

Electronic Supplementary Information

Triphenylbismuth(V) di[(iso)nicotines] - transmetallation agents or divergent organometallogoligands? First organobismuth(V)-based silver(I) coordination polymers

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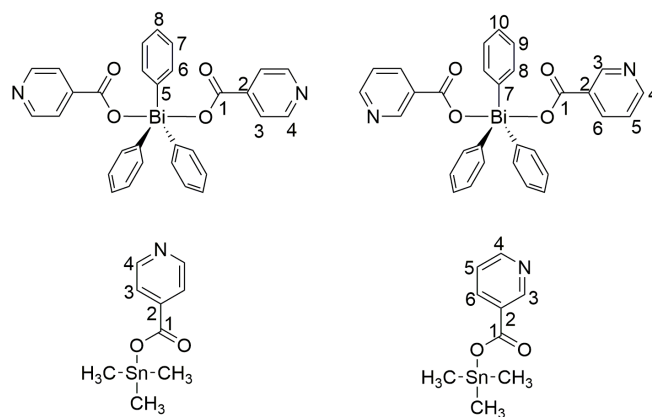
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Scheme S1. Numbering schemes for the assignment of resonances in NMR spectra

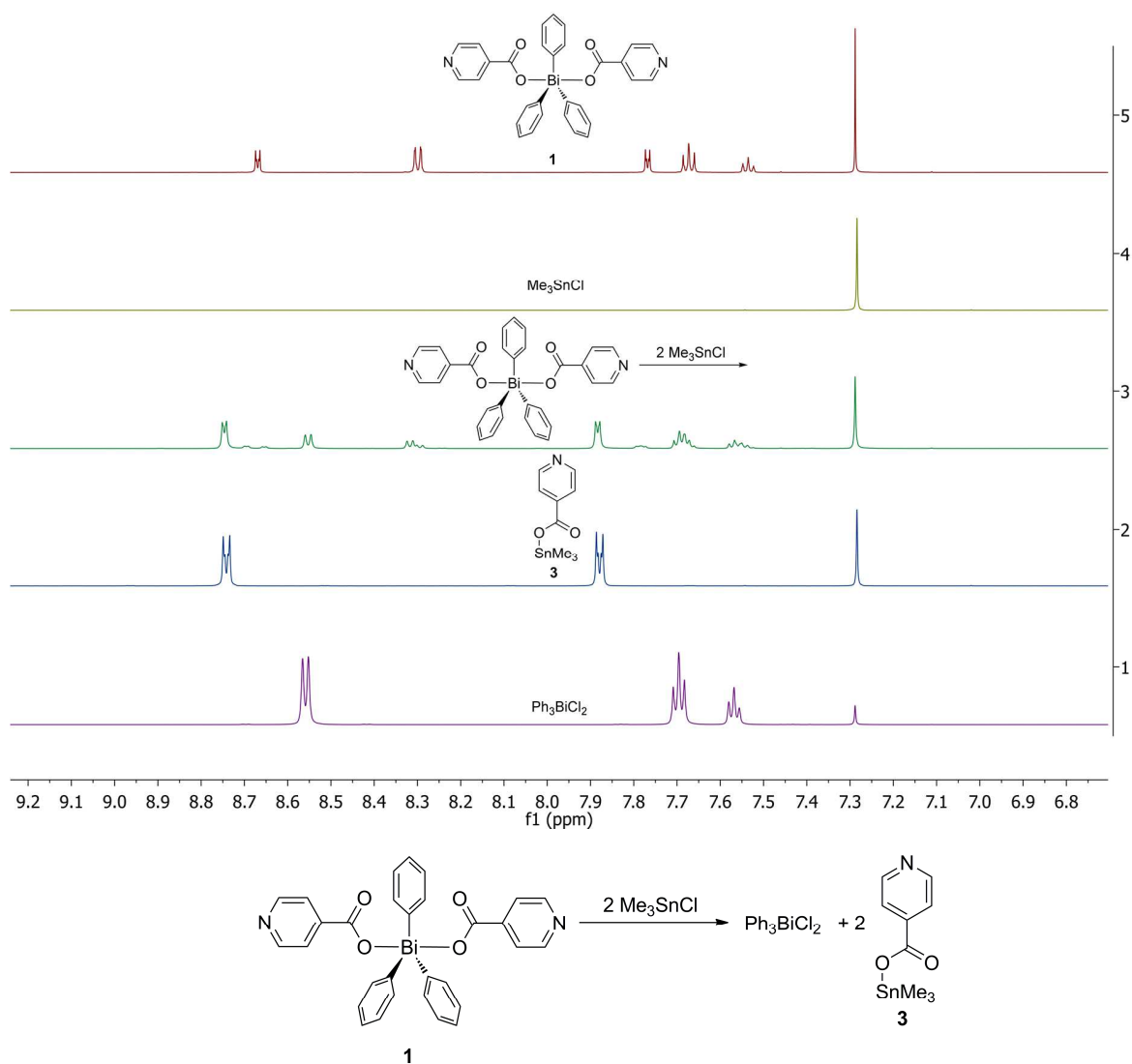


Figure S1. Aromatic region of superimposed ^1H NMR spectra (CDCl₃, 20 °C) of pure $\text{Ph}_3\text{Bi}[\text{O}(\text{O})\text{CC}_5\text{H}_4\text{N}-4]_2$ (**1**) (spectrum 5), pure Me_3SnCl (spectrum 4), the crude product isolated from the reaction between these two organometallic species (spectrum 3), pure $\text{Me}_3\text{Sn}[\text{O}(\text{O})\text{CC}_5\text{H}_4\text{N}-4]$ (**3**) (spectrum 2), and pure Ph_3BiCl_2 (spectrum 1), illustrating the transmetalation reaction taking place according to the depicted scheme.

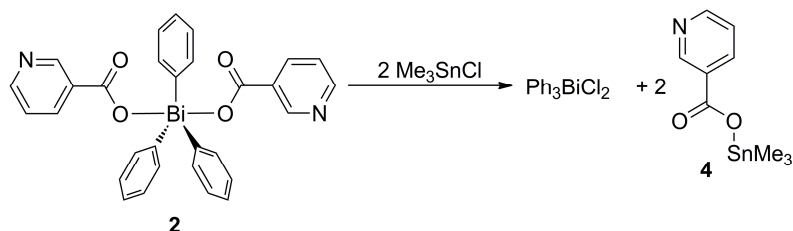
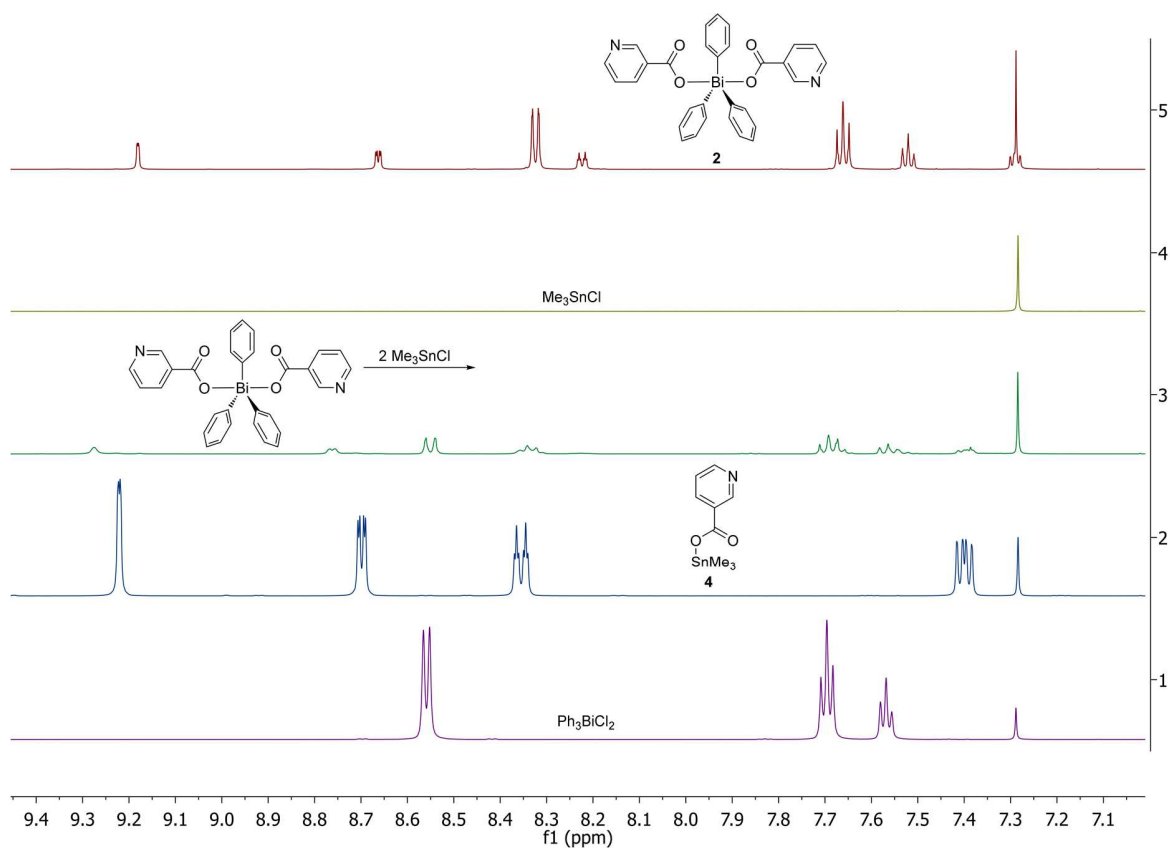


Figure S2. Aromatic region of superimposed ^1H NMR spectra (CDCl_3 , 20°C) of pure $\text{Ph}_3\text{Bi}[\text{O}(\text{O})\text{CC}_5\text{H}_4\text{N}-3]_2$ (**2**) (spectrum 5), pure Me_3SnCl (spectrum 4), the crude product isolated from the reaction between these two organometallic species (spectrum 3), pure $\text{Me}_3\text{Sn}[\text{O}(\text{O})\text{CC}_5\text{H}_4\text{N}-3]$ (**4**) (spectrum 2), and pure Ph_3BiCl_2 (spectrum 1), illustrating the transmetalation reaction taking place according to the depicted scheme.

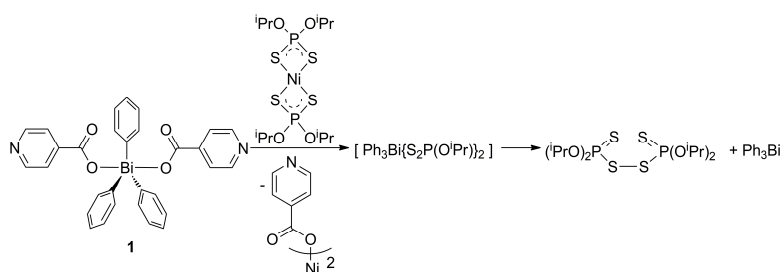
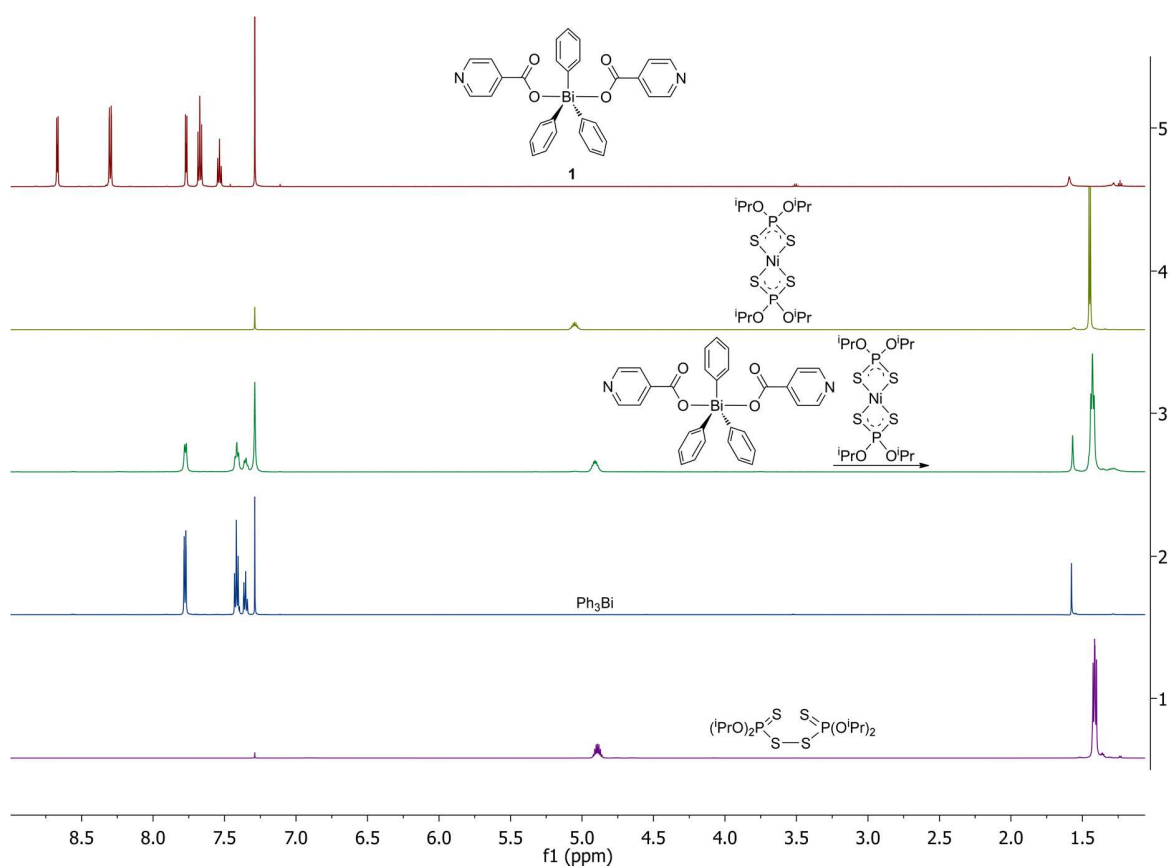


Figure S3. Superimposed ^1H NMR spectra (CDCl_3 , 20°C) of pure $\text{Ph}_3\text{Bi}[\text{O}(\text{O})\text{CC}_5\text{H}_4\text{N}-4]_2$ (**1**) (spectrum 5), pure $\text{Ni}[\text{S}_2\text{P}(\text{O}^i\text{Pr})_2]_2$ (spectrum 4), the crude product isolated from the reaction between these two compounds (spectrum 3), pure Ph_3Bi (spectrum 2), and pure $[(^i\text{PrO})_2\text{P}(\text{S})\text{S}]_2$ (spectrum 2), illustrating the transmetalation reaction taking place according to the depicted scheme.

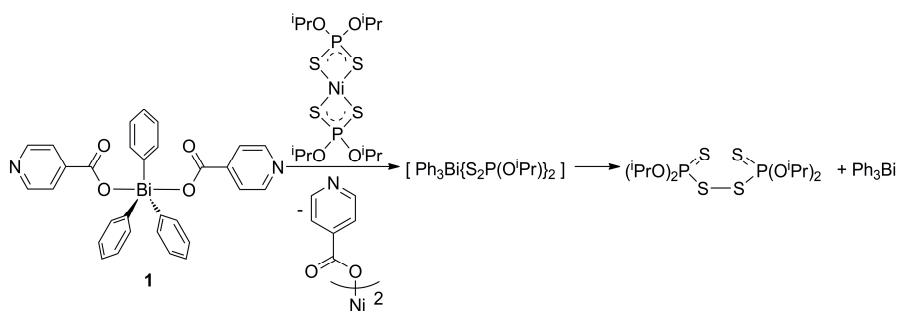
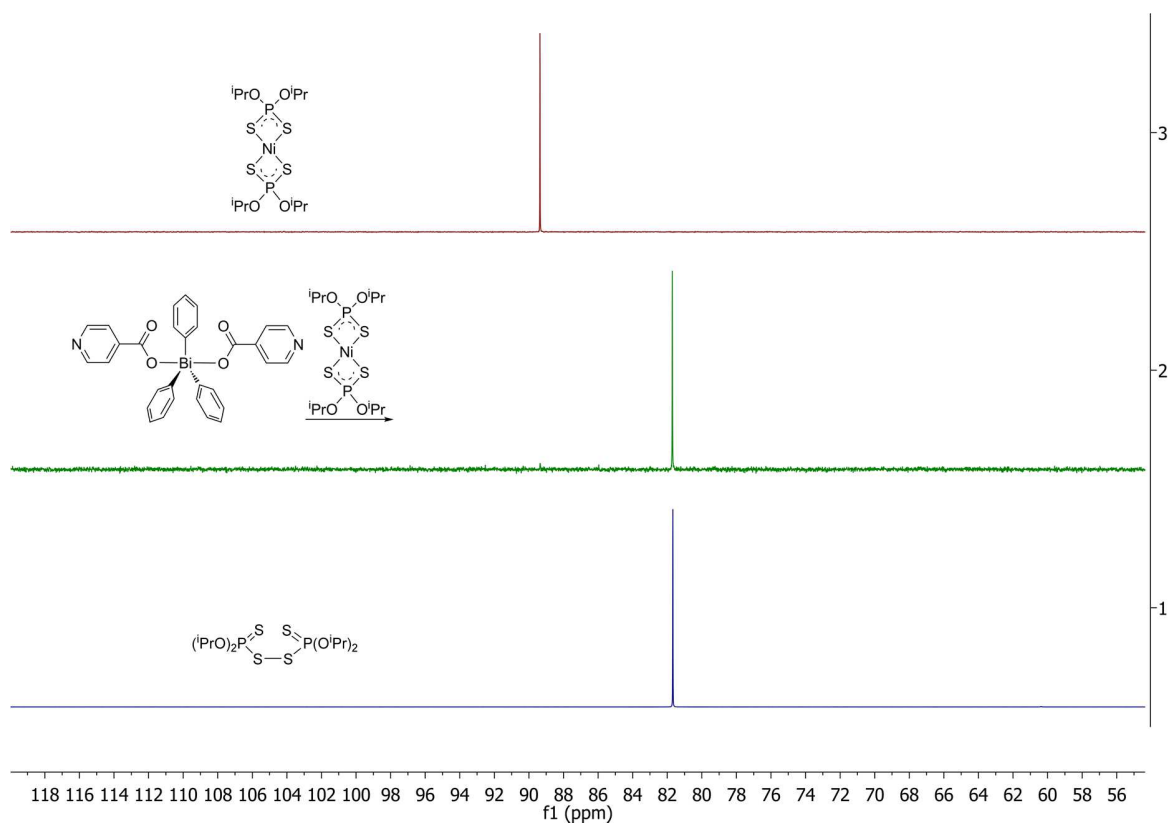


Figure S4. Superimposed ^{31}P NMR spectra (CDCl_3 , 20°C) of pure $\text{Ni}[\text{S}_2\text{P}(\text{O}^i\text{Pr})_2]_2$ (spectrum 3), the crude product isolated from the reaction between **1** and $\text{Ni}[\text{S}_2\text{P}(\text{O}^i\text{Pr})_2]_2$ (spectrum 2), and pure $[(\text{iPrO})_2\text{P}(\text{S})\text{S}]_2$ (spectrum 1), illustrating the transmetalation reaction taking place according to the depicted scheme.

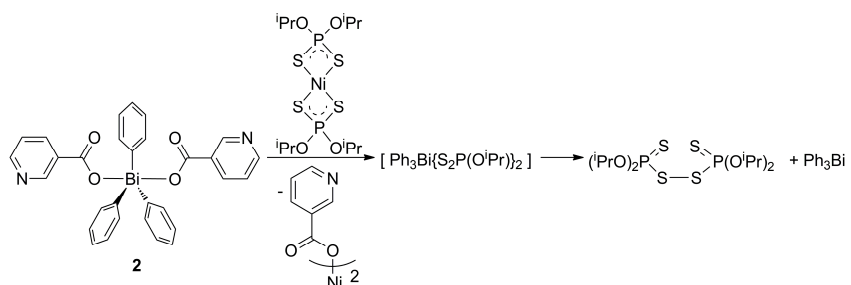
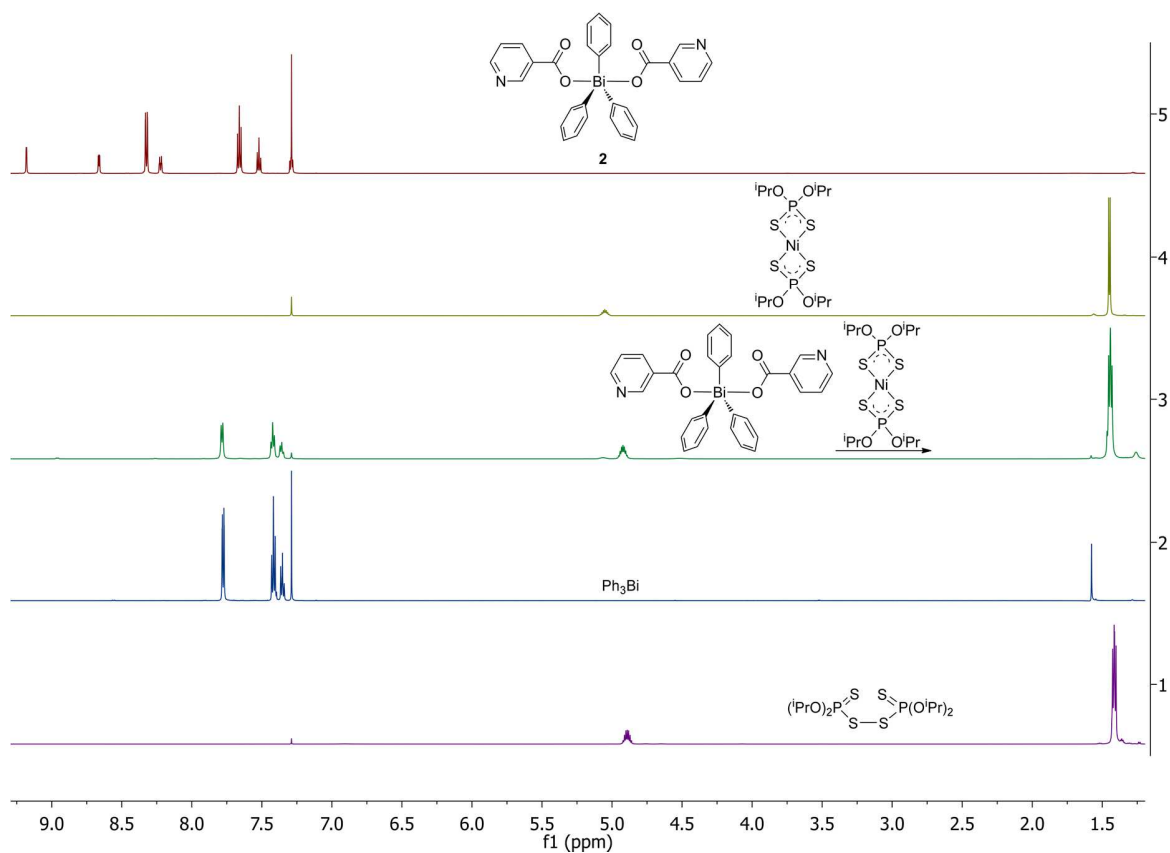


Figure S5. Superimposed ¹H NMR spectra (CDCl₃, 20 °C) of pure Ph₃Bi[O(O)CC₅H₄N-3] (**2**) (spectrum 5), pure Ni[S₂P(OⁱPr)₂]₂ (spectrum 4), the crude product isolated from the reaction between these two compounds (spectrum 3), pure Ph₃Bi (spectrum 2), and pure [(ⁱPrO)₂P(S)S]₂ (spectrum 2), illustrating the transmetalation reaction taking place according to the depicted scheme.

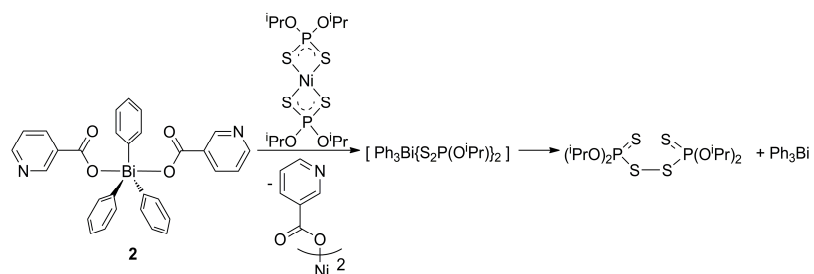
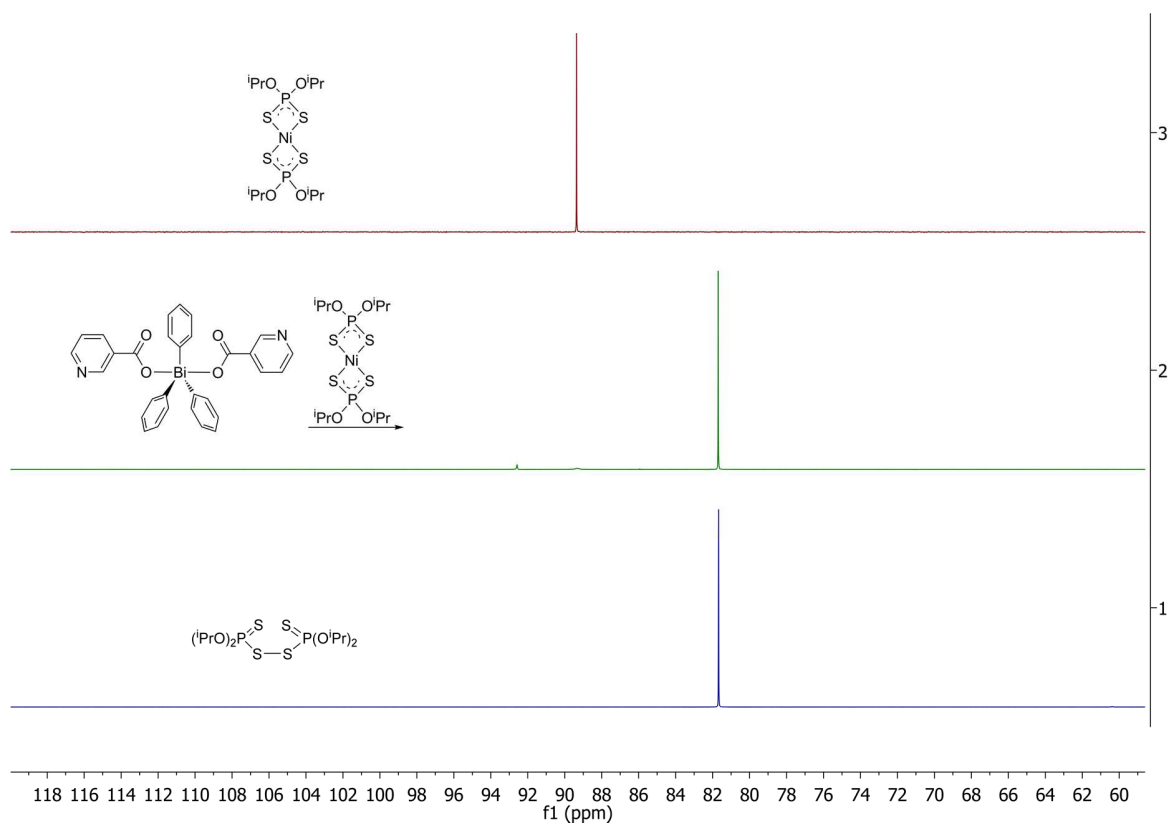


Figure S6. Superimposed ^{31}P NMR spectra (CDCl₃, 20 °C) of pure Ni[S₂P(OiPr)₂]₂ (spectrum 3), the crude product isolated from the reaction between **2** and Ni[S₂P(OiPr)₂]₂ (spectrum 2), and pure [(iPrO)₂P(S)S]₂ (spectrum 1), illustrating the transmetalation reaction taking place according to the depicted scheme.

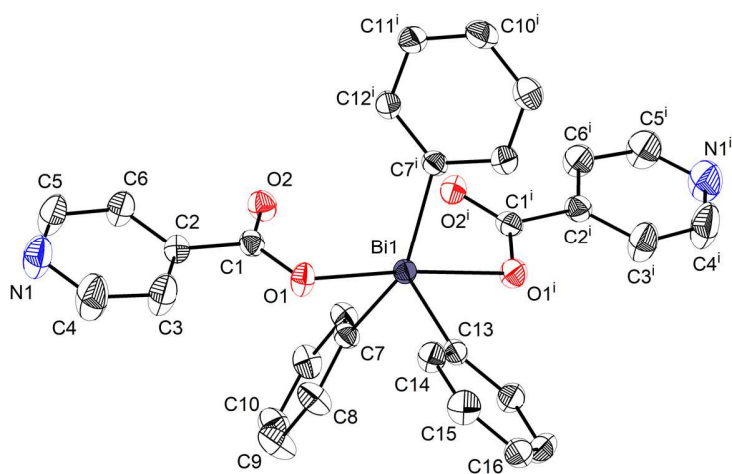


Figure S7. Molecular structure of **1**. Thermal ellipsoids are drawn at the 30% probability. Hydrogen atoms are omitted for clarity [symmetry code: (i) $1.5-x, y, 0.5-z$].

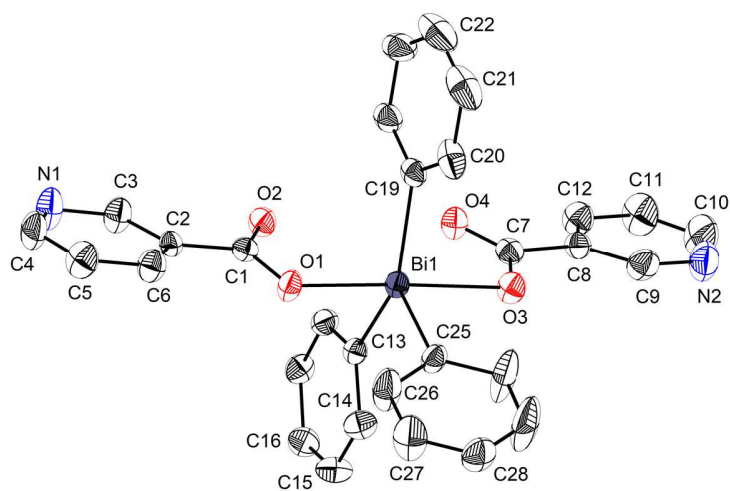


Figure S8. Molecular structure of **2**. Thermal ellipsoids are drawn at the 30% probability. Hydrogen atoms are omitted for clarity.

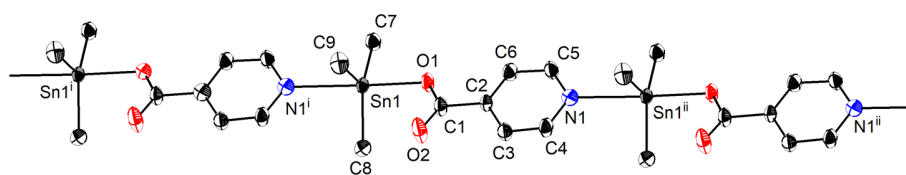


Figure S9. Coordination polymer in the crystal of **3**. Thermal ellipsoids are drawn at the 30% probability. Hydrogen atoms are omitted for clarity [symmetry codes: (i) $-1+x, 1+y, z$; (ii) $1+x, -1+y, z$].

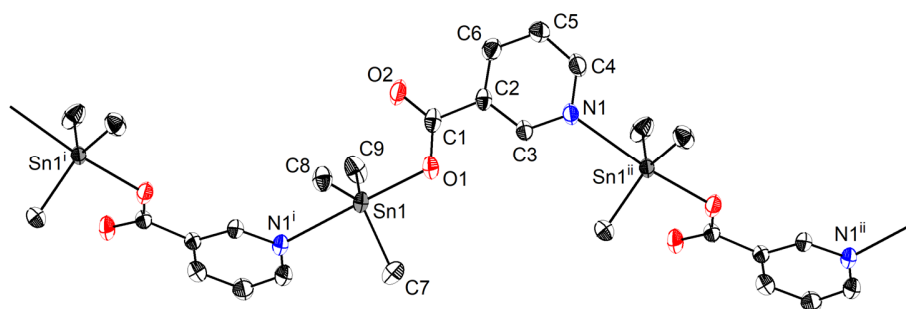


Figure S10. Coordination polymer in the crystal of **4**. Thermal ellipsoids are drawn at the 30% probability. Hydrogen atoms are omitted for clarity [symmetry codes: (i) $x, -y, 0.5+z$; (ii) $x, -y, -0.5+z$].

Table S1. X-ray crystal data and structure refinement for compounds **1–4**.

	1	2	3	4
Empirical formula	C ₃₀ H ₂₃ BiN ₂ O ₄	C ₃₀ H ₂₃ BiN ₂ O ₄	C ₉ H ₁₃ NO ₂ Sn	C ₉ H ₁₃ NO ₂ Sn
Formula weight	684.48	684.48	285.89	285.89
Temperature (K)	297(2)	294(2)	297(2)	294(2)
Crystal system	Monoclinic	Triclinic	Triclinic	Monoclinic
Space group	<i>P2/n</i>	<i>P</i> -1	<i>P</i> -1	<i>C2/c</i>
<i>a</i> (Å)	12.9965(12)	9.0030(12)	6.6686(9)	24.937(10)
<i>b</i> (Å)	9.7492(9)	11.5252(15)	8.9159(12)	7.100(3)
<i>c</i> (Å)	13.4933(13)	12.9642(17)	10.1287(13)	15.414(6)
α (°)	90.00	95.401(2)	79.674(2)	90.00
β (°)	105.707(2)	93.536(2)	72.196(2)	124.647(5)
γ (°)	90.00	101.254(2)	75.429(2)	90.00
Volume (Å ³)	1645.8(3)	1309.0(3)	551.50(13)	2245.1(16)
<i>Z</i>	2	2	2	8
<i>D</i> _{calc} (g cm ⁻³)	1.381	1.737	1.722	1.692
Absorption coefficient (mm ⁻¹)	5.387	6.773	2.287	2.247
<i>F</i> (000)	664	664	280	1120
Crystal size (mm)	0.23x0.25x0.29	0.33x0.39x0.40	0.28x0.33x0.39	0.23x0.28x0.37
ϑ range for data collection (°)	1.93 to 25.01	1.58 to 24.99	2.12 to 24.99	1.98 to 26.37
Reflections collected	15123	12304	5081	11208
Independent reflections	2906	4564	1885	2282
	[<i>R</i> _{int} = 0.0565]	[<i>R</i> _{int} = 0.0516]	[<i>R</i> _{int} = 0.0210]	[<i>R</i> _{int} = 0.0324]
Absorption correction	Multi-Scan ¹	Multi-Scan ¹	Multi-Scan ¹	Multi-Scan ¹
Data / restraints / parameters	2906 / 0 / 169	4564 / 24 / 334	1885 / 0 / 118	2282 / 0 / 119
Goodness-of-fit on <i>F</i> ²	1.071	0.973	1.087	1.111
Final <i>R</i> indices [<i>I</i> > 2 σ (<i>I</i>)]	<i>R</i> ₁ = 0.0270 <i>wR</i> ₂ = 0.0641	<i>R</i> ₁ = 0.0318 <i>wR</i> ₂ = 0.0572	<i>R</i> ₁ = 0.0225 <i>wR</i> ₂ = 0.0569	<i>R</i> ₁ = 0.0266 <i>wR</i> ₂ = 0.0619
<i>R</i> indices (all data)	<i>R</i> ₁ = 0.0297 <i>wR</i> ₂ = 0.0650	<i>R</i> ₁ = 0.0408 <i>wR</i> ₂ = 0.0597	<i>R</i> ₁ = 0.0243 <i>wR</i> ₂ = 0.0579	<i>R</i> ₁ = 0.0298 <i>wR</i> ₂ = 0.0634
Largest difference peak and hole (e Å ⁻³)	0.35 and -1.44	0.92 and -0.95	0.41 and -0.38	0.32 and -0.45
CCDC No.	1585633	1585634	1585635	1585636

¹ G. M. Sheldrick, *SADABS, Program for area detector adsorption correction*, Institute for Inorganic Chemistry, University of Göttingen, Germany, 1996.

Table S2. X-ray crystal data and structure refinement for coordination polymers 5–9.

	5	6·CH ₂ Cl ₂	7·2THF	8·CH ₂ Cl ₂	9·CH ₂ Cl ₂
Empirical formula	C ₃₁ H ₂₃ AgBiF ₃ N ₂ O ₇ S	C ₆₄ H ₅₀ Ag ₂ Bi ₂ Cl ₄ F ₆ N ₄ O ₁₄ S ₂	C ₃₈ H ₃₉ AgBiF ₆ N ₂ O ₆ S ₂ Sb	C ₃₁ H ₂₅ AgBiCl ₂ F ₆ N ₂ O ₄ Sb	C ₃₁ H ₂₅ AgBiCl ₂ N ₃ O ₇
Formula weight	941.42	2052.70	1172.31	1113.03	939.29
Temperature (K)	160.0(1)	130.1(2)	170(2)	199.9(1)	293(2)
Crystal system	Monoclinic	Monoclinic	Monoclinic	Triclinic	Monoclinic
Space group	<i>P</i> 2 ₁ / <i>c</i>	<i>P</i> 2 ₁ / <i>n</i>	<i>P</i> 2 ₁ / <i>n</i>	<i>P</i> -1	<i>P</i> 2 ₁ / <i>c</i>
<i>a</i> (Å)	13.6678(5)	12.3800(4)	16.1795(6)	10.5410(6)	11.0075(15)
<i>b</i> (Å)	15.1649(10)	19.8689(8)	8.7028(4)	10.8892(5)	16.9543(16)
<i>c</i> (Å)	15.0978(6)	28.0926(9)	29.2014(12)	16.7211(7)	17.883(2)
α (°)	90.00	90.00	90.00	83.744(4)	90.00
β (°)	102.116(4)	90.574(3)	98.001(3)	71.929(4)	98.652(11)
γ (°)	90.00	90.00	90.00	76.436(4)	90.00
Volume (Å ³)	3059.6(3)	6909.8(4)	4071.7(3)	1772.38(15)	3299.5(7)
<i>Z</i>	4	4	4	2	4
<i>D</i> _{calc} (g cm ⁻³)	2.044	1.973	1.912	2.086	1.891
Absorption coefficient (mm ⁻¹)	6.522	5.934	5.519	6.476	6.131
<i>F</i> (000)	1808	3952	2256	1052	1808
Crystal size (mm)	0.15x0.20x0.40	0.05x0.10x0.15	0.10x0.17x0.22	0.03x0.10x0.35	0.08x0.12x0.17
ϑ range for data collection (°)	1.93 to 25.03	1.79 to 25.03	2.31 to 25.00	1.92 to 25.03	2.22 to 25.00
Reflections collected	14322	29329	23210	12588	34784
Independent reflections	5402	12047	7144	6253	5798
	[<i>R</i> _{int} = 0.0431]	[<i>R</i> _{int} = 0.0529]	[<i>R</i> _{int} = 0.1492]	[<i>R</i> _{int} = 0.0520]	[<i>R</i> _{int} = 0.2068]
Absorption correction	Multi-Scan ¹	Multi-Scan ¹	Multi-Scan ¹	Multi-Scan ¹	Multi-Scan ¹
Data / restraints / parameters	5402 / 0 / 415	12047 / 66 / 858	7144 / 1 / 496	6253 / 30 / 436	5798 / 0 / 406
Goodness-of-fit on <i>F</i> ²	1.034	1.097	0.962	1.035	0.765
Final <i>R</i> indices [<i>I</i> > 2σ(<i>I</i>)]	<i>R</i> ₁ = 0.0367 <i>wR</i> ₂ = 0.0732	<i>R</i> ₁ = 0.0665 <i>wR</i> ₂ = 0.1365	<i>R</i> ₁ = 0.0524 <i>wR</i> ₂ = 0.0956	<i>R</i> ₁ = 0.0483 <i>wR</i> ₂ = 0.0948	<i>R</i> ₁ = 0.0452 <i>wR</i> ₂ = 0.0908
<i>R</i> indices (all data)	<i>R</i> ₁ = 0.0512 <i>wR</i> ₂ = 0.0794	<i>R</i> ₁ = 0.1017 <i>wR</i> ₂ = 0.1531	<i>R</i> ₁ = 0.1067 <i>wR</i> ₂ = 0.1130	<i>R</i> ₁ = 0.0607 <i>wR</i> ₂ = 0.1013	<i>R</i> ₁ = 0.0992 <i>wR</i> ₂ = 0.1034
Largest difference peak and hole (e Å ⁻³)	1.52 and -0.67	3.28 and -1.65	1.29 and -1.30	1.79 and -1.44	1.20 and -2.83
CCDC No.	1585637	1585638	1585639	1585640	1585641

¹ G. M. Sheldrick, *SADABS, Program for area detector adsorption correction*, Institute for Inorganic Chemistry, University of Göttingen, Germany, 1996.

Table S3. Summary of the experimental results.

Reagents	Product	Ligand coordination pattern / tecton conformation	Ag(I) coordination geometry	Network dimensionality and topology	Supramolecular interactions
Tecton 1 + Me ₃ SnCl (molar ratio 1:2)	Transmetallation - Ph ₃ BiCl ₂ and Me ₃ Sn[O(O)CC ₅ H ₄ N-4] (3)	μ_2 -bridging (O1, N1) carboxylate in 3 / -	-	1-D linear chain in 3	-
Tecton 2 + Me ₃ SnCl (molar ratio 1:2)	Transmetallation - Ph ₃ BiCl ₂ and Me ₃ Sn[O(O)CC ₅ H ₄ N-3] (4)	μ_2 -bridging (O1, N1) carboxylate in 4 / -	-	1-D zig-zag chain in 4	-
Tecton 1 + Ni[S ₂ P(O ⁱ Pr) ₂] ₂ (molar ratio 1:1)	Transmetallation - Ph ₃ Bi, [(ⁱ PrO) ₂ P(S)S] ₂ and Ni[O(O)CC ₅ H ₄ N-4] ₂	- / -	-	-	-
Tecton 2 + Ni[S ₂ P(O ⁱ Pr) ₂] ₂ (molar ratio 1:1)	Transmetallation - Ph ₃ Bi, [(ⁱ PrO) ₂ P(S)S] ₂ and Ni[O(O)CC ₅ H ₄ N-3] ₂	- / -	-	-	-
Tecton 1 + AgOTf (molar ratio 1:1)	[Ag{Ph ₃ Bi[O(O)CC ₅ H ₄ N-4] ₂ }(OTf)] (5)	μ_2 -bridging (N1, N2) of tecton 1 / -	see-saw [$\tau_{4\text{ Ag}(1)}$ 0.63]	1-D double chain built from near-linear chains	Ag $\cdots\pi$ contacts (1-D \rightarrow 2-D)
Tecton 2 + AgOTf (molar ratio 1:1)	[Ag{Ph ₃ Bi[O(O)CC ₅ H ₄ N-3] ₂ }(OTf)] \cdot CH ₂ Cl ₂ (6 \cdot CH ₂ Cl ₂)	μ_2 -bridging (N1, N2) of tecton 2 / cis	see-saw [$\tau_{4\text{ Ag}(1A)}$ 0.48, $\tau_{4\text{ Ag}(1AX)}$ 0.24, and $\tau_{4\text{ Ag}(1B)}$ 0.17, $\tau_{4\text{ Ag}(1BX)}$ 0.40, respectively]	2-D grid-like layer constructed from corrugated chains	intra-layer offset $\pi\cdots\pi$ contacts
Tecton 1 + AgSbF ₆ (molar ratio 1:1)	[Ag{Ph ₃ Bi[O(O)CC ₅ H ₄ N-4] ₂ }(SbF ₆) \cdot 2THF (7 \cdot 2THF)	μ_3 -bridging (N1, O2, N2) of tecton 1 / -	distorted trigonal-bipyramid [$\tau_{5\text{ Ag}(1)}$ 90%]	2-D grid-like network built from linear chains	Ag \cdots O contacts
Tecton 2 + AgSbF ₆ (molar ratio 1:1)	[Ag{Ph ₃ Bi[O(O)CC ₅ H ₄ N-3] ₂ }(SbF ₆) \cdot CH ₂ Cl ₂ (8 \cdot CH ₂ Cl ₂)	μ_2 -bridging (N1, N2) of tecton 2 / trans	Square-planar	1-D zig-zag chain	offset $\pi\cdots\pi$ stacking (1-D \rightarrow 2-D)
Tecton 2 + AgNO ₃ (molar ratio 1:1)	[Ag{Ph ₃ Bi[O(O)CC ₅ H ₄ N-3] ₂ }(NO ₃) \cdot CH ₂ Cl ₂ (9 \cdot CH ₂ Cl ₂)	μ_2 -bridging (N1, N2) of tecton 2 / trans	see-saw [$\tau_{4\text{ Ag}(1)}$ 0.67]	1-D wavy chain	Ag \cdots Ag and $\pi\cdots\pi$ contacts (1-D \rightarrow 3-D)

