

**On the development of a chemical shift-based scale for weak halogen
bonding interactions; a ^{15}N -NMR and DFT study**

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Supporting Information

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1. Observed chemical shifts

Table S1: Chemical shifts of the nitrogen and 2H in pyridine when mixed with various halogen bond donors following 5:1 stoichiometric experiments in toluene-d₈. In addition, the Hammett para-constant σ_{para} and the changes in chemical shift $\Delta\delta$ between pyridine mixed with halogen bond donors and pyridine mixed with the same equivalents of benzene are displayed. The substituents under "R" and halogens under "X" correspond to the structure displayed in Figure 1.

R	X	σ	$^1\text{H}\delta$ (ppm)	$^{15}\text{N}\delta$ (ppm)	$^1\text{H}\Delta\delta$ (ppm)	$^{15}\text{N}\Delta\delta$ (ppm)
<i>(benzene reference)</i>		-	8.1997	-58.0056	0	0
OCH ₃	I	-0.27	7.9363	-59.2809	-0.2634	-1.2753
	Br		8.0569	-58.5915	-0.1428	-0.5859
	Cl		8.1008	-59.0427	-0.0989	-1.0371
CH ₃	I	-0.17	7.9869	-58.7460	-0.2128	-0.7404
	Br		7.9647	-58.5445	-0.2350	-0.5389
	Cl		8.1434	-58.2335	-0.0563	-0.2279
H	I	0	7.9894	-58.9422	-0.2103	-0.9366
	Br		8.0756	-58.5297	-0.1241	-0.5241
	Cl		8.1442	-58.2789	-0.0555	-0.2733
F	I	0.06	7.9898	-59.0868	-0.2099	-1.0812
	Br		8.0848	-58.3691	-0.1149	-0.3635
	Cl		8.1531	-58.2873	-0.0466	-0.2817
OCF ₃	I	0.35	8.0140	-60.1671	-0.1857	-2.1615
	Br		8.0989	-58.7858	-0.1008	-0.7802
	Cl		8.1600	-58.4837	-0.0397	-0.4781
CF ₃	I	0.54	8.0115	-59.3602	-0.1882	-1.3546
	Br		8.0977	-58.4912	-0.1020	-0.4856
	Cl		8.1504	-58.2889	-0.0493	-0.2833
(C ₆ F ₅)	I	-	7.7945	-67.0214	-0.4052	-9.0158
	Br		7.9728	-59.1034	-0.2269	-1.0978
	Cl		8.0309	-58.2871	-0.1688	-0.2815

Table S2: Chemical shifts of the nitrogen and 2H in pyridine when mixed with various halogen bond donors following 5:1 stoichiometric experiments in acetonitrile-d₃. In addition, the Hammett para-constant σ_{para} and the changes in chemical shift $\Delta\delta$ between pyridine mixed with halogen bond donors and pyridine mixed with the same equivalents of benzene are displayed. The substituents under "R" and halogens under "X" correspond to the structure displayed in Figure 1.

R	X	σ	$^1\text{H}\delta$ (ppm)	$^{15}\text{N}\delta$ (ppm)	$^1\text{H}\Delta\delta$ (ppm)	$^{15}\text{N}\Delta\delta$ (ppm)
<i>(benzene reference)</i>		-	9.0142	-60.7775	0	0
OCH ₃	I	-0.27	8.4538	-61.1981	0.5605	0.4206
	Br		8.6299	-60.5959	0.3844	-0.1816
	Cl		8.5376	-60.4980	0.4766	-0.2795
CH ₃	I	-0.17	8.4794	-60.6136	0.5348	-0.1639
	Br		8.4786	-60.1747	0.5357	-0.6028
	Cl		8.6642	-60.4805	0.3500	-0.2970
H	I	0	8.5464	-60.4733	0.4679	-0.3042
	Br		8.6284	-60.4523	0.3858	-0.3252
	Cl		8.5665	-60.5222	0.4477	-0.2553
F	I	0.06	8.5251	-60.9733	0.4892	0.1959
	Br		8.6160	-60.7238	0.3982	-0.0537
	Cl		8.6798	-60.6732	0.3344	-0.1043
OCF ₃	I	0.35	8.6628	-61.3577	0.3514	0.5803
	Br		8.5883	-60.5150	0.4260	-0.2625
	Cl		8.6433	-60.3475	0.3710	-0.4300
CF ₃	I	0.54	8.4714	-61.3331	0.5428	0.5557
	Br		8.5530	-61.2109	0.4612	0.4334
	Cl		8.6428	-60.4614	0.3714	-0.3161
(C ₆ F ₅)	I	-	8.3456	-66.5245	0.6686	5.7471
	Br		8.4560	-61.5097	0.5583	0.7322
	Cl		8.5090	-60.8302	0.5053	0.0527

Table S3: Chemical shifts of the nitrogen and 2H in pyridine when mixed with various halogen bond donors following 5:1 stoichiometric experiments in DMSO-d₃. In addition, the Hammett para-constant σ_{para} and the changes in chemical shift $\Delta\delta$ between pyridine mixed with halogen bond donors and pyridine mixed with the same equivalents of benzene are displayed. The substituents under "R" and halogens under "X" correspond to the structure displayed in Figure 1.

R	X	σ	$^1\text{H}\delta$ (ppm)	$^{15}\text{N}\delta$ (ppm)	$^1\text{H}\Delta\delta$ (ppm)	$^{15}\text{N}\Delta\delta$ (ppm)
<i>(benzene reference)</i>		-	8.7467	-60.6011	0	0
OCH ₃	I	-0.27	8.2698	-60.9608	-0.4770	-0.3597
	Br		8.3851	-60.7012	-0.3616	-0.1001
	Cl		8.4675	-60.6346	-0.2792	-0.0336
CH ₃	I	-0.17	8.3422	-60.1871	-0.4046	0.4140
	Br		8.4405	-59.9424	-0.3063	0.6587
	Cl		8.5454	-60.1418	-0.2014	0.4593
H	I	0	8.3361	-60.7310	-0.4106	-0.1299
	Br		8.4301	-60.5691	-0.3167	0.0320
	Cl		8.5233	-60.4072	-0.2235	0.1939
F	I	0.06	8.3196	-60.4930	-0.4271	0.1081
	Br		8.4160	-60.6623	-0.3307	-0.0612
	Cl		8.4810	-60.5853	-0.2658	0.0158
OCF ₃	I	0.35	8.3089	-60.2588	-0.4378	0.3423
	Br		8.3960	-60.0950	-0.3508	0.5061
	Cl		8.4615	-58.0799	-0.2853	0.5212
CF ₃	I	0.54	8.2951	-60.6759	-0.4517	-0.0749
	Br		8.3688	-60.2988	-0.3779	0.3023
	Cl		8.4324	-60.0330	-0.3144	0.5681
(C ₆ F ₅)	I	-	8.1254	-63.4702	-0.6213	-2.8692
	Br		8.2554	-60.6733	-0.4914	-0.0723
	Cl		8.3334	-60.3015	-0.4133	0.2996

Table S4: Chemical shifts of the nitrogen and 2H in pyridine when mixed with various iodobenzenes following 3:1 stoichiometric experiments in toluene-d₈. In addition, the Hammett para-constant σ_{para} and the changes in chemical shift $\Delta\delta$ between pyridine mixed with halogen bond donors and pyridine mixed with the same equivalents of benzene are displayed. The substituents under "R" and halogens under "X" correspond to the structure displayed in Figure 1.

R	σ	$^1\text{H}\delta$ (ppm)	$^{15}\text{N}\delta$ (ppm)	$^1\text{H}\Delta\delta$ (ppm)	$^{15}\text{N}\Delta\delta$ (ppm)
(benzene reference)	-	8.1920	-58.1670	0	0.0280
		8.1920	-58.1975	0	-0.0025
		8.1930	-58.2215	0.0010	-0.0265
OCH ₃	-0.27	8.0130	-58.9105	-0.1790	-0.7155
		8.0145	-58.9320	-0.1775	-0.7370
		8.0145	-58.9245	-0.1775	-0.7295
CH ₃	-0.17	8.0345	-58.5035	-0.1575	-0.3085
		8.0350	-58.5780	-0.1570	-0.3830
		8.0335	-58.5600	-0.1585	-0.3650
H	0	8.0325	-58.7680	-0.1595	-0.5730
		8.0315	-58.7795	-0.1605	-0.5845
		8.0335	-58.7845	-0.1585	-0.5895
F	0.06	8.0495	-58.8340	-0.1425	-0.6390
		8.0465	-58.8560	-0.1455	-0.6610
		8.0495	-58.8720	-0.1425	-0.6770
Cl	0.23	8.0155	-58.9225	-0.1765	-0.7035
		8.0145	-58.9905	-0.1775	-0.7715
		8.0080	-58.9855	-0.1840	-0.7665
OCF ₃	0.35	8.0540	-59.0215	-0.1380	-0.8265
		8.0550	-59.0095	-0.1370	-0.8145
		8.0565	-59.0075	-0.1355	-0.8125
CF ₃	0.54	8.0535	-59.0510	-0.1385	-0.8560
		8.0535	-59.0220	-0.1385	-0.8270
		8.0515	-59.0175	-0.1405	-0.8225

R	σ	^1H δ (ppm)	^{15}N δ (ppm)	^1H $\Delta\delta$ (ppm)	^{15}N $\Delta\delta$ (ppm)
OCH ₂ CH ₃	-0.24	8.0295	-58.9870	-0.1625	-0.7680
		8.0250	-58.9905	-0.1670	-0.7715
		8.0270	-59.0170	-0.1650	-0.7980

Table S5: Chemical shifts of the nitrogen and 2H in pyridine when mixed with various para-substituted phenols following 3:1 stoichiometric experiments in toluene-*d*₈. In addition, the Hammett para-constant σ_{para} and the changes in chemical shift $\Delta\delta$ between pyridine mixed with halogen bond donors and pyridine mixed with the same equivalents of benzene are displayed. The substituents under "R" and halogens under "X" correspond to the structure displayed in Figure 1.

R	σ	$^1\text{H}\delta$ (ppm)	$^{15}\text{N}\delta$ (ppm)	$^1\text{H}\Delta\delta$ (ppm)	$^{15}\text{N}\Delta\delta$ (ppm)
(benzene reference)	-	8.1920	-58.1670	0	0.0280
		8.1920	-58.1975	0	-0.0025
		8.1930	-58.2215	0.0010	-0.0265
OCH ₃	-0.27	7.8045	-82.4545	-0.3875	-24.2595
		7.8090	-82.4065	-0.3830	-24.2115
		7.8045	-82.5245	-0.3875	-24.3295
CH ₃	-0.17	7.7790	-83.2410	-0.4130	-25.0460
		7.7840	-82.8600	-0.4080	-24.6650
		7.7810	-83.0940	-0.4110	-24.8990
H	0	7.7140	-84.2600	-0.4780	-26.0650
		7.7170	-84.2125	-0.4750	-26.0175
		7.7150	-84.2760	-0.4770	-26.0810
F	0.06	7.7260	-85.5895	-0.4660	-27.3945
		7.7240	-85.6455	-0.4680	-27.4505
		7.7260	-85.5625	-0.4660	-27.3675
Cl	0.23	7.6485	-86.6055	-0.5435	-28.4105
		7.6510	-86.4915	-0.5410	-28.2965
		7.6515	-86.4055	-0.5405	-28.2105
OCF ₃	0.35	7.6945	-86.8615	-0.4975	-28.6665
		7.6945	-86.9165	-0.4975	-28.7215
		7.6990	-86.7195	-0.4930	-28.5245

R	σ	$^1\text{H}\delta$ (ppm)	$^{15}\text{N}\delta$ (ppm)	$^1\text{H}\Delta\delta$ (ppm)	$^{15}\text{N}\Delta\delta$ (ppm)
CF ₃	0.54	7.7060	-88.5670	-0.4860	-30.3720
		7.7160	-87.5290	-0.4760	-29.3340
		7.7090	-88.2060	-0.4830	-30.0110
OCH ₂ CH ₃	-0.24	7.8175	-82.1410	-0.3745	-23.9460
		7.8140	-82.2675	-0.3780	-24.0725
		7.8150	-82.0470	-0.3770	-23.8520
(C ₆ F ₅)	-	7.7140	-124.790	-0.4780	-66.5950

2. Computed electronic energies, Gibbs free energies, BSSE, NAP, chemical shifts and complexations shifts

Table S6: Calculated absolute electronic energies E_e , Gibbs free energies $G(298)$, and basis set superposition errors BSSE for the compounds investigated in this study. The substituents under "R" and halogens under "X" correspond to the structure displayed in Figure 1. All values are given in Hartree units.

		E_e	$G(298)$			
Pyridine		-248.27556669	-248.214115			
		Halobenzene		Halobenzene-pyridine complex		
X	R	E_e	$G(298)$	E_e	$G(298)$	BSSE
I	H	-7150.87892577	-7150.820668	-7399.15920688	-7399.026576	0.000679
	OCH ₃	-7265.40209411	-7265.315926	-7513.68201593	-7513.520539	0.000662
	OCH ₂ CH ₃	-7304.71769091	-7304.605868	-7552.99755110	-7552.810152	0.000660
	CH ₃	-7190.19304383	-7190.109950	-7438.47308046	-7438.315201	0.000655
	CH(CH ₃) ₂	-7268.81371240	-7268.678721	-7517.09363449	-7516.884324	0.000576
	F	-7250.12236332	-7250.073702	-7498.40323600	-7498.279570	0.000706
	Cl	-7610.44454720	-7610.398106	-7858.72558324	-7858.604047	0.000706
	OCF ₃	-7563.16097268	-7563.102881	-7811.44200808	-7811.307596	0.000672
	COOCH ₃	-7378.76322051	-7378.669884	-7627.04416581	-7626.875390	0.000681
	COCH ₃	-7303.52753299	-7303.438180	-7551.80861151	-7551.644139	0.000670
	CF ₃	-7487.94206436	-7487.884464	-7736.22341853	-7736.090498	0.000705
	CN	-7243.12232060	-7243.068202	-7491.40422390	-7491.274551	0.000698
	NO ₂	-7355.39158801	-7355.335520	-7603.67386110	-7603.542469	0.000716
Br	H	-2805.60842779	-2805.548950	-3053.88644195	-3053.751489	0.000458
	OCH ₃	-2920.13113367	-2920.043692	-3168.40894100	-3168.248143	0.000454
	OCH ₂ CH ₃	-2959.44673102	-2959.333564	-3207.72448027	-3207.537645	0.000443
	CH ₃	-2844.92239987	-2844.839932	-3093.20023672	-3093.041703	0.000419
	CH(CH ₃) ₂	-2923.54302209	-2923.406888	-3171.82081334	-3171.607753	0.000411

	F	-2904.85164392	-2904.801751	-3153.13011375	-3153.007568	0.000499
	Cl	-3265.17389558	-3265.126233	-3513.45247580	-3513.331944	0.000477
	OCF ₃	-3217.89105812	-3217.831740	-3466.16987472	-3466.036571	0.000454
	COOCH ₃	-3033.49273543	-3033.398101	-3281.77130994	-3281.603225	0.000439
	COCH ₃	-2958.25706885	-2958.166790	-3206.53574851	-3206.371637	0.000432
	CF ₃	-3142.67154520	-3142.612745	-3390.95045364	-3390.817558	0.000447
	CN	-2897.85188795	-2897.796529	-3146.13114342	-3146.002321	0.000456
	NO ₂	-3010.12114760	-3010.063804	-3258.40070619	-3258.269789	0.000486
Cl	H	-691.80292038	-691.741711	-940.07960339	-939.944792	0.000643
	OCH ₃	-806.32538231	-806.236138	-1054.60194562	-1054.440848	0.001060
	OCH ₂ CH ₃	-845.64099789	-845.525847	-1093.91750095	-1093.730744	0.000650
	CH ₃	-731.11682767	-731.032574	-979.39336250	-979.236786	0.000618
	CH(CH ₃) ₂	-809.73744491	-809.599625	-1058.01400536	-1057.805301	0.000644
	F	-791.04604457	-790.994402	-1039.32304625	-1039.197216	0.000690
	Cl	-1151.36836738	-1151.318266	-1399.64544579	-1399.522156	0.000674
	OCF ₃	-1104.08557295	-1104.024480	-1352.36289519	-1352.225634	0.000736
	COOCH ₃	-919.68734883	-919.591177	-1167.96448198	-1167.797118	0.000661
	COCH ₃	-844.45170183	-844.359858	-1092.72889164	-1092.565212	0.000661
	CF ₃	-1028.86611429	-1028.805566	-1277.14351629	-1277.011746	0.000702
	CN	-784.04652810	-783.989401	-1032.32414890	-1032.192173	0.000699
	NO ₂	-896.31581271	-896.256697	-1144.59363475	-1144.459770	0.000703

Table S7: The calculated change of the natural atomic population (NAP) of the pyridine N atom, $\Delta n(\text{N})$, the π part $\Delta n_{\pi}(\text{N})$ of $\Delta n(\text{N})$, and the changes of the NAP at the halogen atoms $\Delta n(\text{X})$ ($\text{X} = \text{I}, \text{Br}, \text{Cl}$) in the molecules investigated in this study. $\Delta n(\text{N})$ and $\Delta n_{\pi}(\text{N})$ are given relative to free pyridine, $\Delta n(\text{X})$ relative to the corresponding compound with $\text{R} = \text{H}$. For the $\Delta n(\text{X})$, the notations C and H refer to the halobenzene-pyridine complex and the free halobenzene, respectively. All values scaled by a factor of 100. The substituents under "R" and halogens under "X" correspond to the structure displayed in Figure 1.

R	X = I				X = Br				X = Cl			
	$100 \cdot \Delta n(\text{N})$	$100 \cdot \Delta n_{\pi}(\text{N})$	$100 \cdot \Delta n(\text{I})$		$100 \cdot \Delta n(\text{Br})$	$100 \cdot \Delta n_{\pi}(\text{Br})$	$100 \cdot \Delta n(\text{Cl})$		$100 \cdot \Delta n(\text{N})$	$100 \cdot \Delta n_{\pi}(\text{N})$	$100 \cdot \Delta n(\text{Cl})$	
			C	H			C	H			C	H
H	0.80	2.03	0.00	0.00	0.11	0.76	0.00	0.00	-0.31	0.04	0.00	0.00
OCH ₃	0.71	1.90	-0.13	0.59	0.02	0.68	0.65	0.62	-0.38	-0.02	0.67	0.62
OCH ₂ CH ₃	0.69	1.87	-0.16	0.69	0.01	0.67	0.77	0.75	-0.39	-0.03	0.77	0.73
CH ₃	0.72	1.91	-0.12	0.48	0.04	0.69	0.50	0.44	-0.37	-0.01	0.47	0.42
CH(CH ₃) ₂	0.71	1.90	-0.13	0.49	0.03	0.68	0.49	0.50	-0.37	-0.01	0.46	0.43
F	0.97	2.28	0.25	-0.95	0.26	0.92	-0.80	-0.86	-0.20	0.13	-0.71	-0.75
Cl	1.03	2.35	0.32	-1.47	0.32	0.98	-1.32	-1.31	-0.15	0.18	-1.23	-1.22
OCF ₃	1.05	2.40	0.37	-1.53	0.41	1.09	-1.83	-1.41	-0.07	0.26	-1.68	-1.14
COOCH ₃	1.05	2.36	0.33	-2.19	0.35	1.00	-2.01	-1.94	-0.12	0.21	-1.92	-1.75
COCH ₃	1.09	2.41	0.38	-2.40	0.39	1.04	-2.22	-2.17	-0.09	0.23	-2.11	-1.91
CF ₃	1.17	2.55	0.52	-2.62	0.47	1.15	-2.45	-2.47	-0.02	0.29	-2.34	-2.18
CN	1.31	2.75	0.72	-3.65	0.60	1.28	-3.47	-3.39	0.09	0.39	-3.20	-3.00
NO ₂	1.43	2.91	0.88	-4.71	0.71	1.40	-4.36	-4.29	0.18	0.47	-4.04	-3.83

Table S8: The calculated chemical shielding σ and the change in chemical shift upon complexation $\Delta\delta$ at the N atom and the *ortho* H atom in the pyridine moiety for the halobenzene-pyridine complexes investigated in this study. $\Delta\delta$ given relative to the chemical shift in free pyridine. For free pyridine, the calculated shielding values are $\sigma(^{15}\text{N}) = -99.6517$ and $\sigma(^1\text{H}) = 22.5326$. All values are given in ppm. The substituents under "R" and halogens under "X" correspond to the structure displayed in Figure 1.

R	X = I				X = Br				X = Cl			
	$\sigma(^{15}\text{N})$	$\sigma(^1\text{H})$	$\Delta\delta(^{15}\text{N})$	$\Delta\delta(^1\text{H})$	$\sigma(^{15}\text{N})$	$\sigma(^1\text{H})$	$\Delta\delta(^{15}\text{N})$	$\Delta\delta(^1\text{H})$	$\sigma(^{15}\text{N})$	$\sigma(^1\text{H})$	$\Delta\delta(^{15}\text{N})$	$\Delta\delta(^1\text{H})$
H	-95.7897	22.5871	-3.8620	-0.0545	-99.5788	22.5018	-0.0729	0.0308	-101.8064	22.4686	2.1547	0.0640
OCH ₃	-96.3123	22.5921	-3.3394	-0.0595	-99.9298	22.5118	0.2781	0.0208	-102.0816	22.4787	2.4299	0.0539
OCH ₂ CH ₃	-96.3388	22.5918	-3.3129	-0.0592	-99.9801	22.5117	0.3284	0.0209	-102.1045	22.4792	2.4528	0.0534
CH ₃	-96.1964	22.5932	-3.4553	-0.0606	-99.8518	22.5126	0.2001	0.0200	-102.012	22.4782	2.3603	0.0544
CH(CH ₃) ₂	-96.5561	22.6039	-3.0956	-0.0713	-99.7967	22.5093	0.1450	0.0233	-101.974	22.4747	2.3223	0.0579
F	-94.8637	22.6071	-4.7880	-0.0745	-98.8562	22.5229	-0.7955	0.0097	-101.3641	22.4866	1.7124	0.0460
Cl	-94.4717	22.6135	-5.1800	-0.0809	-98.5006	22.5309	-1.1511	0.0017	-101.1319	22.493	1.4802	0.0396
OCF ₃	-94.3032	22.6083	-5.3485	-0.0757	-98.2464	22.5291	-1.4053	0.0035	-100.9009	22.4962	1.2492	0.0364
COOCH ₃	-94.3236	22.5988	-5.3281	-0.0662	-98.3306	22.5157	-1.3211	0.0169	-100.9815	22.4772	1.3298	0.0554
COCH ₃	-94.3101	22.5988	-5.3416	-0.0662	-98.1702	22.5165	-1.4815	0.0161	-100.9063	22.4775	1.2546	0.0551
CF ₃	-93.6587	22.6050	-5.9930	-0.0724	-97.7201	22.5222	-1.9316	0.0104	-100.647	22.4819	0.9953	0.0507
CN	-92.8231	22.6192	-6.8286	-0.0866	-96.9877	22.5352	-2.6640	-0.0026	-100.1296	22.4933	0.4779	0.0393
NO ₂	-92.1983	22.6154	-7.4534	-0.0828	-96.6463	22.5298	-3.0054	0.0028	-99.7796	22.4887	0.1279	0.0439

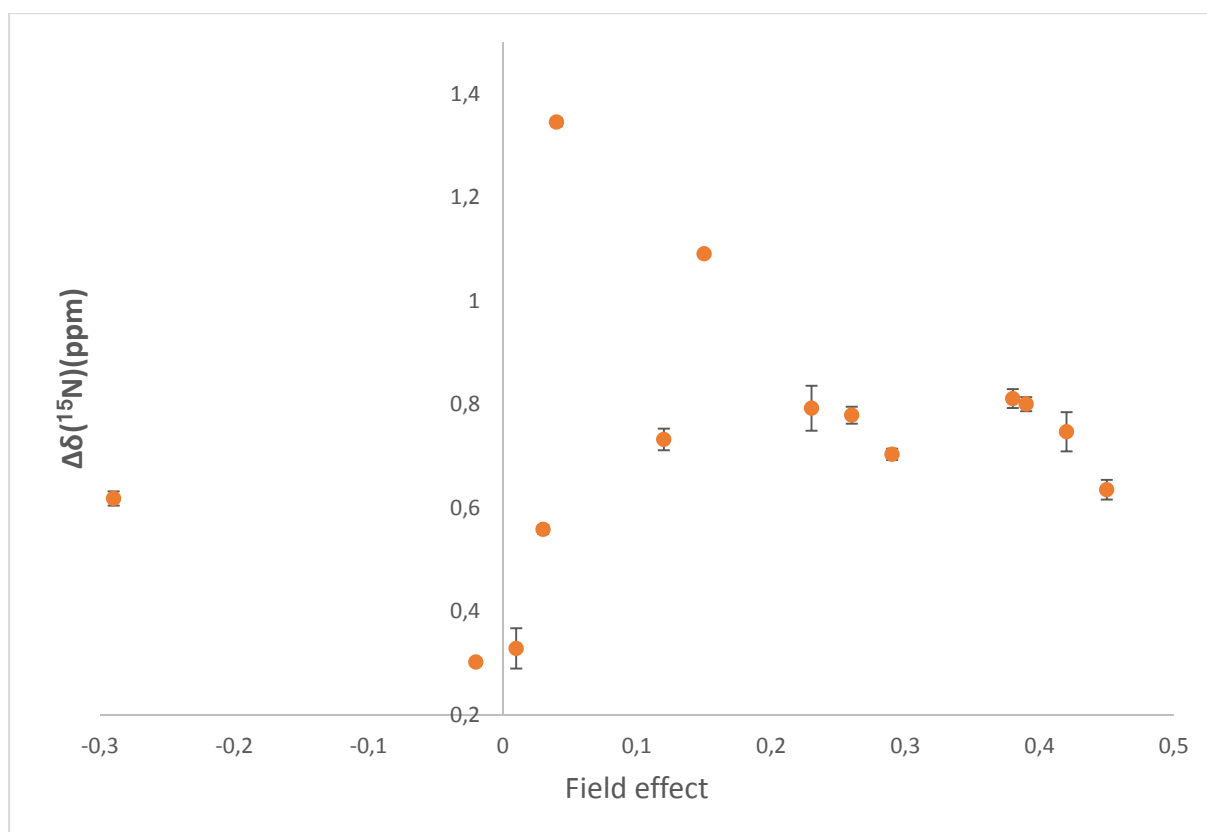


Figure S1. Plot of the change in ^{15}N chemical shift against Swain-Lupton field effect for the various para-substituted iodobenzenes.

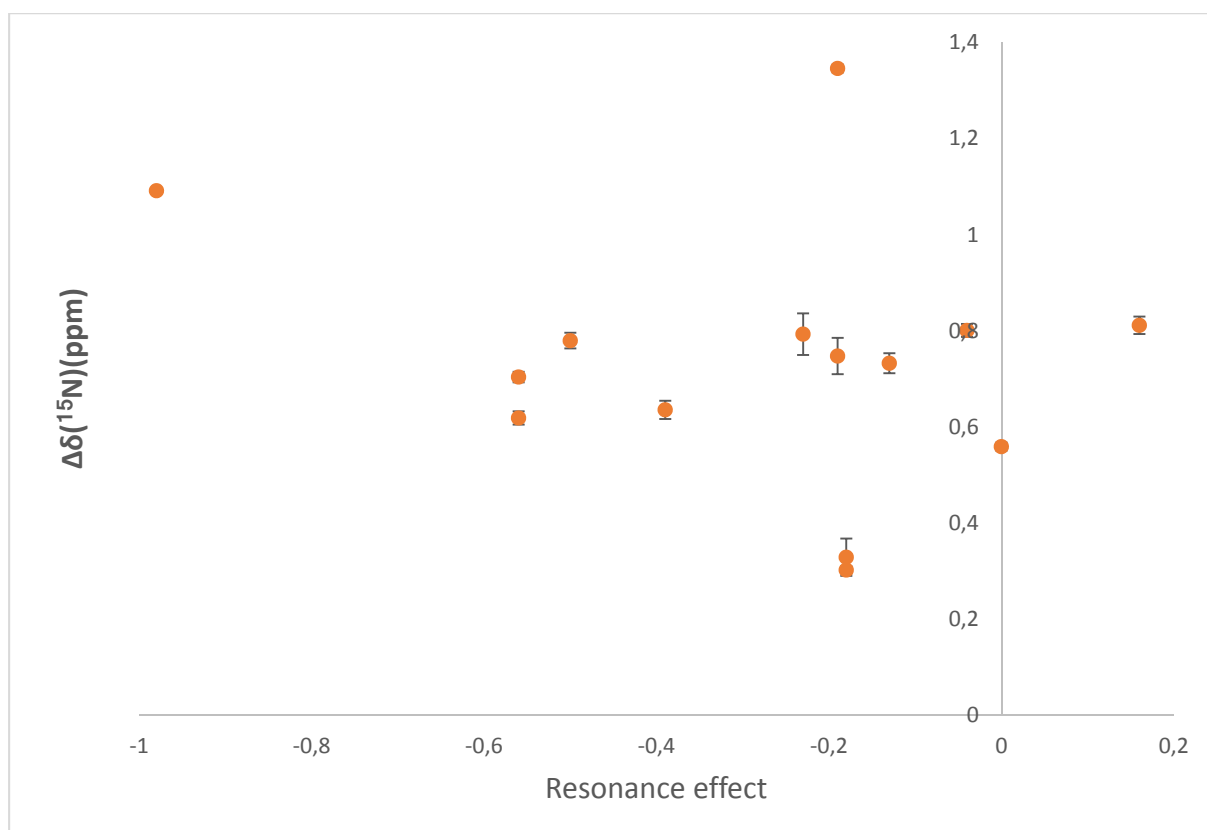


Figure S2. Plot of the change in ^{15}N chemical shift against Swain-Lupton resonance effect for the various para-substituted iodobenzenes.

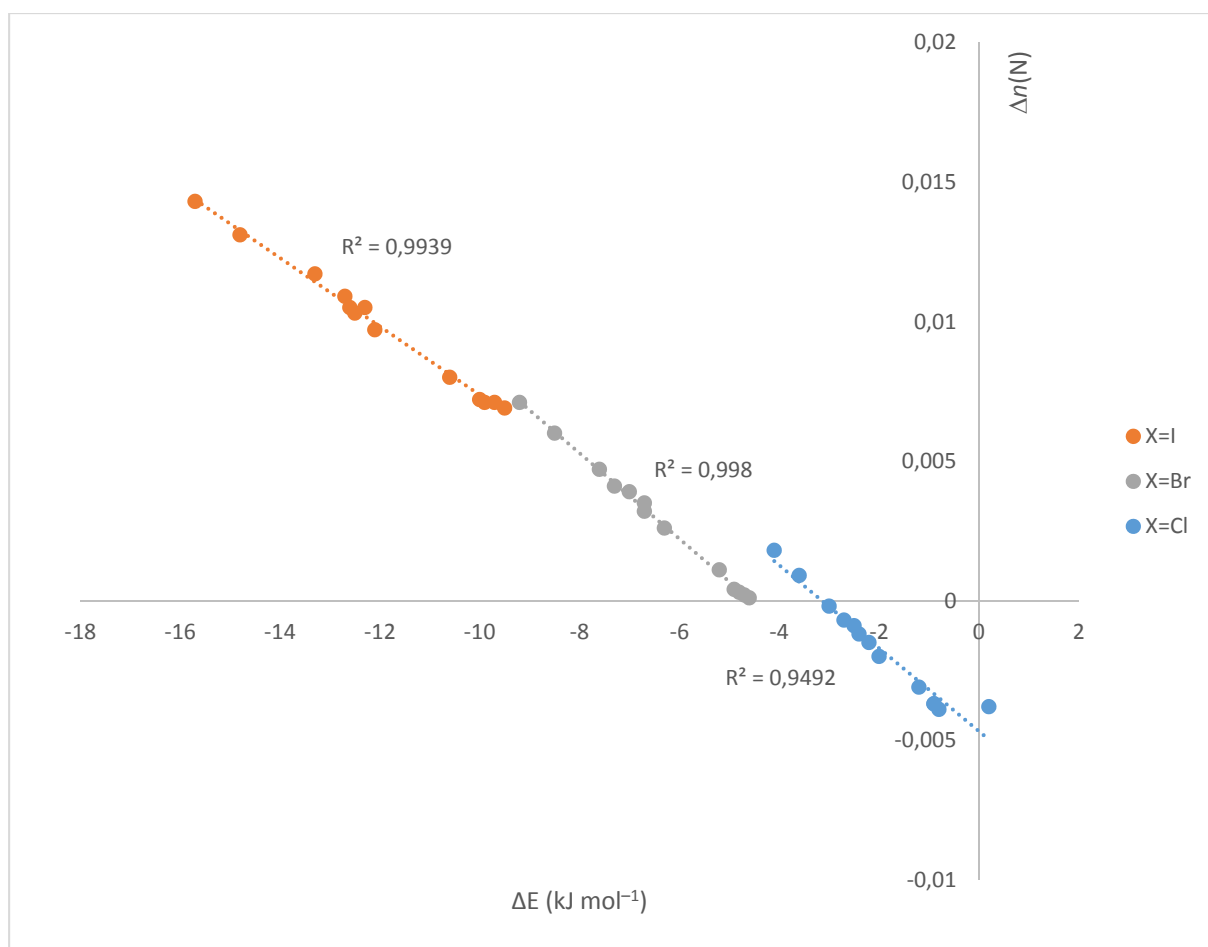


Figure S3. Plot of $\Delta n(N)$ against ΔE for a series of iodo-, bromo- and chlorobenzenes (see table S7), and the R^2 -values for their linear correlation.

3. NMR spectra (pyridine:halobenzene 1:5 mixtures)

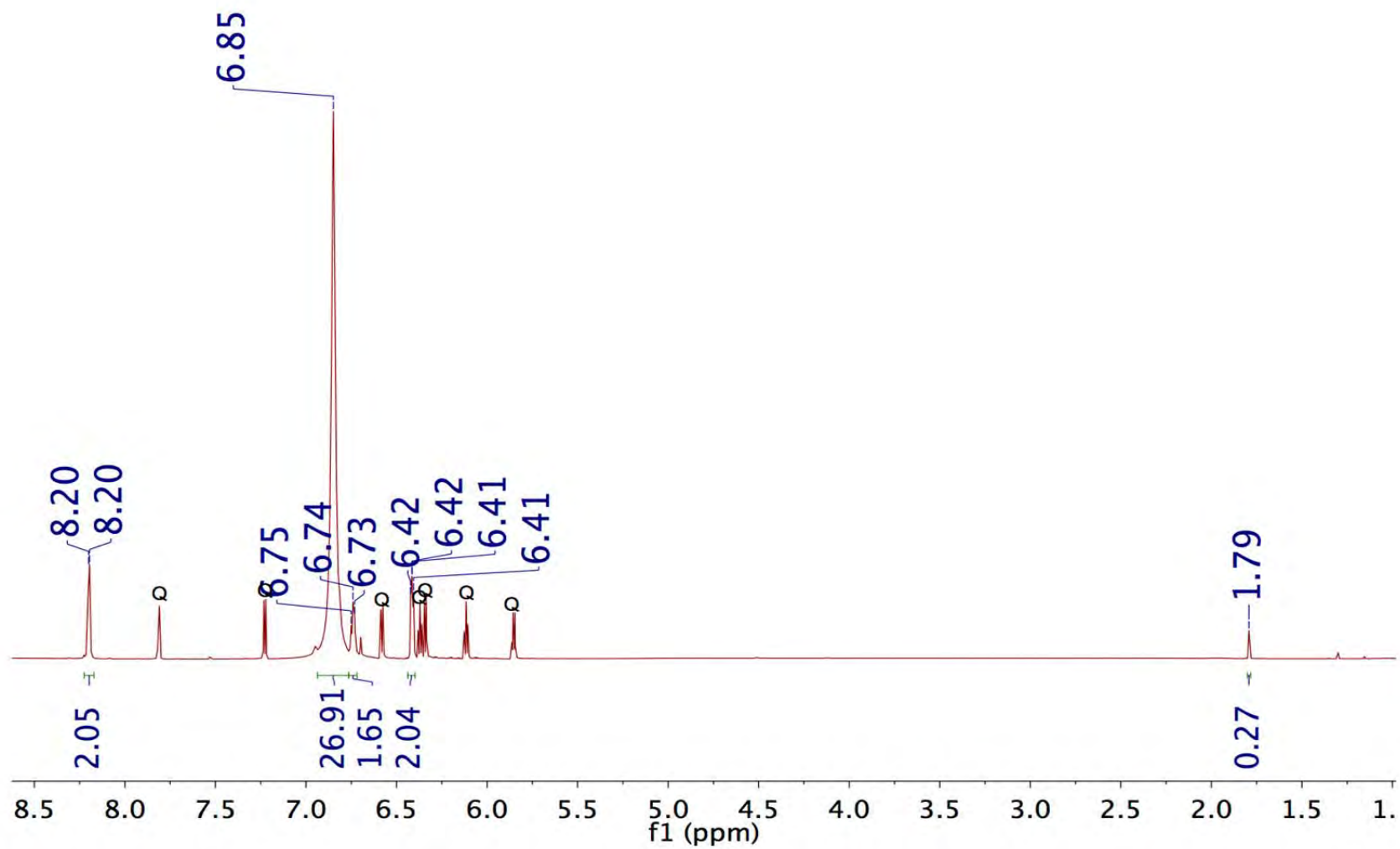


Figure S4: The ^1H NMR-spectrum of pyridine mixed with 5 equivalents of benzene in toluene- d_8 . Quinoline peaks are highlighted with "Q".

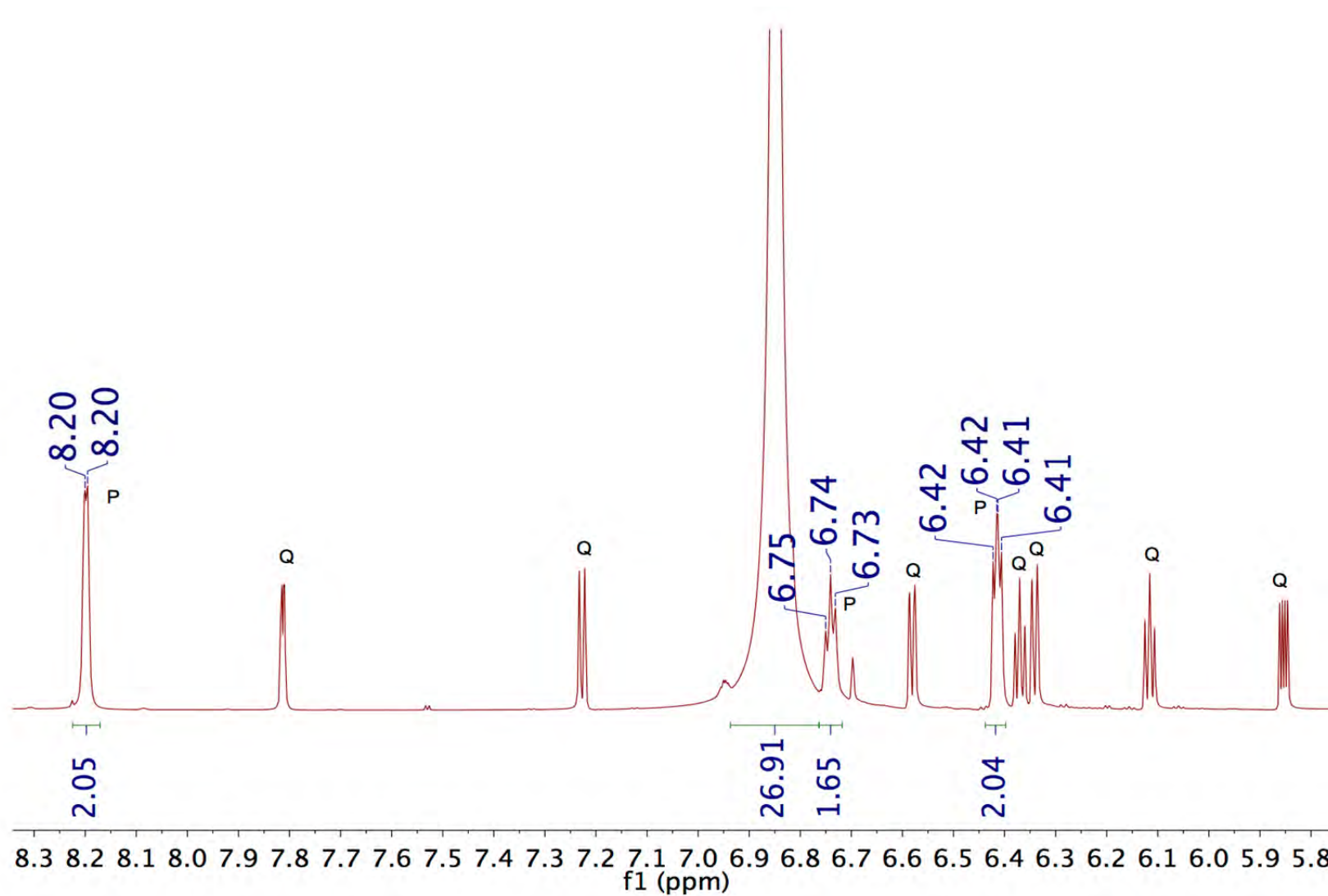


Figure S5: The aromatic region of the ^1H NMR-spectrum of pyridine mixed with 5 equivalents of benzene in toluene- d_8 . Quinoline and pyridine peaks are highlighted with “Q” and “P”, respectively.

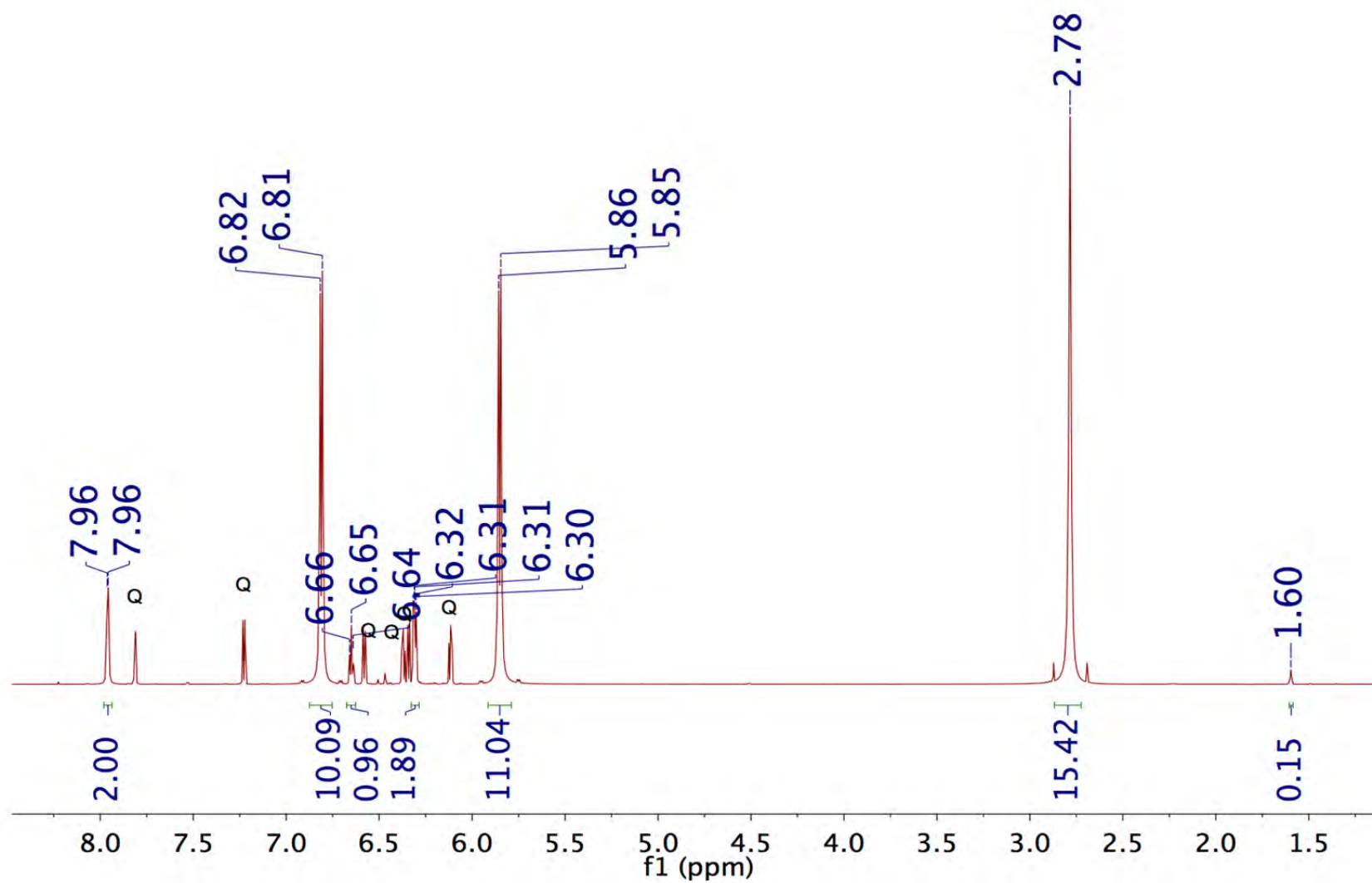


Figure S6: The ¹H NMR-spectrum of pyridine mixed with 5 equivalents of 4-iodoanisole in toluene-*d*₈. Quinoline peaks are highlighted with “Q”.

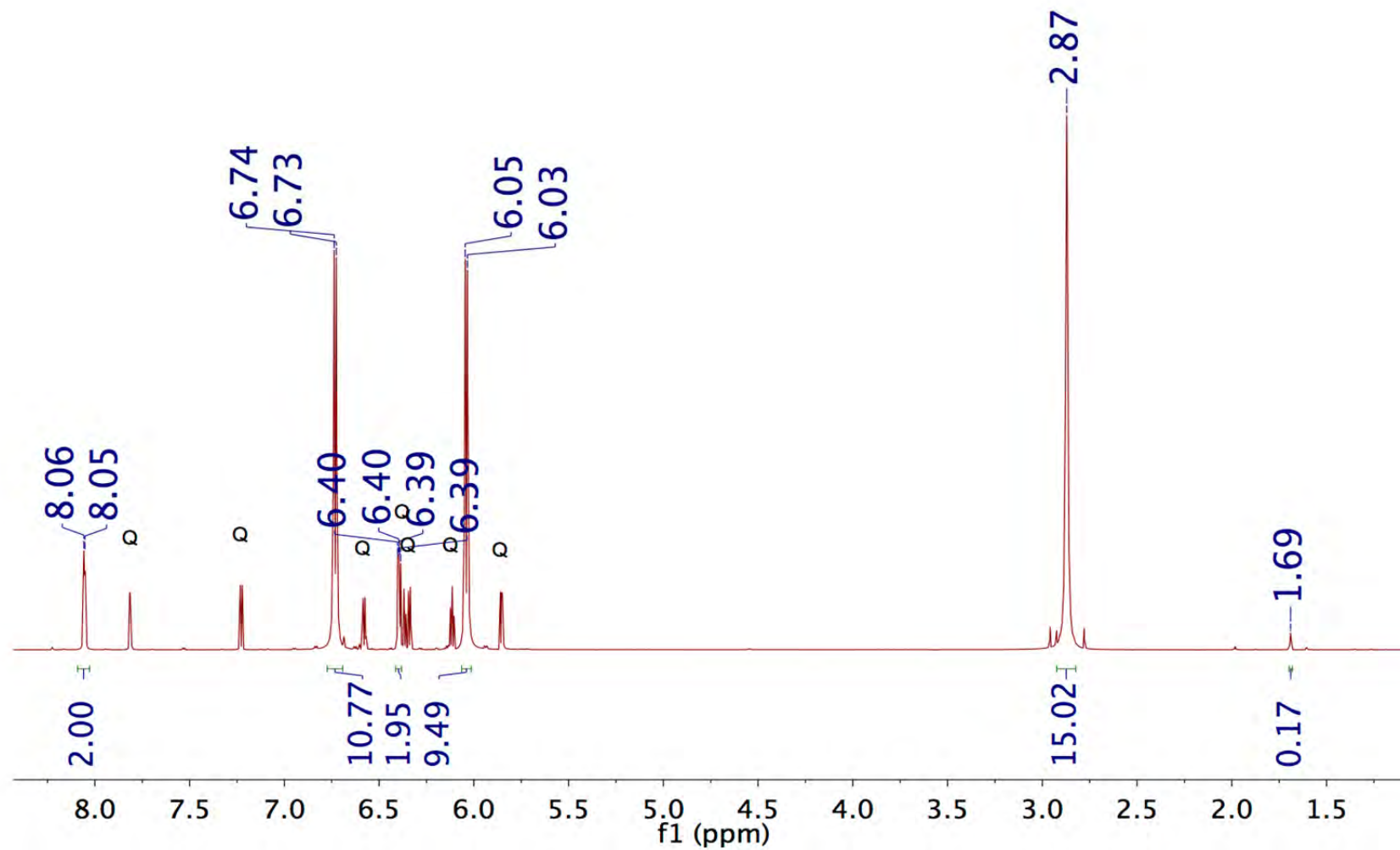


Figure S7: The ¹H NMR-spectrum of pyridine mixed with 5 equivalents of 4-bromoanisole in toluene-*d*₈. Quinoline peaks are highlighted with “Q”.

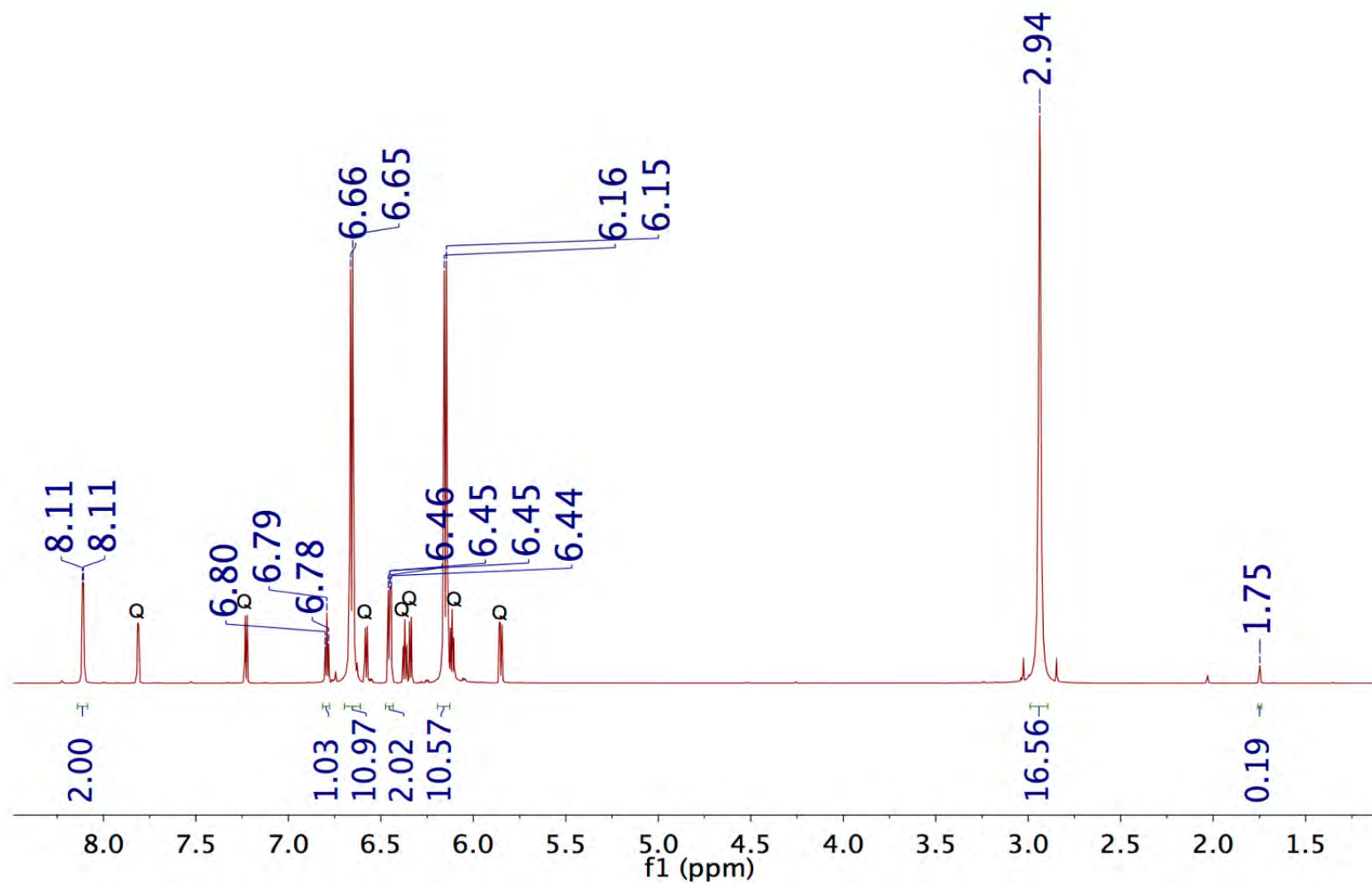


Figure S8: The ¹H NMR-spectrum of pyridine mixed with 5 equivalents of 4-chloroanisole in toluene-*d*₈. Quinoline peaks are highlighted with “Q”.

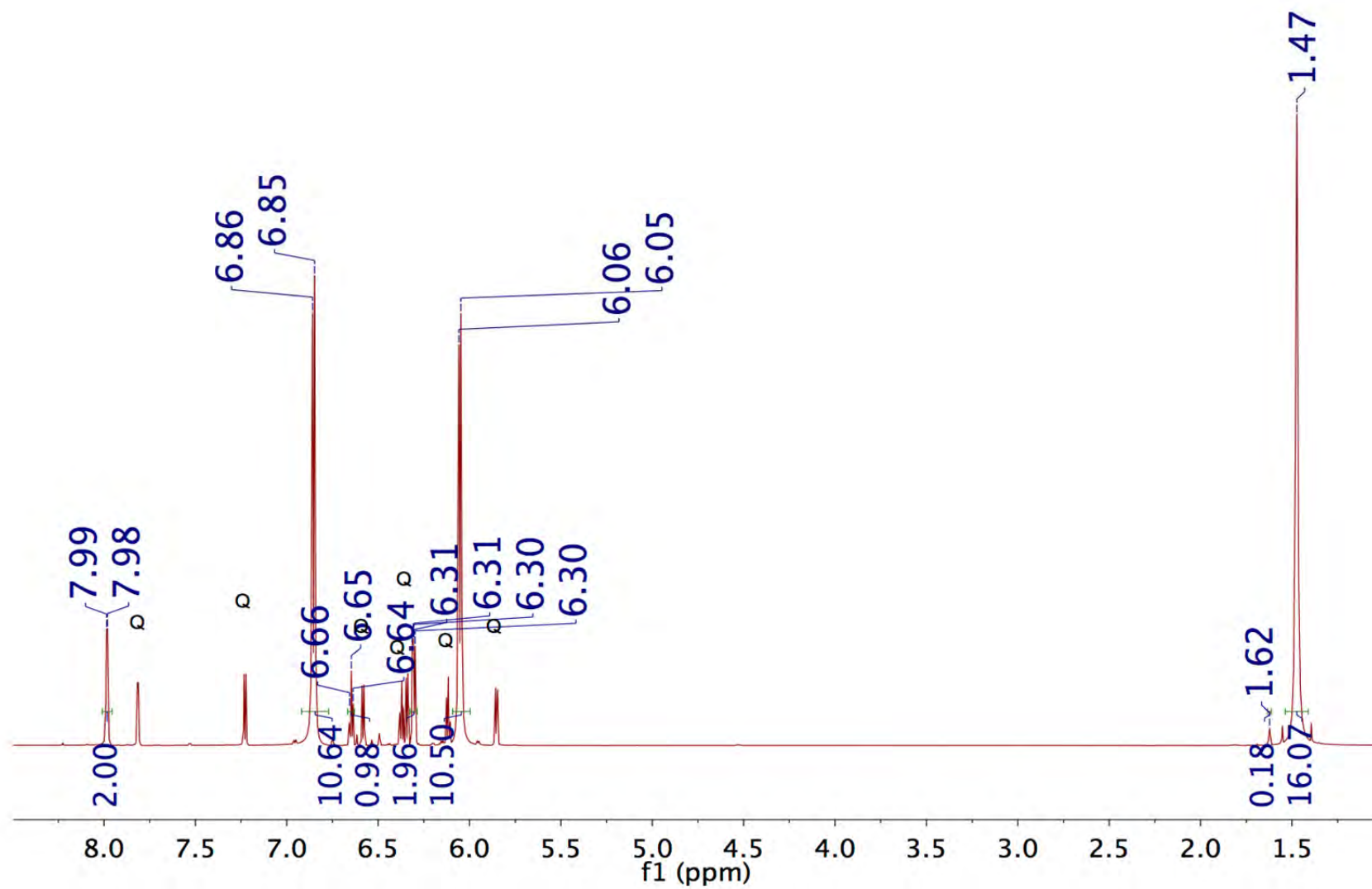


Figure S9: The ¹H NMR-spectrum of pyridine mixed with 5 equivalents of 4-iodotoluene in toluene-*d*₈. Quinoline peaks are highlighted with “Q”.

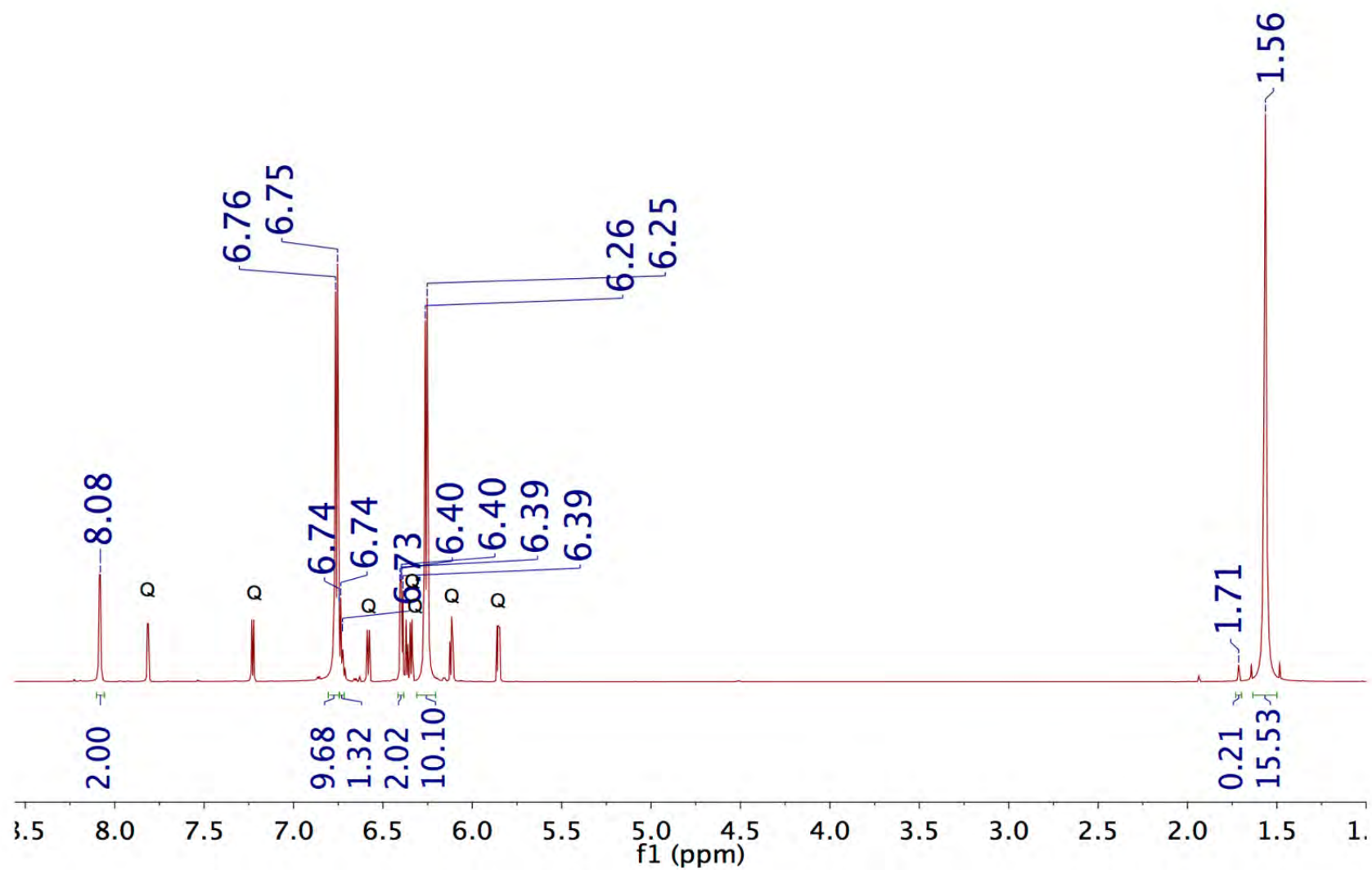


Figure S10:

The ^1H NMR-spectrum of pyridine mixed with 5 equivalents of 4-bromotoluene in toluene- d_8 . Quinoline peaks are highlighted with "Q".

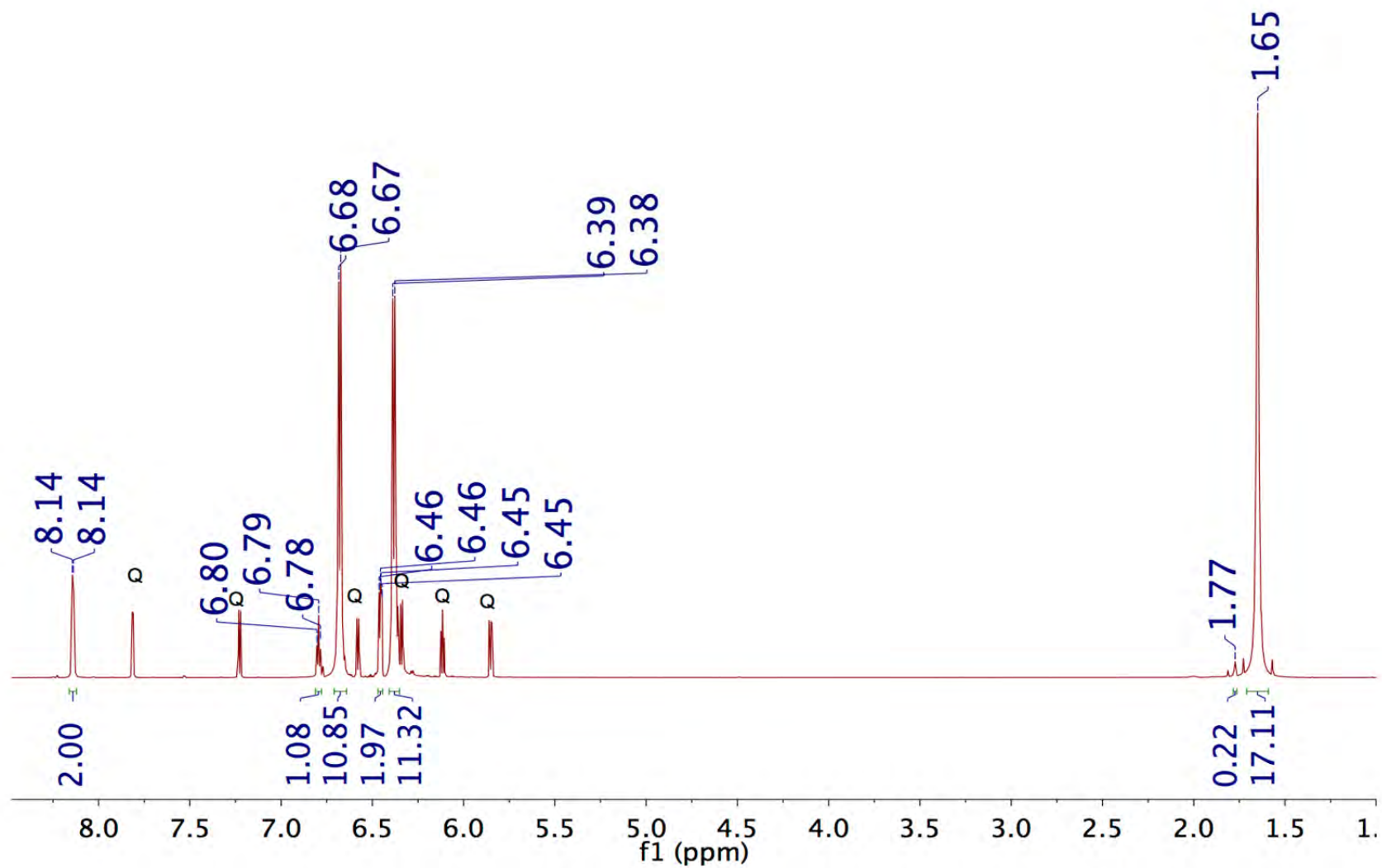


Figure S7: The ^1H NMR-spectrum of pyridine mixed with 5 equivalents of 4-chlorotoluene in toluene- d_8 . Quinoline peaks are highlighted with "Q".

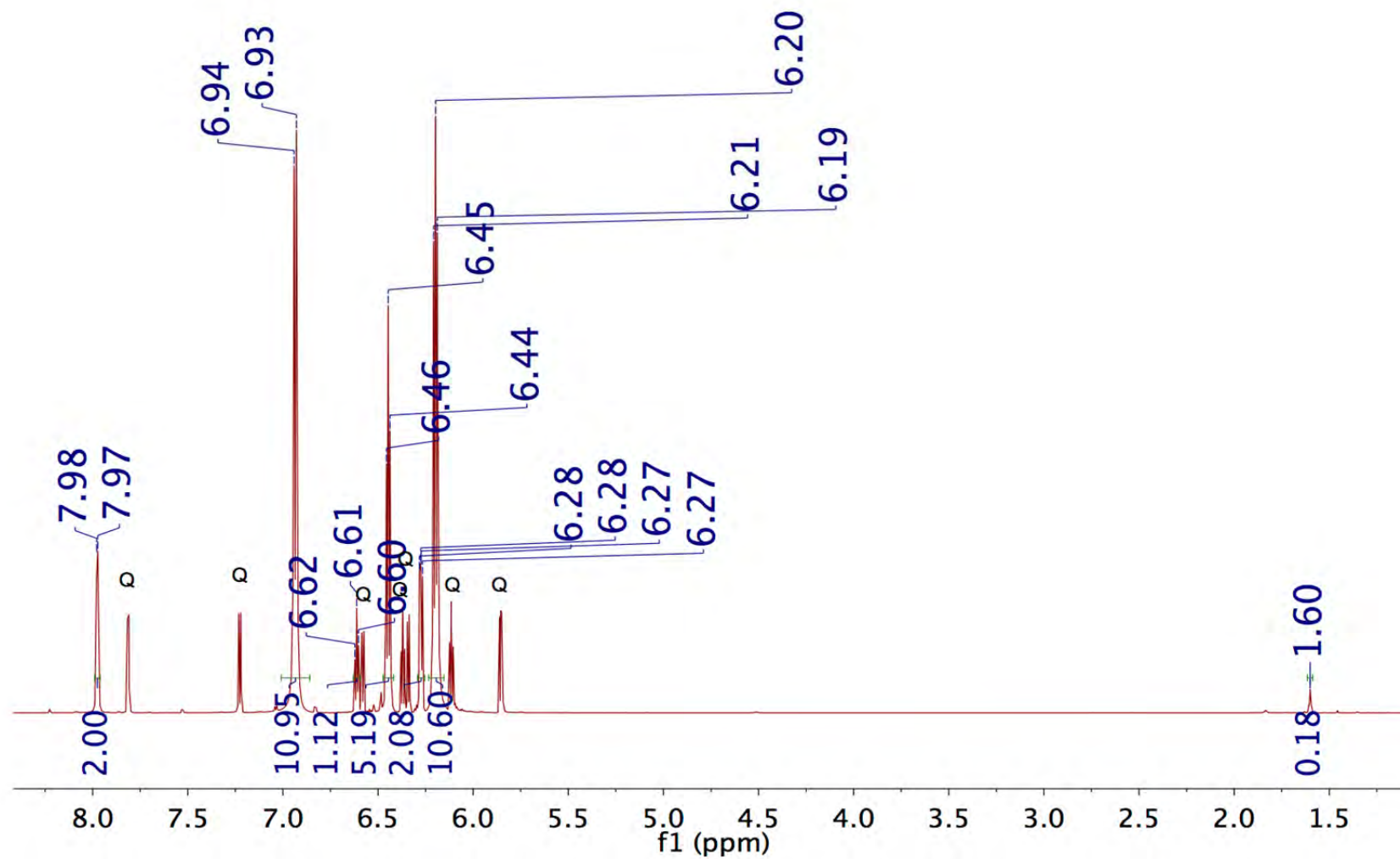


Figure S8: The ^1H NMR-spectrum of pyridine mixed with 5 equivalents of 4-iodobenzene in toluene- d_8 . Quinoline peaks are highlighted with "Q".

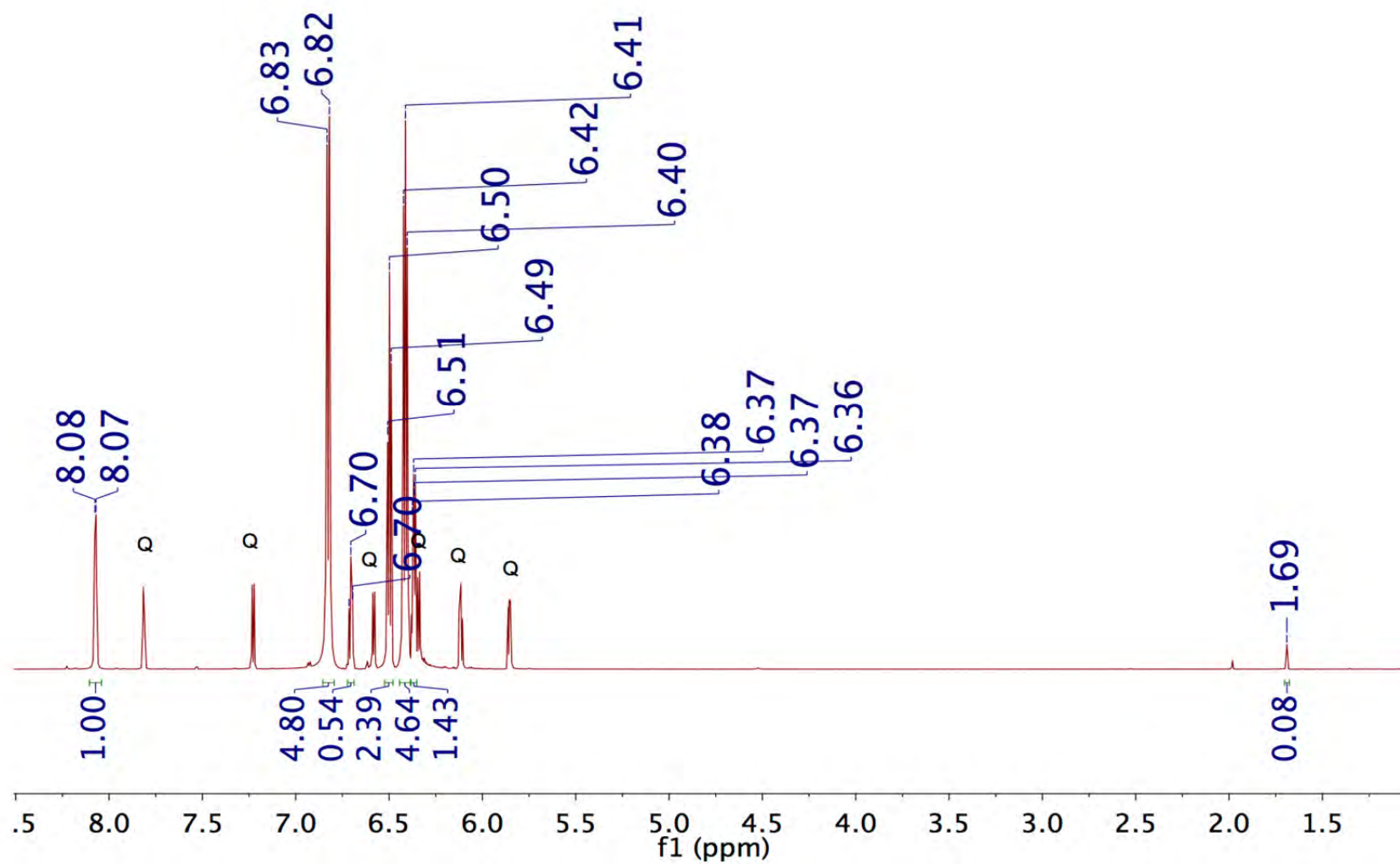


Figure S9: The ¹H NMR-spectrum of pyridine mixed with 5 equivalents of 4-bromobenzene in toluene-*d*₈. Quinoline peaks are highlighted with “Q”.

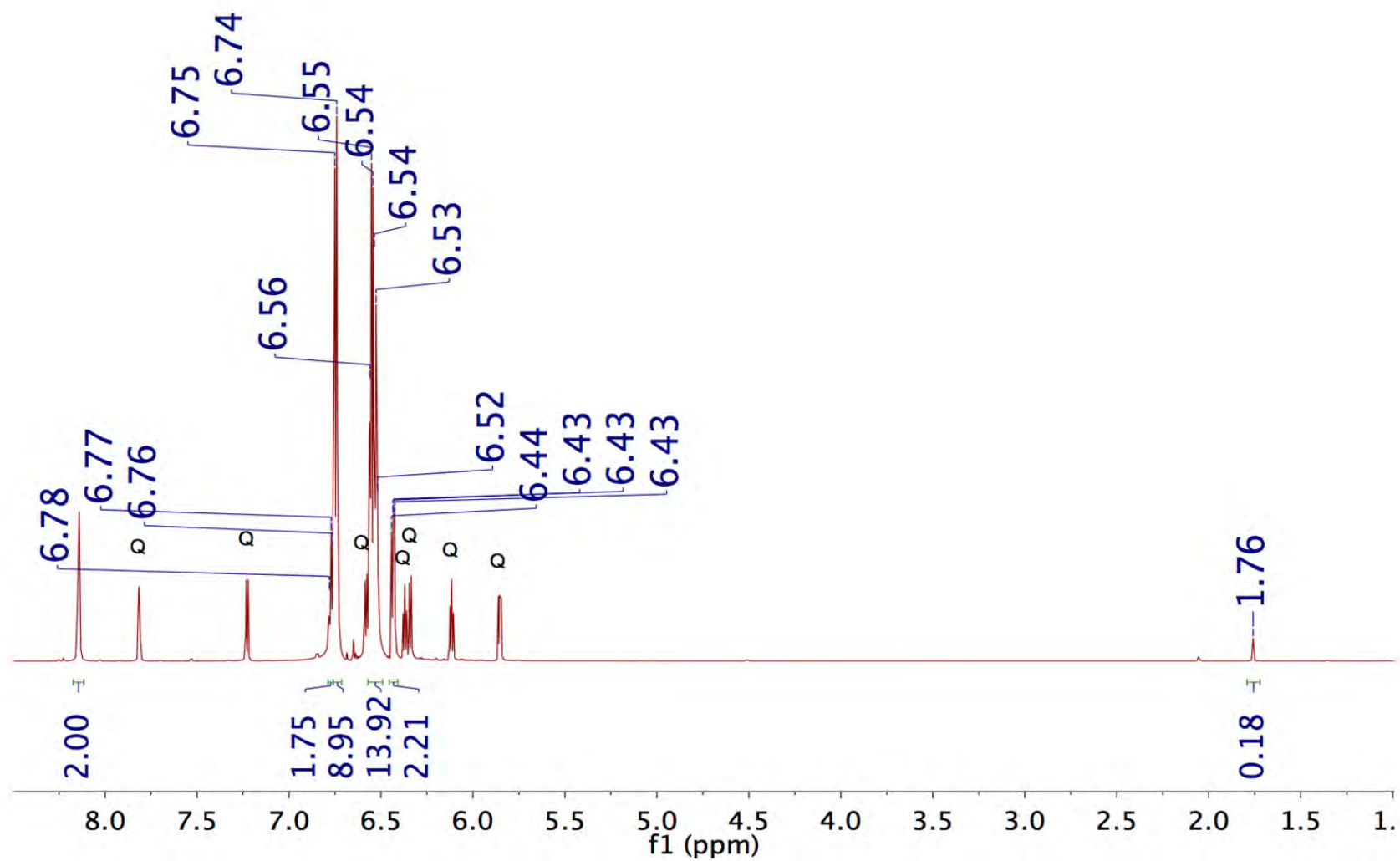


Figure S10: The ^1H NMR-spectrum of pyridine mixed with 5 equivalents of 4-chlorobenzene in toluene- d_8 . Quinoline peaks are highlighted with "Q".

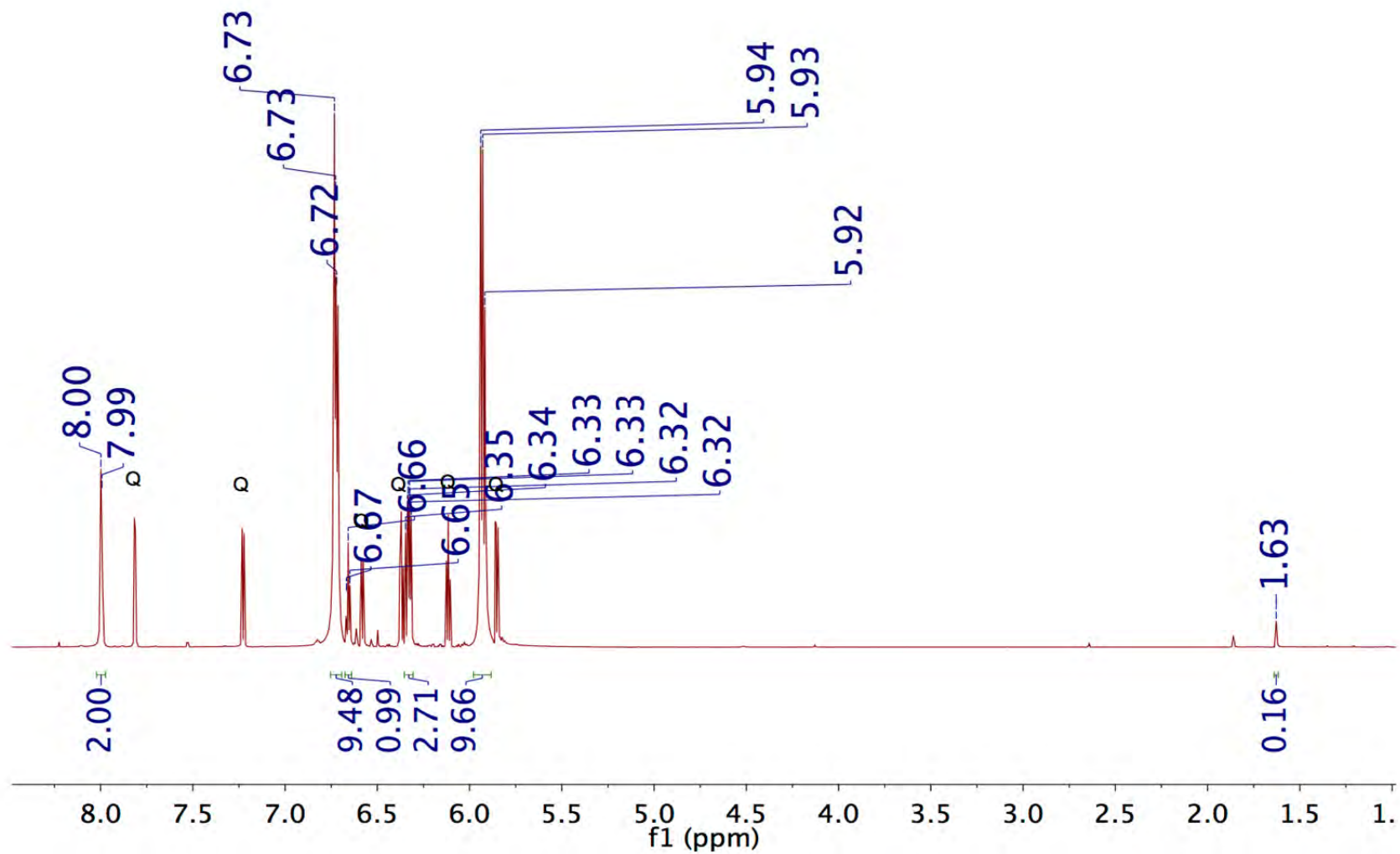


Figure S11: The ^1H NMR-spectrum of pyridine mixed with 5 equivalents of 4-fluoriodobenzene in toluene- d_8 . Quinoline peaks are highlighted with "Q".

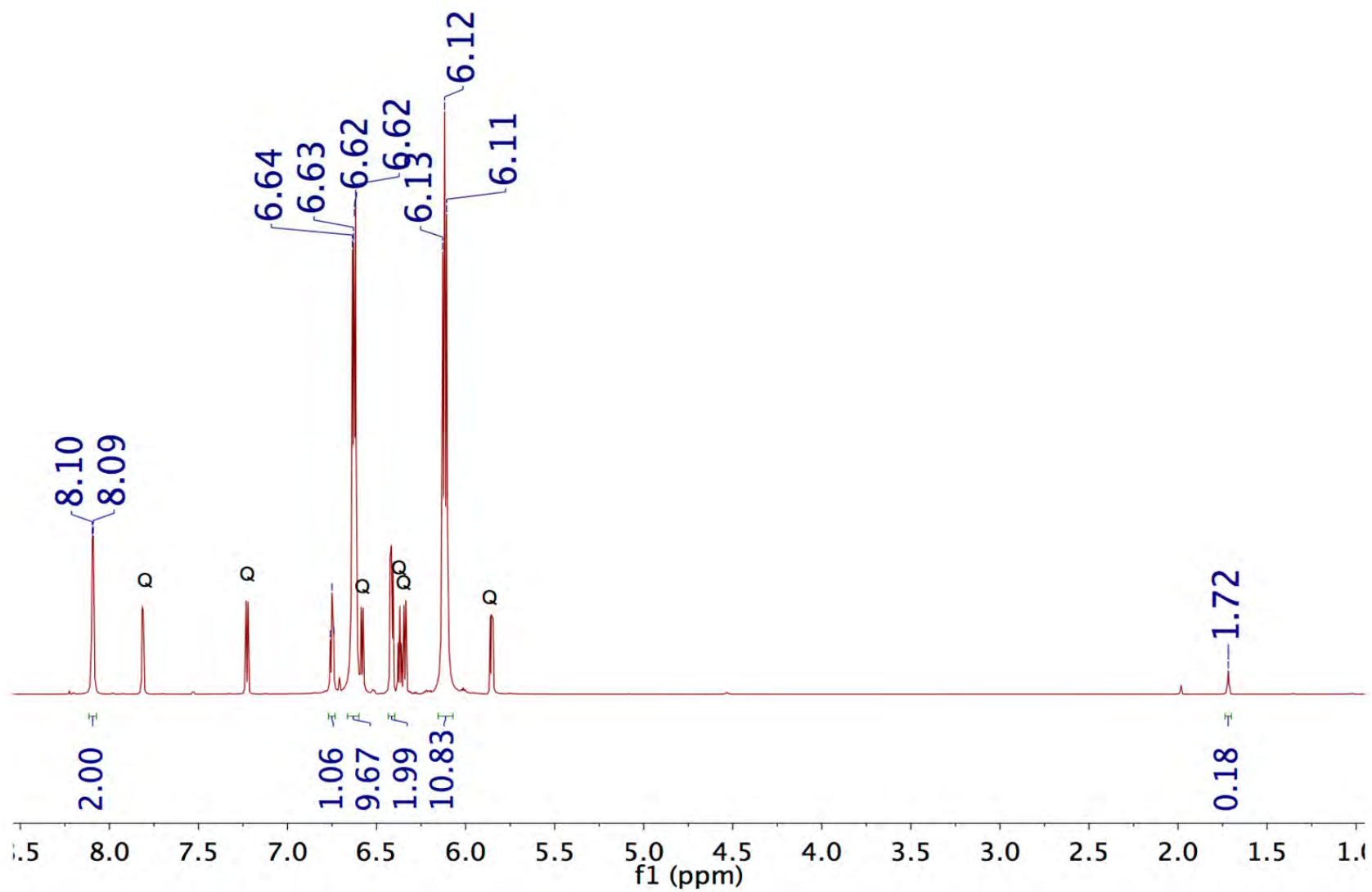


Figure S12: The ^1H NMR-spectrum of pyridine mixed with 5 equivalents of 4-bromofluorobenzene in toluene- d_8 . Quinoline peaks are highlighted with "Q".

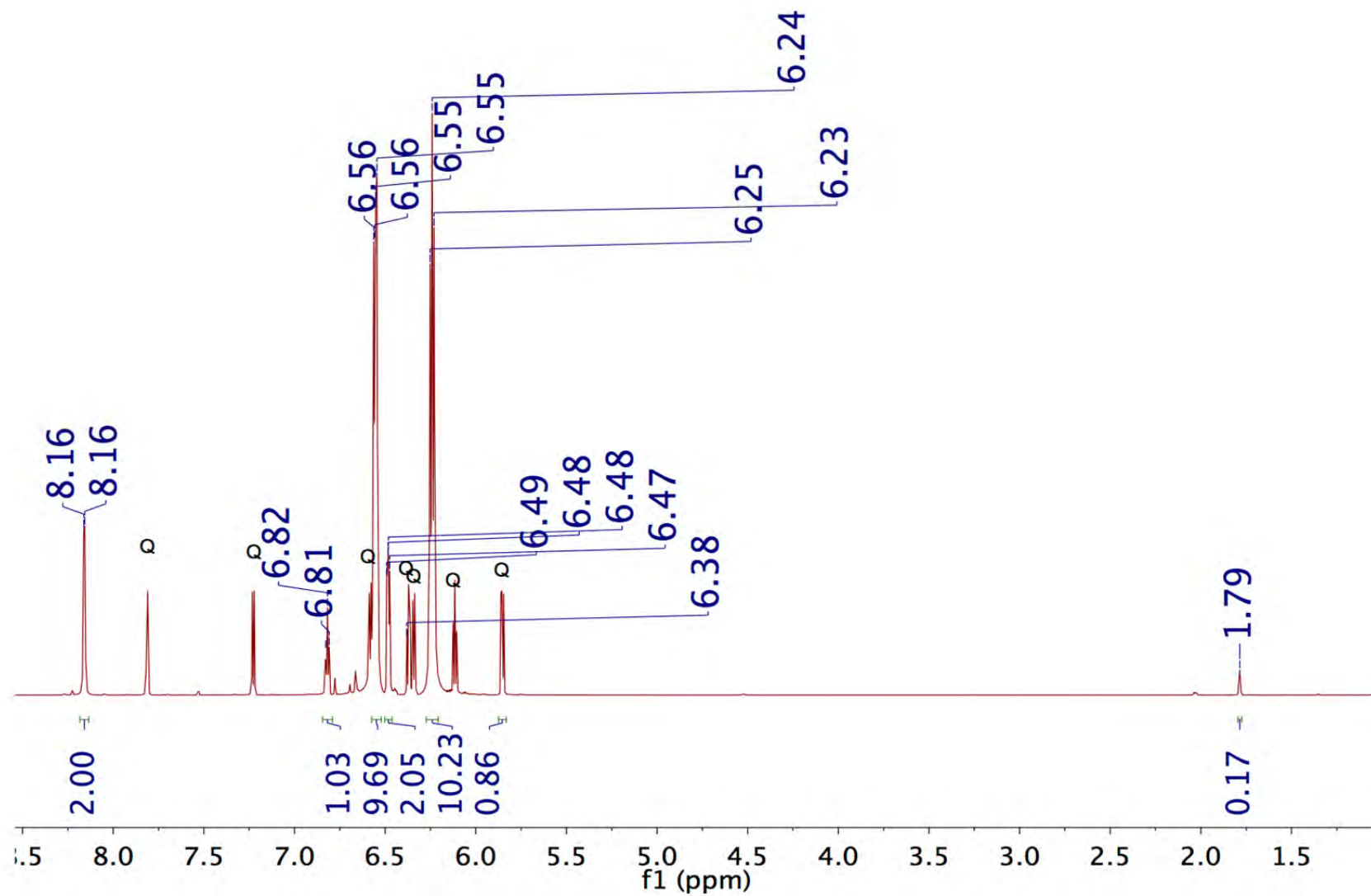


Figure S13: The ^1H NMR-spectrum of pyridine mixed with 5 equivalents of 4-chlorofluorobenzene in toluene- d_8 . Quinoline peaks are highlighted with “Q”.

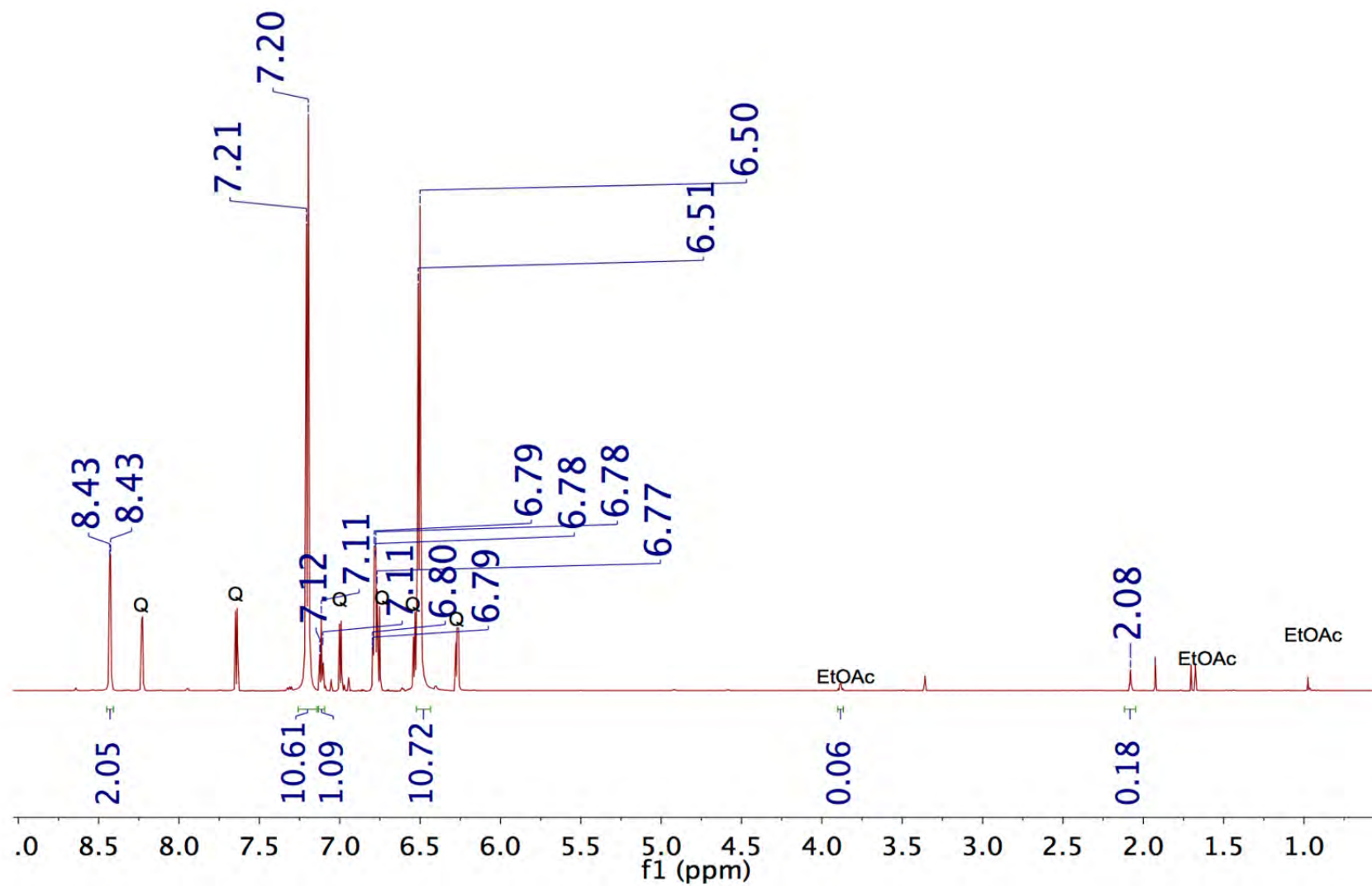


Figure S14: The ^1H NMR-spectrum of pyridine mixed with 5 equivalents of 4-(trifluoromethoxy)iodobenzene in toluene- d_8 . Quinoline peaks are highlighted with “Q”.

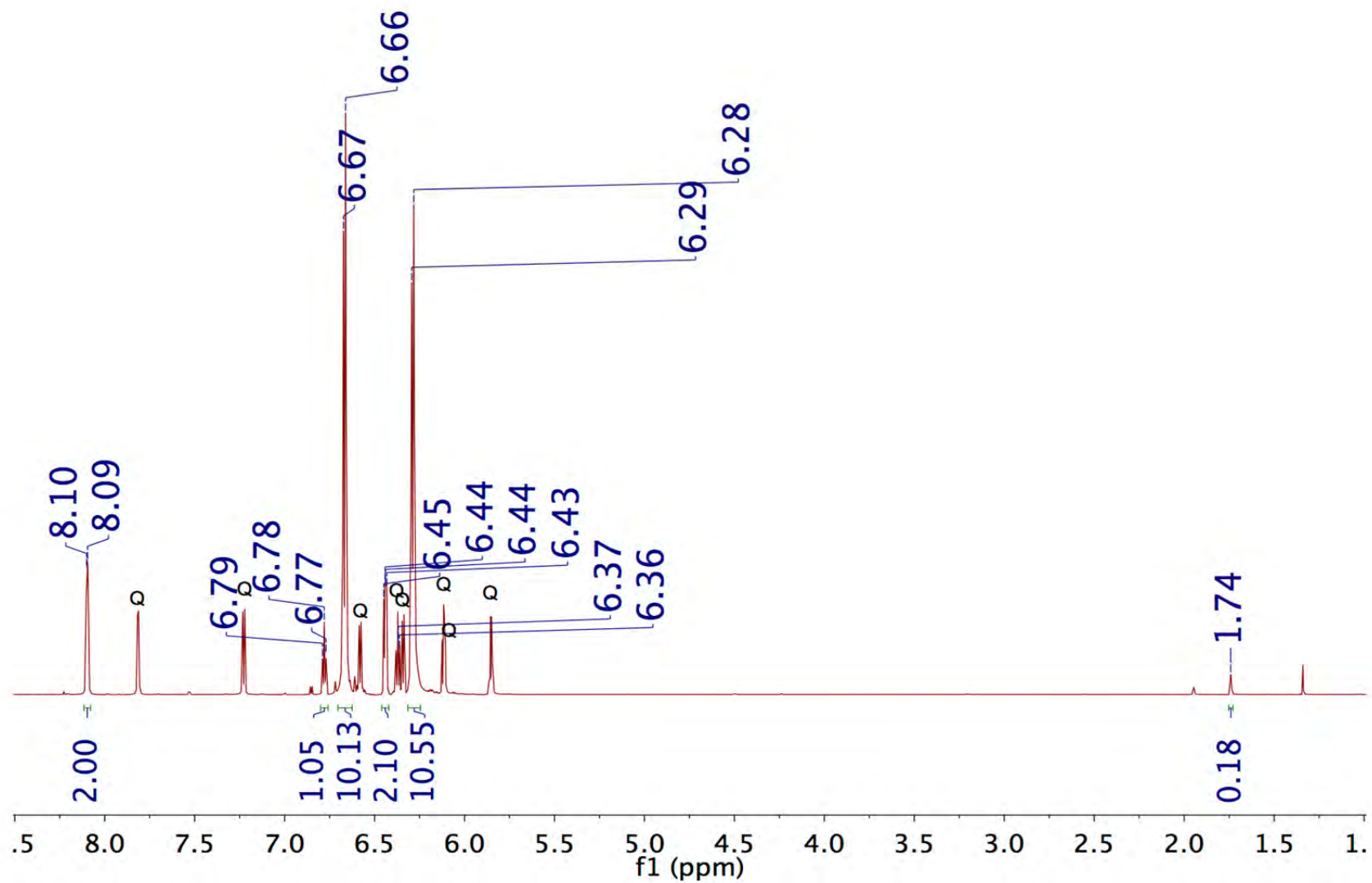


Figure S19: The ^1H NMR-spectrum of pyridine mixed with 5 equivalents of 4-(trifluoromethoxy)bromobenzene in toluene- d_8 . Quinoline peaks are highlighted with "Q".

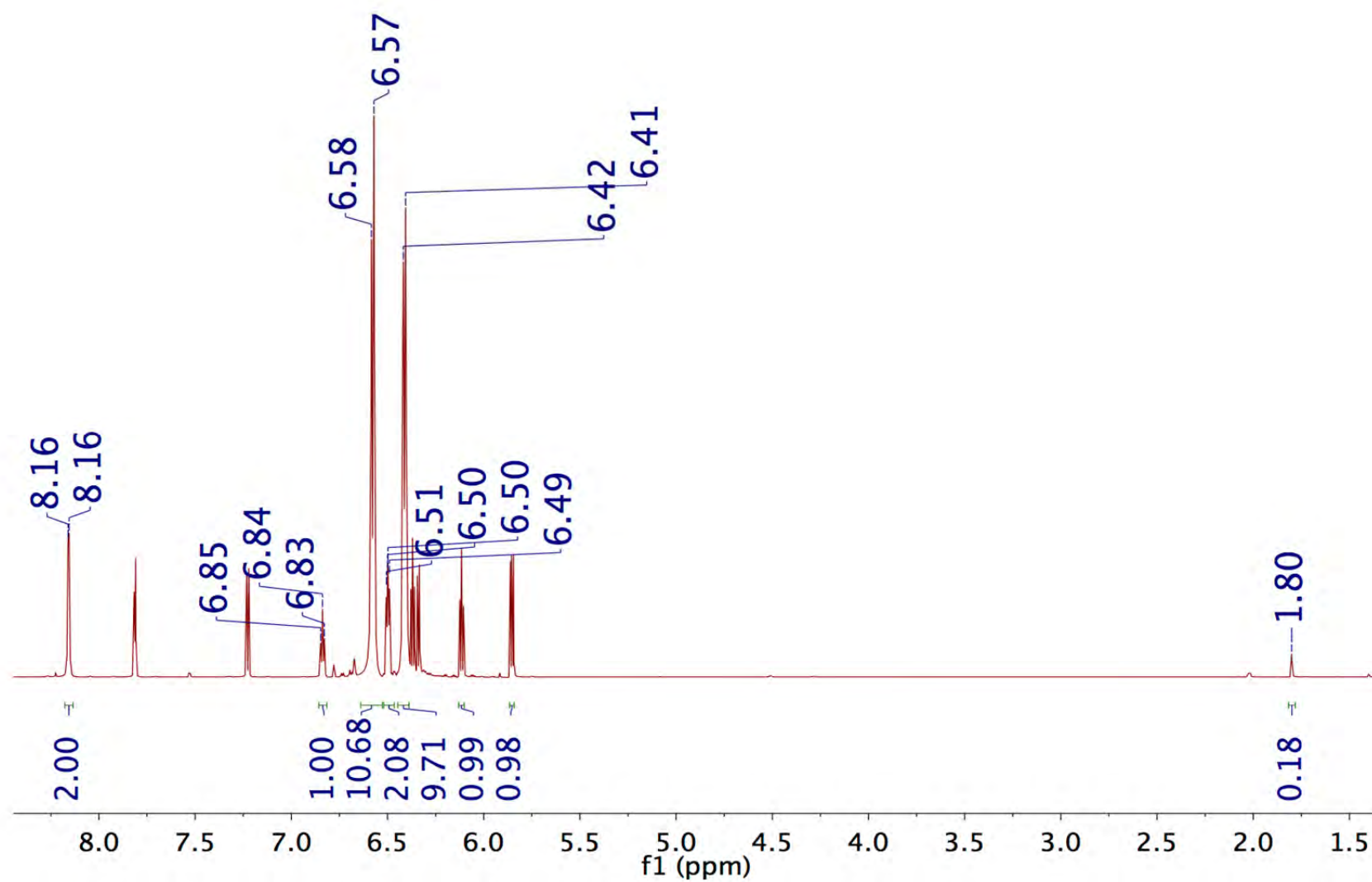


Figure S20: The ¹H NMR-spectrum of pyridine mixed with 5 equivalents of 4-(trifluoromethoxy)chlorobenzene in toluene-*d*₈. Quinoline peaks are highlighted with “Q”.

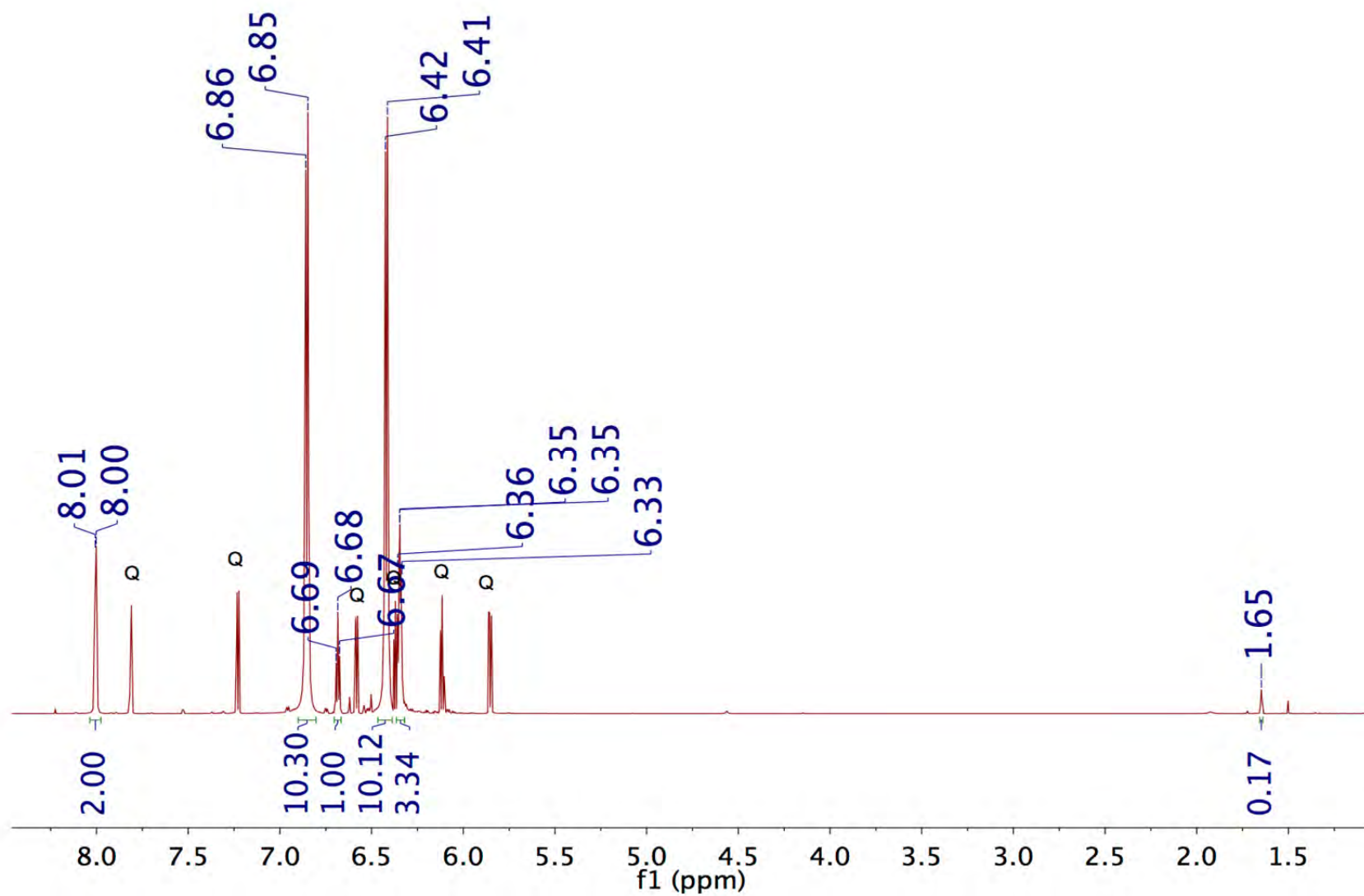


Figure S15: The ^1H NMR-spectrum of pyridine mixed with 5 equivalents of 4-iodobenzotrifluoride in toluene- d_8 . Quinoline peaks are highlighted with “Q”.

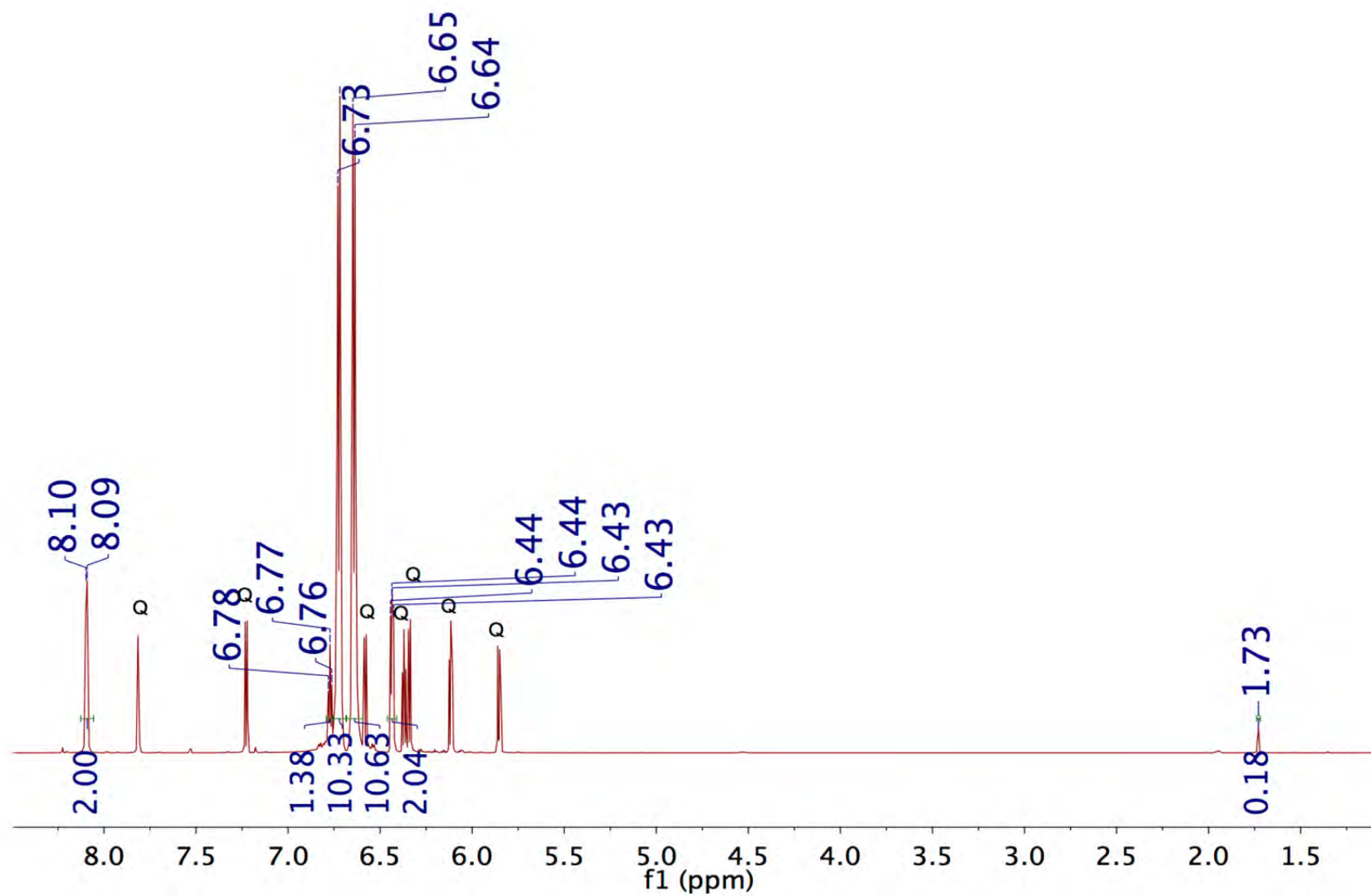


Figure S22: The ^1H NMR-spectrum of pyridine mixed with 5 equivalents of 4-bromobenzotrifluoride in toluene- d_8 . Quinoline peaks are highlighted with "Q".

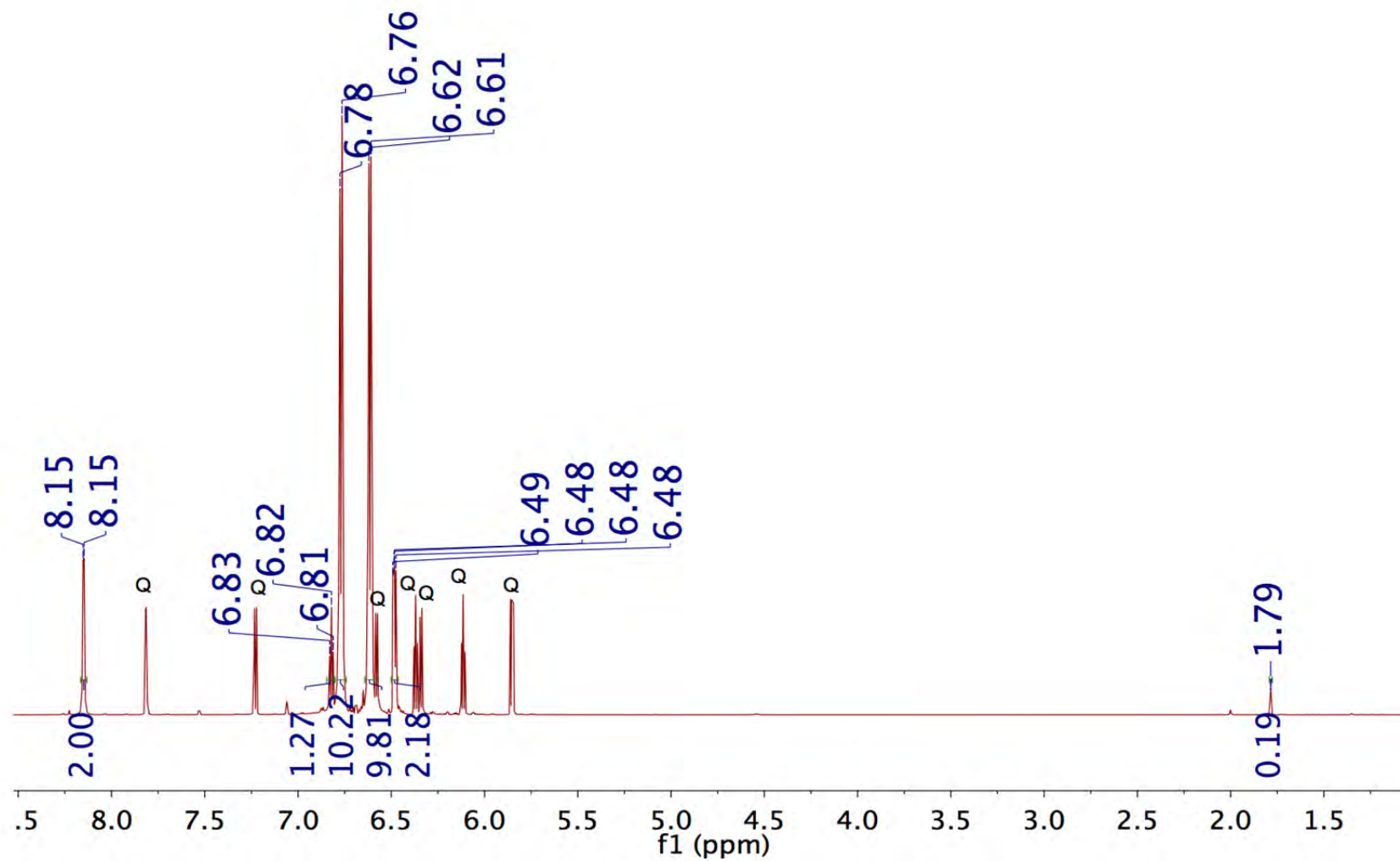


Figure S23: The ^1H NMR-spectrum of pyridine mixed with 5 equivalents of 4-chlorobenzotrifluoride in toluene- d_8 . Quinoline peaks are highlighted with "Q".

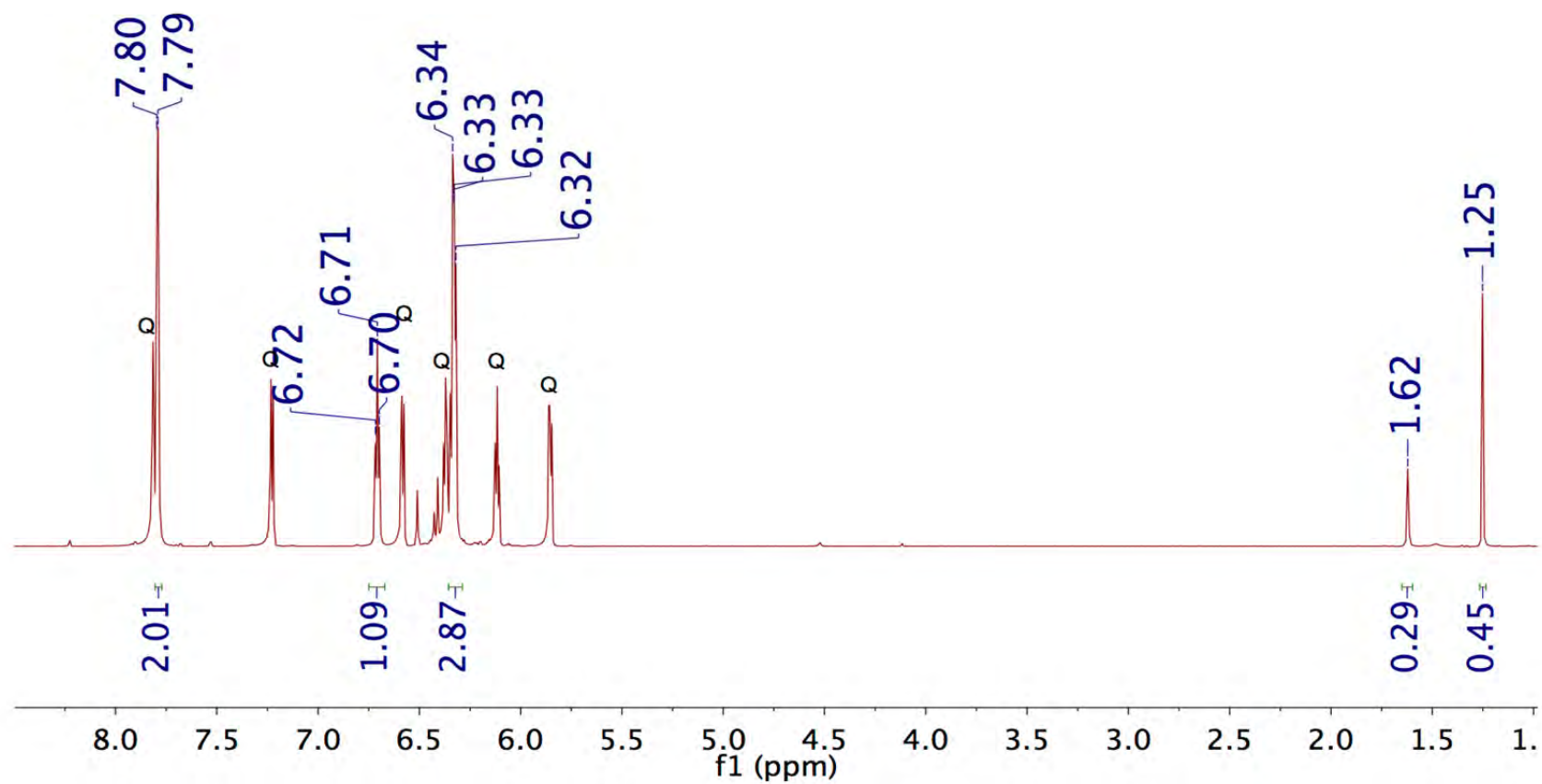


Figure S16: The ^1H NMR-spectrum of pyridine mixed with 5 equivalents of iodopentafluorobenzene in toluene- d_8 . Quinoline peaks are highlighted with "Q".

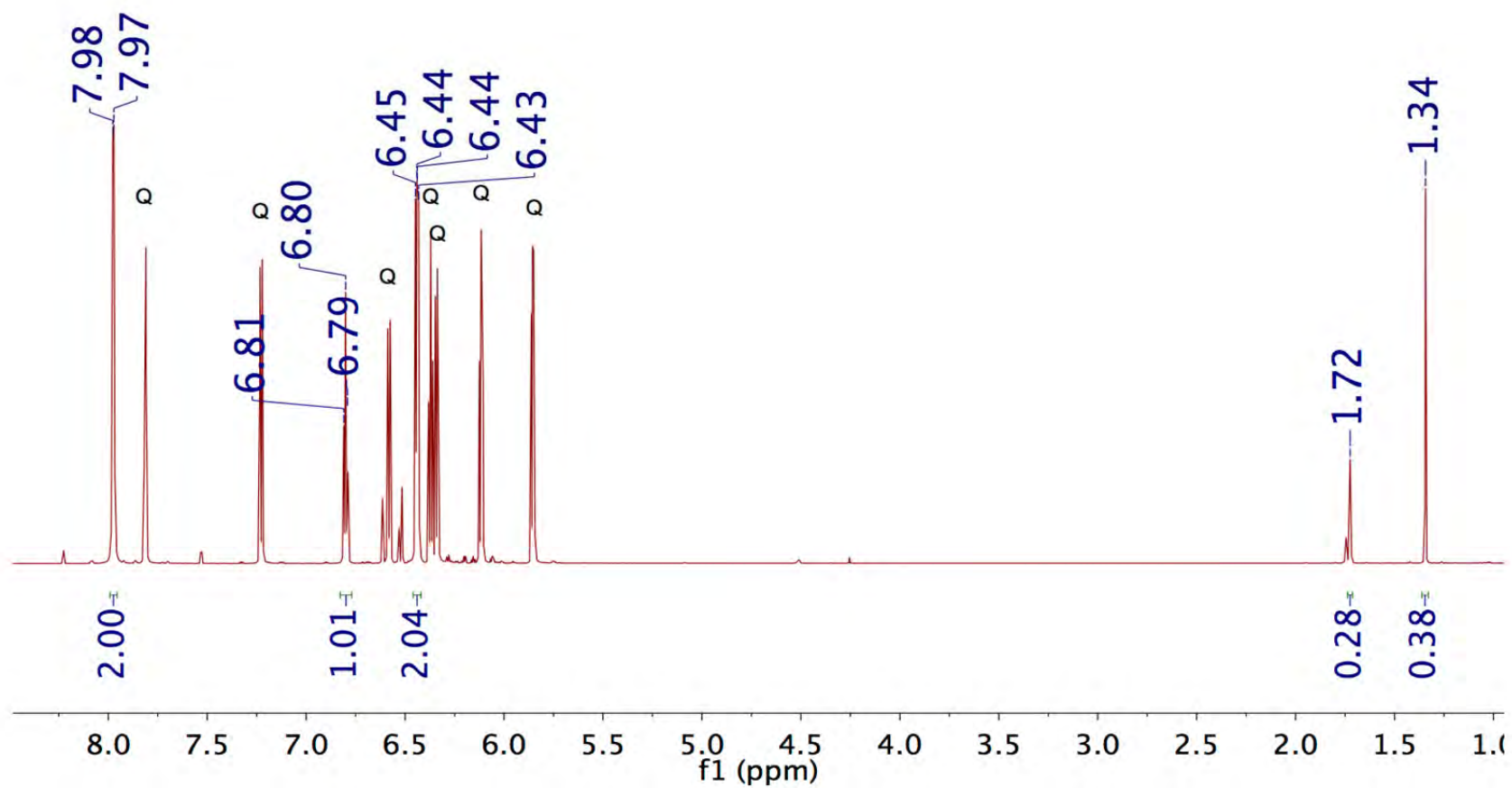


Figure S17: The ^1H NMR-spectrum of pyridine mixed with 5 equivalents of bromopentafluorobenzene in toluene- d_8 . Quinoline peaks are highlighted with “Q”.

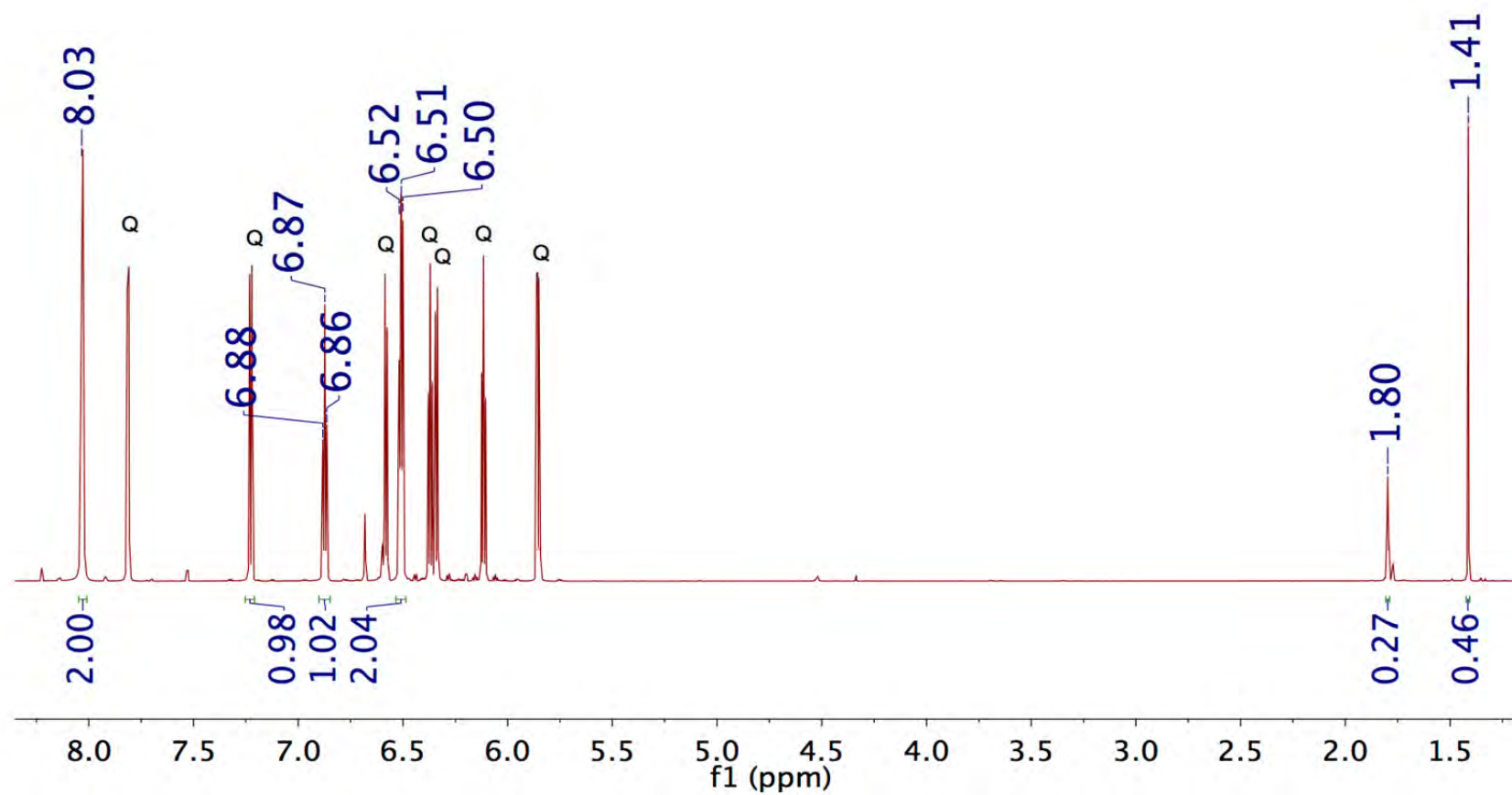


Figure S18: The ¹H NMR-spectrum of pyridine mixed with 5 equivalents of chloropentafluorobenzene in toluene-*d*₈. Quinoline peaks are highlighted with “Q”.

NMR data: pyridine:iodobenzene (1:3) series

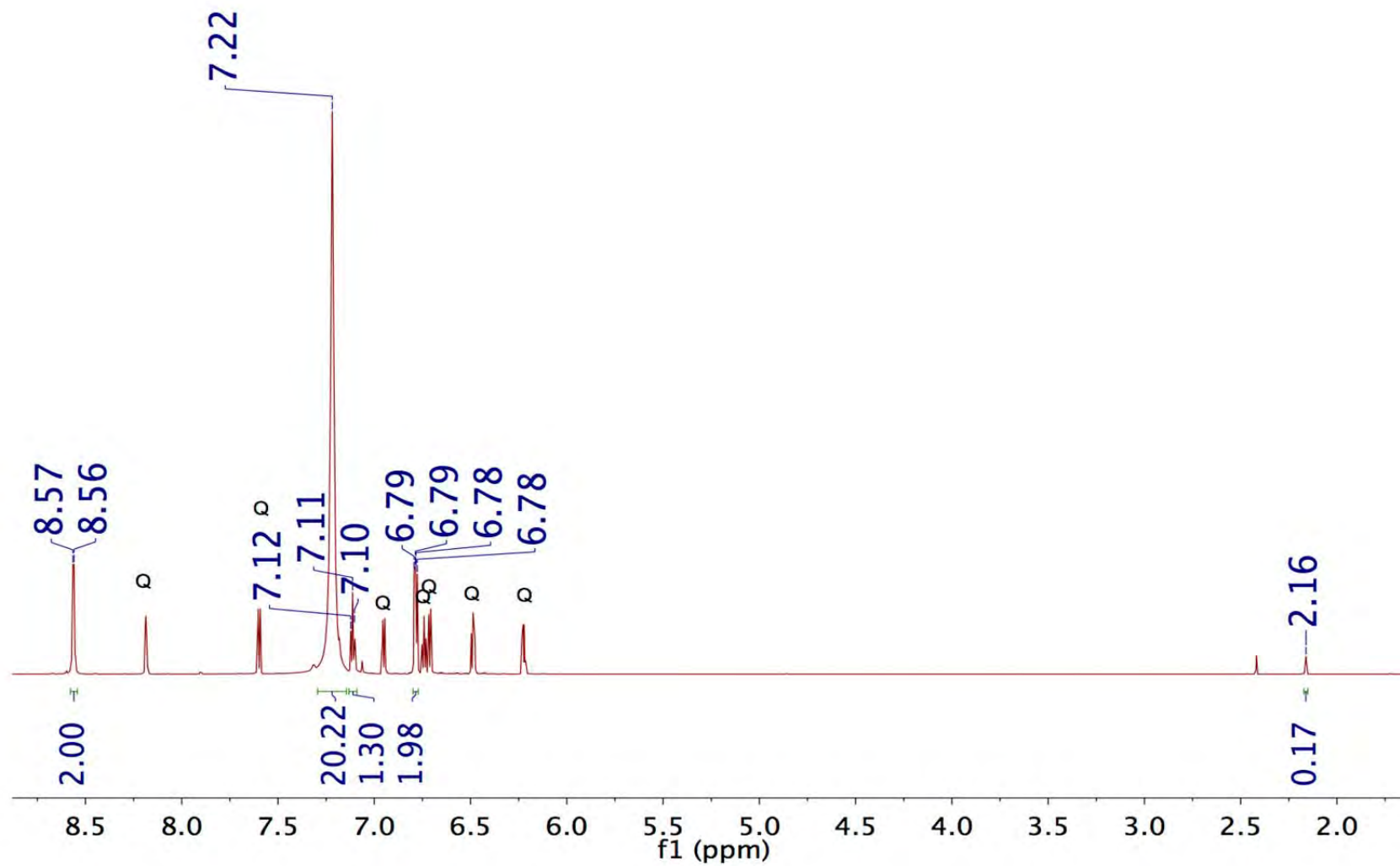


Figure S19: The ¹H NMR-spectrum of pyridine mixed with 3 equivalents of benzene in toluene-*d*₈. Quinoline peaks are highlighted with “Q”.

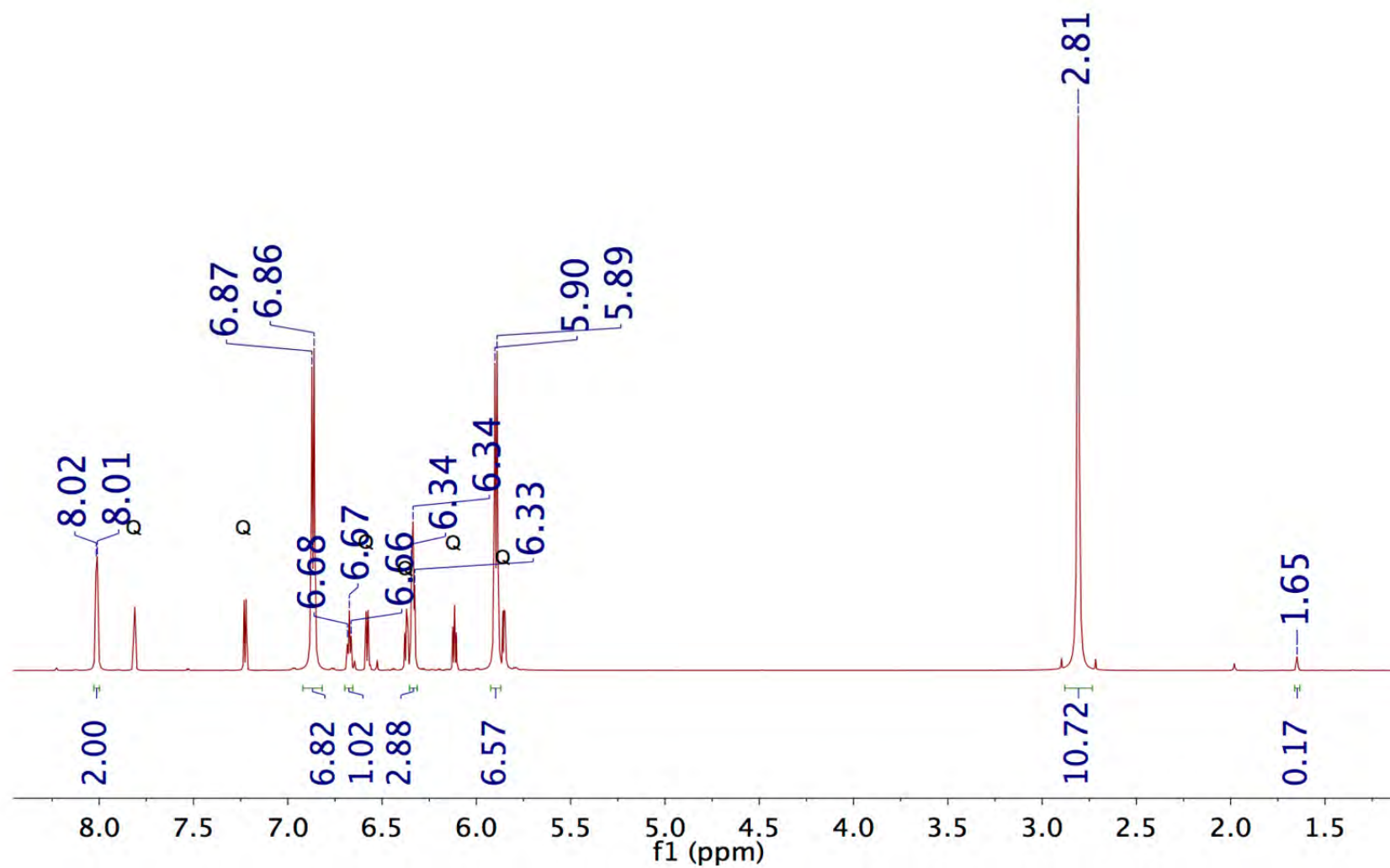


Figure S20: The ¹H NMR-spectrum of pyridine mixed with 3 equivalents of 4-iodoanisole in toluene-*d*₈. Quinoline peaks are highlighted with “Q”.

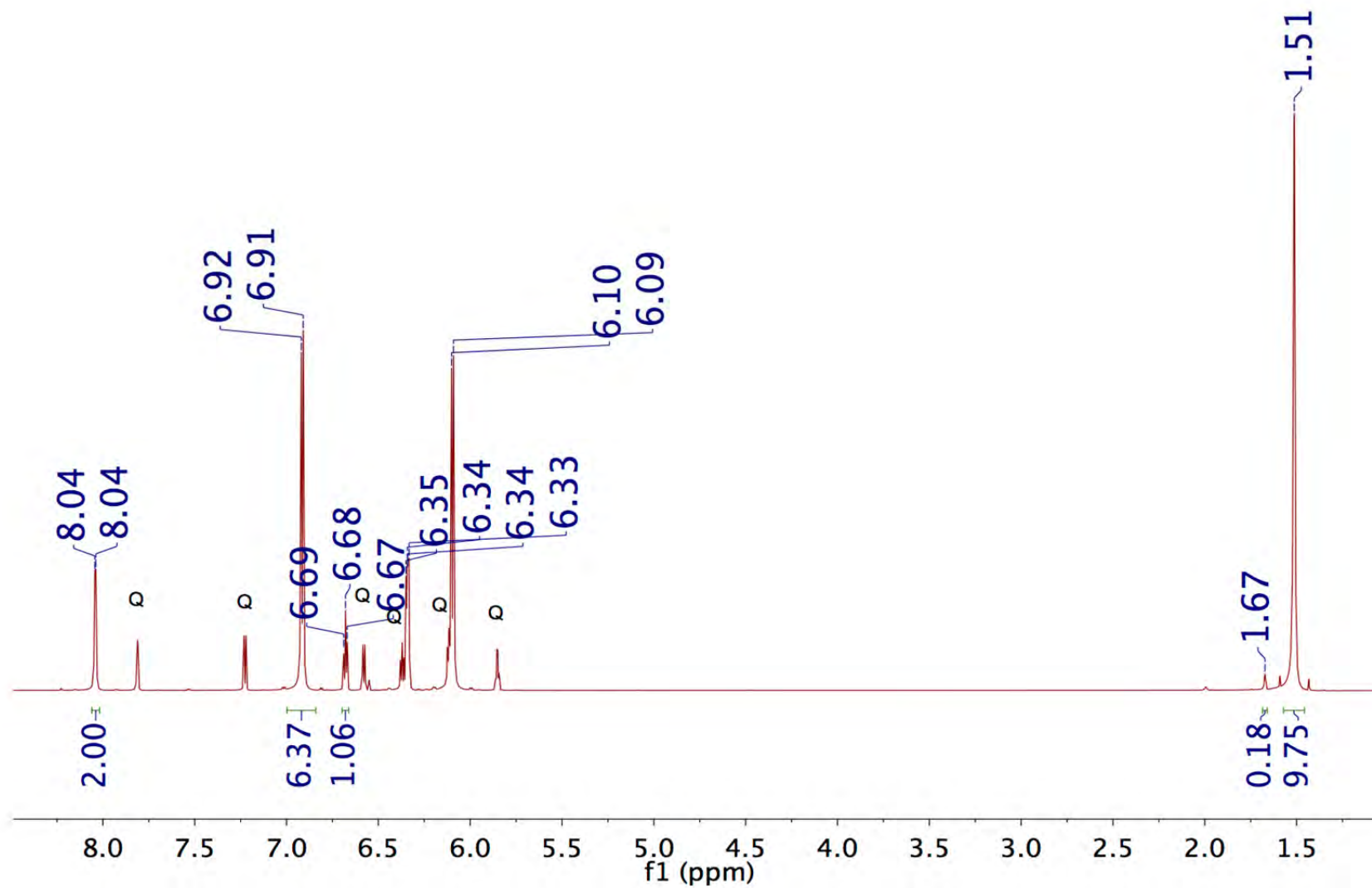


Figure S21: The ^1H NMR-spectrum of pyridine mixed with 3 equivalents of 4-iodotoluene in toluene- d_8 . Quinoline peaks are highlighted with "Q".

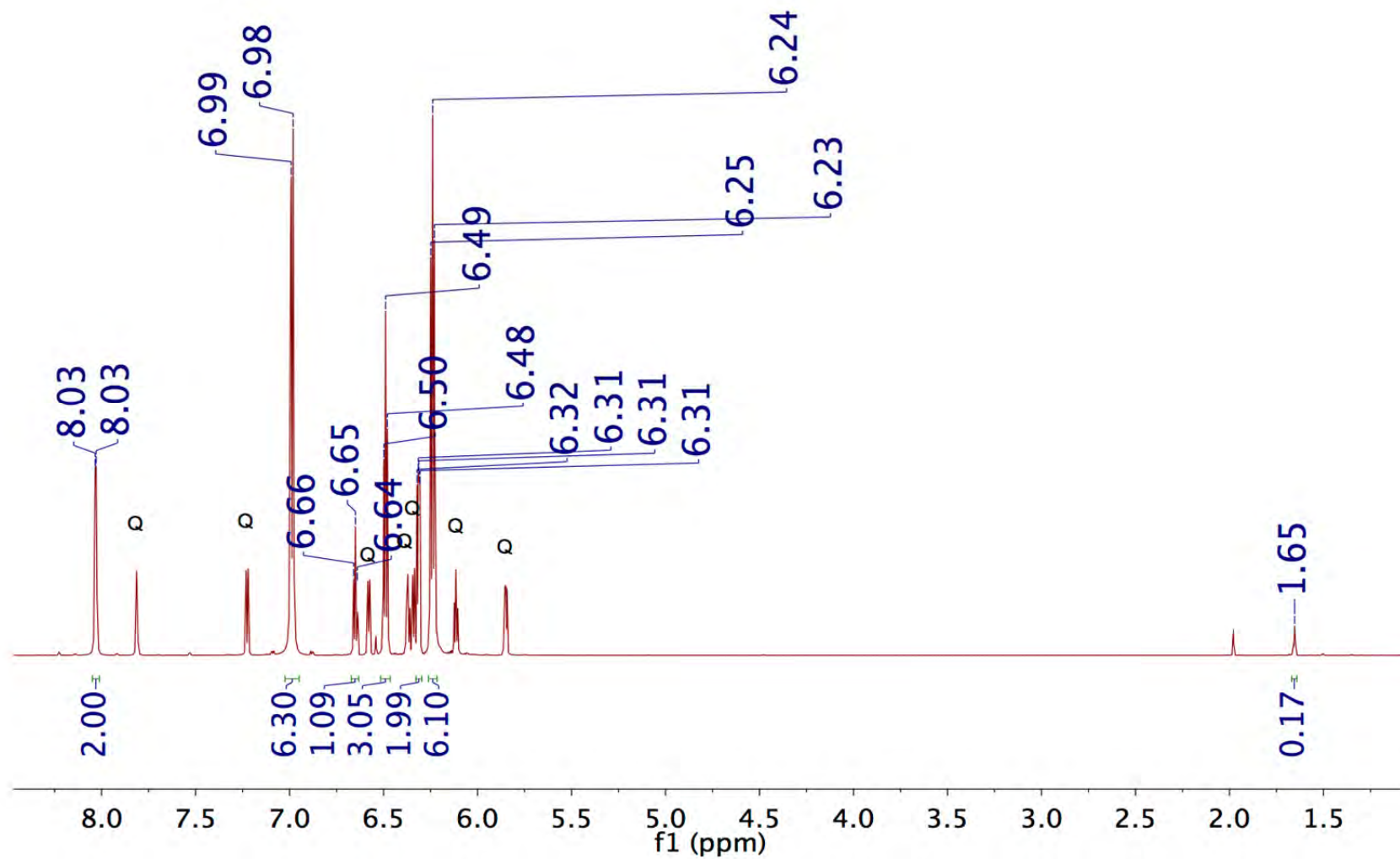


Figure S22: The ¹H NMR-spectrum of pyridine mixed with 3 equivalents of iodobenzene in toluene-*d*₈. Quinoline peaks are highlighted with “Q”.

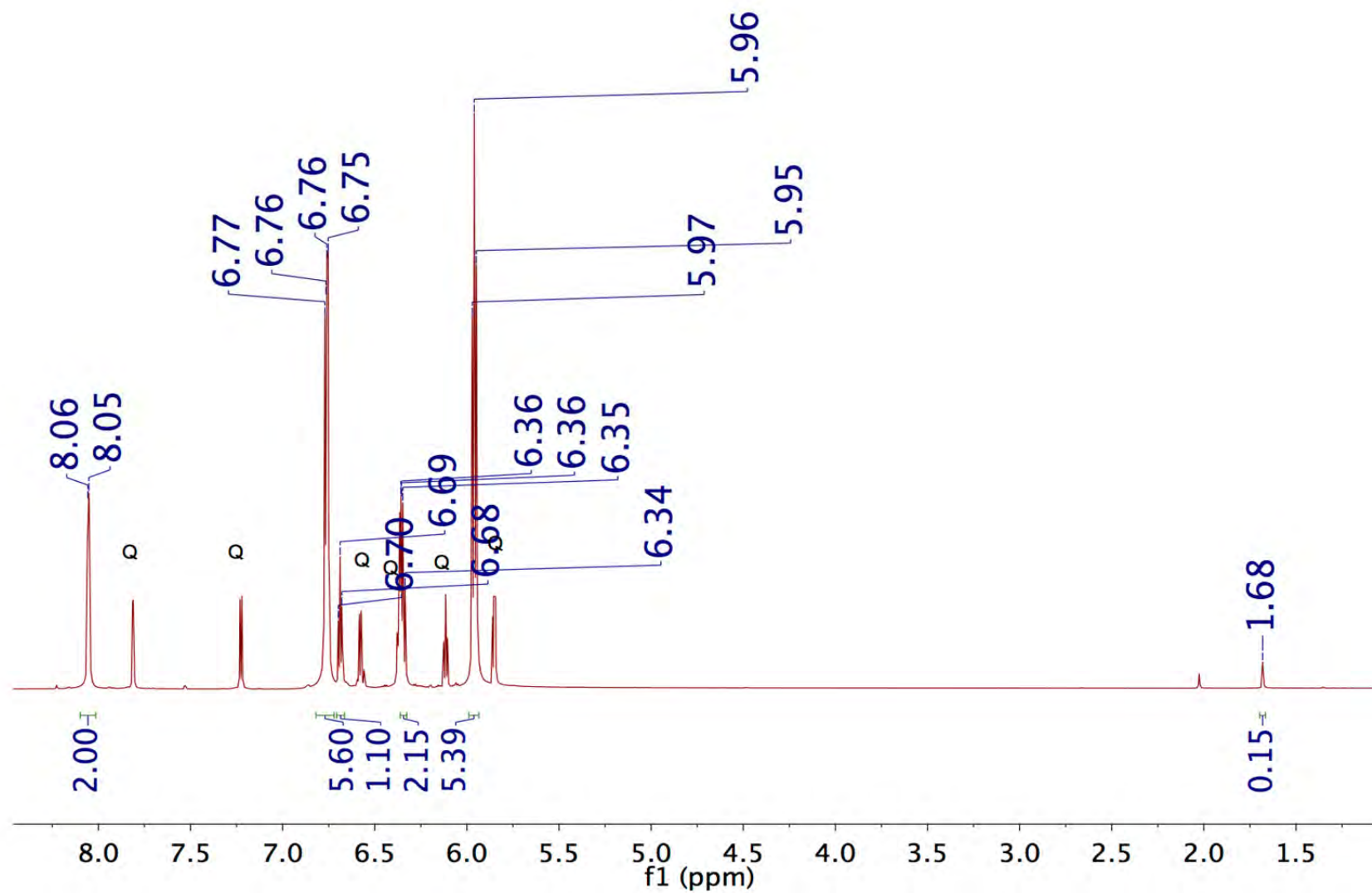


Figure S23: The ^1H NMR-spectrum of pyridine mixed with 3 equivalents of 4-fluoriodobenzene in toluene- d_8 . Quinoline peaks are highlighted with "Q".

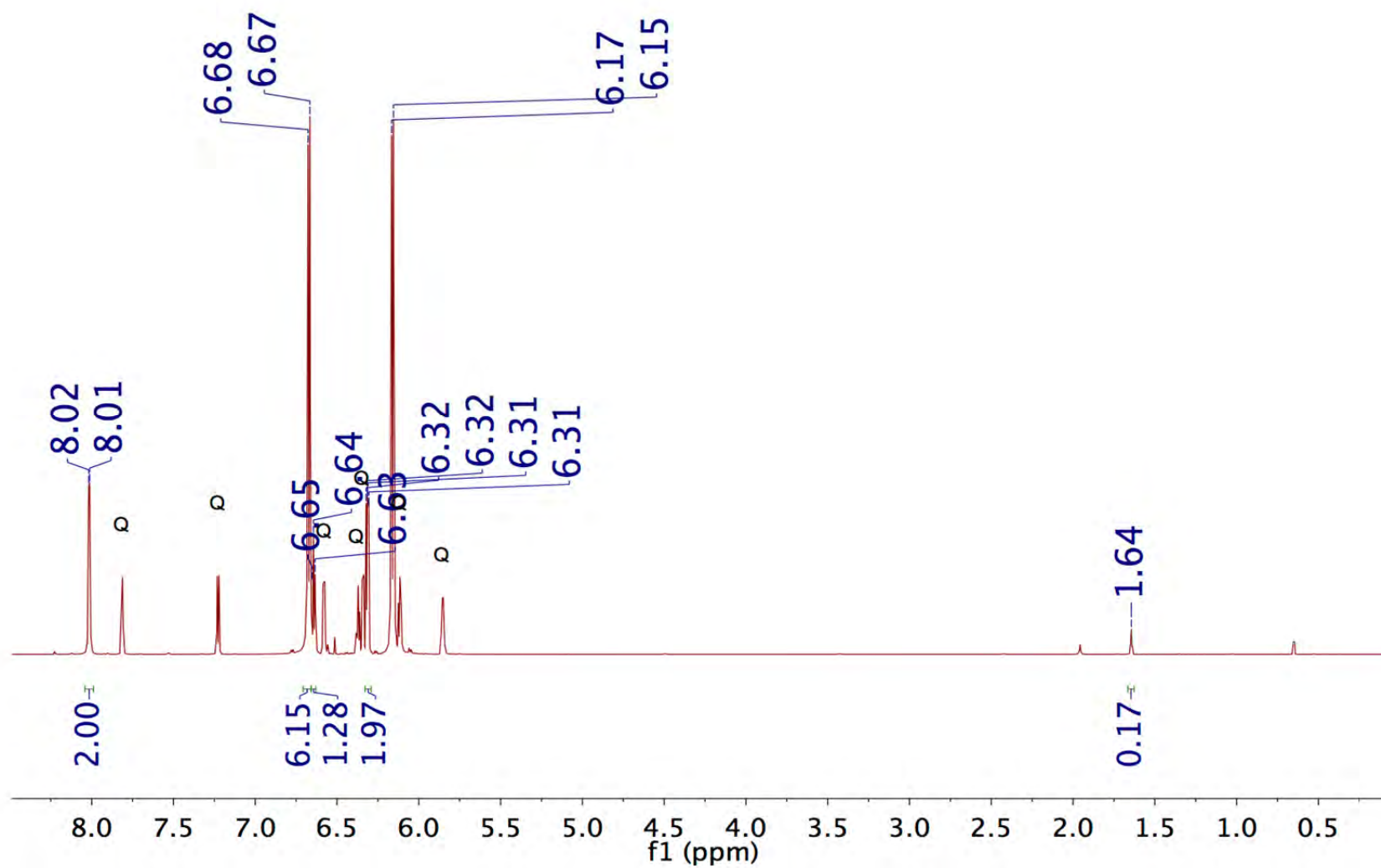


Figure S24: The ^1H NMR-spectrum of pyridine mixed with 3 equivalents of 4-chloriodobenzene in toluene- d_8 . Quinoline peaks are highlighted with "Q".

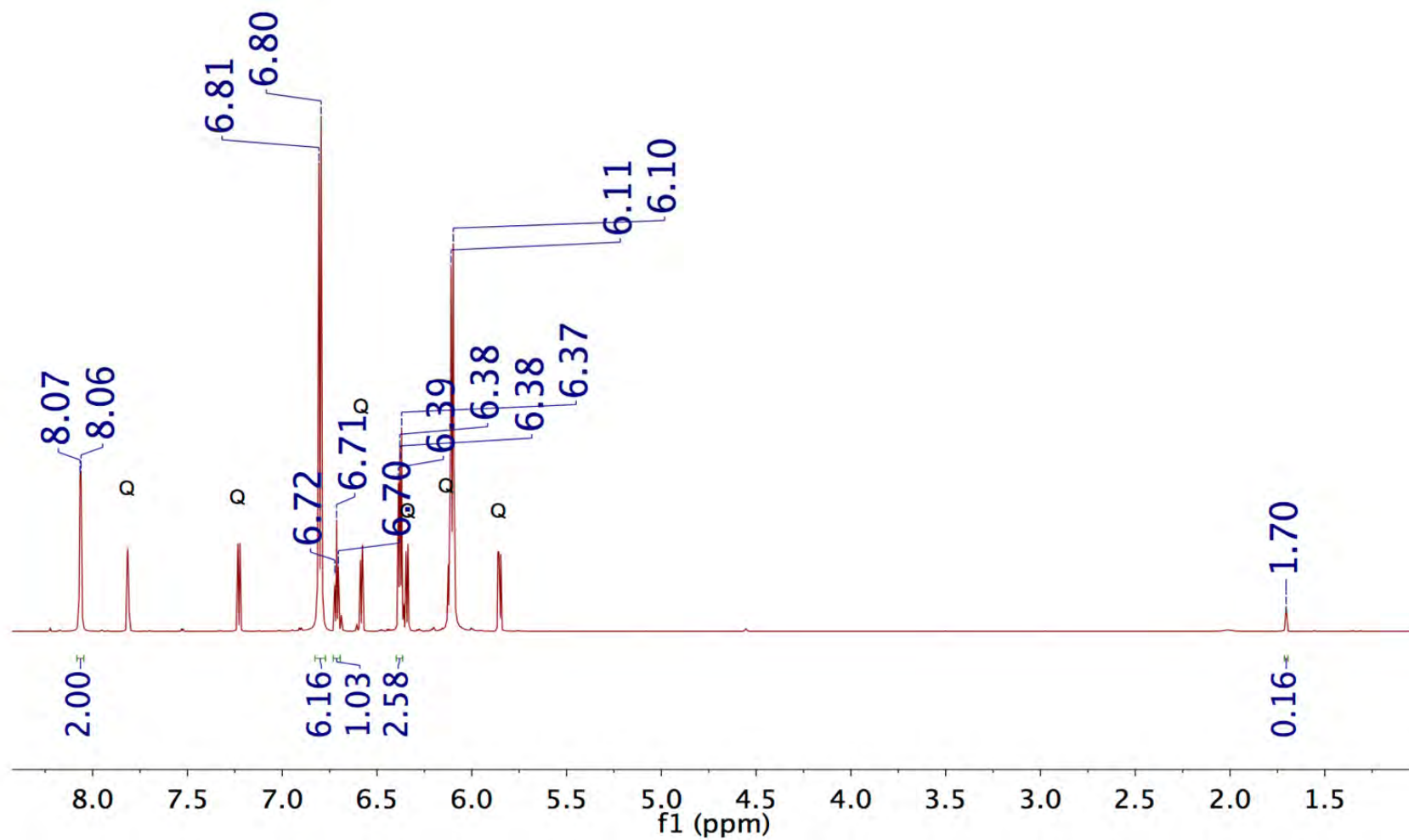


Figure S25: The ^1H NMR-spectrum of pyridine mixed with 3 equivalents of 4-(trifluoromethoxy)iodobenzene in toluene- d_8 . Quinoline peaks are highlighted with “Q”.

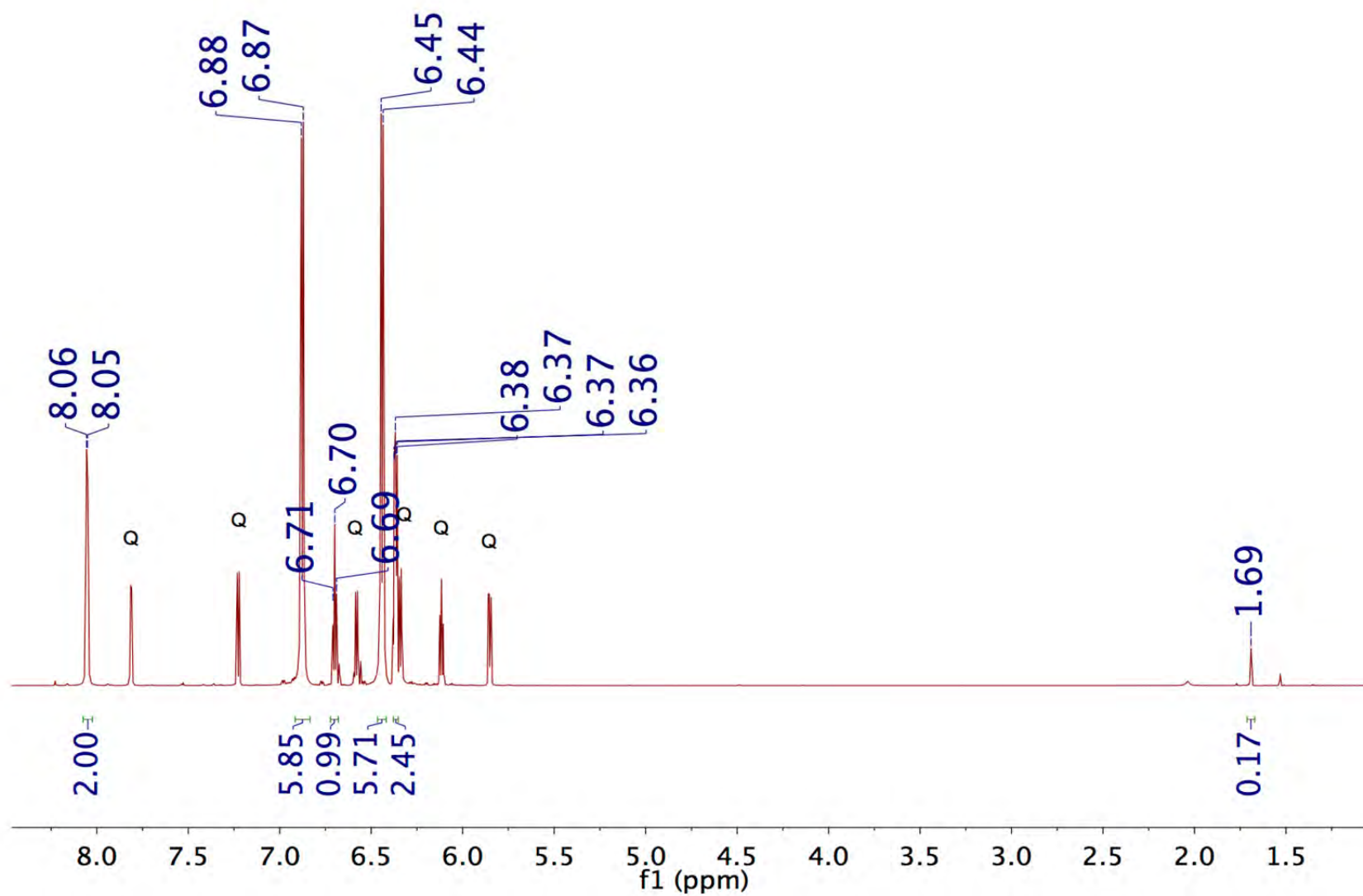


Figure S26: The ^1H NMR-spectrum of pyridine mixed with 3 equivalents of 4-iodobenzotrifluoride in toluene- d_8 . Quinoline peaks are highlighted with “Q”.

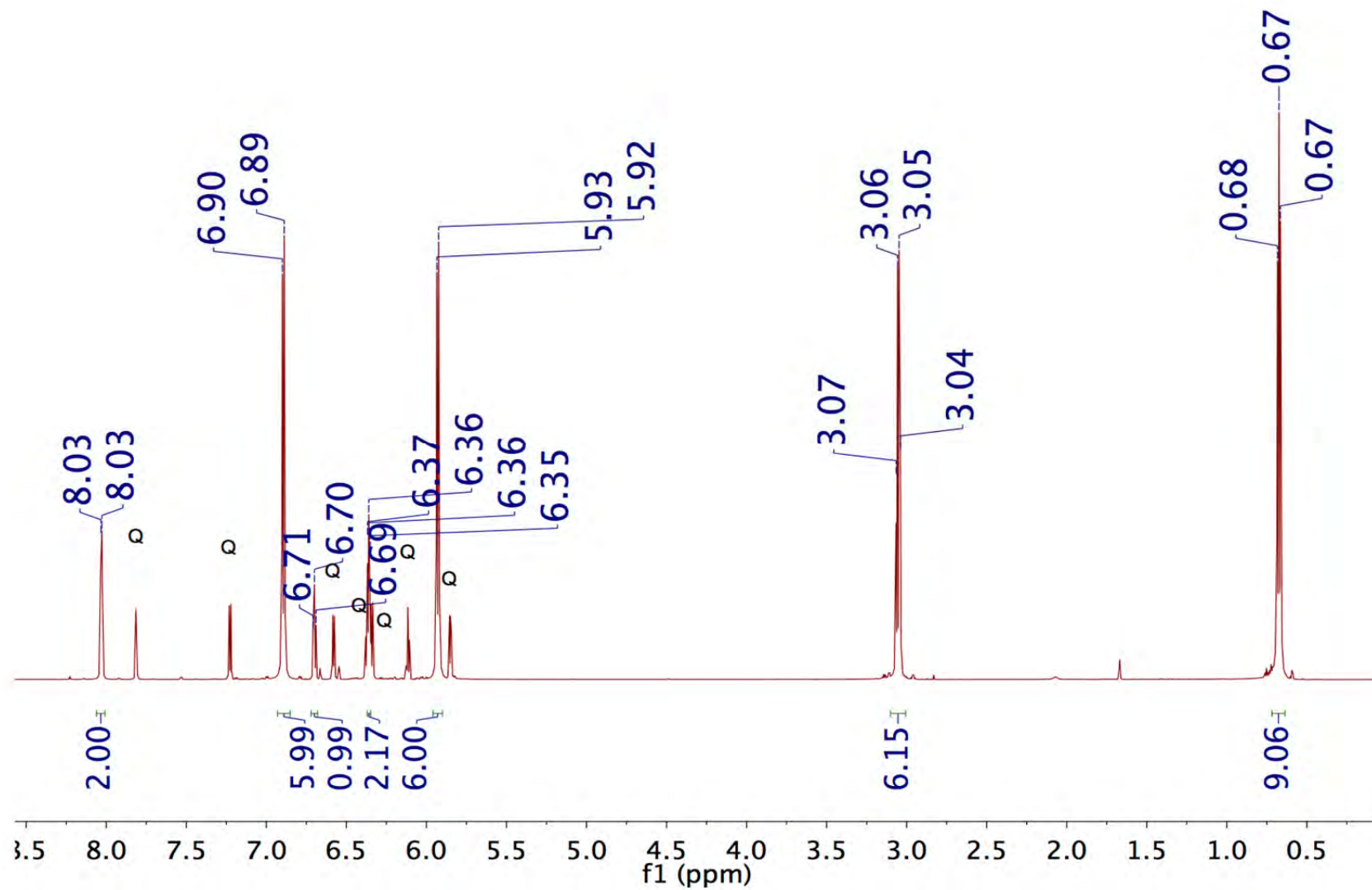


Figure S27: The ^1H NMR-spectrum of pyridine mixed with 3 equivalents of 4-iodophenetole in toluene- d_8 . Quinoline peaks are highlighted with "Q".

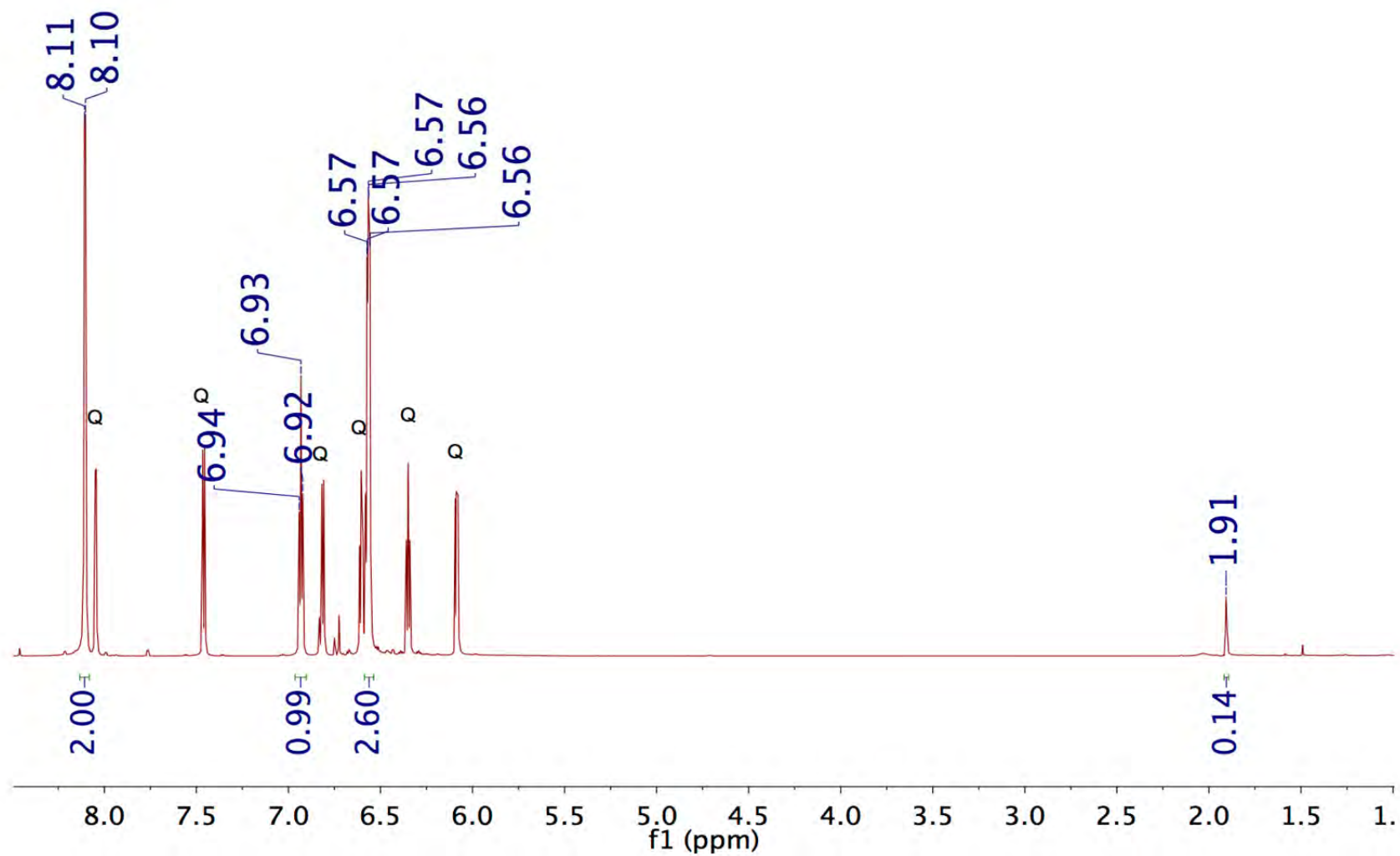


Figure S28: The ¹H NMR-spectrum of pyridine mixed with 3 equivalents of iodopentafluorobenzene in toluene-*d*₈. Quinoline peaks are highlighted with "Q".

NMR data: pyridine:phenol (1:3) series

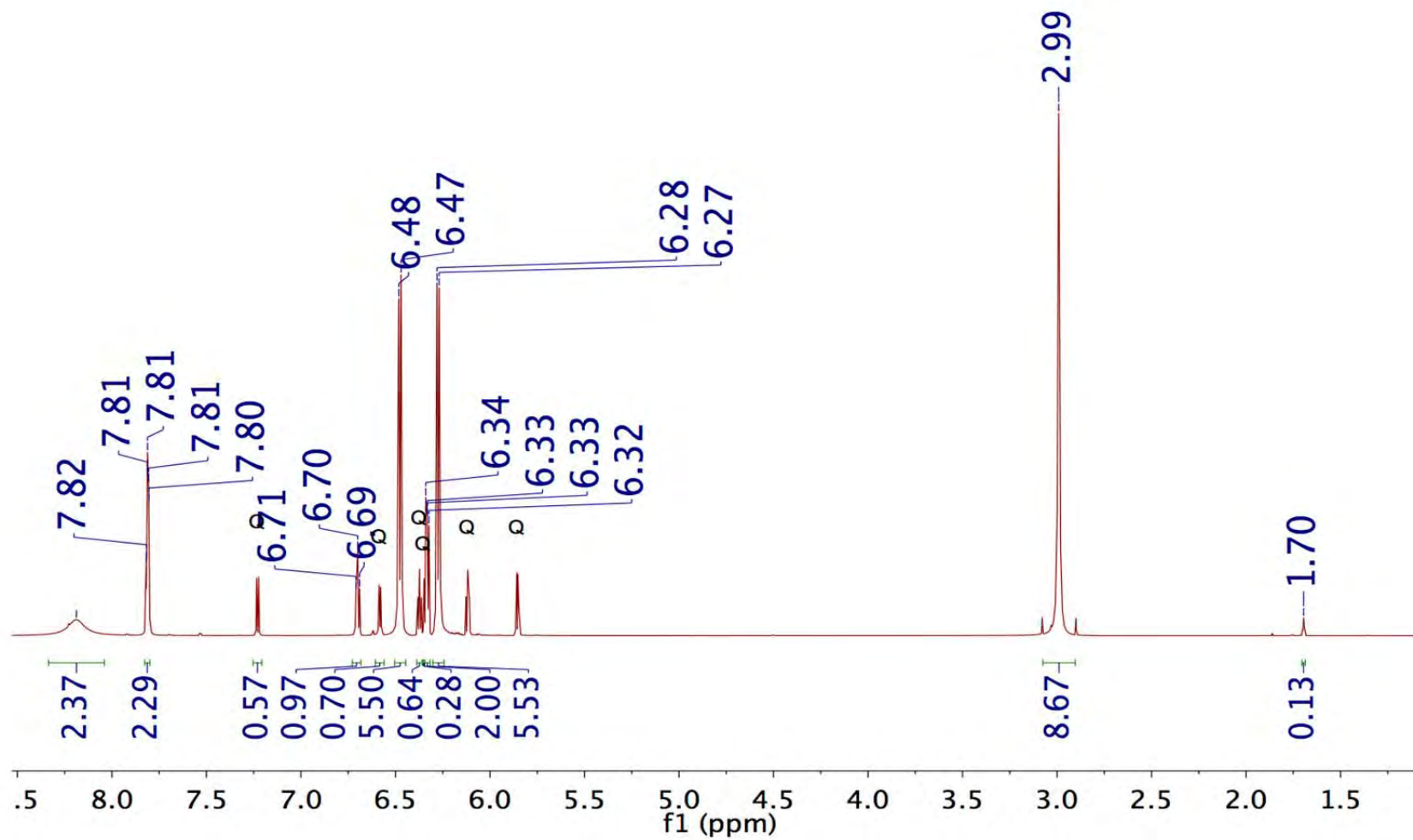


Figure S37: The ¹H NMR-spectrum of pyridine mixed with 3 equivalents of 4-methoxyphenol in toluene-*d*₈. Quinoline peaks are highlighted with "Q".

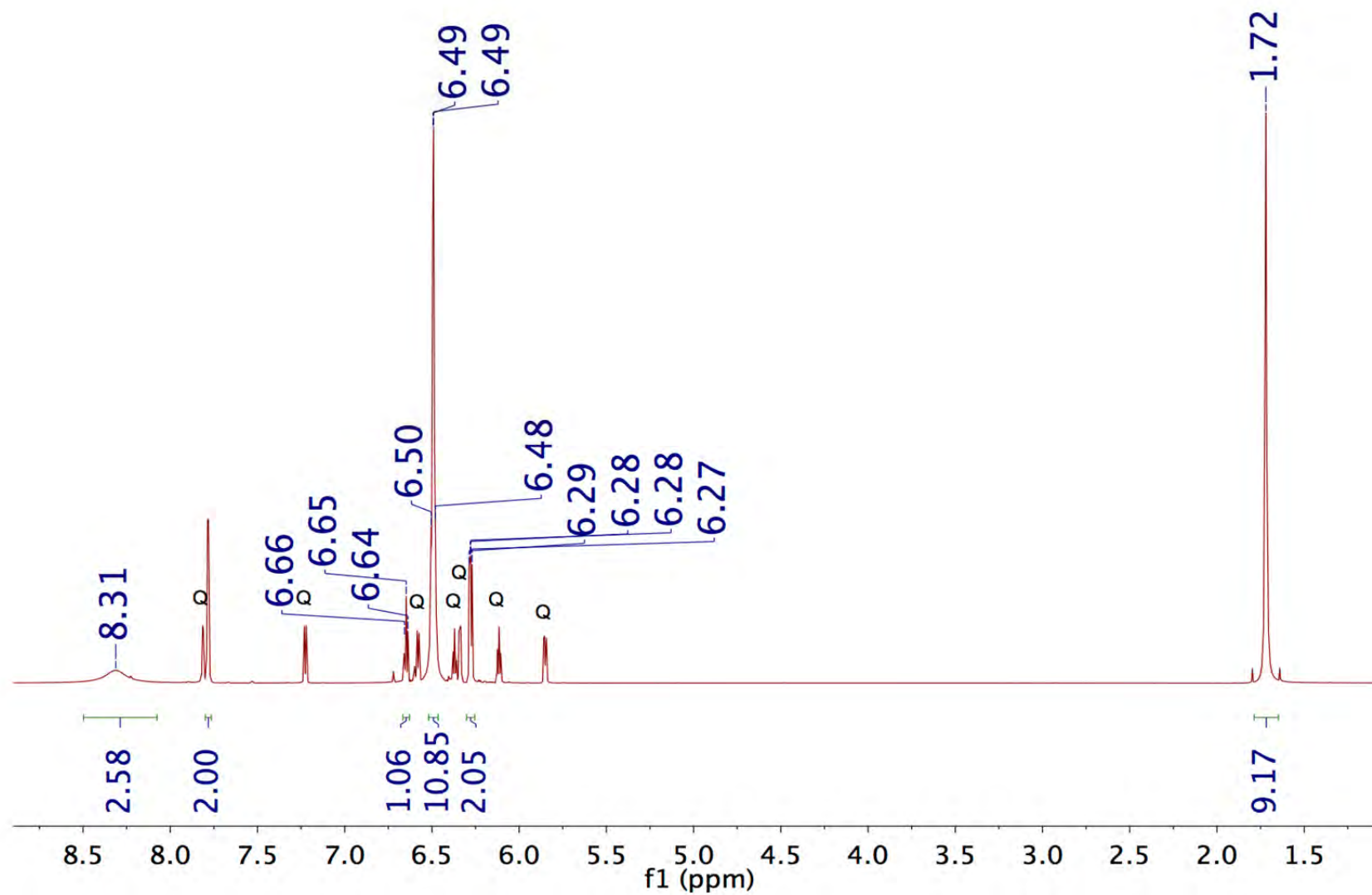


Figure S38: The ^1H NMR-spectrum of pyridine mixed with 3 equivalents of *p*-cresol in toluene- d_8 . Quinoline peaks are highlighted with "Q".

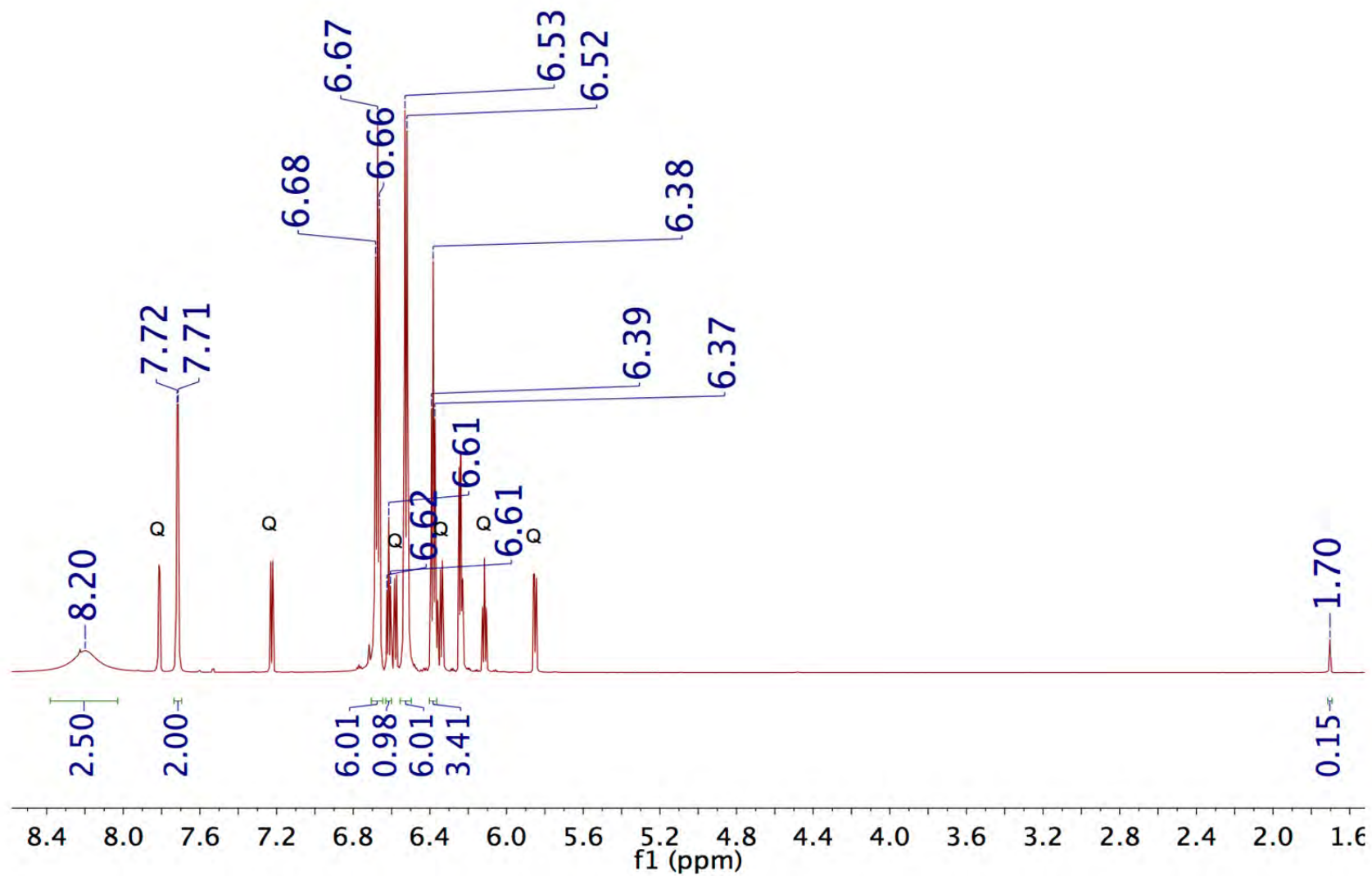


Figure S39: The ^1H NMR-spectrum of pyridine mixed with 3 equivalents of phenol in toluene- d_8 . Quinoline peaks are highlighted with “Q”.

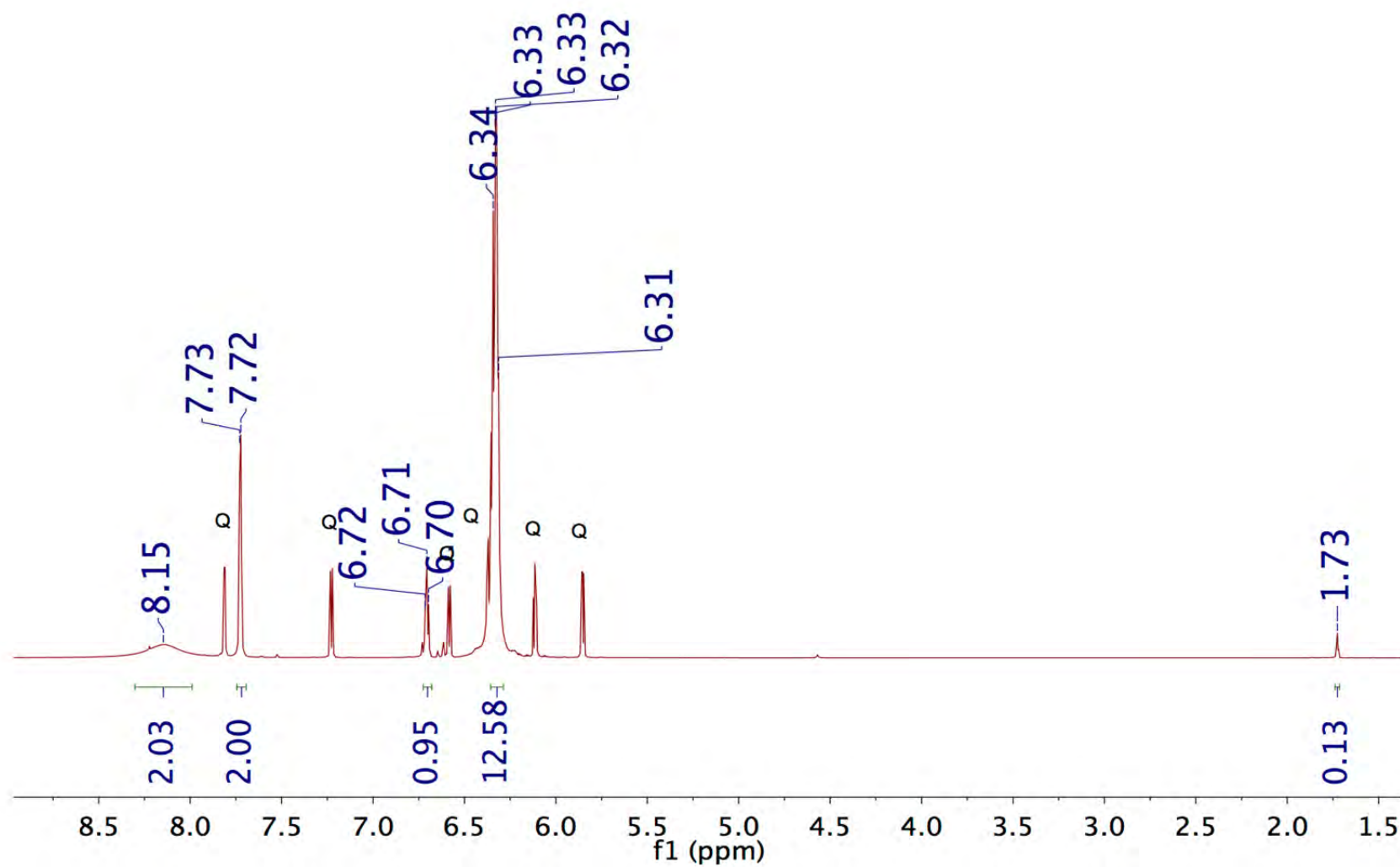


Figure S29: The ¹H NMR-spectrum of pyridine mixed with 3 equivalents of 4-fluorophenol in toluene-*d*₈. Quinoline peaks are highlighted with “Q”.

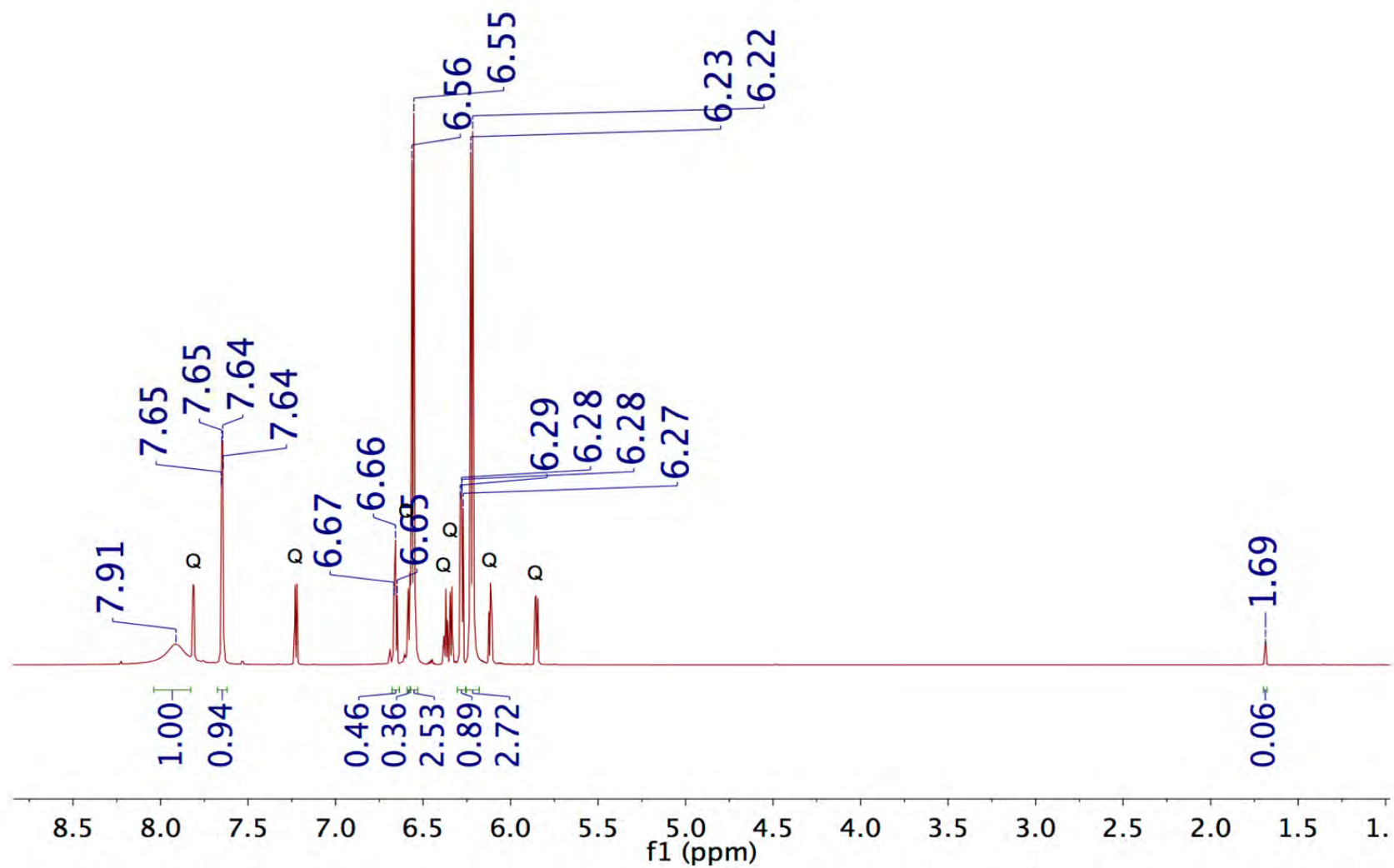


Figure S30: The ^1H NMR-spectrum of pyridine mixed with 3 equivalents of 4-chlorophenol in toluene- d_8 . Quinoline peaks are highlighted with “Q”.

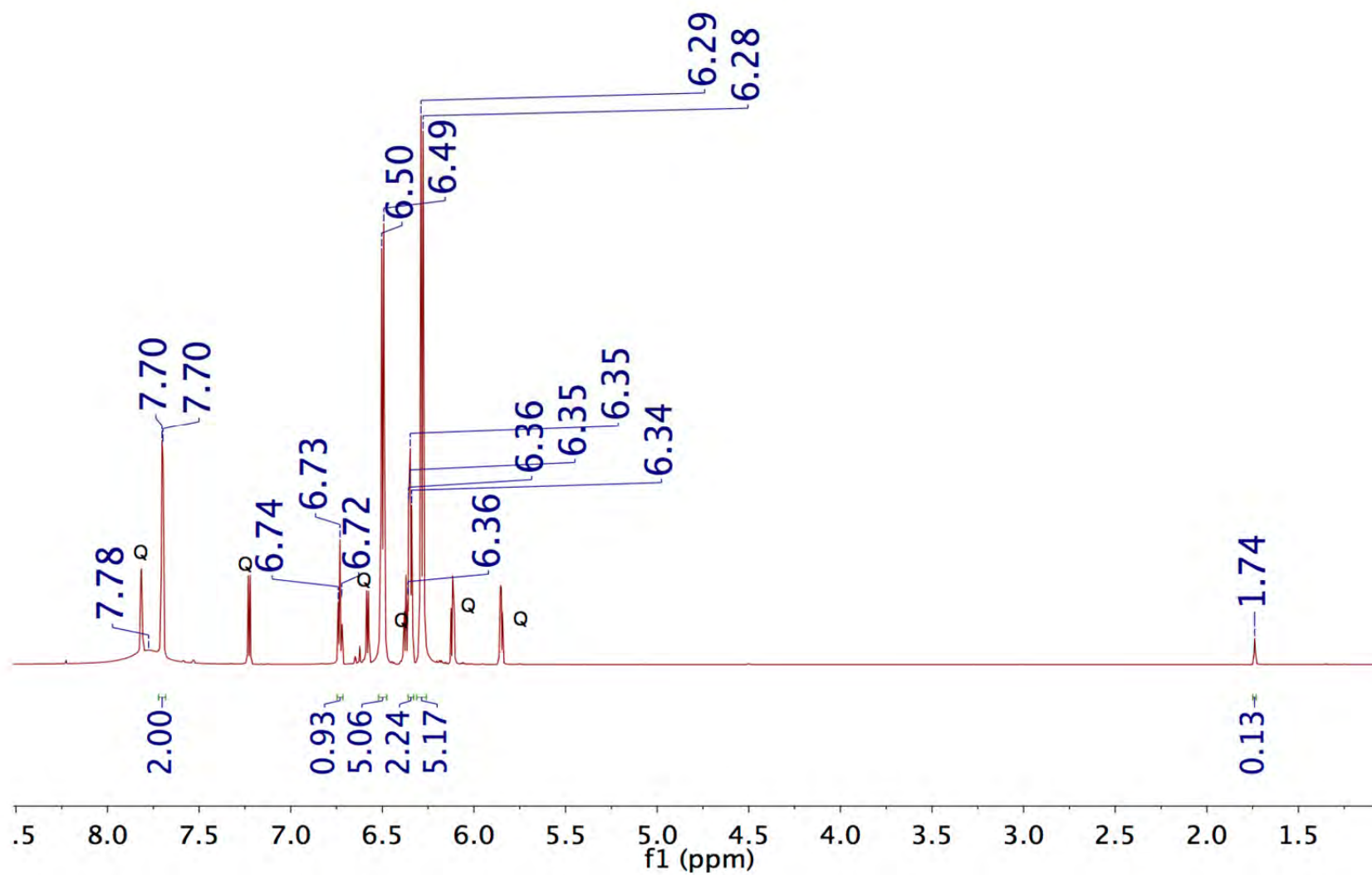


Figure S31: The ¹H NMR-spectrum of pyridine mixed with 3 equivalents of 4-(trifluoromethoxy)phenol in toluene-*d*₈. Quinoline peaks are highlighted with “Q”.

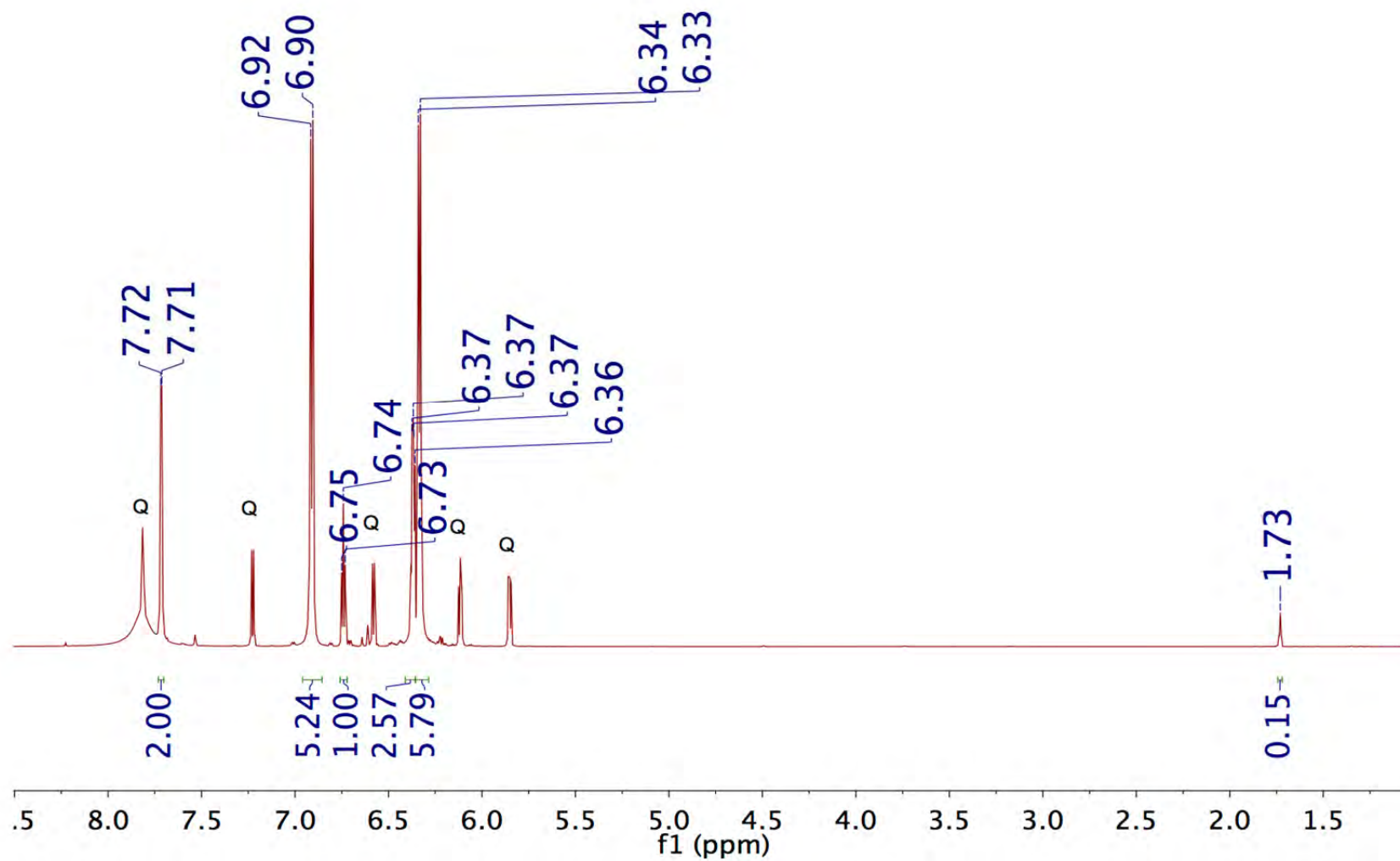


Figure S32: The ¹H NMR-spectrum of pyridine mixed with 3 equivalents of 4-(trifluoromethyl)phenol in toluene-*d*₈. Quinoline peaks are highlighted with “Q”.

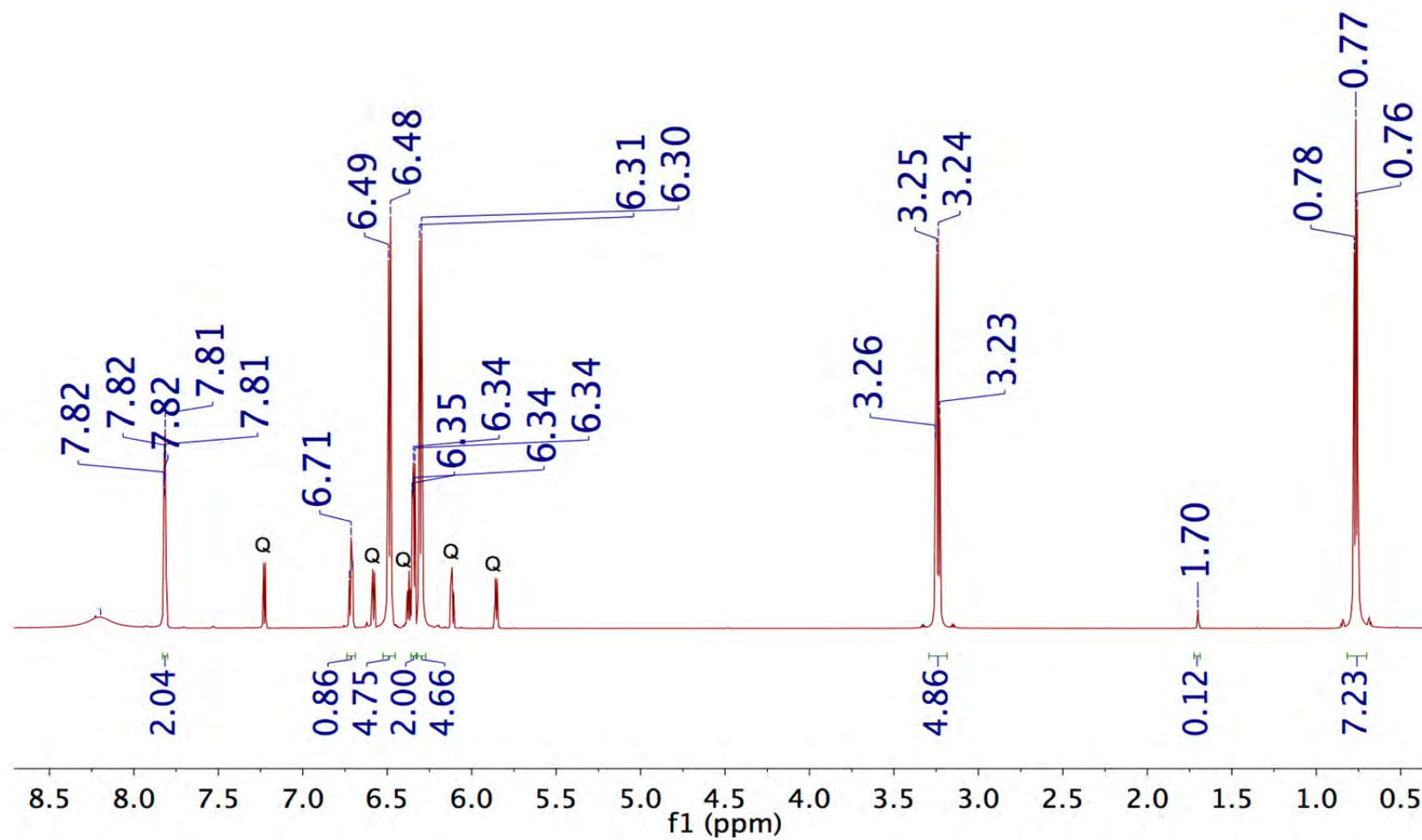


Figure S33: The ^1H NMR-spectrum of pyridine mixed with 3 equivalents of 4-ethoxyphenol in toluene- d_8 . Quinoline peaks are highlighted with “Q”.

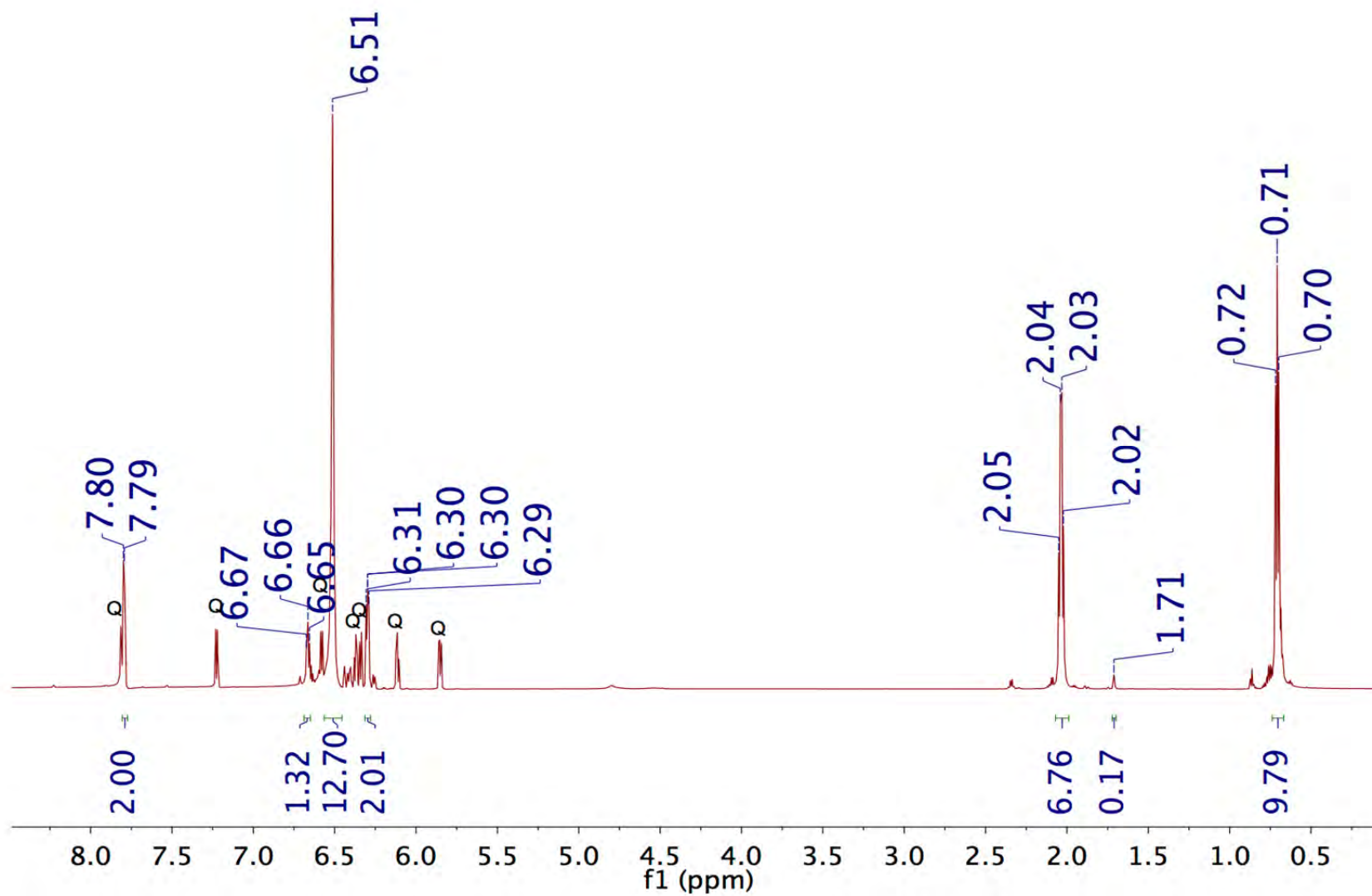


Figure S45: The ^1H NMR-spectrum of pyridine mixed with 3 equivalents of 4-ethylphenol in toluene- d_8 . Quinoline peaks are highlighted with “Q”.

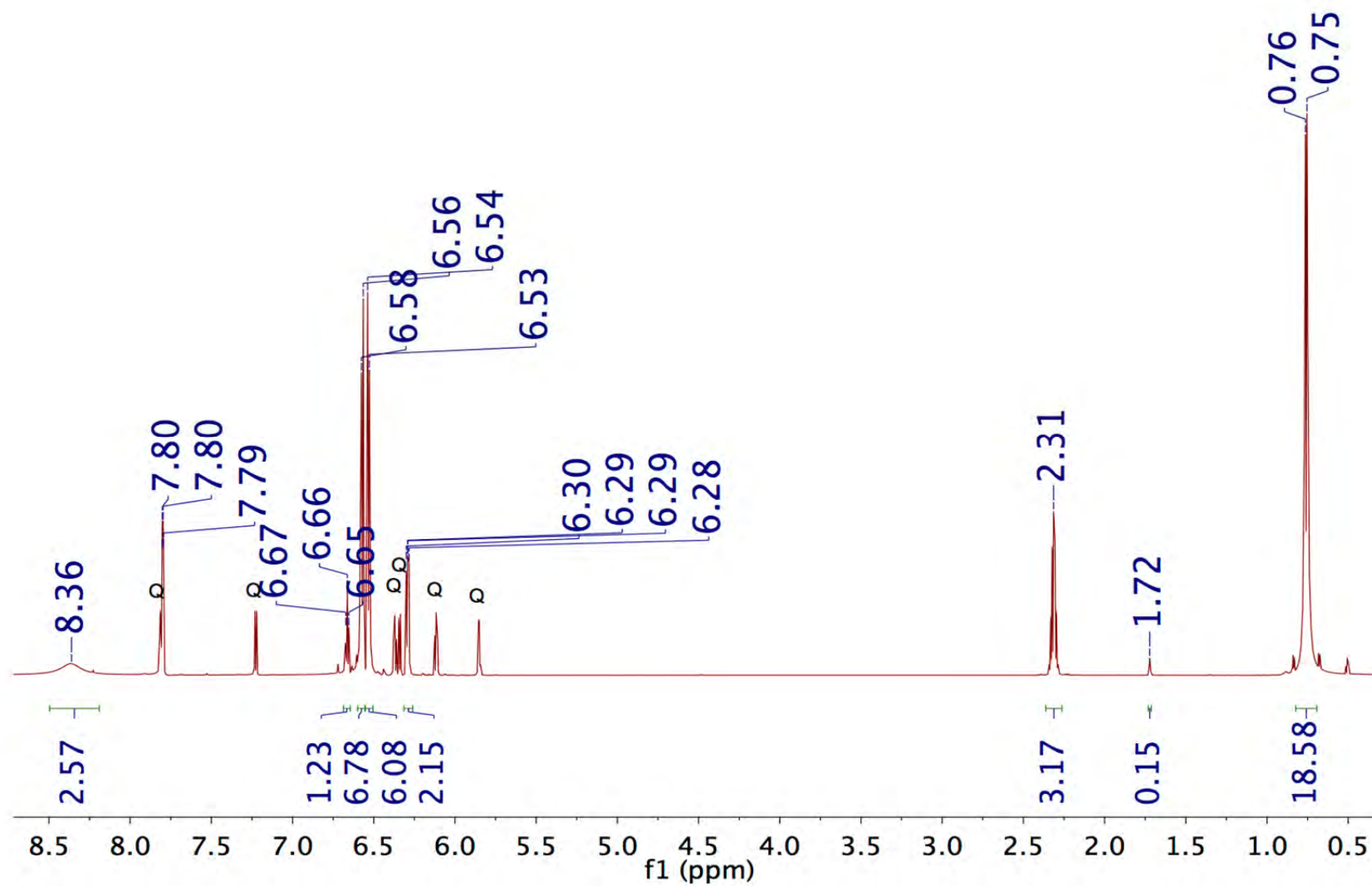


Figure S46: The ¹H NMR-spectrum of pyridine mixed with 3 equivalents of 4-isopropylphenol in toluene-*d*₈. Quinoline peaks are highlighted with “Q”.

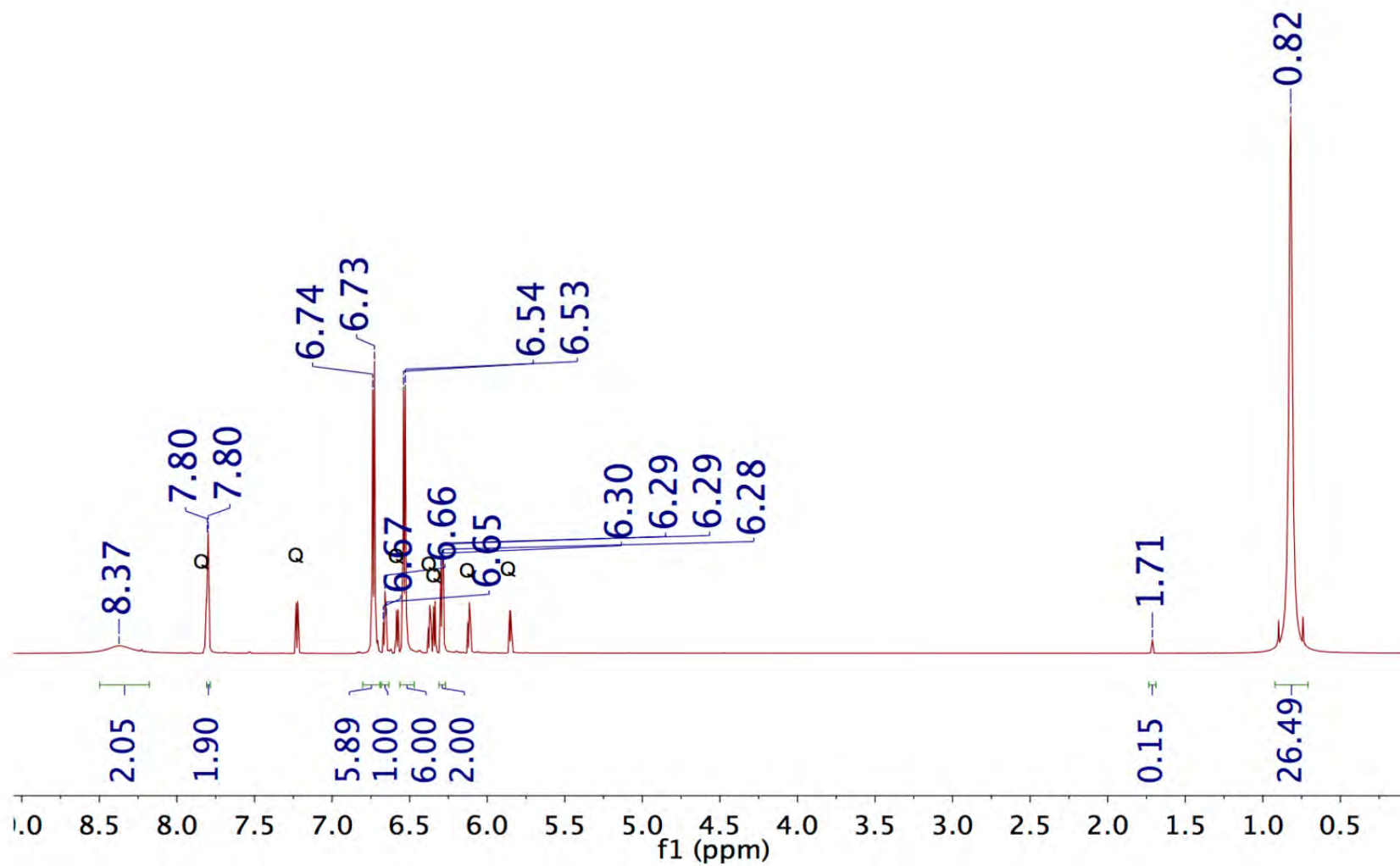


Figure S47: The ^1H NMR-spectrum of pyridine mixed with 3 equivalents of 4-*tert*-butylphenol in toluene- d_8 . Quinoline peaks are highlighted with "Q".

4. Computational data

In the following, equilibrium geometries, thermochemical data, lowest vibration frequencies, data from NBO analysis, and NMR chemical shieldings for the systems investigated in this study are compiled. Energies given in Hartree units, geometries in Å, frequencies in cm^{-1} , and chemical shieldings in ppm.

Geometry optimizations and frequency calculations were performed exploiting molecular symmetry, with the corresponding standard orientation of the molecules. For the NBO analysis, the molecules were rotated into an orientation such that the π orbitals at the pyridine N atom points in z direction. In the following, the geometries are given for the latter orientation. Electronic energies are given for both orientations, the value in parentheses refers to the NBO calculation. Deviations between the electronic energies for different orientations are due to the different orientation of the DFT integration grids.

For some of the compounds, small imaginary lowest frequencies were observed. These imaginary frequencies are related to methyl rotations and could not be alleviated by reorienting the respective methyl group. They are supposed to be an artefact of the SCRF calculation.

Pyridine

```
=====
Pyridine
-----
Energies and NBO populations
-----
```

```
Number of atoms: 11
Full point group          C2V      NOp    4
```

```
N      0.013796  0.000000  0.000000
C      0.712997  1.142358  0.000000
C      2.107606  1.198411  0.000000
C      2.820112  0.000000  0.000000
C      2.107606 -1.198411  0.000000
C      0.712997 -1.142358  0.000000
H      0.127260  2.060220  0.000000
H      2.617005  2.157024  0.000000
H      3.906357  0.000000  0.000000
H      2.617005 -2.157025  0.000000
H      0.127260 -2.060220  0.000000
```

```
Electronic energy:          -248.27556669 ( -248.27555652)
Zero-point correction=      0.088218
Sum of electronic and zero-point Energies= -248.187349
Sum of electronic and thermal Energies=    -248.183045
Sum of electronic and thermal Enthalpies=   -248.182100
Sum of electronic and thermal Free Energies= -248.214115
```

```
Lowest vibration frequency:      379.1106
```

Summary of Natural Population Analysis:

		Natural Population				
Atom	No	Natural Charge	Core	Valence	Rydberg	Total
N	1	-0.47338	1.99946	5.45057	0.02335	7.47338
C	2	0.04782	1.99923	3.92446	0.02849	5.95218
C	3	-0.25026	1.99916	4.23443	0.01667	6.25026
C	4	-0.17004	1.99918	4.15320	0.01766	6.17004
C	5	-0.25026	1.99916	4.23443	0.01667	6.25026
C	6	0.04782	1.99923	3.92446	0.02849	5.95218

H	7	0.19394	0.00000	0.80473	0.00133	0.80606
H	8	0.22127	0.00000	0.77729	0.00144	0.77873
H	9	0.21790	0.00000	0.78078	0.00132	0.78210
H	10	0.22127	0.00000	0.77729	0.00144	0.77873
H	11	0.19394	0.00000	0.80473	0.00133	0.80606
=====						
* Total *		0.00000	11.99543	29.86637	0.13820	42.00000

Contributions to pi population at N atom in pyridine:

Val(2p) 1.17094
Ryd(3p) 0.00074
Ryd(4p) 0.00007
Ryd(5p) 0.00004

Total: 1.17179

NMR properties

Electronic energy (NMR): -248.27556669

1	N	Isotropic =	-99.6517	Anisotropy =	587.1778
2	C	Isotropic =	21.5106	Anisotropy =	181.3627
3	C	Isotropic =	50.9414	Anisotropy =	183.3477
4	C	Isotropic =	37.9582	Anisotropy =	211.9098
5	C	Isotropic =	50.9414	Anisotropy =	183.3477
6	C	Isotropic =	21.5106	Anisotropy =	181.3627
7	H	Isotropic =	22.5326	Anisotropy =	6.4498
8	H	Isotropic =	23.9899	Anisotropy =	5.3831
9	H	Isotropic =	23.5386	Anisotropy =	5.3388
10	H	Isotropic =	23.9899	Anisotropy =	5.3831
11	H	Isotropic =	22.5326	Anisotropy =	6.4498

Substituted halobenzenes

=====

X = I R = H

Energies and NBO populations

Number of atoms: 12
Full point group C2V NOp 4

C	0.041351	0.000000	0.000000
C	0.727635	1.217024	0.000000
C	2.125858	1.208234	0.000000
C	2.826909	0.000000	0.000000
C	2.125858	-1.208234	0.000000
C	0.727635	-1.217024	0.000000
H	0.187002	2.156972	0.000000
H	2.662148	2.152595	0.000000
H	3.912429	0.000000	0.000000
H	2.662148	-2.152595	0.000000
H	0.187002	-2.156972	0.000000
I	-2.085973	0.000000	0.000000

Electronic energy: -7150.87892577 (-7150.87892577)
Zero-point correction= 0.089394
Sum of electronic and zero-point Energies= -7150.789532
Sum of electronic and thermal Energies= -7150.783606
Sum of electronic and thermal Enthalpies= -7150.782662
Sum of electronic and thermal Free Energies= -7150.820668

Lowest vibration frequency: 148.2531

Summary of Natural Population Analysis:

		Natural Population				
Atom	No	Natural Charge	Core	Valence	Rydberg	Total
C	1	-0.20253	1.99874	4.17700	0.02679	6.20253
C	2	-0.23415	1.99899	4.21477	0.02039	6.23415
C	3	-0.19333	1.99917	4.17639	0.01778	6.19333

C	4	-0.20798	1.99918	4.19127	0.01753	6.20798
C	5	-0.19333	1.99917	4.17639	0.01778	6.19333
C	6	-0.23415	1.99899	4.21477	0.02039	6.23415
H	7	0.22819	0.00000	0.77028	0.00154	0.77181
H	8	0.21935	0.00000	0.77928	0.00137	0.78065
H	9	0.21782	0.00000	0.78085	0.00133	0.78218
H	10	0.21935	0.00000	0.77928	0.00137	0.78065
H	11	0.22819	0.00000	0.77028	0.00154	0.77181
I	12	0.15256	45.99798	6.83003	0.01942	52.84744
=====						
* Total *		0.00000	57.99220	35.86057	0.14723	94.00000

X = I R = OCH3

Energies and NBO populations

Number of atoms: 16

Full point group

CS

NOp 2

C	-0.058340	0.019740	0.000000
C	-0.716327	1.247292	0.000000
C	-2.115844	1.293211	0.000000
C	-2.853733	0.102176	0.000000
C	-2.179705	-1.130230	0.000000
C	-0.789530	-1.174747	0.000000
H	-0.156157	2.175723	0.000000
H	-2.605633	2.259610	0.000000
O	-4.211342	0.036049	0.000000
H	-2.762087	-2.045952	0.000000
H	-0.283149	-2.133655	0.000000
I	2.066479	-0.040892	0.000000
C	-4.947613	1.253954	0.000000
H	-5.998807	0.965347	0.000000
H	-4.733112	1.848153	-0.896417
H	-4.733112	1.848153	0.896417

Electronic energy: -7265.40209411 (-7265.40197845)
Zero-point correction= 0.121583
Sum of electronic and zero-point Energies= -7265.280511
Sum of electronic and thermal Energies= -7265.271924
Sum of electronic and thermal Enthalpies= -7265.270979
Sum of electronic and thermal Free Energies= -7265.315926

Lowest vibration frequency: 62.8273

Summary of Natural Population Analysis:

Natural Population						
Atom	No	Natural Charge	Core	Valence	Rydberg	Total
C	1	-0.23347	1.99872	4.20849	0.02626	6.23347
C	2	-0.21268	1.99900	4.19310	0.02058	6.21268
C	3	-0.28344	1.99907	4.26918	0.01519	6.28344
C	4	0.32577	1.99882	3.65229	0.02312	5.67423
C	5	-0.23310	1.99908	4.21679	0.01723	6.23310
C	6	-0.21627	1.99900	4.19713	0.02014	6.21627
H	7	0.22750	0.00000	0.77093	0.00157	0.77250
H	8	0.22590	0.00000	0.77235	0.00175	0.77410
O	9	-0.54100	1.99971	6.52363	0.01766	8.54100
H	10	0.23014	0.00000	0.76812	0.00174	0.76986
H	11	0.22875	0.00000	0.76977	0.00148	0.77125
I	12	0.14634	45.99797	6.83589	0.01980	52.85366
C	13	-0.21427	1.99927	4.19939	0.01561	6.21427
H	14	0.19773	0.00000	0.80127	0.00100	0.80227
H	15	0.17605	0.00000	0.82190	0.00205	0.82395
H	16	0.17605	0.00000	0.82190	0.00205	0.82395
=====						
* Total *		0.00000	61.99065	47.82211	0.18724	110.00000

X = I R = OCH2CH3

Energies and NBO populations

Number of atoms: 19
Full point group

CS NOp 2

C	0.105245	-0.058393	0.000000
C	-0.758593	1.034251	0.000000
C	-2.144396	0.832402	0.000000
C	-2.661970	-0.470162	0.000000
C	-1.780334	-1.564018	0.000000
C	-0.404321	-1.362710	0.000000
H	-0.370236	2.046851	0.000000
H	-2.797647	1.696656	0.000000
O	-3.985579	-0.775657	0.000000
H	-2.191643	-2.568295	0.000000
H	0.262873	-2.217446	0.000000
I	2.209315	0.250582	0.000000
C	-4.940633	0.291966	0.000000
C	-6.327370	-0.328296	0.000000
H	-4.791010	0.917098	-0.890703
H	-4.791010	0.917098	0.890703
H	-7.084351	0.462387	0.000000
H	-6.475055	-0.948751	0.888688
H	-6.475055	-0.948751	-0.888688

Electronic energy:	-7304.71769091 (-7304.71769223)
Zero-point correction=	0.149488
Sum of electronic and zero-point Energies=	-7304.568203
Sum of electronic and thermal Energies=	-7304.558195
Sum of electronic and thermal Enthalpies=	-7304.557250
Sum of electronic and thermal Free Energies=	-7304.605868

Lowest vibration frequency: 37.4409

Summary of Natural Population Analysis:

		Natural Population				
Atom	No	Natural Charge	Core	Valence	Rydberg	Total
C	1	-0.23337	1.99871	4.20850	0.02615	6.23337
C	2	-0.21304	1.99900	4.19359	0.02045	6.21304
C	3	-0.28319	1.99907	4.26867	0.01546	6.28319
C	4	0.32878	1.99883	3.64927	0.02312	5.67122
C	5	-0.23286	1.99908	4.21689	0.01690	6.23286
C	6	-0.21687	1.99900	4.19778	0.02008	6.21687
H	7	0.22738	0.00000	0.77104	0.00158	0.77262
H	8	0.22541	0.00000	0.77279	0.00181	0.77459
O	9	-0.55484	1.99970	6.53445	0.02069	8.55484
H	10	0.22957	0.00000	0.76874	0.00169	0.77043
H	11	0.22839	0.00000	0.77012	0.00149	0.77161
I	12	0.14482	45.99798	6.83746	0.01975	52.85518
C	13	-0.03658	1.99924	4.01887	0.01847	6.03658
C	14	-0.59441	1.99938	4.58646	0.00858	6.59441
H	15	0.17539	0.00000	0.82219	0.00242	0.82461
H	16	0.17539	0.00000	0.82219	0.00242	0.82461
H	17	0.21134	0.00000	0.78751	0.00115	0.78866
H	18	0.20934	0.00000	0.78899	0.00167	0.79066
H	19	0.20934	0.00000	0.78899	0.00167	0.79066
* Total *		0.00000	63.99000	53.80448	0.20553	118.00000

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X = I R = CH3

Energies and NBO populations

Number of atoms: 15
Full point group

CS NOp 2

C	-0.003716	0.000886	0.000000
C	-0.693341	1.213264	0.000000
C	-2.092105	1.202649	0.000000
C	-2.814767	0.003089	0.000000

C	-2.092589	-1.200383	0.000000
C	-0.696800	-1.212883	0.000000
H	-0.157864	2.156174	0.000000
H	-2.623718	2.150703	0.000000
C	-4.326765	-0.003459	0.000000
H	-2.625834	-2.147952	0.000000
H	-0.161585	-2.156023	0.000000
I	2.123076	-0.001613	0.000000
H	-4.727225	1.013722	0.000000
H	-4.717711	-0.521805	-0.882753
H	-4.717711	-0.521805	0.882753

Electronic energy: -7190.19304383 (-7190.19304378)
 Zero-point correction= 0.116324
 Sum of electronic and zero-point Energies= -7190.076720
 Sum of electronic and thermal Energies= -7190.069733
 Sum of electronic and thermal Enthalpies= -7190.068789
 Sum of electronic and thermal Free Energies= -7190.109950

Lowest vibration frequency: -10.0416

Summary of Natural Population Analysis:

Natural Population						
Atom	No	Natural Charge	Core	Valence	Rydberg	Total
C	1	-0.21036	1.99872	4.18534	0.02630	6.21036
C	2	-0.22707	1.99899	4.20729	0.02079	6.22707
C	3	-0.19620	1.99906	4.18061	0.01654	6.19620
C	4	-0.02961	1.99907	4.01516	0.01538	6.02961
C	5	-0.19359	1.99906	4.17791	0.01663	6.19359
C	6	-0.22628	1.99899	4.20662	0.02067	6.22628
H	7	0.22712	0.00000	0.77127	0.00161	0.77288
H	8	0.21573	0.00000	0.78268	0.00160	0.78427
C	9	-0.59774	1.99932	4.58837	0.01005	6.59774
H	10	0.21588	0.00000	0.78257	0.00155	0.78412
H	11	0.22701	0.00000	0.77138	0.00161	0.77299
I	12	0.14725	45.99798	6.83509	0.01968	52.85275
H	13	0.21126	0.00000	0.78747	0.00127	0.78874
H	14	0.21831	0.00000	0.78037	0.00132	0.78169
H	15	0.21831	0.00000	0.78037	0.00132	0.78169
=====						
* Total *		0.00000	59.99118	41.85250	0.15632	102.00000

=====

X = I R = CHCH32

Energies and NBO populations

Number of atoms: 21
 Full point group C1 NOp 1

C	-0.066773	-0.024320	0.042338
C	0.694956	1.143132	0.094302
C	2.088164	1.043170	0.170259
C	2.736047	-0.198685	0.195410
C	1.940685	-1.354189	0.141731
C	0.548992	-1.278425	0.065623
H	0.219383	2.117668	0.076396
H	2.677518	1.955992	0.210553
C	4.255134	-0.285066	0.278508
H	2.405385	-2.336427	0.158997
H	-0.043116	-2.185934	0.025278
I	-2.187891	0.099031	-0.073676
H	4.632281	0.745440	0.307401
C	4.716170	-0.987442	1.572177
C	4.856341	-0.967557	-0.967295
H	5.809817	-0.983494	1.639606
H	4.383537	-2.031583	1.596053
H	4.316062	-0.485879	2.459280
H	4.555991	-0.451772	-1.885109
H	5.950759	-0.963620	-0.913845
H	4.529171	-2.010870	-1.044144

Electronic energy: -7268.81371240 (-7268.81372694)
 Zero-point correction= 0.172900
 Sum of electronic and zero-point Energies= -7268.640812
 Sum of electronic and thermal Energies= -7268.630444
 Sum of electronic and thermal Enthalpies= -7268.629500
 Sum of electronic and thermal Free Energies= -7268.678721

Lowest vibration frequency: 37.3948

Summary of Natural Population Analysis:

		Natural Population				
Atom	No	Natural Charge	Core	Valence	Rydberg	Total
C	1	-0.20862	1.99872	4.18338	0.02652	6.20862
C	2	-0.22741	1.99899	4.20779	0.02063	6.22741
C	3	-0.19341	1.99905	4.17788	0.01649	6.19341
C	4	-0.01576	1.99902	3.99655	0.02019	6.01576
C	5	-0.19588	1.99904	4.18062	0.01622	6.19588
C	6	-0.22700	1.99899	4.20726	0.02075	6.22700
H	7	0.22735	0.00000	0.77107	0.00158	0.77265
H	8	0.21609	0.00000	0.78221	0.00170	0.78391
C	9	-0.24420	1.99920	4.22936	0.01564	6.24420
H	10	0.21617	0.00000	0.78213	0.00170	0.78383
H	11	0.22669	0.00000	0.77172	0.00159	0.77331
I	12	0.14677	45.99798	6.83573	0.01952	52.85323
H	13	0.19961	0.00000	0.79743	0.00295	0.80039
C	14	-0.57024	1.99935	4.56141	0.00948	6.57024
C	15	-0.57025	1.99935	4.56142	0.00948	6.57025
H	16	0.20430	0.00000	0.79452	0.00119	0.79570
H	17	0.19798	0.00000	0.80067	0.00135	0.80202
H	18	0.20778	0.00000	0.79122	0.00100	0.79222
H	19	0.20778	0.00000	0.79121	0.00100	0.79222
H	20	0.20430	0.00000	0.79452	0.00119	0.79570
H	21	0.19797	0.00000	0.80068	0.00135	0.80203
* Total *		0.00000	63.98971	53.81878	0.19152	118.00000

X = I R = H

Energies and NBO populations

Number of atoms: 12
 Full point group C2V NOp 4

C	-0.018117	0.000000	0.000000
C	-0.706870	1.216023	0.000000
C	-2.104209	1.217523	0.000000
C	-2.773121	0.000000	0.000000
C	-2.104209	-1.217523	0.000000
C	-0.706870	-1.216023	0.000000
H	-0.169752	2.157436	0.000000
H	-2.664609	2.146065	0.000000
F	-4.125117	0.000000	0.000000
H	-2.664609	-2.146065	0.000000
H	-0.169752	-2.157436	0.000000
I	2.107236	0.000000	0.000000

Electronic energy: -7250.12236332 (-7250.12236332)
 Zero-point correction= 0.081090
 Sum of electronic and zero-point Energies= -7250.041273
 Sum of electronic and thermal Energies= -7250.034503
 Sum of electronic and thermal Enthalpies= -7250.033559
 Sum of electronic and thermal Free Energies= -7250.073702

Lowest vibration frequency: 99.3474

Summary of Natural Population Analysis:

		Natural Population				
Atom	No	Natural Charge	Core	Valence	Rydberg	Total

C	-0.089556	0.053617	0.011044
C	0.747970	-1.061816	0.029756
C	2.136476	-0.896930	0.048217
C	2.661933	0.393849	0.055099
C	1.830534	1.512830	0.037476
C	0.447325	1.344538	0.013592
H	0.336391	-2.064506	0.026571
H	2.772469	-1.772743	0.053719
O	4.026977	0.697340	0.052793
H	2.270078	2.504268	0.042994
H	-0.198345	2.215135	-0.000012
I	-2.195242	-0.212710	-0.020843
C	4.966030	-0.233952	0.312503
F	6.144175	0.384404	0.346945
F	5.028359	-1.195067	-0.632586
F	4.776899	-0.853849	1.492434

Electronic energy: -7563.16097268 (-7563.16097206)
 Zero-point correction= 0.098069
 Sum of electronic and zero-point Energies= -7563.062904
 Sum of electronic and thermal Energies= -7563.052458
 Sum of electronic and thermal Enthalpies= -7563.051514
 Sum of electronic and thermal Free Energies= -7563.102881

Lowest vibration frequency: 14.4635

Summary of Natural Population Analysis:

		Natural Population				
Atom	No	Natural Charge	Core	Valence	Rydberg	Total
C	1	-0.21726	1.99873	4.19189	0.02664	6.21726
C	2	-0.20795	1.99900	4.18880	0.02015	6.20795
C	3	-0.25135	1.99904	4.23673	0.01557	6.25135
C	4	0.29345	1.99871	3.68542	0.02242	5.70655
C	5	-0.22141	1.99906	4.20526	0.01708	6.22141
C	6	-0.21051	1.99900	4.19165	0.01986	6.21051
H	7	0.23407	0.00000	0.76442	0.00150	0.76593
H	8	0.23959	0.00000	0.75849	0.00192	0.76041
O	9	-0.55041	1.99964	6.53460	0.01617	8.55041
H	10	0.23892	0.00000	0.75947	0.00160	0.76108
H	11	0.23398	0.00000	0.76454	0.00148	0.76602
I	12	0.16924	45.99797	6.81313	0.01967	52.83076
C	13	1.28379	1.99965	2.64626	0.07030	4.71621
F	14	-0.33111	1.99991	7.32489	0.00632	9.33111
F	15	-0.35345	1.99992	7.34670	0.00684	9.35345
F	16	-0.34959	1.99991	7.34292	0.00676	9.34959
* Total *		0.00000	67.99053	65.75518	0.25429	134.00000

X = I R = COOCH3

Energies and NBO populations

Number of atoms: 18
 Full point group CS NOp 2

C	0.042119	0.012006	0.000000
C	-0.617888	1.245227	0.000000
C	-2.011574	1.267811	0.000000
C	-2.743691	0.072699	0.000000
C	-2.066037	-1.154646	0.000000
C	-0.671625	-1.190025	0.000000
H	-0.057674	2.173297	0.000000
H	-2.544094	2.213032	0.000000
C	-4.235796	0.160397	0.000000
H	-2.627496	-2.081835	0.000000
H	-0.152689	-2.141579	0.000000
I	2.165847	-0.038424	0.000000
O	-4.820538	-1.051018	0.000000
C	-6.259561	-1.058482	0.000000
H	-6.543390	-2.110091	0.000000

H	-6.642485	-0.557046	-0.891758
H	-6.642485	-0.557046	0.891758
O	-4.857351	1.201549	0.000000

Electronic energy:	-7378.76322051 (-7378.76312785)
Zero-point correction=	0.131918
Sum of electronic and zero-point Energies=	-7378.631303
Sum of electronic and thermal Energies=	-7378.620809
Sum of electronic and thermal Enthalpies=	-7378.619865
Sum of electronic and thermal Free Energies=	-7378.669884

Lowest vibration frequency: 49.4715

Summary of Natural Population Analysis:

Natural Population						
Atom	No	Natural Charge	Core	Valence	Rydberg	Total
C	1	-0.18406	1.99875	4.15842	0.02690	6.18406
C	2	-0.23371	1.99899	4.21449	0.02023	6.23371
C	3	-0.14559	1.99910	4.12944	0.01705	6.14559
C	4	-0.16080	1.99893	4.14376	0.01811	6.16080
C	5	-0.15119	1.99910	4.13572	0.01637	6.15119
C	6	-0.23448	1.99899	4.21527	0.02022	6.23448
H	7	0.23247	0.00000	0.76600	0.00152	0.76753
H	8	0.23983	0.00000	0.75856	0.00161	0.76017
C	9	0.80370	1.99930	3.15035	0.04665	5.19630
H	10	0.23657	0.00000	0.76189	0.00154	0.76343
H	11	0.23181	0.00000	0.76666	0.00154	0.76819
I	12	0.17309	45.99796	6.80953	0.01941	52.82691
O	13	-0.54966	1.99972	6.53584	0.01410	8.54966
C	14	-0.21740	1.99926	4.20243	0.01570	6.21740
H	15	0.19243	0.00000	0.80650	0.00107	0.80757
H	16	0.19049	0.00000	0.80782	0.00169	0.80951
H	17	0.19049	0.00000	0.80782	0.00169	0.80951
O	18	-0.61399	1.99975	6.60304	0.01120	8.61399
=====						
* Total *		0.00000	65.98985	57.77356	0.23659	124.00000

X = I R = COCH3

Energies and NBO populations

Number of atoms: 17
Full point group CS NOp 2

C	0.000506	0.019377	0.000000
C	-0.649641	1.258950	0.000000
C	-2.041752	1.292533	0.000000
C	-2.793674	0.106421	0.000000
C	-2.118943	-1.124072	0.000000
C	-0.723978	-1.174965	0.000000
H	-0.080694	2.181809	0.000000
H	-2.566330	2.242374	0.000000
C	-4.294961	0.204747	0.000000
H	-2.670542	-2.058447	0.000000
H	-0.214468	-2.131513	0.000000
I	2.122829	-0.051840	0.000000
C	-5.109984	-1.075257	0.000000
O	-4.841699	1.294324	0.000000
H	-6.169715	-0.819096	0.000000
H	-4.881891	-1.681590	0.883636
H	-4.881891	-1.681590	-0.883636

Electronic energy:	-7303.52753299 (-7303.52749341)
Zero-point correction=	0.126348
Sum of electronic and zero-point Energies=	-7303.401185
Sum of electronic and thermal Energies=	-7303.391682
Sum of electronic and thermal Enthalpies=	-7303.390738
Sum of electronic and thermal Free Energies=	-7303.438180

Lowest vibration frequency: 56.3927

Summary of Natural Population Analysis:

		Natural Population				
Atom	No	Natural Charge	Core	Valence	Rydberg	Total
C	1	-0.18327	1.99875	4.15746	0.02706	6.18327
C	2	-0.23198	1.99899	4.21284	0.02015	6.23198
C	3	-0.14185	1.99909	4.12470	0.01806	6.14185
C	4	-0.15529	1.99895	4.13881	0.01753	6.15529
C	5	-0.16076	1.99910	4.14620	0.01546	6.16076
C	6	-0.23475	1.99899	4.21522	0.02054	6.23475
H	7	0.23222	0.00000	0.76626	0.00153	0.76778
H	8	0.23985	0.00000	0.75828	0.00187	0.76015
C	9	0.56607	1.99931	3.39597	0.03865	5.43393
H	10	0.22324	0.00000	0.77530	0.00145	0.77676
H	11	0.23162	0.00000	0.76681	0.00156	0.76838
I	12	0.17532	45.99796	6.80730	0.01942	52.82468
C	13	-0.67871	1.99930	4.67109	0.00832	6.67871
O	14	-0.57344	1.99976	6.56006	0.01362	8.57344
H	15	0.23274	0.00000	0.76542	0.00184	0.76726
H	16	0.22949	0.00000	0.76912	0.00140	0.77051
H	17	0.22949	0.00000	0.76912	0.00140	0.77051
* Total *		0.00000	63.99019	51.79994	0.20986	116.00000

X = I R = CF3

Energies and NBO populations

Number of atoms: 15

Full point group

CS

NOp 2

C	0.005904	0.001189	0.000000
C	-0.685581	1.214047	0.000000
C	-2.082804	1.205217	0.000000
C	-2.773461	-0.007806	0.000000
C	-2.071802	-1.219146	0.000000
C	-0.678468	-1.219063	0.000000
H	-0.151750	2.157383	0.000000
H	-2.624906	2.144155	0.000000
C	-4.281686	-0.043007	0.000000
H	-2.606015	-2.164120	0.000000
H	-0.137261	-2.157898	0.000000
I	2.128800	0.003031	0.000000
F	-4.830279	1.187480	0.000000
F	-4.769009	-0.693223	-1.083113
F	-4.769009	-0.693223	1.083113

Electronic energy:

-7487.94206436 (-7487.94192516)

Zero-point correction=

0.093701

Sum of electronic and zero-point Energies=

-7487.848363

Sum of electronic and thermal Energies=

-7487.839689

Sum of electronic and thermal Enthalpies=

-7487.838744

Sum of electronic and thermal Free Energies=

-7487.884464

Lowest vibration frequency:

-12.3412

Summary of Natural Population Analysis:

		Natural Population				
Atom	No	Natural Charge	Core	Valence	Rydberg	Total
C	1	-0.19023	1.99875	4.16428	0.02720	6.19023
C	2	-0.22455	1.99899	4.20543	0.02013	6.22455
C	3	-0.16671	1.99908	4.15089	0.01673	6.16671
C	4	-0.15263	1.99891	4.13618	0.01753	6.15263
C	5	-0.16287	1.99908	4.14669	0.01710	6.16287
C	6	-0.22376	1.99900	4.20459	0.02017	6.22376
H	7	0.23469	0.00000	0.76381	0.00149	0.76531
H	8	0.23829	0.00000	0.76014	0.00157	0.76171
C	9	1.06719	1.99913	2.87350	0.06017	4.93281
H	10	0.23432	0.00000	0.76412	0.00155	0.76568

H	11	0.23479	0.000000	0.76370	0.00152	0.76521
I	12	0.17964	45.99796	6.80299	0.01940	52.82036
F	13	-0.35426	1.99992	7.34797	0.00637	9.35426
F	14	-0.35696	1.99992	7.35051	0.00653	9.35696
F	15	-0.35696	1.99992	7.35051	0.00653	9.35696
=====						
* Total *		0.00000	65.99066	59.78532	0.22402	126.00000

=====

X = I R = CN

Energies and NBO populations

Number of atoms: 13
Full point group C2V NOp 4

C	-0.035153	0.000000	0.000000
C	-0.723369	1.217803	0.000000
C	-2.116504	1.217079	0.000000
C	-2.818056	0.000000	0.000000
C	-2.116504	-1.217079	0.000000
C	-0.723369	-1.217803	0.000000
H	-0.184810	2.158253	0.000000
H	-2.659682	2.155953	0.000000
C	-4.251598	0.000000	0.000000
H	-2.659682	-2.155953	0.000000
H	-0.184810	-2.158253	0.000000
I	2.083774	0.000000	0.000000
N	-5.410235	0.000000	0.000000

Electronic energy:	-7243.12232060 (-7243.12232060)
Zero-point correction=	0.087947
Sum of electronic and zero-point Energies=	-7243.034374
Sum of electronic and thermal Energies=	-7243.026645
Sum of electronic and thermal Enthalpies=	-7243.025701
Sum of electronic and thermal Free Energies=	-7243.068202

Lowest vibration frequency: 70.5975

Summary of Natural Population Analysis:

		Natural Population				
Atom	No	Natural Charge	Core	Valence	Rydberg	Total
C	1	-0.18811	1.99876	4.16205	0.02731	6.18811
C	2	-0.22564	1.99899	4.20664	0.02001	6.22564
C	3	-0.13786	1.99910	4.12320	0.01556	6.13786
C	4	-0.18695	1.99888	4.17113	0.01694	6.18695
C	5	-0.13786	1.99910	4.12320	0.01556	6.13786
C	6	-0.22564	1.99899	4.20664	0.02001	6.22564
H	7	0.23700	0.00000	0.76152	0.00148	0.76300
H	8	0.23536	0.00000	0.76338	0.00126	0.76464
C	9	0.30521	1.99927	3.66164	0.03388	5.69479
H	10	0.23536	0.00000	0.76338	0.00126	0.76464
H	11	0.23700	0.00000	0.76152	0.00148	0.76300
I	12	0.19112	45.99794	6.79153	0.01941	52.80888
N	13	-0.33900	1.99959	5.31940	0.02001	7.33900
=====						
* Total *		0.00000	61.99061	43.81523	0.19417	106.00000

=====

X = I R = N02

Energies and NBO populations

Number of atoms: 14
Full point group C2V NOp 4

C	0.009124	0.000000	0.000000
C	0.695773	1.219571	0.000000
C	2.089327	1.220376	0.000000
C	2.763878	0.000000	0.000000

C	2.089327	-1.220376	0.000000
C	0.695773	-1.219571	0.000000
H	0.156385	2.159224	0.000000
H	2.648081	2.148282	0.000000
N	4.238219	0.000000	0.000000
H	2.648081	-2.148282	0.000000
H	0.156385	-2.159224	0.000000
I	-2.109328	0.000000	0.000000
O	4.810710	1.084692	0.000000
O	4.810710	-1.084692	0.000000

Electronic energy:	-7355.39158801 (-7355.39158801)
Zero-point correction=	0.091655
Sum of electronic and zero-point Energies=	-7355.299933
Sum of electronic and thermal Energies=	-7355.291422
Sum of electronic and thermal Enthalpies=	-7355.290478
Sum of electronic and thermal Free Energies=	-7355.335520

Lowest vibration frequency: 47.4789

Summary of Natural Population Analysis:

		Natural Population				
Atom	No	Natural Charge	Core	Valence	Rydberg	Total
C	1	-0.18159	1.99876	4.15540	0.02743	6.18159
C	2	-0.22261	1.99900	4.20364	0.01997	6.22261
C	3	-0.17062	1.99906	4.15308	0.01847	6.17062
C	4	0.06520	1.99879	3.91514	0.02087	5.93480
C	5	-0.17062	1.99906	4.15308	0.01847	6.17062
C	6	-0.22261	1.99900	4.20364	0.01997	6.22261
H	7	0.23837	0.00000	0.76018	0.00144	0.76163
H	8	0.25466	0.00000	0.74347	0.00187	0.74534
N	9	0.48981	1.99946	4.45564	0.05510	6.51019
H	10	0.25466	0.00000	0.74347	0.00187	0.74534
H	11	0.23837	0.00000	0.76018	0.00144	0.76163
I	12	0.20045	45.99794	6.78206	0.01956	52.79955
O	13	-0.38674	1.99980	6.37205	0.01489	8.38674
O	14	-0.38674	1.99980	6.37205	0.01489	8.38674
=====						
* Total *		0.00000	63.99066	51.77309	0.23625	116.00000

X = Br R = H

Energies and NBO populations

Number of atoms: 12
Full point group C2V NOp 4

C	0.026165	0.000000	0.015106
C	0.006109	0.000000	1.410053
C	1.221552	0.000000	2.100878
C	2.433001	0.000000	1.404694
C	2.430190	0.000000	0.007456
C	1.224196	0.000000	-0.699736
H	-0.936240	0.000000	1.946289
H	1.214481	0.000000	3.186666
H	3.372919	0.000000	1.947356
H	3.366974	0.000000	-0.541561
H	1.217415	0.000000	-1.783952
Br	-1.633754	0.000000	-0.943249

Electronic energy:	-2805.60842779 (-2805.60846734)
Zero-point correction=	0.089762
Sum of electronic and zero-point Energies=	-2805.518666
Sum of electronic and thermal Energies=	-2805.512896
Sum of electronic and thermal Enthalpies=	-2805.511952
Sum of electronic and thermal Free Energies=	-2805.548950

Lowest vibration frequency: 165.4216

Summary of Natural Population Analysis:

Atom	No	Natural Charge	Natural Population			
			Core	Valence	Rydberg	Total
C	1	-0.10983	1.99865	4.08836	0.02281	6.10983
C	2	-0.23357	1.99897	4.21534	0.01926	6.23357
C	3	-0.19218	1.99917	4.17532	0.01769	6.19218
C	4	-0.21102	1.99918	4.19429	0.01755	6.21102
C	5	-0.19218	1.99917	4.17532	0.01769	6.19218
C	6	-0.23353	1.99897	4.21530	0.01926	6.23353
H	7	0.23021	0.00000	0.76820	0.00159	0.76979
H	8	0.21906	0.00000	0.77954	0.00140	0.78094
H	9	0.21780	0.00000	0.78089	0.00131	0.78220
H	10	0.21907	0.00000	0.77953	0.00140	0.78093
H	11	0.23022	0.00000	0.76819	0.00159	0.76978
Br	12	0.05595	27.99926	6.92540	0.01939	34.94405
=====						
* Total *		0.00000	39.99338	35.86568	0.14094	76.00000

=====

X = Br R = OCH3

Energies and NBO populations

Number of atoms: 16
Full point group CS NOp 2

C	-0.021145	0.000000	-0.055180
C	-0.073976	0.000000	1.334325
C	1.113866	0.000000	2.075922
C	2.349902	0.000000	1.416132
C	2.385946	0.000000	0.011860
C	1.206431	0.000000	-0.724641
H	-1.028145	0.000000	1.849285
H	1.053278	0.000000	3.157552
O	3.558312	0.000000	2.039672
H	3.349305	0.000000	-0.487523
H	1.242124	0.000000	-1.808298
Br	-1.649021	0.000000	-1.067231
C	3.585599	0.000000	3.462562
H	4.640090	0.000000	3.739028
H	3.102705	-0.896540	3.869812
H	3.102705	0.896540	3.869812

Electronic energy:	-2920.13113367 (-2920.13111411)
Zero-point correction=	0.121964
Sum of electronic and zero-point Energies=	-2920.009170
Sum of electronic and thermal Energies=	-2920.000772
Sum of electronic and thermal Enthalpies=	-2919.999828
Sum of electronic and thermal Free Energies=	-2920.043692

Lowest vibration frequency: 67.4077

Summary of Natural Population Analysis:

Atom	No	Natural Charge	Natural Population			
			Core	Valence	Rydberg	Total
C	1	-0.13961	1.99863	4.11858	0.02240	6.13961
C	2	-0.21228	1.99899	4.19406	0.01924	6.21228
C	3	-0.28212	1.99907	4.26794	0.01511	6.28212
C	4	0.32306	1.99882	3.65509	0.02304	5.67694
C	5	-0.23180	1.99908	4.21562	0.01709	6.23180
C	6	-0.21554	1.99899	4.19774	0.01881	6.21554
H	7	0.22950	0.00000	0.76886	0.00164	0.77050
H	8	0.22559	0.00000	0.77262	0.00179	0.77441
O	9	-0.54195	1.99971	6.52451	0.01772	8.54195
H	10	0.22983	0.00000	0.76841	0.00176	0.77017
H	11	0.23062	0.00000	0.76784	0.00154	0.76938
Br	12	0.04968	27.99927	6.93148	0.01956	34.95032
C	13	-0.21422	1.99928	4.19929	0.01565	6.21422
H	14	0.19760	0.00000	0.80140	0.00100	0.80240
H	15	0.17582	0.00000	0.82212	0.00206	0.82418
H	16	0.17582	0.00000	0.82212	0.00206	0.82418

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=====
* Total *      0.00000      43.99184      47.82770      0.18047      92.00000
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X = Br      R = OCH2CH3
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Energies and NBO populations
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Number of atoms: 19
Full point group      CS      NOp      2
```

```
C      -0.045510      0.000000      -0.114531
C      -0.125519      0.000000      1.273649
C      1.047249      0.000000      2.038886
C      2.297252      0.000000      1.404304
C      2.359501      0.000000      0.000383
C      1.194809      0.000000      -0.759732
H      -1.089740      0.000000      1.769619
H      0.964026      0.000000      3.118952
O      3.492367      0.000000      2.051477
H      3.332482      0.000000      -0.480157
H      1.252233      0.000000      -1.842510
Br     -1.653964      0.000000      -1.158330
C      3.499637      0.000000      3.483837
C      4.948791      0.000000      3.939528
H      2.972883      -0.890756      3.852099
H      2.972883      0.890756      3.852099
H      4.990954      0.000000      5.033365
H      5.470763      0.888694      3.572967
H      5.470763      -0.888694      3.572967
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Electronic energy:                               -2959.44673102 ( -2959.44668799)
Zero-point correction=                           0.149913
Sum of electronic and zero-point Energies=       -2959.296818
Sum of electronic and thermal Energies=          -2959.286984
Sum of electronic and thermal Enthalpies=        -2959.286040
Sum of electronic and thermal Free Energies=     -2959.333564
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Lowest vibration frequency:                      43.3331
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Summary of Natural Population Analysis:
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                                     Natural Population
      Atom  No  Natural Charge  ----- Core ----- Valence ----- Rydberg ----- Total -----
-----
      C      1  -0.13974      1.99863      4.11878      0.02232      6.13974
      C      2  -0.21258      1.99899      4.19451      0.01908      6.21258
      C      3  -0.28190      1.99908      4.26746      0.01536      6.28190
      C      4   0.32599      1.99883      3.65216      0.02302      5.67401
      C      5  -0.23158      1.99909      4.21577      0.01673      6.23158
      C      6  -0.21612      1.99899      4.19839      0.01874      6.21612
      H      7   0.22921      0.00000      0.76914      0.00165      0.77079
      H      8   0.22511      0.00000      0.77305      0.00184      0.77489
      O      9  -0.55564      1.99970      6.53516      0.02078      8.55564
      H     10   0.22925      0.00000      0.76904      0.00171      0.77075
      H     11   0.23038      0.00000      0.76809      0.00154      0.76962
      Br     12   0.04838     27.99927      6.93282      0.01953     34.95162
      C     13  -0.03650      1.99924      4.01873      0.01852      6.03650
      C     14  -0.59441      1.99938      4.58645      0.00858      6.59441
      H     15   0.17513      0.00000      0.82244      0.00243      0.82487
      H     16   0.17513      0.00000      0.82244      0.00243      0.82487
      H     17   0.21126      0.00000      0.78759      0.00115      0.78874
      H     18   0.20931      0.00000      0.78902      0.00167      0.79069
      H     19   0.20931      0.00000      0.78902      0.00167      0.79069
=====
* Total *      0.00000      45.99118      53.81005      0.19877     100.00000
=====
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=====
X = Br      R = CH3
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Energies and NBO populations
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Number of atoms: 15
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Full point group CS NOp 2

C	-0.004197	0.000000	-0.001259
C	-0.019803	0.000000	1.391377
C	1.196396	0.000000	2.081792
C	2.422808	0.000000	1.405122
C	2.400371	0.000000	0.001636
C	1.198864	0.000000	-0.708350
H	-0.959566	0.000000	1.932430
H	1.182445	0.000000	3.168541
C	3.734961	0.000000	2.156354
H	3.336339	0.000000	-0.551504
H	1.199725	0.000000	-1.792748
Br	-1.661419	0.000000	-0.963865
H	3.572055	0.000000	3.237353
H	4.333208	-0.882694	1.903727
H	4.333208	0.882694	1.903727

Electronic energy:	-2844.92239987 (-2844.92244554)
Zero-point correction=	0.116777
Sum of electronic and zero-point Energies=	-2844.805623
Sum of electronic and thermal Energies=	-2844.797921
Sum of electronic and thermal Enthalpies=	-2844.796977
Sum of electronic and thermal Free Energies=	-2844.839932

Lowest vibration frequency: 28.1949

Summary of Natural Population Analysis:

		Natural Population				
Atom	No	Natural Charge	Core	Valence	Rydberg	Total
C	1	-0.11762	1.99864	4.09654	0.02244	6.11762
C	2	-0.22655	1.99897	4.20810	0.01948	6.22655
C	3	-0.19503	1.99906	4.17952	0.01645	6.19503
C	4	-0.03275	1.99907	4.01834	0.01534	6.03275
C	5	-0.19240	1.99906	4.17679	0.01655	6.19240
C	6	-0.22556	1.99897	4.20724	0.01935	6.22556
H	7	0.22910	0.00000	0.76922	0.00168	0.77090
H	8	0.21537	0.00000	0.78300	0.00162	0.78463
C	9	-0.59744	1.99932	4.58808	0.01004	6.59744
H	10	0.21556	0.00000	0.78287	0.00157	0.78444
H	11	0.22895	0.00000	0.76937	0.00167	0.77105
Br	12	0.05141	27.99926	6.92981	0.01951	34.94859
H	13	0.21111	0.00000	0.78762	0.00128	0.78889
H	14	0.21792	0.00000	0.78075	0.00133	0.78208
H	15	0.21792	0.00000	0.78075	0.00133	0.78208
=====						
* Total *		0.00000	41.99236	41.85801	0.14963	84.00000

=====

X = Br R = CHCH32

Energies and NBO populations

Number of atoms: 21

Full point group CS NOp 2

C	-0.067713	0.000000	-0.006790
C	-0.064865	0.000000	1.386061
C	1.161346	0.000000	2.058654
C	2.379769	0.000000	1.366803
C	2.337294	0.000000	-0.036688
C	1.125624	0.000000	-0.729789
H	-0.997169	0.000000	1.939963
H	1.162350	0.000000	3.145780
C	3.704132	0.000000	2.120596
H	3.260850	0.000000	-0.609304
H	1.111670	0.000000	-1.814054
Br	-1.738168	0.000000	-0.948665
H	3.461619	0.000000	3.191234
C	4.527208	-1.271668	1.829333
C	4.527208	1.271668	1.829333

H	5.444881	-1.277776	2.428074
H	4.816742	-1.323147	0.773403
H	3.956826	-2.175525	2.066716
H	5.444881	1.277776	2.428074
H	3.956826	2.175525	2.066716
H	4.816742	1.323147	0.773403

Electronic energy:	-2923.54302209 (-2923.54303672)
Zero-point correction=	0.173214
Sum of electronic and zero-point Energies=	-2923.369808
Sum of electronic and thermal Energies=	-2923.359598
Sum of electronic and thermal Enthalpies=	-2923.358654
Sum of electronic and thermal Free Energies=	-2923.406888

Lowest vibration frequency: 38.0837

Summary of Natural Population Analysis:

		Natural Population				
Atom	No	Natural Charge	Core	Valence	Rydberg	Total
C	1	-0.11606	1.99864	4.09484	0.02258	6.11606
C	2	-0.22690	1.99897	4.20853	0.01939	6.22690
C	3	-0.19218	1.99906	4.17673	0.01639	6.19218
C	4	-0.01896	1.99902	3.99983	0.02011	6.01896
C	5	-0.19461	1.99905	4.17943	0.01614	6.19461
C	6	-0.22609	1.99898	4.20763	0.01948	6.22609
H	7	0.22922	0.00000	0.76914	0.00164	0.77078
H	8	0.21567	0.00000	0.78260	0.00172	0.78433
C	9	-0.24380	1.99920	4.22896	0.01563	6.24380
H	10	0.21592	0.00000	0.78236	0.00173	0.78408
H	11	0.22872	0.00000	0.76964	0.00164	0.77128
Br	12	0.05089	27.99927	6.93044	0.01941	34.94911
H	13	0.19934	0.00000	0.79769	0.00298	0.80066
C	14	-0.57026	1.99935	4.56143	0.00948	6.57026
C	15	-0.57026	1.99935	4.56143	0.00948	6.57026
H	16	0.20411	0.00000	0.79470	0.00119	0.79589
H	17	0.19789	0.00000	0.80076	0.00136	0.80211
H	18	0.20769	0.00000	0.79131	0.00100	0.79231
H	19	0.20411	0.00000	0.79470	0.00119	0.79589
H	20	0.20769	0.00000	0.79131	0.00100	0.79231
H	21	0.19789	0.00000	0.80076	0.00136	0.80211
=====						
* Total *		0.00000	45.99089	53.82421	0.18490	100.00000

=====
X = Br R = H

Energies and NBO populations

Number of atoms: 12
Full point group C2V NOp 4

C	0.005801	0.000000	0.003349
C	-0.011256	0.000000	1.398757
C	1.198216	0.000000	2.097969
C	2.386344	0.000000	1.377756
C	2.416003	0.000000	-0.011299
C	1.205731	0.000000	-0.709127
H	-0.951281	0.000000	1.938440
H	1.219449	0.000000	3.182212
F	3.557232	0.000000	2.053769
H	3.365601	0.000000	-0.535032
H	1.203098	0.000000	-1.793054
Br	-1.651928	0.000000	-0.953741

Electronic energy:	-2904.85164392 (-2904.85168888)
Zero-point correction=	0.081468
Sum of electronic and zero-point Energies=	-2904.770176
Sum of electronic and thermal Energies=	-2904.763569
Sum of electronic and thermal Enthalpies=	-2904.762625
Sum of electronic and thermal Free Energies=	-2904.801751

Lowest vibration frequency: 110.7970

Summary of Natural Population Analysis:

		Natural Population				
Atom	No	Natural Charge	Core	Valence	Rydberg	Total
C	1	-0.12711	1.99865	4.10596	0.02250	6.12711
C	2	-0.21243	1.99899	4.19485	0.01859	6.21243
C	3	-0.25601	1.99906	4.24012	0.01683	6.25601
C	4	0.40659	1.99850	3.56845	0.02646	5.59341
C	5	-0.25599	1.99906	4.24010	0.01683	6.25599
C	6	-0.21241	1.99899	4.19482	0.01859	6.21241
H	7	0.23463	0.00000	0.76379	0.00159	0.76537
H	8	0.23765	0.00000	0.76080	0.00156	0.76235
F	9	-0.35166	1.99993	7.34367	0.00806	9.35166
H	10	0.23765	0.00000	0.76080	0.00156	0.76235
H	11	0.23465	0.00000	0.76377	0.00158	0.76535
Br	12	0.06445	27.99926	6.91674	0.01955	34.93555
* Total *		0.00000	41.99243	41.85386	0.15371	84.00000

X = Br R = Cl

Energies and NBO populations

Number of atoms: 12
Full point group C2V NOp 4

C	-0.024354	0.000000	-0.014061
C	-0.040321	0.000000	1.380842
C	1.169481	0.000000	2.079081
C	2.368697	0.000000	1.367568
C	2.385277	0.000000	-0.026740
C	1.175684	0.000000	-0.725340
H	-0.979300	0.000000	1.922529
H	1.174046	0.000000	3.163292
Cl	3.893114	0.000000	2.247690
H	3.326514	0.000000	-0.564892
H	1.175308	0.000000	-1.809363
Br	-1.681137	0.000000	-0.970605

Electronic energy: -3265.17389558 (-3265.17394054)
Zero-point correction= 0.080164
Sum of electronic and zero-point Energies= -3265.093732
Sum of electronic and thermal Energies= -3265.086746
Sum of electronic and thermal Enthalpies= -3265.085802
Sum of electronic and thermal Free Energies= -3265.126233

Lowest vibration frequency: 87.7719

Summary of Natural Population Analysis:

		Natural Population				
Atom	No	Natural Charge	Core	Valence	Rydberg	Total
C	1	-0.11544	1.99865	4.09421	0.02258	6.11544
C	2	-0.21403	1.99897	4.19606	0.01900	6.21403
C	3	-0.21398	1.99898	4.19645	0.01856	6.21398
C	4	-0.04373	1.99861	4.02234	0.02279	6.04373
C	5	-0.21402	1.99898	4.19648	0.01856	6.21402
C	6	-0.21400	1.99897	4.19603	0.01900	6.21400
H	7	0.23587	0.00000	0.76252	0.00161	0.76413
H	8	0.23637	0.00000	0.76198	0.00165	0.76363
Cl	9	0.00171	9.99962	6.97869	0.01997	16.99829
H	10	0.23634	0.00000	0.76201	0.00166	0.76366
H	11	0.23589	0.00000	0.76250	0.00161	0.76411
Br	12	0.06901	27.99926	6.91215	0.01958	34.93099
* Total *		0.00000	49.99203	41.84141	0.16656	92.00000

X = Br R = OCF3

 Energies and NBO populations

Number of atoms: 16

Full point group C1 NOp 1

C	-0.033094	-0.073812	-0.108558
C	-0.112570	-0.064990	1.281903
C	1.055017	0.030240	2.045531
C	2.283793	0.107756	1.392269
C	2.364574	0.098531	0.000218
C	1.199465	0.009162	-0.758549
H	-1.073037	-0.127927	1.780025
H	0.979390	0.044068	3.125097
O	3.521223	0.225984	2.033573
H	3.335986	0.160737	-0.477785
H	1.255492	0.001207	-1.840839
Br	-1.636102	-0.202201	-1.142975
C	3.664457	0.003788	3.355608
F	3.215123	-1.206762	3.735477
F	4.963758	0.076144	3.634723
F	3.025032	0.918077	4.113956

Electronic energy: -3217.89105812 (-3217.89013680)
 Zero-point correction= 0.098316
 Sum of electronic and zero-point Energies= -3217.792742
 Sum of electronic and thermal Energies= -3217.782463
 Sum of electronic and thermal Enthalpies= -3217.781519
 Sum of electronic and thermal Free Energies= -3217.831740

Lowest vibration frequency: 26.4590

Summary of Natural Population Analysis:

		Natural Population				
Atom	No	Natural Charge	Core	Valence	Rydberg	Total
C	1	-0.12310	1.99864	4.10184	0.02262	6.12310
C	2	-0.20727	1.99899	4.18941	0.01888	6.20727
C	3	-0.25000	1.99905	4.23550	0.01545	6.25000
C	4	0.29093	1.99871	3.68807	0.02230	5.70907
C	5	-0.21997	1.99907	4.20398	0.01692	6.21997
C	6	-0.20973	1.99898	4.19216	0.01858	6.20973
H	7	0.23620	0.00000	0.76224	0.00156	0.76380
H	8	0.23940	0.00000	0.75865	0.00195	0.76060
O	9	-0.55080	1.99964	6.53497	0.01619	8.55080
H	10	0.23873	0.00000	0.75965	0.00162	0.76127
H	11	0.23615	0.00000	0.76232	0.00153	0.76385
Br	12	0.07003	27.99926	6.91114	0.01958	34.92997
C	13	1.28368	1.99965	2.64633	0.07035	4.71632
F	14	-0.34955	1.99991	7.34287	0.00677	9.34955
F	15	-0.33125	1.99991	7.32502	0.00632	9.33125
F	16	-0.35343	1.99992	7.34667	0.00685	9.35343
=====						
* Total *		0.00000	49.99172	65.76080	0.24748	116.00000

=====

X = Br R = COOCH3

Energies and NBO populations

Number of atoms: 18

Full point group CS NOp 2

C	-0.047267	0.000000	-0.014245
C	-0.087120	0.000000	1.381823
C	1.113783	0.000000	2.088425
C	2.340653	0.000000	1.410256
C	2.357346	0.000000	0.008015
C	1.162510	0.000000	-0.711011
H	-1.036141	0.000000	1.905532
H	1.110555	0.000000	3.173227

C	3.593895	0.000000	2.224466
H	3.303158	0.000000	-0.521247
H	1.172482	0.000000	-1.794840
Br	-1.687320	0.000000	-0.995302
O	4.701003	0.000000	1.461114
C	5.954513	0.000000	2.168204
H	6.722280	0.000000	1.395597
H	6.038102	-0.891785	2.793564
H	6.038102	0.891785	2.793564
O	3.619673	0.000000	3.436858

Electronic energy:	-3033.49273543 (-3033.49273040)
Zero-point correction=	0.132370
Sum of electronic and zero-point Energies=	-3033.360365
Sum of electronic and thermal Energies=	-3033.350051
Sum of electronic and thermal Enthalpies=	-3033.349107
Sum of electronic and thermal Free Energies=	-3033.398101

Lowest vibration frequency: 51.0117

Summary of Natural Population Analysis:

		Natural Population				
Atom	No	Natural Charge	Core	Valence	Rydberg	Total
C	1	-0.09154	1.99867	4.06995	0.02292	6.09154
C	2	-0.23254	1.99897	4.21453	0.01904	6.23254
C	3	-0.14442	1.99910	4.12839	0.01693	6.14442
C	4	-0.16311	1.99893	4.14621	0.01797	6.16311
C	5	-0.15007	1.99911	4.13472	0.01624	6.15007
C	6	-0.23328	1.99897	4.21528	0.01903	6.23328
H	7	0.23456	0.00000	0.76386	0.00157	0.76544
H	8	0.23963	0.00000	0.75873	0.00164	0.76037
C	9	0.80371	1.99930	3.15028	0.04671	5.19629
H	10	0.23637	0.00000	0.76207	0.00156	0.76363
H	11	0.23391	0.00000	0.76450	0.00159	0.76609
Br	12	0.07529	27.99925	6.90597	0.01949	34.92471
O	13	-0.54962	1.99972	6.53577	0.01413	8.54962
C	14	-0.21749	1.99926	4.20251	0.01571	6.21749
H	15	0.19235	0.00000	0.80658	0.00107	0.80765
H	16	0.19050	0.00000	0.80780	0.00169	0.80950
H	17	0.19050	0.00000	0.80780	0.00169	0.80950
O	18	-0.61479	1.99975	6.60384	0.01120	8.61479
=====						
* Total *		0.00000	47.99104	57.77881	0.23016	106.00000

X = Br R = COCH3

Energies and NBO populations

Number of atoms: 17
Full point group CS NOp 2

C	-0.015158	0.000000	0.011597
C	-0.030856	0.000000	1.409387
C	1.180461	0.000000	2.095029
C	2.402858	0.000000	1.403319
C	2.388432	0.000000	-0.000228
C	1.183041	0.000000	-0.703334
H	-0.971712	0.000000	1.948187
H	1.195251	0.000000	3.180048
C	3.679677	0.000000	2.199199
H	3.315978	0.000000	-0.563090
H	1.176598	0.000000	-1.787313
Br	-1.670167	0.000000	-0.941605
C	5.002803	0.000000	1.456125
O	3.646507	0.000000	3.417805
H	5.815988	0.000000	2.182525
H	5.088327	0.883680	0.814008
H	5.088327	-0.883680	0.814008

Electronic energy:	-2958.25706885 (-2958.25708368)
Zero-point correction=	0.126577

Sum of electronic and zero-point Energies= -2958.130492
 Sum of electronic and thermal Energies= -2958.121102
 Sum of electronic and thermal Enthalpies= -2958.120158
 Sum of electronic and thermal Free Energies= -2958.166790

Lowest vibration frequency: 53.8616

Summary of Natural Population Analysis:

		Natural Population				
Atom	No	Natural Charge	Core	Valence	Rydberg	Total
C	1	-0.09101	1.99867	4.06933	0.02302	6.09101
C	2	-0.23056	1.99897	4.21265	0.01894	6.23056
C	3	-0.14092	1.99909	4.12387	0.01796	6.14092
C	4	-0.15753	1.99895	4.14117	0.01741	6.15753
C	5	-0.15976	1.99910	4.14531	0.01534	6.15976
C	6	-0.23340	1.99898	4.21506	0.01937	6.23340
H	7	0.23425	0.00000	0.76418	0.00157	0.76575
H	8	0.23973	0.00000	0.75837	0.00190	0.76027
C	9	0.56615	1.99931	3.39587	0.03867	5.43385
H	10	0.22312	0.00000	0.77539	0.00149	0.77688
H	11	0.23375	0.00000	0.76464	0.00161	0.76625
Br	12	0.07762	27.99924	6.90361	0.01953	34.92238
C	13	-0.67866	1.99930	4.67103	0.00833	6.67866
O	14	-0.57426	1.99976	6.56085	0.01365	8.57426
H	15	0.23276	0.00000	0.76540	0.00184	0.76724
H	16	0.22936	0.00000	0.76924	0.00140	0.77064
H	17	0.22936	0.00000	0.76924	0.00140	0.77064
* Total *		0.00000	45.99138	51.80519	0.20342	98.00000

X = Br R = CF3

Energies and NBO populations

Number of atoms: 15
 Full point group CS NOp 2

C	-0.012356	0.000000	-0.006527
C	-0.038468	0.000000	1.387399
C	1.169170	0.000000	2.089201
C	2.380523	0.000000	1.395107
C	2.392072	0.000000	-0.005081
C	1.192679	0.000000	-0.713573
H	-0.981439	0.000000	1.921619
H	1.159142	0.000000	3.173241
C	3.696661	0.000000	2.131480
H	3.332594	0.000000	-0.546885
H	1.195495	0.000000	-1.797227
Br	-1.659306	0.000000	-0.973623
F	3.543526	0.000000	3.469847
F	4.447552	-1.083087	1.819677
F	4.447552	1.083087	1.819677

Electronic energy: -3142.67154520 (-3142.67148747)
 Zero-point correction= 0.094075
 Sum of electronic and zero-point Energies= -3142.577470
 Sum of electronic and thermal Energies= -3142.568961
 Sum of electronic and thermal Enthalpies= -3142.568017
 Sum of electronic and thermal Free Energies= -3142.612745

Lowest vibration frequency: -10.7226

Summary of Natural Population Analysis:

		Natural Population				
Atom	No	Natural Charge	Core	Valence	Rydberg	Total
C	1	-0.09714	1.99866	4.07535	0.02312	6.09714
C	2	-0.22346	1.99898	4.20553	0.01896	6.22346
C	3	-0.16558	1.99909	4.14986	0.01663	6.16558

C	4	-0.15510	1.99891	4.13885	0.01734	6.15510
C	5	-0.16167	1.99909	4.14560	0.01698	6.16167
C	6	-0.22247	1.99898	4.20449	0.01901	6.22247
H	7	0.23687	0.00000	0.76159	0.00155	0.76313
H	8	0.23820	0.00000	0.76020	0.00160	0.76180
C	9	1.06721	1.99913	2.87340	0.06026	4.93279
H	10	0.23424	0.00000	0.76418	0.00159	0.76576
H	11	0.23707	0.00000	0.76136	0.00156	0.76293
Br	12	0.08057	27.99925	6.90066	0.01952	34.91943
F	13	-0.35426	1.99992	7.34797	0.00638	9.35426
F	14	-0.35723	1.99992	7.35077	0.00654	9.35723
F	15	-0.35723	1.99992	7.35077	0.00654	9.35723
=====						
* Total *		0.00000	47.99185	59.79058	0.21757	108.00000

=====

X = Br R = CN

Energies and NBO populations

Number of atoms: 13
Full point group C2V NOp 4

C	0.021815	0.000000	0.012595
C	0.003311	0.000000	1.409324
C	1.210186	0.000000	2.104506
C	2.426504	0.000000	1.400943
C	2.427649	0.000000	-0.004201
C	1.222165	0.000000	-0.701795
H	-0.937539	0.000000	1.947335
H	1.211104	0.000000	3.189073
C	3.667845	0.000000	2.117631
H	3.367370	0.000000	-0.545690
H	1.217672	0.000000	-1.785600
Br	-1.630032	0.000000	-0.941099
N	4.671303	0.000000	2.696978

Electronic energy: -2897.85188795 (-2897.85192561)
Zero-point correction= 0.088334
Sum of electronic and zero-point Energies= -2897.763554
Sum of electronic and thermal Energies= -2897.755992
Sum of electronic and thermal Enthalpies= -2897.755048
Sum of electronic and thermal Free Energies= -2897.796529

Lowest vibration frequency: 77.8866

Summary of Natural Population Analysis:

Natural Population						
Atom	No	Natural Charge	Core	Valence	Rydberg	Total

C	1	-0.09323	1.99867	4.07135	0.02320	6.09323
C	2	-0.22457	1.99897	4.20672	0.01888	6.22457
C	3	-0.13659	1.99911	4.12208	0.01541	6.13659
C	4	-0.18925	1.99888	4.17346	0.01691	6.18925
C	5	-0.13657	1.99911	4.12206	0.01541	6.13657
C	6	-0.22456	1.99897	4.20671	0.01887	6.22456
H	7	0.23938	0.00000	0.75910	0.00152	0.76062
H	8	0.23538	0.00000	0.76334	0.00127	0.76462
C	9	0.30526	1.99927	3.66159	0.03388	5.69474
H	10	0.23536	0.00000	0.76336	0.00127	0.76464
H	11	0.23941	0.00000	0.75907	0.00152	0.76059
Br	12	0.08979	27.99924	6.89148	0.01949	34.91021
N	13	-0.33981	1.99959	5.32008	0.02014	7.33981
=====						
* Total *		0.00000	43.99181	43.82040	0.18779	88.00000

=====

X = Br R = N02

Energies and NBO populations

Number of atoms: 14

Full point group C2V NOp 4

C	-0.000015	0.000000	-0.000009
C	-0.020401	0.000000	1.397739
C	1.185803	0.000000	2.094306
C	2.380559	0.000000	1.374416
C	2.406624	0.000000	-0.020217
C	1.200277	0.000000	-0.716537
H	-0.961575	0.000000	1.934715
H	1.206161	0.000000	3.177189
N	3.656829	0.000000	2.111271
H	3.354607	0.000000	-0.544028
H	1.194725	0.000000	-1.800106
Br	-1.649654	0.000000	-0.952428
O	3.610036	0.000000	3.337058
O	4.694995	0.000000	1.457854

Electronic energy:	-3010.12114760 (-3010.12117230)
Zero-point correction=	0.092054
Sum of electronic and zero-point Energies=	-3010.029094
Sum of electronic and thermal Energies=	-3010.020755
Sum of electronic and thermal Enthalpies=	-3010.019811
Sum of electronic and thermal Free Energies=	-3010.063804

Lowest vibration frequency: 49.6593

Summary of Natural Population Analysis:

		Natural Population				
Atom	No	Natural Charge	Core	Valence	Rydberg	Total
C	1	-0.08735	1.99868	4.06541	0.02326	6.08735
C	2	-0.22125	1.99898	4.20341	0.01886	6.22125
C	3	-0.16933	1.99907	4.15192	0.01833	6.16933
C	4	0.06328	1.99879	3.91714	0.02079	5.93672
C	5	-0.16927	1.99907	4.15186	0.01833	6.16927
C	6	-0.22124	1.99898	4.20340	0.01886	6.22124
H	7	0.24071	0.00000	0.75780	0.00148	0.75929
H	8	0.25463	0.00000	0.74346	0.00190	0.74537
N	9	0.48989	1.99945	4.45559	0.05506	6.51011
H	10	0.25465	0.00000	0.74345	0.00190	0.74535
H	11	0.24075	0.00000	0.75776	0.00148	0.75925
Br	12	0.09875	27.99924	6.88235	0.01967	34.90125
O	13	-0.38715	1.99980	6.37247	0.01488	8.38715
O	14	-0.38708	1.99980	6.37240	0.01488	8.38708
=====						
* Total *		0.00000	45.99186	51.77844	0.22970	98.00000

=====
X = Cl R = H

Energies and NBO populations

Number of atoms: 12
Full point group C2V NOp 4

C	-0.043963	0.000000	0.000000
C	-0.723614	1.217682	0.000000
C	-2.121114	1.208564	0.000000
C	-2.822280	0.000000	0.000000
C	-2.121114	-1.208564	0.000000
C	-0.723614	-1.217682	0.000000
H	-0.171493	2.150993	0.000000
H	-2.657561	2.152529	0.000000
H	-3.907571	0.000000	0.000000
H	-2.657561	-2.152529	0.000000
H	-0.171493	-2.150993	0.000000
Cl	1.721378	0.000000	0.000000

Electronic energy:	-691.80292038 (-691.80292038)
Zero-point correction=	0.090370
Sum of electronic and zero-point Energies=	-691.712550
Sum of electronic and thermal Energies=	-691.707015

Sum of electronic and thermal Enthalpies= -691.706071
 Sum of electronic and thermal Free Energies= -691.741711

Lowest vibration frequency: 185.7069

Summary of Natural Population Analysis:

		Natural Population				
Atom	No	Natural Charge	Core	Valence	Rydberg	Total
C	1	-0.04048	1.99861	4.01894	0.02293	6.04048
C	2	-0.23363	1.99898	4.21587	0.01878	6.23363
C	3	-0.19214	1.99918	4.17543	0.01753	6.19214
C	4	-0.21322	1.99918	4.19647	0.01758	6.21322
C	5	-0.19214	1.99918	4.17543	0.01753	6.19214
C	6	-0.23363	1.99898	4.21587	0.01878	6.23363
H	7	0.23039	0.00000	0.76799	0.00162	0.76961
H	8	0.21880	0.00000	0.77981	0.00139	0.78120
H	9	0.21762	0.00000	0.78107	0.00131	0.78238
H	10	0.21880	0.00000	0.77981	0.00139	0.78120
H	11	0.23039	0.00000	0.76799	0.00162	0.76961
Cl	12	-0.01075	9.99963	6.99138	0.01975	17.01075
* Total *		0.00000	21.99372	35.86606	0.14022	58.00000

X = Cl R = OCH3

Energies and NBO populations

Number of atoms: 16
 Full point group CS NOp 2

C	-0.009832	0.000000	-0.051017
C	-0.064438	0.000000	1.337727
C	1.122586	0.000000	2.080124
C	2.358627	0.000000	1.420587
C	2.395899	0.000000	0.016550
C	1.216808	0.000000	-0.720388
H	-1.021577	0.000000	1.847479
H	1.062451	0.000000	3.161773
O	3.567429	0.000000	2.045398
H	3.359490	0.000000	-0.482146
H	1.248161	0.000000	-1.804342
Cl	-1.508824	0.000000	-0.984413
C	3.591416	0.000000	3.468291
H	4.645093	0.000000	3.747938
H	3.107249	-0.896364	3.874763
H	3.107249	0.896364	3.874763

Electronic energy: -806.32538231 (-806.32538409)
 Zero-point correction= 0.122602
 Sum of electronic and zero-point Energies= -806.202780
 Sum of electronic and thermal Energies= -806.194650
 Sum of electronic and thermal Enthalpies= -806.193706
 Sum of electronic and thermal Free Energies= -806.236138

Lowest vibration frequency: 76.8919

Summary of Natural Population Analysis:

		Natural Population				
Atom	No	Natural Charge	Core	Valence	Rydberg	Total
C	1	-0.06972	1.99858	4.04864	0.02251	6.06972
C	2	-0.21244	1.99900	4.19473	0.01871	6.21244
C	3	-0.28189	1.99908	4.26783	0.01498	6.28189
C	4	0.32078	1.99882	3.65745	0.02295	5.67922
C	5	-0.23164	1.99909	4.21559	0.01696	6.23164
C	6	-0.21550	1.99900	4.19827	0.01823	6.21550
H	7	0.22969	0.00000	0.76862	0.00168	0.77031
H	8	0.22542	0.00000	0.77280	0.00177	0.77458
O	9	-0.54262	1.99971	6.52520	0.01770	8.54262

H	10	0.22962	0.000000	0.76863	0.00175	0.77038
H	11	0.23082	0.000000	0.76760	0.00158	0.76918
Cl	12	-0.01692	9.99963	6.99739	0.01989	17.01692
C	13	-0.21423	1.99928	4.19927	0.01569	6.21423
H	14	0.19748	0.000000	0.80152	0.00100	0.80252
H	15	0.17557	0.000000	0.82236	0.00207	0.82443
H	16	0.17557	0.000000	0.82236	0.00207	0.82443
=====						
* Total *		0.00000	25.99218	47.82825	0.17957	74.00000

=====

X = Cl R = OCH2CH3

Energies and NBO populations

Number of atoms: 19

Full point group

CS

NOp 2

C	-0.034666	0.000000	-0.113462
C	-0.118900	0.000000	1.273738
C	1.051903	0.000000	2.041550
C	2.302866	0.000000	1.409338
C	2.368906	0.000000	0.005978
C	1.205934	0.000000	-0.756242
H	-1.086722	0.000000	1.762942
H	0.968103	0.000000	3.121439
O	3.497785	0.000000	2.058307
H	3.342823	0.000000	-0.472285
H	1.260528	0.000000	-1.839264
Cl	-1.513823	0.000000	-1.078923
C	3.504983	0.000000	3.489647
C	4.954494	0.000000	3.944523
H	2.978505	-0.890537	3.859154
H	2.978505	0.890537	3.859154
H	4.997779	0.000000	5.038308
H	5.476333	0.888586	3.577485
H	5.476333	-0.888586	3.577485

Electronic energy:

-845.64099789 (-845.64096100)

Zero-point correction=

0.150629

Sum of electronic and zero-point Energies=

-845.490369

Sum of electronic and thermal Energies=

-845.480839

Sum of electronic and thermal Enthalpies=

-845.479895

Sum of electronic and thermal Free Energies=

-845.525847

Lowest vibration frequency:

51.6906

Summary of Natural Population Analysis:

Natural Population						
Atom	No	Natural Charge	Core	Valence	Rydberg	Total

C	1	-0.07000	1.99858	4.04899	0.02244	6.07000
C	2	-0.21272	1.99900	4.19517	0.01856	6.21272
C	3	-0.28159	1.99908	4.26726	0.01525	6.28159
C	4	0.32378	1.99883	3.65447	0.02293	5.67622
C	5	-0.23135	1.99909	4.21565	0.01661	6.23135
C	6	-0.21605	1.99900	4.19888	0.01817	6.21605
H	7	0.22939	0.00000	0.76892	0.00169	0.77061
H	8	0.22487	0.00000	0.77330	0.00183	0.77513
O	9	-0.55630	1.99970	6.53580	0.02079	8.55630
H	10	0.22904	0.00000	0.76926	0.00170	0.77096
H	11	0.23050	0.00000	0.76791	0.00158	0.76950
Cl	12	-0.01800	9.99963	6.99850	0.01987	17.01800
C	13	-0.03636	1.99924	4.01856	0.01855	6.03636
C	14	-0.59430	1.99938	4.58634	0.00859	6.59430
H	15	0.17480	0.00000	0.82276	0.00244	0.82520
H	16	0.17480	0.00000	0.82276	0.00244	0.82520
H	17	0.21109	0.00000	0.78776	0.00115	0.78891
H	18	0.20920	0.00000	0.78912	0.00168	0.79080
H	19	0.20920	0.00000	0.78912	0.00168	0.79080
=====						
* Total *		0.00000	27.99153	53.81053	0.19794	82.00000

```
=====
X = Cl    R = CH3
-----
```

Energies and NBO populations -----

```
Number of atoms: 15
Full point group          CS      NOp   2
```

```
C      0.005429  0.000000  0.004297
C     -0.010451  0.000000  1.396127
C      1.205487  0.000000  2.086858
C      2.431694  0.000000  1.410297
C      2.409357  0.000000  0.006816
C      1.207924  0.000000 -0.702722
H     -0.952627  0.000000  1.933329
H      1.190941  0.000000  3.173465
C      3.744535  0.000000  2.161958
H      3.345141  0.000000 -0.546383
H      1.203956  0.000000 -1.787305
Cl     -1.521139  0.000000 -0.882715
H      3.580571  0.000000  3.242666
H      4.342192 -0.882888  1.908822
H      4.342192  0.882888  1.908822
```

```
Electronic energy: -731.11682767 ( -731.11684150)
Zero-point correction= 0.117404
Sum of electronic and zero-point Energies= -730.999424
Sum of electronic and thermal Energies= -730.991974
Sum of electronic and thermal Enthalpies= -730.991030
Sum of electronic and thermal Free Energies= -731.032574
```

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Lowest vibration frequency: 29.6021
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Summary of Natural Population Analysis:

		Natural Population				
Atom	No	Natural Charge	Core	Valence	Rydberg	Total
C	1	-0.04820	1.99859	4.02705	0.02256	6.04820
C	2	-0.22662	1.99898	4.20870	0.01893	6.22662
C	3	-0.19508	1.99907	4.17971	0.01630	6.19508
C	4	-0.03499	1.99907	4.02062	0.01530	6.03499
C	5	-0.19244	1.99907	4.17696	0.01641	6.19244
C	6	-0.22549	1.99898	4.20771	0.01880	6.22549
H	7	0.22921	0.00000	0.76908	0.00171	0.77079
H	8	0.21511	0.00000	0.78328	0.00161	0.78489
C	9	-0.59701	1.99932	4.58769	0.01000	6.59701
H	10	0.21531	0.00000	0.78313	0.00156	0.78469
H	11	0.22907	0.00000	0.76922	0.00171	0.77093
Cl	12	-0.01493	9.99963	6.99548	0.01982	17.01493
H	13	0.21093	0.00000	0.78779	0.00127	0.78907
H	14	0.21757	0.00000	0.78110	0.00133	0.78243
H	15	0.21757	0.00000	0.78110	0.00133	0.78243
* Total *		0.00000	23.99270	41.85863	0.14866	66.00000

```
=====
X = Cl    R = CHCH32
-----
```

Energies and NBO populations -----

```
Number of atoms: 21
Full point group          C1      NOp   1
```

```
C     -0.036137 -0.041962 -0.054017
C     -0.061056  0.003771  1.337235
C      1.151251  0.026851  2.033219
C      2.382974  0.005050  1.365562
C      2.367982 -0.041094 -0.037819
C      1.170164 -0.064759 -0.753596
H     -1.006486  0.021128  1.868307
H      1.131386  0.062585  3.119491
```

C	3.692176	0.030831	2.145287
H	3.302683	-0.059143	-0.591631
H	1.172519	-0.100365	-1.837548
Cl	-1.557214	-0.071564	-0.949864
H	3.428031	0.065743	3.210237
C	4.521169	-1.248909	1.912633
C	4.520737	1.292988	1.829307
H	5.426325	-1.235130	2.530051
H	4.832391	-1.334381	0.865094
H	3.945981	-2.144980	2.167336
H	3.945161	2.203570	2.024760
H	5.425830	1.320020	2.446373
H	4.831993	1.309749	0.778436

Electronic energy:	-809.73744491 (	-809.73745098)
Zero-point correction=	0.173837	
Sum of electronic and zero-point Energies=	-809.563608	
Sum of electronic and thermal Energies=	-809.553653	
Sum of electronic and thermal Enthalpies=	-809.552709	
Sum of electronic and thermal Free Energies=	-809.599625	

Lowest vibration frequency: 36.1584

Summary of Natural Population Analysis:

			Natural Population			
Atom	No	Natural Charge	Core	Valence	Rydberg	Total
C	1	-0.04718	1.99859	4.02595	0.02264	6.04718
C	2	-0.22696	1.99898	4.20907	0.01891	6.22696
C	3	-0.19210	1.99906	4.17679	0.01624	6.19210
C	4	-0.02135	1.99902	4.00226	0.02007	6.02135
C	5	-0.19449	1.99906	4.17944	0.01599	6.19449
C	6	-0.22600	1.99899	4.20803	0.01899	6.22600
H	7	0.22931	0.00000	0.76902	0.00167	0.77069
H	8	0.21541	0.00000	0.78287	0.00172	0.78459
C	9	-0.24355	1.99920	4.22873	0.01562	6.24355
H	10	0.21564	0.00000	0.78265	0.00172	0.78436
H	11	0.22885	0.00000	0.76947	0.00167	0.77115
Cl	12	-0.01496	9.99963	6.99552	0.01982	17.01496
H	13	0.19921	0.00000	0.79780	0.00298	0.80079
C	14	-0.57031	1.99935	4.56148	0.00948	6.57031
C	15	-0.57029	1.99935	4.56145	0.00948	6.57029
H	16	0.20398	0.00000	0.79482	0.00119	0.79602
H	17	0.19780	0.00000	0.80084	0.00136	0.80220
H	18	0.20760	0.00000	0.79139	0.00101	0.79240
H	19	0.20761	0.00000	0.79138	0.00101	0.79239
H	20	0.20399	0.00000	0.79482	0.00119	0.79601
H	21	0.19777	0.00000	0.80087	0.00136	0.80223
=====						
* Total *		0.00000	27.99123	53.82464	0.18413	82.00000

X = Cl R = H

Energies and NBO populations

Number of atoms:	12
Full point group	C2V NOp 4

C	0.017519	0.000000	0.010115
C	0.000206	0.000000	1.404975
C	1.209099	0.000000	2.104098
C	2.397635	0.000000	1.384275
C	2.426752	0.000000	-0.004938
C	1.216847	0.000000	-0.702309
H	-0.942140	0.000000	1.940761
H	1.230301	0.000000	3.188273
F	3.568753	0.000000	2.060420
H	3.376276	0.000000	-0.528665
H	1.209679	0.000000	-1.786298
Cl	-1.508109	0.000000	-0.870707

Electronic energy:	-791.04604457 (	-791.04605484)
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Zero-point correction= 0.082103
Sum of electronic and zero-point Energies= -790.963942
Sum of electronic and thermal Energies= -790.957585
Sum of electronic and thermal Enthalpies= -790.956641
Sum of electronic and thermal Free Energies= -790.994402

Lowest vibration frequency: 125.8834

Summary of Natural Population Analysis:

		Natural Population				
Atom	No	Natural Charge	Core	Valence	Rydberg	Total
C	1	-0.05734	1.99860	4.03612	0.02261	6.05734
C	2	-0.21236	1.99900	4.19535	0.01802	6.21236
C	3	-0.25591	1.99906	4.24015	0.01670	6.25591
C	4	0.40466	1.99850	3.57046	0.02638	5.59534
C	5	-0.25590	1.99906	4.24013	0.01670	6.25590
C	6	-0.21233	1.99900	4.19532	0.01802	6.21233
H	7	0.23482	0.00000	0.76355	0.00163	0.76518
H	8	0.23737	0.00000	0.76108	0.00155	0.76263
F	9	-0.35201	1.99993	7.34404	0.00805	9.35201
H	10	0.23737	0.00000	0.76108	0.00155	0.76263
H	11	0.23485	0.00000	0.76352	0.00163	0.76515
Cl	12	-0.00321	9.99963	6.98367	0.01991	17.00321
* Total *		0.00000	23.99278	41.85446	0.15276	66.00000

X = Cl R = Cl

Energies and NBO populations

Number of atoms: 12
Full point group D2H NOp 8

C	0.016193	0.000000	0.009349
C	0.000075	0.000000	1.403449
C	1.209411	0.000000	2.101660
C	2.408678	0.000000	1.390651
C	2.424796	0.000000	-0.003449
C	1.215460	0.000000	-0.701660
H	-0.941140	0.000000	1.941754
H	1.213833	0.000000	3.185928
Cl	3.932334	0.000000	2.270334
H	3.366011	0.000000	-0.541754
H	1.211038	0.000000	-1.785928
Cl	-1.507463	0.000000	-0.870334

Electronic energy: -1151.36836738 (-1151.36837741)
Zero-point correction= 0.080840
Sum of electronic and zero-point Energies= -1151.287527
Sum of electronic and thermal Energies= -1151.280803
Sum of electronic and thermal Enthalpies= -1151.279859
Sum of electronic and thermal Free Energies= -1151.318266

Lowest vibration frequency: 99.8754

Summary of Natural Population Analysis:

		Natural Population				
Atom	No	Natural Charge	Core	Valence	Rydberg	Total
C	1	-0.04612	1.99860	4.02477	0.02275	6.04612
C	2	-0.21378	1.99898	4.19637	0.01843	6.21378
C	3	-0.21375	1.99898	4.19634	0.01843	6.21375
C	4	-0.04612	1.99860	4.02477	0.02275	6.04612
C	5	-0.21378	1.99898	4.19637	0.01843	6.21378
C	6	-0.21375	1.99898	4.19634	0.01843	6.21375
H	7	0.23606	0.00000	0.76229	0.00165	0.76394
H	8	0.23609	0.00000	0.76226	0.00165	0.76391
Cl	9	0.00149	9.99962	6.97887	0.02002	16.99851
H	10	0.23606	0.00000	0.76229	0.00165	0.76394

H	11	0.23609	0.00000	0.76226	0.00165	0.76391
Cl	12	0.00149	9.99962	6.97887	0.02002	16.99851
=====						
* Total *		0.00000	31.99238	41.84181	0.16581	74.00000

=====

X = Cl R = OCF3

Energies and NBO populations

Number of atoms: 16

Full point group C1 NOp 1

C	-0.002621	-0.148220	-0.065928
C	-0.226654	0.260379	1.248989
C	0.866126	0.547787	2.068429
C	2.152396	0.411128	1.554185
C	2.378989	0.011456	0.240666
C	1.287633	-0.273530	-0.581403
H	-1.237651	0.358032	1.627434
H	0.719795	0.876283	3.091146
O	3.256542	0.769046	2.350018
H	3.393139	-0.070987	-0.134159
H	1.439617	-0.585996	-1.608090
Cl	-1.379317	-0.505457	-1.100886
C	3.761469	-0.168606	3.180333
F	4.174355	-1.273175	2.530513
F	4.808576	0.366145	3.806691
F	2.875920	-0.574285	4.111735

Electronic energy: -1104.08557295 (-1104.08555830)

Zero-point correction= 0.098956

Sum of electronic and zero-point Energies= -1103.986617

Sum of electronic and thermal Energies= -1103.976604

Sum of electronic and thermal Enthalpies= -1103.975660

Sum of electronic and thermal Free Energies= -1104.024480

Lowest vibration frequency: 27.3433

Summary of Natural Population Analysis:

		Natural Population				
Atom	No	Natural Charge	Core	Valence	Rydberg	Total

C	1	-0.04125	1.99861	4.01987	0.02277	6.04125
C	2	-0.21523	1.99899	4.19805	0.01819	6.21523
C	3	-0.21417	1.99904	4.19838	0.01675	6.21417
C	4	0.26322	1.99869	3.71487	0.02323	5.73678
C	5	-0.21174	1.99904	4.19574	0.01696	6.21174
C	6	-0.21534	1.99899	4.19819	0.01817	6.21534
H	7	0.23709	0.00000	0.76128	0.00163	0.76291
H	8	0.23945	0.00000	0.75886	0.00169	0.76055
O	9	-0.56341	1.99965	6.54659	0.01717	8.56341
H	10	0.23880	0.00000	0.75953	0.00166	0.76120
H	11	0.23707	0.00000	0.76130	0.00163	0.76293
Cl	12	0.00556	9.99962	6.97484	0.01998	16.99444
C	13	1.28126	1.99965	2.64811	0.07099	4.71874
F	14	-0.35225	1.99991	7.34570	0.00663	9.35225
F	15	-0.33497	1.99991	7.32871	0.00635	9.33497
F	16	-0.35409	1.99991	7.34756	0.00662	9.35409
=====						
* Total *		0.00000	31.99201	65.75757	0.25042	98.00000

=====

X = Cl R = COOCH3

Energies and NBO populations

Number of atoms: 18

Full point group CS NOp 2

C	-0.039050	0.000000	-0.009127
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C	-0.078630	0.000000	1.386324
C	1.122050	0.000000	2.092146
C	2.348604	0.000000	1.413218
C	2.364279	0.000000	0.010715
C	1.169310	0.000000	-0.707259
H	-1.029741	0.000000	1.906310
H	1.119929	0.000000	3.176865
C	3.601397	0.000000	2.227445
H	3.309701	0.000000	-0.519124
H	1.173601	0.000000	-1.791264
Cl	-1.549217	0.000000	-0.909991
O	4.709269	0.000000	1.465087
C	5.961590	0.000000	2.173139
H	6.730556	0.000000	1.401737
H	6.044775	-0.891609	2.798894
H	6.044775	0.891609	2.798894
O	3.626818	0.000000	3.439990

Electronic energy:	-919.68734883 (	-919.68735257)
Zero-point correction=	0.132933	
Sum of electronic and zero-point Energies=	-919.554416	
Sum of electronic and thermal Energies=	-919.544331	
Sum of electronic and thermal Enthalpies=	-919.543387	
Sum of electronic and thermal Free Energies=	-919.591177	

Lowest vibration frequency: 49.2247

Summary of Natural Population Analysis:

Atom	No	Natural Charge	Natural Population			
			Core	Valence	Rydberg	Total
C	1	-0.02131	1.99862	3.99961	0.02308	6.02131
C	2	-0.23225	1.99898	4.21472	0.01855	6.23225
C	3	-0.14417	1.99911	4.12826	0.01680	6.14417
C	4	-0.16519	1.99893	4.14834	0.01791	6.16519
C	5	-0.14983	1.99911	4.13461	0.01612	6.14983
C	6	-0.23301	1.99898	4.21549	0.01854	6.23301
H	7	0.23482	0.00000	0.76357	0.00161	0.76518
H	8	0.23951	0.00000	0.75886	0.00164	0.76049
C	9	0.80371	1.99930	3.15028	0.04671	5.19629
H	10	0.23615	0.00000	0.76230	0.00155	0.76385
H	11	0.23423	0.00000	0.76414	0.00162	0.76577
Cl	12	0.00683	9.99961	6.97362	0.01994	16.99317
O	13	-0.54967	1.99972	6.53582	0.01414	8.54967
C	14	-0.21753	1.99926	4.20252	0.01575	6.21753
H	15	0.19227	0.00000	0.80666	0.00107	0.80773
H	16	0.19042	0.00000	0.80789	0.00169	0.80958
H	17	0.19042	0.00000	0.80789	0.00169	0.80958
O	18	-0.61538	1.99975	6.60442	0.01121	8.61538
=====						
* Total *		0.00000	29.99138	57.77901	0.22961	88.00000

=====
X = Cl R = COCH3

Energies and NBO populations

Number of atoms: 17
Full point group CS NOp 2

C	-0.006401	0.000000	0.018023
C	-0.022356	0.000000	1.415217
C	1.188737	0.000000	2.099921
C	2.410732	0.000000	1.406673
C	2.395530	0.000000	0.003296
C	1.189714	0.000000	-0.698767
H	-0.965472	0.000000	1.950135
H	1.205214	0.000000	3.184738
C	3.686978	0.000000	2.202184
H	3.322813	0.000000	-0.559575
H	1.177787	0.000000	-1.782788
Cl	-1.530833	0.000000	-0.857236
C	5.011528	0.000000	1.459421

O	3.653925	0.000000	3.420773
H	5.823480	0.000000	2.187235
H	5.097393	0.884250	0.818210
H	5.097393	-0.884250	0.818210

Electronic energy:	-844.45170183 (	-844.45169824)
Zero-point correction=	0.127149	
Sum of electronic and zero-point Energies=	-844.324553	
Sum of electronic and thermal Energies=	-844.315389	
Sum of electronic and thermal Enthalpies=	-844.314445	
Sum of electronic and thermal Free Energies=	-844.359858	

Lowest vibration frequency: 50.6968

Summary of Natural Population Analysis:

		Natural Population				
Atom	No	Natural Charge	Core	Valence	Rydberg	Total
C	1	-0.02033	1.99862	3.99854	0.02317	6.02033
C	2	-0.23028	1.99898	4.21287	0.01843	6.23028
C	3	-0.14052	1.99910	4.12359	0.01783	6.14052
C	4	-0.15975	1.99895	4.14345	0.01735	6.15975
C	5	-0.15929	1.99911	4.14498	0.01521	6.15929
C	6	-0.23322	1.99898	4.21533	0.01890	6.23322
H	7	0.23458	0.00000	0.76381	0.00161	0.76542
H	8	0.23962	0.00000	0.75848	0.00190	0.76038
C	9	0.56619	1.99931	3.39582	0.03868	5.43381
H	10	0.22297	0.00000	0.77555	0.00148	0.77703
H	11	0.23415	0.00000	0.76420	0.00165	0.76585
Cl	12	0.00842	9.99961	6.97201	0.01996	16.99158
C	13	-0.67846	1.99930	4.67085	0.00832	6.67846
O	14	-0.57501	1.99976	6.56157	0.01367	8.57501
H	15	0.23263	0.00000	0.76552	0.00185	0.76737
H	16	0.22914	0.00000	0.76946	0.00140	0.77086
H	17	0.22914	0.00000	0.76946	0.00140	0.77086
* Total *		0.00000	27.99173	51.80548	0.20279	80.00000

X = Cl R = CF3

Energies and NBO populations

Number of atoms: 15
Full point group CS NOp 2

C	-0.001878	0.000000	-0.000952
C	-0.029438	0.000000	1.392249
C	1.177294	0.000000	2.094578
C	2.388806	0.000000	1.400616
C	2.400897	0.000000	0.000483
C	1.202296	0.000000	-0.708311
H	-0.975184	0.000000	1.921561
H	1.166733	0.000000	3.178534
C	3.705379	0.000000	2.136025
H	3.341979	0.000000	-0.540003
H	1.200398	0.000000	-1.792130
Cl	-1.517307	0.000000	-0.890808
F	3.552827	0.000000	3.474794
F	4.456201	-1.082838	1.823849
F	4.456201	1.082838	1.823849

Electronic energy:	-1028.86611429 (	-1028.86608880)
Zero-point correction=	0.094740	
Sum of electronic and zero-point Energies=	-1028.771374	
Sum of electronic and thermal Energies=	-1028.763127	
Sum of electronic and thermal Enthalpies=	-1028.762183	
Sum of electronic and thermal Free Energies=	-1028.805566	

Lowest vibration frequency: -19.4527

Summary of Natural Population Analysis:

Atom	No	Natural Charge	Natural Population			
			Core	Valence	Rydberg	Total
C	1	-0.02660	1.99862	4.00473	0.02325	6.02660
C	2	-0.22327	1.99898	4.20581	0.01848	6.22327
C	3	-0.16541	1.99909	4.14979	0.01653	6.16541
C	4	-0.15705	1.99891	4.14086	0.01728	6.15705
C	5	-0.16153	1.99910	4.14556	0.01687	6.16153
C	6	-0.22213	1.99899	4.20460	0.01854	6.22213
H	7	0.23719	0.00000	0.76123	0.00158	0.76281
H	8	0.23808	0.00000	0.76033	0.00160	0.76192
C	9	1.06722	1.99913	2.87336	0.06028	4.93278
H	10	0.23404	0.00000	0.76437	0.00159	0.76596
H	11	0.23748	0.00000	0.76092	0.00160	0.76252
Cl	12	0.01109	9.99962	6.96935	0.01995	16.98891
F	13	-0.35452	1.99992	7.34823	0.00637	9.35452
F	14	-0.35730	1.99992	7.35084	0.00654	9.35730
F	15	-0.35730	1.99992	7.35084	0.00654	9.35730
=====						
* Total *		0.00000	29.99219	59.79082	0.21699	90.00000

=====

X = Cl R = CN

Energies and NBO populations

Number of atoms: 13

Full point group C2V NOp 4

C	0.033602	0.000000	0.019400
C	0.013558	0.000000	1.415520
C	1.219937	0.000000	2.110422
C	2.436206	0.000000	1.406544
C	2.437647	0.000000	0.001285
C	1.232656	0.000000	-0.696019
H	-0.929812	0.000000	1.949204
H	1.221767	0.000000	3.194902
C	3.677551	0.000000	2.123235
H	3.377750	0.000000	-0.539370
H	1.223154	0.000000	-1.779843
Cl	-1.485739	0.000000	-0.857792
N	4.680886	0.000000	2.702511

Electronic energy:	-784.04652810 (	-784.04653071)
Zero-point correction=	0.088996	
Sum of electronic and zero-point Energies=	-783.957532	
Sum of electronic and thermal Energies=	-783.950226	
Sum of electronic and thermal Enthalpies=	-783.949282	
Sum of electronic and thermal Free Energies=	-783.989401	

Lowest vibration frequency: 88.0705

Summary of Natural Population Analysis:

Atom	No	Natural Charge	Natural Population			
			Core	Valence	Rydberg	Total
C	1	-0.02209	1.99863	4.00010	0.02336	6.02209
C	2	-0.22427	1.99898	4.20687	0.01842	6.22427
C	3	-0.13636	1.99911	4.12195	0.01529	6.13636
C	4	-0.19100	1.99888	4.17530	0.01682	6.19100
C	5	-0.13633	1.99911	4.12193	0.01529	6.13633
C	6	-0.22425	1.99898	4.20685	0.01842	6.22425
H	7	0.23984	0.00000	0.75861	0.00156	0.76016
H	8	0.23529	0.00000	0.76345	0.00127	0.76471
C	9	0.30533	1.99927	3.66151	0.03390	5.69467
H	10	0.23527	0.00000	0.76347	0.00127	0.76473
H	11	0.23988	0.00000	0.75856	0.00156	0.76012
Cl	12	0.01930	9.99961	6.96110	0.02000	16.98070
N	13	-0.34059	1.99959	5.32080	0.02020	7.34059
=====						
* Total *		0.00000	25.99216	43.82048	0.18736	70.00000

```
=====
X = Cl      R = NO2
-----
```

Energies and NBO populations -----

```
Number of atoms: 14
Full point group          C2V      NOp    4
```

```
C      0.010495  0.000000  0.006059
C     -0.010524  0.000000  1.403377
C      1.195016  0.000000  2.099604
C      2.390304  0.000000  1.380042
C      2.415818  0.000000 -0.014888
C      1.210098  0.000000 -0.710803
H     -0.954045  0.000000  1.936393
H      1.215257  0.000000  3.182413
N      3.665594  0.000000  2.116332
H      3.363679  0.000000 -0.538763
H      1.199943  0.000000 -1.794424
Cl     -1.506067  0.000000 -0.869528
O      3.619163  0.000000  3.342242
O      4.704048  0.000000  1.463166
```

```
Electronic energy: -896.31581271 ( -896.31580214)
Zero-point correction= 0.092727
Sum of electronic and zero-point Energies= -896.223086
Sum of electronic and thermal Energies= -896.215012
Sum of electronic and thermal Enthalpies= -896.214068
Sum of electronic and thermal Free Energies= -896.256697
```

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Lowest vibration frequency: 49.2593
```

Summary of Natural Population Analysis:

		Natural Population				
Atom	No	Natural Charge	Core	Valence	Rydberg	Total
C	1	-0.01596	1.99864	3.99385	0.02347	6.01596
C	2	-0.22087	1.99899	4.20350	0.01838	6.22087
C	3	-0.16877	1.99908	4.15148	0.01821	6.16877
C	4	0.06151	1.99879	3.91900	0.02070	5.93849
C	5	-0.16871	1.99908	4.15142	0.01821	6.16871
C	6	-0.22085	1.99899	4.20348	0.01837	6.22085
H	7	0.24114	0.00000	0.75734	0.00152	0.75886
H	8	0.25449	0.00000	0.74361	0.00190	0.74551
N	9	0.48985	1.99945	4.45566	0.05503	6.51015
H	10	0.25450	0.00000	0.74360	0.00190	0.74550
H	11	0.24119	0.00000	0.75729	0.00152	0.75881
Cl	12	0.02762	9.99961	6.95261	0.02016	16.97238
O	13	-0.38761	1.99980	6.37293	0.01488	8.38761
O	14	-0.38753	1.99980	6.37286	0.01488	8.38753
=====						
* Total *		0.00000	27.99222	51.77863	0.22915	80.00000

Substituted halobenzenes complexed with pyridine

```
=====
X = I      R = H
-----
```

Energies and NBO populations -----

```
Number of atoms: 23
Full point group          C2V      NOp    4
```

```
C      0.000000 -0.562419  0.000000
C      0.000000 -1.257290  1.214244
C      0.000000 -2.655555  1.207955
C      0.000000 -3.357632  0.000000
C      0.000000 -2.655555 -1.207955
C      0.000000 -1.257290 -1.214244
H      0.000000 -0.717817  2.155519
```

H	0.000000	-3.192575	2.152060
H	0.000000	-4.443259	0.000000
H	0.000000	-3.192575	-2.152060
H	0.000000	-0.717817	-2.155519
I	0.000000	1.571499	0.000000
N	0.000000	4.682508	0.000000
C	-1.146506	5.375047	0.000000
C	-1.200285	6.770506	0.000000
C	0.000000	7.482636	0.000000
C	1.200285	6.770506	0.000000
C	1.146506	5.375047	0.000000
H	-2.060542	4.785097	0.000000
H	-2.158500	7.279550	0.000000
H	0.000000	8.568443	0.000000
H	2.158500	7.279550	0.000000
H	2.060542	4.785097	0.000000

Electronic energy: -7399.15920688 (-7399.15920688)
 Zero-point correction= 0.178216
 Sum of electronic and zero-point Energies= -7398.980991
 Sum of electronic and thermal Energies= -7398.968439
 Sum of electronic and thermal Enthalpies= -7398.967495
 Sum of electronic and thermal Free Energies= -7399.026576

Lowest vibration frequency: 10.0685

Counterpoise: BSSE energy = 0.000678761722

Summary of Natural Population Analysis:

		Natural Population				
Atom	No	Natural Charge	Core	Valence	Rydberg	Total
C	1	-0.22464	1.99876	4.19852	0.02736	6.22464
C	2	-0.23569	1.99900	4.21616	0.02054	6.23569
C	3	-0.19788	1.99917	4.18081	0.01791	6.19788
C	4	-0.21348	1.99918	4.19662	0.01768	6.21348
C	5	-0.19788	1.99917	4.18081	0.01791	6.19788
C	6	-0.23569	1.99900	4.21616	0.02054	6.23569
H	7	0.22365	0.00000	0.77479	0.00157	0.77635
H	8	0.21613	0.00000	0.78247	0.00140	0.78387
H	9	0.21502	0.00000	0.78362	0.00136	0.78498
H	10	0.21613	0.00000	0.78247	0.00140	0.78387
H	11	0.22365	0.00000	0.77479	0.00157	0.77635
I	12	0.18056	45.99809	6.79853	0.02282	52.81944
N	13	-0.48138	1.99942	5.45954	0.02242	7.48138
C	14	0.05479	1.99923	3.91942	0.02657	5.94521
C	15	-0.24634	1.99917	4.23087	0.01630	6.24634
C	16	-0.16663	1.99918	4.15010	0.01735	6.16663
C	17	-0.24634	1.99917	4.23087	0.01630	6.24634
C	18	0.05479	1.99923	3.91942	0.02657	5.94521
H	19	0.19826	0.00000	0.80026	0.00148	0.80174
H	20	0.22277	0.00000	0.77574	0.00149	0.77723
H	21	0.21921	0.00000	0.77952	0.00128	0.78079
H	22	0.22277	0.00000	0.77574	0.00149	0.77723
H	23	0.19826	0.00000	0.80026	0.00148	0.80174
* Total *		0.00000	69.98774	65.72749	0.28477	136.00000

Contributions to pi population at N atom in pyridine:

Val(2p)	1.19091
Ryd(3p)	0.00080
Ryd(4p)	0.00038
Ryd(5p)	0.00002

Total: 1.19211

 NMR properties

Electronic energy (NMR): -7399.15920688

1	C	Isotropic =	35.2211	Anisotropy =	142.1813
2	C	Isotropic =	36.1974	Anisotropy =	164.8949
3	C	Isotropic =	44.3749	Anisotropy =	185.5380
4	C	Isotropic =	46.9259	Anisotropy =	192.7416

5	C	Isotropic =	44.3749	Anisotropy =	185.5380
6	C	Isotropic =	36.1974	Anisotropy =	164.8949
7	H	Isotropic =	23.6405	Anisotropy =	11.9992
8	H	Isotropic =	23.9781	Anisotropy =	5.1795
9	H	Isotropic =	24.0016	Anisotropy =	4.0698
10	H	Isotropic =	23.9781	Anisotropy =	5.1795
11	H	Isotropic =	23.6405	Anisotropy =	11.9992
12	I	Isotropic =	3194.3713	Anisotropy =	3144.1720
13	N	Isotropic =	-95.7897	Anisotropy =	564.2350
14	C	Isotropic =	22.8215	Anisotropy =	180.9776
15	C	Isotropic =	50.3616	Anisotropy =	183.7978
16	C	Isotropic =	37.0154	Anisotropy =	212.7685
17	C	Isotropic =	50.3616	Anisotropy =	183.7978
18	C	Isotropic =	22.8215	Anisotropy =	180.9776
19	H	Isotropic =	22.5871	Anisotropy =	10.6447
20	H	Isotropic =	23.9449	Anisotropy =	5.3530
21	H	Isotropic =	23.5092	Anisotropy =	4.8547
22	H	Isotropic =	23.9449	Anisotropy =	5.3530
23	H	Isotropic =	22.5871	Anisotropy =	10.6447

=====

X = I R = OCH3

Energies and NBO populations

Number of atoms: 27
Full point group CS NOp 2

C	0.000000	-0.507423	0.034652
C	0.000000	-1.249893	1.214800
C	0.000000	-2.650139	1.173800
C	0.000000	-3.311152	-0.061016
C	0.000000	-2.560242	-1.247361
C	0.000000	-1.169829	-1.200957
H	0.000000	-0.751063	2.178321
H	0.000000	-3.201949	2.106510
O	0.000000	-4.664367	-0.212911
H	0.000000	-3.083496	-2.198325
H	0.000000	-0.605004	-2.127286
I	0.000000	1.622910	0.104425
C	0.000000	-5.472710	0.957598
H	0.000000	-6.504929	0.606683
H	-0.896000	-5.294464	1.564418
H	0.896000	-5.294464	1.564418
N	0.000000	4.748081	0.176765
C	1.146436	5.440648	0.176765
C	1.200348	6.836179	0.176784
C	0.000000	7.548333	0.176683
C	-1.200348	6.836179	0.176784
C	-1.146436	5.440648	0.176765
H	2.060496	4.850650	0.176267
H	2.158638	7.345124	0.176498
H	0.000000	8.634167	0.176327
H	-2.158638	7.345124	0.176498
H	-2.060496	4.850650	0.176267

Electronic energy:	-7513.68201593 (-7513.68201594)
Zero-point correction=	0.210550
Sum of electronic and zero-point Energies=	-7513.471466
Sum of electronic and thermal Energies=	-7513.456307
Sum of electronic and thermal Enthalpies=	-7513.455362
Sum of electronic and thermal Free Energies=	-7513.520539

Lowest vibration frequency: 12.7347

Counterpoise: BSSE energy = 0.000662309277

Summary of Natural Population Analysis:

		Natural Population				
Atom	No	Natural Charge	Core	Valence	Rydberg	Total
C	1	-0.25516	1.99874	4.22952	0.02690	6.25516
C	2	-0.21401	1.99901	4.19428	0.02072	6.21401

C	3	-0.28726	1.99907	4.27294	0.01526	6.28726
C	4	0.32103	1.99881	3.65707	0.02308	5.67897
C	5	-0.23753	1.99908	4.22113	0.01732	6.23753
C	6	-0.21690	1.99901	4.19767	0.02022	6.21690
H	7	0.22328	0.00000	0.77512	0.00160	0.77672
H	8	0.22319	0.00000	0.77503	0.00179	0.77681
O	9	-0.54396	1.99971	6.52648	0.01777	8.54396
H	10	0.22726	0.00000	0.77098	0.00177	0.77274
H	11	0.22442	0.00000	0.77407	0.00151	0.77558
I	12	0.17458	45.99808	6.80430	0.02304	52.82542
C	13	-0.21367	1.99928	4.19871	0.01568	6.21367
H	14	0.19622	0.00000	0.80276	0.00101	0.80378
H	15	0.17499	0.00000	0.82291	0.00210	0.82501
H	16	0.17499	0.00000	0.82291	0.00210	0.82501
N	17	-0.48053	1.99942	5.45860	0.02251	7.48053
C	18	0.05449	1.99923	3.91962	0.02666	5.94551
C	19	-0.24664	1.99917	4.23116	0.01631	6.24664
C	20	-0.16706	1.99918	4.15052	0.01736	6.16706
C	21	-0.24664	1.99917	4.23116	0.01631	6.24664
C	22	0.05449	1.99923	3.91962	0.02666	5.94551
H	23	0.19816	0.00000	0.80036	0.00149	0.80184
H	24	0.22256	0.00000	0.77595	0.00149	0.77744
H	25	0.21901	0.00000	0.77972	0.00128	0.78099
H	26	0.22256	0.00000	0.77595	0.00149	0.77744
H	27	0.19816	0.00000	0.80036	0.00149	0.80184
=====						
* Total *		0.00000	73.98617	77.68892	0.32491	152.00000

Contributions to pi population at N atom in pyridine:

Val(2p) 1.18960
Ryd(3p) 0.00080
Ryd(4p) 0.00037
Ryd(5p) 0.00002

Total: 1.19079

----- NMR properties -----

Electronic energy (NMR): -7513.68201593

1	C	Isotropic =	49.6286	Anisotropy =	121.3590
2	C	Isotropic =	35.6841	Anisotropy =	162.8120
3	C	Isotropic =	65.2034	Anisotropy =	159.4900
4	C	Isotropic =	11.5499	Anisotropy =	148.0149
5	C	Isotropic =	55.9970	Anisotropy =	148.8774
6	C	Isotropic =	35.4107	Anisotropy =	161.8443
7	H	Isotropic =	23.7675	Anisotropy =	11.9335
8	H	Isotropic =	24.5772	Anisotropy =	7.8242
9	O	Isotropic =	222.8598	Anisotropy =	84.4183
10	H	Isotropic =	24.4112	Anisotropy =	5.9833
11	H	Isotropic =	23.8426	Anisotropy =	11.5354
12	I	Isotropic =	3259.7586	Anisotropy =	3105.4221
13	C	Isotropic =	124.6639	Anisotropy =	75.4529
14	H	Isotropic =	27.5012	Anisotropy =	8.5082
15	H	Isotropic =	27.9001	Anisotropy =	8.0508
16	H	Isotropic =	27.9001	Anisotropy =	8.0508
17	N	Isotropic =	-96.3123	Anisotropy =	565.9891
18	C	Isotropic =	22.7399	Anisotropy =	180.8905
19	C	Isotropic =	50.4278	Anisotropy =	183.6736
20	C	Isotropic =	37.0851	Anisotropy =	212.5243
21	C	Isotropic =	50.4278	Anisotropy =	183.6736
22	C	Isotropic =	22.7399	Anisotropy =	180.8905
23	H	Isotropic =	22.5921	Anisotropy =	10.7105
24	H	Isotropic =	23.9566	Anisotropy =	5.3717
25	H	Isotropic =	23.5212	Anisotropy =	4.8402
26	H	Isotropic =	23.9566	Anisotropy =	5.3717
27	H	Isotropic =	22.5921	Anisotropy =	10.7105

=====

X = I R = OCH2CH3

----- Energies and NBO populations -----

Number of atoms: 30
Full point group CS NOp 2

C	0.000000	-0.305755	-0.054597
C	0.000000	-1.042804	1.128943
C	0.000000	-2.443236	1.094838
C	0.000000	-3.111533	-0.136813
C	0.000000	-2.364765	-1.326313
C	0.000000	-0.974151	-1.286851
H	0.000000	-0.539440	2.090145
H	0.000000	-2.989903	2.030447
O	0.000000	-4.464373	-0.283558
H	0.000000	-2.892173	-2.275024
H	0.000000	-0.413965	-2.216019
I	0.000000	1.824792	0.004116
C	0.000000	-5.282803	0.890386
C	0.000000	-6.734669	0.442095
H	-0.890258	-5.059231	1.493601
H	0.890258	-5.059231	1.493601
H	0.000000	-7.391386	1.317932
H	0.889067	-6.955323	-0.156428
H	-0.889067	-6.955323	-0.156428
N	0.000000	4.953759	0.055085
C	1.146385	5.646241	0.055085
C	1.200485	7.041772	0.055208
C	0.000000	7.754075	0.055199
C	-1.200485	7.041772	0.055208
C	-1.146385	5.646241	0.055085
H	2.060383	5.056132	0.054471
H	2.158853	7.550582	0.054933
H	0.000000	8.839915	0.054959
H	-2.158853	7.550582	0.054933
H	-2.060383	5.056132	0.054471

Electronic energy:	-7552.99755110 (-7552.99755118)
Zero-point correction=	0.238492
Sum of electronic and zero-point Energies=	-7552.759059
Sum of electronic and thermal Energies=	-7552.742474
Sum of electronic and thermal Enthalpies=	-7552.741530
Sum of electronic and thermal Free Energies=	-7552.810152

Lowest vibration frequency: 12.5858

Counterpoise: BSSE energy = 0.000660112364

Summary of Natural Population Analysis:

Atom	No	Natural Charge	Natural Population			Total
			Core	Valence	Rydberg	
C	1	-0.25553	1.99874	4.23001	0.02678	6.25553
C	2	-0.21423	1.99901	4.19464	0.02058	6.21423
C	3	-0.28691	1.99907	4.27232	0.01552	6.28691
C	4	0.32405	1.99882	3.65405	0.02308	5.67595
C	5	-0.23721	1.99908	4.22116	0.01697	6.23721
C	6	-0.21737	1.99901	4.19819	0.02017	6.21737
H	7	0.22300	0.00000	0.77539	0.00161	0.77700
H	8	0.22264	0.00000	0.77552	0.00184	0.77736
O	9	-0.55751	1.99970	6.53702	0.02078	8.55751
H	10	0.22669	0.00000	0.77160	0.00172	0.77331
H	11	0.22415	0.00000	0.77434	0.00152	0.77585
I	12	0.17359	45.99808	6.80532	0.02302	52.82641
C	13	-0.03608	1.99924	4.01825	0.01859	6.03608
C	14	-0.59419	1.99938	4.58625	0.00856	6.59419
H	15	0.17436	0.00000	0.82318	0.00246	0.82564
H	16	0.17436	0.00000	0.82318	0.00246	0.82564
H	17	0.21036	0.00000	0.78848	0.00116	0.78964
H	18	0.20880	0.00000	0.78952	0.00169	0.79120
H	19	0.20880	0.00000	0.78952	0.00169	0.79120
N	20	-0.48033	1.99942	5.45839	0.02252	7.48033
C	21	0.05443	1.99923	3.91968	0.02667	5.94557
C	22	-0.24672	1.99917	4.23123	0.01631	6.24672
C	23	-0.16711	1.99918	4.15058	0.01735	6.16711
C	24	-0.24672	1.99917	4.23123	0.01631	6.24672
C	25	0.05443	1.99923	3.91968	0.02667	5.94557
H	26	0.19812	0.00000	0.80039	0.00149	0.80188
H	27	0.22252	0.00000	0.77599	0.00149	0.77748

H	28	0.21897	0.00000	0.77976	0.00128	0.78103
H	29	0.22252	0.00000	0.77599	0.00149	0.77748
H	30	0.19812	0.00000	0.80039	0.00149	0.80188
=====						
* Total *		0.00000	75.98552	83.67122	0.34326	160.00000

Contributions to pi population at N atom in pyridine:

Val(2p)	1.18932
Ryd(3p)	0.00080
Ryd(4p)	0.00037
Ryd(5p)	0.00002

Total: 1.19051

----- NMR properties -----

Electronic energy (NMR): -7552.99755110

1	C	Isotropic =	49.9653	Anisotropy =	120.4776
2	C	Isotropic =	35.7204	Anisotropy =	162.3232
3	C	Isotropic =	64.6013	Anisotropy =	157.4116
4	C	Isotropic =	11.9341	Anisotropy =	147.3212
5	C	Isotropic =	55.9670	Anisotropy =	147.8607
6	C	Isotropic =	35.4161	Anisotropy =	161.5224
7	H	Isotropic =	23.7987	Anisotropy =	11.9557
8	H	Isotropic =	24.6185	Anisotropy =	8.4346
9	O	Isotropic =	192.8679	Anisotropy =	69.7989
10	H	Isotropic =	24.3987	Anisotropy =	6.4715
11	H	Isotropic =	23.8436	Anisotropy =	11.5412
12	I	Isotropic =	3262.6606	Anisotropy =	3102.9789
13	C	Isotropic =	113.6376	Anisotropy =	63.1547
14	C	Isotropic =	165.9159	Anisotropy =	25.7178
15	H	Isotropic =	27.6511	Anisotropy =	5.3576
16	H	Isotropic =	27.6511	Anisotropy =	5.3576
17	H	Isotropic =	30.2647	Anisotropy =	10.1316
18	H	Isotropic =	30.0311	Anisotropy =	8.2201
19	H	Isotropic =	30.0311	Anisotropy =	8.2201
20	N	Isotropic =	-96.3388	Anisotropy =	566.3019
21	C	Isotropic =	22.7350	Anisotropy =	180.8412
22	C	Isotropic =	50.4245	Anisotropy =	183.6693
23	C	Isotropic =	37.0499	Anisotropy =	212.4983
24	C	Isotropic =	50.4245	Anisotropy =	183.6693
25	C	Isotropic =	22.7350	Anisotropy =	180.8412
26	H	Isotropic =	22.5918	Anisotropy =	10.7381
27	H	Isotropic =	23.9572	Anisotropy =	5.3804
28	H	Isotropic =	23.5217	Anisotropy =	4.8329
29	H	Isotropic =	23.9572	Anisotropy =	5.3804
30	H	Isotropic =	22.5918	Anisotropy =	10.7381

=====

X = I R = CH3

Energies and NBO populations -----

Number of atoms: 26
Full point group CS NOp 2

C	0.000000	-0.469356	-0.008260
C	0.000000	-1.147834	1.212332
C	0.000000	-2.546711	1.226595
C	0.000000	-3.289516	0.039371
C	0.000000	-2.586006	-1.175073
C	0.000000	-1.190391	-1.207559
H	0.000000	-0.598449	2.148022
H	0.000000	-3.063694	2.183012
C	0.000000	-4.801864	0.057229
H	0.000000	-3.134913	-2.113962
H	0.000000	-0.671770	-2.160663
I	0.000000	1.662931	-0.043976
H	0.000000	-5.185615	1.081043
H	-0.882455	-5.202453	-0.454276
H	0.882455	-5.202453	-0.454276
N	0.000000	4.788135	-0.069128
C	1.146420	5.480648	-0.069128

C	1.200372	6.876171	-0.069121
C	0.000000	7.588360	-0.069087
C	-1.200372	6.876171	-0.069121
C	-1.146420	5.480648	-0.069128
H	2.060498	4.890666	-0.068793
H	2.158666	7.385096	-0.068959
H	0.000000	8.674186	-0.068926
H	-2.158666	7.385096	-0.068959
H	-2.060498	4.890666	-0.068793

Electronic energy:	-7438.47308046 (-7438.47308046)
Zero-point correction=	0.205160
Sum of electronic and zero-point Energies=	-7438.267920
Sum of electronic and thermal Energies=	-7438.254314
Sum of electronic and thermal Enthalpies=	-7438.253370
Sum of electronic and thermal Free Energies=	-7438.315201

Lowest vibration frequency: -18.7009

Counterpoise: BSSE energy = 0.000655306574

Summary of Natural Population Analysis:

		Natural Population				
Atom	No	Natural Charge	Core	Valence	Rydberg	Total
C	1	-0.23238	1.99874	4.20673	0.02691	6.23238
C	2	-0.22855	1.99900	4.20862	0.02093	6.22855
C	3	-0.20026	1.99906	4.18456	0.01664	6.20026
C	4	-0.03500	1.99907	4.02049	0.01544	6.03500
C	5	-0.19784	1.99906	4.18206	0.01672	6.19784
C	6	-0.22738	1.99900	4.20758	0.02080	6.22738
H	7	0.22274	0.00000	0.77561	0.00165	0.77726
H	8	0.21268	0.00000	0.78569	0.00163	0.78732
C	9	-0.59692	1.99932	4.58760	0.01000	6.59692
H	10	0.21281	0.00000	0.78561	0.00158	0.78719
H	11	0.22263	0.00000	0.77573	0.00164	0.77737
I	12	0.17570	45.99808	6.80319	0.02303	52.82430
H	13	0.20999	0.00000	0.78872	0.00129	0.79001
H	14	0.21650	0.00000	0.78216	0.00134	0.78350
H	15	0.21650	0.00000	0.78216	0.00134	0.78350
N	16	-0.48061	1.99942	5.45874	0.02246	7.48061
C	17	0.05453	1.99923	3.91961	0.02663	5.94547
C	18	-0.24659	1.99917	4.23111	0.01631	6.24659
C	19	-0.16699	1.99918	4.15046	0.01735	6.16699
C	20	-0.24659	1.99917	4.23111	0.01631	6.24659
C	21	0.05453	1.99923	3.91961	0.02663	5.94547
H	22	0.19814	0.00000	0.80038	0.00148	0.80186
H	23	0.22259	0.00000	0.77592	0.00149	0.77741
H	24	0.21903	0.00000	0.77969	0.00128	0.78097
H	25	0.22259	0.00000	0.77592	0.00149	0.77741
H	26	0.19814	0.00000	0.80038	0.00148	0.80186
=====						
* Total *		0.00000	71.98671	71.71945	0.29384	144.00000

Contributions to pi population at N atom in pyridine:

Val(2p)	1.18974
Ryd(3p)	0.00080
Ryd(4p)	0.00037
Ryd(5p)	0.00002

Total: 1.19093

NMR properties

Electronic energy (NMR): -7438.47308046

1	C	Isotropic =	40.4880	Anisotropy =	134.3040
2	C	Isotropic =	36.4073	Anisotropy =	161.0137
3	C	Isotropic =	43.4759	Anisotropy =	160.2570
4	C	Isotropic =	33.9686	Anisotropy =	206.9717
5	C	Isotropic =	43.4497	Anisotropy =	171.1440
6	C	Isotropic =	36.6850	Anisotropy =	161.4147
7	H	Isotropic =	23.8228	Anisotropy =	11.8107
8	H	Isotropic =	24.1970	Anisotropy =	6.7157

9	C	Isotropic =	159.6141	Anisotropy =	41.6860
10	H	Isotropic =	24.0732	Anisotropy =	7.2358
11	H	Isotropic =	23.7758	Anisotropy =	11.8221
12	I	Isotropic =	3232.0591	Anisotropy =	3113.2495
13	H	Isotropic =	29.5226	Anisotropy =	7.6382
14	H	Isotropic =	28.9901	Anisotropy =	8.6033
15	H	Isotropic =	28.9901	Anisotropy =	8.6033
16	N	Isotropic =	-96.1964	Anisotropy =	565.6524
17	C	Isotropic =	22.7710	Anisotropy =	180.9132
18	C	Isotropic =	50.4044	Anisotropy =	183.7083
19	C	Isotropic =	37.0622	Anisotropy =	212.5981
20	C	Isotropic =	50.4044	Anisotropy =	183.7083
21	C	Isotropic =	22.7710	Anisotropy =	180.9132
22	H	Isotropic =	22.5932	Anisotropy =	10.6783
23	H	Isotropic =	23.9548	Anisotropy =	5.3661
24	H	Isotropic =	23.5186	Anisotropy =	4.8464
25	H	Isotropic =	23.9548	Anisotropy =	5.3661
26	H	Isotropic =	22.5932	Anisotropy =	10.6783

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X = I R = CHCH32

Energies and NBO populations

Number of atoms: 32

Full point group

C1

NOp 1

C	-0.007974	-0.313207	-0.068934
C	-0.380435	-0.949786	1.116951
C	-0.463483	-2.345481	1.154638
C	-0.181069	-3.128145	0.027483
C	0.192462	-2.464958	-1.152263
C	0.280077	-1.072436	-1.208204
H	-0.605687	-0.370167	2.006134
H	-0.755046	-2.830603	2.083317
C	-0.278250	-4.647814	0.090852
H	0.419949	-3.036262	-2.048752
H	0.569881	-0.585582	-2.133613
I	0.116302	1.814874	-0.143626
H	-0.593936	-4.903682	1.110775
C	-1.346503	-5.193742	-0.878461
C	1.088495	-5.318892	-0.155424
H	-1.441071	-6.280296	-0.771635
H	-1.080743	-4.982051	-1.920583
H	-2.324966	-4.743233	-0.682569
H	1.840816	-4.954665	0.551671
H	1.007114	-6.405746	-0.041045
H	1.452667	-5.115839	-1.169133
N	0.226534	4.939401	-0.215029
C	1.372925	5.632276	-0.215029
C	1.426691	7.027832	-0.215055
C	0.226385	7.739927	-0.215349
C	-0.973851	7.027778	-0.215422
C	-0.919863	5.632274	-0.215029
H	2.287337	5.042883	-0.214537
H	2.384938	7.536894	-0.214761
H	0.226407	8.825761	-0.215274
H	-1.932218	7.536618	-0.215521
H	-1.834311	5.042808	-0.213953

Electronic energy:	-7517.09363449 (-7517.09365451)
Zero-point correction=	0.261707
Sum of electronic and zero-point Energies=	-7516.831927
Sum of electronic and thermal Energies=	-7516.814932
Sum of electronic and thermal Enthalpies=	-7516.813988
Sum of electronic and thermal Free Energies=	-7516.884324

Lowest vibration frequency: 6.6128

Counterpoise: BSSE energy = 0.000575744854

Summary of Natural Population Analysis:

	Natural Population
Natural	-----

Atom	No	Charge	Core	Valence	Rydberg	Total
C	1	-0.23088	1.99874	4.20502	0.02712	6.23088
C	2	-0.22887	1.99900	4.20912	0.02076	6.22887
C	3	-0.19736	1.99905	4.18174	0.01658	6.19736
C	4	-0.02117	1.99902	4.00195	0.02020	6.02117
C	5	-0.19996	1.99905	4.18462	0.01629	6.19996
C	6	-0.22804	1.99900	4.20817	0.02086	6.22804
H	7	0.22284	0.00000	0.77554	0.00162	0.77716
H	8	0.21305	0.00000	0.78522	0.00173	0.78695
C	9	-0.24351	1.99920	4.22863	0.01568	6.24351
H	10	0.21334	0.00000	0.78494	0.00173	0.78666
H	11	0.22244	0.00000	0.77595	0.00161	0.77756
I	12	0.17565	45.99808	6.80338	0.02290	52.82435
H	13	0.19836	0.00000	0.79865	0.00299	0.80164
C	14	-0.57025	1.99935	4.56144	0.00946	6.57025
C	15	-0.57010	1.99935	4.56130	0.00945	6.57010
H	16	0.20292	0.00000	0.79587	0.00120	0.79708
H	17	0.19768	0.00000	0.80096	0.00137	0.80232
H	18	0.20735	0.00000	0.79163	0.00101	0.79265
H	19	0.20736	0.00000	0.79163	0.00102	0.79264
H	20	0.20283	0.00000	0.79597	0.00121	0.79717
H	21	0.19774	0.00000	0.80090	0.00137	0.80226
N	22	-0.48052	1.99942	5.45867	0.02244	7.48052
C	23	0.05470	1.99923	3.91944	0.02664	5.94530
C	24	-0.24669	1.99917	4.23121	0.01631	6.24669
C	25	-0.16700	1.99918	4.15047	0.01736	6.16700
C	26	-0.24657	1.99917	4.23109	0.01631	6.24657
C	27	0.05433	1.99922	3.91983	0.02662	5.94567
H	28	0.19801	0.00000	0.80053	0.00145	0.80199
H	29	0.22256	0.00000	0.77595	0.00149	0.77744
H	30	0.21901	0.00000	0.77971	0.00128	0.78099
H	31	0.22258	0.00000	0.77593	0.00149	0.77742
H	32	0.19818	0.00000	0.80031	0.00151	0.80182
=====						
* Total *		0.00000	75.98524	83.68572	0.32905	160.00000

Contributions to pi population at N atom in pyridine:

```

Val( 2p)    1.18963
Ryd( 3p)    0.00081
Ryd( 4p)    0.00038
Ryd( 5p)    0.00002

Total:      1.19084

```

NMR properties

Electronic energy (NMR): -7517.09363449

1	C	Isotropic =	39.7539	Anisotropy =	133.3445
2	C	Isotropic =	36.3021	Anisotropy =	160.3427
3	C	Isotropic =	44.3931	Anisotropy =	154.7170
4	C	Isotropic =	22.2269	Anisotropy =	215.9476
5	C	Isotropic =	47.0963	Anisotropy =	183.8068
6	C	Isotropic =	36.2478	Anisotropy =	161.9927
7	H	Isotropic =	23.8122	Anisotropy =	11.6856
8	H	Isotropic =	24.1956	Anisotropy =	7.3744
9	C	Isotropic =	139.5764	Anisotropy =	24.8615
10	H	Isotropic =	24.0372	Anisotropy =	9.6800
11	H	Isotropic =	23.7229	Anisotropy =	12.0042
12	I	Isotropic =	3228.3591	Anisotropy =	3117.3709
13	H	Isotropic =	28.7570	Anisotropy =	6.8543
14	C	Isotropic =	155.8020	Anisotropy =	36.7633
15	C	Isotropic =	155.7741	Anisotropy =	36.8192
16	H	Isotropic =	30.2875	Anisotropy =	10.2732
17	H	Isotropic =	30.3108	Anisotropy =	6.4795
18	H	Isotropic =	30.2959	Anisotropy =	8.3954
19	H	Isotropic =	30.3303	Anisotropy =	8.3932
20	H	Isotropic =	30.2690	Anisotropy =	10.2949
21	H	Isotropic =	30.3388	Anisotropy =	6.5646
22	N	Isotropic =	-96.5561	Anisotropy =	565.1519
23	C	Isotropic =	22.8281	Anisotropy =	180.9213
24	C	Isotropic =	50.4037	Anisotropy =	183.3566
25	C	Isotropic =	37.2471	Anisotropy =	212.9671
26	C	Isotropic =	50.3728	Anisotropy =	183.4464
27	C	Isotropic =	22.8033	Anisotropy =	181.4146

28	H	Isotropic =	22.6039	Anisotropy =	10.5111
29	H	Isotropic =	23.9572	Anisotropy =	5.3316
30	H	Isotropic =	23.5122	Anisotropy =	4.8359
31	H	Isotropic =	23.9534	Anisotropy =	5.4322
32	H	Isotropic =	22.5843	Anisotropy =	10.9688

=====

X = I R = H

Energies and NBO populations

Number of atoms: 23
Full point group C2V NOp 4

C	0.000000	-0.525753	0.000000
C	0.000000	-1.223880	1.212723
C	0.000000	-2.621398	1.216732
C	0.000000	-3.291435	0.000000
C	0.000000	-2.621398	-1.216732
C	0.000000	-1.223880	-1.212723
H	0.000000	-0.688501	2.155942
H	0.000000	-3.181463	2.145729
F	0.000000	-4.645207	0.000000
H	0.000000	-3.181463	-2.145729
H	0.000000	-0.688501	-2.155942
I	0.000000	1.605928	0.000000
N	0.000000	4.690377	0.000000
C	1.146782	5.382637	0.000000
C	1.200337	6.777975	0.000000
C	0.000000	7.490019	0.000000
C	-1.200337	6.777975	0.000000
C	-1.146782	5.382637	0.000000
H	2.060700	4.792534	0.000000
H	2.158534	7.287022	0.000000
H	0.000000	8.575809	0.000000
H	-2.158534	7.287022	0.000000
H	-2.060700	4.792534	0.000000

Electronic energy:	-7498.40323600 (-7498.40323600)
Zero-point correction=	0.169980
Sum of electronic and zero-point Energies=	-7498.233256
Sum of electronic and thermal Energies=	-7498.219893
Sum of electronic and thermal Enthalpies=	-7498.218949
Sum of electronic and thermal Free Energies=	-7498.279570

Lowest vibration frequency: 11.1323

Counterpoise: BSSE energy = 0.000705938760

Summary of Natural Population Analysis:

		Natural Population				
Atom	No	Natural Charge	Core	Valence	Rydberg	Total
C	1	-0.24366	1.99876	4.21775	0.02715	6.24366
C	2	-0.21463	1.99901	4.19564	0.01997	6.21463
C	3	-0.26223	1.99905	4.24613	0.01705	6.26223
C	4	0.40460	1.99849	3.57039	0.02652	5.99540
C	5	-0.26223	1.99905	4.24613	0.01705	6.26223
C	6	-0.21463	1.99901	4.19564	0.01997	6.21463
H	7	0.22784	0.00000	0.77060	0.00156	0.77216
H	8	0.23464	0.00000	0.76380	0.00156	0.76536
F	9	-0.35494	1.99993	7.34701	0.00800	9.35494
H	10	0.23464	0.00000	0.76380	0.00156	0.76536
H	11	0.22784	0.00000	0.77060	0.00156	0.77216
I	12	0.18998	45.99810	6.78873	0.02319	52.81002
N	13	-0.48308	1.99941	5.46117	0.02249	7.48308
C	14	0.05518	1.99922	3.91907	0.02652	5.94482
C	15	-0.24582	1.99917	4.23037	0.01629	6.24582
C	16	-0.16585	1.99918	4.14933	0.01733	6.16585
C	17	-0.24582	1.99917	4.23037	0.01629	6.24582
C	18	0.05518	1.99922	3.91907	0.02652	5.94482
H	19	0.19852	0.00000	0.80000	0.00148	0.80148
H	20	0.22316	0.00000	0.77536	0.00148	0.77684

H	21	0.21958	0.00000	0.77914	0.00127	0.78042
H	22	0.22316	0.00000	0.77536	0.00148	0.77684
H	23	0.19852	0.00000	0.80000	0.00148	0.80148
=====						
* Total *		0.00000	71.98679	71.71546	0.29775	144.00000

Contributions to pi population at N atom in pyridine:

Val(2p)	1.19334
Ryd(3p)	0.00081
Ryd(4p)	0.00038
Ryd(5p)	0.00002

Total:	1.19455
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NMR properties

Electronic energy (NMR): -7498.40323600

1	C	Isotropic =	42.3971	Anisotropy =	137.4914
2	C	Isotropic =	34.8483	Anisotropy =	163.9808
3	C	Isotropic =	58.2884	Anisotropy =	152.3919
4	C	Isotropic =	6.4124	Anisotropy =	108.1165
5	C	Isotropic =	58.2884	Anisotropy =	152.3919
6	C	Isotropic =	34.8483	Anisotropy =	163.9808
7	H	Isotropic =	23.7140	Anisotropy =	11.5719
8	H	Isotropic =	24.3110	Anisotropy =	4.7880
9	F	Isotropic =	294.0431	Anisotropy =	97.3839
10	H	Isotropic =	24.3110	Anisotropy =	4.7880
11	H	Isotropic =	23.7140	Anisotropy =	11.5719
12	I	Isotropic =	3202.7053	Anisotropy =	3155.9798
13	N	Isotropic =	-94.8637	Anisotropy =	561.5538
14	C	Isotropic =	22.9406	Anisotropy =	180.8756
15	C	Isotropic =	50.2887	Anisotropy =	183.9685
16	C	Isotropic =	36.8846	Anisotropy =	213.0350
17	C	Isotropic =	50.2887	Anisotropy =	183.9685
18	C	Isotropic =	22.9406	Anisotropy =	180.8756
19	H	Isotropic =	22.6071	Anisotropy =	10.7080
20	H	Isotropic =	23.9423	Anisotropy =	5.3590
21	H	Isotropic =	23.5055	Anisotropy =	4.8533
22	H	Isotropic =	23.9423	Anisotropy =	5.3590
23	H	Isotropic =	22.6071	Anisotropy =	10.7080

=====

X = I R = Cl

Energies and NBO populations

Number of atoms: 23
Full point group C2V NOp 4

C	0.000000	-0.487592	0.000000
C	0.000000	-1.186437	1.211671
C	0.000000	-2.583736	1.214941
C	0.000000	-3.267011	0.000000
C	0.000000	-2.583736	-1.214941
C	0.000000	-1.186437	-1.211671
H	0.000000	-0.652840	2.156049
H	0.000000	-3.130133	2.151722
Cl	0.000000	-5.029181	0.000000
H	0.000000	-3.130133	-2.151722
H	0.000000	-0.652840	-2.156049
I	0.000000	1.643142	0.000000
N	0.000000	4.720311	0.000000
C	1.146952	5.412338	0.000000
C	1.200413	6.807678	0.000000
C	0.000000	7.519591	0.000000
C	-1.200413	6.807678	0.000000
C	-1.146952	5.412338	0.000000
H	2.060879	4.822253	0.000000
H	2.158595	7.316704	0.000000
H	0.000000	8.605376	0.000000
H	-2.158595	7.316704	0.000000
H	-2.060879	4.822253	0.000000

Electronic energy: -7858.72558324 (-7858.72558324)
 Zero-point correction= 0.168677
 Sum of electronic and zero-point Energies= -7858.556906
 Sum of electronic and thermal Energies= -7858.543161
 Sum of electronic and thermal Enthalpies= -7858.542217
 Sum of electronic and thermal Free Energies= -7858.604047

Lowest vibration frequency: 11.3387

Counterpoise: BSSE energy = 0.000705924157

Summary of Natural Population Analysis:

		Natural Population				
Atom	No	Natural Charge	Core	Valence	Rydberg	Total
C	1	-0.23217	1.99876	4.20628	0.02714	6.23217
C	2	-0.21619	1.99900	4.19683	0.02037	6.21619
C	3	-0.21998	1.99897	4.20226	0.01875	6.21998
C	4	-0.04387	1.99860	4.02256	0.02271	6.04387
C	5	-0.21998	1.99897	4.20226	0.01875	6.21998
C	6	-0.21619	1.99900	4.19683	0.02037	6.21619
H	7	0.22905	0.00000	0.76937	0.00158	0.77095
H	8	0.23349	0.00000	0.76486	0.00165	0.76651
Cl	9	-0.00552	9.99962	6.98593	0.01997	17.00552
H	10	0.23349	0.00000	0.76486	0.00165	0.76651
H	11	0.22905	0.00000	0.76937	0.00158	0.77095
I	12	0.19522	45.99810	6.78340	0.02329	52.80478
N	13	-0.48370	1.99941	5.46182	0.02247	7.48370
C	14	0.05535	1.99922	3.91894	0.02648	5.94465
C	15	-0.24561	1.99917	4.23017	0.01628	6.24561
C	16	-0.16562	1.99918	4.14911	0.01733	6.16562
C	17	-0.24561	1.99917	4.23017	0.01628	6.24561
C	18	0.05535	1.99922	3.91894	0.02648	5.94465
H	19	0.19860	0.00000	0.79992	0.00148	0.80140
H	20	0.22328	0.00000	0.77523	0.00148	0.77672
H	21	0.21969	0.00000	0.77904	0.00127	0.78031
H	22	0.22328	0.00000	0.77523	0.00148	0.77672
H	23	0.19860	0.00000	0.79992	0.00148	0.80140
=====						
* Total *		0.00000	79.98639	71.70331	0.31030	152.00000

Contributions to pi population at N atom in pyridine:

Val(2p) 1.19414
 Ryd(3p) 0.00081
 Ryd(4p) 0.00038
 Ryd(5p) 0.00002
 Total: 1.19535

NMR properties

Electronic energy (NMR): -7858.72558324

1	C	Isotropic =	37.2790	Anisotropy =	146.5006
2	C	Isotropic =	35.3379	Anisotropy =	161.4783
3	C	Isotropic =	44.2592	Anisotropy =	157.1247
4	C	Isotropic =	31.3266	Anisotropy =	123.1087
5	C	Isotropic =	44.2592	Anisotropy =	157.1247
6	C	Isotropic =	35.3379	Anisotropy =	161.4783
7	H	Isotropic =	23.7477	Anisotropy =	11.6351
8	H	Isotropic =	24.0417	Anisotropy =	7.4124
9	Cl	Isotropic =	686.8649	Anisotropy =	501.5431
10	H	Isotropic =	24.0417	Anisotropy =	7.4124
11	H	Isotropic =	23.7477	Anisotropy =	11.6351
12	I	Isotropic =	3183.4077	Anisotropy =	3141.3335
13	N	Isotropic =	-94.4717	Anisotropy =	560.5017
14	C	Isotropic =	22.9814	Anisotropy =	180.8153
15	C	Isotropic =	50.2504	Anisotropy =	184.0500
16	C	Isotropic =	36.8299	Anisotropy =	213.1007
17	C	Isotropic =	50.2504	Anisotropy =	184.0500
18	C	Isotropic =	22.9814	Anisotropy =	180.8153
19	H	Isotropic =	22.6135	Anisotropy =	10.7543
20	H	Isotropic =	23.9406	Anisotropy =	5.3714
21	H	Isotropic =	23.5046	Anisotropy =	4.8489

22 H Isotropic = 23.9406 Anisotropy = 5.3714
 23 H Isotropic = 22.6135 Anisotropy = 10.7543

=====

X = I R = OCF3

Energies and NBO populations

Number of atoms: 27

Full point group C1 NOp 1

C	0.004787	-0.346961	0.065111
C	0.093874	-1.039194	-1.144838
C	0.103264	-2.437658	-1.165469
C	0.010407	-3.127606	0.041465
C	-0.080474	-2.450614	1.256091
C	-0.080432	-1.056342	1.268479
H	0.162019	-0.500271	-2.083475
H	0.185650	-2.955827	-2.112600
O	0.039532	-4.524004	0.160559
H	-0.150144	-3.017436	2.178305
H	-0.150913	-0.531552	2.214907
I	-0.002023	1.784020	0.075795
C	-0.257999	-5.329484	-0.876549
F	-0.309234	-6.578016	-0.414036
F	0.670309	-5.291908	-1.856757
F	-1.441948	-5.038652	-1.446437
N	-0.012639	4.855820	0.072622
C	-1.159687	5.547801	0.072622
C	-1.213059	6.943091	0.072637
C	-0.012679	7.655009	0.072905
C	1.187777	6.943201	0.072734
C	1.134340	5.547814	0.072622
H	-2.073732	4.957924	0.072650
H	-2.171232	7.452123	0.072635
H	-0.012749	8.740793	0.073269
H	2.145929	7.452268	0.072790
H	2.048249	4.957713	0.072523

Electronic energy: -7811.44200808 (-7811.44194046)
 Zero-point correction= 0.186987
 Sum of electronic and zero-point Energies= -7811.255021
 Sum of electronic and thermal Energies= -7811.238035
 Sum of electronic and thermal Enthalpies= -7811.237091
 Sum of electronic and thermal Free Energies= -7811.307596

Lowest vibration frequency: 11.8355

Counterpoise: BSSE energy = 0.000671572644

Summary of Natural Population Analysis:

		Natural Population				
Atom	No	Natural Charge	Core	Valence	Rydberg	Total
C	1	-0.23854	1.99875	4.21256	0.02723	6.23854
C	2	-0.20993	1.99901	4.19069	0.02023	6.20993
C	3	-0.25417	1.99904	4.23937	0.01576	6.25417
C	4	0.28687	1.99870	3.69196	0.02247	5.71313
C	5	-0.22548	1.99906	4.20930	0.01713	6.22548
C	6	-0.21222	1.99901	4.19329	0.01992	6.21222
H	7	0.22939	0.00000	0.76907	0.00153	0.77061
H	8	0.23718	0.00000	0.76092	0.00190	0.76282
O	9	-0.55297	1.99964	6.53713	0.01620	8.55297
H	10	0.23586	0.00000	0.76251	0.00163	0.76414
H	11	0.22941	0.00000	0.76908	0.00151	0.77059
I	12	0.19578	45.99810	6.78292	0.02320	52.80422
C	13	1.28354	1.99965	2.64649	0.07031	4.71646
F	14	-0.33353	1.99991	7.32731	0.00632	9.33353
F	15	-0.35570	1.99992	7.34893	0.00685	9.35570
F	16	-0.34964	1.99991	7.34299	0.00673	9.34964
N	17	-0.48396	1.99941	5.46207	0.02249	7.48396
C	18	0.05545	1.99922	3.91885	0.02647	5.94455
C	19	-0.24551	1.99917	4.23007	0.01627	6.24551

C	20	-0.16548	1.99918	4.14897	0.01733	6.16548
C	21	-0.24551	1.99917	4.23006	0.01627	6.24551
C	22	0.05540	1.99922	3.91891	0.02647	5.94460
H	23	0.19867	0.00000	0.79986	0.00147	0.80133
H	24	0.22335	0.00000	0.77517	0.00148	0.77665
H	25	0.21976	0.00000	0.77897	0.00127	0.78024
H	26	0.22335	0.00000	0.77516	0.00148	0.77665
H	27	0.19863	0.00000	0.79989	0.00148	0.80137
=====						
* Total *		0.00000	79.98607	95.62250	0.39142	176.00000

Contributions to pi population at N atom in pyridine:

Val(2p)	1.19459
Ryd(3p)	0.00081
Ryd(4p)	0.00038
Ryd(5p)	0.00002

Total: 1.19580

----- NMR properties -----

Electronic energy (NMR): -7811.44200808

1	C	Isotropic =	40.0252	Anisotropy =	141.9064
2	C	Isotropic =	35.5102	Anisotropy =	164.6885
3	C	Isotropic =	58.9082	Anisotropy =	159.6043
4	C	Isotropic =	20.9675	Anisotropy =	119.7546
5	C	Isotropic =	55.0788	Anisotropy =	155.4456
6	C	Isotropic =	34.6864	Anisotropy =	165.3147
7	H	Isotropic =	23.6696	Anisotropy =	11.9798
8	H	Isotropic =	24.0978	Anisotropy =	9.6473
9	O	Isotropic =	156.6522	Anisotropy =	41.9549
10	H	Isotropic =	24.2879	Anisotropy =	6.2964
11	H	Isotropic =	23.6664	Anisotropy =	11.4372
12	I	Isotropic =	3186.5554	Anisotropy =	3159.7843
13	C	Isotropic =	47.0420	Anisotropy =	9.7578
14	F	Isotropic =	236.1858	Anisotropy =	177.1019
15	F	Isotropic =	231.0784	Anisotropy =	167.1652
16	F	Isotropic =	245.6397	Anisotropy =	168.5918
17	N	Isotropic =	-94.3032	Anisotropy =	560.3131
18	C	Isotropic =	22.9956	Anisotropy =	180.7874
19	C	Isotropic =	50.2466	Anisotropy =	184.0423
20	C	Isotropic =	36.8264	Anisotropy =	213.1414
21	C	Isotropic =	50.2342	Anisotropy =	184.0982
22	C	Isotropic =	22.9639	Anisotropy =	180.7720
23	H	Isotropic =	22.6083	Anisotropy =	10.7850
24	H	Isotropic =	23.9420	Anisotropy =	5.3667
25	H	Isotropic =	23.5013	Anisotropy =	4.8254
26	H	Isotropic =	23.9300	Anisotropy =	5.3870
27	H	Isotropic =	22.6041	Anisotropy =	10.8383

=====

X = I R = COOCH3

Energies and NBO populations -----

Number of atoms: 29
Full point group CS NOp 2

C	0.000000	-0.356497	0.015723
C	0.000000	-1.065449	1.223487
C	0.000000	-2.458901	1.203960
C	0.000000	-3.154107	-0.013554
C	0.000000	-2.435388	-1.217981
C	0.000000	-1.040670	-1.205697
H	0.000000	-0.537254	2.170789
H	0.000000	-3.020828	2.132204
C	0.000000	-4.646712	0.027732
H	0.000000	-2.965740	-2.163456
H	0.000000	-0.492619	-2.141695
I	0.000000	1.772640	0.035205
O	0.000000	-5.195492	-1.201297
C	0.000000	-6.633077	-1.248897
H	0.000000	-6.888004	-2.307967

H	-0.891393	-7.030417	-0.757967
H	0.891393	-7.030417	-0.757967
O	0.000000	-5.303506	1.047880
N	0.000000	4.851533	0.048233
C	1.146915	5.543603	0.048233
C	1.200414	6.938927	0.048174
C	0.000000	7.650894	0.048119
C	-1.200414	6.938927	0.048174
C	-1.146915	5.543603	0.048233
H	2.060869	4.953558	0.048231
H	2.158618	7.447910	0.048180
H	0.000000	8.736676	0.048047
H	-2.158618	7.447910	0.048180
H	-2.060869	4.953558	0.048231

Electronic energy:	-7627.04416581 (-7627.04416579)
Zero-point correction=	0.220793
Sum of electronic and zero-point Energies=	-7626.823373
Sum of electronic and thermal Energies=	-7626.806286
Sum of electronic and thermal Enthalpies=	-7626.805341
Sum of electronic and thermal Free Energies=	-7626.875390

Lowest vibration frequency: 11.7425

Counterpoise: BSSE energy = 0.000680913739

Summary of Natural Population Analysis:

		Natural Population				
Atom	No	Natural Charge	Core	Valence	Rydberg	Total
C	1	-0.20682	1.99877	4.18066	0.02738	6.20682
C	2	-0.23552	1.99900	4.21620	0.02032	6.23552
C	3	-0.14897	1.99910	4.13275	0.01712	6.14897
C	4	-0.16680	1.99893	4.14981	0.01806	6.16680
C	5	-0.15469	1.99910	4.13915	0.01644	6.15469
C	6	-0.23627	1.99900	4.21695	0.02032	6.23627
H	7	0.22777	0.00000	0.77068	0.00155	0.77223
H	8	0.23663	0.00000	0.76174	0.00163	0.76337
C	9	0.80358	1.99930	3.15045	0.04667	5.19642
H	10	0.23337	0.00000	0.76508	0.00155	0.76663
H	11	0.22714	0.00000	0.77129	0.00157	0.77286
I	12	0.20239	45.99807	6.77633	0.02320	52.79761
O	13	-0.55068	1.99972	6.53677	0.01420	8.55068
C	14	-0.21746	1.99926	4.20242	0.01577	6.21746
H	15	0.19151	0.00000	0.80741	0.00108	0.80849
H	16	0.18973	0.00000	0.80856	0.00171	0.81027
H	17	0.18973	0.00000	0.80856	0.00171	0.81027
O	18	-0.61821	1.99975	6.60723	0.01123	8.61821
N	19	-0.48395	1.99941	5.46209	0.02244	7.48395
C	20	0.05535	1.99922	3.91895	0.02647	5.94465
C	21	-0.24556	1.99917	4.23012	0.01627	6.24556
C	22	-0.16554	1.99918	4.14903	0.01733	6.16554
C	23	-0.24556	1.99917	4.23012	0.01627	6.24556
C	24	0.05535	1.99922	3.91895	0.02647	5.94465
H	25	0.19857	0.00000	0.79996	0.00147	0.80143
H	26	0.22331	0.00000	0.77521	0.00148	0.77669
H	27	0.21971	0.00000	0.77902	0.00127	0.78029
H	28	0.22331	0.00000	0.77521	0.00148	0.77669
H	29	0.19857	0.00000	0.79996	0.00147	0.80143
=====						
* Total *		0.00000	77.98539	87.64066	0.37395	166.00000

Contributions to pi population at N atom in pyridine:

Val(2p)	1.19422
Ryd(3p)	0.00081
Ryd(4p)	0.00038
Ryd(5p)	0.00002
Total:	1.19543

NMR properties

Electronic energy (NMR):	-7627.04416581
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1	C	Isotropic =	25.9984	Anisotropy =	158.2798
2	C	Isotropic =	36.5210	Anisotropy =	163.9242
3	C	Isotropic =	42.2466	Anisotropy =	192.5237
4	C	Isotropic =	44.6939	Anisotropy =	167.2227
5	C	Isotropic =	43.7275	Anisotropy =	187.3074
6	C	Isotropic =	36.9932	Anisotropy =	163.2878
7	H	Isotropic =	23.5543	Anisotropy =	12.0425
8	H	Isotropic =	23.1291	Anisotropy =	7.8667
9	C	Isotropic =	6.1051	Anisotropy =	85.7470
10	H	Isotropic =	23.3346	Anisotropy =	8.8208
11	H	Isotropic =	23.6084	Anisotropy =	12.1824
12	I	Isotropic =	3139.1645	Anisotropy =	3098.8820
13	O	Isotropic =	145.4559	Anisotropy =	141.8150
14	C	Isotropic =	126.7688	Anisotropy =	73.1295
15	H	Isotropic =	27.7780	Anisotropy =	7.6484
16	H	Isotropic =	27.6141	Anisotropy =	8.5179
17	H	Isotropic =	27.6141	Anisotropy =	8.5179
18	O	Isotropic =	-65.9414	Anisotropy =	603.4103
19	N	Isotropic =	-94.3236	Anisotropy =	560.4446
20	C	Isotropic =	22.9686	Anisotropy =	180.8228
21	C	Isotropic =	50.2321	Anisotropy =	184.1228
22	C	Isotropic =	36.7838	Anisotropy =	213.1060
23	C	Isotropic =	50.2321	Anisotropy =	184.1228
24	C	Isotropic =	22.9686	Anisotropy =	180.8228
25	H	Isotropic =	22.5988	Anisotropy =	10.7448
26	H	Isotropic =	23.9311	Anisotropy =	5.3667
27	H	Isotropic =	23.4971	Anisotropy =	4.8320
28	H	Isotropic =	23.9311	Anisotropy =	5.3667
29	H	Isotropic =	22.5988	Anisotropy =	10.7448

=====

X = I R = COCH3

Energies and NBO populations

Number of atoms: 28

Full point group

CS

NOp 2

C	0.000000	-0.410150	-0.002704
C	0.000000	-1.069407	1.234387
C	0.000000	-2.460848	1.272499
C	0.000000	-3.215238	0.087354
C	0.000000	-2.539935	-1.143369
C	0.000000	-1.144943	-1.193048
H	0.000000	-0.501015	2.158359
H	0.000000	-2.984273	2.223137
C	0.000000	-4.714264	0.186448
H	0.000000	-3.092812	-2.077201
H	0.000000	-0.637783	-2.151812
I	0.000000	1.717835	-0.068150
C	0.000000	-5.531142	-1.093917
O	0.000000	-5.263941	1.275406
H	0.000000	-6.590592	-0.836035
H	0.883548	-5.303998	-1.700251
H	-0.883548	-5.303998	-1.700251
N	0.000000	4.792325	-0.131973
C	1.147089	5.484446	-0.131973
C	1.200320	6.879788	-0.132131
C	0.000000	7.591481	-0.132161
C	-1.200320	6.879788	-0.132131
C	-1.147089	5.484446	-0.131973
H	2.061145	4.894573	-0.131274
H	2.158410	7.388963	-0.131782
H	0.000000	8.677268	-0.131868
H	-2.158410	7.388963	-0.131782
H	-2.061145	4.894573	-0.131274

Electronic energy:

Zero-point correction=

Sum of electronic and zero-point Energies=

Sum of electronic and thermal Energies=

Sum of electronic and thermal Enthalpies=

Sum of electronic and thermal Free Energies=

-7551.80861151 (-7551.80861158)

0.215117

-7551.593495

-7551.577344

-7551.576400

-7551.644139

Lowest vibration frequency:

11.9137

Counterpoise: BSSE energy = 0.000669636196

Summary of Natural Population Analysis:

		Natural Population				
Atom	No	Natural Charge	Core	Valence	Rydberg	Total
C	1	-0.20575	1.99877	4.17942	0.02755	6.20575
C	2	-0.23369	1.99900	4.21446	0.02023	6.23369
C	3	-0.14535	1.99909	4.12812	0.01814	6.14535
C	4	-0.16126	1.99895	4.14484	0.01747	6.16126
C	5	-0.16395	1.99910	4.14931	0.01554	6.16395
C	6	-0.23675	1.99900	4.21709	0.02066	6.23675
H	7	0.22744	0.00000	0.77100	0.00156	0.77256
H	8	0.23677	0.00000	0.76135	0.00188	0.76323
C	9	0.56535	1.99931	3.39663	0.03870	5.43465
H	10	0.22018	0.00000	0.77834	0.00148	0.77982
H	11	0.22698	0.00000	0.77142	0.00160	0.77302
I	12	0.20454	45.99807	6.77416	0.02322	52.79546
C	13	-0.67809	1.99930	4.67047	0.00832	6.67809
O	14	-0.57861	1.99976	6.56514	0.01370	8.57861
H	15	0.23114	0.00000	0.76698	0.00188	0.76886
H	16	0.22854	0.00000	0.77005	0.00141	0.77146
H	17	0.22854	0.00000	0.77005	0.00141	0.77146
N	18	-0.48433	1.99941	5.46248	0.02244	7.48433
C	19	0.05541	1.99922	3.91891	0.02645	5.94459
C	20	-0.24543	1.99917	4.22999	0.01627	6.24543
C	21	-0.16542	1.99918	4.14891	0.01733	6.16542
C	22	-0.24543	1.99917	4.22999	0.01627	6.24543
C	23	0.05541	1.99922	3.91891	0.02645	5.94459
H	24	0.19859	0.00000	0.79994	0.00147	0.80141
H	25	0.22339	0.00000	0.77513	0.00148	0.77661
H	26	0.21980	0.00000	0.77893	0.00127	0.78020
H	27	0.22339	0.00000	0.77513	0.00148	0.77661
H	28	0.19859	0.00000	0.79994	0.00147	0.80141
* Total *		0.00000	75.98574	81.66712	0.34714	158.00000

Contributions to pi population at N atom in pyridine:

Val(2p) 1.19466
Ryd(3p) 0.00081
Ryd(4p) 0.00038
Ryd(5p) 0.00002

Total: 1.19587

NMR properties

Electronic energy (NMR): -7551.80861151

1	C	Isotropic =	25.2543	Anisotropy =	159.1643
2	C	Isotropic =	36.1183	Anisotropy =	163.5639
3	C	Isotropic =	45.4630	Anisotropy =	196.4004
4	C	Isotropic =	38.1239	Anisotropy =	180.3063
5	C	Isotropic =	43.2512	Anisotropy =	181.6166
6	C	Isotropic =	37.1010	Anisotropy =	162.8520
7	H	Isotropic =	23.5510	Anisotropy =	12.1352
8	H	Isotropic =	23.1541	Anisotropy =	7.0652
9	C	Isotropic =	-26.0274	Anisotropy =	173.8296
10	H	Isotropic =	23.4624	Anisotropy =	8.5816
11	H	Isotropic =	23.5655	Anisotropy =	12.2120
12	I	Isotropic =	3133.6217	Anisotropy =	3089.6734
13	C	Isotropic =	153.8732	Anisotropy =	46.4303
14	O	Isotropic =	-290.7385	Anisotropy =	967.7174
15	H	Isotropic =	29.3605	Anisotropy =	7.0408
16	H	Isotropic =	28.7365	Anisotropy =	4.6607
17	H	Isotropic =	28.7365	Anisotropy =	4.6607
18	N	Isotropic =	-94.3101	Anisotropy =	560.0627
19	C	Isotropic =	22.9695	Anisotropy =	180.8706
20	C	Isotropic =	50.2234	Anisotropy =	184.1533
21	C	Isotropic =	36.8134	Anisotropy =	213.1478
22	C	Isotropic =	50.2234	Anisotropy =	184.1533
23	C	Isotropic =	22.9695	Anisotropy =	180.8706
24	H	Isotropic =	22.5988	Anisotropy =	10.7143

25	H	Isotropic =	23.9290	Anisotropy =	5.3582
26	H	Isotropic =	23.4959	Anisotropy =	4.8406
27	H	Isotropic =	23.9290	Anisotropy =	5.3582
28	H	Isotropic =	22.5988	Anisotropy =	10.7143

=====

X = I R = CF3

Energies and NBO populations

Number of atoms: 26
Full point group CS NOp 2

C	0.000281	-0.430734	-0.015907
C	0.000127	-1.088714	1.217430
C	-0.000344	-2.485032	1.260827
C	-0.000609	-3.218929	0.073242
C	-0.000428	-2.559180	-1.161576
C	0.000020	-1.166369	-1.208105
H	0.000353	-0.523425	2.142891
H	-0.000487	-2.993703	2.218308
C	-0.001647	-4.725748	0.092002
H	-0.000528	-3.127435	-2.086622
H	0.000198	-0.659787	-2.166944
I	0.000927	1.698059	-0.081613
F	0.002130	-5.231818	1.341122
F	-1.087197	-5.238673	-0.536779
F	1.078794	-5.240271	-0.544097
N	0.001442	4.757427	-0.142844
C	1.148616	5.449172	-0.142844
C	1.201909	6.844455	-0.142652
C	0.001440	7.556261	-0.142495
C	-1.199027	6.844454	-0.142647
C	-1.145731	5.449173	-0.142844
H	2.062524	4.859111	-0.142584
H	2.160040	7.353513	-0.142422
H	0.001443	8.642032	-0.142198
H	-2.157161	7.353509	-0.142434
H	-2.059642	4.859114	-0.142586

Electronic energy:	-7736.22341853 (-7736.22343597)
Zero-point correction=	0.182513
Sum of electronic and zero-point Energies=	-7736.040906
Sum of electronic and thermal Energies=	-7736.025626
Sum of electronic and thermal Enthalpies=	-7736.024681
Sum of electronic and thermal Free Energies=	-7736.090498

Lowest vibration frequency: -22.8678

Counterpoise: BSSE energy = 0.000704977204

Summary of Natural Population Analysis:

		Natural Population				
Atom	No	Natural Charge	Core	Valence	Rydberg	Total
C	1	-0.21305	1.99877	4.18663	0.02765	6.21305
C	2	-0.22668	1.99900	4.20746	0.02023	6.22668
C	3	-0.17059	1.99908	4.15470	0.01681	6.17059
C	4	-0.15913	1.99891	4.14277	0.01745	6.15913
C	5	-0.16695	1.99909	4.15070	0.01716	6.16695
C	6	-0.22548	1.99901	4.20623	0.02024	6.22548
H	7	0.22972	0.00000	0.76875	0.00153	0.77028
H	8	0.23502	0.00000	0.76339	0.00159	0.76498
C	9	1.06722	1.99912	2.87356	0.06009	4.93278
H	10	0.23103	0.00000	0.76739	0.00158	0.76897
H	11	0.23005	0.00000	0.76840	0.00155	0.76995
I	12	0.20672	45.99809	6.77195	0.02324	52.79328
F	13	-0.35566	1.99992	7.34937	0.00638	9.35566
F	14	-0.35899	1.99992	7.35255	0.00653	9.35899
F	15	-0.35898	1.99992	7.35253	0.00653	9.35898
N	16	-0.48513	1.99941	5.46327	0.02245	7.48513
C	17	0.05567	1.99922	3.91869	0.02641	5.94433
C	18	-0.24512	1.99917	4.22969	0.01626	6.24512

C	19	-0.16496	1.99918	4.14846	0.01732	6.16496
C	20	-0.24512	1.99917	4.22969	0.01626	6.24512
C	21	0.05567	1.99922	3.91870	0.02641	5.94433
H	22	0.19876	0.00000	0.79977	0.00147	0.80124
H	23	0.22362	0.00000	0.77490	0.00148	0.77638
H	24	0.22001	0.00000	0.77873	0.00127	0.77999
H	25	0.22362	0.00000	0.77490	0.00148	0.77638
H	26	0.19876	0.00000	0.79977	0.00147	0.80124
=====						
* Total *		0.00000	77.98621	89.65296	0.36084	168.00000

Contributions to pi population at N atom in pyridine:

Val(2p) 1.19610
Ryd(3p) 0.00082
Ryd(4p) 0.00038
Ryd(5p) 0.00002

Total: 1.19732

----- NMR properties -----

Electronic energy (NMR): -7736.22341853

1	C	Isotropic =	28.0395	Anisotropy =	158.9877
2	C	Isotropic =	36.6836	Anisotropy =	166.6052
3	C	Isotropic =	48.4892	Anisotropy =	184.3964
4	C	Isotropic =	42.4941	Anisotropy =	166.9481
5	C	Isotropic =	46.4282	Anisotropy =	186.3705
6	C	Isotropic =	35.9429	Anisotropy =	168.0859
7	H	Isotropic =	23.5457	Anisotropy =	11.9150
8	H	Isotropic =	23.6874	Anisotropy =	8.5717
9	C	Isotropic =	43.1080	Anisotropy =	10.6489
10	H	Isotropic =	23.7493	Anisotropy =	8.0971
11	H	Isotropic =	23.5044	Anisotropy =	11.8923
12	I	Isotropic =	3133.0780	Anisotropy =	3153.0123
13	F	Isotropic =	261.0141	Anisotropy =	114.4206
14	F	Isotropic =	232.4075	Anisotropy =	148.7354
15	F	Isotropic =	232.6172	Anisotropy =	148.6238
16	N	Isotropic =	-93.6587	Anisotropy =	558.3359
17	C	Isotropic =	23.0600	Anisotropy =	180.7629
18	C	Isotropic =	50.1589	Anisotropy =	184.2541
19	C	Isotropic =	36.6918	Anisotropy =	213.3203
20	C	Isotropic =	50.1591	Anisotropy =	184.2534
21	C	Isotropic =	23.0602	Anisotropy =	180.7656
22	H	Isotropic =	22.6050	Anisotropy =	10.7852
23	H	Isotropic =	23.9240	Anisotropy =	5.3704
24	H	Isotropic =	23.4900	Anisotropy =	4.8363
25	H	Isotropic =	23.9242	Anisotropy =	5.3706
26	H	Isotropic =	22.6049	Anisotropy =	10.7873

=====

X = I R = CN

----- Energies and NBO populations -----

Number of atoms: 24
Full point group C2V NOp 4

C	0.000000	-0.503559	0.000000
C	0.000000	-1.200815	1.214823
C	0.000000	-2.593817	1.216830
C	0.000000	-3.296618	0.000000
C	0.000000	-2.593817	-1.216830
C	0.000000	-1.200815	-1.214823
H	0.000000	-0.663448	2.156748
H	0.000000	-3.137523	2.155609
C	0.000000	-4.729802	0.000000
H	0.000000	-3.137523	-2.155609
H	0.000000	-0.663448	-2.156748
I	0.000000	1.624639	0.000000
N	0.000000	-5.888669	0.000000
N	0.000000	4.667512	0.000000
C	1.147356	5.359101	0.000000
C	1.200490	6.754271	0.000000

C	0.000000	7.466042	0.000000
C	-1.200490	6.754271	0.000000
C	-1.147356	5.359101	0.000000
H	2.061216	4.769007	0.000000
H	2.158608	7.263314	0.000000
H	0.000000	8.551787	0.000000
H	-2.158608	7.263314	0.000000
H	-2.061216	4.769007	0.000000

Electronic energy:	-7491.40422390 (-7491.40422390)
Zero-point correction=	0.176882
Sum of electronic and zero-point Energies=	-7491.227342
Sum of electronic and thermal Energies=	-7491.213051
Sum of electronic and thermal Enthalpies=	-7491.212107
Sum of electronic and thermal Free Energies=	-7491.274551

Lowest vibration frequency: 12.6196

Counterpoise: BSSE energy = 0.000697647821

Summary of Natural Population Analysis:

		Natural Population				
Atom	No	Natural Charge	Core	Valence	Rydberg	Total
C	1	-0.20958	1.99878	4.18304	0.02776	6.20958
C	2	-0.22797	1.99900	4.20889	0.02008	6.22797
C	3	-0.14188	1.99910	4.12714	0.01564	6.14188
C	4	-0.19350	1.99888	4.17757	0.01705	6.19350
C	5	-0.14188	1.99910	4.12714	0.01564	6.14188
C	6	-0.22797	1.99900	4.20889	0.02008	6.22797
H	7	0.23216	0.00000	0.76633	0.00151	0.76784
H	8	0.23201	0.00000	0.76671	0.00128	0.76799
C	9	0.30785	1.99927	3.65885	0.03403	5.69215
H	10	0.23201	0.00000	0.76671	0.00128	0.76799
H	11	0.23216	0.00000	0.76633	0.00151	0.76784
I	12	0.21701	45.99809	6.76161	0.02329	52.78299
N	13	-0.34827	1.99959	5.32846	0.02022	7.34827
N	14	-0.48653	1.99941	5.46467	0.02246	7.48653
C	15	0.05598	1.99922	3.91845	0.02635	5.94402
C	16	-0.24468	1.99917	4.22926	0.01625	6.24468
C	17	-0.16429	1.99918	4.14780	0.01731	6.16429
C	18	-0.24468	1.99917	4.22926	0.01625	6.24468
C	19	0.05598	1.99922	3.91845	0.02635	5.94402
H	20	0.19892	0.00000	0.79961	0.00147	0.80108
H	21	0.22395	0.00000	0.77458	0.00148	0.77605
H	22	0.22033	0.00000	0.77840	0.00126	0.77967
H	23	0.22395	0.00000	0.77458	0.00148	0.77605
H	24	0.19892	0.00000	0.79961	0.00147	0.80108
=====						
* Total *		0.00000	73.98618	73.68235	0.33148	148.00000

Contributions to pi population at N atom in pyridine:

Val(2p)	1.19804
Ryd(3p)	0.00082
Ryd(4p)	0.00039
Ryd(5p)	0.00002
Total:	1.19927

NMR properties

Electronic energy (NMR): -7491.40422390

1	C	Isotropic =	25.0253	Anisotropy =	166.4518
2	C	Isotropic =	35.9882	Anisotropy =	169.0891
3	C	Isotropic =	39.9301	Anisotropy =	187.0703
4	C	Isotropic =	63.5325	Anisotropy =	151.0618
5	C	Isotropic =	39.9301	Anisotropy =	187.0703
6	C	Isotropic =	35.9882	Anisotropy =	169.0891
7	H	Isotropic =	23.5507	Anisotropy =	11.8377
8	H	Isotropic =	23.7459	Anisotropy =	6.3628
9	C	Isotropic =	55.3911	Anisotropy =	353.1877
10	H	Isotropic =	23.7459	Anisotropy =	6.3628
11	H	Isotropic =	23.5507	Anisotropy =	11.8377

12	I	Isotropic =	3085.3909	Anisotropy =	3129.1333
13	N	Isotropic =	-43.0783	Anisotropy =	417.4651
14	N	Isotropic =	-92.8231	Anisotropy =	555.7247
15	C	Isotropic =	23.1796	Anisotropy =	180.7567
16	C	Isotropic =	50.0684	Anisotropy =	184.4227
17	C	Isotropic =	36.5759	Anisotropy =	213.6631
18	C	Isotropic =	50.0684	Anisotropy =	184.4227
19	C	Isotropic =	23.1796	Anisotropy =	180.7567
20	H	Isotropic =	22.6192	Anisotropy =	10.7435
21	H	Isotropic =	23.9175	Anisotropy =	5.3645
22	H	Isotropic =	23.4812	Anisotropy =	4.8570
23	H	Isotropic =	23.9175	Anisotropy =	5.3645
24	H	Isotropic =	22.6192	Anisotropy =	10.7435

=====

X = I R = NO2

Energies and NBO populations

Number of atoms: 25
Full point group C2V NOp 4

C	0.000000	-0.450351	0.000000
C	0.000000	-1.147316	1.216275
C	0.000000	-2.540412	1.220080
C	0.000000	-3.217085	0.000000
C	0.000000	-2.540412	-1.220080
C	0.000000	-1.147316	-1.216275
H	0.000000	-0.609487	2.157701
H	0.000000	-3.098819	2.148274
N	0.000000	-4.688947	0.000000
H	0.000000	-3.098819	-2.148274
H	0.000000	-0.609487	-2.157701
I	0.000000	1.676189	0.000000
O	0.000000	-5.263610	1.084693
O	0.000000	-5.263610	-1.084693
N	0.000000	4.703519	0.000000
C	1.147544	5.394844	0.000000
C	1.200540	6.789960	0.000000
C	0.000000	7.501641	0.000000
C	-1.200540	6.789960	0.000000
C	-1.147544	5.394844	0.000000
H	2.061356	4.804722	0.000000
H	2.158618	7.299028	0.000000
H	0.000000	8.587374	0.000000
H	-2.158618	7.299028	0.000000
H	-2.061356	4.804722	0.000000

Electronic energy:	-7603.67386110 (-7603.67386110)
Zero-point correction=	0.180566
Sum of electronic and zero-point Energies=	-7603.493295
Sum of electronic and thermal Energies=	-7603.478201
Sum of electronic and thermal Enthalpies=	-7603.477257
Sum of electronic and thermal Free Energies=	-7603.542469

Lowest vibration frequency: 12.5144

Counterpoise: BSSE energy = 0.000715535538

Summary of Natural Population Analysis:

		Natural Population				
Atom	No	Natural Charge	Core	Valence	Rydberg	Total
C	1	-0.20371	1.99879	4.17709	0.02783	6.20371
C	2	-0.22506	1.99901	4.20603	0.02003	6.22506
C	3	-0.17426	1.99907	4.15669	0.01851	6.17426
C	4	0.06044	1.99879	3.91993	0.02084	5.93956
C	5	-0.17426	1.99907	4.15669	0.01851	6.17426
C	6	-0.22506	1.99901	4.20603	0.02003	6.22506
H	7	0.23335	0.00000	0.76518	0.00147	0.76665
H	8	0.25133	0.00000	0.74678	0.00188	0.74867
N	9	0.48921	1.99945	4.45637	0.05498	6.51079
H	10	0.25133	0.00000	0.74678	0.00188	0.74867

H	11	0.23335	0.00000	0.76518	0.00147	0.76665
I	12	0.22758	45.99809	6.75074	0.02360	52.77242
O	13	-0.39197	1.99980	6.37723	0.01494	8.39197
O	14	-0.39197	1.99980	6.37723	0.01494	8.39197
N	15	-0.48774	1.99941	5.46587	0.02246	7.48774
C	16	0.05631	1.99922	3.91817	0.02629	5.94369
C	17	-0.24429	1.99917	4.22889	0.01623	6.24429
C	18	-0.16377	1.99918	4.14729	0.01730	6.16377
C	19	-0.24429	1.99917	4.22889	0.01623	6.24429
C	20	0.05631	1.99922	3.91817	0.02629	5.94369
H	21	0.19909	0.00000	0.79945	0.00146	0.80091
H	22	0.22421	0.00000	0.77432	0.00147	0.77579
H	23	0.22057	0.00000	0.77816	0.00126	0.77943
H	24	0.22421	0.00000	0.77432	0.00147	0.77579
H	25	0.19909	0.00000	0.79945	0.00146	0.80091
=====						
* Total *		0.00000	75.98624	81.64092	0.37285	158.00000

Contributions to pi population at N atom in pyridine:

Val(2p) 1.19971
Ryd(3p) 0.00083
Ryd(4p) 0.00039
Ryd(5p) 0.00002

Total: 1.20095

NMR properties

Electronic energy (NMR): -7603.67386110

1	C	Isotropic =	19.7682	Anisotropy =	173.6822
2	C	Isotropic =	36.2578	Anisotropy =	165.0509
3	C	Isotropic =	49.7180	Anisotropy =	182.1551
4	C	Isotropic =	25.0905	Anisotropy =	117.4083
5	C	Isotropic =	49.7180	Anisotropy =	182.1551
6	C	Isotropic =	36.2578	Anisotropy =	165.0509
7	H	Isotropic =	23.4942	Anisotropy =	11.9803
8	H	Isotropic =	23.0071	Anisotropy =	6.0738
9	N	Isotropic =	-155.0186	Anisotropy =	291.6879
10	H	Isotropic =	23.0071	Anisotropy =	6.0738
11	H	Isotropic =	23.4942	Anisotropy =	11.9803
12	I	Isotropic =	3067.4532	Anisotropy =	3075.1417
13	O	Isotropic =	-307.6753	Anisotropy =	723.9530
14	O	Isotropic =	-307.6753	Anisotropy =	723.9530
15	N	Isotropic =	-92.1983	Anisotropy =	554.0301
16	C	Isotropic =	23.2568	Anisotropy =	180.7688
17	C	Isotropic =	50.0021	Anisotropy =	184.6005
18	C	Isotropic =	36.4737	Anisotropy =	213.8564
19	C	Isotropic =	50.0021	Anisotropy =	184.6005
20	C	Isotropic =	23.2568	Anisotropy =	180.7688
21	H	Isotropic =	22.6154	Anisotropy =	10.7242
22	H	Isotropic =	23.9062	Anisotropy =	5.3473
23	H	Isotropic =	23.4713	Anisotropy =	4.8519
24	H	Isotropic =	23.9062	Anisotropy =	5.3473
25	H	Isotropic =	22.6154	Anisotropy =	10.7242

=====

X = Br R = H

Energies and NBO populations

Number of atoms: 23
Full point group C2V NOp 4

C	0.000000	-0.568430	0.000000
C	0.000000	-1.255166	1.216061
C	0.000000	-2.653139	1.208412
C	0.000000	-3.354816	0.000000
C	0.000000	-2.653139	-1.208412
C	0.000000	-1.255166	-1.216061
H	0.000000	-0.708191	2.152621
H	0.000000	-3.190401	2.152157
H	0.000000	-4.440314	0.000000
H	0.000000	-3.190401	-2.152157

H	0.000000	-0.708191	-2.152621
Br	0.000000	1.349170	0.000000
N	0.000000	4.496990	0.000000
C	1.144942	5.192448	0.000000
C	1.199851	6.588363	0.000000
C	0.000000	7.301117	0.000000
C	-1.199851	6.588363	0.000000
C	-1.144942	5.192448	0.000000
H	2.060403	4.604414	0.000000
H	2.158409	7.097040	0.000000
H	0.000000	8.387014	0.000000
H	-2.158409	7.097040	0.000000
H	-2.060403	4.604414	0.000000

Electronic energy:	-3053.88644195 (-3053.88644195)
Zero-point correction=	0.178365
Sum of electronic and zero-point Energies=	-3053.708077
Sum of electronic and thermal Energies=	-3053.696487
Sum of electronic and thermal Enthalpies=	-3053.695543
Sum of electronic and thermal Free Energies=	-3053.751489

Lowest vibration frequency: -2.0199

Counterpoise: BSSE energy = 0.000457795755

Summary of Natural Population Analysis:

		Natural Population				
Atom	No	Natural Charge	Core	Valence	Rydberg	Total
C	1	-0.12216	1.99867	4.10029	0.02319	6.12216
C	2	-0.23490	1.99898	4.21652	0.01941	6.23490
C	3	-0.19539	1.99917	4.17849	0.01773	6.19539
C	4	-0.21535	1.99918	4.19851	0.01767	6.21535
C	5	-0.19539	1.99917	4.17849	0.01773	6.19539
C	6	-0.23490	1.99898	4.21652	0.01941	6.23490
H	7	0.22718	0.00000	0.77121	0.00161	0.77282
H	8	0.21674	0.00000	0.78184	0.00142	0.78326
H	9	0.21567	0.00000	0.78300	0.00133	0.78433
H	10	0.21674	0.00000	0.78184	0.00142	0.78326
H	11	0.22718	0.00000	0.77121	0.00161	0.77282
Br	12	0.07950	27.99929	6.89911	0.02211	34.92050
N	13	-0.47447	1.99944	5.45263	0.02241	7.47447
C	14	0.05160	1.99923	3.92199	0.02718	5.94840
C	15	-0.24863	1.99917	4.23302	0.01644	6.24863
C	16	-0.16948	1.99918	4.15287	0.01744	6.16948
C	17	-0.24863	1.99917	4.23302	0.01644	6.24863
C	18	0.05160	1.99923	3.92199	0.02718	5.94840
H	19	0.19606	0.00000	0.80248	0.00146	0.80394
H	20	0.22143	0.00000	0.77708	0.00149	0.77857
H	21	0.21812	0.00000	0.78059	0.00129	0.78188
H	22	0.22143	0.00000	0.77708	0.00149	0.77857
H	23	0.19606	0.00000	0.80248	0.00146	0.80394
=====						
* Total *		0.00000	51.98884	65.73225	0.27891	118.00000

Contributions to pi population at N atom in pyridine:

Val(2p)	1.17833
Ryd(3p)	0.00073
Ryd(4p)	0.00034
Ryd(5p)	0.00002

Total: 1.17942

NMR properties

Electronic energy (NMR): -3053.88644195

1	C	Isotropic =	29.0803	Anisotropy =	133.8628
2	C	Isotropic =	41.2152	Anisotropy =	163.5715
3	C	Isotropic =	44.1792	Anisotropy =	187.0894
4	C	Isotropic =	47.6054	Anisotropy =	191.9693
5	C	Isotropic =	44.1792	Anisotropy =	187.0894
6	C	Isotropic =	41.2152	Anisotropy =	163.5715
7	H	Isotropic =	23.7302	Anisotropy =	9.8219

8	H	Isotropic =	23.9203	Anisotropy =	5.0113
9	H	Isotropic =	23.9892	Anisotropy =	4.1870
10	H	Isotropic =	23.9203	Anisotropy =	5.0113
11	H	Isotropic =	23.7302	Anisotropy =	9.8219
12	Br	Isotropic =	1939.1207	Anisotropy =	1467.8390
13	N	Isotropic =	-99.5788	Anisotropy =	578.6408
14	C	Isotropic =	21.8922	Anisotropy =	180.9487
15	C	Isotropic =	50.6225	Anisotropy =	183.4124
16	C	Isotropic =	37.5102	Anisotropy =	211.8561
17	C	Isotropic =	50.6225	Anisotropy =	183.4124
18	C	Isotropic =	21.8922	Anisotropy =	180.9487
19	H	Isotropic =	22.5018	Anisotropy =	9.4377
20	H	Isotropic =	23.9486	Anisotropy =	5.3271
21	H	Isotropic =	23.5209	Anisotropy =	4.9419
22	H	Isotropic =	23.9486	Anisotropy =	5.3271
23	H	Isotropic =	22.5018	Anisotropy =	9.4377

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X = Br R = OCH3

Energies and NBO populations

Number of atoms: 27
Full point group CS NOp 2

C	0.000000	-0.414219	-0.024155
C	0.000000	-1.158280	1.152084
C	0.000000	-2.557858	1.098366
C	0.000000	-3.208408	-0.142203
C	0.000000	-2.448188	-1.322856
C	0.000000	-1.058364	-1.266519
H	0.000000	-0.659549	2.115329
H	0.000000	-3.117316	2.026333
O	0.000000	-4.560453	-0.305409
H	0.000000	-2.963953	-2.277721
H	0.000000	-0.478958	-2.183419
Br	0.000000	1.501625	0.054654
C	0.000000	-5.378357	0.858375
H	0.000000	-6.407658	0.499034
H	-0.896132	-5.205061	1.466634
H	0.896132	-5.205061	1.466634
N	0.000000	4.659423	0.169458
C	1.144871	5.354957	0.169458
C	1.199836	6.750906	0.169758
C	0.000000	7.463689	0.169889
C	-1.199836	6.750906	0.169758
C	-1.144871	5.354957	0.169458
H	2.060313	4.766881	0.168909
H	2.158406	7.259589	0.169581
H	0.000000	8.549595	0.169888
H	-2.158406	7.259589	0.169581
H	-2.060313	4.766881	0.168909

Electronic energy:	-3168.40894100 (-3168.40894108)
Zero-point correction=	0.210768
Sum of electronic and zero-point Energies=	-3168.198173
Sum of electronic and thermal Energies=	-3168.183089
Sum of electronic and thermal Enthalpies=	-3168.182144
Sum of electronic and thermal Free Energies=	-3168.248143

Lowest vibration frequency: 4.7384

Counterpoise: BSSE energy = 0.000453931277

Summary of Natural Population Analysis:

Atom	No	Natural Charge	Natural Population			
			Core	Valence	Rydberg	Total
C	1	-0.15140	1.99865	4.13000	0.02275	6.15140
C	2	-0.21354	1.99899	4.19514	0.01940	6.21354
C	3	-0.28474	1.99907	4.27057	0.01509	6.28474
C	4	0.31920	1.99881	3.65899	0.02300	5.68080
C	5	-0.23496	1.99908	4.21876	0.01712	6.23496

C	6	-0.21625	1.99899	4.19834	0.01892	6.21625
H	7	0.22670	0.00000	0.77164	0.00166	0.77330
H	8	0.22366	0.00000	0.77454	0.00181	0.77634
O	9	-0.54411	1.99971	6.52659	0.01780	8.54411
H	10	0.22775	0.00000	0.77048	0.00178	0.77225
H	11	0.22780	0.00000	0.77064	0.00156	0.77220
Br	12	0.07289	27.99929	6.90555	0.02227	34.92711
C	13	-0.21371	1.99928	4.19871	0.01572	6.21371
H	14	0.19642	0.00000	0.80256	0.00101	0.80358
H	15	0.17497	0.00000	0.82293	0.00210	0.82503
H	16	0.17497	0.00000	0.82293	0.00210	0.82503
N	17	-0.47376	1.99944	5.45185	0.02247	7.47376
C	18	0.05147	1.99923	3.92206	0.02724	5.94853
C	19	-0.24884	1.99917	4.23322	0.01645	6.24884
C	20	-0.16976	1.99918	4.15314	0.01744	6.16976
C	21	-0.24884	1.99917	4.23322	0.01645	6.24884
C	22	0.05147	1.99923	3.92206	0.02724	5.94853
H	23	0.19604	0.00000	0.80250	0.00146	0.80396
H	24	0.22128	0.00000	0.77723	0.00149	0.77872
H	25	0.21797	0.00000	0.78073	0.00129	0.78203
H	26	0.22128	0.00000	0.77723	0.00149	0.77872
H	27	0.19604	0.00000	0.80250	0.00146	0.80396
=====						
* Total *		0.00000	55.98729	77.69413	0.31857	134.00000

Contributions to pi population at N atom in pyridine:

Val(2p)	1.17756
Ryd(3p)	0.00073
Ryd(4p)	0.00033
Ryd(5p)	0.00002

Total: 1.17864

NMR properties

Electronic energy (NMR): -3168.40894100

1	C	Isotropic =	41.2234	Anisotropy =	115.6101
2	C	Isotropic =	40.7702	Anisotropy =	161.7672
3	C	Isotropic =	65.5272	Anisotropy =	160.4591
4	C	Isotropic =	12.3273	Anisotropy =	146.3826
5	C	Isotropic =	56.1651	Anisotropy =	149.9131
6	C	Isotropic =	40.2763	Anisotropy =	161.2150
7	H	Isotropic =	23.8453	Anisotropy =	9.7712
8	H	Isotropic =	24.5539	Anisotropy =	7.6159
9	O	Isotropic =	223.2488	Anisotropy =	84.6280
10	H	Isotropic =	24.3837	Anisotropy =	5.8321
11	H	Isotropic =	23.9160	Anisotropy =	9.3721
12	Br	Isotropic =	1982.9693	Anisotropy =	1452.3829
13	C	Isotropic =	124.5430	Anisotropy =	75.5031
14	H	Isotropic =	27.4854	Anisotropy =	8.4007
15	H	Isotropic =	27.8837	Anisotropy =	8.0317
16	H	Isotropic =	27.8837	Anisotropy =	8.0317
17	N	Isotropic =	-99.9298	Anisotropy =	579.7036
18	C	Isotropic =	21.8408	Anisotropy =	180.8837
19	C	Isotropic =	50.6866	Anisotropy =	183.3014
20	C	Isotropic =	37.5617	Anisotropy =	211.6456
21	C	Isotropic =	50.6866	Anisotropy =	183.3014
22	C	Isotropic =	21.8408	Anisotropy =	180.8837
23	H	Isotropic =	22.5118	Anisotropy =	9.5107
24	H	Isotropic =	23.9606	Anisotropy =	5.3449
25	H	Isotropic =	23.5329	Anisotropy =	4.9292
26	H	Isotropic =	23.9606	Anisotropy =	5.3449
27	H	Isotropic =	22.5118	Anisotropy =	9.5107

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X = Br R = OCH2CH3

Energies and NBO populations

Number of atoms: 30
Full point group CS NOp 2

C	0.000000	-0.312698	-0.053010
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C	0.000000	-1.058061	1.122391
C	0.000000	-2.457599	1.067582
C	0.000000	-3.108252	-0.173640
C	0.000000	-2.345501	-1.353152
C	0.000000	-0.955671	-1.295840
H	0.000000	-0.560273	2.086149
H	0.000000	-3.017111	1.995376
O	0.000000	-4.459225	-0.339247
H	0.000000	-2.860057	-2.308709
H	0.000000	-0.375542	-2.212301
Br	0.000000	1.603396	0.027688
C	0.000000	-5.293293	0.823453
C	0.000000	-6.739035	0.355852
H	-0.890389	-5.077746	1.429736
H	0.890389	-5.077746	1.429736
H	0.000000	-7.407316	1.222870
H	0.888631	-6.951787	-0.245576
H	-0.888631	-6.951787	-0.245576
N	0.000000	4.763546	0.143309
C	1.144879	5.459034	0.143309
C	1.199856	6.855016	0.143491
C	0.000000	7.567771	0.143568
C	-1.199856	6.855016	0.143491
C	-1.144879	5.459034	0.143309
H	2.060348	4.871007	0.142822
H	2.158403	7.363731	0.143292
H	0.000000	8.653684	0.143533
H	-2.158403	7.363731	0.143292
H	-2.060348	4.871007	0.142822

Electronic energy:	-3207.72448027 (-3207.72446349)
Zero-point correction=	0.238755
Sum of electronic and zero-point Energies=	-3207.485725
Sum of electronic and thermal Energies=	-3207.469220
Sum of electronic and thermal Enthalpies=	-3207.468276
Sum of electronic and thermal Free Energies=	-3207.537645

Lowest vibration frequency: 5.3586

Counterpoise: BSSE energy = 0.000442848601

Summary of Natural Population Analysis:

		Natural Population				
Atom	No	Natural Charge	Core	Valence	Rydberg	Total
C	1	-0.15163	1.99864	4.13033	0.02266	6.15163
C	2	-0.21383	1.99899	4.19555	0.01928	6.21383
C	3	-0.28435	1.99907	4.26997	0.01530	6.28435
C	4	0.32217	1.99882	3.65604	0.02297	5.67783
C	5	-0.23463	1.99909	4.21878	0.01676	6.23463
C	6	-0.21676	1.99899	4.19891	0.01886	6.21676
H	7	0.22643	0.00000	0.77191	0.00166	0.77357
H	8	0.22313	0.00000	0.77501	0.00186	0.77687
O	9	-0.55778	1.99970	6.53721	0.02087	8.55778
H	10	0.22718	0.00000	0.77110	0.00173	0.77282
H	11	0.22753	0.00000	0.77091	0.00156	0.77247
Br	12	0.07182	27.99929	6.90665	0.02224	34.92818
C	13	-0.03610	1.99925	4.01823	0.01862	6.03610
C	14	-0.59417	1.99938	4.58622	0.00857	6.59417
H	15	0.17432	0.00000	0.82322	0.00246	0.82568
H	16	0.17432	0.00000	0.82322	0.00246	0.82568
H	17	0.21047	0.00000	0.78838	0.00115	0.78953
H	18	0.20883	0.00000	0.78949	0.00169	0.79117
H	19	0.20883	0.00000	0.78949	0.00169	0.79117
N	20	-0.47361	1.99944	5.45171	0.02246	7.47361
C	21	0.05146	1.99923	3.92207	0.02724	5.94854
C	22	-0.24888	1.99917	4.23325	0.01646	6.24888
C	23	-0.16984	1.99918	4.15322	0.01744	6.16984
C	24	-0.24888	1.99917	4.23325	0.01646	6.24888
C	25	0.05146	1.99923	3.92207	0.02724	5.94854
H	26	0.19604	0.00000	0.80250	0.00146	0.80396
H	27	0.22124	0.00000	0.77726	0.00150	0.77876
H	28	0.21794	0.00000	0.78077	0.00129	0.78206
H	29	0.22124	0.00000	0.77726	0.00150	0.77876
H	30	0.19604	0.00000	0.80250	0.00146	0.80396

```
=====
* Total *      0.00000      57.98664      83.67647      0.33689      142.00000
```

Contributions to pi population at N atom in pyridine:

```
Val( 2p)      1.17740
Ryd( 3p)      0.00073
Ryd( 4p)      0.00033
Ryd( 5p)      0.00002
```

```
Total:        1.17848
```

NMR properties

Electronic energy (NMR): -3207.72448027

1	C	Isotropic =	41.4395	Anisotropy =	114.8869
2	C	Isotropic =	40.8045	Anisotropy =	161.2611
3	C	Isotropic =	64.9359	Anisotropy =	158.4218
4	C	Isotropic =	12.7025	Anisotropy =	145.7050
5	C	Isotropic =	56.1547	Anisotropy =	148.8789
6	C	Isotropic =	40.3110	Anisotropy =	160.9023
7	H	Isotropic =	23.8745	Anisotropy =	9.7997
8	H	Isotropic =	24.5952	Anisotropy =	8.2315
9	O	Isotropic =	193.3481	Anisotropy =	70.2450
10	H	Isotropic =	24.3710	Anisotropy =	6.3233
11	H	Isotropic =	23.9169	Anisotropy =	9.3768
12	Br	Isotropic =	1983.7882	Anisotropy =	1452.9244
13	C	Isotropic =	113.5093	Anisotropy =	63.1869
14	C	Isotropic =	165.9502	Anisotropy =	25.6620
15	H	Isotropic =	27.6335	Anisotropy =	5.3802
16	H	Isotropic =	27.6335	Anisotropy =	5.3802
17	H	Isotropic =	30.2564	Anisotropy =	10.0870
18	H	Isotropic =	30.0324	Anisotropy =	8.2204
19	H	Isotropic =	30.0324	Anisotropy =	8.2204
20	N	Isotropic =	-99.9801	Anisotropy =	579.9339
21	C	Isotropic =	21.8279	Anisotropy =	180.8523
22	C	Isotropic =	50.6950	Anisotropy =	183.2913
23	C	Isotropic =	37.5633	Anisotropy =	211.5787
24	C	Isotropic =	50.6950	Anisotropy =	183.2913
25	C	Isotropic =	21.8279	Anisotropy =	180.8523
26	H	Isotropic =	22.5117	Anisotropy =	9.5410
27	H	Isotropic =	23.9620	Anisotropy =	5.3525
28	H	Isotropic =	23.5352	Anisotropy =	4.9212
29	H	Isotropic =	23.9620	Anisotropy =	5.3525
30	H	Isotropic =	22.5117	Anisotropy =	9.5410

=====

X = Br R = CH3

Energies and NBO populations

Number of atoms: 26
Full point group CS NOp 2

C	0.000000	-0.477013	-0.007327
C	0.000000	-1.111407	1.234037
C	0.000000	-2.509052	1.287760
C	0.000000	-3.285847	0.122256
C	0.000000	-2.618554	-1.112743
C	0.000000	-1.224866	-1.187215
H	0.000000	-0.527730	2.148336
H	0.000000	-2.997923	2.258599
C	0.000000	-4.797133	0.183878
H	0.000000	-3.194796	-2.034886
H	0.000000	-0.726538	-2.150746
Br	0.000000	1.438743	-0.096918
H	0.000000	-5.151161	1.218329
H	-0.882698	-5.212372	-0.315637
H	0.882698	-5.212372	-0.315637
N	0.000000	4.597416	-0.218676
C	1.144873	5.292968	-0.218676
C	1.199835	6.688912	-0.218835
C	0.000000	7.401694	-0.218884
C	-1.199835	6.688912	-0.218835

C	-1.144873	5.292968	-0.218676
H	2.060344	4.704939	-0.218271
H	2.158404	7.197591	-0.218697
H	0.000000	8.487599	-0.218816
H	-2.158404	7.197591	-0.218697
H	-2.060344	4.704939	-0.218271

Electronic energy:	-3093.20023672 (-3093.20023677)
Zero-point correction=	0.205418
Sum of electronic and zero-point Energies=	-3092.994819
Sum of electronic and thermal Energies=	-3092.981325
Sum of electronic and thermal Enthalpies=	-3092.980381
Sum of electronic and thermal Free Energies=	-3093.041703

Lowest vibration frequency: -3.6078

Counterpoise: BSSE energy = 0.000418767595

Summary of Natural Population Analysis:

		Natural Population				
Atom	No	Natural Charge	Core	Valence	Rydberg	Total
C	1	-0.12937	1.99865	4.10793	0.02278	6.12937
C	2	-0.22793	1.99898	4.20932	0.01962	6.22793
C	3	-0.19788	1.99906	4.18234	0.01648	6.19788
C	4	-0.03695	1.99907	4.02253	0.01535	6.03695
C	5	-0.19536	1.99906	4.17972	0.01658	6.19536
C	6	-0.22671	1.99898	4.20823	0.01949	6.22671
H	7	0.22620	0.00000	0.77210	0.00170	0.77380
H	8	0.21323	0.00000	0.78513	0.00164	0.78677
C	9	-0.59682	1.99932	4.58751	0.00999	6.59682
H	10	0.21335	0.00000	0.78506	0.00159	0.78665
H	11	0.22610	0.00000	0.77220	0.00170	0.77390
Br	12	0.07446	27.99929	6.90405	0.02221	34.92554
H	13	0.21019	0.00000	0.78852	0.00129	0.78981
H	14	0.21652	0.00000	0.78214	0.00134	0.78348
H	15	0.21652	0.00000	0.78214	0.00134	0.78348
N	16	-0.47386	1.99944	5.45201	0.02241	7.47386
C	17	0.05150	1.99923	3.92206	0.02721	5.94850
C	18	-0.24881	1.99917	4.23319	0.01645	6.24881
C	19	-0.16972	1.99918	4.15310	0.01744	6.16972
C	20	-0.24881	1.99917	4.23319	0.01645	6.24881
C	21	0.05150	1.99923	3.92206	0.02721	5.94850
H	22	0.19602	0.00000	0.80252	0.00146	0.80398
H	23	0.22130	0.00000	0.77721	0.00149	0.77870
H	24	0.21799	0.00000	0.78071	0.00129	0.78201
H	25	0.22130	0.00000	0.77721	0.00149	0.77870
H	26	0.19602	0.00000	0.80252	0.00146	0.80398
=====						
* Total *		0.00000	53.98782	71.72469	0.28749	126.00000

Contributions to pi population at N atom in pyridine:

Val(2p)	1.17764
Ryd(3p)	0.00073
Ryd(4p)	0.00034
Ryd(5p)	0.00002
Total:	1.17873

NMR properties

Electronic energy (NMR): -3093.20023672

1	C	Isotropic =	33.3237	Anisotropy =	126.8832
2	C	Isotropic =	41.5298	Anisotropy =	159.8372
3	C	Isotropic =	43.4765	Anisotropy =	161.4204
4	C	Isotropic =	34.6683	Anisotropy =	205.6111
5	C	Isotropic =	43.3582	Anisotropy =	172.3976
6	C	Isotropic =	41.6428	Anisotropy =	160.3252
7	H	Isotropic =	23.9176	Anisotropy =	9.6356
8	H	Isotropic =	24.1539	Anisotropy =	6.5537
9	C	Isotropic =	159.6196	Anisotropy =	41.4511
10	H	Isotropic =	24.0187	Anisotropy =	7.0995
11	H	Isotropic =	23.8509	Anisotropy =	9.6567

=====

X = Br R = CHCH32

Energies and NBO populations

=====

C	0.000000	-0.330631	0.048206
C	0.000000	-1.071106	1.229488
C	0.000000	-2.467901	1.159121
C	0.000000	-3.141782	-0.069240
C	0.000000	-2.367707	-1.240688
C	0.000000	-0.972665	-1.192131
H	0.000000	-0.569834	2.191491
H	0.000000	-3.040493	2.083525
C	0.000000	-4.664920	-0.123211
H	0.000000	-2.853207	-2.213072
H	0.000000	-0.391696	-2.108266
Br	0.000000	1.585759	0.123981
H	0.000000	-5.020519	0.915481
C	-1.271447	-5.212984	-0.803214
C	1.271447	-5.212984	-0.803214
H	-1.278515	-6.308482	-0.775156
H	-1.322423	-4.905046	-1.853989
H	-2.175185	-4.851188	-0.302170
H	1.278515	-6.308482	-0.775156
H	2.175185	-4.851188	-0.302170
H	1.322423	-4.905046	-1.853989
N	0.000000	4.747277	0.212694
C	1.144857	5.442869	0.212694
C	1.199820	6.838823	0.212793
C	0.000000	7.551627	0.212946
C	-1.199820	6.838823	0.212793
C	-1.144857	5.442869	0.212694
H	2.060340	4.854868	0.212402
H	2.158405	7.347484	0.212670
H	0.000000	8.637530	0.213070
H	-2.158405	7.347484	0.212670
H	-2.060340	4.854868	0.212402

Lowest vibration frequency: -7.3245
Counterpoise: BSSE energy = 0.000410560388
Summary of Natural Population Analysis:

S118

C	2	-0.22815	1.99898	4.20966	0.01951	6.22815
C	3	-0.19496	1.99906	4.17947	0.01643	6.19496
C	4	-0.02320	1.99902	4.00408	0.02010	6.02320
C	5	-0.19749	1.99905	4.18227	0.01617	6.19749
C	6	-0.22714	1.99898	4.20855	0.01961	6.22714
H	7	0.22631	0.00000	0.77203	0.00166	0.77369
H	8	0.21352	0.00000	0.78474	0.00174	0.78648
C	9	-0.24333	1.99920	4.22846	0.01567	6.24333
H	10	0.21371	0.00000	0.78455	0.00174	0.78629
H	11	0.22583	0.00000	0.77251	0.00166	0.77417
Br	12	0.07461	27.99929	6.90395	0.02216	34.92539
H	13	0.19852	0.00000	0.79847	0.00301	0.80148
C	14	-0.57022	1.99935	4.56140	0.00947	6.57022
C	15	-0.57022	1.99935	4.56140	0.00947	6.57022
H	16	0.20305	0.00000	0.79575	0.00120	0.79695
H	17	0.19763	0.00000	0.80101	0.00137	0.80237
H	18	0.20739	0.00000	0.79159	0.00101	0.79261
H	19	0.20305	0.00000	0.79575	0.00120	0.79695
H	20	0.20739	0.00000	0.79159	0.00101	0.79261
H	21	0.19763	0.00000	0.80101	0.00137	0.80237
N	22	-0.47378	1.99944	5.45193	0.02241	7.47378
C	23	0.05146	1.99923	3.92208	0.02723	5.94854
C	24	-0.24882	1.99917	4.23320	0.01645	6.24882
C	25	-0.16974	1.99918	4.15312	0.01744	6.16974
C	26	-0.24882	1.99917	4.23320	0.01645	6.24882
C	27	0.05146	1.99923	3.92208	0.02723	5.94854
H	28	0.19602	0.00000	0.80252	0.00146	0.80398
H	29	0.22129	0.00000	0.77722	0.00149	0.77871
H	30	0.21799	0.00000	0.78072	0.00129	0.78201
H	31	0.22129	0.00000	0.77722	0.00149	0.77871
H	32	0.19602	0.00000	0.80252	0.00146	0.80398
=====						
* Total *		0.00000	57.98635	83.69079	0.32287	142.00000

Contributions to pi population at N atom in pyridine:

Val(2p) 1.17756
Ryd(3p) 0.00073
Ryd(4p) 0.00034
Ryd(5p) 0.00002

Total: 1.17865

NMR properties

Electronic energy (NMR): -3171.82081334

1	C	Isotropic =	32.8472	Anisotropy =	125.8137
2	C	Isotropic =	41.4597	Anisotropy =	159.2673
3	C	Isotropic =	44.5323	Anisotropy =	156.1379
4	C	Isotropic =	22.9064	Anisotropy =	214.5764
5	C	Isotropic =	47.1852	Anisotropy =	185.3806
6	C	Isotropic =	41.2384	Anisotropy =	160.9324
7	H	Isotropic =	23.8975	Anisotropy =	9.4498
8	H	Isotropic =	24.1485	Anisotropy =	7.2185
9	C	Isotropic =	139.6199	Anisotropy =	24.5509
10	H	Isotropic =	23.9925	Anisotropy =	9.5436
11	H	Isotropic =	23.8162	Anisotropy =	9.8788
12	Br	Isotropic =	1959.8952	Anisotropy =	1458.3613
13	H	Isotropic =	28.7371	Anisotropy =	6.9217
14	C	Isotropic =	155.7508	Anisotropy =	36.8608
15	C	Isotropic =	155.7508	Anisotropy =	36.8608
16	H	Isotropic =	30.2582	Anisotropy =	10.2055
17	H	Isotropic =	30.3331	Anisotropy =	6.5431
18	H	Isotropic =	30.3078	Anisotropy =	8.5172
19	H	Isotropic =	30.2582	Anisotropy =	10.2055
20	H	Isotropic =	30.3078	Anisotropy =	8.5172
21	H	Isotropic =	30.3331	Anisotropy =	6.5431
22	N	Isotropic =	-99.7967	Anisotropy =	579.7513
23	C	Isotropic =	21.8353	Anisotropy =	180.8599
24	C	Isotropic =	50.6791	Anisotropy =	183.2978
25	C	Isotropic =	37.5861	Anisotropy =	211.7039
26	C	Isotropic =	50.6791	Anisotropy =	183.2978
27	C	Isotropic =	21.8353	Anisotropy =	180.8599
28	H	Isotropic =	22.5093	Anisotropy =	9.5619
29	H	Isotropic =	23.9597	Anisotropy =	5.3487
30	H	Isotropic =	23.5325	Anisotropy =	4.9016

31 H Isotropic = 23.9597 Anisotropy = 5.3487
 32 H Isotropic = 22.5093 Anisotropy = 9.5619

=====

X = Br R = H

Energies and NBO populations

Number of atoms: 23
 Full point group C2V NOp 4

C	0.000000	-0.533417	0.000000
C	0.000000	-1.223190	1.214697
C	0.000000	-2.620369	1.217194
C	0.000000	-3.290044	0.000000
C	0.000000	-2.620369	-1.217194
C	0.000000	-1.223190	-1.214698
H	0.000000	-0.680227	2.153188
H	0.000000	-3.180688	2.145842
F	0.000000	-4.643773	0.000000
H	0.000000	-3.180688	-2.145842
H	0.000000	-0.680227	-2.153188
Br	0.000000	1.381934	0.000000
N	0.000000	4.506001	0.000000
C	1.145101	5.201305	0.000000
C	1.199878	6.597139	0.000000
C	0.000000	7.309837	0.000000
C	-1.199878	6.597139	0.000000
C	-1.145101	5.201305	0.000000
H	2.060514	4.613230	0.000000
H	2.158411	7.105830	0.000000
H	0.000000	8.395714	0.000000
H	-2.158411	7.105830	0.000000
H	-2.060514	4.613230	0.000000

Electronic energy:	-3153.13011375 (-3153.13011375)
Zero-point correction=	0.170141
Sum of electronic and zero-point Energies=	-3152.959973
Sum of electronic and thermal Energies=	-3152.946637
Sum of electronic and thermal Enthalpies=	-3152.945693
Sum of electronic and thermal Free Energies=	-3153.007568

Lowest vibration frequency: 5.4020

Counterpoise: BSSE energy = 0.000499154561

Summary of Natural Population Analysis:

Atom	No	Natural Charge	Natural Population			Total
			Core	Valence	Rydberg	
C	1	-0.13959	1.99867	4.11806	0.02286	6.13959
C	2	-0.21373	1.99900	4.19601	0.01872	6.21373
C	3	-0.25941	1.99906	4.24353	0.01683	6.25941
C	4	0.40292	1.99849	3.57215	0.02643	5.59708
C	5	-0.25941	1.99906	4.24353	0.01683	6.25941
C	6	-0.21372	1.99900	4.19601	0.01872	6.21372
H	7	0.23152	0.00000	0.76687	0.00161	0.76848
H	8	0.23530	0.00000	0.76312	0.00157	0.76470
F	9	-0.35452	1.99993	7.34656	0.00803	9.35452
H	10	0.23530	0.00000	0.76312	0.00157	0.76470
H	11	0.23152	0.00000	0.76687	0.00161	0.76848
Br	12	0.08748	27.99929	6.89091	0.02232	34.91252
N	13	-0.47600	1.99943	5.45407	0.02250	7.47600
C	14	0.05178	1.99923	3.92184	0.02715	5.94822
C	15	-0.24823	1.99917	4.23263	0.01643	6.24823
C	16	-0.16891	1.99918	4.15230	0.01743	6.16891
C	17	-0.24823	1.99917	4.23263	0.01643	6.24823
C	18	0.05178	1.99923	3.92184	0.02715	5.94822
H	19	0.19613	0.00000	0.80241	0.00146	0.80387
H	20	0.22173	0.00000	0.77678	0.00149	0.77827
H	21	0.21843	0.00000	0.78028	0.00129	0.78157
H	22	0.22173	0.00000	0.77678	0.00149	0.77827
H	23	0.19613	0.00000	0.80241	0.00146	0.80387

```
=====
* Total *      0.00000      53.98789      71.72072      0.29139      126.00000
```

Contributions to pi population at N atom in pyridine:

```
Val( 2p)      1.17992
Ryd( 3p)      0.00073
Ryd( 4p)      0.00034
Ryd( 5p)      0.00002
```

```
Total:      1.18101
```

----- NMR properties -----

```
Electronic energy (NMR):      -3153.13011375
```

1	C	Isotropic =	35.7256	Anisotropy =	128.4925
2	C	Isotropic =	39.7298	Anisotropy =	163.1044
3	C	Isotropic =	58.2448	Anisotropy =	153.9198
4	C	Isotropic =	7.3155	Anisotropy =	106.5773
5	C	Isotropic =	58.2448	Anisotropy =	153.9198
6	C	Isotropic =	39.7298	Anisotropy =	163.1044
7	H	Isotropic =	23.8026	Anisotropy =	9.4074
8	H	Isotropic =	24.2712	Anisotropy =	4.6345
9	F	Isotropic =	294.5860	Anisotropy =	99.1838
10	H	Isotropic =	24.2712	Anisotropy =	4.6345
11	H	Isotropic =	23.8026	Anisotropy =	9.4074
12	Br	Isotropic =	1951.8230	Anisotropy =	1468.1497
13	N	Isotropic =	-98.8562	Anisotropy =	576.9210
14	C	Isotropic =	21.9792	Anisotropy =	180.8271
15	C	Isotropic =	50.5665	Anisotropy =	183.5619
16	C	Isotropic =	37.4101	Anisotropy =	212.0478
17	C	Isotropic =	50.5665	Anisotropy =	183.5619
18	C	Isotropic =	21.9792	Anisotropy =	180.8271
19	H	Isotropic =	22.5229	Anisotropy =	9.4820
20	H	Isotropic =	23.9483	Anisotropy =	5.3336
21	H	Isotropic =	23.5197	Anisotropy =	4.9424
22	H	Isotropic =	23.9483	Anisotropy =	5.3336
23	H	Isotropic =	22.5229	Anisotropy =	9.4820

```
=====
X = Br      R = Cl
```

----- Energies and NBO populations -----

```
Number of atoms: 23
Full point group      C2V      NOp      4
```

C	0.000000	-0.494724	0.000000
C	0.000000	-1.185303	1.213717
C	0.000000	-2.582160	1.215355
C	0.000000	-3.265320	0.000000
C	0.000000	-2.582160	-1.215355
C	0.000000	-1.185303	-1.213717
H	0.000000	-0.644074	2.153359
H	0.000000	-3.128717	2.151866
Cl	0.000000	-5.027014	0.000000
H	0.000000	-3.128717	-2.151866
H	0.000000	-0.644074	-2.153359
Br	0.000000	1.419123	0.000000
N	0.000000	4.535620	0.000000
C	1.145156	5.230877	0.000000
C	1.199889	6.626687	0.000000
C	0.000000	7.339361	0.000000
C	-1.199889	6.626687	0.000000
C	-1.145156	5.230877	0.000000
H	2.060575	4.642825	0.000000
H	2.158412	7.135380	0.000000
H	0.000000	8.425232	0.000000
H	-2.158412	7.135380	0.000000
H	-2.060575	4.642825	0.000000

```
Electronic energy:      -3513.45247580 ( -3513.45247580)
Zero-point correction=      0.168823
Sum of electronic and zero-point Energies= -3513.283653
```

Sum of electronic and thermal Energies= -3513.269931
 Sum of electronic and thermal Enthalpies= -3513.268986
 Sum of electronic and thermal Free Energies= -3513.331944

Lowest vibration frequency: 6.4860

Counterpoise: BSSE energy = 0.000477108347

Summary of Natural Population Analysis:

		Natural Population				
Atom	No	Natural Charge	Core	Valence	Rydberg	Total
C	1	-0.12811	1.99867	4.10651	0.02293	6.12811
C	2	-0.21531	1.99898	4.19721	0.01912	6.21531
C	3	-0.21727	1.99898	4.19972	0.01857	6.21727
C	4	-0.04592	1.99860	4.02465	0.02266	6.04592
C	5	-0.21727	1.99898	4.19972	0.01857	6.21727
C	6	-0.21531	1.99898	4.19721	0.01912	6.21531
H	7	0.23277	0.00000	0.76560	0.00163	0.76723
H	8	0.23414	0.00000	0.76419	0.00167	0.76586
Cl	9	-0.00413	9.99962	6.98454	0.01997	17.00413
H	10	0.23414	0.00000	0.76419	0.00167	0.76586
H	11	0.23277	0.00000	0.76560	0.00163	0.76723
Br	12	0.09267	27.99929	6.88569	0.02236	34.90733
N	13	-0.47657	1.99943	5.45465	0.02248	7.47657
C	14	0.05188	1.99923	3.92177	0.02712	5.94812
C	15	-0.24808	1.99917	4.23249	0.01642	6.24808
C	16	-0.16871	1.99918	4.15211	0.01742	6.16871
C	17	-0.24808	1.99917	4.23249	0.01642	6.24808
C	18	0.05188	1.99923	3.92177	0.02712	5.94812
H	19	0.19616	0.00000	0.80238	0.00146	0.80384
H	20	0.22183	0.00000	0.77668	0.00149	0.77817
H	21	0.21852	0.00000	0.78019	0.00129	0.78148
H	22	0.22183	0.00000	0.77668	0.00149	0.77817
H	23	0.19616	0.00000	0.80238	0.00146	0.80384
=====						
* Total *		0.00000	61.98749	71.70845	0.30406	134.00000

Contributions to pi population at N atom in pyridine:

Val(2p) 1.18049
 Ryd(3p) 0.00073
 Ryd(4p) 0.00034
 Ryd(5p) 0.00002

Total: 1.18158

 NMR properties

Electronic energy (NMR): -3513.45247580

1	C	Isotropic =	31.2728	Anisotropy =	136.3778
2	C	Isotropic =	40.1508	Anisotropy =	160.6059
3	C	Isotropic =	44.0364	Anisotropy =	158.6931
4	C	Isotropic =	32.3766	Anisotropy =	121.9799
5	C	Isotropic =	44.0364	Anisotropy =	158.6931
6	C	Isotropic =	40.1508	Anisotropy =	160.6059
7	H	Isotropic =	23.8349	Anisotropy =	9.4769
8	H	Isotropic =	23.9887	Anisotropy =	7.3208
9	Cl	Isotropic =	687.0565	Anisotropy =	502.4561
10	H	Isotropic =	23.9887	Anisotropy =	7.3208
11	H	Isotropic =	23.8349	Anisotropy =	9.4769
12	Br	Isotropic =	1938.5272	Anisotropy =	1453.8918
13	N	Isotropic =	-98.5006	Anisotropy =	576.1537
14	C	Isotropic =	22.0110	Anisotropy =	180.7586
15	C	Isotropic =	50.5421	Anisotropy =	183.6230
16	C	Isotropic =	37.3601	Anisotropy =	212.0979
17	C	Isotropic =	50.5421	Anisotropy =	183.6230
18	C	Isotropic =	22.0110	Anisotropy =	180.7586
19	H	Isotropic =	22.5309	Anisotropy =	9.5279
20	H	Isotropic =	23.9480	Anisotropy =	5.3443
21	H	Isotropic =	23.5196	Anisotropy =	4.9376
22	H	Isotropic =	23.9480	Anisotropy =	5.3443
23	H	Isotropic =	22.5309	Anisotropy =	9.5279

=====

X = Br R = OCF3

Energies and NBO populations

Number of atoms: 27
Full point group C1 NOp 1

C	-0.126249	-0.371179	-0.051567
C	-0.573703	-1.056765	-1.183617
C	-0.548480	-2.452814	-1.198687
C	-0.069803	-3.132630	-0.082885
C	0.368090	-2.459558	1.053472
C	0.341572	-1.063549	1.068058
H	-0.941469	-0.511334	-2.045333
H	-0.897839	-3.005196	-2.064031
O	-0.107043	-4.541205	-0.070219
H	0.720916	-3.017601	1.913782
H	0.679834	-0.523391	1.945043
Br	-0.161528	1.541734	-0.031455
C	0.978254	-5.197724	-0.530290
F	2.095517	-4.917457	0.168998
F	0.738165	-6.504591	-0.426288
F	1.249847	-4.920896	-1.820924
N	-0.206661	4.632312	0.003213
C	-1.351951	5.327556	0.003213
C	-1.406540	6.723365	0.002936
C	-0.206669	7.435943	0.002978
C	0.993214	6.723359	0.003227
C	0.938589	5.327564	0.003213
H	-2.267633	4.739973	0.003535
H	-2.365022	7.232123	0.002814
H	-0.206731	8.521810	0.002875
H	1.951728	7.232076	0.003236
H	1.854234	4.739878	0.002919

Electronic energy:	-3466.16987472 (-3466.16984624)
Zero-point correction=	0.186972
Sum of electronic and zero-point Energies=	-3465.982903
Sum of electronic and thermal Energies=	-3465.965933
Sum of electronic and thermal Enthalpies=	-3465.964989
Sum of electronic and thermal Free Energies=	-3466.036571

Lowest vibration frequency: 8.2656

Counterpoise: BSSE energy = 0.000454462194

Summary of Natural Population Analysis:

		Natural Population				
Atom	No	Natural Charge	Core	Valence	Rydberg	Total
C	1	-0.12393	1.99868	4.10230	0.02295	6.12393
C	2	-0.21709	1.99899	4.19928	0.01882	6.21709
C	3	-0.21634	1.99903	4.20032	0.01698	6.21634
C	4	0.26082	1.99868	3.71723	0.02328	5.73918
C	5	-0.21599	1.99903	4.19994	0.01702	6.21599
C	6	-0.21713	1.99899	4.19931	0.01883	6.21713
H	7	0.23351	0.00000	0.76488	0.00161	0.76649
H	8	0.23702	0.00000	0.76128	0.00170	0.76298
O	9	-0.56407	1.99965	6.54722	0.01719	8.56407
H	10	0.23688	0.00000	0.76142	0.00170	0.76312
H	11	0.23351	0.00000	0.76488	0.00161	0.76649
Br	12	0.09776	27.99928	6.88059	0.02237	34.90224
C	13	1.28119	1.99964	2.64817	0.07099	4.71881
F	14	-0.35385	1.99991	7.34732	0.00662	9.35385
F	15	-0.33622	1.99991	7.32997	0.00634	9.33622
F	16	-0.35405	1.99991	7.34752	0.00661	9.35405
N	17	-0.47747	1.99943	5.45551	0.02253	7.47747
C	18	0.05219	1.99923	3.92151	0.02707	5.94781
C	19	-0.24785	1.99917	4.23226	0.01642	6.24785
C	20	-0.16835	1.99918	4.15176	0.01742	6.16835
C	21	-0.24777	1.99917	4.23219	0.01642	6.24777
C	22	0.05214	1.99923	3.92157	0.02707	5.94786

H	23	0.19614	0.00000	0.80241	0.00145	0.80386
H	24	0.22198	0.00000	0.77654	0.00149	0.77802
H	25	0.21869	0.00000	0.78002	0.00129	0.78131
H	26	0.22202	0.00000	0.77649	0.00149	0.77798
H	27	0.19625	0.00000	0.80229	0.00146	0.80375
=====						
* Total *		0.00000	61.98712	95.62417	0.38871	158.00000

Contributions to pi population at N atom in pyridine:

Val(2p)	1.18163
Ryd(3p)	0.00075
Ryd(4p)	0.00035
Ryd(5p)	0.00002

Total: 1.18275

----- NMR properties -----

Electronic energy (NMR): -3466.16987472

1	C	Isotropic =	29.5377	Anisotropy =	139.1897
2	C	Isotropic =	40.0842	Anisotropy =	163.0517
3	C	Isotropic =	49.2751	Anisotropy =	164.4337
4	C	Isotropic =	23.2880	Anisotropy =	112.7649
5	C	Isotropic =	49.2500	Anisotropy =	164.2909
6	C	Isotropic =	40.0754	Anisotropy =	163.0688
7	H	Isotropic =	23.6968	Anisotropy =	9.5398
8	H	Isotropic =	24.0285	Anisotropy =	6.9843
9	O	Isotropic =	167.9679	Anisotropy =	88.6697
10	H	Isotropic =	24.0339	Anisotropy =	6.8867
11	H	Isotropic =	23.6919	Anisotropy =	9.5274
12	Br	Isotropic =	1926.4558	Anisotropy =	1465.1125
13	C	Isotropic =	47.3860	Anisotropy =	10.8789
14	F	Isotropic =	240.2241	Anisotropy =	165.0153
15	F	Isotropic =	241.3086	Anisotropy =	170.1156
16	F	Isotropic =	239.9621	Anisotropy =	165.4027
17	N	Isotropic =	-98.2464	Anisotropy =	574.7700
18	C	Isotropic =	22.1848	Anisotropy =	180.9495
19	C	Isotropic =	50.4006	Anisotropy =	183.2865
20	C	Isotropic =	37.4914	Anisotropy =	212.6939
21	C	Isotropic =	50.4480	Anisotropy =	183.4432
22	C	Isotropic =	22.1509	Anisotropy =	181.1515
23	H	Isotropic =	22.5291	Anisotropy =	9.5014
24	H	Isotropic =	23.9407	Anisotropy =	5.3290
25	H	Isotropic =	23.5060	Anisotropy =	4.9418
26	H	Isotropic =	23.9341	Anisotropy =	5.3908
27	H	Isotropic =	22.5220	Anisotropy =	9.6543

=====

X = Br R = COOCH3

Energies and NBO populations -----

Number of atoms: 29
Full point group CS NOp 2

C	0.000000	-0.366483	0.016488
C	0.000000	-1.071911	1.223631
C	0.000000	-2.464790	1.197212
C	0.000000	-3.154879	-0.023469
C	0.000000	-2.431918	-1.225618
C	0.000000	-1.037668	-1.209488
H	0.000000	-0.539692	2.168318
H	0.000000	-3.030613	2.122903
C	0.000000	-4.647564	0.012271
H	0.000000	-2.958947	-2.172772
H	0.000000	-0.478530	-2.138519
Br	0.000000	1.545114	0.042113
O	0.000000	-5.191545	-1.218742
C	0.000000	-6.629045	-1.271893
H	0.000000	-6.879823	-2.331934
H	-0.891477	-7.028148	-0.782490
H	0.891477	-7.028148	-0.782490
O	0.000000	-5.307733	1.030156

N	0.000000	4.662142	0.073352
C	1.145180	5.357398	0.073352
C	1.199895	6.753203	0.073256
C	0.000000	7.465862	0.073180
C	-1.199895	6.753203	0.073256
C	-1.145180	5.357398	0.073352
H	2.060646	4.769428	0.073362
H	2.158406	7.261900	0.073241
H	0.000000	8.551728	0.073083
H	-2.158406	7.261900	0.073241
H	-2.060646	4.769428	0.073362

Electronic energy: -3281.77130994 (-3281.77130991)
 Zero-point correction= 0.220955
 Sum of electronic and zero-point Energies= -3281.550355
 Sum of electronic and thermal Energies= -3281.533302
 Sum of electronic and thermal Enthalpies= -3281.532358
 Sum of electronic and thermal Free Energies= -3281.603225

Lowest vibration frequency: 7.5818

Counterpoise: BSSE energy = 0.000438904398

Summary of Natural Population Analysis:

		Natural Population				
Atom	No	Natural Charge	Core	Valence	Rydberg	Total
C	1	-0.10329	1.99868	4.08139	0.02321	6.10329
C	2	-0.23426	1.99898	4.21614	0.01914	6.23426
C	3	-0.14665	1.99910	4.13060	0.01695	6.14665
C	4	-0.16791	1.99893	4.15108	0.01790	6.16791
C	5	-0.15236	1.99911	4.13698	0.01627	6.15236
C	6	-0.23499	1.99898	4.21688	0.01913	6.23499
H	7	0.23153	0.00000	0.76688	0.00159	0.76847
H	8	0.23739	0.00000	0.76097	0.00164	0.76261
C	9	0.80363	1.99930	3.15036	0.04671	5.19637
H	10	0.23410	0.00000	0.76433	0.00157	0.76590
H	11	0.23091	0.00000	0.76748	0.00161	0.76909
Br	12	0.09967	27.99927	6.87871	0.02235	34.90033
O	13	-0.55065	1.99972	6.53674	0.01419	8.55065
C	14	-0.21746	1.99926	4.20243	0.01577	6.21746
H	15	0.19167	0.00000	0.80726	0.00108	0.80833
H	16	0.18987	0.00000	0.80842	0.00171	0.81013
H	17	0.18987	0.00000	0.80842	0.00171	0.81013
O	18	-0.61789	1.99975	6.60691	0.01122	8.61789
N	19	-0.47695	1.99943	5.45506	0.02246	7.47695
C	20	0.05193	1.99923	3.92173	0.02711	5.94807
C	21	-0.24802	1.99917	4.23243	0.01642	6.24802
C	22	-0.16863	1.99918	4.15203	0.01742	6.16863
C	23	-0.24802	1.99917	4.23243	0.01642	6.24802
C	24	0.05193	1.99923	3.92173	0.02711	5.94807
H	25	0.19616	0.00000	0.80239	0.00146	0.80384
H	26	0.22186	0.00000	0.77665	0.00149	0.77814
H	27	0.21854	0.00000	0.78017	0.00129	0.78146
H	28	0.22186	0.00000	0.77665	0.00149	0.77814
H	29	0.19616	0.00000	0.80239	0.00146	0.80384
=====						
* Total *		0.00000	59.98650	87.64564	0.36786	148.00000

Contributions to pi population at N atom in pyridine:

Val(2p)	1.18070
Ryd(3p)	0.00074
Ryd(4p)	0.00034
Ryd(5p)	0.00002
Total:	1.18180

 NMR properties

Electronic energy (NMR): -3281.77130994

1	C	Isotropic =	21.8327	Anisotropy =	145.7629
2	C	Isotropic =	41.4069	Anisotropy =	162.5556
3	C	Isotropic =	41.5350	Anisotropy =	194.5684

4	C	Isotropic =	45.2493	Anisotropy =	165.9925
5	C	Isotropic =	43.0586	Anisotropy =	189.4220
6	C	Isotropic =	41.8815	Anisotropy =	161.9002
7	H	Isotropic =	23.6587	Anisotropy =	9.8715
8	H	Isotropic =	23.0492	Anisotropy =	7.7344
9	C	Isotropic =	6.3387	Anisotropy =	85.4927
10	H	Isotropic =	23.2587	Anisotropy =	8.6630
11	H	Isotropic =	23.7158	Anisotropy =	9.9977
12	Br	Isotropic =	1904.1767	Anisotropy =	1426.8812
13	O	Isotropic =	145.2928	Anisotropy =	141.8710
14	C	Isotropic =	126.7136	Anisotropy =	73.1082
15	H	Isotropic =	27.7651	Anisotropy =	7.5903
16	H	Isotropic =	27.6018	Anisotropy =	8.4582
17	H	Isotropic =	27.6018	Anisotropy =	8.4582
18	O	Isotropic =	-66.0550	Anisotropy =	603.0071
19	N	Isotropic =	-98.3306	Anisotropy =	576.0406
20	C	Isotropic =	21.9915	Anisotropy =	180.7630
21	C	Isotropic =	50.5245	Anisotropy =	183.6987
22	C	Isotropic =	37.3196	Anisotropy =	212.0982
23	C	Isotropic =	50.5245	Anisotropy =	183.6987
24	C	Isotropic =	21.9915	Anisotropy =	180.7630
25	H	Isotropic =	22.5157	Anisotropy =	9.5239
26	H	Isotropic =	23.9386	Anisotropy =	5.3378
27	H	Isotropic =	23.5127	Anisotropy =	4.9199
28	H	Isotropic =	23.9386	Anisotropy =	5.3378
29	H	Isotropic =	22.5157	Anisotropy =	9.5239

=====

X = Br R = COCH3

Energies and NBO populations

Number of atoms: 28

Full point group

CS

NOp 2

C	0.000000	-0.418626	0.002964
C	0.000000	-1.057119	1.248301
C	0.000000	-2.447694	1.298754
C	0.000000	-3.213624	0.120766
C	0.000000	-2.551240	-1.117198
C	0.000000	-1.157351	-1.182328
H	0.000000	-0.472259	2.161547
H	0.000000	-2.962039	2.254143
C	0.000000	-4.711636	0.235722
H	0.000000	-3.113557	-2.045183
H	0.000000	-0.651989	-2.141693
Br	0.000000	1.490938	-0.076419
C	0.000000	-5.541767	-1.035909
O	0.000000	-5.249287	1.330544
H	0.000000	-6.598422	-0.766995
H	0.883830	-5.320901	-1.644530
H	-0.883830	-5.320901	-1.644530
N	0.000000	4.601031	-0.184643
C	1.145200	5.296265	-0.184643
C	1.199903	6.692061	-0.184939
C	0.000000	7.404709	-0.185075
C	-1.199903	6.692061	-0.184939
C	-1.145200	5.296265	-0.184643
H	2.060702	4.708352	-0.183988
H	2.158405	7.200757	-0.184668
H	0.000000	8.490572	-0.184993
H	-2.158405	7.200757	-0.184668
H	-2.060702	4.708352	-0.183988

Electronic energy:	-3206.53574851 (	-3206.53574860)
Zero-point correction=	0.215343	
Sum of electronic and zero-point Energies=	-3206.320406	
Sum of electronic and thermal Energies=	-3206.304331	
Sum of electronic and thermal Enthalpies=	-3206.303387	
Sum of electronic and thermal Free Energies=	-3206.371637	

Lowest vibration frequency: 6.7724

Counterpoise: BSSE energy = 0.000431990517

Summary of Natural Population Analysis:

Atom	No	Natural Charge	Natural Population			
			Core	Valence	Rydberg	Total
C	1	-0.10231	1.99868	4.08030	0.02333	6.10231
C	2	-0.23227	1.99898	4.21426	0.01903	6.23227
C	3	-0.14305	1.99909	4.12598	0.01798	6.14305
C	4	-0.16232	1.99895	4.14603	0.01733	6.16232
C	5	-0.16165	1.99911	4.14718	0.01536	6.16165
C	6	-0.23540	1.99898	4.21692	0.01950	6.23540
H	7	0.23123	0.00000	0.76718	0.00159	0.76877
H	8	0.23751	0.00000	0.76058	0.00190	0.76249
C	9	0.56546	1.99931	3.39652	0.03871	5.43454
H	10	0.22093	0.00000	0.77756	0.00150	0.77907
H	11	0.23076	0.00000	0.76760	0.00163	0.76924
Br	12	0.10167	27.99927	6.87669	0.02236	34.89833
C	13	-0.67816	1.99930	4.67054	0.00833	6.67816
O	14	-0.57824	1.99976	6.56477	0.01371	8.57824
H	15	0.23147	0.00000	0.76666	0.00187	0.76853
H	16	0.22865	0.00000	0.76994	0.00141	0.77135
H	17	0.22865	0.00000	0.76994	0.00141	0.77135
N	18	-0.47738	1.99943	5.45547	0.02248	7.47738
C	19	0.05197	1.99923	3.92170	0.02710	5.94803
C	20	-0.24790	1.99917	4.23232	0.01642	6.24790
C	21	-0.16847	1.99918	4.15187	0.01742	6.16847
C	22	-0.24790	1.99917	4.23232	0.01642	6.24790
C	23	0.05197	1.99923	3.92170	0.02710	5.94803
H	24	0.19613	0.00000	0.80241	0.00145	0.80387
H	25	0.22194	0.00000	0.77657	0.00149	0.77806
H	26	0.21864	0.00000	0.78007	0.00129	0.78136
H	27	0.22194	0.00000	0.77657	0.00149	0.77806
H	28	0.19613	0.00000	0.80241	0.00145	0.80387
=====						
* Total *		0.00000	57.98685	81.67209	0.34106	140.00000

Contributions to pi population at N atom in pyridine:

Val(2p) 1.18110
 Ryd(3p) 0.00074
 Ryd(4p) 0.00034
 Ryd(5p) 0.00002

 Total: 1.18220

NMR properties

Electronic energy (NMR): -3206.53574851

1	C	Isotropic =	21.3563	Anisotropy =	146.3676
2	C	Isotropic =	41.0054	Anisotropy =	162.3133
3	C	Isotropic =	44.6574	Anisotropy =	198.6626
4	C	Isotropic =	38.6063	Anisotropy =	179.1058
5	C	Isotropic =	42.5948	Anisotropy =	183.7303
6	C	Isotropic =	41.9697	Anisotropy =	161.2630
7	H	Isotropic =	23.6613	Anisotropy =	9.9529
8	H	Isotropic =	23.0705	Anisotropy =	6.9124
9	C	Isotropic =	-25.6912	Anisotropy =	173.4674
10	H	Isotropic =	23.3817	Anisotropy =	8.3971
11	H	Isotropic =	23.6719	Anisotropy =	10.0439
12	Br	Isotropic =	1900.9906	Anisotropy =	1420.3293
13	C	Isotropic =	153.8211	Anisotropy =	46.3282
14	O	Isotropic =	-290.6812	Anisotropy =	967.0196
15	H	Isotropic =	29.3397	Anisotropy =	6.9460
16	H	Isotropic =	28.7171	Anisotropy =	4.7002
17	H	Isotropic =	28.7171	Anisotropy =	4.7002
18	N	Isotropic =	-98.1702	Anisotropy =	575.5936
19	C	Isotropic =	22.0193	Anisotropy =	180.7786
20	C	Isotropic =	50.4957	Anisotropy =	183.7458
21	C	Isotropic =	37.2959	Anisotropy =	212.2015
22	C	Isotropic =	50.4957	Anisotropy =	183.7458
23	C	Isotropic =	22.0193	Anisotropy =	180.7786
24	H	Isotropic =	22.5165	Anisotropy =	9.4927
25	H	Isotropic =	23.9353	Anisotropy =	5.3308
26	H	Isotropic =	23.5087	Anisotropy =	4.9286
27	H	Isotropic =	23.9353	Anisotropy =	5.3308

28 H Isotropic = 22.5165 Anisotropy = 9.4927

=====

X = Br R = CF3

Energies and NBO populations

Number of atoms: 26

Full point group CS NOp 2

C	0.000000	-0.436711	-0.002367
C	0.000000	-1.118666	1.215334
C	0.000000	-2.515168	1.220049
C	0.000000	-3.217185	0.013037
C	0.000000	-2.525553	-1.204607
C	0.000000	-1.132558	-1.215674
H	0.000000	-0.570559	2.150697
H	0.000000	-3.049326	2.163411
C	0.000000	-4.724055	-0.007238
H	0.000000	-3.068946	-2.144347
H	0.000000	-0.593106	-2.156043
Br	0.000000	1.474773	-0.013836
F	0.000000	-5.261841	1.228327
F	-1.083035	-5.220834	-0.652717
F	1.083035	-5.220834	-0.652717
N	0.000000	4.564201	-0.027170
C	1.145330	5.259255	-0.027170
C	1.199932	6.655015	-0.027177
C	0.000000	7.367615	-0.027088
C	-1.199932	6.655015	-0.027177
C	-1.145330	5.259255	-0.027170
H	2.060785	4.671296	-0.027116
H	2.158409	7.163731	-0.027163
H	0.000000	8.453467	-0.026882
H	-2.158409	7.163731	-0.027163
H	-2.060785	4.671296	-0.027116

Electronic energy:	-3390.95045364 (-3390.95040819)
Zero-point correction=	0.182781
Sum of electronic and zero-point Energies=	-3390.767673
Sum of electronic and thermal Energies=	-3390.752474
Sum of electronic and thermal Enthalpies=	-3390.751529
Sum of electronic and thermal Free Energies=	-3390.817558

Lowest vibration frequency: -17.8666

Counterpoise: BSSE energy = 0.000446598965

Summary of Natural Population Analysis:

		Natural Population				
Atom	No	Natural Charge	Core	Valence	Rydberg	Total
C	1	-0.10942	1.99868	4.08734	0.02340	6.10942
C	2	-0.22541	1.99898	4.20737	0.01906	6.22541
C	3	-0.16822	1.99909	4.15248	0.01666	6.16822
C	4	-0.16011	1.99891	4.14393	0.01727	6.16011
C	5	-0.16449	1.99909	4.14838	0.01701	6.16449
C	6	-0.22419	1.99899	4.20610	0.01910	6.22419
H	7	0.23362	0.00000	0.76481	0.00157	0.76638
H	8	0.23581	0.00000	0.76258	0.00161	0.76419
C	9	1.06723	1.99912	2.87343	0.06021	4.93277
H	10	0.23184	0.00000	0.76656	0.00160	0.76816
H	11	0.23387	0.00000	0.76455	0.00158	0.76613
Br	12	0.10401	27.99928	6.87433	0.02238	34.89599
F	13	-0.35538	1.99992	7.34908	0.00638	9.35538
F	14	-0.35871	1.99992	7.35226	0.00653	9.35871
F	15	-0.35871	1.99992	7.35226	0.00653	9.35871
N	16	-0.47814	1.99943	5.45620	0.02251	7.47814
C	17	0.05216	1.99923	3.92154	0.02707	5.94784
C	18	-0.24766	1.99917	4.23208	0.01641	6.24766
C	19	-0.16815	1.99918	4.15155	0.01741	6.16815
C	20	-0.24766	1.99917	4.23208	0.01641	6.24766
C	21	0.05216	1.99923	3.92154	0.02707	5.94784

H	22	0.19625	0.00000	0.80230	0.00145	0.80375
H	23	0.22212	0.00000	0.77640	0.00149	0.77788
H	24	0.21881	0.00000	0.77991	0.00128	0.78119
H	25	0.22212	0.00000	0.77640	0.00149	0.77788
H	26	0.19625	0.00000	0.80230	0.00145	0.80375
=====						
* Total *		0.00000	59.98731	89.65777	0.35492	150.00000

Contributions to pi population at N atom in pyridine:

Val(2p)	1.18217
Ryd(3p)	0.00074
Ryd(4p)	0.00034
Ryd(5p)	0.00002

Total: 1.18327

----- NMR properties -----

Electronic energy (NMR): -3390.95045364

1	C	Isotropic =	23.6871	Anisotropy =	146.4333
2	C	Isotropic =	41.5398	Anisotropy =	165.1695
3	C	Isotropic =	47.9104	Anisotropy =	186.2480
4	C	Isotropic =	43.0928	Anisotropy =	165.9477
5	C	Isotropic =	45.7804	Anisotropy =	188.2045
6	C	Isotropic =	40.6052	Anisotropy =	166.5692
7	H	Isotropic =	23.6461	Anisotropy =	9.7216
8	H	Isotropic =	23.6154	Anisotropy =	8.4467
9	C	Isotropic =	43.2287	Anisotropy =	10.9000
10	H	Isotropic =	23.6683	Anisotropy =	8.0035
11	H	Isotropic =	23.6020	Anisotropy =	9.7582
12	Br	Isotropic =	1904.5113	Anisotropy =	1455.1189
13	F	Isotropic =	261.2689	Anisotropy =	113.1866
14	F	Isotropic =	232.4671	Anisotropy =	148.4153
15	F	Isotropic =	232.4671	Anisotropy =	148.4153
16	N	Isotropic =	-97.7201	Anisotropy =	574.4351
17	C	Isotropic =	22.0837	Anisotropy =	180.7128
18	C	Isotropic =	50.4584	Anisotropy =	183.8129
19	C	Isotropic =	37.2429	Anisotropy =	212.3278
20	C	Isotropic =	50.4584	Anisotropy =	183.8129
21	C	Isotropic =	22.0837	Anisotropy =	180.7128
22	H	Isotropic =	22.5222	Anisotropy =	9.5678
23	H	Isotropic =	23.9324	Anisotropy =	5.3398
24	H	Isotropic =	23.5061	Anisotropy =	4.9218
25	H	Isotropic =	23.9324	Anisotropy =	5.3398
26	H	Isotropic =	22.5222	Anisotropy =	9.5678

=====

X = Br R = CN

Energies and NBO populations -----

Number of atoms: 24
Full point group C2V NOp 4

C	0.000000	-0.506352	0.000000
C	0.000000	-1.196232	1.216783
C	0.000000	-2.588631	1.217122
C	0.000000	-3.291337	0.000000
C	0.000000	-2.588631	-1.217122
C	0.000000	-1.196232	-1.216783
H	0.000000	-0.651031	2.153866
H	0.000000	-3.132122	2.155855
C	0.000000	-4.724408	0.000000
H	0.000000	-3.132122	-2.155855
H	0.000000	-0.651031	-2.153866
Br	0.000000	1.402052	0.000000
N	0.000000	-5.883218	0.000000
N	0.000000	4.475791	0.000000
C	1.145223	5.170993	0.000000
C	1.199860	6.566714	0.000000
C	0.000000	7.279468	0.000000
C	-1.199860	6.566714	0.000000
C	-1.145223	5.170993	0.000000

H	2.061115	4.583692	0.000000
H	2.158525	7.075048	0.000000
H	0.000000	8.365324	0.000000
H	-2.158525	7.075048	0.000000
H	-2.061115	4.583692	0.000000

Electronic energy:	-3146.13114342 (-3146.13114342)
Zero-point correction=	0.177037
Sum of electronic and zero-point Energies=	-3145.954106
Sum of electronic and thermal Energies=	-3145.939838
Sum of electronic and thermal Enthalpies=	-3145.938894
Sum of electronic and thermal Free Energies=	-3146.002321

Lowest vibration frequency: 8.1378

Counterpoise: BSSE energy = 0.000455830399

Summary of Natural Population Analysis:

		Natural Population				
Atom	No	Natural Charge	Core	Valence	Rydberg	Total
C	1	-0.10571	1.99870	4.08355	0.02347	6.10571
C	2	-0.22645	1.99898	4.20850	0.01897	6.22645
C	3	-0.13944	1.99911	4.12487	0.01546	6.13944
C	4	-0.19430	1.99888	4.17843	0.01699	6.19430
C	5	-0.13944	1.99911	4.12487	0.01546	6.13944
C	6	-0.22645	1.99898	4.20850	0.01897	6.22645
H	7	0.23605	0.00000	0.76241	0.00154	0.76395
H	8	0.23290	0.00000	0.76581	0.00129	0.76710
C	9	0.30731	1.99927	3.65943	0.03400	5.69269
H	10	0.23290	0.00000	0.76581	0.00129	0.76710
H	11	0.23605	0.00000	0.76241	0.00154	0.76395
Br	12	0.11419	27.99927	6.86413	0.02241	34.88581
N	13	-0.34694	1.99959	5.32705	0.02030	7.34694
N	14	-0.47941	1.99943	5.45743	0.02255	7.47941
C	15	0.05234	1.99923	3.92140	0.02702	5.94766
C	16	-0.24733	1.99917	4.23176	0.01640	6.24733
C	17	-0.16763	1.99918	4.15105	0.01740	6.16763
C	18	-0.24733	1.99917	4.23176	0.01640	6.24733
C	19	0.05234	1.99923	3.92140	0.02702	5.94766
H	20	0.19623	0.00000	0.80231	0.00145	0.80377
H	21	0.22238	0.00000	0.77614	0.00148	0.77762
H	22	0.21909	0.00000	0.77962	0.00128	0.78091
H	23	0.22238	0.00000	0.77614	0.00148	0.77762
H	24	0.19623	0.00000	0.80231	0.00145	0.80377
* Total *		0.00000	55.98727	73.68712	0.32561	130.00000

Contributions to pi population at N atom in pyridine:

Val(2p)	1.18349
Ryd(3p)	0.00074
Ryd(4p)	0.00035
Ryd(5p)	0.00002

Total: 1.18460

NMR properties

Electronic energy (NMR): -3146.13114342

1	C	Isotropic =	21.5464	Anisotropy =	152.5853
2	C	Isotropic =	40.5905	Anisotropy =	167.5468
3	C	Isotropic =	39.1564	Anisotropy =	188.9566
4	C	Isotropic =	64.0755	Anisotropy =	150.3059
5	C	Isotropic =	39.1564	Anisotropy =	188.9566
6	C	Isotropic =	40.5905	Anisotropy =	167.5468
7	H	Isotropic =	23.6566	Anisotropy =	9.6805
8	H	Isotropic =	23.6650	Anisotropy =	6.2854
9	C	Isotropic =	55.5818	Anisotropy =	352.7752
10	H	Isotropic =	23.6650	Anisotropy =	6.2854
11	H	Isotropic =	23.6566	Anisotropy =	9.6805
12	Br	Isotropic =	1877.8646	Anisotropy =	1426.9361
13	N	Isotropic =	-43.1950	Anisotropy =	418.3365
14	N	Isotropic =	-96.9877	Anisotropy =	572.4676

15	C	Isotropic =	22.1796	Anisotropy =	180.6913
16	C	Isotropic =	50.3617	Anisotropy =	183.9565
17	C	Isotropic =	37.1421	Anisotropy =	212.5867
18	C	Isotropic =	50.3617	Anisotropy =	183.9565
19	C	Isotropic =	22.1796	Anisotropy =	180.6913
20	H	Isotropic =	22.5352	Anisotropy =	9.5175
21	H	Isotropic =	23.9261	Anisotropy =	5.3384
22	H	Isotropic =	23.4976	Anisotropy =	4.9406
23	H	Isotropic =	23.9261	Anisotropy =	5.3384
24	H	Isotropic =	22.5352	Anisotropy =	9.5175

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X = Br R = N02

Energies and NBO populations

Number of atoms: 25

Full point group

C2V

NOp

4

C	0.000000	-0.453692	0.000000
C	0.000000	-1.142269	1.218524
C	0.000000	-2.534780	1.220572
C	0.000000	-3.211063	0.000000
C	0.000000	-2.534780	-1.220572
C	0.000000	-1.142269	-1.218524
H	0.000000	-0.596763	2.155148
H	0.000000	-3.093554	2.148415
N	0.000000	-4.682826	0.000000
H	0.000000	-3.093554	-2.148415
H	0.000000	-0.596763	-2.155148
Br	0.000000	1.453199	0.000000
O	0.000000	-5.257086	1.084745
O	0.000000	-5.257086	-1.084746
N	0.000000	4.508190	0.000000
C	1.145563	5.202982	0.000000
C	1.199985	6.598625	0.000000
C	0.000000	7.311126	0.000000
C	-1.199985	6.598625	0.000000
C	-1.145563	5.202982	0.000000
H	2.060990	4.615022	0.000000
H	2.158410	7.107368	0.000000
H	0.000000	8.396950	0.000000
H	-2.158410	7.107368	0.000000
H	-2.060990	4.615022	0.000000

Electronic energy:	-3258.40070619 (-3258.40070619)
Zero-point correction=	0.180758
Sum of electronic and zero-point Energies=	-3258.219948
Sum of electronic and thermal Energies=	-3258.204901
Sum of electronic and thermal Enthalpies=	-3258.203956
Sum of electronic and thermal Free Energies=	-3258.269789

Lowest vibration frequency: 9.1902

Counterpoise: BSSE energy = 0.000485888843

Summary of Natural Population Analysis:

Atom	No	Natural Charge	Natural Population			
			Core	Valence	Rydberg	Total
C	1	-0.09916	1.99870	4.07691	0.02354	6.09916
C	2	-0.22344	1.99899	4.20551	0.01895	6.22344
C	3	-0.17166	1.99907	4.15425	0.01834	6.17166
C	4	0.05947	1.99879	3.92098	0.02075	5.94053
C	5	-0.17166	1.99907	4.15425	0.01834	6.17166
C	6	-0.22344	1.99899	4.20551	0.01895	6.22344
H	7	0.23739	0.00000	0.76111	0.00150	0.76261
H	8	0.25222	0.00000	0.74587	0.00191	0.74778
N	9	0.48946	1.99945	4.45611	0.05497	6.51054
H	10	0.25222	0.00000	0.74587	0.00191	0.74778
H	11	0.23739	0.00000	0.76111	0.00150	0.76261
Br	12	0.12309	27.99927	6.85506	0.02258	34.87691
O	13	-0.39117	1.99980	6.37645	0.01492	8.39117

O	14	-0.39117	1.99980	6.37646	0.01492	8.39117
N	15	-0.48053	1.99943	5.45853	0.02257	7.48053
C	16	0.05256	1.99923	3.92123	0.02698	5.94744
C	17	-0.24702	1.99917	4.23146	0.01639	6.24702
C	18	-0.16724	1.99918	4.15067	0.01739	6.16724
C	19	-0.24702	1.99917	4.23146	0.01639	6.24702
C	20	0.05256	1.99923	3.92123	0.02698	5.94744
H	21	0.19635	0.00000	0.80220	0.00145	0.80365
H	22	0.22258	0.00000	0.77594	0.00148	0.77742
H	23	0.21927	0.00000	0.77945	0.00128	0.78073
H	24	0.22258	0.00000	0.77594	0.00148	0.77742
H	25	0.19635	0.00000	0.80220	0.00145	0.80365
=====						
* Total *		0.00000	57.98733	81.64574	0.36692	140.00000

Contributions to pi population at N atom in pyridine:

Val(2p)	1.18471
Ryd(3p)	0.00074
Ryd(4p)	0.00035
Ryd(5p)	0.00002

Total: 1.18582

----- NMR properties -----

Electronic energy (NMR): -3258.40070619

1	C	Isotropic =	17.3987	Anisotropy =	157.5252
2	C	Isotropic =	40.9393	Anisotropy =	163.7352
3	C	Isotropic =	48.7081	Anisotropy =	184.6939
4	C	Isotropic =	25.8144	Anisotropy =	115.9778
5	C	Isotropic =	48.7081	Anisotropy =	184.6939
6	C	Isotropic =	40.9393	Anisotropy =	163.7352
7	H	Isotropic =	23.6108	Anisotropy =	9.8215
8	H	Isotropic =	22.9157	Anisotropy =	5.8956
9	N	Isotropic =	-154.2969	Anisotropy =	290.7412
10	H	Isotropic =	22.9157	Anisotropy =	5.8956
11	H	Isotropic =	23.6108	Anisotropy =	9.8215
12	Br	Isotropic =	1865.5914	Anisotropy =	1396.7917
13	O	Isotropic =	-308.0818	Anisotropy =	723.9839
14	O	Isotropic =	-308.0818	Anisotropy =	723.9839
15	N	Isotropic =	-96.6463	Anisotropy =	571.4852
16	C	Isotropic =	22.2360	Anisotropy =	180.6929
17	C	Isotropic =	50.3301	Anisotropy =	184.1171
18	C	Isotropic =	37.0635	Anisotropy =	212.7326
19	C	Isotropic =	50.3301	Anisotropy =	184.1171
20	C	Isotropic =	22.2360	Anisotropy =	180.6929
21	H	Isotropic =	22.5298	Anisotropy =	9.4931
22	H	Isotropic =	23.9168	Anisotropy =	5.3191
23	H	Isotropic =	23.4902	Anisotropy =	4.9376
24	H	Isotropic =	23.9168	Anisotropy =	5.3191
25	H	Isotropic =	22.5298	Anisotropy =	9.4931

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X = Cl R = H

----- Energies and NBO populations -----

Number of atoms: 23

Full point group C2V NOp 4

C	0.000000	-0.667475	0.000000
C	0.000000	-1.351090	1.216661
C	0.000000	-2.748591	1.208548
C	0.000000	-3.450178	0.000000
C	0.000000	-2.748591	-1.208548
C	0.000000	-1.351090	-1.216661
H	0.000000	-0.799225	2.150260
H	0.000000	-3.285858	2.152129
H	0.000000	-4.535584	0.000000
H	0.000000	-3.285858	-2.152129
H	0.000000	-0.799225	-2.150260
Cl	0.000000	1.094747	0.000000
N	0.000000	4.420709	0.000000

C	1.143941	5.117997	0.000000
C	1.199508	6.514126	0.000000
C	0.000000	7.227358	0.000000
C	-1.199508	6.514126	0.000000
C	-1.143941	5.117997	0.000000
H	2.060275	4.531207	0.000000
H	2.158273	7.022579	0.000000
H	0.000000	8.313297	0.000000
H	-2.158273	7.022579	0.000000
H	-2.060275	4.531207	0.000000

Electronic energy:	-940.07960339 (	-940.07960339)
Zero-point correction=	0.178880	
Sum of electronic and zero-point Energies=	-939.900723	
Sum of electronic and thermal Energies=	-939.889324	
Sum of electronic and thermal Enthalpies=	-939.888380	
Sum of electronic and thermal Free Energies=	-939.944792	

Lowest vibration frequency: -11.1580

Counterpoise: BSSE energy = 0.000642529394

Summary of Natural Population Analysis:

Atom	No	Natural Charge	Natural Population			
			Core	Valence	Rydberg	Total
C	1	-0.04637	1.99861	4.02429	0.02347	6.04637
C	2	-0.23463	1.99898	4.21673	0.01892	6.23463
C	3	-0.19425	1.99918	4.17753	0.01754	6.19425
C	4	-0.21659	1.99918	4.19974	0.01767	6.21659
C	5	-0.19425	1.99918	4.17753	0.01754	6.19425
C	6	-0.23463	1.99898	4.21673	0.01892	6.23463
H	7	0.22878	0.00000	0.76959	0.00163	0.77122
H	8	0.21719	0.00000	0.78141	0.00140	0.78281
H	9	0.21607	0.00000	0.78260	0.00132	0.78393
H	10	0.21719	0.00000	0.78141	0.00140	0.78281
H	11	0.22878	0.00000	0.76959	0.00163	0.77122
Cl	12	0.00708	9.99963	6.97075	0.02254	16.99292
N	13	-0.47031	1.99945	5.44858	0.02228	7.47031
C	14	0.04915	1.99923	3.92375	0.02788	5.95085
C	15	-0.24991	1.99917	4.23423	0.01651	6.24991
C	16	-0.17101	1.99918	4.15432	0.01751	6.17101
C	17	-0.24991	1.99917	4.23423	0.01651	6.24991
C	18	0.04915	1.99923	3.92375	0.02788	5.95085
H	19	0.19478	0.00000	0.80376	0.00147	0.80522
H	20	0.22070	0.00000	0.77782	0.00149	0.77930
H	21	0.21753	0.00000	0.78117	0.00130	0.78247
H	22	0.22070	0.00000	0.77782	0.00149	0.77930
H	23	0.19478	0.00000	0.80376	0.00147	0.80522
=====						
* Total *		0.00000	33.98917	65.73108	0.27975	100.00000

Contributions to pi population at N atom in pyridine:

Val(2p)	1.17111
Ryd(3p)	0.00071
Ryd(4p)	0.00033
Ryd(5p)	0.00003

Total: 1.17218

NMR properties

Electronic energy (NMR): -940.07960339

1	C	Isotropic =	28.1751	Anisotropy =	130.3792
2	C	Isotropic =	44.3823	Anisotropy =	164.0224
3	C	Isotropic =	44.0096	Anisotropy =	188.8469
4	C	Isotropic =	48.0597	Anisotropy =	191.3286
5	C	Isotropic =	44.0096	Anisotropy =	188.8469
6	C	Isotropic =	44.3823	Anisotropy =	164.0224
7	H	Isotropic =	23.7733	Anisotropy =	8.2162
8	H	Isotropic =	23.8717	Anisotropy =	4.9375
9	H	Isotropic =	23.9841	Anisotropy =	4.2844
10	H	Isotropic =	23.8717	Anisotropy =	4.9375

11	H	Isotropic =	23.7733	Anisotropy =	8.2162
12	Cl	Isotropic =	663.7295	Anisotropy =	530.3794
13	N	Isotropic =	-101.8064	Anisotropy =	587.1353
14	C	Isotropic =	21.4366	Anisotropy =	181.0841
15	C	Isotropic =	50.7711	Anisotropy =	183.2567
16	C	Isotropic =	37.7710	Anisotropy =	211.3243
17	C	Isotropic =	50.7711	Anisotropy =	183.2567
18	C	Isotropic =	21.4366	Anisotropy =	181.0841
19	H	Isotropic =	22.4686	Anisotropy =	8.5699
20	H	Isotropic =	23.9689	Anisotropy =	5.3188
21	H	Isotropic =	23.5265	Anisotropy =	5.0200
22	H	Isotropic =	23.9689	Anisotropy =	5.3188
23	H	Isotropic =	22.4686	Anisotropy =	8.5699

=====

X = Cl R = OCH3

Energies and NBO populations

Number of atoms: 27
Full point group CS NOp 2

C	0.000000	-0.498765	-0.003914
C	0.000000	-1.242496	1.171243
C	0.000000	-2.641535	1.113479
C	0.000000	-3.288952	-0.128799
C	0.000000	-2.525870	-1.307707
C	0.000000	-1.136600	-1.248302
H	0.000000	-0.741278	2.133111
H	0.000000	-3.203277	2.039939
O	0.000000	-4.640572	-0.295553
H	0.000000	-3.039198	-2.263746
H	0.000000	-0.550248	-2.160701
Cl	0.000000	1.262785	0.071544
C	0.000000	-5.461475	0.866110
H	0.000000	-6.489826	0.504112
H	-0.896165	-5.289732	1.474771
H	0.896165	-5.289732	1.474771
N	0.000000	4.602116	0.125794
C	1.143924	5.299368	0.125794
C	1.199508	6.695534	0.125990
C	0.000000	7.408777	0.126153
C	-1.199508	6.695534	0.125990
C	-1.143924	5.299368	0.125794
H	2.060229	4.712524	0.125448
H	2.158285	7.203978	0.125914
H	0.000000	8.494723	0.126279
H	-2.158285	7.203978	0.125914
H	-2.060229	4.712524	0.125448

Electronic energy: -1054.60194562 (-1054.60194561)
Zero-point correction= 0.211263
Sum of electronic and zero-point Energies= -1054.390683
Sum of electronic and thermal Energies= -1054.375770
Sum of electronic and thermal Enthalpies= -1054.374826
Sum of electronic and thermal Free Energies= -1054.440848

Lowest vibration frequency: 5.3006

Counterpoise: BSSE energy = 0.001059994696

Summary of Natural Population Analysis:

		Natural Population				
Atom	No	Natural Charge	Core	Valence	Rydberg	Total
C	1	-0.07510	1.99859	4.05350	0.02302	6.07510
C	2	-0.21349	1.99900	4.19561	0.01888	6.21349
C	3	-0.28367	1.99908	4.26964	0.01495	6.28367
C	4	0.31794	1.99881	3.66032	0.02293	5.68206
C	5	-0.23379	1.99909	4.21773	0.01697	6.23379
C	6	-0.21609	1.99900	4.19875	0.01833	6.21609
H	7	0.22818	0.00000	0.77012	0.00169	0.77182
H	8	0.22403	0.00000	0.77418	0.00179	0.77597

O	9	-0.54407	1.99971	6.52658	0.01778	8.54407
H	10	0.22816	0.00000	0.77008	0.00176	0.77184
H	11	0.22937	0.00000	0.76904	0.00158	0.77063
Cl	12	0.00042	9.99964	6.97726	0.02268	16.99958
C	13	-0.21373	1.99928	4.19872	0.01573	6.21373
H	14	0.19659	0.00000	0.80240	0.00101	0.80341
H	15	0.17498	0.00000	0.82293	0.00210	0.82502
H	16	0.17498	0.00000	0.82293	0.00210	0.82502
N	17	-0.46961	1.99945	5.44783	0.02233	7.46961
C	18	0.04907	1.99923	3.92378	0.02793	5.95093
C	19	-0.25007	1.99917	4.23438	0.01652	6.25007
C	20	-0.17125	1.99918	4.15455	0.01752	6.17125
C	21	-0.25007	1.99917	4.23438	0.01652	6.25007
C	22	0.04907	1.99923	3.92378	0.02793	5.95093
H	23	0.19481	0.00000	0.80373	0.00147	0.80519
H	24	0.22057	0.00000	0.77795	0.00149	0.77943
H	25	0.21740	0.00000	0.78129	0.00131	0.78260
H	26	0.22057	0.00000	0.77795	0.00149	0.77943
H	27	0.19481	0.00000	0.80373	0.00147	0.80519
=====						
* Total *		0.00000	37.98762	77.69315	0.31923	116.00000

Contributions to pi population at N atom in pyridine:

Val(2p)	1.17054
Ryd(3p)	0.00071
Ryd(4p)	0.00033
Ryd(5p)	0.00003

Total: 1.17161

NMR properties

Electronic energy (NMR): -1054.60194562

1	C	Isotropic =	39.1382	Anisotropy =	113.4934
2	C	Isotropic =	44.0029	Anisotropy =	162.3874
3	C	Isotropic =	65.6431	Anisotropy =	161.7752
4	C	Isotropic =	12.8669	Anisotropy =	145.2195
5	C	Isotropic =	56.1508	Anisotropy =	151.2869
6	C	Isotropic =	43.3237	Anisotropy =	162.0189
7	H	Isotropic =	23.8863	Anisotropy =	8.1816
8	H	Isotropic =	24.5286	Anisotropy =	7.5045
9	O	Isotropic =	223.3923	Anisotropy =	84.6627
10	H	Isotropic =	24.3518	Anisotropy =	5.7834
11	H	Isotropic =	23.9381	Anisotropy =	7.7942
12	Cl	Isotropic =	685.5880	Anisotropy =	527.5842
13	C	Isotropic =	124.4901	Anisotropy =	75.5374
14	H	Isotropic =	27.4774	Anisotropy =	8.3299
15	H	Isotropic =	27.8757	Anisotropy =	8.0214
16	H	Isotropic =	27.8757	Anisotropy =	8.0214
17	N	Isotropic =	-102.0816	Anisotropy =	588.0022
18	C	Isotropic =	21.4002	Anisotropy =	181.0145
19	C	Isotropic =	50.8262	Anisotropy =	183.1693
20	C	Isotropic =	37.8113	Anisotropy =	211.1192
21	C	Isotropic =	50.8262	Anisotropy =	183.1693
22	C	Isotropic =	21.4002	Anisotropy =	181.0145
23	H	Isotropic =	22.4787	Anisotropy =	8.6452
24	H	Isotropic =	23.9801	Anisotropy =	5.3347
25	H	Isotropic =	23.5380	Anisotropy =	5.0061
26	H	Isotropic =	23.9801	Anisotropy =	5.3347
27	H	Isotropic =	22.4787	Anisotropy =	8.6452

X = Cl R = OCH2CH3

Energies and NBO populations

Number of atoms: 30

Full point group CS NOp 2

C	0.000000	-0.389535	-0.041209
C	0.000000	-1.141658	1.128570
C	0.000000	-2.540239	1.061349
C	0.000000	-3.180381	-0.185453

C	0.000000	-2.407643	-1.358575
C	0.000000	-1.018747	-1.289860
H	0.000000	-0.647215	2.093970
H	0.000000	-3.107431	1.984321
O	0.000000	-4.529809	-0.362712
H	0.000000	-2.913993	-2.318363
H	0.000000	-0.426265	-2.198320
Cl	0.000000	1.371691	0.047156
C	0.000000	-5.373868	0.792891
C	0.000000	-6.815469	0.312874
H	-0.890416	-5.163370	1.400892
H	0.890416	-5.163370	1.400892
H	0.000000	-7.491210	1.174067
H	0.888633	-7.022992	-0.290358
H	-0.888633	-7.022992	-0.290358
N	0.000000	4.716345	0.141656
C	1.143915	5.413603	0.141656
C	1.199503	6.809778	0.141895
C	0.000000	7.523026	0.142098
C	-1.199503	6.809778	0.141895
C	-1.143915	5.413603	0.141656
H	2.060214	4.826752	0.141345
H	2.158282	7.318224	0.141867
H	0.000000	8.608973	0.142308
H	-2.158282	7.318224	0.141867
H	-2.060214	4.826752	0.141345

Electronic energy:	-1093.91750095 (-1093.91750091)
Zero-point correction=	0.239256
Sum of electronic and zero-point Energies=	-1093.678245
Sum of electronic and thermal Energies=	-1093.661897
Sum of electronic and thermal Enthalpies=	-1093.660953
Sum of electronic and thermal Free Energies=	-1093.730744

Lowest vibration frequency: 6.5179

Counterpoise: BSSE energy = 0.000649677576

Summary of Natural Population Analysis:

Atom	No	Natural Charge	Natural Population			
			Core	Valence	Rydberg	Total
C	1	-0.07544	1.99859	4.05392	0.02293	6.07544
C	2	-0.21372	1.99900	4.19600	0.01872	6.21372
C	3	-0.28334	1.99908	4.26905	0.01522	6.28334
C	4	0.32094	1.99882	3.65734	0.02290	5.67906
C	5	-0.23351	1.99909	4.21781	0.01661	6.23351
C	6	-0.21661	1.99900	4.19934	0.01827	6.21661
H	7	0.22790	0.00000	0.77041	0.00170	0.77210
H	8	0.22350	0.00000	0.77466	0.00184	0.77650
O	9	-0.55774	1.99970	6.53721	0.02083	8.55774
H	10	0.22758	0.00000	0.77071	0.00171	0.77242
H	11	0.22907	0.00000	0.76934	0.00159	0.77093
Cl	12	-0.00056	9.99964	6.97825	0.02267	17.00056
C	13	-0.03606	1.99925	4.01820	0.01862	6.03606
C	14	-0.59420	1.99938	4.58625	0.00858	6.59420
H	15	0.17432	0.00000	0.82322	0.00246	0.82568
H	16	0.17432	0.00000	0.82322	0.00246	0.82568
H	17	0.21058	0.00000	0.78827	0.00115	0.78942
H	18	0.20889	0.00000	0.78943	0.00168	0.79111
H	19	0.20889	0.00000	0.78943	0.00168	0.79111
N	20	-0.46944	1.99945	5.44767	0.02233	7.46944
C	21	0.04904	1.99923	3.92379	0.02793	5.95096
C	22	-0.25010	1.99917	4.23441	0.01652	6.25010
C	23	-0.17130	1.99918	4.15460	0.01752	6.17130
C	24	-0.25010	1.99917	4.23441	0.01652	6.25010
C	25	0.04904	1.99923	3.92379	0.02793	5.95096
H	26	0.19482	0.00000	0.80372	0.00147	0.80518
H	27	0.22054	0.00000	0.77798	0.00149	0.77946
H	28	0.21737	0.00000	0.78132	0.00131	0.78263
H	29	0.22054	0.00000	0.77798	0.00149	0.77946
H	30	0.19482	0.00000	0.80372	0.00147	0.80518
=====						
* Total *		0.00000	39.98697	83.67544	0.33759	124.00000

Contributions to pi population at N atom in pyridine:

Val(2p) 1.17040
 Ryd(3p) 0.00071
 Ryd(4p) 0.00033
 Ryd(5p) 0.00003

Total: 1.17147

 NMR properties

Electronic energy (NMR): -1093.91750095

1	C	Isotropic =	39.3512	Anisotropy =	112.8807
2	C	Isotropic =	44.0309	Anisotropy =	161.8641
3	C	Isotropic =	65.0511	Anisotropy =	159.7368
4	C	Isotropic =	13.2334	Anisotropy =	144.5902
5	C	Isotropic =	56.1533	Anisotropy =	150.2302
6	C	Isotropic =	43.3727	Anisotropy =	161.6931
7	H	Isotropic =	23.9133	Anisotropy =	8.2133
8	H	Isotropic =	24.5692	Anisotropy =	8.1192
9	O	Isotropic =	193.5023	Anisotropy =	70.3399
10	H	Isotropic =	24.3401	Anisotropy =	6.2741
11	H	Isotropic =	23.9405	Anisotropy =	7.8010
12	Cl	Isotropic =	686.0261	Anisotropy =	528.0881
13	C	Isotropic =	113.4400	Anisotropy =	63.2440
14	C	Isotropic =	165.9397	Anisotropy =	25.6094
15	H	Isotropic =	27.6255	Anisotropy =	5.3967
16	H	Isotropic =	27.6255	Anisotropy =	5.3967
17	H	Isotropic =	30.2506	Anisotropy =	10.0578
18	H	Isotropic =	30.0278	Anisotropy =	8.2155
19	H	Isotropic =	30.0278	Anisotropy =	8.2155
20	N	Isotropic =	-102.1045	Anisotropy =	588.2228
21	C	Isotropic =	21.3828	Anisotropy =	180.9788
22	C	Isotropic =	50.8395	Anisotropy =	183.1583
23	C	Isotropic =	37.8117	Anisotropy =	211.0535
24	C	Isotropic =	50.8395	Anisotropy =	183.1583
25	C	Isotropic =	21.3828	Anisotropy =	180.9788
26	H	Isotropic =	22.4792	Anisotropy =	8.6712
27	H	Isotropic =	23.9817	Anisotropy =	5.3409
28	H	Isotropic =	23.5403	Anisotropy =	4.9977
29	H	Isotropic =	23.9817	Anisotropy =	5.3409
30	H	Isotropic =	22.4792	Anisotropy =	8.6712

=====

X = Cl R = CH3

Energies and NBO populations

Number of atoms: 26
 Full point group CS NOp 2

C	0.000000	-0.565336	0.026605
C	0.000000	-1.312020	1.202577
C	0.000000	-2.708080	1.122757
C	0.000000	-3.370075	-0.111727
C	0.000000	-2.588137	-1.277652
C	0.000000	-1.194209	-1.219657
H	0.000000	-0.812988	2.165637
H	0.000000	-3.287161	2.042563
C	0.000000	-4.880235	-0.195240
H	0.000000	-3.073926	-2.250339
H	0.000000	-0.601802	-2.128288
Cl	0.000000	1.195750	0.110073
H	0.000000	-5.332139	0.800322
H	-0.882740	-5.245740	-0.732107
H	0.882740	-5.245740	-0.732107
N	0.000000	4.535869	0.173089
C	1.143928	5.233139	0.173089
C	1.199509	6.629303	0.173128
C	0.000000	7.342539	0.173175
C	-1.199509	6.629303	0.173128
C	-1.143928	5.233139	0.173089
H	2.060248	4.646319	0.173000
H	2.158287	7.137741	0.173131

```

H      0.000000    8.428483    0.173234
H     -2.158287    7.137741    0.173131
H     -2.060248    4.646319    0.173000

```

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Electronic energy:                               -979.39336250 (  -979.39336250)
Zero-point correction=                           0.205825
Sum of electronic and zero-point Energies=       -979.187537
Sum of electronic and thermal Energies=          -979.174150
Sum of electronic and thermal Enthalpies=        -979.173206
Sum of electronic and thermal Free Energies=      -979.236786

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Lowest vibration frequency: -30.5971

Counterpoise: BSSE energy = 0.000617787437

Summary of Natural Population Analysis:

		Natural Population				
Atom	No	Natural Charge	Core	Valence	Rydberg	Total
C	1	-0.05363	1.99860	4.03195	0.02308	6.05363
C	2	-0.22779	1.99898	4.20974	0.01906	6.22779
C	3	-0.19681	1.99907	4.18144	0.01630	6.19681
C	4	-0.03827	1.99907	4.02389	0.01531	6.03827
C	5	-0.19430	1.99907	4.17882	0.01641	6.19430
C	6	-0.22642	1.99899	4.20850	0.01893	6.22642
H	7	0.22770	0.00000	0.77058	0.00172	0.77230
H	8	0.21365	0.00000	0.78473	0.00162	0.78635
C	9	-0.59674	1.99932	4.58744	0.00998	6.59674
H	10	0.21376	0.00000	0.78467	0.00157	0.78624
H	11	0.22764	0.00000	0.77064	0.00172	0.77236
Cl	12	0.00235	9.99963	6.97541	0.02261	16.99765
H	13	0.21027	0.00000	0.78845	0.00128	0.78973
H	14	0.21660	0.00000	0.78206	0.00134	0.78340
H	15	0.21660	0.00000	0.78206	0.00134	0.78340
N	16	-0.46972	1.99945	5.44798	0.02228	7.46972
C	17	0.04910	1.99923	3.92377	0.02790	5.95090
C	18	-0.25003	1.99917	4.23435	0.01651	6.25003
C	19	-0.17121	1.99918	4.15451	0.01751	6.17121
C	20	-0.25003	1.99917	4.23435	0.01651	6.25003
C	21	0.04910	1.99923	3.92377	0.02790	5.95090
H	22	0.19479	0.00000	0.80374	0.00147	0.80521
H	23	0.22059	0.00000	0.77793	0.00149	0.77941
H	24	0.21742	0.00000	0.78128	0.00131	0.78258
H	25	0.22059	0.00000	0.77793	0.00149	0.77941
H	26	0.19479	0.00000	0.80374	0.00147	0.80521
=====						
* Total *		0.00000	35.98815	71.72373	0.28811	108.00000

Contributions to pi population at N atom in pyridine:

```

Val( 2p)    1.17062
Ryd( 3p)    0.00071
Ryd( 4p)    0.00033
Ryd( 5p)    0.00003

```

Total: 1.17169

NMR properties

Electronic energy (NMR): -979.39336250

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1 C   Isotropic = 31.9568   Anisotropy = 124.0752
2 C   Isotropic = 44.8454   Anisotropy = 160.4289
3 C   Isotropic = 43.3129   Anisotropy = 163.0132
4 C   Isotropic = 35.2087   Anisotropy = 204.7690
5 C   Isotropic = 43.2882   Anisotropy = 173.9594
6 C   Isotropic = 44.7100   Anisotropy = 160.9874
7 H   Isotropic = 23.9504   Anisotropy = 8.0106
8 H   Isotropic = 24.1040   Anisotropy = 6.4765
9 C   Isotropic = 159.7698   Anisotropy = 41.0857
10 H  Isotropic = 24.0032   Anisotropy = 7.0222
11 H  Isotropic = 23.8980   Anisotropy = 8.0818
12 Cl Isotropic = 674.4166   Anisotropy = 527.7188
13 H  Isotropic = 29.4673   Anisotropy = 7.5835
14 H  Isotropic = 28.9648   Anisotropy = 8.4437

```


C	5	-0.19637	1.99906	4.18131	0.01600	6.19637
C	6	-0.22689	1.99899	4.20879	0.01911	6.22689
H	7	0.22782	0.00000	0.77050	0.00168	0.77218
H	8	0.21393	0.00000	0.78434	0.00173	0.78607
C	9	-0.24321	1.99920	4.22835	0.01566	6.24321
H	10	0.21408	0.00000	0.78420	0.00172	0.78592
H	11	0.22732	0.00000	0.77100	0.00168	0.77268
Cl	12	0.00247	9.99963	6.97531	0.02258	16.99753
H	13	0.19862	0.00000	0.79838	0.00300	0.80138
C	14	-0.57029	1.99935	4.56146	0.00947	6.57029
C	15	-0.57029	1.99935	4.56146	0.00947	6.57029
H	16	0.20318	0.00000	0.79562	0.00120	0.79682
H	17	0.19764	0.00000	0.80099	0.00137	0.80236
H	18	0.20741	0.00000	0.79158	0.00101	0.79259
H	19	0.20741	0.00000	0.79158	0.00101	0.79259
H	20	0.20318	0.00000	0.79562	0.00120	0.79682
H	21	0.19764	0.00000	0.80099	0.00137	0.80236
N	22	-0.46969	1.99945	5.44794	0.02229	7.46969
C	23	0.04908	1.99923	3.92377	0.02792	5.95092
C	24	-0.25004	1.99917	4.23435	0.01652	6.25004
C	25	-0.17121	1.99918	4.15452	0.01752	6.17121
C	26	-0.25004	1.99917	4.23435	0.01652	6.25004
C	27	0.04908	1.99923	3.92377	0.02792	5.95092
H	28	0.19480	0.00000	0.80374	0.00146	0.80520
H	29	0.22058	0.00000	0.77793	0.00149	0.77942
H	30	0.21741	0.00000	0.78128	0.00131	0.78259
H	31	0.22058	0.00000	0.77793	0.00149	0.77942
H	32	0.19480	0.00000	0.80374	0.00146	0.80520
=====						
* Total *		0.00000	39.98668	83.68971	0.32361	124.00000

Contributions to pi population at N atom in pyridine:

Val(2p) 1.17061
Ryd(3p) 0.00071
Ryd(4p) 0.00033
Ryd(5p) 0.00003

Total: 1.17168

NMR properties

Electronic energy (NMR): -1058.01400536

1	C	Isotropic =	31.5345	Anisotropy =	123.0756
2	C	Isotropic =	44.7029	Anisotropy =	159.7379
3	C	Isotropic =	44.4389	Anisotropy =	157.7650
4	C	Isotropic =	23.3905	Anisotropy =	213.5175
5	C	Isotropic =	47.1380	Anisotropy =	187.1395
6	C	Isotropic =	44.4256	Anisotropy =	161.6259
7	H	Isotropic =	23.9426	Anisotropy =	7.8888
8	H	Isotropic =	24.1068	Anisotropy =	7.1791
9	C	Isotropic =	139.6481	Anisotropy =	24.3247
10	H	Isotropic =	23.9584	Anisotropy =	9.4514
11	H	Isotropic =	23.8595	Anisotropy =	8.2741
12	Cl	Isotropic =	673.0855	Anisotropy =	529.5101
13	H	Isotropic =	28.7223	Anisotropy =	6.9767
14	C	Isotropic =	155.7735	Anisotropy =	36.8306
15	C	Isotropic =	155.7737	Anisotropy =	36.8304
16	H	Isotropic =	30.2508	Anisotropy =	10.1418
17	H	Isotropic =	30.3292	Anisotropy =	6.5829
18	H	Isotropic =	30.3041	Anisotropy =	8.5588
19	H	Isotropic =	30.3040	Anisotropy =	8.5591
20	H	Isotropic =	30.2510	Anisotropy =	10.1417
21	H	Isotropic =	30.3292	Anisotropy =	6.5830
22	N	Isotropic =	-101.9740	Anisotropy =	587.9129
23	C	Isotropic =	21.3928	Anisotropy =	180.9730
24	C	Isotropic =	50.8189	Anisotropy =	183.1849
25	C	Isotropic =	37.7960	Anisotropy =	211.1106
26	C	Isotropic =	50.8189	Anisotropy =	183.1842
27	C	Isotropic =	21.3925	Anisotropy =	180.9724
28	H	Isotropic =	22.4747	Anisotropy =	8.7063
29	H	Isotropic =	23.9774	Anisotropy =	5.3412
30	H	Isotropic =	23.5362	Anisotropy =	4.9843
31	H	Isotropic =	23.9774	Anisotropy =	5.3412
32	H	Isotropic =	22.4747	Anisotropy =	8.7063

=====

X = Cl R = H

Energies and NBO populations

Number of atoms: 23
Full point group C2V NOp 4

C	0.000000	-0.629188	0.000000
C	0.000000	-1.315733	1.215395
C	0.000000	-2.712382	1.217297
C	0.000000	-3.382085	0.000000
C	0.000000	-2.712382	-1.217297
C	0.000000	-1.315733	-1.215395
H	0.000000	-0.767899	2.150966
H	0.000000	-3.272603	2.145871
F	0.000000	-4.735618	0.000000
H	0.000000	-3.272603	-2.145871
H	0.000000	-0.767899	-2.150966
Cl	0.000000	1.130611	0.000000
N	0.000000	4.424939	0.000000
C	1.144030	5.122176	0.000000
C	1.199529	6.518271	0.000000
C	0.000000	7.231466	0.000000
C	-1.199529	6.518271	0.000000
C	-1.144030	5.122176	0.000000
H	2.060419	4.535473	0.000000
H	2.158275	7.026718	0.000000
H	0.000000	8.317387	0.000000
H	-2.158275	7.026718	0.000000
H	-2.060419	4.535473	0.000000

Electronic energy:	-1039.32304625 (-1039.32304625)
Zero-point correction=	0.170632
Sum of electronic and zero-point Energies=	-1039.152414
Sum of electronic and thermal Energies=	-1039.140188
Sum of electronic and thermal Enthalpies=	-1039.139244
Sum of electronic and thermal Free Energies=	-1039.197216

Lowest vibration frequency: -8.9286

Counterpoise: BSSE energy = 0.000689659556

Summary of Natural Population Analysis:

		Natural Population				
Atom	No	Natural Charge	Core	Valence	Rydberg	Total
C	1	-0.06282	1.99861	4.04104	0.02317	6.06282
C	2	-0.21340	1.99900	4.19625	0.01815	6.21340
C	3	-0.25817	1.99906	4.24242	0.01670	6.25817
C	4	0.40177	1.99849	3.57337	0.02636	5.9823
C	5	-0.25817	1.99906	4.24242	0.01670	6.25817
C	6	-0.21340	1.99900	4.19625	0.01815	6.21340
H	7	0.23323	0.00000	0.76514	0.00163	0.76677
H	8	0.23579	0.00000	0.76265	0.00156	0.76421
F	9	-0.35410	1.99993	7.34614	0.00804	9.35410
H	10	0.23579	0.00000	0.76265	0.00156	0.76421
H	11	0.23323	0.00000	0.76514	0.00163	0.76677
Cl	12	0.01419	9.99963	6.96354	0.02264	16.98581
N	13	-0.47145	1.99945	5.44966	0.02235	7.47145
C	14	0.04916	1.99923	3.92375	0.02787	5.95084
C	15	-0.24961	1.99917	4.23394	0.01650	6.24961
C	16	-0.17061	1.99918	4.15392	0.01750	6.17061
C	17	-0.24961	1.99917	4.23394	0.01650	6.24961
C	18	0.04916	1.99923	3.92375	0.02787	5.95084
H	19	0.19472	0.00000	0.80381	0.00147	0.80528
H	20	0.22092	0.00000	0.77760	0.00148	0.77908
H	21	0.21777	0.00000	0.78092	0.00130	0.78223
H	22	0.22092	0.00000	0.77760	0.00148	0.77908
H	23	0.19472	0.00000	0.80381	0.00147	0.80528
=====						
* Total *		0.00000	35.98822	71.71970	0.29208	108.00000

Contributions to pi population at N atom in pyridine:

Val(2p) 1.17208
 Ryd(3p) 0.00071
 Ryd(4p) 0.00033
 Ryd(5p) 0.00003

Total: 1.17315

 NMR properties

Electronic energy (NMR): -1039.32304625

1	C	Isotropic =	34.3613	Anisotropy =	124.5064
2	C	Isotropic =	42.8546	Anisotropy =	163.7865
3	C	Isotropic =	58.1468	Anisotropy =	155.4927
4	C	Isotropic =	7.9797	Anisotropy =	105.5043
5	C	Isotropic =	58.1468	Anisotropy =	155.4927
6	C	Isotropic =	42.8546	Anisotropy =	163.7865
7	H	Isotropic =	23.8425	Anisotropy =	7.8244
8	H	Isotropic =	24.2352	Anisotropy =	4.5838
9	F	Isotropic =	294.5829	Anisotropy =	99.7427
10	H	Isotropic =	24.2352	Anisotropy =	4.5838
11	H	Isotropic =	23.8425	Anisotropy =	7.8244
12	Cl	Isotropic =	671.8913	Anisotropy =	528.6766
13	N	Isotropic =	-101.3641	Anisotropy =	586.1500
14	C	Isotropic =	21.4994	Anisotropy =	180.9615
15	C	Isotropic =	50.7247	Anisotropy =	183.3783
16	C	Isotropic =	37.6933	Anisotropy =	211.4531
17	C	Isotropic =	50.7247	Anisotropy =	183.3783
18	C	Isotropic =	21.4994	Anisotropy =	180.9615
19	H	Isotropic =	22.4866	Anisotropy =	8.6153
20	H	Isotropic =	23.9697	Anisotropy =	5.3257
21	H	Isotropic =	23.5267	Anisotropy =	5.0196
22	H	Isotropic =	23.9697	Anisotropy =	5.3257
23	H	Isotropic =	22.4866	Anisotropy =	8.6153

=====

X = Cl R = Cl

 Energies and NBO populations

Number of atoms: 23
 Full point group C2V NOp 4

C	0.000000	-0.615804	0.000000
C	0.000000	-1.303177	1.214455
C	0.000000	-2.699482	1.215484
C	0.000000	-3.382670	0.000000
C	0.000000	-2.699482	-1.215484
C	0.000000	-1.303177	-1.214455
H	0.000000	-0.757014	2.151147
H	0.000000	-3.246002	2.151894
Cl	0.000000	-5.144136	0.000000
H	0.000000	-3.246002	-2.151895
H	0.000000	-0.757014	-2.151147
Cl	0.000000	1.142299	0.000000
N	0.000000	4.411328	0.000000
C	1.144063	5.108544	0.000000
C	1.199540	6.504624	0.000000
C	0.000000	7.217799	0.000000
C	-1.199540	6.504624	0.000000
C	-1.144063	5.108544	0.000000
H	2.060462	4.521853	0.000000
H	2.158280	7.013067	0.000000
H	0.000000	8.303716	0.000000
H	-2.158280	7.013067	0.000000
H	-2.060462	4.521853	0.000000

Electronic energy: -1399.64544579 (-1399.64544579)
 Zero-point correction= 0.169298
 Sum of electronic and zero-point Energies= -1399.476148
 Sum of electronic and thermal Energies= -1399.463521
 Sum of electronic and thermal Enthalpies= -1399.462577
 Sum of electronic and thermal Free Energies= -1399.522156

Lowest vibration frequency: -7.3093
Counterpoise: BSSE energy = 0.000674292501
Summary of Natural Population Analysis:

		Natural Population				
Atom	No	Natural Charge	Core	Valence	Rydberg	Total
C	1	-0.05129	1.99861	4.02936	0.02332	6.05129
C	2	-0.21495	1.99898	4.19743	0.01854	6.21495
C	3	-0.21607	1.99898	4.19867	0.01842	6.21607
C	4	-0.04731	1.99860	4.02608	0.02263	6.04731
C	5	-0.21607	1.99898	4.19867	0.01842	6.21607
C	6	-0.21496	1.99898	4.19743	0.01854	6.21496
H	7	0.23446	0.00000	0.76389	0.00165	0.76554
H	8	0.23460	0.00000	0.76375	0.00165	0.76540
Cl	9	-0.00329	9.99963	6.98371	0.01996	17.00329
H	10	0.23460	0.00000	0.76375	0.00165	0.76540
H	11	0.23446	0.00000	0.76389	0.00165	0.76554
Cl	12	0.01935	9.99963	6.95835	0.02267	16.98065
N	13	-0.47190	1.99945	5.45010	0.02235	7.47190
C	14	0.04926	1.99923	3.92368	0.02783	5.95074
C	15	-0.24949	1.99917	4.23383	0.01649	6.24949
C	16	-0.17045	1.99918	4.15377	0.01750	6.17045
C	17	-0.24949	1.99917	4.23383	0.01649	6.24949
C	18	0.04926	1.99923	3.92368	0.02783	5.95074
H	19	0.19473	0.00000	0.80380	0.00147	0.80527
H	20	0.22099	0.00000	0.77752	0.00148	0.77901
H	21	0.21785	0.00000	0.78085	0.00130	0.78215
H	22	0.22099	0.00000	0.77752	0.00148	0.77901
H	23	0.19473	0.00000	0.80380	0.00147	0.80527
=====						
* Total *		0.00000	43.98782	71.70739	0.30479	116.00000

Contributions to pi population at N atom in pyridine:

Val(2p) 1.17253
Ryd(3p) 0.00071
Ryd(4p) 0.00034
Ryd(5p) 0.00003

Total: 1.17361

NMR properties

Electronic energy (NMR): -1399.64544579

1	C	Isotropic =	30.3094	Anisotropy =	131.9130
2	C	Isotropic =	43.2424	Anisotropy =	161.2584
3	C	Isotropic =	43.8364	Anisotropy =	160.3613
4	C	Isotropic =	33.0881	Anisotropy =	120.9988
5	C	Isotropic =	43.8364	Anisotropy =	160.3613
6	C	Isotropic =	43.2424	Anisotropy =	161.2584
7	H	Isotropic =	23.8769	Anisotropy =	7.9017
8	H	Isotropic =	23.9452	Anisotropy =	7.2981
9	Cl	Isotropic =	686.8304	Anisotropy =	503.3459
10	H	Isotropic =	23.9452	Anisotropy =	7.2981
11	H	Isotropic =	23.8769	Anisotropy =	7.9017
12	Cl	Isotropic =	664.8804	Anisotropy =	519.9607
13	N	Isotropic =	-101.1319	Anisotropy =	585.5892
14	C	Isotropic =	21.5355	Anisotropy =	180.8999
15	C	Isotropic =	50.7048	Anisotropy =	183.4272
16	C	Isotropic =	37.6487	Anisotropy =	211.4858
17	C	Isotropic =	50.7048	Anisotropy =	183.4272
18	C	Isotropic =	21.5355	Anisotropy =	180.8999
19	H	Isotropic =	22.4930	Anisotropy =	8.6803
20	H	Isotropic =	23.9692	Anisotropy =	5.3345
21	H	Isotropic =	23.5266	Anisotropy =	5.0117
22	H	Isotropic =	23.9692	Anisotropy =	5.3345
23	H	Isotropic =	22.4930	Anisotropy =	8.6803

=====

X = Cl R = OCF3

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Energies and NBO populations

Number of atoms: 27

Full point group

C1

NOp 1

C	-0.302942	0.533297	0.010942
C	-0.210076	1.048964	-1.283116
C	0.055949	2.407340	-1.461633
C	0.226937	3.216640	-0.342598
C	0.125650	2.711680	0.950295
C	-0.139347	1.352965	1.129215
H	-0.347294	0.399608	-2.140221
H	0.125637	2.833803	-2.456176
O	0.422052	4.599488	-0.526994
H	0.247952	3.370077	1.803065
H	-0.222513	0.937222	2.126935
Cl	-0.643437	-1.176012	0.234538
C	1.692933	5.049871	-0.582570
F	2.380610	4.819441	0.553530
F	1.649155	6.366548	-0.782720
F	2.402296	4.493644	-1.583841
N	-1.124747	-4.347166	0.535040
C	-2.268971	-5.044270	0.535040
C	-2.324470	-6.440351	0.534793
C	-1.124879	-7.153342	0.534841
C	0.074734	-6.440282	0.535088
C	0.019377	-5.044290	0.535040
H	-3.185564	-4.458013	0.534955
H	-3.283281	-6.948657	0.534438
H	-1.124744	-8.239255	0.534622
H	1.033376	-6.948893	0.534949
H	0.935617	-4.457209	0.534527

Electronic energy: -1352.36289519 (-1352.36290264)
 Zero-point correction= 0.187528
 Sum of electronic and zero-point Energies= -1352.175367
 Sum of electronic and thermal Energies= -1352.159531
 Sum of electronic and thermal Enthalpies= -1352.158587
 Sum of electronic and thermal Free Energies= -1352.225634

Lowest vibration frequency: -9.3817

Counterpoise: BSSE energy = 0.000736166393

Summary of Natural Population Analysis:

Atom	No	Natural Charge	Natural Population			
			Core	Valence	Rydberg	Total
C	1	-0.04686	1.99862	4.02492	0.02332	6.04686
C	2	-0.21670	1.99900	4.19942	0.01828	6.21670
C	3	-0.21447	1.99904	4.19853	0.01690	6.21447
C	4	0.25995	1.99868	3.71817	0.02320	5.74005
C	5	-0.21550	1.99904	4.19965	0.01681	6.21550
C	6	-0.21666	1.99900	4.19937	0.01830	6.21666
H	7	0.23533	0.00000	0.76304	0.00163	0.76467
H	8	0.23736	0.00000	0.76096	0.00168	0.76264
O	9	-0.56390	1.99965	6.54705	0.01719	8.56390
H	10	0.23767	0.00000	0.76064	0.00169	0.76233
H	11	0.23539	0.00000	0.76299	0.00163	0.76461
Cl	12	0.02384	9.99963	6.95387	0.02266	16.97616
C	13	1.28121	1.99964	2.64816	0.07098	4.71879
F	14	-0.35429	1.99991	7.34776	0.00661	9.35429
F	15	-0.33590	1.99991	7.32965	0.00634	9.33590
F	16	-0.35346	1.99991	7.34693	0.00662	9.35346
N	17	-0.47272	1.99945	5.45085	0.02242	7.47272
C	18	0.04966	1.99923	3.92336	0.02775	5.95034
C	19	-0.24939	1.99917	4.23371	0.01651	6.24939
C	20	-0.17020	1.99918	4.15352	0.01749	6.17020
C	21	-0.24923	1.99917	4.23359	0.01647	6.24923
C	22	0.04907	1.99923	3.92385	0.02785	5.95093
H	23	0.19460	0.00000	0.80399	0.00142	0.80540
H	24	0.22112	0.00000	0.77740	0.00148	0.77888
H	25	0.21797	0.00000	0.78073	0.00130	0.78203

H	26	0.22115	0.00000	0.77736	0.00149	0.77885
H	27	0.19494	0.00000	0.80352	0.00154	0.80506
=====						
* Total *		0.00000	43.98745	95.62300	0.38955	140.00000

Contributions to pi population at N atom in pyridine:

Val(2p)	1.17335
Ryd(3p)	0.00072
Ryd(4p)	0.00033
Ryd(5p)	0.00003

Total:	1.17443
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NMR properties

Electronic energy (NMR): -1352.36289519

1	C	Isotropic =	28.9296	Anisotropy =	133.8644
2	C	Isotropic =	43.0873	Anisotropy =	163.4936
3	C	Isotropic =	48.9323	Anisotropy =	165.7374
4	C	Isotropic =	23.9033	Anisotropy =	112.0735
5	C	Isotropic =	49.0459	Anisotropy =	166.3582
6	C	Isotropic =	43.2458	Anisotropy =	163.6983
7	H	Isotropic =	23.7749	Anisotropy =	8.0143
8	H	Isotropic =	24.0289	Anisotropy =	6.8015
9	O	Isotropic =	168.2266	Anisotropy =	89.4486
10	H	Isotropic =	23.9760	Anisotropy =	6.9712
11	H	Isotropic =	23.7539	Anisotropy =	7.9261
12	Cl	Isotropic =	659.4290	Anisotropy =	522.8748
13	C	Isotropic =	47.3609	Anisotropy =	10.8939
14	F	Isotropic =	240.1274	Anisotropy =	166.1093
15	F	Isotropic =	241.1578	Anisotropy =	170.2513
16	F	Isotropic =	239.7250	Anisotropy =	165.9679
17	N	Isotropic =	-100.9009	Anisotropy =	585.2597
18	C	Isotropic =	21.5782	Anisotropy =	180.6296
19	C	Isotropic =	50.6227	Anisotropy =	183.0022
20	C	Isotropic =	37.7556	Anisotropy =	212.0828
21	C	Isotropic =	50.6197	Anisotropy =	183.3533
22	C	Isotropic =	21.6806	Anisotropy =	181.4287
23	H	Isotropic =	22.4962	Anisotropy =	8.2633
24	H	Isotropic =	23.9805	Anisotropy =	5.1806
25	H	Isotropic =	23.5022	Anisotropy =	5.0154
26	H	Isotropic =	23.9419	Anisotropy =	5.5042
27	H	Isotropic =	22.4699	Anisotropy =	9.2892

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X = Cl R = COOCH3

Energies and NBO populations

Number of atoms: 29
Full point group CS NOp 2

C	0.000000	-0.431432	0.036346
C	0.000000	-1.147700	1.235996
C	0.000000	-2.539594	1.192801
C	0.000000	-3.215216	-0.036049
C	0.000000	-2.478488	-1.229886
C	0.000000	-1.085089	-1.198046
H	0.000000	-0.621512	2.183985
H	0.000000	-3.116231	2.111647
C	0.000000	-4.708153	-0.017457
H	0.000000	-2.994629	-2.182893
H	0.000000	-0.510281	-2.117382
Cl	0.000000	1.323841	0.079878
O	0.000000	-5.238008	-1.254514
C	0.000000	-6.674923	-1.323911
H	0.000000	-6.913688	-2.386715
H	-0.891506	-7.079412	-0.839034
H	0.891506	-7.079412	-0.839034
O	0.000000	-5.379612	0.992943
N	0.000000	4.580093	0.103990
C	1.144085	5.277294	0.103990
C	1.199543	6.673362	0.103918

C	0.000000	7.386529	0.103865
C	-1.199543	6.673362	0.103918
C	-1.144085	5.277294	0.103990
H	2.060494	4.690615	0.104105
H	2.158281	7.181801	0.103986
H	0.000000	8.472444	0.103871
H	-2.158281	7.181801	0.103986
H	-2.060494	4.690615	0.104105

Electronic energy: -1167.96448198 (-1167.96445984)
 Zero-point correction= 0.221453
 Sum of electronic and zero-point Energies= -1167.743029
 Sum of electronic and thermal Energies= -1167.726142
 Sum of electronic and thermal Enthalpies= -1167.725198
 Sum of electronic and thermal Free Energies= -1167.797118

Lowest vibration frequency: 3.5024

Counterpoise: BSSE energy = 0.000660771622

Summary of Natural Population Analysis:

		Natural Population				
Atom	No	Natural Charge	Core	Valence	Rydberg	Total
C	1	-0.02672	1.99863	4.00445	0.02364	6.02672
C	2	-0.23370	1.99899	4.21604	0.01867	6.23370
C	3	-0.14566	1.99911	4.12976	0.01679	6.14566
C	4	-0.16885	1.99893	4.15210	0.01782	6.16885
C	5	-0.15140	1.99911	4.13616	0.01613	6.15140
C	6	-0.23440	1.99899	4.21676	0.01864	6.23440
H	7	0.23318	0.00000	0.76521	0.00161	0.76682
H	8	0.23788	0.00000	0.76048	0.00164	0.76212
C	9	0.80365	1.99930	3.15032	0.04673	5.19635
H	10	0.23459	0.00000	0.76385	0.00156	0.76541
H	11	0.23260	0.00000	0.76578	0.00163	0.76740
Cl	12	0.02626	9.99962	6.95145	0.02266	16.97374
O	13	-0.55064	1.99972	6.53675	0.01417	8.55064
C	14	-0.21745	1.99926	4.20243	0.01576	6.21745
H	15	0.19177	0.00000	0.80715	0.00107	0.80823
H	16	0.18996	0.00000	0.80833	0.00171	0.81004
H	17	0.18996	0.00000	0.80833	0.00171	0.81004
O	18	-0.61770	1.99975	6.60672	0.01123	8.61770
N	19	-0.47223	1.99945	5.45044	0.02234	7.47223
C	20	0.04935	1.99923	3.92362	0.02780	5.95065
C	21	-0.24942	1.99917	4.23376	0.01649	6.24942
C	22	-0.17036	1.99918	4.15368	0.01750	6.17036
C	23	-0.24942	1.99917	4.23376	0.01649	6.24942
C	24	0.04935	1.99923	3.92362	0.02780	5.95065
H	25	0.19474	0.00000	0.80379	0.00147	0.80526
H	26	0.22102	0.00000	0.77749	0.00149	0.77898
H	27	0.21787	0.00000	0.78083	0.00130	0.78213
H	28	0.22102	0.00000	0.77749	0.00149	0.77898
H	29	0.19474	0.00000	0.80379	0.00147	0.80526
* Total *		0.00000	41.98683	87.64435	0.36882	130.00000

Contributions to pi population at N atom in pyridine:

Val(2p)	1.17280
Ryd(3p)	0.00071
Ryd(4p)	0.00034
Ryd(5p)	0.00002
Total:	1.17387

 NMR properties

Electronic energy (NMR): -1167.96448198

1	C	Isotropic =	21.9938	Anisotropy =	140.1884
2	C	Isotropic =	44.5711	Anisotropy =	162.9121
3	C	Isotropic =	41.0481	Anisotropy =	196.6241
4	C	Isotropic =	45.6684	Anisotropy =	164.9862
5	C	Isotropic =	42.6244	Anisotropy =	191.4735
6	C	Isotropic =	45.0040	Anisotropy =	162.2187

7	H	Isotropic =	23.7183	Anisotropy =	8.2774
8	H	Isotropic =	22.9863	Anisotropy =	7.6766
9	C	Isotropic =	6.5032	Anisotropy =	85.3582
10	H	Isotropic =	23.2030	Anisotropy =	8.5805
11	H	Isotropic =	23.7742	Anisotropy =	8.4070
12	Cl	Isotropic =	646.4984	Anisotropy =	506.0322
13	O	Isotropic =	145.1998	Anisotropy =	141.8686
14	C	Isotropic =	126.6954	Anisotropy =	73.0670
15	H	Isotropic =	27.7568	Anisotropy =	7.5522
16	H	Isotropic =	27.5972	Anisotropy =	8.4249
17	H	Isotropic =	27.5972	Anisotropy =	8.4249
18	O	Isotropic =	-66.1035	Anisotropy =	602.5807
19	N	Isotropic =	-100.9815	Anisotropy =	585.3997
20	C	Isotropic =	21.5267	Anisotropy =	180.9012
21	C	Isotropic =	50.6891	Anisotropy =	183.4968
22	C	Isotropic =	37.6058	Anisotropy =	211.4846
23	C	Isotropic =	50.6891	Anisotropy =	183.4968
24	C	Isotropic =	21.5267	Anisotropy =	180.9012
25	H	Isotropic =	22.4772	Anisotropy =	8.6934
26	H	Isotropic =	23.9595	Anisotropy =	5.3261
27	H	Isotropic =	23.5197	Anisotropy =	4.9915
28	H	Isotropic =	23.9595	Anisotropy =	5.3261
29	H	Isotropic =	22.4772	Anisotropy =	8.6934

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X = Cl R = COCH3

Energies and NBO populations

Number of atoms: 28
Full point group CS NOp 2

C	0.000000	-0.481943	-0.013748
C	0.000000	-1.114731	1.233438
C	0.000000	-2.504654	1.286159
C	0.000000	-3.272745	0.109468
C	0.000000	-2.613057	-1.130014
C	0.000000	-1.219830	-1.198479
H	0.000000	-0.523167	2.142273
H	0.000000	-3.017289	2.242342
C	0.000000	-4.770478	0.227819
H	0.000000	-3.177307	-2.056682
H	0.000000	-0.711388	-2.156137
Cl	0.000000	1.271740	-0.088965
C	0.000000	-5.603578	-1.041774
O	0.000000	-5.305253	1.323980
H	0.000000	-6.659552	-0.770283
H	0.883869	-5.384225	-1.650889
H	-0.883869	-5.384225	-1.650889
N	0.000000	4.517036	-0.147820
C	1.144092	5.214239	-0.147820
C	1.199543	6.610294	-0.148032
C	0.000000	7.323459	-0.148192
C	-1.199543	6.610294	-0.148032
C	-1.144092	5.214239	-0.147820
H	2.060514	4.627583	-0.147352
H	2.158273	7.118748	-0.147861
H	0.000000	8.409370	-0.148232
H	-2.158273	7.118748	-0.147861
H	-2.060514	4.627583	-0.147352

Electronic energy:	-1092.72889164 (-1092.72889163)
Zero-point correction=	0.215826
Sum of electronic and zero-point Energies=	-1092.513066
Sum of electronic and thermal Energies=	-1092.497154
Sum of electronic and thermal Enthalpies=	-1092.496210
Sum of electronic and thermal Free Energies=	-1092.565212

Lowest vibration frequency: 3.6457

Counterpoise: BSSE energy = 0.000660964606

Summary of Natural Population Analysis:

Natural Population

Atom	No	Natural Charge	Core	Valence	Rydberg	Total
C	1	-0.02563	1.99863	4.00327	0.02374	6.02563
C	2	-0.23163	1.99899	4.21411	0.01853	6.23163
C	3	-0.14206	1.99910	4.12514	0.01782	6.14206
C	4	-0.16328	1.99895	4.14705	0.01728	6.16328
C	5	-0.16066	1.99911	4.14635	0.01520	6.16066
C	6	-0.23486	1.99899	4.21683	0.01904	6.23486
H	7	0.23292	0.00000	0.76547	0.00161	0.76708
H	8	0.23801	0.00000	0.76009	0.00190	0.76199
C	9	0.56552	1.99931	3.39647	0.03870	5.43448
H	10	0.22143	0.00000	0.77708	0.00149	0.77857
H	11	0.23240	0.00000	0.76594	0.00165	0.76760
Cl	12	0.02812	9.99962	6.94960	0.02266	16.97188
C	13	-0.67819	1.99930	4.67057	0.00833	6.67819
O	14	-0.57799	1.99976	6.56452	0.01371	8.57799
H	15	0.23168	0.00000	0.76645	0.00187	0.76832
H	16	0.22870	0.00000	0.76989	0.00141	0.77130
H	17	0.22870	0.00000	0.76989	0.00141	0.77130
N	18	-0.47249	1.99945	5.45069	0.02236	7.47249
C	19	0.04935	1.99923	3.92363	0.02779	5.95065
C	20	-0.24933	1.99917	4.23367	0.01649	6.24933
C	21	-0.17024	1.99918	4.15357	0.01749	6.17024
C	22	-0.24933	1.99917	4.23367	0.01649	6.24933
C	23	0.04935	1.99923	3.92363	0.02779	5.95065
H	24	0.19470	0.00000	0.80383	0.00147	0.80530
H	25	0.22109	0.00000	0.77743	0.00149	0.77891
H	26	0.21795	0.00000	0.78075	0.00130	0.78205
H	27	0.22109	0.00000	0.77743	0.00149	0.77891
H	28	0.19470	0.00000	0.80383	0.00147	0.80530
=====						
* Total *		0.00000	39.98718	81.67084	0.34198	122.00000

Contributions to pi population at N atom in pyridine:

Val(2p) 1.17305
Ryd(3p) 0.00071
Ryd(4p) 0.00034
Ryd(5p) 0.00002

Total: 1.17412

NMR properties

Electronic energy (NMR): -1092.72889164

1	C	Isotropic =	21.6608	Anisotropy =	140.6032
2	C	Isotropic =	44.1279	Anisotropy =	162.7140
3	C	Isotropic =	44.1431	Anisotropy =	200.8762
4	C	Isotropic =	38.9715	Anisotropy =	178.0566
5	C	Isotropic =	42.1423	Anisotropy =	185.7860
6	C	Isotropic =	45.1391	Anisotropy =	161.5029
7	H	Isotropic =	23.7139	Anisotropy =	8.3628
8	H	Isotropic =	23.0059	Anisotropy =	6.8433
9	C	Isotropic =	-25.4626	Anisotropy =	173.2921
10	H	Isotropic =	23.3223	Anisotropy =	8.2842
11	H	Isotropic =	23.7382	Anisotropy =	8.4444
12	Cl	Isotropic =	644.6731	Anisotropy =	503.6200
13	C	Isotropic =	153.8157	Anisotropy =	46.2593
14	O	Isotropic =	-290.5811	Anisotropy =	966.6546
15	H	Isotropic =	29.3320	Anisotropy =	6.8790
16	H	Isotropic =	28.7107	Anisotropy =	4.7318
17	H	Isotropic =	28.7107	Anisotropy =	4.7318
18	N	Isotropic =	-100.9063	Anisotropy =	585.1174
19	C	Isotropic =	21.5566	Anisotropy =	180.9173
20	C	Isotropic =	50.6662	Anisotropy =	183.5312
21	C	Isotropic =	37.5895	Anisotropy =	211.5769
22	C	Isotropic =	50.6662	Anisotropy =	183.5312
23	C	Isotropic =	21.5566	Anisotropy =	180.9173
24	H	Isotropic =	22.4775	Anisotropy =	8.6674
25	H	Isotropic =	23.9566	Anisotropy =	5.3194
26	H	Isotropic =	23.5159	Anisotropy =	4.9999
27	H	Isotropic =	23.9566	Anisotropy =	5.3194
28	H	Isotropic =	22.4775	Anisotropy =	8.6674

X = Cl R = CF3

 Energies and NBO populations

Number of atoms: 26
 Full point group CS NOp 2

C	0.000000	-0.504189	0.003614
C	0.000000	-1.154517	1.237438
C	0.000000	-2.549947	1.273832
C	0.000000	-3.279655	0.083272
C	0.000000	-2.616575	-1.150230
C	0.000000	-1.224819	-1.194102
H	0.000000	-0.580006	2.156737
H	0.000000	-3.062089	2.229191
C	0.000000	-4.786605	0.098022
H	0.000000	-3.181471	-2.077053
H	0.000000	-0.702280	-2.143921
Cl	0.000000	1.250039	-0.048432
F	0.000000	-5.295159	1.345801
F	-1.083009	-5.298069	-0.535705
F	1.083009	-5.298069	-0.535705
N	0.000000	4.469259	-0.152330
C	1.144140	5.166423	-0.152330
C	1.199552	6.562450	-0.152678
C	0.000000	7.275607	-0.152827
C	-1.199552	6.562450	-0.152678
C	-1.144140	5.166423	-0.152330
H	2.060600	4.579799	-0.151799
H	2.158292	7.070861	-0.152587
H	0.000000	8.361511	-0.152727
H	-2.158292	7.070861	-0.152587
H	-2.060600	4.579799	-0.151799

Electronic energy:	-1277.14351629 (-1277.14349265)
Zero-point correction=	0.183262
Sum of electronic and zero-point Energies=	-1276.960254
Sum of electronic and thermal Energies=	-1276.945197
Sum of electronic and thermal Enthalpies=	-1276.944253
Sum of electronic and thermal Free Energies=	-1277.011746

Lowest vibration frequency: -20.6925

Counterpoise: BSSE energy = 0.000702091274

Summary of Natural Population Analysis:

		Natural Population				
Atom	No	Natural Charge	Core	Valence	Rydberg	Total
C	1	-0.03251	1.99863	4.01008	0.02380	6.03251
C	2	-0.22492	1.99899	4.20732	0.01861	6.22492
C	3	-0.16720	1.99909	4.15158	0.01653	6.16720
C	4	-0.16100	1.99891	4.14488	0.01721	6.16100
C	5	-0.16341	1.99910	4.14743	0.01688	6.16341
C	6	-0.22357	1.99899	4.20594	0.01864	6.22357
H	7	0.23538	0.00000	0.76303	0.00159	0.76462
H	8	0.23635	0.00000	0.76205	0.00160	0.76365
C	9	1.06724	1.99913	2.87339	0.06024	4.93276
H	10	0.23241	0.00000	0.76599	0.00159	0.76759
H	11	0.23566	0.00000	0.76274	0.00160	0.76434
Cl	12	0.03049	9.99962	6.94726	0.02262	16.96951
F	13	-0.35519	1.99992	7.34890	0.00638	9.35519
F	14	-0.35851	1.99992	7.35206	0.00653	9.35851
F	15	-0.35851	1.99992	7.35206	0.00653	9.35851
N	16	-0.47310	1.99945	5.45126	0.02240	7.47310
C	17	0.04940	1.99923	3.92359	0.02778	5.95060
C	18	-0.24917	1.99917	4.23352	0.01649	6.24917
C	19	-0.17001	1.99918	4.15335	0.01749	6.17001
C	20	-0.24917	1.99917	4.23352	0.01649	6.24917
C	21	0.04940	1.99923	3.92359	0.02778	5.95060
H	22	0.19471	0.00000	0.80382	0.00147	0.80529
H	23	0.22121	0.00000	0.77730	0.00148	0.77879
H	24	0.21809	0.00000	0.78062	0.00130	0.78191

H	25	0.22121	0.00000	0.77730	0.00148	0.77879
H	26	0.19471	0.00000	0.80382	0.00147	0.80529
=====						
* Total *		0.00000	41.98764	89.65637	0.35599	132.00000

Contributions to pi population at N atom in pyridine:

Val(2p)	1.17366
Ryd(3p)	0.00071
Ryd(4p)	0.00035
Ryd(5p)	0.00002

Total:	1.17474
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NMR properties

Electronic energy (NMR): -1277.14351629

1	C	Isotropic =	23.6832	Anisotropy =	140.6107
2	C	Isotropic =	44.6953	Anisotropy =	165.3909
3	C	Isotropic =	47.4982	Anisotropy =	188.0796
4	C	Isotropic =	43.5520	Anisotropy =	165.0126
5	C	Isotropic =	45.3527	Anisotropy =	190.0943
6	C	Isotropic =	43.6433	Anisotropy =	166.8862
7	H	Isotropic =	23.7090	Anisotropy =	8.1524
8	H	Isotropic =	23.5607	Anisotropy =	8.3974
9	C	Isotropic =	43.3245	Anisotropy =	11.1560
10	H	Isotropic =	23.6108	Anisotropy =	7.9677
11	H	Isotropic =	23.6522	Anisotropy =	8.1844
12	Cl	Isotropic =	648.1771	Anisotropy =	517.0619
13	F	Isotropic =	261.1212	Anisotropy =	113.2111
14	F	Isotropic =	232.4395	Anisotropy =	148.1981
15	F	Isotropic =	232.4395	Anisotropy =	148.1981
16	N	Isotropic =	-100.6470	Anisotropy =	584.5135
17	C	Isotropic =	21.5954	Anisotropy =	180.8371
18	C	Isotropic =	50.6402	Anisotropy =	183.5792
19	C	Isotropic =	37.5455	Anisotropy =	211.6399
20	C	Isotropic =	50.6402	Anisotropy =	183.5792
21	C	Isotropic =	21.5954	Anisotropy =	180.8371
22	H	Isotropic =	22.4819	Anisotropy =	8.7353
23	H	Isotropic =	23.9543	Anisotropy =	5.3290
24	H	Isotropic =	23.5141	Anisotropy =	4.9943
25	H	Isotropic =	23.9543	Anisotropy =	5.3290
26	H	Isotropic =	22.4819	Anisotropy =	8.7353

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X = Cl	R = CN
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Energies and NBO populations

Number of atoms:	24			
Full point group		C2V	NOp	4

C	0.000000	-0.580025	0.000000
C	0.000000	-1.265573	1.217808
C	0.000000	-2.657539	1.217479
C	0.000000	-3.359784	0.000000
C	0.000000	-2.657539	-1.217479
C	0.000000	-1.265573	-1.217808
H	0.000000	-0.715693	2.152006
H	0.000000	-3.201334	2.155875
C	0.000000	-4.792848	0.000000
H	0.000000	-3.201334	-2.155875
H	0.000000	-0.715693	-2.152006
Cl	0.000000	1.172339	0.000000
N	0.000000	-5.951610	0.000000
N	0.000000	4.375484	0.000000
C	1.144216	5.072633	0.000000
C	1.199568	6.468615	0.000000
C	0.000000	7.181731	0.000000
C	-1.199568	6.468615	0.000000
C	-1.144216	5.072633	0.000000
H	2.060688	4.486066	0.000000
H	2.158260	6.977084	0.000000
H	0.000000	8.267619	0.000000

H -2.158260 6.977084 0.000000
H -2.060688 4.486066 0.000000

Electronic energy: -1032.32414890 (-1032.32414890)
Zero-point correction= 0.177563
Sum of electronic and zero-point Energies= -1032.146586
Sum of electronic and thermal Energies= -1032.133446
Sum of electronic and thermal Enthalpies= -1032.132502
Sum of electronic and thermal Free Energies= -1032.192173

Lowest vibration frequency: -4.5567

Counterpoise: BSSE energy = 0.000699445229

Summary of Natural Population Analysis:

		Natural Population				
Atom	No	Natural Charge	Core	Valence	Rydberg	Total
C	1	-0.02765	1.99864	4.00509	0.02392	6.02765
C	2	-0.22591	1.99899	4.20838	0.01854	6.22591
C	3	-0.13822	1.99911	4.12379	0.01531	6.13822
C	4	-0.19491	1.99888	4.17917	0.01686	6.19491
C	5	-0.13822	1.99911	4.12379	0.01531	6.13822
C	6	-0.22591	1.99899	4.20838	0.01854	6.22591
H	7	0.23802	0.00000	0.76042	0.00156	0.76198
H	8	0.23354	0.00000	0.76519	0.00127	0.76646
C	9	0.30686	1.99927	3.65991	0.03397	5.69314
H	10	0.23354	0.00000	0.76519	0.00127	0.76646
H	11	0.23802	0.00000	0.76042	0.00156	0.76198
Cl	12	0.03905	9.99962	6.93870	0.02263	16.96095
N	13	-0.34594	1.99959	5.32604	0.02031	7.34594
N	14	-0.47432	1.99945	5.45245	0.02242	7.47432
C	15	0.04946	1.99923	3.92356	0.02775	5.95054
C	16	-0.24889	1.99917	4.23324	0.01648	6.24889
C	17	-0.16960	1.99918	4.15294	0.01748	6.16960
C	18	-0.24889	1.99917	4.23324	0.01648	6.24889
C	19	0.04946	1.99923	3.92356	0.02775	5.95054
H	20	0.19466	0.00000	0.80386	0.00147	0.80534
H	21	0.22143	0.00000	0.77709	0.00148	0.77857
H	22	0.21832	0.00000	0.78039	0.00130	0.78168
H	23	0.22143	0.00000	0.77709	0.00148	0.77857
H	24	0.19466	0.00000	0.80386	0.00147	0.80534
=====						
* Total *		0.00000	37.98761	73.68576	0.32663	112.00000

Contributions to pi population at N atom in pyridine:

Val(2p) 1.17464
Ryd(3p) 0.00071
Ryd(4p) 0.00035
Ryd(5p) 0.00002

Total: 1.17572

NMR properties

Electronic energy (NMR): -1032.32414890

1	C	Isotropic =	22.1177	Anisotropy =	145.4172
2	C	Isotropic =	43.6752	Anisotropy =	167.7108
3	C	Isotropic =	38.5504	Anisotropy =	190.9058
4	C	Isotropic =	64.4477	Anisotropy =	149.6068
5	C	Isotropic =	38.5504	Anisotropy =	190.9058
6	C	Isotropic =	43.6752	Anisotropy =	167.7108
7	H	Isotropic =	23.7129	Anisotropy =	8.1021
8	H	Isotropic =	23.5999	Anisotropy =	6.2722
9	C	Isotropic =	55.7019	Anisotropy =	352.4979
10	H	Isotropic =	23.5999	Anisotropy =	6.2722
11	H	Isotropic =	23.7129	Anisotropy =	8.1021
12	Cl	Isotropic =	635.4230	Anisotropy =	501.9735
13	N	Isotropic =	-43.1212	Anisotropy =	418.8143
14	N	Isotropic =	-100.1296	Anisotropy =	583.1468
15	C	Isotropic =	21.6740	Anisotropy =	180.8288
16	C	Isotropic =	50.5690	Anisotropy =	183.7129
17	C	Isotropic =	37.4713	Anisotropy =	211.8935

18	C	Isotropic =	50.5690	Anisotropy =	183.7129
19	C	Isotropic =	21.6740	Anisotropy =	180.8288
20	H	Isotropic =	22.4933	Anisotropy =	8.6844
21	H	Isotropic =	23.9497	Anisotropy =	5.3256
22	H	Isotropic =	23.5071	Anisotropy =	5.0138
23	H	Isotropic =	23.9497	Anisotropy =	5.3256
24	H	Isotropic =	22.4933	Anisotropy =	8.6844

=====

X = Cl R = NO2

Energies and NBO populations

Number of atoms: 25
Full point group C2V NOp 4

C	0.000000	-0.523839	0.000000
C	0.000000	-1.208898	1.219473
C	0.000000	-2.600805	1.220719
C	0.000000	-3.276925	0.000000
C	0.000000	-2.600805	-1.220719
C	0.000000	-1.208898	-1.219473
H	0.000000	-0.658447	2.153103
H	0.000000	-3.159663	2.148415
N	0.000000	-4.748707	0.000000
H	0.000000	-3.159663	-2.148415
H	0.000000	-0.658447	-2.153103
Cl	0.000000	1.225940	0.000000
O	0.000000	-5.322659	1.084754
O	0.000000	-5.322659	-1.084754
N	0.000000	4.410788	0.000000
C	1.144287	5.107911	0.000000
C	1.199588	6.503863	0.000000
C	0.000000	7.216948	0.000000
C	-1.199588	6.503863	0.000000
C	-1.144287	5.107911	0.000000
H	2.060809	4.521437	0.000000
H	2.158259	7.012336	0.000000
H	0.000000	8.302823	0.000000
H	-2.158259	7.012336	0.000000
H	-2.060809	4.521437	0.000000

Electronic energy:	-1144.59363475 (-1144.59363475)
Zero-point correction=	0.181246
Sum of electronic and zero-point Energies=	-1144.412389
Sum of electronic and thermal Energies=	-1144.398448
Sum of electronic and thermal Enthalpies=	-1144.397504
Sum of electronic and thermal Free Energies=	-1144.459770

Lowest vibration frequency: -2.6025

Counterpoise: BSSE energy = 0.000703144081

Summary of Natural Population Analysis:

		Natural Population				
Atom	No	Natural Charge	Core	Valence	Rydberg	Total
C	1	-0.02105	1.99865	3.99835	0.02405	6.02105
C	2	-0.22263	1.99899	4.20515	0.01849	6.22263
C	3	-0.17049	1.99908	4.15320	0.01821	6.17049
C	4	0.05864	1.99879	3.92190	0.02067	5.94136
C	5	-0.17049	1.99908	4.15320	0.01821	6.17049
C	6	-0.22263	1.99899	4.20515	0.01849	6.22263
H	7	0.23935	0.00000	0.75912	0.00152	0.76065
H	8	0.25285	0.00000	0.74525	0.00190	0.74715
N	9	0.48964	1.99945	4.45591	0.05500	6.51036
H	10	0.25285	0.00000	0.74525	0.00190	0.74715
H	11	0.23935	0.00000	0.75912	0.00152	0.76065
Cl	12	0.04747	9.99962	6.93015	0.02277	16.95253
O	13	-0.39053	1.99980	6.37582	0.01491	8.39053
O	14	-0.39053	1.99980	6.37582	0.01491	8.39053
N	15	-0.47525	1.99945	5.45336	0.02244	7.47525
C	16	0.04956	1.99923	3.92349	0.02772	5.95044

C	17	-0.24867	1.99917	4.23304	0.01647	6.24867
C	18	-0.16931	1.99918	4.15265	0.01748	6.16931
C	19	-0.24867	1.99917	4.23304	0.01647	6.24867
C	20	0.04956	1.99923	3.92349	0.02772	5.95044
H	21	0.19466	0.00000	0.80387	0.00147	0.80534
H	22	0.22159	0.00000	0.77693	0.00148	0.77841
H	23	0.21847	0.00000	0.78023	0.00129	0.78153
H	24	0.22159	0.00000	0.77693	0.00148	0.77841
H	25	0.19466	0.00000	0.80387	0.00147	0.80534
=====						
* Total *		0.00000	39.98767	81.64428	0.36805	122.00000

Contributions to pi population at N atom in pyridine:

Val(2p) 1.17542
Ryd(3p) 0.00071
Ryd(4p) 0.00035
Ryd(5p) 0.00002

Total: 1.17650

NMR properties

Electronic energy (NMR): -1144.59363475

1	C	Isotropic =	18.5101	Anisotropy =	149.2264
2	C	Isotropic =	44.0109	Anisotropy =	164.0343
3	C	Isotropic =	48.0320	Anisotropy =	186.9534
4	C	Isotropic =	26.3551	Anisotropy =	114.8458
5	C	Isotropic =	48.0320	Anisotropy =	186.9534
6	C	Isotropic =	44.0109	Anisotropy =	164.0343
7	H	Isotropic =	23.6762	Anisotropy =	8.2354
8	H	Isotropic =	22.8504	Anisotropy =	5.7965
9	N	Isotropic =	-153.8258	Anisotropy =	290.0595
10	H	Isotropic =	22.8504	Anisotropy =	5.7965
11	H	Isotropic =	23.6762	Anisotropy =	8.2354
12	Cl	Isotropic =	630.0636	Anisotropy =	485.7990
13	O	Isotropic =	-308.1745	Anisotropy =	723.9803
14	O	Isotropic =	-308.1745	Anisotropy =	723.9803
15	N	Isotropic =	-99.7796	Anisotropy =	582.4600
16	C	Isotropic =	21.7005	Anisotropy =	180.8271
17	C	Isotropic =	50.5275	Anisotropy =	183.8457
18	C	Isotropic =	37.4027	Anisotropy =	212.0086
19	C	Isotropic =	50.5275	Anisotropy =	183.8457
20	C	Isotropic =	21.7005	Anisotropy =	180.8271
21	H	Isotropic =	22.4887	Anisotropy =	8.6534
22	H	Isotropic =	23.9414	Anisotropy =	5.3096
23	H	Isotropic =	23.5003	Anisotropy =	5.0097
24	H	Isotropic =	23.9414	Anisotropy =	5.3096
25	H	Isotropic =	22.4887	Anisotropy =	8.6534