

ELECTRONIC SUPPLEMENTARY INFORMATION (ESI)

Title:

A 2D rhomboidal system of manganese(II) $[\text{Mn}(\text{3-MeC}_6\text{H}_4\text{COO})_2(\text{H}_2\text{O})_2]_n$ with spin canting: rationalization of the magnetic exchange

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Content:

Figure S1. Asymmetric unit of compound 1	2
Figure S2. Hydrogen bonds in the layer of compound 1	2
Figure S3. View of two layers of compound 1	3
Figure S4. Field dependence of the magnetization of compound 1	4
Figure S5. AC magnetic susceptibility graphs of compound 1	4
Figura S6. Density of states for antiferromagnetic ground state of compound 1	5
Table S1. X-ray crystallographic data details for compound 1	6

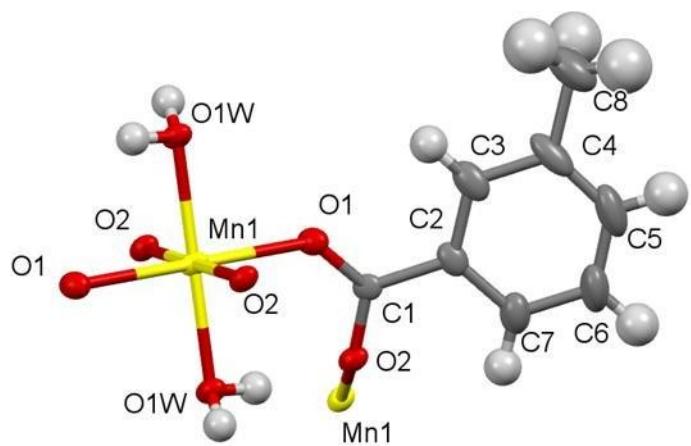


Figure S1. Asymmetric unit of compound 1.

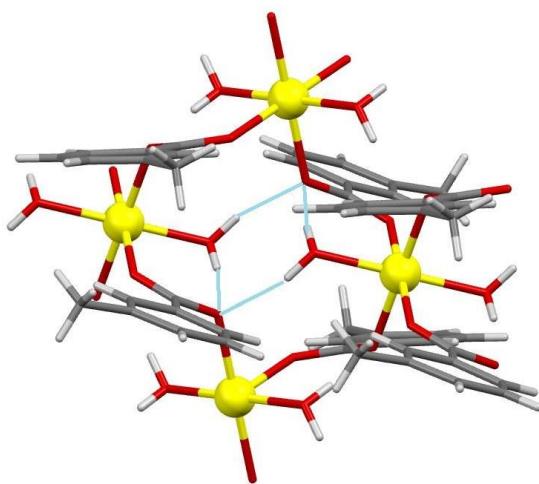


Figure S2. Hydrogen bonds between the water ligand and the carboxylate group.

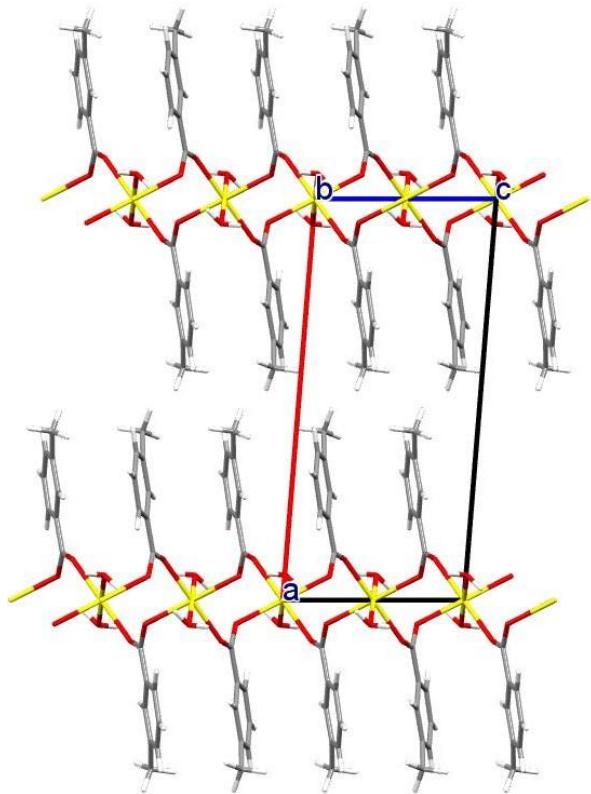


Figure S3. View down through crystallographic *b* axis of two layers of compound 1.

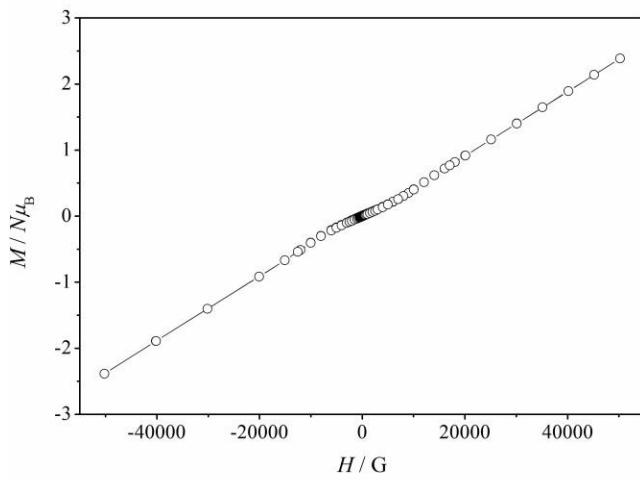


Figure S4. The whole field dependence of the magnetization of **1** between -50.2 to 50.2 kG at 2 K.

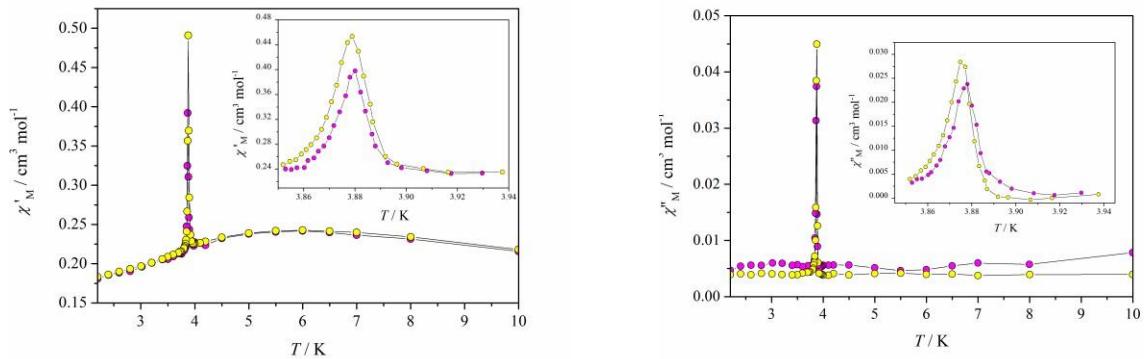


Figure S5. AC susceptibility measurements at 997 Hz (900 Hz in the inset) (yellow) and 10 Hz (magenta). Real (χ'_M) and imaginary (χ''_M) AC susceptibility components are shown in the left and right respectively.

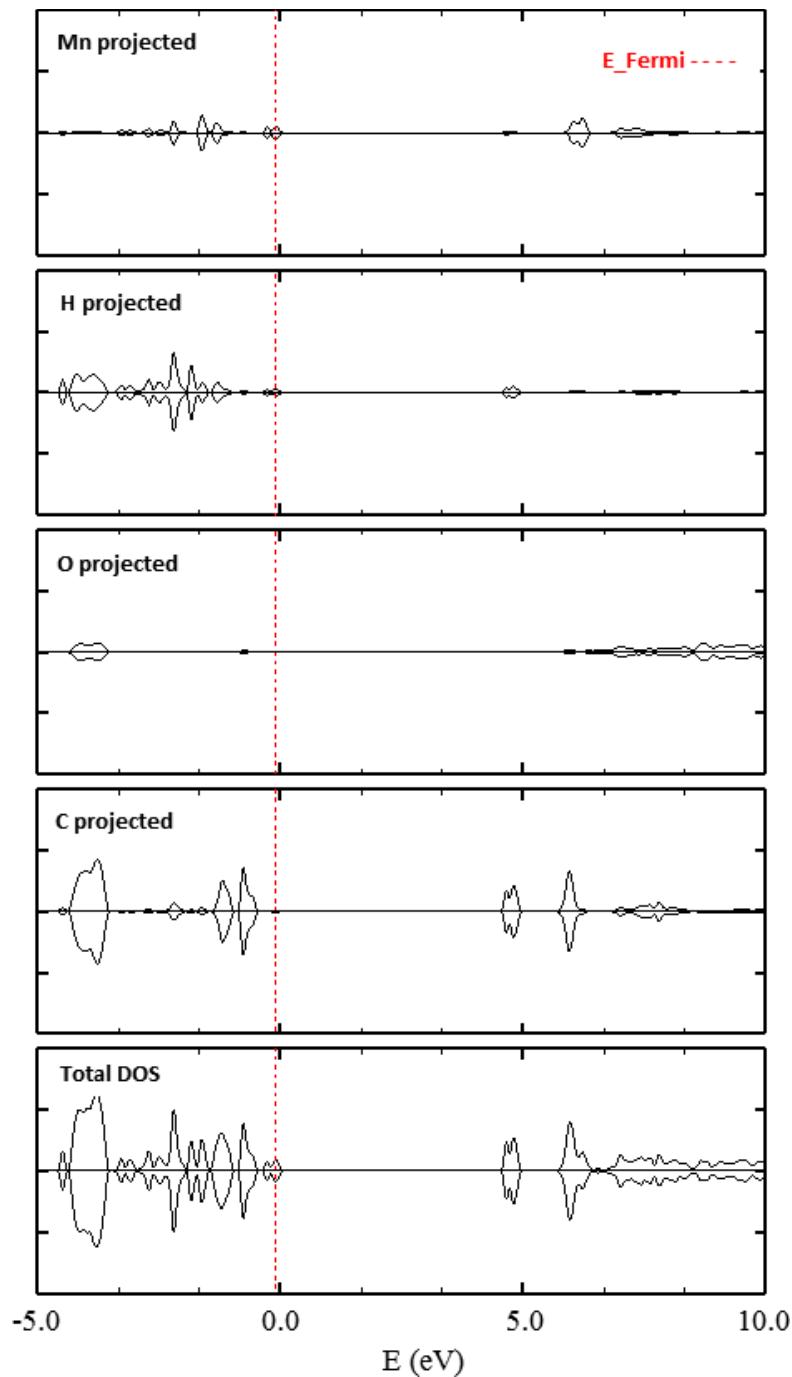


Figure S6. Total and atom projected density of states (DOS, arb. units) for AF ground state of compound **1** obtained using B3LYP functional and all electron basis sets. $E_{\text{F}} \text{ is set at } 0.0 \text{ eV}$.

Table S1. X-ray Crystallographic Data for compound crystallographic data collection and structure refinement details for compound [Mn(3-MeC₆H₄COO)₂(H₂O)₂] (**1**).

chemical formula	C ₁₆ H ₁₈ MnO ₆
formula weight /g mol ⁻¹	361.24
T / K	100(2)
λ (Mo Kα) / Å	0.71073
crystal system	P2 ₁ /c (No. 14)
space group	Monoclinic
crystal size / mm	0.1 x 0.1 x 0.2
a / Å	16.057(2)
b / Å	6.8909(9)
c / Å	7.2259(10)
β / deg.	94.608(5)
V / Å ³	796.94(18)
Z	2
ρ _{calcd} / g cm ⁻³	1.505
μ / mm ⁻¹	0.856
F(000)	374
Θ range / deg.	2.5 to 24.7
limiting indices	<i>h</i> = -18→18, <i>k</i> = -8→8, <i>l</i> = -7→8
data / restraints / parameters	1361 / 103 / 143
goodness-of-fit on <i>F</i> ²	1.081
final <i>R</i> indices [<i>I</i> >2σ(<i>I</i>)]	<i>R</i> _{1a} = 0.0425, <i>wR</i> ₂ = 0.1016
<i>R</i> indices (all data)	<i>R</i> _{1a} = 0.0589, <i>wR</i> ₂ = 0.1130

^a *R*1 = $\sum(|F_o| - |F_c|)/\sum|F_o|$. ^b *wR*2 = $\{\sum[\omega(F_o^2 - F_c^2)^2] / \sum[\omega(F_o^2)^2]\}^{1/2}$, $\omega = 1/[d^2(F_o^2) + (0.0675P)^2 + 1.4805P]$, where $P = (F_o^2 + 2F_c^2) / 3$.