

Supporting Information:

Ligand modification on UiO-66 with unusual visible-light photocatalytic behavior for RhB degradation

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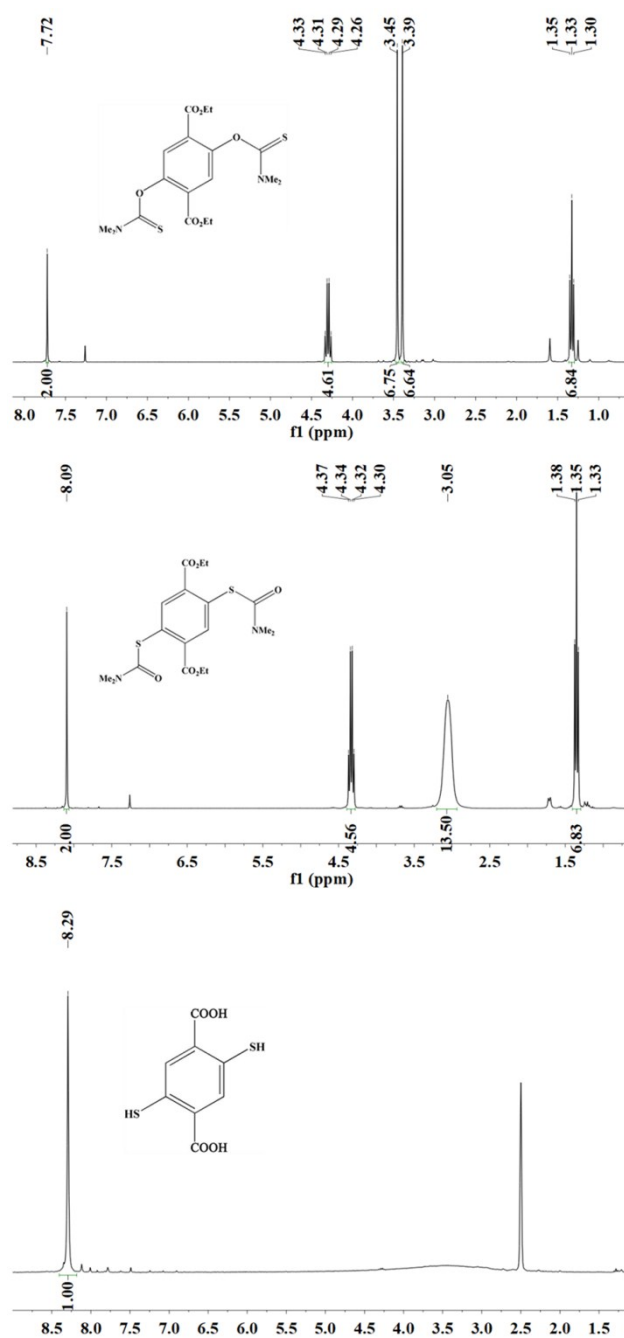


Figure S1. ^1H NMR spectra of synthesized $\text{H}_2\text{BDC-2,5SH}$ and two intermediates.

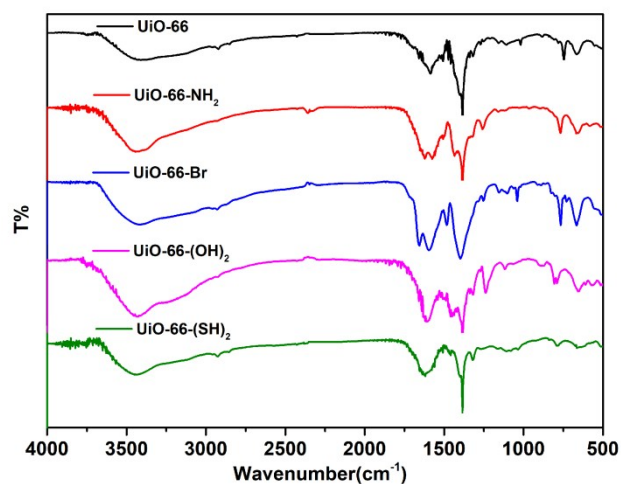


Figure S2. IR spectra of the UiO-66-X (X = H, NH₂, Br, (OH)₂, (SH)₂)

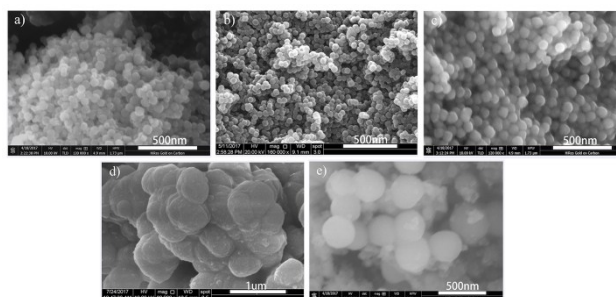


Figure S3. SEM images of the a) UiO-66, b) UiO-66-NH₂, c) UiO-66-Br, d) UiO-66-(OH)₂, e) UiO-66-(SH)₂

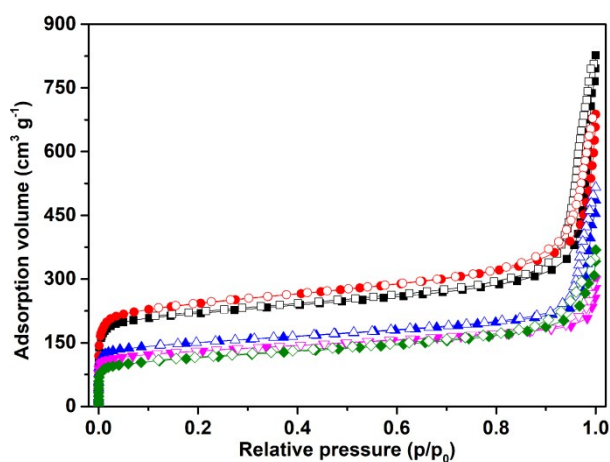


Figure S4. The N₂ adsorptions isotherms of UiO-66 (black), UiO-66-NH₂ (red), UiO-66-Br (blue), UiO-66-(OH)₂ (magenta), UiO-66-(SH)₂ (olive); Solid and hollow data correspond to the adsorption and desorption branches, respectively.

Table S1. Porosity properties of the UiO-66-X (X = H, NH₂, Br, (OH)₂, (SH)₂).

	UiO-66	UiO-66-NH ₂	UiO-66-Br	UiO-66-(OH) ₂	UiO-66-(SH) ₂
BET surface area (m ² g ⁻¹)	835	917	563	482	418
Pore volume (cm ³ g ⁻¹)	0.56	0.60	0.36	0.29	0.33

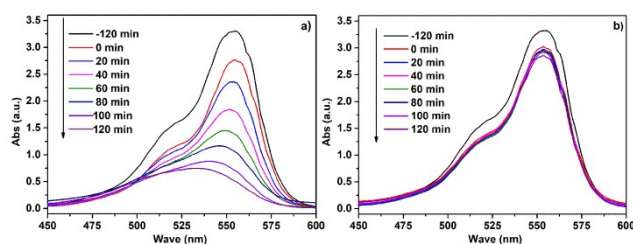


Figure S5. Absorption spectra of RhB solution in different irradiation time under visible light (400-780nm). a) 15mg UiO-66 in 100 mL 20 mg L⁻¹ RhB aqueous solution. b) 15mg UiO-66-NH₂ in 100 mL 20 mg L⁻¹ RhB aqueous solution.

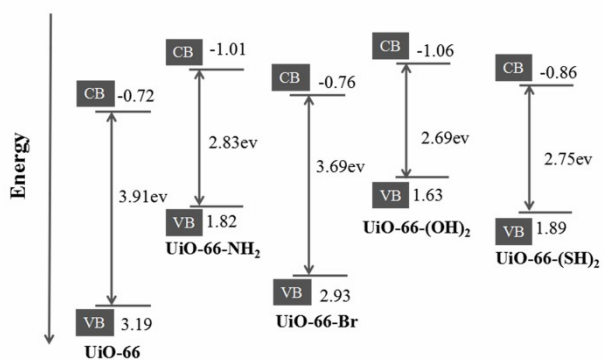


Figure S6. Diagram for the potential of UiO-66-X (X = H, NH₂, Br, (OH)₂, (SH)₂).

Table S2. The potential (V_{fb} , V_{CB} and V_{VB}) of the UiO-66-X (X = H, NH_2 , Br, $(OH)_2$, $(SH)_2$).

Potential (V) \ Sample	UiO-66	UiO-66- NH_2	UiO-66-Br	UiO-66- $(OH)_2$	UiO-66- $(SH)_2$
Flat band (V_{fb})	-0.83	-1.12	-0.87	-1.17	-0.97
Conduction band (V_{CB})	-0.72	-1.01	-0.76	-1.06	-0.86
Valence band (V_{VB})	3.19	1.82	2.93	1.63	1.89

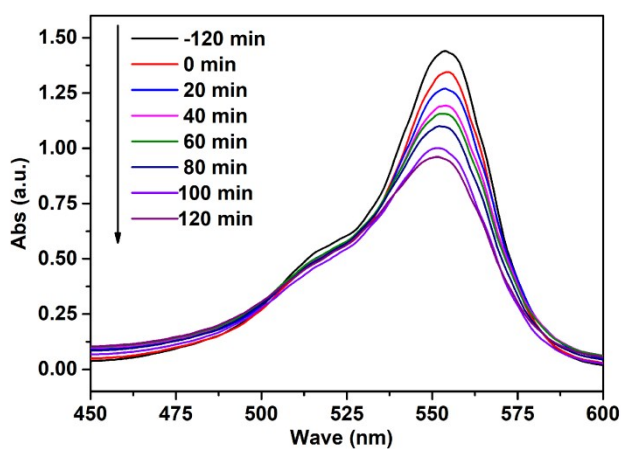


Figure S7. Absorption spectra of RhB in different irradiation time under visible light (400-780 nm). Reaction solution: 20 mg UiO-66- NH_2 in 100 mL 10 mg L^{-1} RhB aqueous solution.

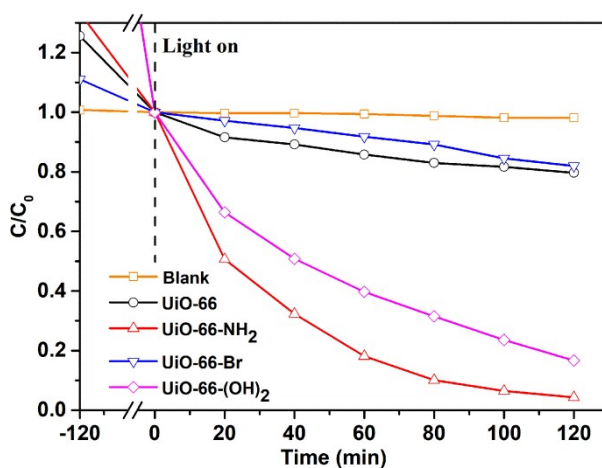


Figure S8. Photocatalytic reaction of Cr(VI) solution in the presence and absence (blank) of catalysts under visible-light irradiation (400-780 nm). Reaction condition: 15 mg catalyst in 100 mL 20 mg L^{-1} Cr(VI) solution, PH=2.