## SUPPLEMENTARY MATERIALS

Influence of the Pseudohalide Ligands on the SIM Behaviour of Four-Coordinate Benzylimidazole-Containing Cobalt(II) Complexes

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## **Figures:**

**Figure S1.** X–ray powder diffraction pattern of **1** at room temperature, together with the calculated pattern from the single crystal data.

**Figure S2.** X–ray powder diffraction pattern of **2** at room temperature, together with the calculated pattern from the single crystal data.

**Figure S3.** X–ray powder diffraction pattern of **3** at room temperature, together with the calculated pattern from the single crystal data.

Figure S4. IR spectra of 1.

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Figure S7. The solid reflectance spectra UV–Vis (a) and NIR (b) of Co(II) complexes

**Figure S8.** Frequency dependence of the (left) out-of-phase ( $\chi_M$ '') and (right) in-phase ( $\chi_M$ ') components of the ac susceptibility for **1** under an applied static field of  $H_{dc}$  = 500 G with a ±5.0 G oscillating field at frequencies in the range 1000-10000 Hz.

**Figure S9.** Frequency dependence of the (left) out-of-phase ( $\chi_M''$ ) and (right) in-phase ( $\chi_M'$ ) components of the ac susceptibility for **1** under an applied static field of  $H_{dc}$  = 1000 G with a ±5.0 G oscillating field at frequencies in the range 1000-10000 Hz.

**Figure S10.** Frequency dependence of the (left) out-of-phase ( $\chi_M$ ") and (right) in-phase ( $\chi_M$ ) components of the ac susceptibility for **1** under an applied static field of  $H_{dc}$  = 2500 G with a ±5.0 G oscillating field at frequencies in the range 1000-10000 Hz.

**Figure S11.** Frequency dependence of the (left) out-of-phase ( $\chi_M$ ") and (right) in-phase ( $\chi_M$ ) components of the ac susceptibility for **2** under an applied static field of  $H_{dc}$  = 1000 G with a ±5.0 G oscillating field at frequencies in the range 1000-10000 Hz.

**Figure S12.** Frequency dependence of the (left) out-of-phase ( $\chi_M$ ") and (right) in-phase ( $\chi_M$ ) components of the ac susceptibility for **2** under an applied static field of  $H_{dc}$  = 2500 G with a ±5.0 G oscillating field at frequencies in the range 1000-10000 Hz.

**Figure S13.** Frequency dependence of the (left) out-of-phase ( $\chi_M$ ") and (right) in-phase ( $\chi_M$ ) components of the ac susceptibility for **3** under an applied static field of  $H_{dc}$  = 1000 G with a ±5.0 G oscillating field at frequencies in the range 1000-10000 Hz.

**Figure S14.** Frequency dependence of the (left) out-of-phase ( $\chi_M$ ") and (right) in-phase ( $\chi_M$ ) components of the ac susceptibility for **3** under an applied static field of  $H_{dc}$  = 2000 G with a ±5.0 G oscillating field at frequencies in the range 1000-10000 Hz.

Figure S15. Arrhenius plots for 1 under applied fields of (left) 1000 and (right) 2500 G.

**Figure S16**. Cole-Cole plots for **1** in the temperature range 2.0-2.6 K at the indicated frequencies and under applied fields of (top) 500, (middle) 1000 and (bottom) 2500 G.

**Figure S17**. Natural logarithm of the  $\chi_M''/\chi_M'$  ratio vs. 1/T for **2** under a dc magnetic field of 1000 G and a ±5.0 G oscillating field at the indicated frequencies.

**Figure S18.** Natural logarithm of the  $\chi_M''/\chi_M'$  ratio vs. 1/T for **3** under dc magnetic fields of (left) 1000 and (right) 2000 G and a ±5.0 G oscillating field at the indicated frequencies

## Tables:

Table S1. Selected bond lengths [Å] and angles [°] for 1-3

Table S2. UV-Vis-NIR data for compounds 1–3.

Table S3. Short intra- and intermolecular contacts detected in structures 2 and 3.

CShM allows to determine a distance between the real metal environment and the ideal polyhedron. Mathematically, CShM of the original structure (Q) is a normalized root-mean-square deviation from the referenced structure with the desired symmetry (P) and it is expressed by eqn (1)

$$S_{\mathcal{Q}}(P) = \min\left[\frac{\sum_{i=1}^{n} |\vec{\mathbf{q}}_{i} - \vec{\mathbf{p}}_{i}|^{2}}{\sum_{i=1}^{n} |\vec{\mathbf{q}}_{i} - \vec{\mathbf{q}}_{0}|^{2}}\right] \times 100$$
 (1)

where,  $\vec{q_i}$  are *N* vectors that contain the 3*N* Cartesian coordinates of the problem structure *Q*,  $\vec{p_i}$  contain the coordinates of the ideal polyhedron *P* and  $\vec{q_0}$  is the position vector of the geometric center that is chosen to be the same for the two polyhedra. The minimum is taken for all possible relative orientations in space, isotropic scaling, and for all possible pairings of the vertexes of the problem and reference polyhedra. The value  $S_Q(P)$  tends to zero, when the polyhedron *Q* is close to the ideal one. The maximum allowed value  $S_Q(P)$  is 100, although in practice the values found for severely distorted chemical structures are very rarely larger than 40. Generally, it is assumed that shape measures of less than 1.0 indicate minor distortions from the reference shape; values of up to 3.0 units indicate important distortions, but the reference shape provides still a good stereochemical description.

The geometry of structure Q, which is intermediate between two polyhedra T and P, can be described by its position along the minimal distortion path given by the *generalized polyhedral interconversion coordinate*  $\varphi_{T \to P}^Q$ . It is derived from the shape measure of structure Q relative to the initial shape of the path, T, according to following equation

$$\varphi_{T \to P}^{Q} = \frac{100}{\theta_{TP}} \arcsin(\frac{\sqrt{S_{Q}(T)}}{10})$$
(2)

When the structure is coincident with T,  $\varphi_{T \to P}^{Q}$  is zero, while for a structure coincident with the end point of the *P* path, it amounts to 100%. Structures along the path have intermediate  $\varphi_{T \to P}^{Q}$  values that correspond to the portion of the covered path (in percentage).<sup>1-3</sup>

<sup>1.</sup> J. Cirera, P. Alemany and S. Alvarez, Chem. Eur. J., 2004, 10, 190.

<sup>2.</sup> D. Casanova, J. Cirera, M. Llunell, P. Alemany, D. Avnir and S. Alvarez, J. Am. Chem. Soc., 2004, 126, 1755.

<sup>3.</sup> J. Cirera, E. Ruiz and S. Alvarez, Inorg. Chem., 2008, 47, 2871

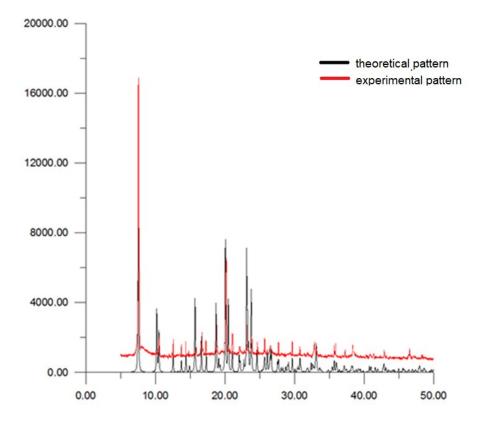


Figure S1. X-ray powder diffraction pattern of 1 at room temperature, together with the calculated pattern from the single crystal data.

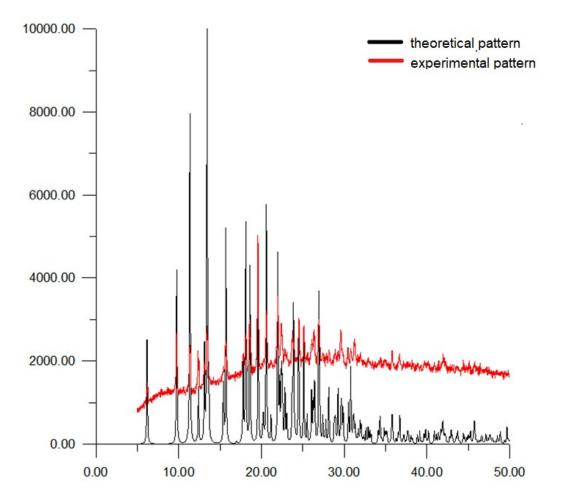


Figure S2. X-ray powder diffraction pattern of 2 at room temperature, together with the calculated pattern from the single crystal data.

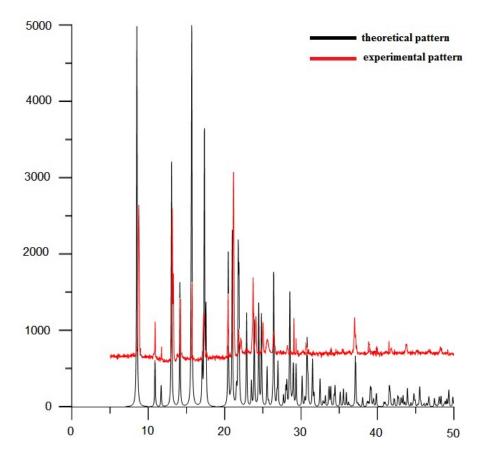


Figure S3. X-ray powder diffraction pattern of 3 at room temperature, together with the calculated pattern from the single crystal data.

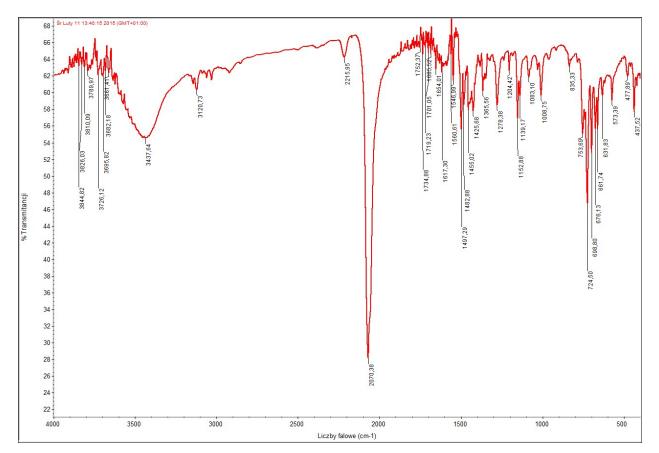


Figure S4. IR spectra of 1.

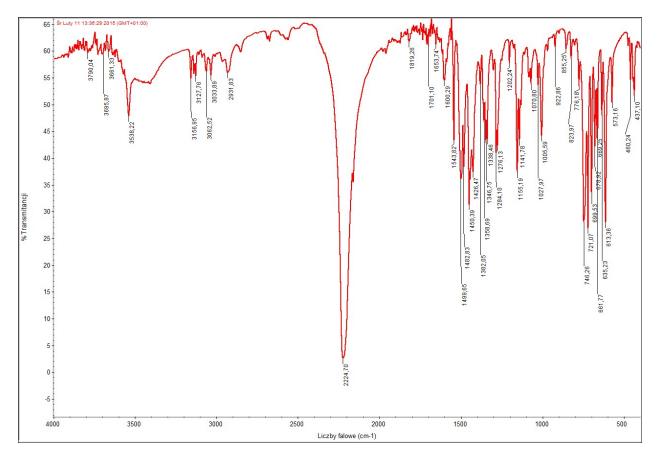


Figure S5. IR spectra of 2.

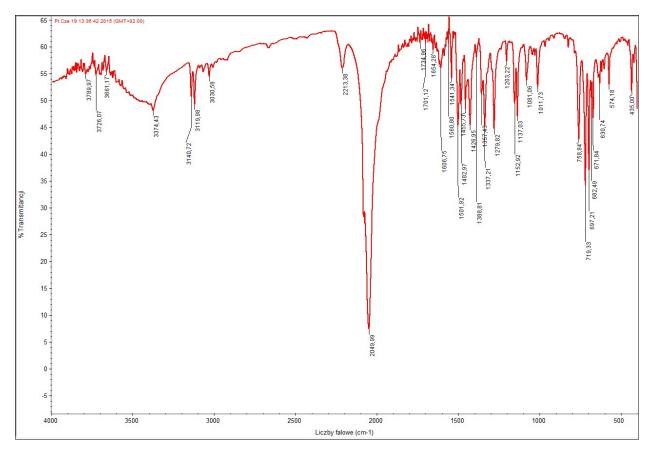


Figure S6. IR spectra of 3.

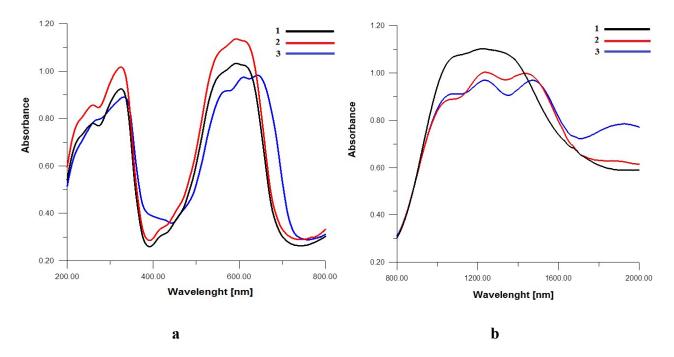
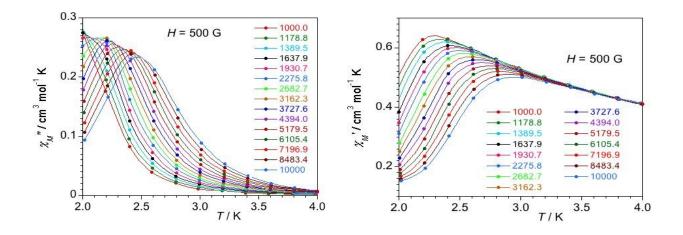
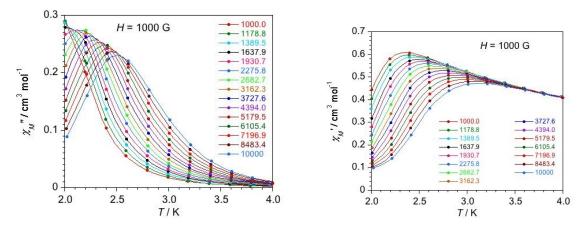


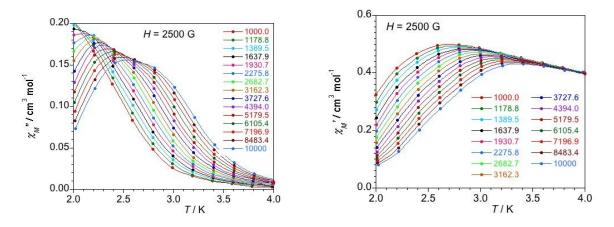
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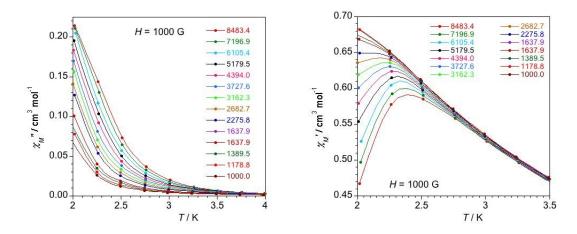
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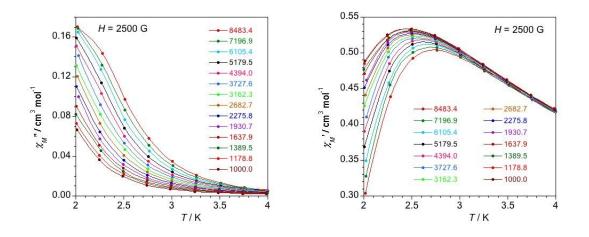
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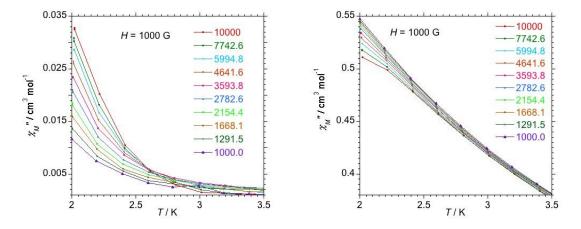
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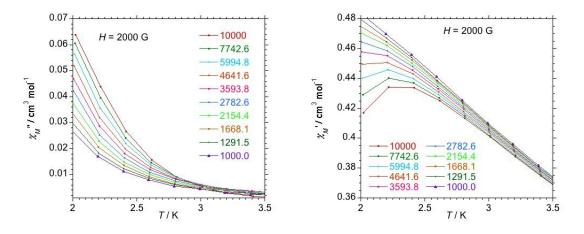
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**Figure S13.** Frequency dependence of the (left) out-of-phase ( $\chi_M$ '') and (right) in-phase ( $\chi_M$ ') components of the ac susceptibility for **3** under an applied static field of  $H_{dc} = 1000$  G with a ±5.0 G oscillating field at frequencies in the range 1000-10000 Hz.



**Figure S14**. Frequency dependence of the (left) out-of-phase ( $\chi_M$ '') and (right) in-phase ( $\chi_M$ ') components of the ac susceptibility for **3** under an applied static field of  $H_{dc} = 2000$  G with a ±5.0 G oscillating field at frequencies in the range 1000-10000 Hz.

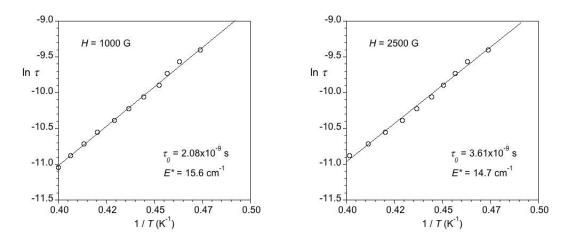
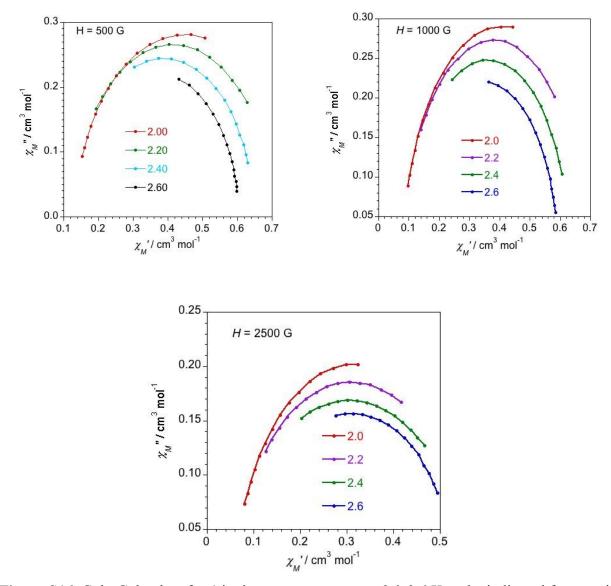
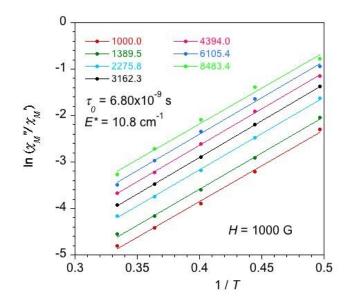


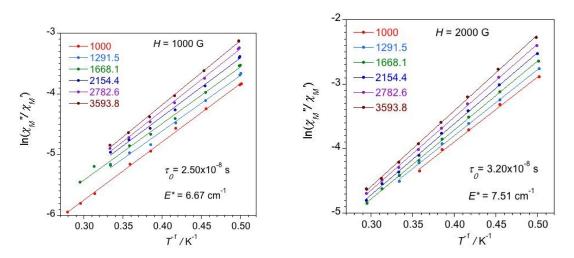
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**Figure S18**. Natural logarithm of the  $\chi_M''/\chi_M'$  ratio vs. 1/T for **3** under dc magnetic fields of (left) 1000 and (right) 2000 G and a ±5.0 G oscillating field at the indicated frequencies

Bond lengths		Bond angles			
		1	·		
Co(1)–N(1)	1.984(6)	N(1)-Co(1)-N(3)	111.3(4)		
Co(1) - N(3)	1.961(11)	N(1)-Co(1)-N(3')	114.8(4)		
Co(1) - N(3')	2.093(13)	N(1)-Co(1)-N(98)	114.2(3)		
Co(1)–N(98)	1.924(6)	N(1)-Co(1)-N(99)	99.5(3)		
Co(1) - N(99) 1.948(7)		N(3)-Co(1)-N(98)	119.2(4)		
		N(3')-Co(1)-N(98)	95.6(4)		
		N(3)-Co(1)-N(99)	98.7(4)		
		N(3')-Co(1)-N(99)	122.7(4)		
		N(98)–Co(1)–N(99)	110.9(3)		
		Co(1)-N(99)-C(99)	164.9(6)		
		Co(1)–N(98)–C(98)	169.5(6)		
		S(98)-C(98)-N(98)	178.9(6)		
		S(99)–C(99)–N(99)	178.7(7)		
		2			
Co(1)–N(1)	2.012(2)	N(1)-Co(1)-N(3)	105.19(8)		
Co(1) - N(3)	2.017(2)	N(1)-Co(1)-N(98)	111.19(9)		
Co(1) - N(98)	1.931(2)	N(1)-Co(1)-N(99)	108.58(9)		
Co(1)–N(99)	1.939(2)	N(3)-Co(1)-N(98)	113.28(9)		
		N(3)-Co(1)-N(99)	105.17(9)		
		N(98)–Co(1)–N(99)	112.95(10)		
		Co(1)–N(98)–C(98)	165.0(2)		
		Co(1)-N(99)-C(99)	170.5(3)		
		O(98)-C(98)-N(98)	178.9(3)		
		O(99)–C(99)–N(99)	178.7(4)		
		3			
Co(1)–N(1)	2.011(2)	N(1)-Co(1)-N(1)a	108.25(13)		
Co(1)-N(1)a	2.011(2)	N(1)-Co(1)-N(99)	112.34(10)		
Co(1)–N(99)	1.951(3)	N(1)-Co(1)-N(99)a	106.83(10)		
Co(1)–N(99)a	1.951(3)	N(1)a - Co(1) - N(99)	112.34(10)		
	~ /	N(99)-Co(1)-N(99)a	110.3(2)		
		N(1)a –Co(1)–N(99)a	106.83(10)		
		N(97)–N(98)–N(99)	176.4(3)		
		Co(1)–N(99)–N(98)	128.3(2)		

Table S1. Selected bond lengths [Å] and angles [°] for 1-3

Symmetry transformations used to generate equivalent atoms: (a) = 1-x, y, 1/2-z.

Compound	$\lambda_{max}(cm^{-1})/(nm)$					Dq [cm <sup>-1</sup> ]
	<b>v</b> <sub>1</sub>	<b>v</b> <sub>2</sub>	v <sub>3</sub>	$n \rightarrow \pi^* \pi \rightarrow \pi^*$		
1	7468 (1339)	16891 (592)	17921 (558)	30769 (325)	660	859
	8156 (1226)			39370 (254)		
	9285 (1077)					
	Average: 8636					
2	6973 (1434)	16891 (592)	18148 (551)	30769 (325)	693	868
	8097 (1235)			38910 (257)		
	9560 (1046)					
	Average: 8210					
3	6793 (1472)	15576 (642)	17825 (561)	30211 (331)	604	746
	8090 (1236)			37593 (266)		
	9451 (1058)					
	Average 8111					

Table S2. UV-VIS-NIR data for compounds 1–3

Table S3. Short intra- and intermolecular contacts in the structures of 2 and 3

D—H•••A	D—H	Н•••А	D•••A	D—H•••A				
			[Å]	[°]				
2								
C(1)-H(1)•••O(98)#a	0.93	2.50	3.1422	126.00				
C(5)–H(5B)•••O(99)#b	0.93	2.53	3.4259	154.00				
C(7)–H(7)•••O(99)#b	0.93	2.52	3.3423	147.00				
C(18)-H(18)•••N(4)#c	0.93	2.60	2.908(4)	100.00				
3								
C(4)–H(4A)•••N(99)	0.96	2.56	3.406(5)	146.0				
C(4)-H(4B)•••N(97)#d	0.96	2.50	3.429(4)	163.0				

Symmetry transformations used to generate equivalent atoms: #a: -1+x,y,z; #b: -1+x,-1+y,z; #c: 1-x,1-y,1-z; #d: x,1-y,1/2+z