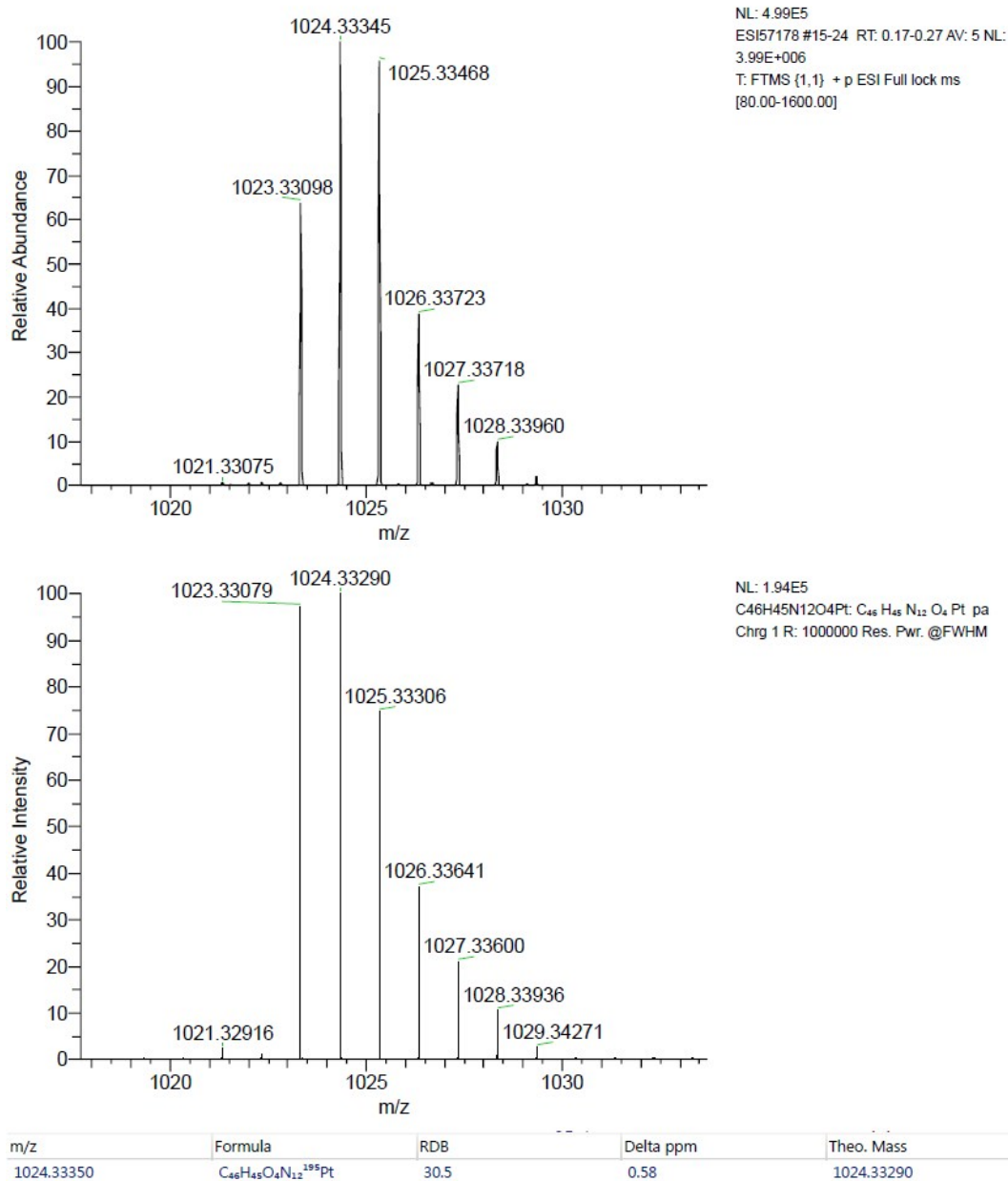


NL: 1.15E6  
 ESI67178 #15-24 RT: 0.17-0.27 AV: 5 NL  
 3.99E+006  
 T: FTMS {1,1} + p ESI Full lock ms  
 [80.00-1600.00]



NL: 2.39E5  
 C28H29N10O3Pt: C<sub>28</sub>H<sub>29</sub>N<sub>10</sub>O<sub>3</sub>Pt pa  
 Chrg 1 R: 1000000 Res. Pwr. @FWHM

m/z	Formula	RDB	Delta ppm	Theo. Mass
748.20673	C <sub>28</sub> H <sub>29</sub> O <sub>3</sub> N <sub>10</sub> <sup>195</sup> Pt	19.5	0.12	748.20664



**Figure S8.** HRMS of DBCO products **12a** (748.2067  $m/z$ , top) and **12b** (1024.3335  $m/z$ , bottom) with isotope predictions.

## References

- 1 N. J. Farrer, P. Gierth and P. J. Sadler, *Chem. Eur. J.*, 2011, **17**, 12059–66.