

Electronic Supplementary Information

**LiCs₂La(BO₃)₂ and Li₃K₉La₃(BO₃)₇: new mixed alkali metal lanthanum borates
with three-dimensional open frameworks and short cut-off edges**

Ruru Ma,^{a,b} Yun Yang,^{a,*} Cong Hu,^{a,b} Xiaoyu Dong,^a Zhihua Yang,^a Shilie Pan,^{a,*}

^aKey Laboratory of Functional Materials and Devices for Special Environments,
Xinjiang Technical Institute of Physics & Chemistry, Chinese Academy of Sciences;
Xinjiang Key Laboratory of Electronic Information Materials and Devices, 40-1
South Beijing Road, Urumqi 830011, China.

^bUniversity of Chinese Academy of Sciences, Beijing 100049, China.

*To whom correspondence should be addressed. *E-mail: yangyun@ms.xjb.ac.cn
(Yun Yang), slpan@ms.xjb.ac.cn (Shilie Pan). Phone: (86)-991-3810816. Fax: (86)-
991-3838957.

CONTENTS

Table S1 Bond lengths [$\text{\AA}$] and angles [deg] for (a) $\text{LiCs}_2\text{La}(\text{BO}_3)_2$ and (b) $\text{Li}_3\text{K}_9\text{La}_3(\text{BO}_3)_7$.

Table S2 Atomic coordinates ($\times 10^4$) and equivalent isotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for $\text{LiCs}_2\text{La}(\text{BO}_3)_2$ and calculated bond valence sum (BVS).

Table S3 Atomic coordinates ($\times 10^4$) and equivalent isotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for $\text{Li}_3\text{K}_9\text{La}_3(\text{BO}_3)_7$ and calculated bond valence sum (BVS).

Table S4 Anisotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for $\text{LiCs}_2\text{La}(\text{BO}_3)_2$.

Table S5 Anisotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for $\text{Li}_3\text{K}_9\text{La}_3(\text{BO}_3)_7$.

Table S6 Existing complex alkaline and/or alkaline earth lanthanum borates, space group, cation/boron ratio and B-O configuration.

Fig. S1 Experimental and calculated X-ray powder diffraction patterns of (a) $\text{LiCs}_2\text{La}(\text{BO}_3)_2$ and (b) $\text{Li}_3\text{K}_9\text{La}_3(\text{BO}_3)_7$.

Fig. S2 The unit cells and asymmetric units of $\text{LiCs}_2\text{La}(\text{BO}_3)_2$ and $\text{Li}_3\text{K}_9\text{La}_3(\text{BO}_3)_7$.

Fig. S3 1D infinite $[\text{LiB}_2\text{O}_6]^{5-}$ chain extending along c direction.

Fig. S4 The arrangement of the BO_3 units in $\text{LiCs}_2\text{La}(\text{BO}_3)_2$.

Fig. S5 The arrangement of the BO_3 units in $\text{Li}_3\text{K}_9\text{La}_3(\text{BO}_3)_7$.

Fig. S6 The La-B-O layered structure along the $[100]$ and $[010]$ directions in $\text{Li}_3\text{K}_9\text{La}_3(\text{BO}_3)_7$.

Fig. S7 Isolated $[\text{Li}_3\text{O}_{10}]$ clusters distributed on the bc plane.

Fig. S8 The coordination environment of cations in $\text{Li}_3\text{K}_9\text{La}_3(\text{BO}_3)_7$.

Fig. S9 The 1D infinite large channels along the c axis in $\text{Li}_3\text{K}_9\text{La}_3(\text{BO}_3)_7$.

Fig. S10 The crystal structures of (a)-(d) $\text{LiCs}_2\text{La}(\text{BO}_3)_2$ and (e)-(h) $\text{LiRb}_2\text{La}(\text{BO}_3)_2$.

Fig. S11 The infrared spectra of (a) $\text{LiCs}_2\text{La}(\text{BO}_3)_2$ and (b) $\text{Li}_3\text{K}_9\text{La}_3(\text{BO}_3)_7$.

Fig. S12 Band structures of $\text{LiCs}_2\text{La}(\text{BO}_3)_2$. (b) DOS and PDOS for $\text{LiCs}_2\text{La}(\text{BO}_3)_2$.

Fig. S13 The dihedral angles between $\text{B}(1)\text{O}_3$ and $\text{B}(2)\text{O}_3$ groups of (a) $\text{Na}_3\text{La}_2(\text{BO}_3)_3$ and (b) $\text{LiCs}_2\text{La}(\text{BO}_3)_2$.

Table S1 Bond lengths [Å] and angles [deg] for (a) LiCs₂La(BO₃)₂ and (b) Li₃K₉La₃(BO₃)₇.

(a) LiCs ₂ La(BO ₃) ₂			
Li(1)-O(4)#1	1.878(5)	O(2)-Cs(1)-O(4)#10	102.98(7)
Li(1)-O(4)#2	1.878(5)	O(4)#9-Cs(1)-O(4)#10	105.75(5)
Li(1)-O(2)	1.992(8)	O(1)-Cs(1)-O(4)#2	123.44(6)
Li(1)-O(3)#3	2.016(8)	O(2)-Cs(1)-O(4)#2	64.71(7)
Cs(1)-O(1)	2.933(2)	O(4)#9-Cs(1)-O(4)#2	111.38(4)
Cs(1)-O(2)	3.061(2)	O(4)#10-Cs(1)-O(4)#2	44.50(8)
Cs(1)-O(4)#9	3.093(3)	O(1)-Cs(1)-O(3)#11	129.00(8)
Cs(1)-O(4)#10	3.183(2)	O(2)-Cs(1)-O(3)#11	148.03(7)
Cs(1)-O(4)#2	3.213(2)	O(4)#9-Cs(1)-O(3)#11	84.63(7)
Cs(1)-O(3)#11	3.291(14)	O(4)#10-Cs(1)-O(3)#11	63.05(7)
Cs(1)-O(5)	3.451(3)	O(4)#2-Cs(1)-O(3)#11	107.55(7)
Cs(1)-O(5)#2	3.505(15)	O(1)-Cs(1)-O(5)	73.86(7)
Cs(1)-O(2)#13	3.485(17)	O(2)-Cs(1)-O(5)	63.74(5)
La(1)-O(2)	2.448(3)	O(4)#9-Cs(1)-O(5)	139.31(5)
La(1)-O(3)	2.473(3)	O(4)#10-Cs(1)-O(5)	108.86(5)
La(1)-O(4)#6	2.483(2)	O(4)#2-Cs(1)-O(5)	108.18(6)
La(1)-O(4)	2.483(2)	O(3)#11-Cs(1)-O(5)	92.41(6)
La(1)-O(5)	2.6126(9)	O(1)-Cs(1)-O(2)#13	128.15(8)
La(1)-O(5)#6	2.6126(9)	O(2)-Cs(1)-O(2)#13	107.10(4)
La(1)-O(1)	2.682(3)	O(4)#9-Cs(1)-O(2)#13	125.42(6)
La(1)-O(1)#3	2.759(3)	O(4)#10-Cs(1)-O(2)#13	60.31(7)
B(1)-O(4)	1.376(3)	O(4)#2-Cs(1)-O(2)#13	93.62(7)
B(1)-O(4)#14	1.376(3)	O(3)#11-Cs(1)-O(2)#13	41.00(8)
B(1)-O(5)#6	1.377(6)	O(5)-Cs(1)-O(2)#13	59.73(5)
B(2)-O(1)	1.374(6)	O(5)#6-La(1)-O(1)	93.53(3)
B(2)-O(2)#9	1.377(6)	O(5)-La(1)-O(1)	93.53(3)
B(2)-O(3)	1.379(6)	O(4)#6-La(1)-O(1)	124.11(7)
O(4)#1-Li(1)-O(4)#2	101.5(4)	O(3)-La(1)-O(1)	54.75(10)
O(4)#1-Li(1)-O(2)	120.4(3)	O(2)-La(1)-O(3)	135.27(11)
O(4)#2-Li(1)-O(2)	120.4(3)	O(2)-La(1)-O(4)#6	132.03(7)
O(4)#1-Li(1)-O(3)#3	120.8(3)	O(3)-La(1)-O(4)#6	81.13(8)
O(4)#2-Li(1)-O(3)#3	120.8(3)	O(2)-La(1)-O(4)	132.03(7)
O(2)-Li(1)-O(3)#3	72.8(3)	O(3)-La(1)-O(4)	81.13(8)
O(1)-Cs(1)-O(2)	67.21(8)	O(4)#6-La(1)-O(4)	71.74(11)

O(1)-Cs(1)-O(4)#9	76.81(8)	O(2)-La(1)-O(5)	86.03(7)
O(2)-Cs(1)-O(4)#9	127.33(6)	O(3)-La(1)-O(5)	95.82(4)
O(1)-Cs(1)-O(4)#10	167.92(6)	O(2)-La(1)-O(1)#3	54.39(10)
O(4)#6-La(1)-O(5)	55.22(9)	O(3)-La(1)-O(1)#3	170.34(10)
O(4)-La(1)-O(5)	126.54(9)	O(4)#6-La(1)-O(1)#3	91.06(8)
O(2)-La(1)-O(5)#6	86.03(7)	O(4)-La(1)-O(1)#3	91.06(8)
O(3)-La(1)-O(5)#6	95.82(4)	O(5)-La(1)-O(1)#3	84.23(2)
O(4)#6-La(1)-O(5)#6	126.54(9)	O(5)#6-La(1)-O(1)#3	84.23(2)
O(4)-La(1)-O(5)#6	55.22(9)	O(1)-La(1)-O(1)#3	134.90(2)
O(5)-La(1)-O(5)#6	168.36(7)	O(4)-B(1)-O(4)#14	123.3(5)
O(2)-La(1)-O(1)	80.51(10)	O(4)-B(1)-O(5)#6	118.4(2)
O(1)-B(2)-O(2)#9	121.1(4)	O(4)#14-B(1)-O(5)#6	118.4(2)
O(2)#9-B(2)-O(3)	119.4(4)	O(1)-B(2)-O(3)	119.5(4)

Symmetry transformations used to generate equivalent atoms:

#1 $x-1, y, z$	#2 $x-1, y, -z+1/2$	#3 $-x+1, y+1/2, -z+1/2$
#4 $x, -y+1/2, -z+1$	#5 $x, -y+1/2, z-1/2$	#6 $x, y, -z+1/2$
#7 $-x, y+1/2, z$	#8 $-x, y+1/2, -z+1/2$	#9 $-x+1, y-1/2, -z+1/2$
#10 $x-1, -y+1/2, z+1/2$	#11 $-x+1, -y, -z+1$	#12 $-x+1, -y, z+1/2$
#13 $x, -y+1/2, z+1/2$	#14 $x, -y+1/2, -z$	#15 $-x+1, -y, z-1/2$
#16 $x+1, -y+1/2, z-1/2$	#17 $x+1, y, -z+1/2$	#18 $-x+1, y+1/2, z$
#19 $x+1, y, z$		

(b) $\text{Li}_3\text{K}_9\text{La}_3(\text{BO}_3)_7$			
Li(1)-O(11B)#1	1.732(7)	K(4)-O(5)#4	2.832(2)
Li(1)-O(11B)	1.732(7)	K(4)-O(8)#11	2.849(2)
Li(1)-O(11A)#1	1.935(8)	K(4)-O(5)	2.929(2)
Li(1)-O(11A)	1.935(8)	K(4)-O(3)#12	3.084(2)
Li(1)-O(10)#2	1.944(7)	K(5)-O(2)	2.817(2)
Li(1)-O(10)#3	1.944(7)	K(5)-O(7)	2.860(2)
Li(2)-O(11A)	1.926(8)	K(5)-O(11B)#4	2.953(9)
Li(2)-O(6)#5	1.972(7)	K(5)-O(11B)#14	2.989(10)
Li(2)-O(2)#6	2.030(7)	K(5)-O(9)#14	3.027(3)
Li(2)-O(11B)	2.094(10)	K(5)-O(8)#4	3.083(3)
Li(2)-O11#1	2.102(10)	La(1)-O(7)#4	2.451(2)
Li(2)-O(9)	2.333(9)	La(1)-O(7)	2.451(2)
K(1)-O(8)	2.681(2)	La(1)-O(2)#4	2.511(2)
K(1)-O(8)#4	2.681(2)	La(1)-O(2)	2.511(2)
K(1)-O(7)	2.681(2)	La(1)-O(5)	2.549(2)
K(1)-O(7)#4	2.681(2)	La(1)-O(5)#4	2.549(2)
K(1)-O(4)	2.702(2)	La(1)-O(3)	2.626(2)
K(1)-O(4)#4	2.702(2)	La(1)-O(3)#4	2.626(2)
K(2)-O(9)	2.658(2)	La(2)-O(4)	2.446(2)
K(2)-O(1)	2.709(2)	La(2)-O(9)	2.470(2)
K(2)-O(9)#5	2.779(2)	La(2)-O(8)	2.480(2)
K(2)-O(6)#8	2.795(2)	La(2)-O(6)	2.487(2)
K(2)-O(6)	2.851(2)	La(2)-O(1)	2.5165(5)
K(2)-O(3)#6	2.862(2)	La(2)-O(10)#8	2.526(3)
K(2)-O(5)#6	3.284(2)	La(2)-O(5)#16	2.597(2)
K(3)-O(10)#8	2.797(3)	La(2)-O(3)#7	2.772(2)
K(3)-O(4)	2.798(2)	B(1)-O(4)	1.376(4)
K(3)-O(4)#7	2.810(2)	B(1)-O(7)#7	1.376(4)
K(3)-O(11A)#9	2.820(5)	B(1)-O(3)#7	1.393(4)
K(3)-O(7)#7	2.846(2)	B(2)-O(6)#11	1.364(4)
K(3)-O(2)#4	2.878(2)	B(2)-O(2)	1.380(4)
K(3)-O(7)	2.915(2)	B(2)-O(5)	1.385(4)
K(3)-O(8)#9	2.961(2)	B(3)-O(11B)	1.361(7)
K(3)-O(11B)#9	3.049(7)	B(3)-O(8)	1.369(4)
K(4)-O(3)	2.772(3)	B(3)-O(9)	1.372(4)
K(4)-O(4)#10	2.793(2)	B(3)-O(11A)	1.437(6)

B(4)-O(10)#8	1.362(4)	O(11B)-Li(2)-O(11A)#1	83.6(3)
B(4)-O(10)	1.362(4)	O(11A)-Li(2)-O(6)#5	136.0(4)
B(4)-O(1)	1.369(6)	O(11A)-Li(2)-O(2)#6	150.3(4)
O(11B)#1-Li(1)-O(11B)	124.6(8)	O(6)#5-Li(2)-O(2)#6	73.0(2)
O(11B)#1-Li(1)-O(11A)#1	29.0(4)	O(11A)-Li(2)-O(11B)	26.6(3)
O(11B)-Li(1)-O(11A)#1	99.3(5)	O(6)#5-Li(2)-O(11B)	158.4(5)
O(11B)#1-Li(1)-O(11A)	99.3(5)	O(2)#6-Li(2)-O(11B)	123.9(4)
O(11B)-Li(1)-O(11A)	29.0(4)	O(11A)-Li(2)-O(11A)#1	75.9(4)
O(11A)#1-Li(1)-O(11A)	79.7(4)	O(6)#5-Li(2)-O(11A)#1	106.0(4)
O(11B)#1-Li(1)-O(10)#2	116.6(3)	O(2)#6-Li(2)-O(11A)#1	105.3(4)
O(11B)-Li(1)-O(10)#2	106.6(4)	O(11B)-Li(2)-O(11A)#1	83.6(3)
O(11A)#1-Li(1)-O(10)#2	120.03(18)	O(8)-K(1)-O(8)#4	128.50(10)
O(11A)-Li(1)-O(10)#2	135.35(18)	O(8)-K(1)-O(7)	132.83(7)
O(11B)#1-Li(1)-O(10)#3	106.6(4)	O(8)#4-K(1)-O(7)	90.73(7)
O(11B)-Li(1)-O(10)#3	116.6(3)	O(8)-K(1)-O(7)#4	90.73(7)
O(11A)#1-Li(1)-O(10)#3	135.35(18)	O(8)#4-K(1)-O(7)#4	132.83(7)
O(11A)-Li(1)-O(10)#3	120.03(18)	O(7)-K(1)-O(7)#4	74.29(9)
O(10)#2-Li(1)-O(10)#3	75.7(3)	O(8)-K(1)-O(4)	76.09(7)
O(11A)-Li(2)-O(6)#5	136.0(4)	O(8)#4-K(1)-O(4)	84.89(7)
O(11A)-Li(2)-O(2)#6	150.3(4)	O(7)-K(1)-O(4)	83.90(6)
O(6)#5-Li(2)-O(2)#6	73.0(2)	O(7)#4-K(1)-O(4)	135.29(7)
O(11A)-Li(2)-O(11B)	26.6(3)	O(8)-K(1)-O(4)#4	84.89(7)
O(6)#5-Li(2)-O(11B)	158.4(5)	O(8)#4-K(1)-O(4)#4	76.09(7)
O(2)#6-Li(2)-O(11B)	123.9(4)	O(7)-K(1)-O(4)#4	135.29(7)
O(11A)-Li(2)-O(11A)#1	75.9(4)	O(7)#4-K(1)-O(4)#4	83.90(6)
O(6)#5-Li(2)-O(11A)#1	106.0(4)	O(4)-K(1)-O(4)#4	135.43(10)
O(2)#6-Li(2)-O(11A)#1	105.3(4)	O(9)-K(2)-O(1)	74.91(6)
O(11B)-Li(2)-O(11A)#1	83.6(3)	O(9)-K(2)-O(9)#5	98.10(7)
O(11A)-Li(2)-O(6)#5	136.0(4)	O(1)-K(2)-O(9)#5	140.63(7)
O(11A)-Li(2)-O(2)#6	150.3(4)	O(9)-K(2)-O(6)#8	141.23(7)
O(6)#5-Li(2)-O(2)#6	73.0(2)	O(1)-K(2)-O(6)#8	68.00(6)
O(11A)-Li(2)-O(11B)	26.6(3)	O(9)#5-K(2)-O(6)#8	103.29(7)
O(6)#5-Li(2)-O(11B)	158.4(5)	O(9)-K(2)-O(6)	76.90(7)
O(2)#6-Li(2)-O(11B)	123.9(4)	O(1)-K(2)-O(6)	67.19(6)
O(11A)-Li(2)-O(11A)#1	75.9(4)	O(9)#5-K(2)-O(6)	73.48(7)
O(6)#5-Li(2)-O(11A)#1	106.0(4)	O(6)#8-K(2)-O(6)	78.72(7)
O(2)#6-Li(2)-O(11A)#1	105.3(4)	O(9)-K(2)-O(3)#6	114.85(8)

O(1)-K(2)-O(3)#6	84.05(5)	O(7)-K(3)-O(8)#9	125.23(6)
O(9)#5-K(2)-O(3)#6	131.66(7)	O(10)#8-K(3)-O(11B)#9	64.59(15)
O(6)#8-K(2)-O(3)#6	72.72(6)	O(4)-K(3)-O(11B)#9	127.29(19)
O(6)-K(2)-O(3)#6	145.42(6)	O(4)#7-K(3)-O(11B)#9	122.70(13)
O(9)-K(2)-O(5)#6	98.18(7)	O(11A)#9-K(3)-O(11B)#9	17.86(18)
O(1)-K(2)-O(5)#6	144.58(6)	O(7)#7-K(3)-O(11B)#9	102.28(19)
O(9)#5-K(2)-O(5)#6	74.24(6)	O(2)#4-K(3)-O(11B)#9	87.6(2)
O(6)#8-K(2)-O(5)#6	118.51(6)	O(7)-K(3)-O(11B)#9	153.5(2)
O(6)-K(2)-O(5)#6	146.22(6)	O(8)#9-K(3)-O(11B)#9	46.55(12)
O(3)#6-K(2)-O(5)#6	67.16(6)	B(1)#7-K(3)-O(11B)#9	134.94(18)
O(10)#8-K(3)-O(4)	70.50(7)	O(3)-K(4)-O(4)#10	146.99(7)
O(10)#8-K(3)-O(4)#7	167.70(8)	O(3)-K(4)-O(5)#4	78.61(6)
O(4)-K(3)-O(4)#7	98.10(6)	O(4)#10-K(4)-O(5)#4	81.77(6)
O(10)#8-K(3)-O(11A)#9	73.48(11)	O(3)-K(4)-O(8)#11	132.72(7)
O(4)-K(3)-O(11A)#9	142.17(11)	O(4)#10-K(4)-O(8)#11	80.16(6)
O(4)#7-K(3)-O(11A)#9	116.51(11)	O(5)#4-K(4)-O(8)#11	119.00(7)
O(10)#8-K(3)-O(7)#7	89.88(9)	O(3)-K(4)-O(5)	73.62(6)
O(4)-K(3)-O(7)#7	50.80(6)	O(4)#10-K(4)-O(5)	122.45(6)
O(4)#7-K(3)-O(7)#7	79.01(6)	O(5)#4-K(4)-O(5)	68.17(7)
O(11A)#9-K(3)-O(7)#7	118.88(10)	O(8)#11-K(4)-O(5)	74.16(6)
O(10)#8-K(3)-O(2)#4	99.29(10)	O(3)-K(4)-O(3)#12	101.47(6)
O(4)-K(3)-O(2)#4	126.44(7)	O(4)#10-K(4)-O(3)#12	47.32(6)
O(4)#7-K(3)-O(2)#4	91.20(6)	O(5)#4-K(4)-O(3)#12	74.56(6)
O(11A)#9-K(3)-O(2)#4	70.42(10)	O(8)#11-K(4)-O(3)#12	124.95(7)
O(7)#7-K(3)-O(2)#4	168.75(6)	O(5)-K(4)-O(3)#12	142.67(6)
O(10)#8-K(3)-O(7)	128.58(8)	O(4)#10-K(4)-O(6)#7	121.34(6)
O(4)-K(3)-O(7)	78.06(6)	O(5)#4-K(4)-O(6)#7	141.20(6)
O(4)#7-K(3)-O(7)	50.00(6)	O(8)#11-K(4)-O(6)#7	96.80(6)
O(11A)#9-K(3)-O(7)	135.79(10)	O(5)-K(4)-O(6)#7	112.29(6)
O(7)#7-K(3)-O(7)	100.72(6)	O(3)#12-K(4)-O(6)#7	97.89(6)
O(2)#4-K(3)-O(7)	68.45(6)	O(2)-K(5)-O(7)	75.65(6)
O(10)#8-K(3)-O(8)#9	105.96(8)	O(2)-K(5)-O(11B)#4	169.17(13)
O(4)-K(3)-O(8)#9	132.14(7)	O(7)-K(5)-O(11B)#4	104.34(17)
O(4)#7-K(3)-O(8)#9	77.96(7)	O(2)-K(5)-O(11B)#14	77.57(16)
O(11A)#9-K(3)-O(8)#9	49.99(11)	O(7)-K(5)-O(11B)#14	153.14(16)
O(7)#7-K(3)-O(8)#9	82.11(6)	O(11B)#4-K(5)-O(11B)#14	101.6(3)
O(2)#4-K(3)-O(8)#9	101.41(7)	O(2)-K(5)-O(9)#14	72.62(6)

O(7)-K(5)-O(9)#14	126.23(7)	O(7)-La(1)-O(3)	55.84(7)
O(11B)#4-K(5)-O(9)#14	114.51(16)	O(2)#4-La(1)-O(3)	91.38(7)
O(11B)#14-K(5)-O(9)#14	44.74(15)	O(2)-La(1)-O(3)	90.02(7)
O(2)-K(5)-O(8)#4	123.98(7)	O(5)-La(1)-O(3)	82.66(7)
O(7)-K(5)-O(8)#4	79.79(6)	O(5)#4-La(1)-O(3)	86.61(7)
O(11B)#4-K(5)-O(8)#4	46.31(15)	O(7)#4-La(1)-O(3)#4	55.84(7)
O(11B)#14-K(5)-O(8)#4	114.27(16)	O(7)-La(1)-O(3)#4	137.97(7)
O(9)#14-K(5)-O(8)#4	153.67(6)	O(2)#4-La(1)-O(3)#4	90.02(7)
O(2)-K(5)-O(10)#7	135.04(7)	O(2)-La(1)-O(3)#4	91.38(7)
O(7)-K(5)-O(10)#7	118.12(7)	O(5)-La(1)-O(3)#4	86.61(7)
O(11B)#4-K(5)-O(10)#7	54.79(14)	O(5)#4-La(1)-O(3)#4	82.66(7)
O(11B)#14-K(5)-O(10)#7	82.94(13)	O(3)-La(1)-O(3)#4	166.12(10)
O(9)#14-K(5)-O(10)#7	65.11(7)	O(4)-La(2)-O(9)	140.76(7)
O(8)#4-K(5)-O(10)#7	100.95(6)	O(4)-La(2)-O(8)	84.67(7)
O(2)-K(5)-O(10)#15	87.50(8)	O(9)-La(2)-O(8)	57.25(7)
O(7)-K(5)-O(10)#15	121.81(7)	O(4)-La(2)-O(6)	127.36(7)
O(11B)#4-K(5)-O(10)#15	83.36(14)	O(9)-La(2)-O(6)	87.53(7)
O(11B)#14-K(5)-O(10)#15	54.47(14)	O(8)-La(2)-O(6)	127.88(7)
O(9)#14-K(5)-O(10)#15	99.01(7)	O(4)-La(2)-O(1)	120.13(7)
O(8)#4-K(5)-O(10)#15	64.67(6)	O(9)-La(2)-O(1)	81.79(6)
O(10)#7-K(5)-O(10)#15	112.99(9)	O(8)-La(2)-O(1)	126.91(7)
O(7)#4-La(1)-O(7)	82.69(10)	O(6)-La(2)-O(1)	75.95(8)
O(7)#4-La(1)-O(2)#4	89.13(7)	O(4)-La(2)-O(10)#8	80.97(9)
O(7)#4-La(1)-O(2)#4	89.13(7)	O(9)-La(2)-O(10)#8	88.07(11)
O(7)-La(1)-O(2)#4	82.09(7)	O(8)-La(2)-O(10)#8	88.13(11)
O(7)#4-La(1)-O(2)	82.09(7)	O(6)-La(2)-O(10)#8	131.48(8)
O(7)-La(1)-O(2)	89.13(7)	O(1)-La(2)-O(10)#8	55.60(9)
O(2)#4-La(1)-O(2)	168.33(10)	O(4)-La(2)-O(5)#16	93.72(7)
O(7)-La(1)-O(5)	127.02(7)	O(9)-La(2)-O(5)#16	93.14(7)
O(2)#4-La(1)-O(5)	135.10(7)	O(8)-La(2)-O(5)#16	86.63(7)
O(2)-La(1)-O(5)	56.56(7)	O(6)-La(2)-O(5)#16	55.63(7)
O(7)#4-La(1)-O(5)#4	127.02(7)	O(1)-La(2)-O(5)#16	131.53(8)
O(7)-La(1)-O(5)#4	124.03(7)	O(10)#8-La(2)-O(5)#16	172.87(8)
O(2)#4-La(1)-O(5)#4	56.56(7)	O(4)-La(2)-O(3)#7	53.68(7)
O(2)-La(1)-O(5)#4	135.10(7)	O(9)-La(2)-O(3)#7	165.55(7)
O(5)-La(1)-O(5)#4	78.63(9)	O(8)-La(2)-O(3)#7	136.28(7)
O(7)#4-La(1)-O(3)	137.97(7)	O(6)-La(2)-O(3)#7	79.07(7)

O(1)-La(2)-O(3)#7	89.64(5)	O(11B)-B(3)-O(8)	121.1(4)
O(10)#8-La(2)-O(3)#7	96.69(11)	O(11B)-B(3)-O(9)	113.9(4)
O(5)#16-La(2)-O(3)#7	83.83(7)	O(8)-B(3)-O(9)	119.8(3)
O(4)-B(1)-O(7)#7	123.2(3)	O(8)-B(3)-O(11A)	121.3(3)
O(4)-B(1)-O(3)#7	118.1(3)	O(9)-B(3)-O(11A)	116.3(3)
O(7)#7-B(1)-O(3)#7	118.7(3)	O(10)#8-B(4)-O(10)	122.3(5)
O(6)#11-B(2)-O(2)	120.3(3)	O(10)#8-B(4)-O(1)	118.9(2)
O(6)#11-B(2)-O(5)	119.5(3)	O(10)-B(4)-O(1)	118.9(2)
O(2)-B(2)-O(5)	120.2(3)		

Symmetry transformations used to generate equivalent atoms:

- | | |
|-------------------|--------------------|
| #1 -x+1,y,-z+3/2 | #2 x,-y+2,z+1/2 |
| #3 -x+1,-y+2,-z+1 | #4 -x,y,-z+1/2 |
| #5 -x+1,-y+3,-z+1 | #6 x+1,-y+2,z+1/2 |
| #7 -x,-y+2,-z | #8 -x+1, y,-z+1/2 |
| #9 x,-y+2,z-1/2 | #10 x,y-1,z |
| #11 -x,y-1,-z+1/2 | #12 -x,-y+1,-z |
| #13 x,-y+1,z-1/2 | #14 x-1,-y+2,z-1/2 |
| #15 x-1,y,z | #16 -x,y+1,-z+1/2 |
| #17 x,y+1,z | #18 x,-y+1,z+1/2 |
| #19 x+1,y,z | |

Table S2 Atomic coordinates ($\times 10^4$) and equivalent isotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for $\text{LiCs}_2\text{La}(\text{BO}_3)_2$ and calculated bond valence sum (BVS).

	x/a	y/b	z/c	U(eq)	BVS
Li(1)	3899(1)	12500	2500	14(2)	1.124
Cs(1)	5887(1)	771(1)	4461(1)	22(1)	1.091
La(1)	6426(1)	2265(1)	2500	10(1)	2.847
B(1)	8149(8)	2500	0	13(1)	2.952
B(2)	6566(7)	-702(5)	2500	12(1)	2.954
O(1)	4844(4)	-147(3)	2500	17(1)	1.697
O(2)	3199(4)	2937(3)	2500	15(1)	1.901
O(3)	8085(5)	116(3)	2500	21(1)	1.741
O(4)	9046(3)	2895(2)	1101(2)	16(1)	2.148
O(5)	6261(5)	2500	5000	19(1)	1.707

$U(\text{eq})$ is defined as one third of the trace of the orthogonalized U_{ij} tensor.

Table S3 Atomic coordinates ($\times 10^4$) and equivalent isotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for $\text{Li}_3\text{K}_9\text{La}_3(\text{BO}_3)_7$.

	S.O.F	x/b	y/c	z/c	U(eq)	BVS
Li(1)	1	5000	10636(8)	7500	23(2)	1.320
Li(2)	1	5841(7)	12907(6)	7044(8)	44(2)	0.894
K(1)	1	771(1)	4461(1)	22(1)	1.091	1.361
K(2)	1	6098(1)	14581(1)	4441(1)	21(1)	1.119
K(3)	1	1722(1)	9286(1)	870(1)	20(1)	1.116
K(4)	1	-934(1)	3917(1)	496(1)	24(1)	0.832
K(5)	1	-3381(1)	9060(1)	531(1)	38(1)	0.789
La(1)	1	0	6892(1)	2500	11(1)	3.049
La(2)	1	2839(1)	12975(1)	2546(1)	13(1)	3.078
B(1)	1	949(3)	12143(3)	-13(3)	14(1)	2.988
B(2)	1	-2363(3)	5805(4)	2466(3)	15(1)	2.958
B(3)	1	3689(4)	12243(4)	5075(3)	22(1)	2.935
B(4)	1	5000	11511(6)	2500	36(2)	3.055
O(1)	1	5000	12852(3)	2500	21(1)	—
O(2)	1	2206(2)	7142(2)	2441(2)	20(1)	—
O(3)	1	1365(2)	6582(2)	37(2)	19(1)	—
O(4)	1	799(2)	11789(2)	976(2)	16(1)	—
O(5)	1	1460(2)	4962(2)	2397(2)	19(1)	—
O(6)	1	3421(2)	15304(2)	2463(2)	19(1)	—
O(7)	1	-694(2)	8694(2)	981(2)	17(1)	—
O(8)	1	2403(2)	11926(2)	4102(2)	23(1)	—
O(9)	1	4494(2)	13052(2)	4847(2)	24(1)	—

O(10)	1	6046(3)	10867(3)	2514(4)	72(1)	—
O(11A)	0.5	4082(5)	12090(4)	6367(5)	13(1)	—
O(11B)	0.5	4407(7)	11424(9)	6070(6)	73(3)	—

$U(\text{eq})$ is defined as one third of the trace of the orthogonalized U_{ij} tensor.

Table S4 Anisotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for $\text{LiCs}_2\text{La}(\text{BO}_3)_2$.

$\text{LiCs}_2\text{La}(\text{BO}_3)_2$						
	U11	U22	U33	U23	U13	U12
Li(1)	9(4)	14(4)	20(4)	0	0	0(3)
Cs(1)	28(1)	14(1)	23(1)	0(1)	11(1)	1(1)
La(1)	8(1)	10(1)	11(1)	0	0	0(1)
B(1)	18(3)	9(2)	12(2)	-2(2)	0	0
B(2)	12(2)	13(2)	11(2)	0	0	-1(2)
O(1)	13(2)	16(2)	21(2)	0	0	5(1)
O(2)	10(2)	11(2)	24(2)	0	0	0(1)
O(3)	14(2)	14(2)	34(2)	0	0	1(1)
O(4)	14(1)	20(1)	13(1)	1(1)	-1(1)	-3(1)
O(5)	13(2)	29(2)	15(2)	3(1)	0	0

Table S5 Anisotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for $\text{Li}_3\text{K}_9\text{La}_3(\text{BO}_3)_7$.

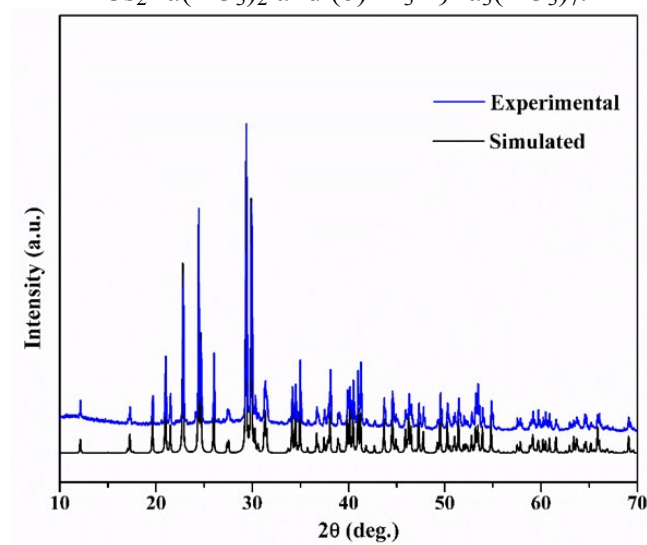
$\text{Li}_3\text{K}_9\text{La}_3(\text{BO}_3)_7$						
	U11	U22	U33	U23	U13	U12
Li(1)	29(4)	20(4)	22(4)	0	15(4)	0
Li(2)	29(4)	31(4)	83(6)	-26(4)	38(4)	-11(3)
K(1)	16(1)	15(1)	14(1)	0	8(1)	0
K(2)	19(1)	23(1)	24(1)	-6(1)	13(1)	-6(1)
K(3)	20(1)	16(1)	22(1)	0(1)	10(1)	-1(1)
K(4)	22(1)	18(1)	31(1)	-1(1)	13(1)	1(1)
K(5)	28(1)	46(1)	41(1)	17(1)	18(1)	7(1)
La(1)	11(1)	10(1)	11(1)	0	6(1)	0
La(2)	11(1)	12(1)	14(1)	-1(1)	6(1)	-2 (1)
B(1)	9(2)	17(2)	13(2)	1(1)	4(1)	0(1)
B(2)	14(2)	18(2)	12(2)	1(1)	7(1)	1(1)
B(3)	18(2)	24(2)	18(2)	4(2)	6(2)	-5(2)
B(4)	20(3)	15(3)	67(5)	0	19(3)	0
O(1)	28(2)	13(2)	29(2)	0	19(2)	0
O(2)	18(1)	14(1)	31(1)	-1(1)	16(1)	-1(1)
O(3)	24(1)	15(1)	16(1)	-1(1)	9(1)	-6(1)
O(4)	18(1)	18(1)	15(1)	15(1)	10(1)	-5(1)
O(5)	17(1)	14(1)	30(1)	3(1)	15(1)	1(1)
O(6)	16(1)	17(1)	31(1)	-4(1)	16(1)	-1(1)
O(7)	3(1)	12(1)	15(1)	-1(1)	10(1)	1(1)
O(8)	16(1)	31(1)	19(1)	0(1)	7(1)	-8(1)

O(9)	18(1)	30(1)	20(1)	1(1)	8(1)	-10(1)
O(10)	40(2)	18(1)	183(4)	28(2)	75(2)	12(1)
O(11 A)	16(2)	11(2)	18(2)	-1(2)	13(2)	1(2)
O(11 B)	44(4)	101(7)	32(4)	39(4)	-10(3)	-35(4)

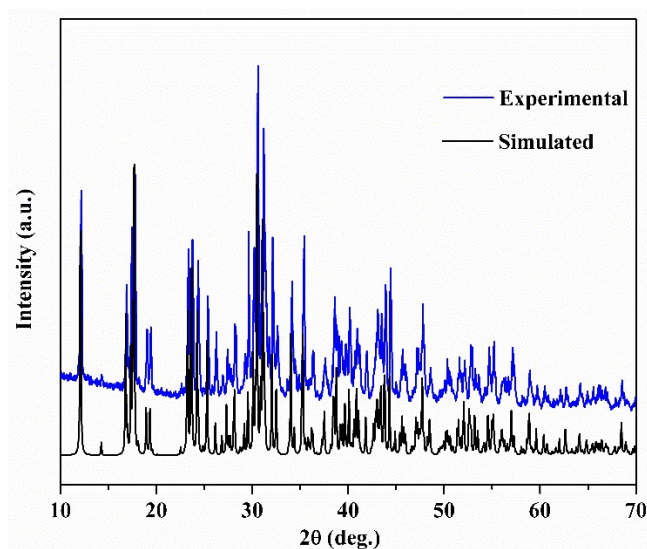
Table S6. Existing complex alkaline and/or alkaline earth lanthanum borates, space group, cation/boron ratio and B-O configuration

	Compounds	Space group	A/B ratio (A=metal ion)	B-O configuration
1	$\text{Li}_3\text{K}_9\text{La}_3(\text{BO}_3)_7$	$P2_1/c$ (no.13)	2.14	Isolated BO_3
2	$\text{LiCs}_2\text{La}(\text{BO}_3)_2$	$Pbcm$ (no.57)	2	Isolated BO_3
3	$\text{LiRb}_2\text{La}(\text{BO}_3)_2$	$P2_1/c$ (no.14)	2	Isolated BO_3
4	$\text{K}_2\text{La}_2\text{O}(\text{BO}_3)_2$	$P2_1/c$ (no.14)	2	Isolated BO_3
5	$\text{Na}_3\text{La}_2(\text{BO}_3)_3$	$Amm2$ (no.38)	1.7	Isolated BO_3
6	$\text{Ca}_4\text{LaO}(\text{BO}_3)_3$	Cm (no.8)	1.7	Isolated BO_3
7	$\text{Na}_3\text{La}_9\text{O}_3(\text{BO}_3)_8$	$P-62m$ (no.189)	1.5	Isolated BO_3
8	$\text{LiSr}_4\text{La}_3(\text{BO}_3)_6$	$R-3$ (no.148)	1.3	Isolated BO_3
9	$\text{NaSr}_4\text{La}_3(\text{BO}_3)_6$	$R-3$ (no.148)	1.3	Isolated BO_3
10	$\text{Ba}_3\text{La}_2(\text{BO}_3)_4$	$Pnma$ (no.62)	1.25	Isolated BO_3
11	$\text{Ca}_3\text{La}_2(\text{BO}_3)_4$	$Pnma$ (no.62)	1.25	Isolated BO_3
12	$\text{Sr}_3\text{La}_2(\text{BO}_3)_4$	$Pna2_1$ (no.3)	1.25	Isolated BO_3
13	$\text{Ca}_3\text{La}_3(\text{BO}_3)_5$	$P6_3mc$ (no.186)	1.2	Isolated BO_3
14	LaBeB_3O_7	$Pmn2_1$ (no.31)	0.67	$\text{BeB}_5\text{O}_{18}$ ring
15	$\text{LaMgB}_5\text{O}_{10}$	$P2_1/c$ (no.14)	0.5	B_5O_{10} layer
16	$\text{La}_2\text{CaB}_{10}\text{O}_{19}$	$C2$ (no.5)	0.3	B_5O_{12} double-layer

Fig. S1 Experimental and calculated X-ray powder diffraction patterns of (a) $\text{LiCs}_2\text{La}(\text{BO}_3)_2$ and (b) $\text{Li}_3\text{K}_9\text{La}_3(\text{BO}_3)_7$.



(a)



(b)

Fig. S2 The unit cells and asymmetric units of $\text{LiCs}_2\text{La}(\text{BO}_3)_2$ and $\text{Li}_3\text{K}_9\text{La}_3(\text{BO}_3)_7$

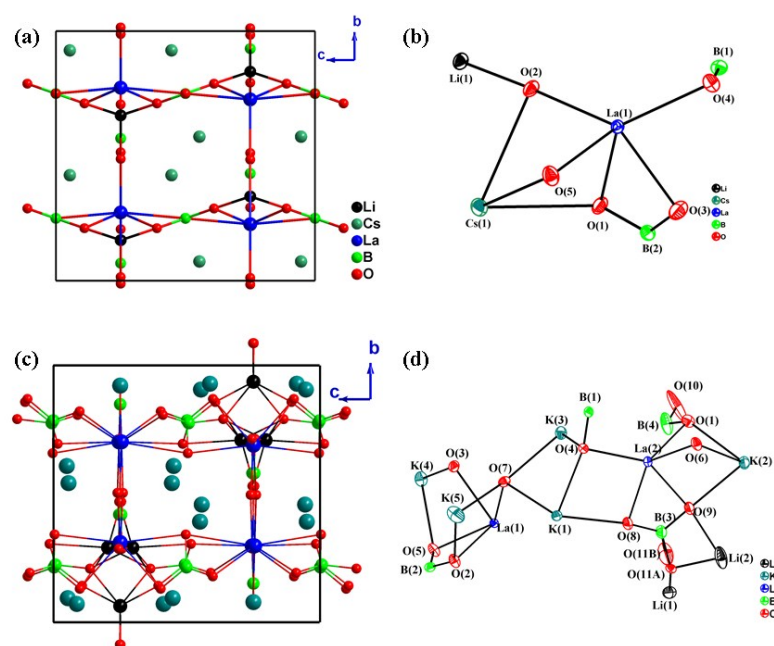


Fig. S3 1D infinite $[\text{LiB}_2\text{O}_6]^{5-}$ chain extending along c direction.

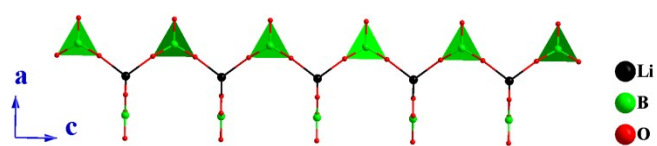


Fig. S4 The arrangement of the BO_3 units in $\text{LiCs}_2\text{La}(\text{BO}_3)_2$.

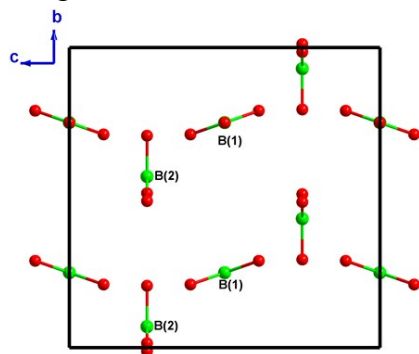


Fig. S5 The arrangement of the BO_3 units in $\text{Li}_3\text{K}_9\text{La}_3(\text{BO}_3)_7$.

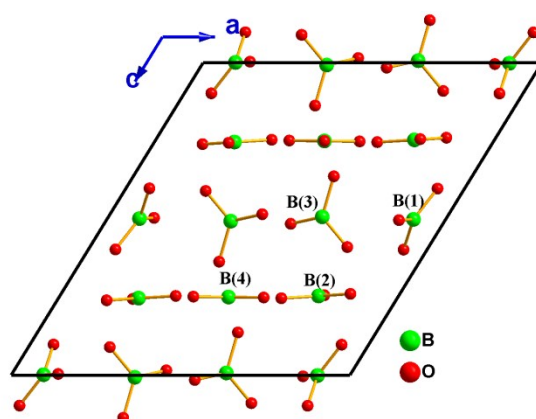


Fig. S6 The La-B-O layered structure along the $[100]$ and $[010]$ directions in $\text{Li}_3\text{K}_9\text{La}_3(\text{BO}_3)_7$.

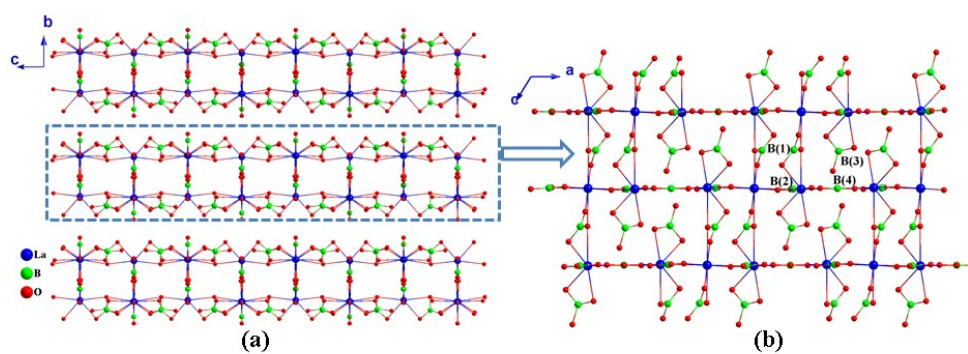


Fig. S7 Isolated $[\text{Li}_3\text{O}_{10}]$ clusters distributed on the bc plane.

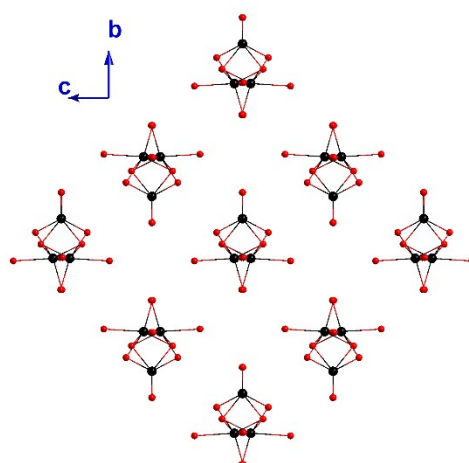


Fig. S10 The crystal structures of (a)-(d) $\text{LiCs}_2\text{La}(\text{BO}_3)_2$ and (e)-(h) $\text{LiRb}_2\text{La}(\text{BO}_3)_2$.

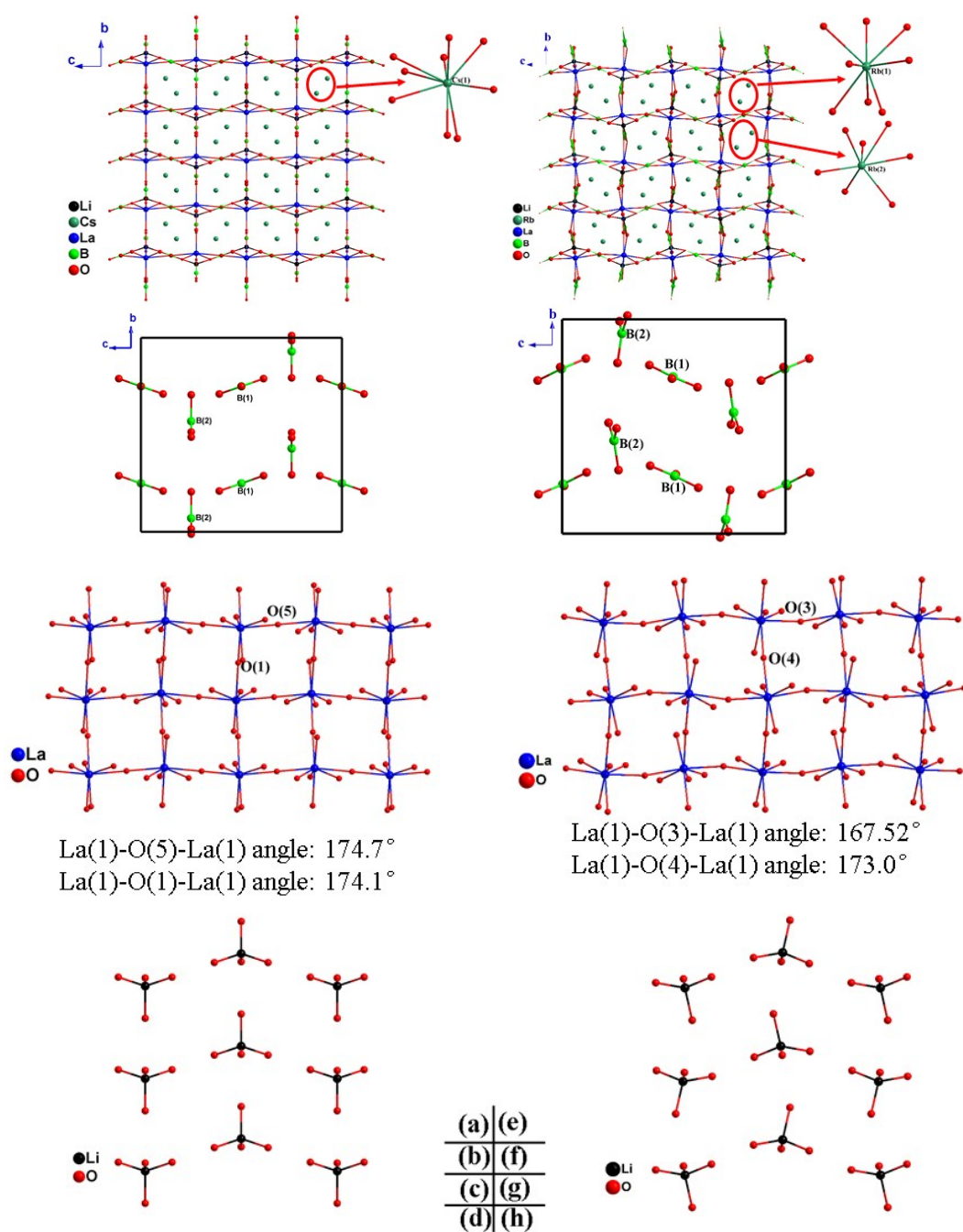


Fig. S11 The infrared spectra of (a) $\text{LiCs}_2\text{La}(\text{BO}_3)_2$ and (b) $\text{Li}_3\text{K}_9\text{La}_3(\text{BO}_3)_7$.

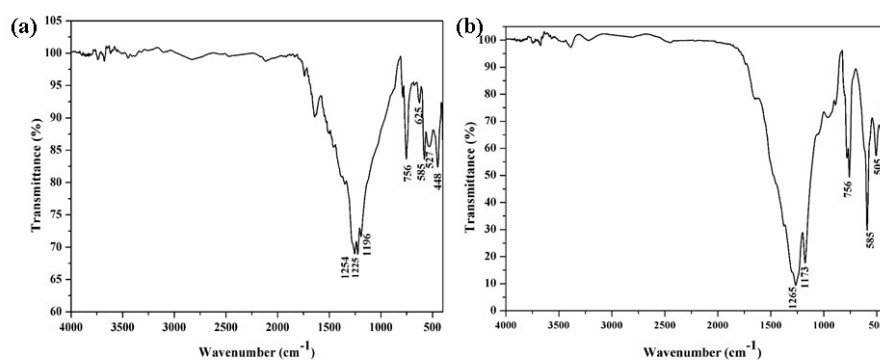


Fig.S12 Band structures of $\text{LiCs}_2\text{La}(\text{BO}_3)_2$. (b) DOS and PDOS for $\text{LiCs}_2\text{La}(\text{BO}_3)_2$.

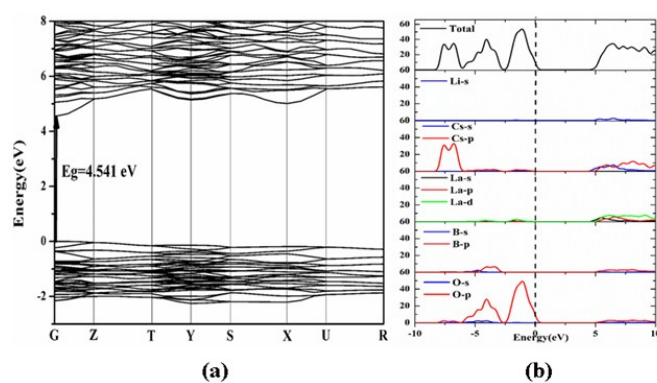


Fig. S13 The dihedral angles between $\text{B}(1)\text{O}_3$ and $\text{B}(2)\text{O}_3$ groups of (a) $\text{Na}_3\text{La}_2(\text{BO}_3)_3$ and (b) $\text{LiCs}_2\text{La}(\text{BO}_3)_2$.

