

**Supporting information:** **A.** sampling sites and bulk chemical and mineralogical composition of the soil samples Figs. S1-S6, Table S1; **B.** SEM-BSE images of MSC and MOM (Figs. S7-S10); **C.** TEM, SAED and STEM-EDS images of all identified phases (S11-S28); **D.** Table S29, Point of zero charges for selected phases identified in this study.

# Supplementary data A

Sampling sites Figure S1 and S2

Bulk mineralogical composition Figures S3-S6

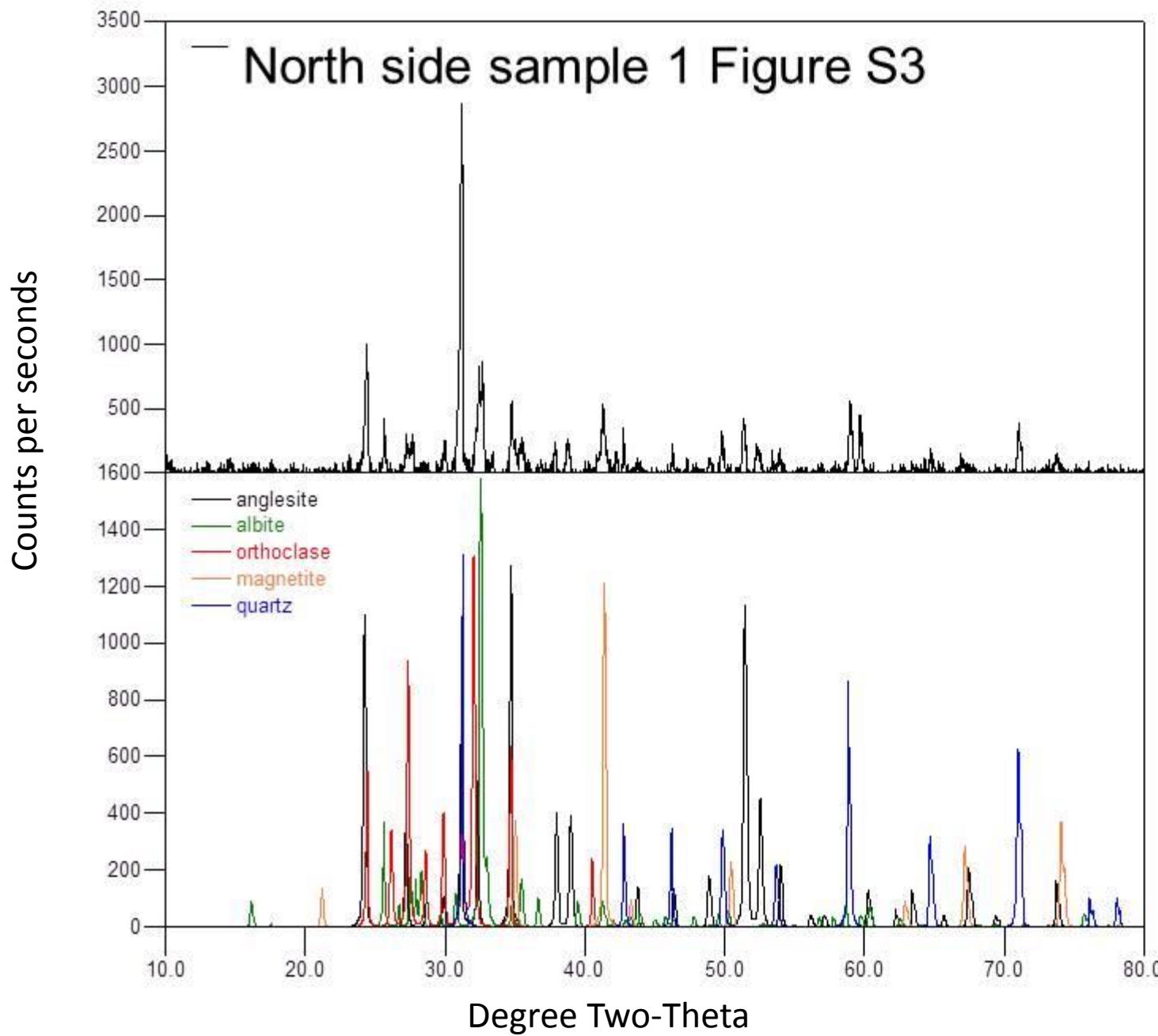
Bulk composition Table S1

Figure S1, Sampling site North, May 2016

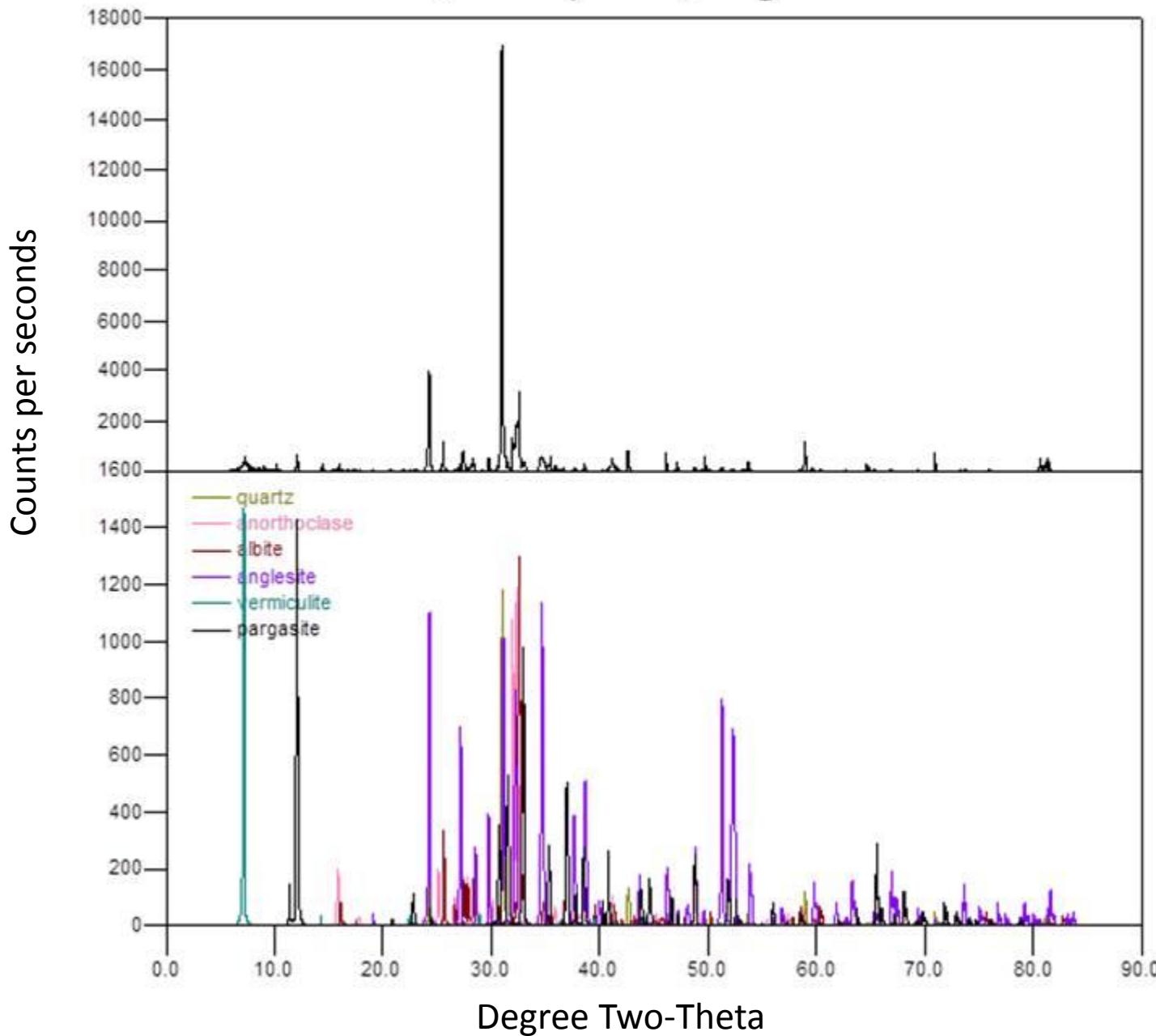


Figure S2, Sampling site South East, May 2016

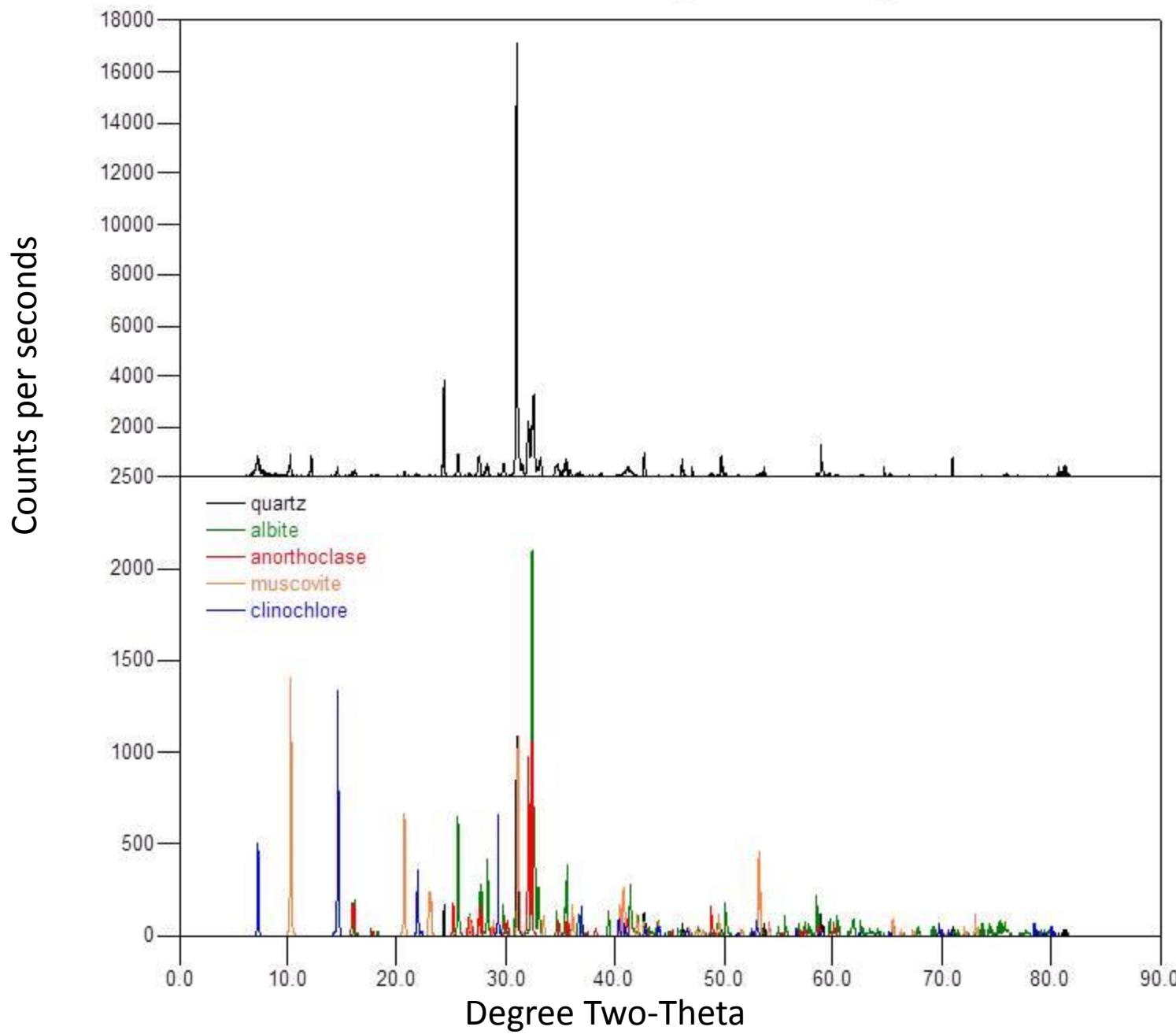


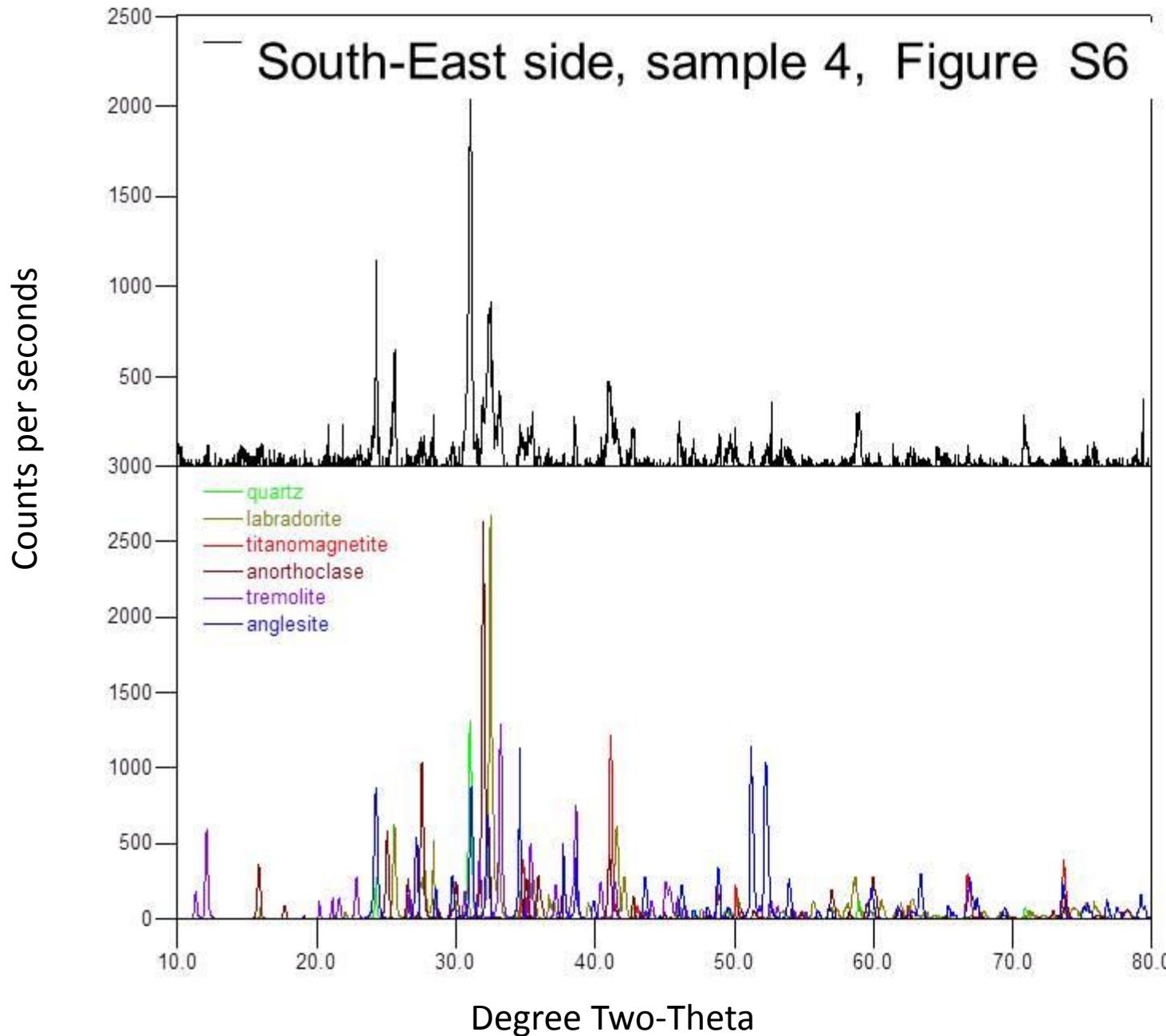


— North side, sample 2, Figure S4



# — South-East side, sample 3, Figure S5





# Table S1. Bulk chemical analysis

	Analyte:	Al	As	B	Ba	Bi	Ca	Cd	Ce	Co	Cs	Cu	Fe
	Unit:	%	ppm	ppm	ppm	ppm	%	ppm	ppm	ppm	ppm	ppm	%
Sample Description	RDL:	0.01	30	20	0.5	0.1	0.05	0.2	0.1	0.5	0.1	5	0.01
North site 1		5.78	843	28	1050	35	1.27	28.5	61	19.4	2.7	566	3.66
North site 2		6.11	820	24	1130	32.7	1.36	35.5	53	10.3	3.2	610	3.75
South East site 3		6.6	440	21	950	17.5	1.51	24.2	61	10.2	3.2	450	3.6
South East site 4		6.32	470	23	1060	20.2	1.42	23.1	76	23.3	3	465	3.47
	Analyte:	K	Mg	Mn	P	Pb	S	Sb	Si	Sn	Sr	Ti	Zn
	Unit:	%	%	ppm	%	ppm	%	ppm	%	ppm	ppm	%	ppm
Sample Description	RDL:	0.05	0.01	10	0.01	5	0.01	0.1	0.01	1	0.1	0.01	5
North site 1		1.4	0.61	590	0.34	16600	0.32	766	22.6	146	343	0.33	3520
North site 2		1.62	0.71	695	0.37	14000	0.26	720	21.5	125	382	0.38	3850
South East site 3		1.61	0.72	745	0.44	8800	0.21	440	23.2	112	352	0.35	2440
South East site 4		1.55	0.67	638	0.48	10100	0.24	360	21.9	80	375	0.38	2590

# Supplementary data B1

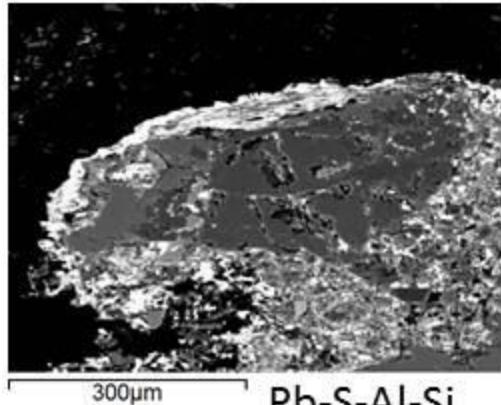
## SEM-BSE images of Mineral surface coatings

### Figures

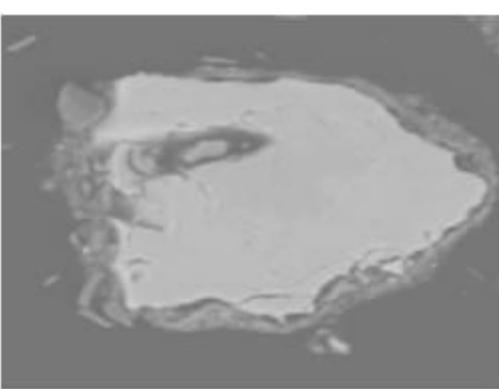
S7 North site

S8 South East site

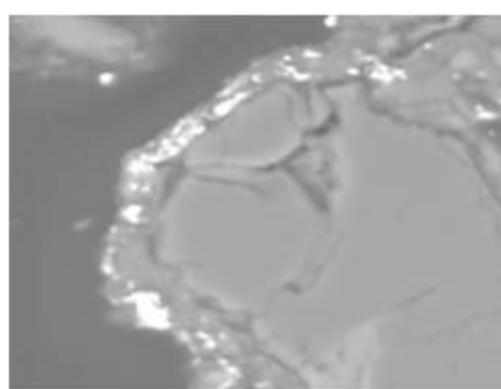
Elements are listed if their concentrations  
are above 1 at%;



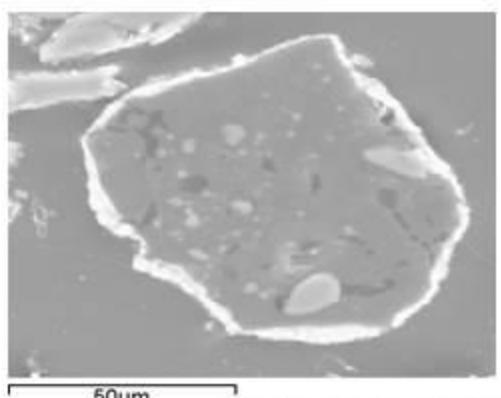
300μm  
Pb-S-Al-Si



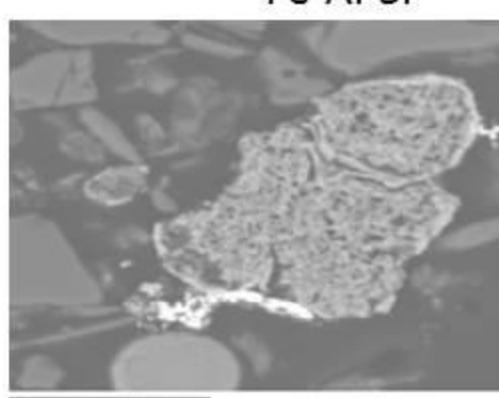
60μm  
Fe-Al-Si



20μm  
Pb-S-Na-Al-Si



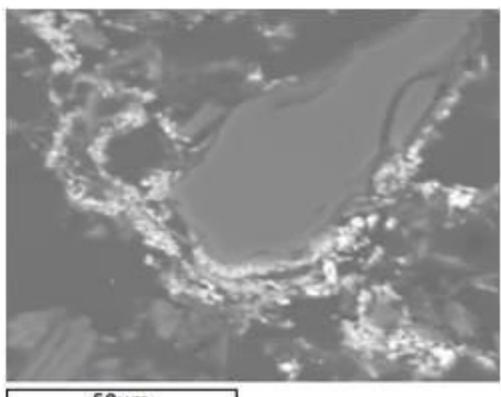
50μm  
Pb-S-Fe-P-Al-Si



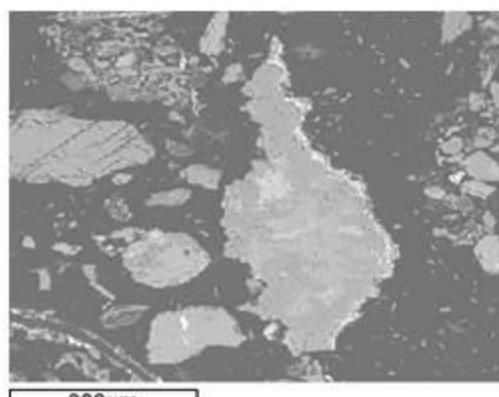
40μm  
Pb-S-Fe-Al-Si



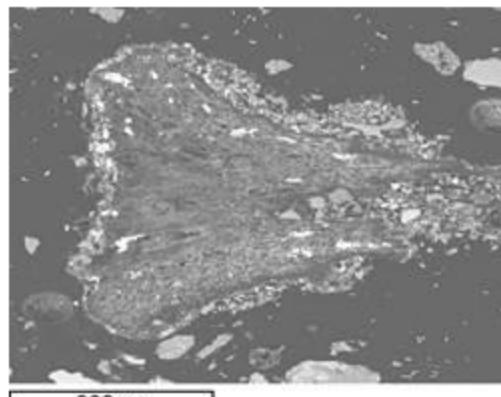
100μm  
Pb-S-Al-Si



50μm  
Pb-S-Al-Si

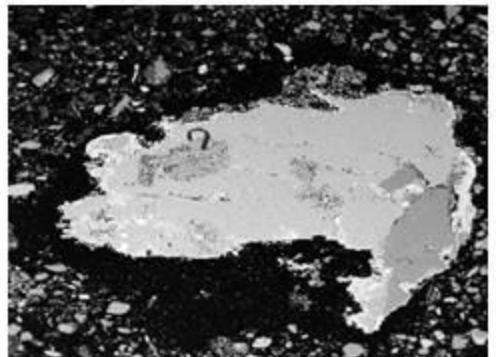


300μm  
Pb-S-Fe-Al-Si

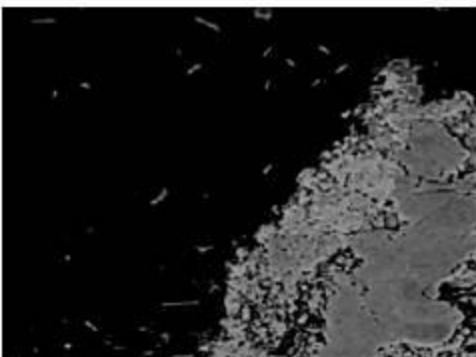


300μm  
Pb-S-Na-Al-Si

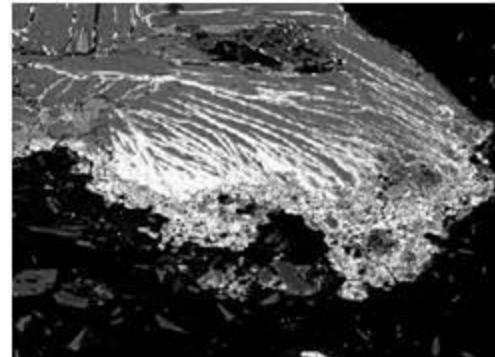
S7



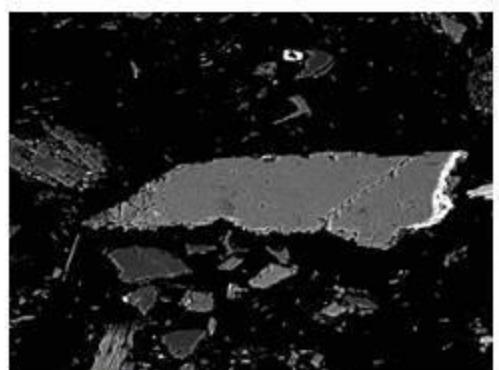
Pb-S-Si



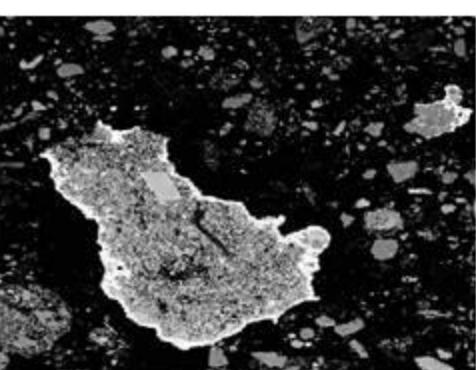
Ca-Fe-Ti-Si



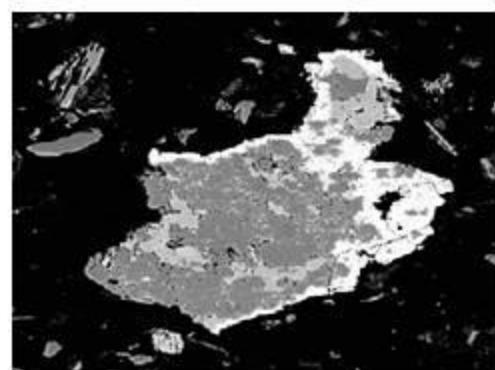
Pb-Zn-Si



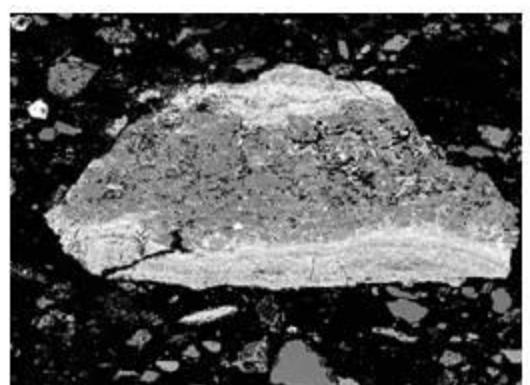
Pb-S-Si



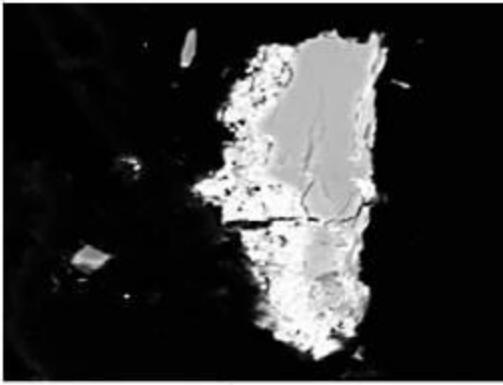
Pb-S-Zn-Si



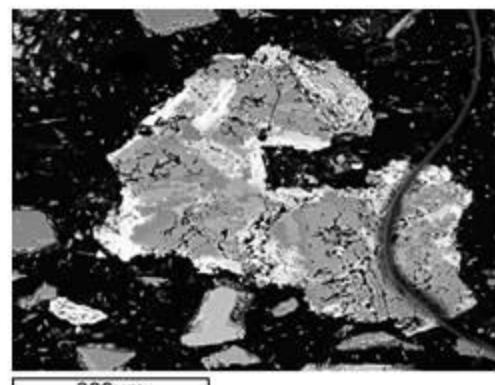
Pb-S-Zn-Si



Pb-S-Si



Pb-S-Zn-Fe-Ti-Si



Pb-S-Si

S8

Supplementary data B2  
SEM-BSE images of  
(partly) mineralized organic features in  
the soils at both sampling sites

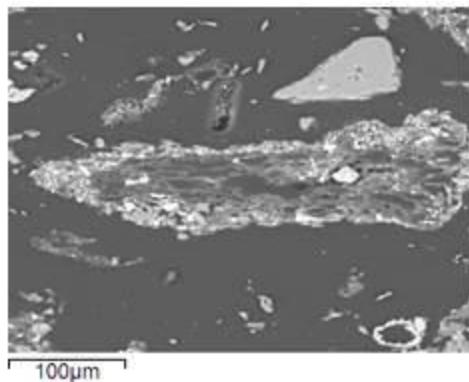
Figures

S9 North site

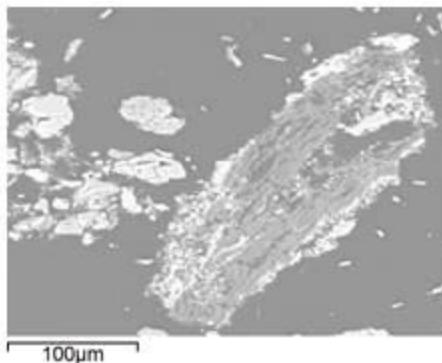
S10 North site / South-East site

All MOM are mainly composed of C, O  
and most likely H.

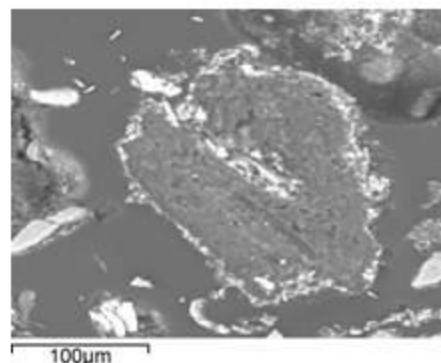
Other elements are only listed  
if their concentrations are above 1 at%



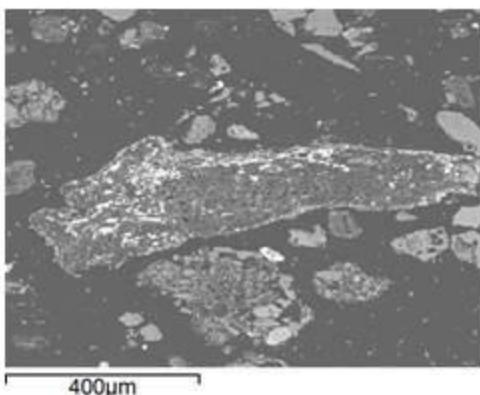
Ba-S-Al-Si



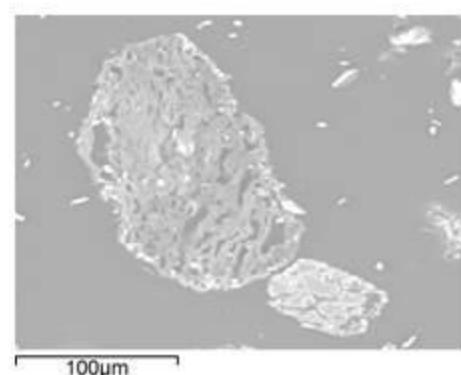
Pb-S-Fe-P-Al-Si



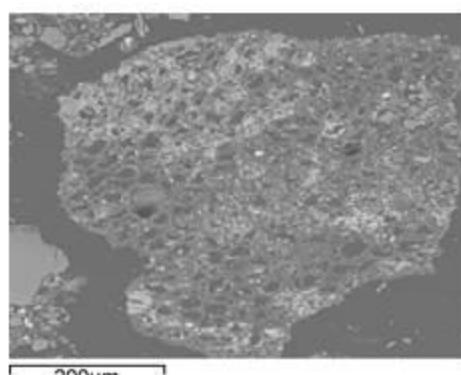
K-Mg-Fe-Al-Si



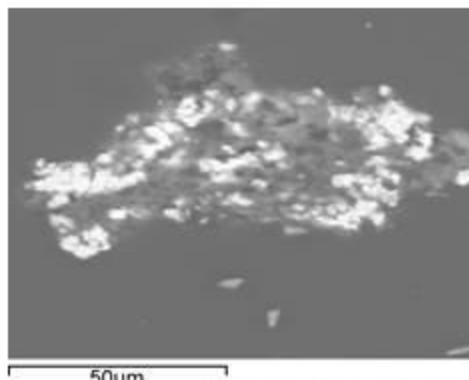
Pb-S-Al-Si



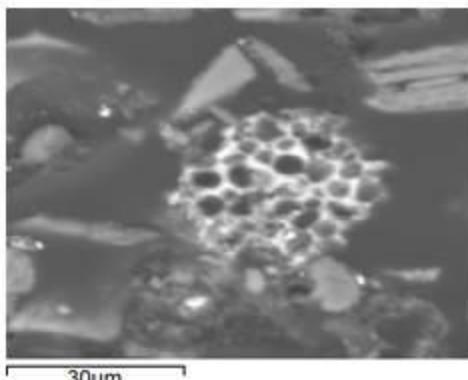
Pb-S-Fe-P-Al-Si



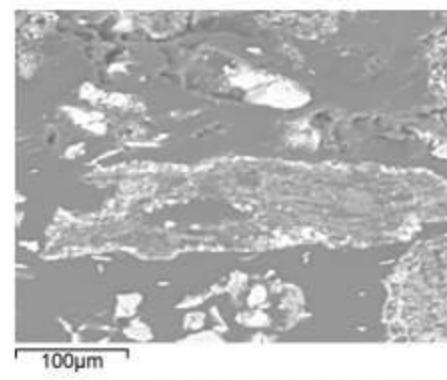
Pb-S-Fe-P-Al-Si



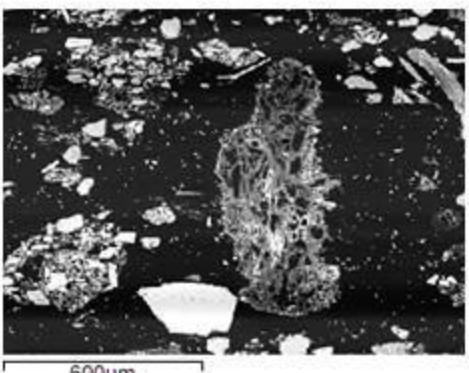
Pb-S-Fe-Al-Si



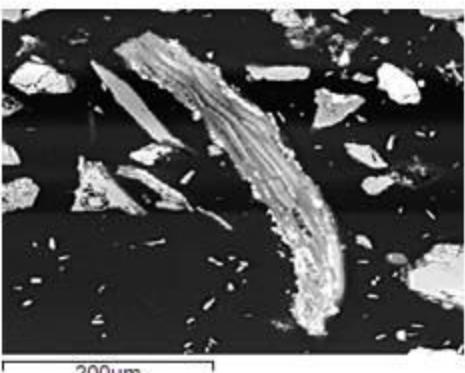
Fe-S-Ca-P-Al-Si



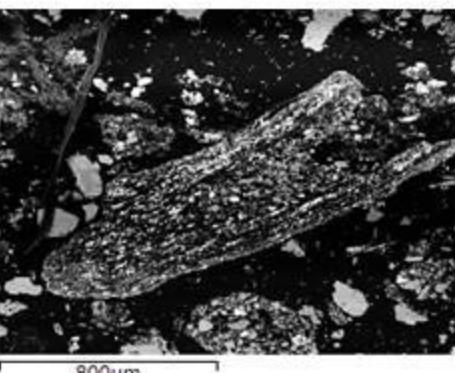
Fe-As-P-Al-Si



Pb-Sb-Fe-Mn-P-Al-Si

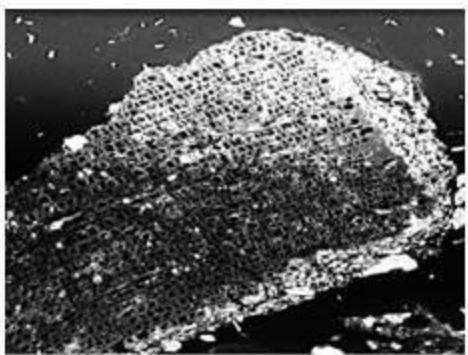


Pb-S-Fe-Si

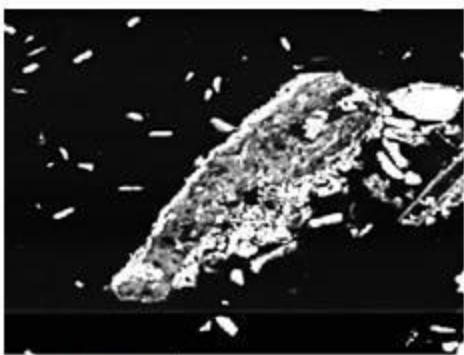


Pb-S-Ca-K-Na-P-Al-Si

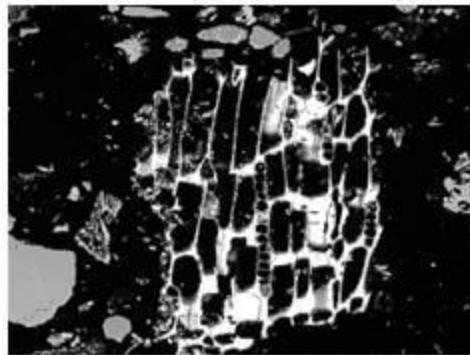
NE-site



Fe-P-Al-Si

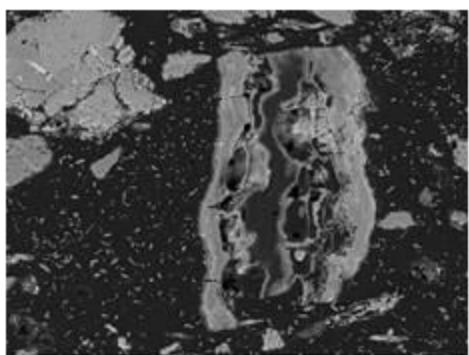


Fe-S-P-Al-Si

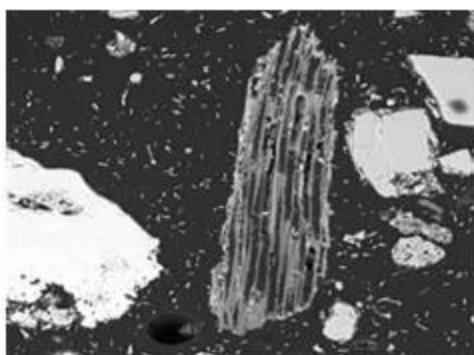


Pb-S-Fe-Al-Si

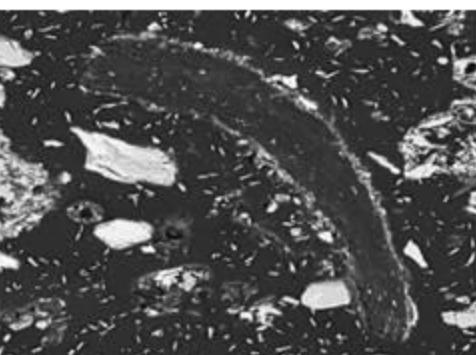
SE-site



Pb



Pb-Fe-Al-P-Si

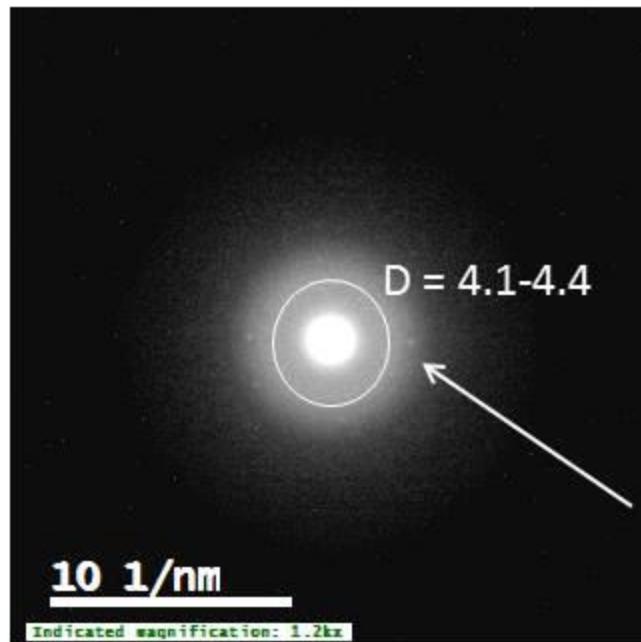
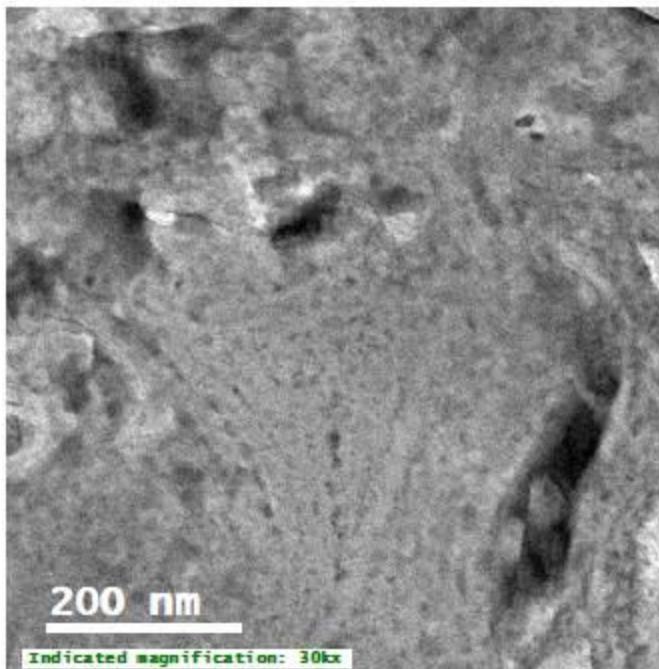
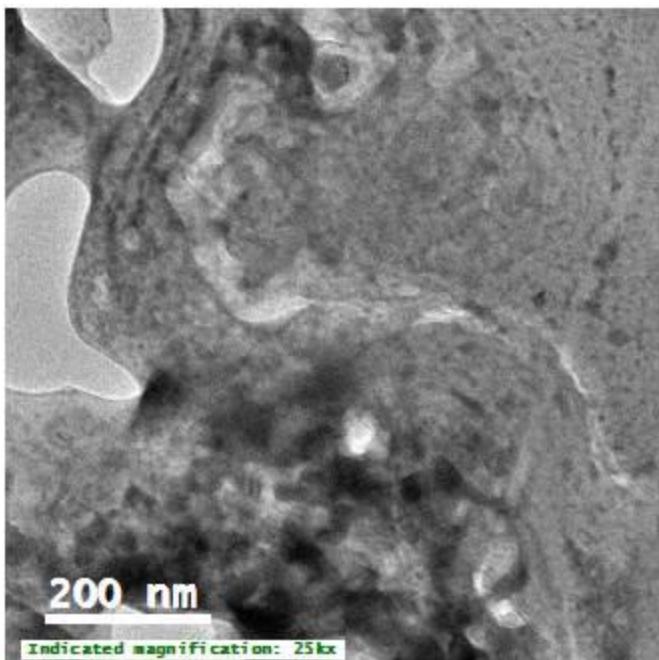


Zn-S-Al-Si

S10

# Supplementary data C1

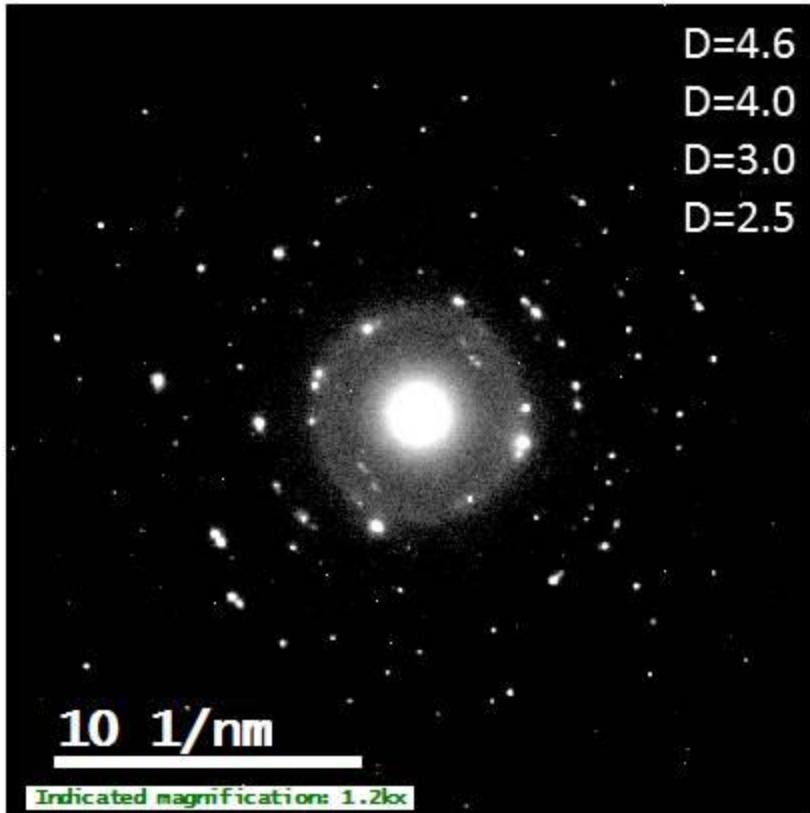
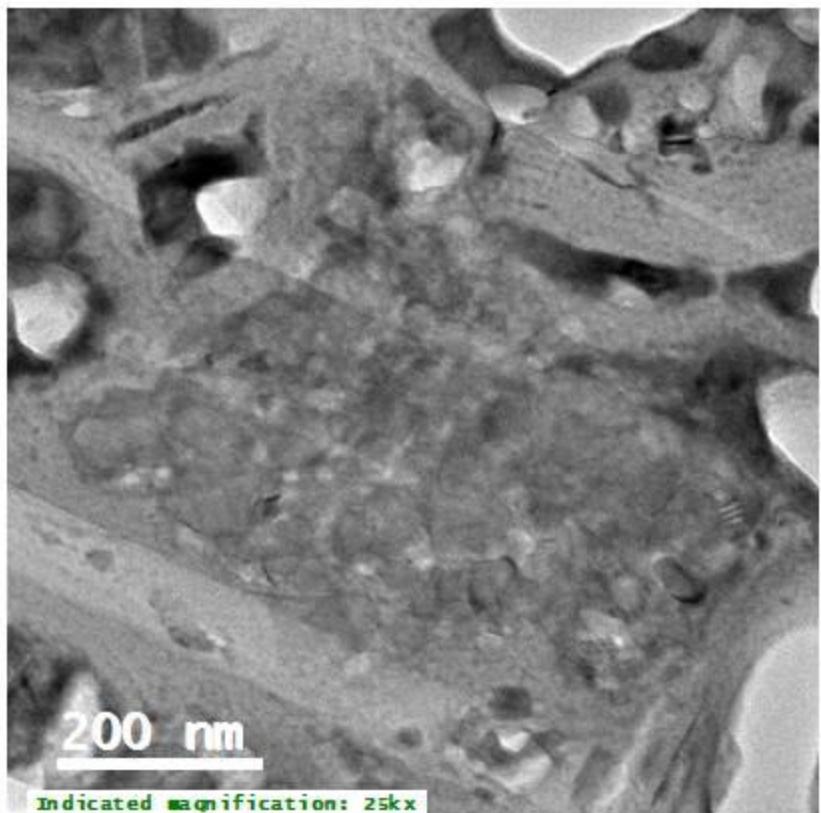
Nano-phases TEM images and  
diffraction pattern for the mineral  
surface coating



## Silica matrix

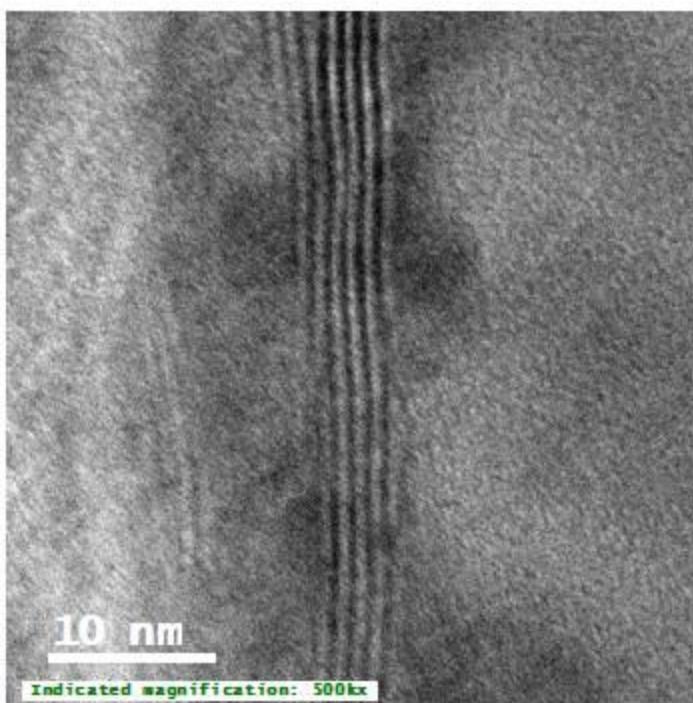
Broad peak between  $d = 4.1$  and  $4.4 \text{ \AA}$   
characteristic for Opal A/C/CT

Diffraction spots for anglesite nanoparticles



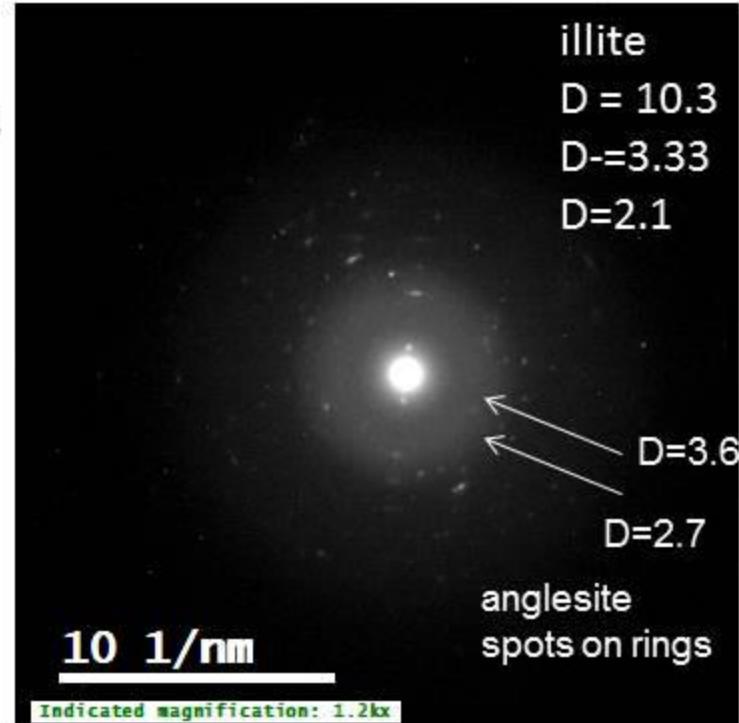
Fluorapatite,  $(\text{Ca}+\text{Pb}) : \text{P} : \text{F} = 5 : 3 : 1$   
 $\text{Ca} : \text{Pb} = 10 : 1$

# Illite+ Nanoparticles of anglesite



10 nm

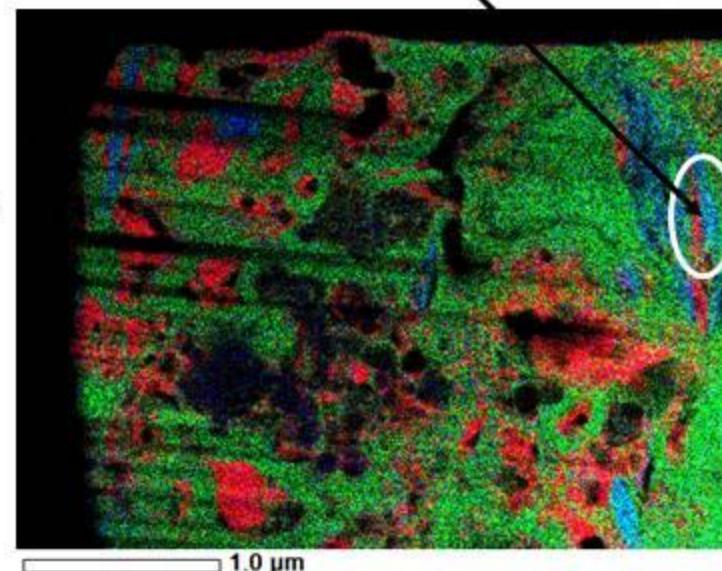
Indicated magnification: 500kx



10 1/nm

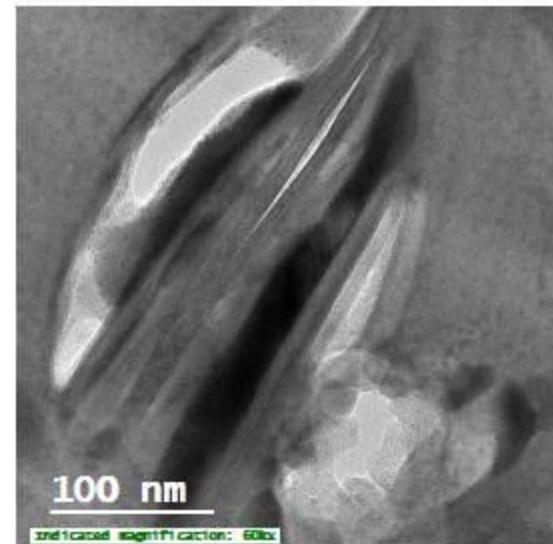
Indicated magnification: 1.2kx

K: Al: Si = 1: 3: 7

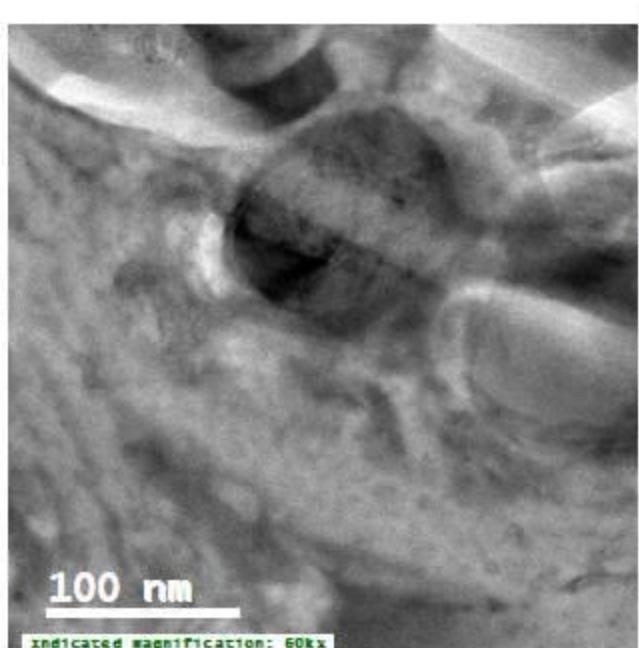
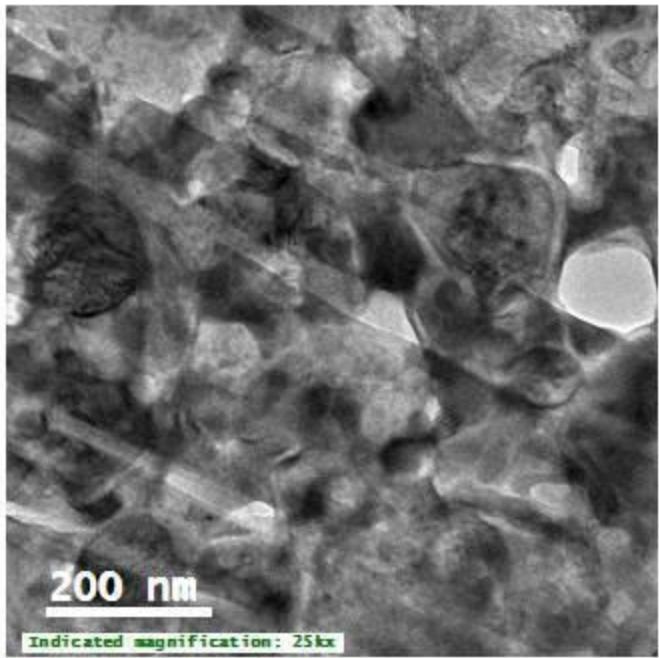


Si: green  
Al: Blue  
Pb: red

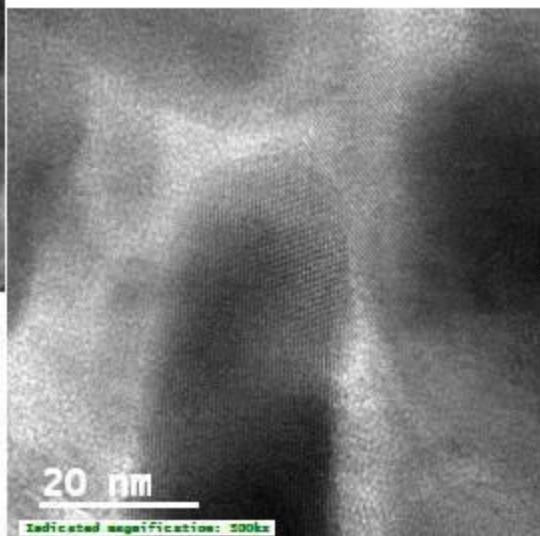
Parallel growth of anglesite  
on the surface of illite



S13

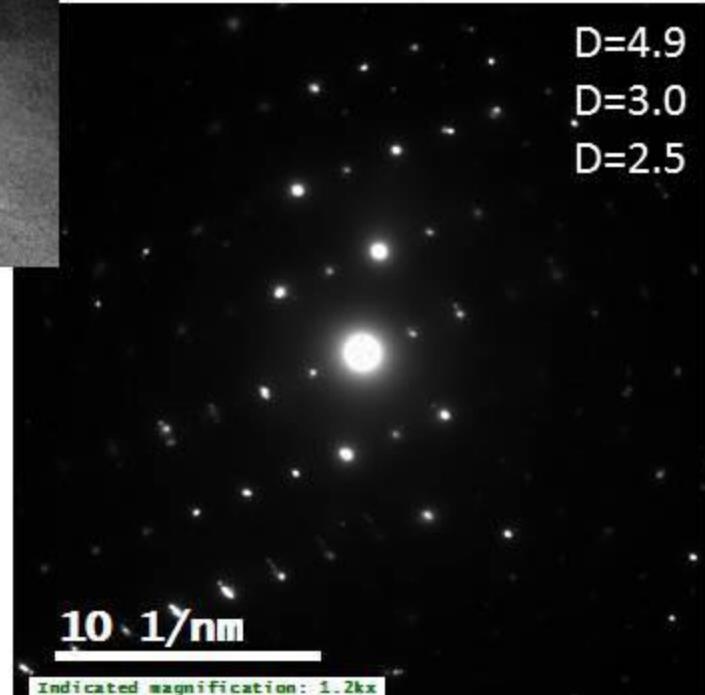
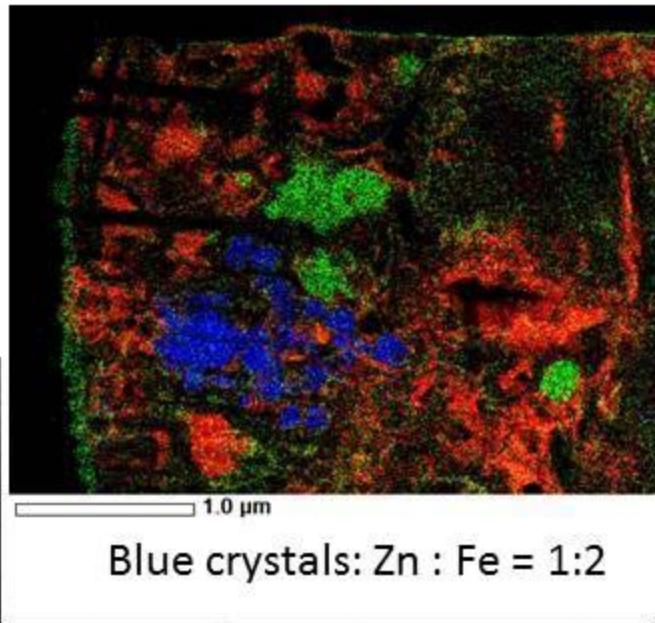


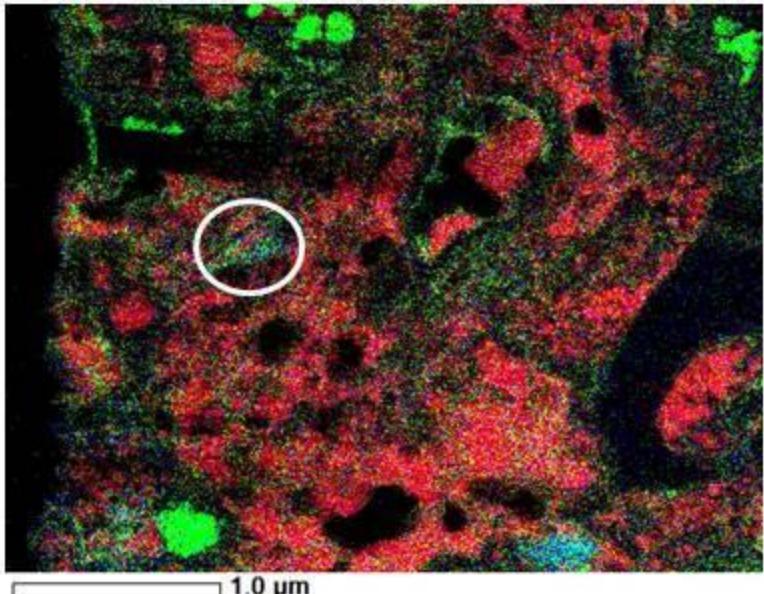
Pb: red  
P: green  
Zn: blue



Detrital rounded grains of  
Franklinite,  
 $\text{ZnFe}_2\text{O}_4$

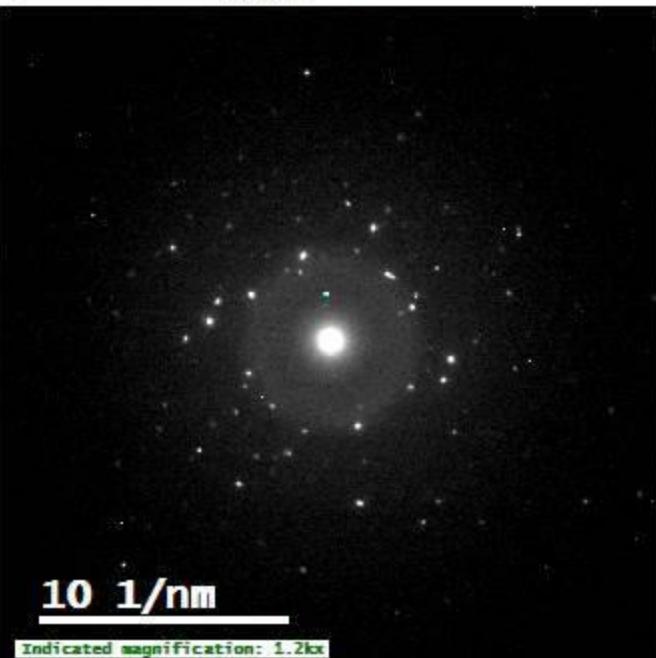
S14





Pb: red  
Fe: green  
As: blue

Fe : (As+S) = 1: 1



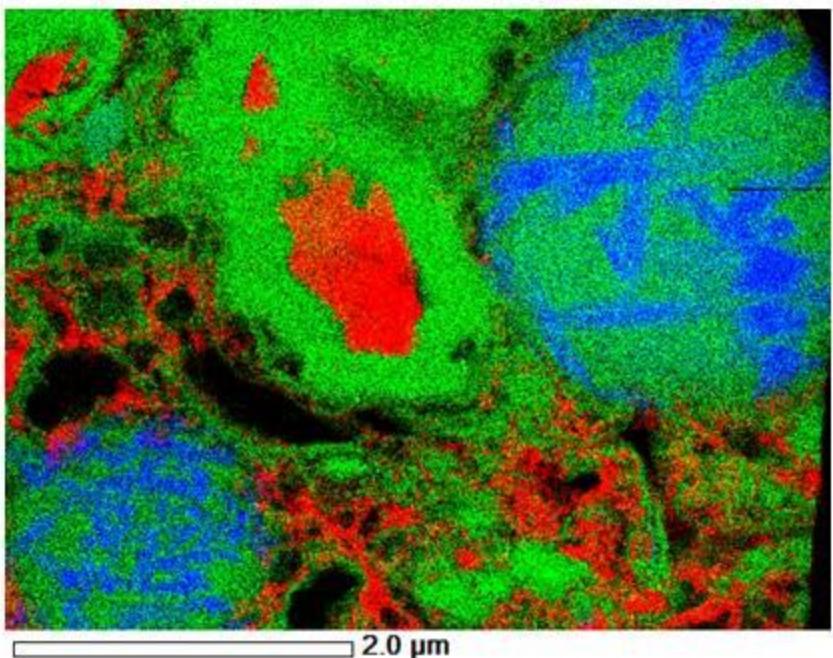
### Tsumcorite Group

**Formula:**  $AM(XO_4)^2(OH, H_2O)^2$

D=6.86  
D=3.4  
D=2.8  
D=2.5  
A group of complex phosphates, arsenates, vanadates and sulphates with a complicated crystal chemistry (symmetry either triclinic or monoclinic, depending on various ordering schemes).

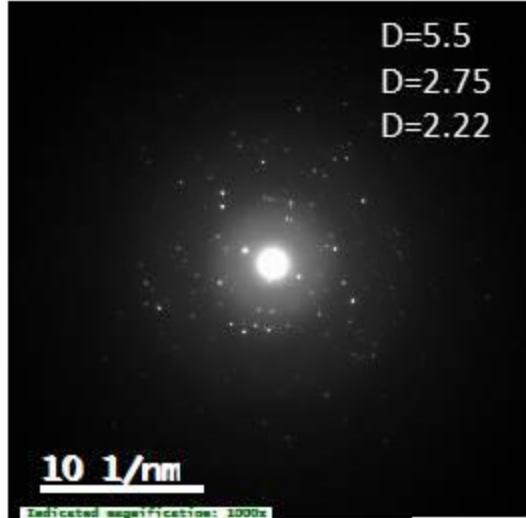
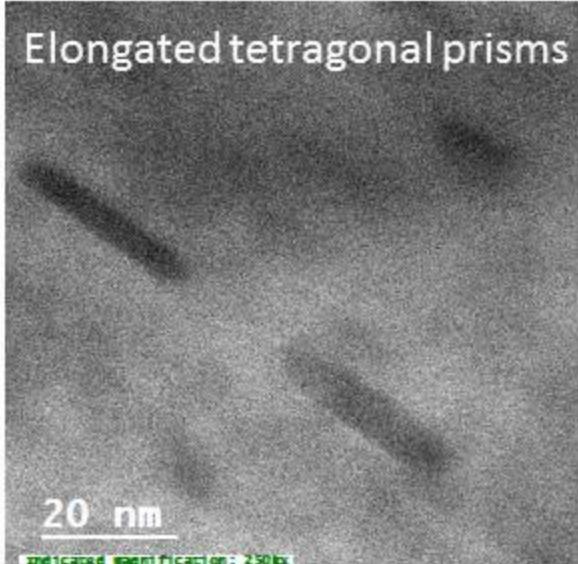
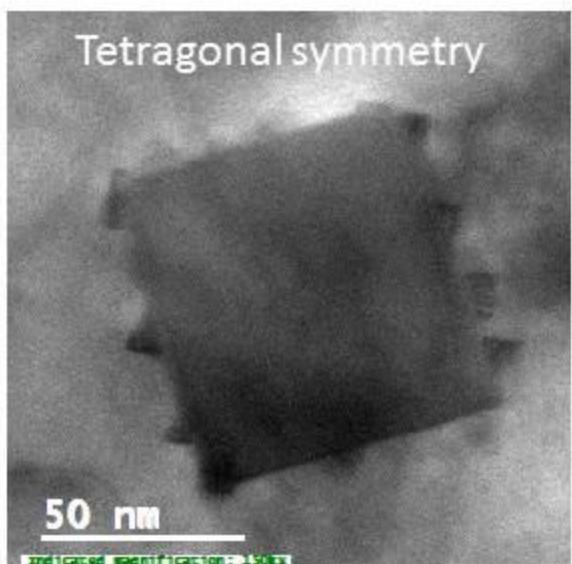
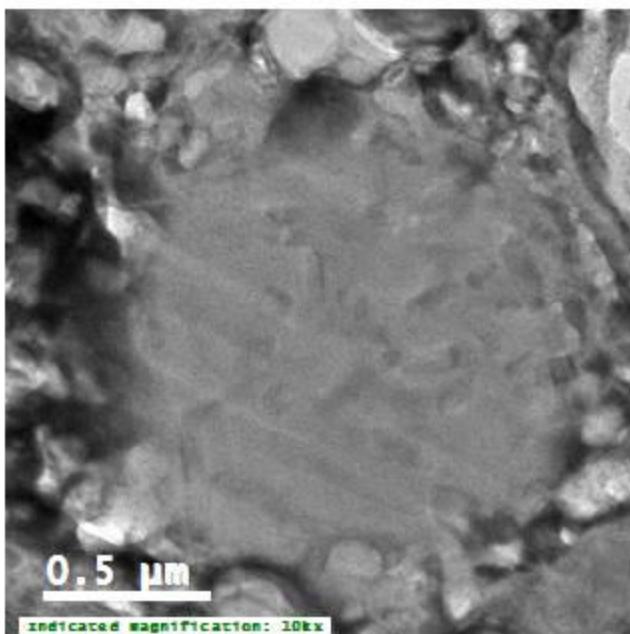
In the general formula given above, A is typically Pb or Ca, rarely Bi, M is typically  $Fe^{3+}$  Mn, Cu, Zn, Co or Ni and X is typically P, As or V

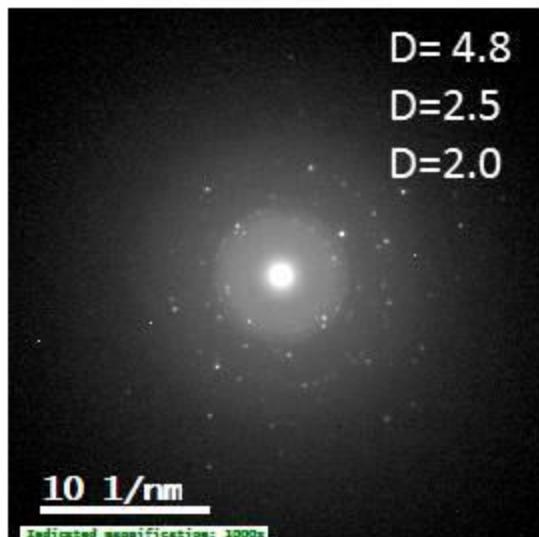
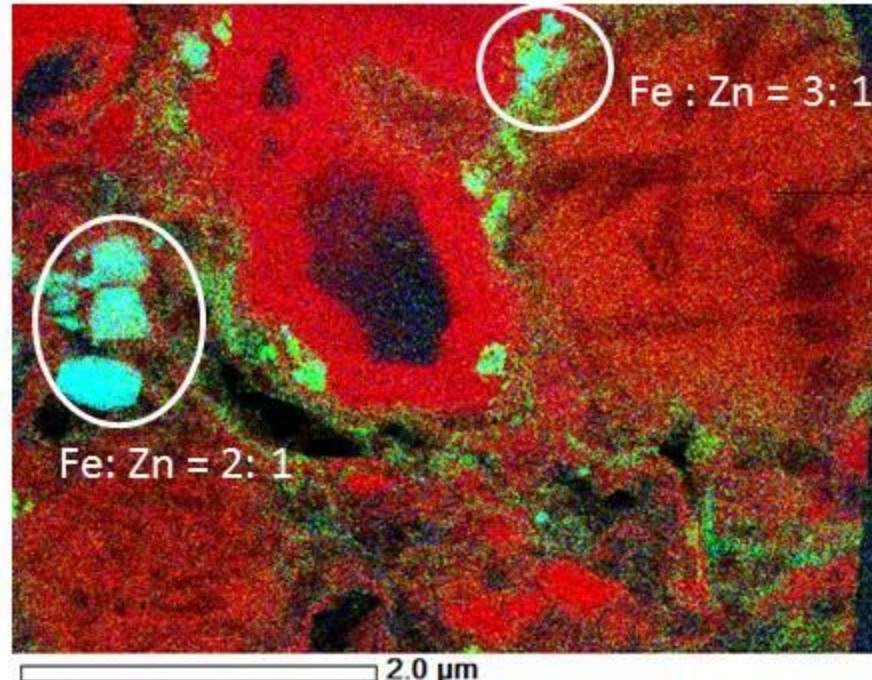
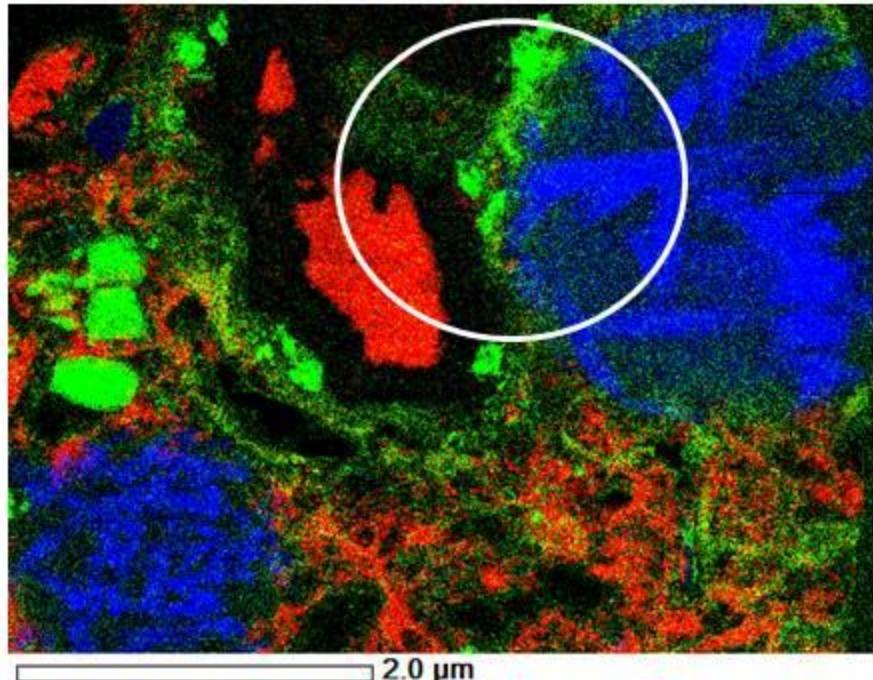
# andalusite crystals (blue) in spherical quartz grains (green)



Pb: red  
Si: green  
Al: blue

Blue crystals:  
Al : Si = 2 : 1

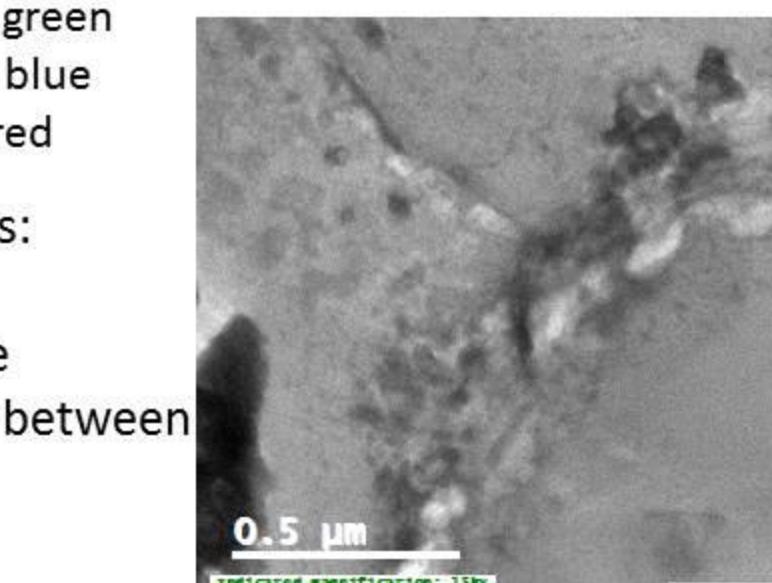


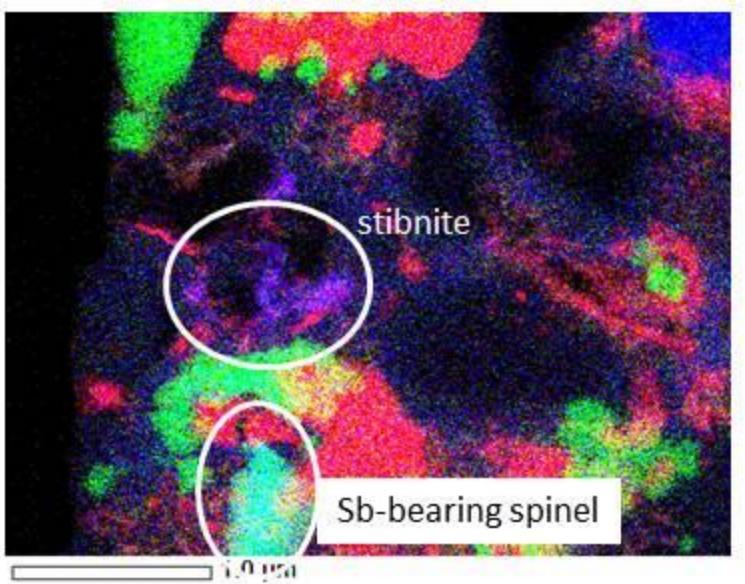


Fe: green  
Al: blue  
Pb: red

Fe: green  
Zn: blue  
Si: red

Zn-bearing spinels:  
Franklinite  
and Zn-magnetite  
Fe: Zn ratios vary between  
2 : 1 and 3 : 1





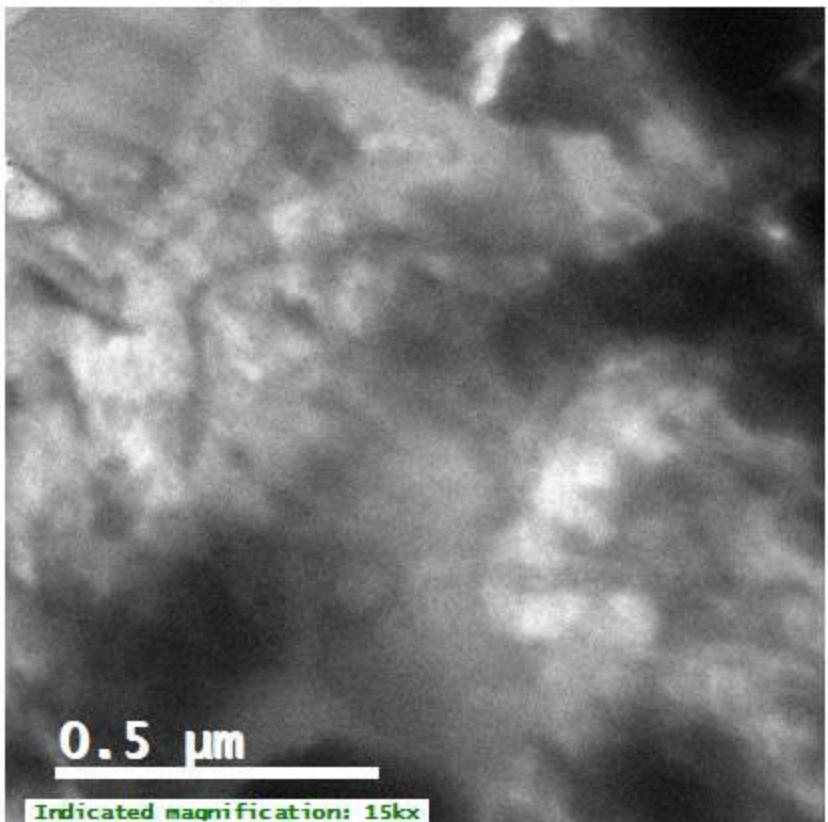
Pb: red  
Zn: green  
Sb: blue

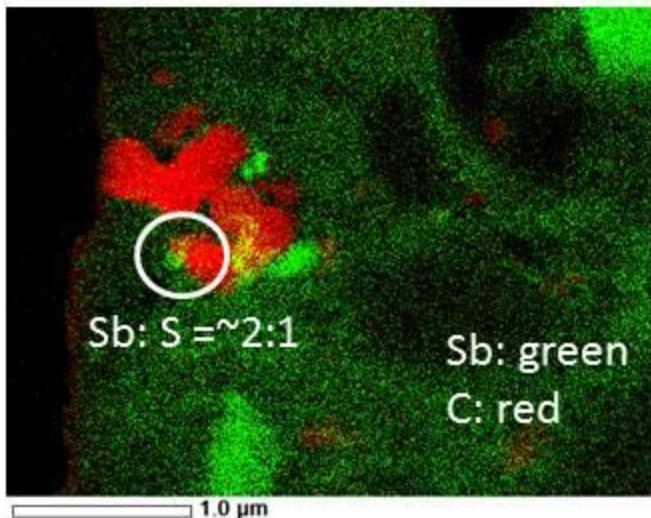
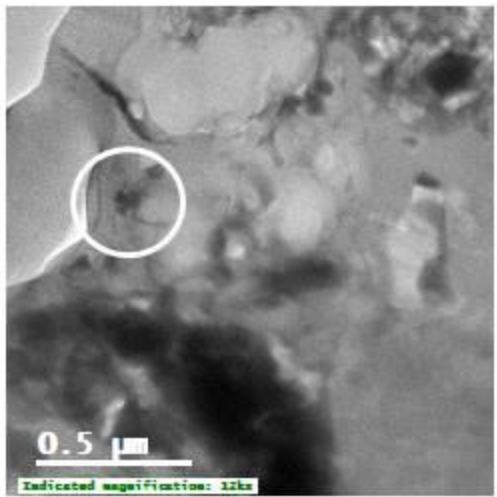
Mixture of various spinels

Fe : Zn : Sb = 4 : 7 : 2

Possible Sb-spinel:

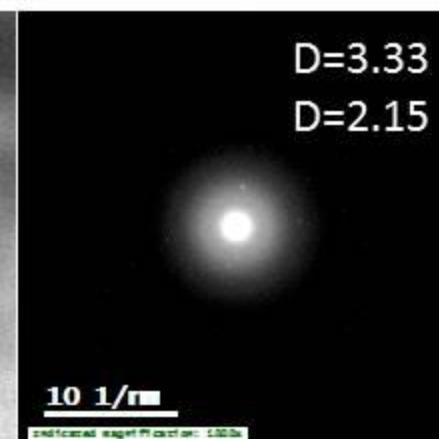
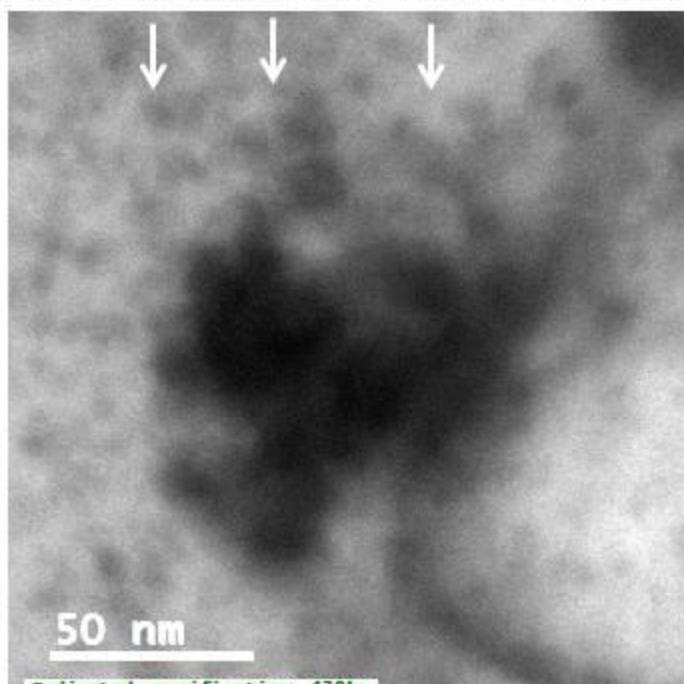
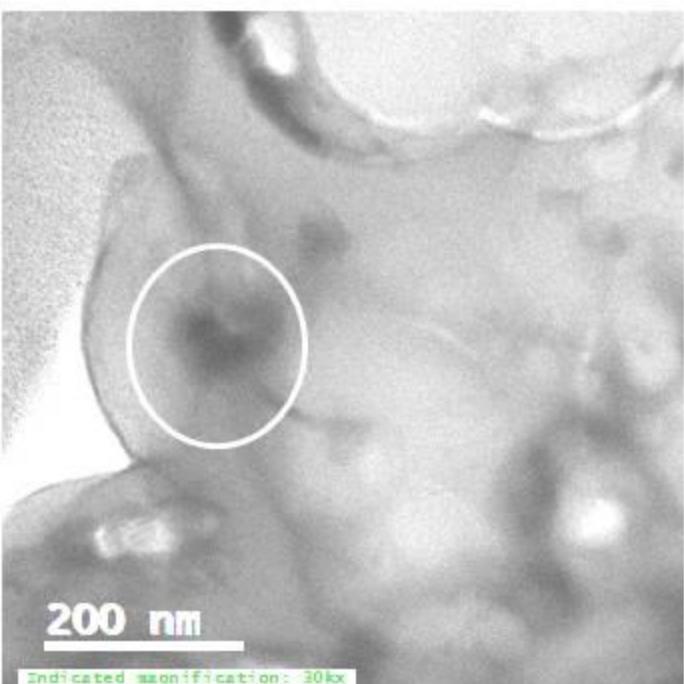
Zn-Sb(V) spinel  $\text{Zn}_7\text{Sb}_2\text{O}_{12}$





carbonaceous material (red) with Sb-bearing nanoparticles inclusions (green)

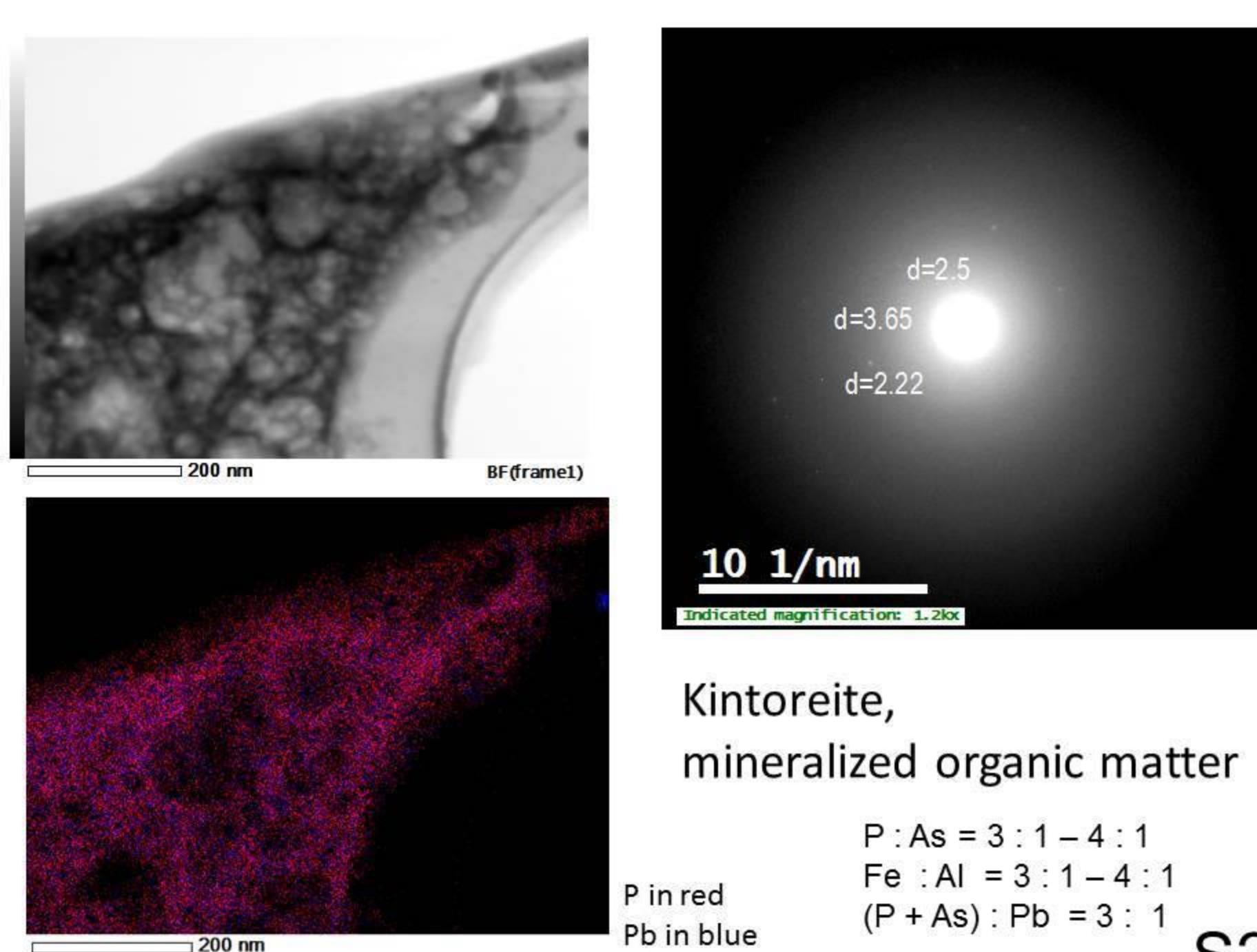
Unidentified Sb-bearing Nanoparticles

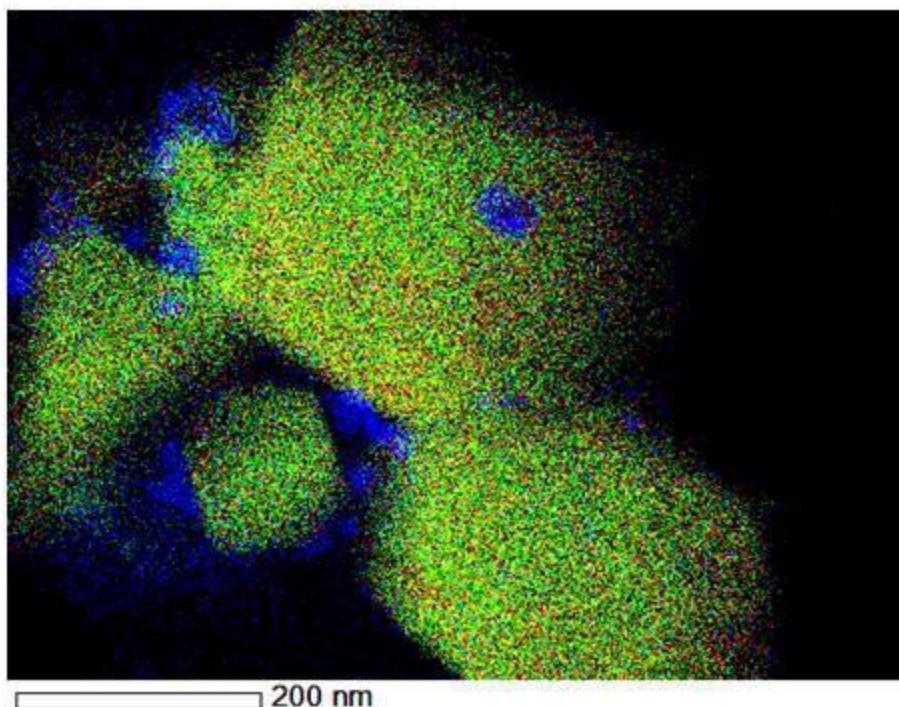
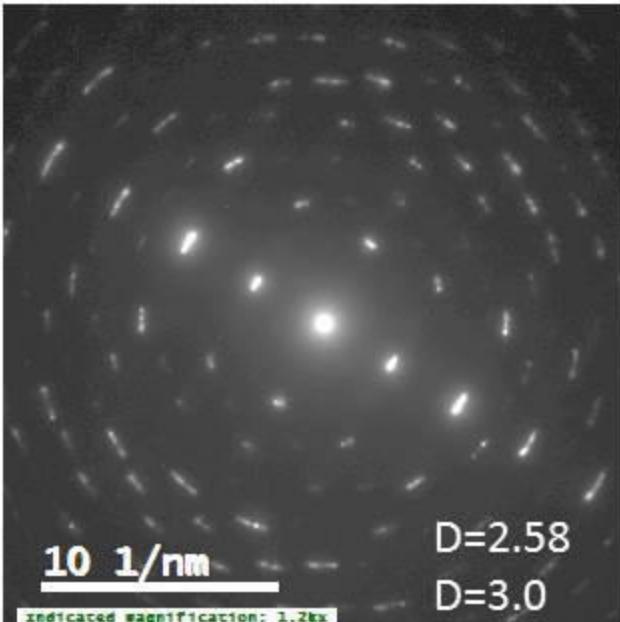
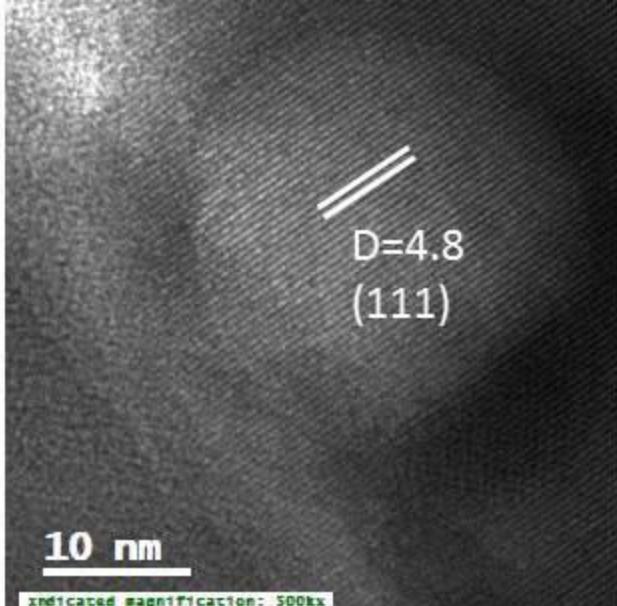
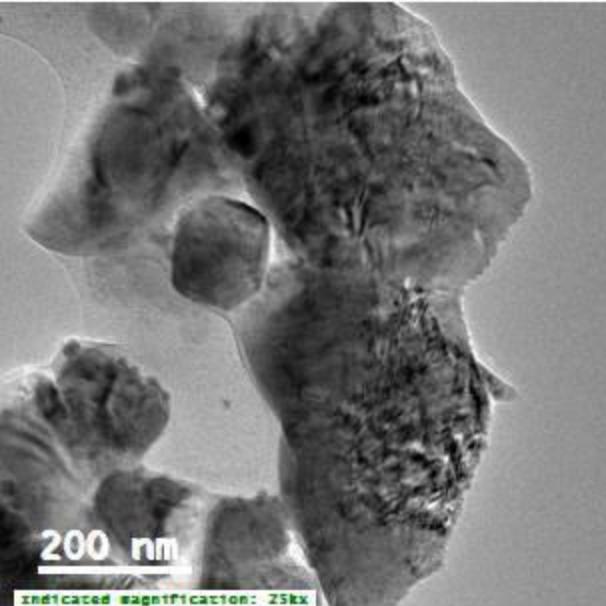


# Supplementary data C1

TEM, SAED and EDS-STEM images of  
nano-size phases in the mineralized  
organic matter

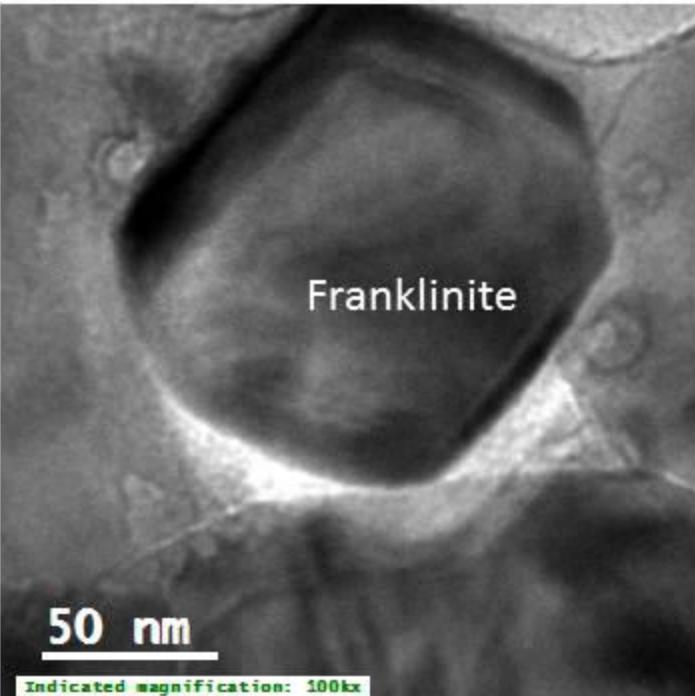
All listed elemental ratios are given as atomic ratios



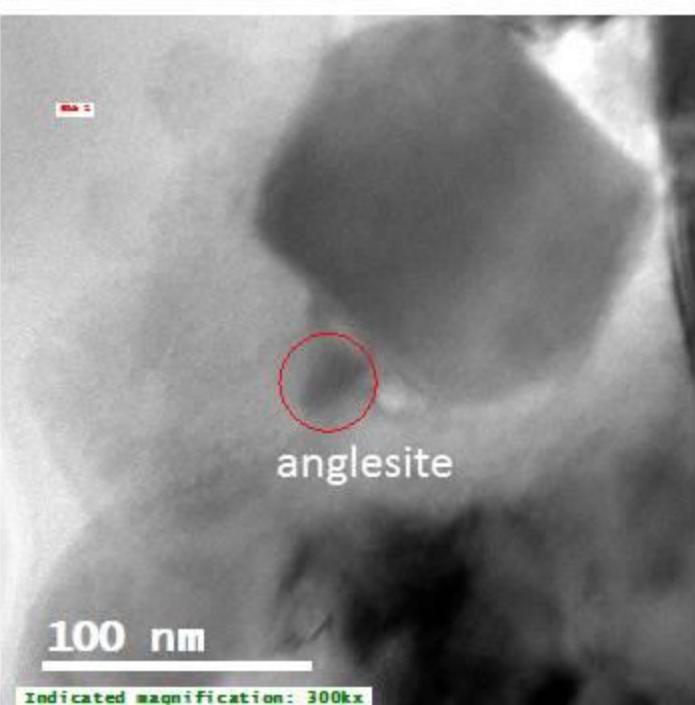


Franklinite with anglesite nanoparticles in the mineralized organic matter

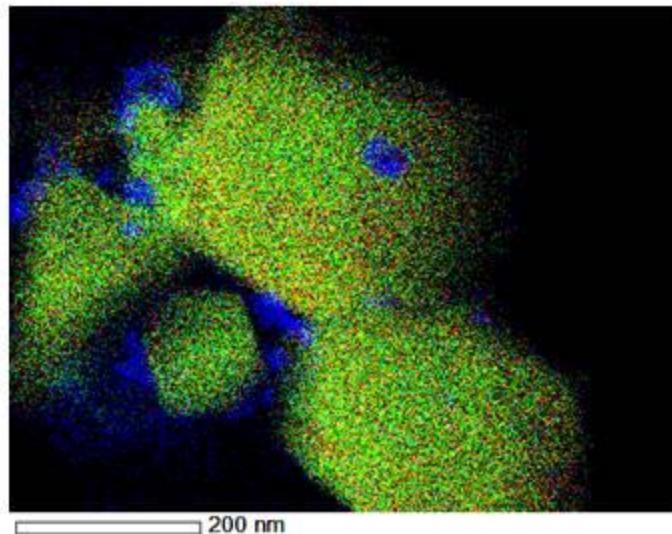
Zn in red       $\text{Zn} : (\text{Al} + \text{Fe}) = 1 : 2$   
Fe in green  
Pb in blue



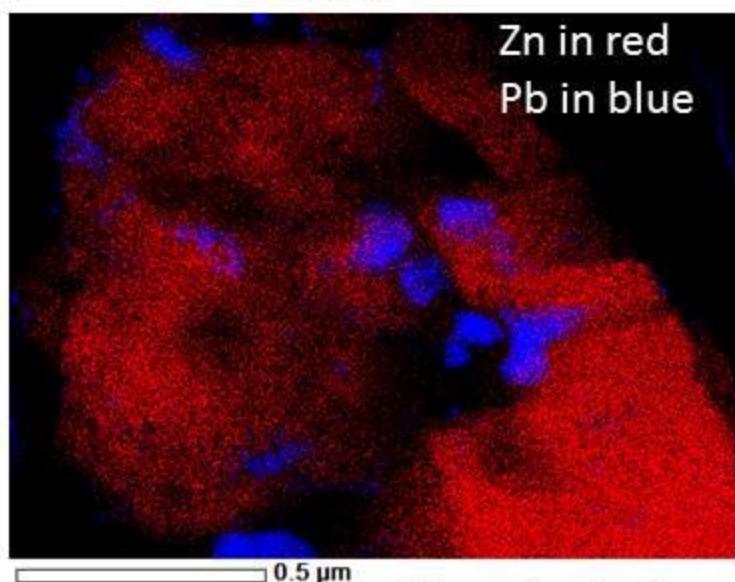
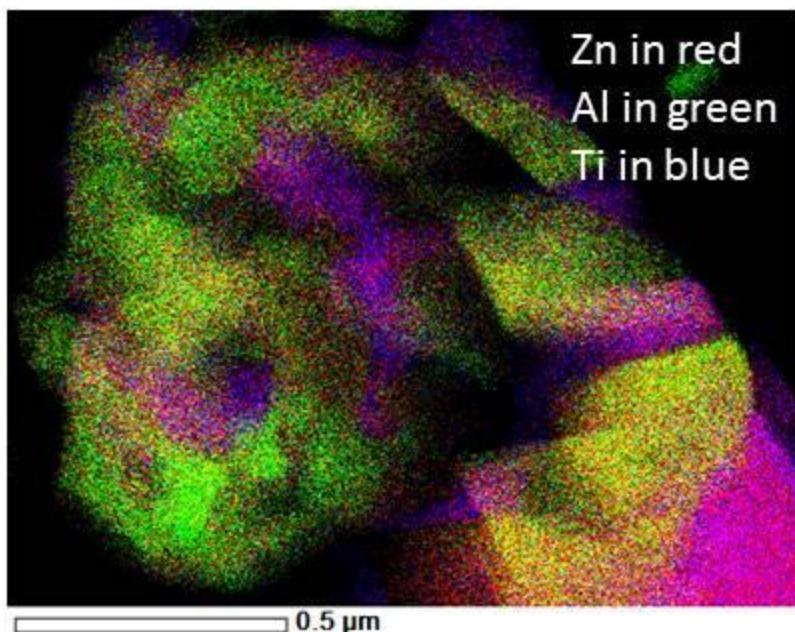
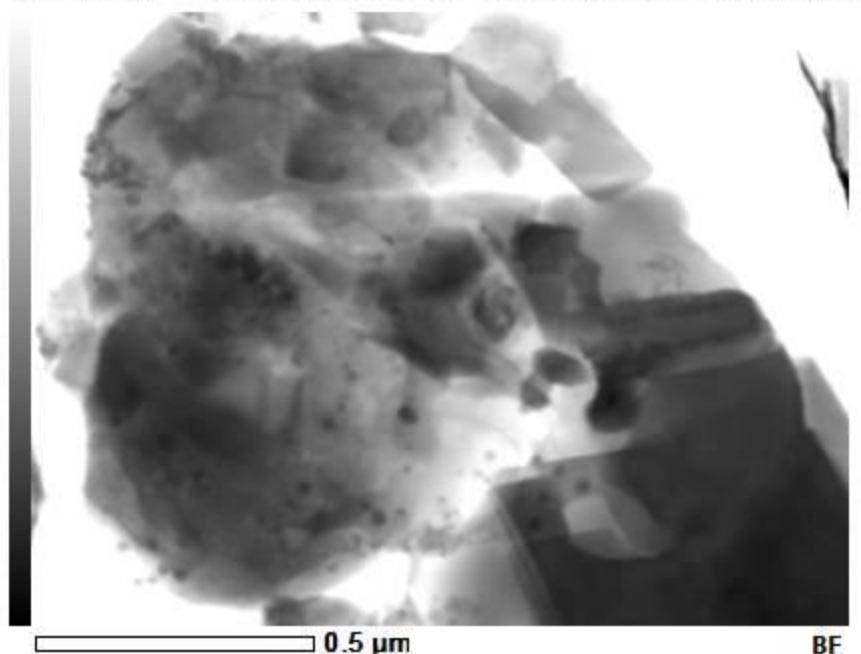
Anglesite spots  
D=5.4  
D=3.8  
D=1.4



Nano-diffraction of anglesite nanoparticles on the surface of franklinite with spots belong to franklinite

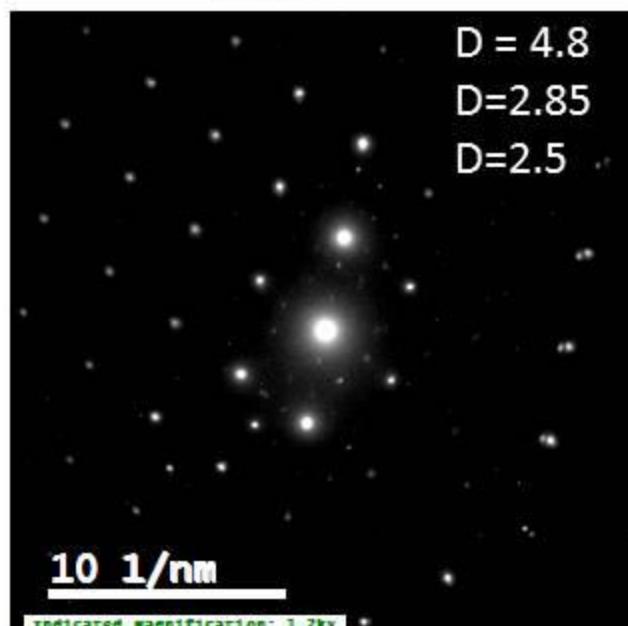


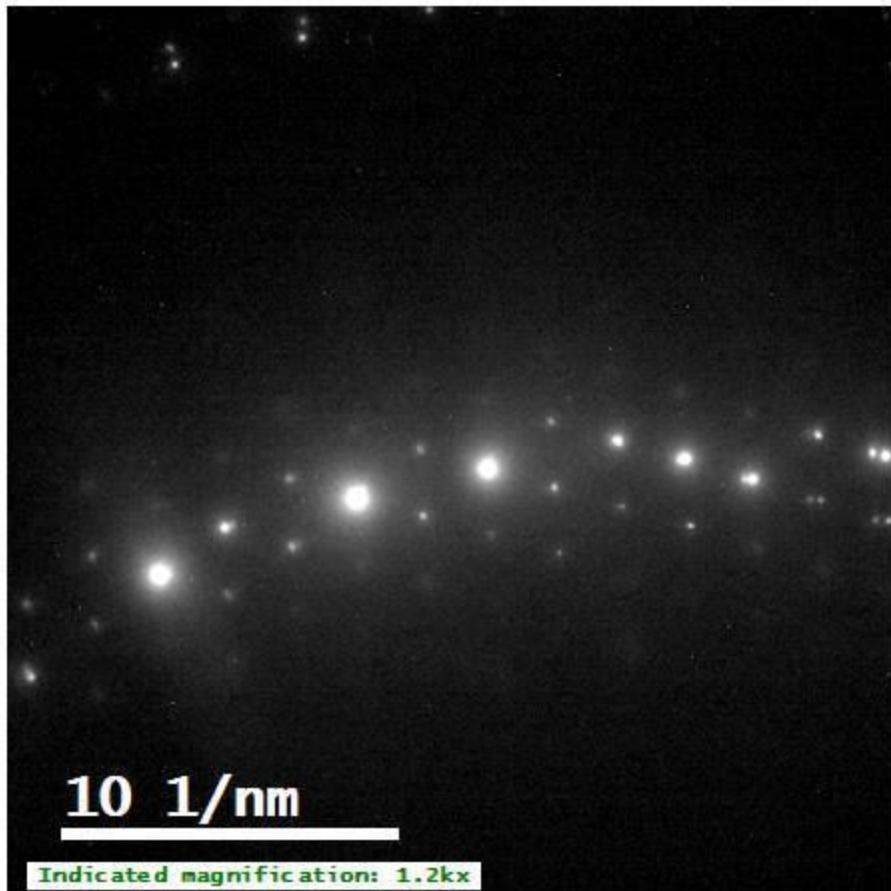
Intergrowth of crystals of gahnite ( $\text{Zn} : \text{Al} = 1:2$ , yellow/green) and a  $\text{Zn-Al-Ti-(hydr)oxide}$  phase ( $\text{Zn} : \text{Al} = 4:1$ ; violet) SAED pattern display most commonly diffraction spots of both phases



blue: Anglesite nanoparticles

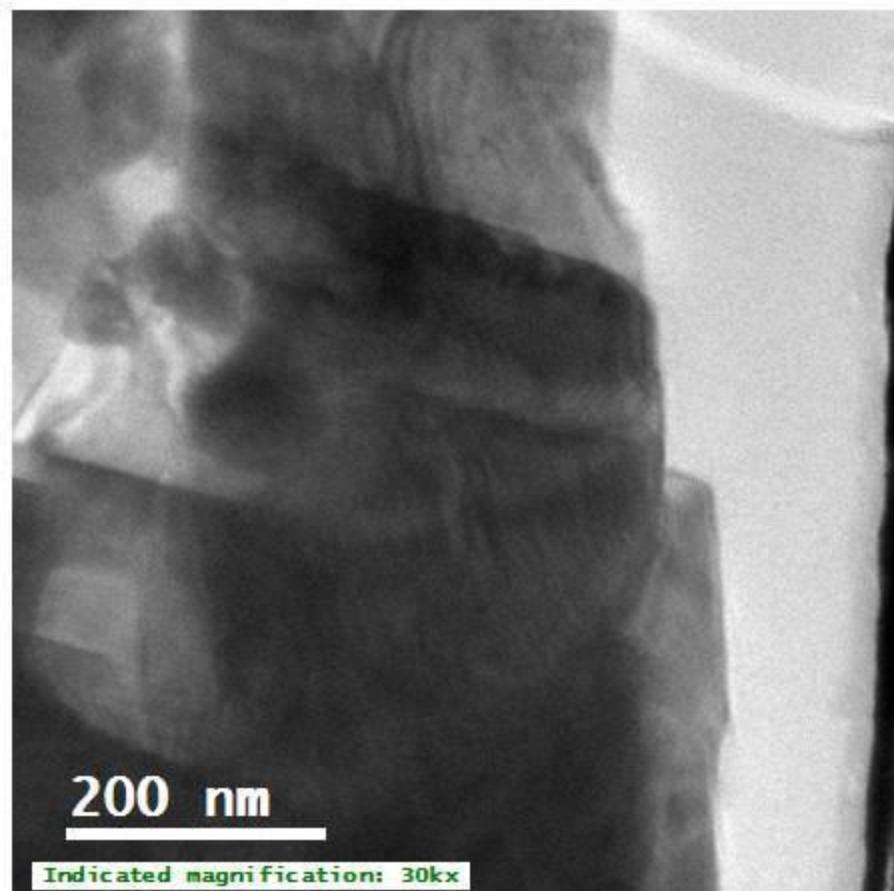
Diffraction pattern of Gahnite,  $\text{ZnAl}_2\text{O}_4$





10 nm

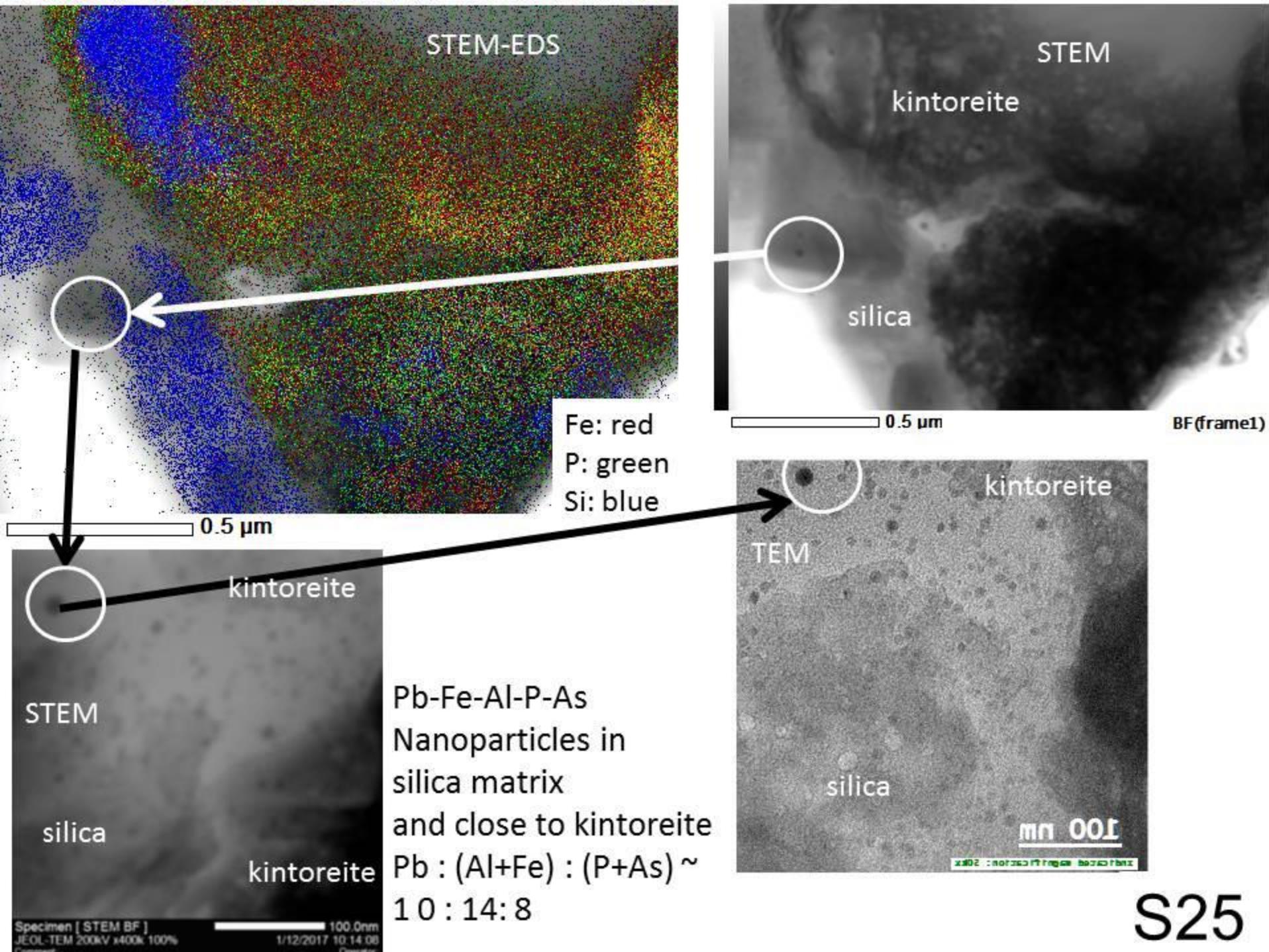
Indicated magnification: 1.2kx



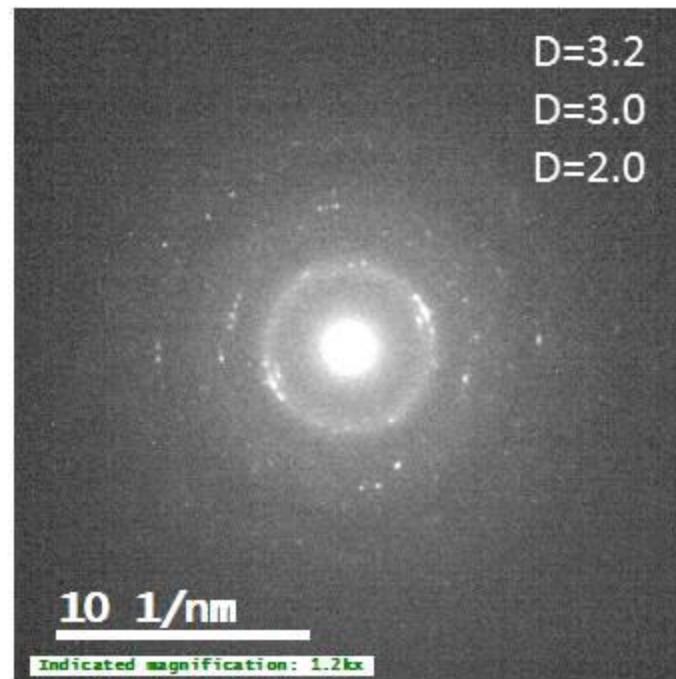
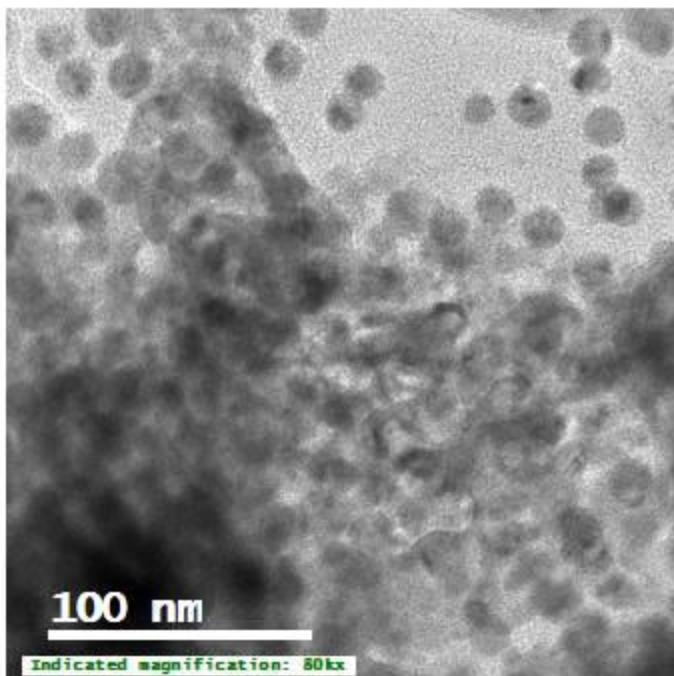
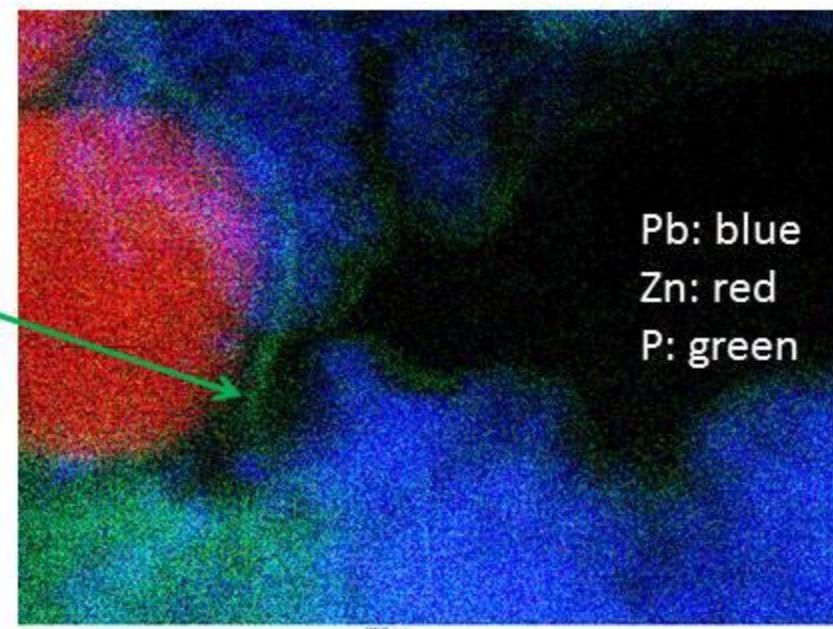
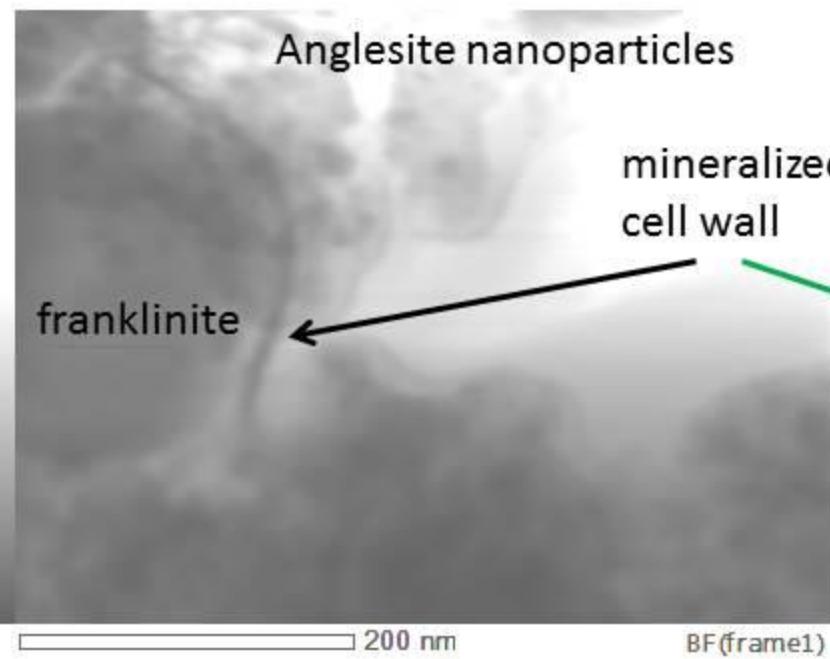
200 nm

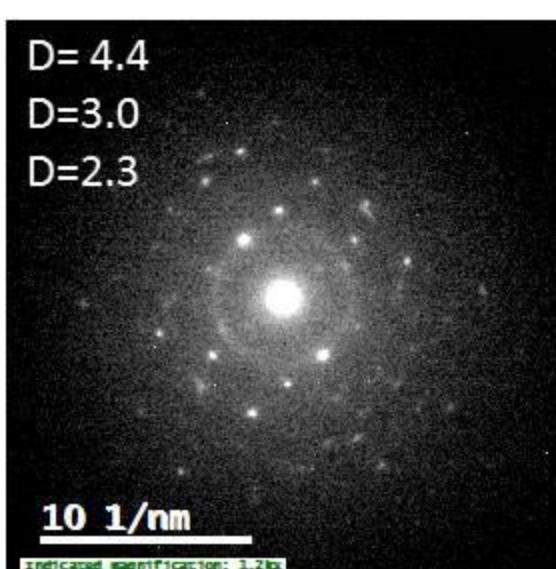
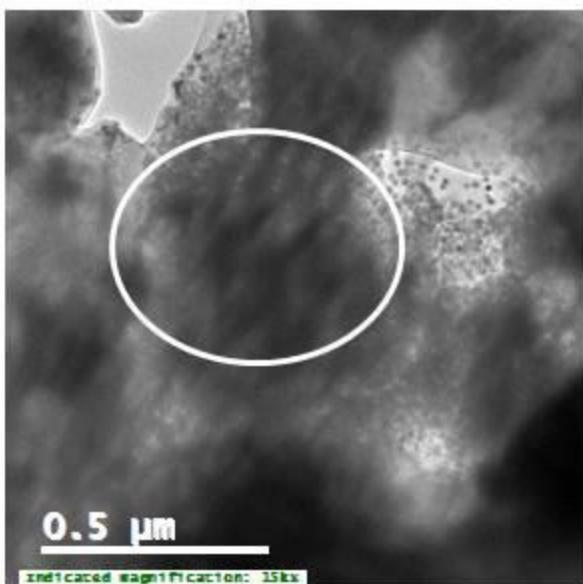
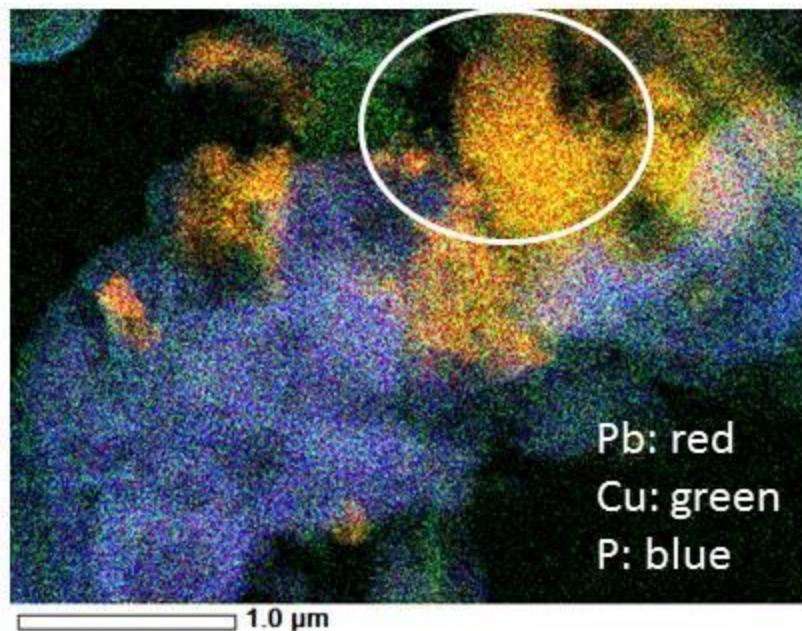
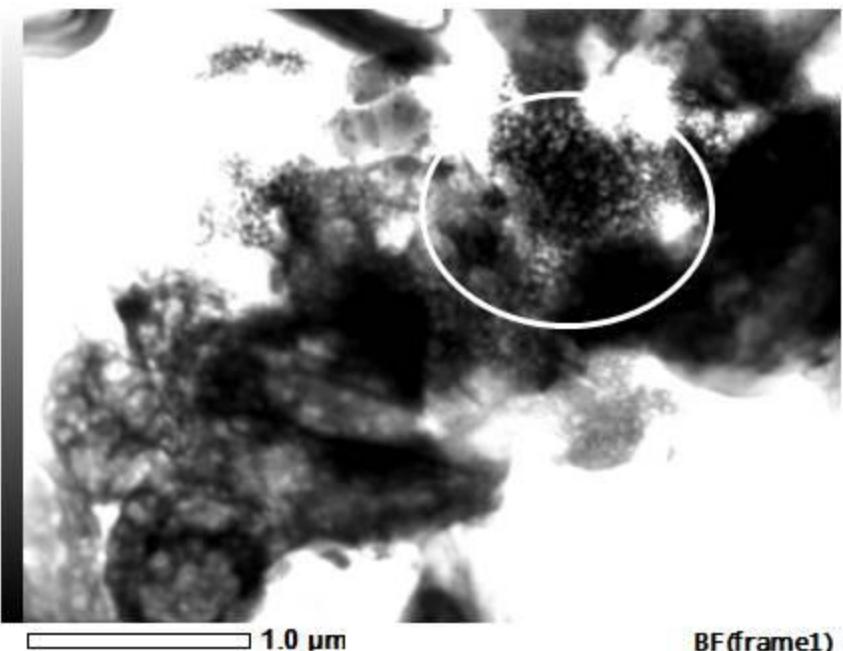
Indicated magnification: 30kx

The Zn-Al-Ti oxides in this sections are composed of hydroxides with mainly Zn. The latter layers are decorated with tetrahdra or OH groups and have characteristics d-spacings of  $d = 7.65$  and  $d = 4.0$

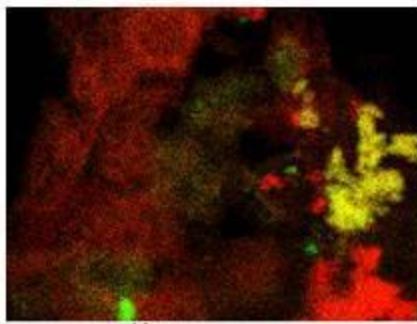
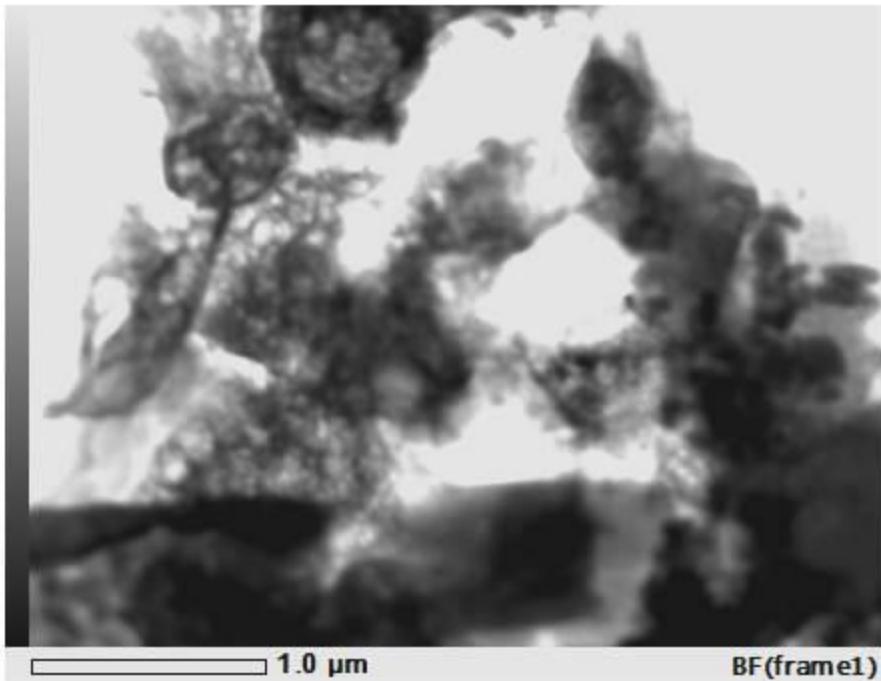


Anglesite nanoparticles

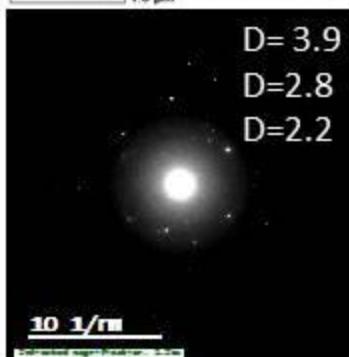




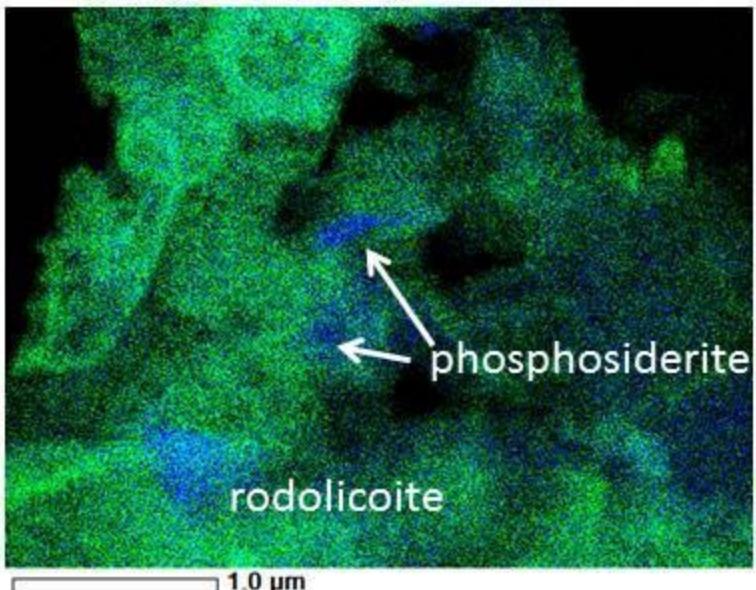
Intergrowth of linarite  
 $\text{PbCu}(\text{OH})_2(\text{SO}_4)$   
(Yellow, Cu : S ~ 1: 1)  
with kintoreite (blue)



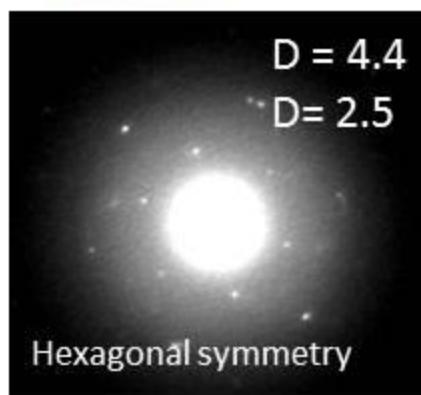
Pb  
Ti



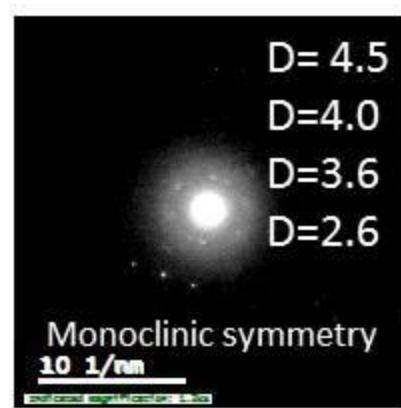
yellow Pb : Ti ~ 1: 1  
macedonite,  $\text{PbTiO}_3$



As  
P  
As : P ~ 2 : 1



Blue:  
rodolicoite  
 $\text{Fe(PO}_4\text{)}$



Blue:  
phosphosiderite,  
 $\text{Fe(PO}_4\text{)(H}_2\text{O})_2$   
Fe : (As+P) ~ 1 : 1

# Supplementary data D, Table S29

Selected minerals identified in the MSC and MOM and their point of zero charge

Mineral	Point of zero charge	
anglesite	4 <sup>3</sup>	1 M. Kosmulski, <i>J. Coll. Interface Sci.</i> (2009), <b>377</b> , 439-448.
Synthetic hydroxyapatite	7.7-8.1 <sup>1</sup>	2 M. Kosmulski, <i>Adv. Coll. and Interface Sci.</i> (2016), <b>238</b> , 1-61.
Kintoreite	~3-6	
Fe <sub>3</sub> (PO <sub>4</sub> ) <sub>3</sub>	3.3 <sup>5</sup>	3 M.C. Fuerstenau, S.A. Olivas, R. Herrera-Urbina, and K.N. Kank, <i>Inter. J. Mineral Process.</i> (1987), <b>20</b> , 73-85.
Fe <sub>3</sub> (PO <sub>4</sub> ) <sub>2</sub> (H <sub>2</sub> O) <sub>8</sub>	4.0-5.6 <sup>1</sup>	
FePO <sub>4</sub> (am)	3 <sup>4</sup>	4. G. De Tommaso, and M. Luliano, <i>J. Chem. Eng. Data</i> (2012) , <b>57</b> , 52–59
Al-rich spinel (amorphous MgAl <sub>2</sub> O <sub>4</sub> )	9 <sup>1</sup>	5. D. Luna-Zaragoza, E.T. Romero-Guzmán and L.R. Reyes-Gutiérrez, <i>J. Minerals &amp; Materials Character &amp; Eng.</i> (2009), <b>8</b> , 591-609.
Fe-rich spinel (magnetite)	6.2-8.5 <sup>1</sup>	
hydrous silica (listed as precipitated silica)	4.1 <sup>1</sup>	
illite	2.5 <sup>1</sup>	
Quartz	2-3 <sup>2</sup>	