### **Electronic Supplementary Information**

# Effects of the Ether Oxygen Atom in Alkyl Side Chains on the Physical Properties of Piperidinium Ionic Liquids

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#### Contents

1.	Preparation of Ionic liquids 3-7	S2-S5
2.	Linear Sweep Voltammetry of Ionic Liquids 2-7	S6
3.	DFT calculations of the Piperidinium Cations	S7-S10
4.	References	S11
5.	<sup>1</sup> H NMR spectra of Alkyl Halide Intermediates S1, S8, S12, and S15	S12-S13
6.	<sup>1</sup> H and <sup>13</sup> C NMR spectra of Ionic Liquids <b>3-7</b>	S14-S18

#### 1. Preparation of ionic liquids 3-7





CaCl<sub>2</sub> (22.5 mmol, 2.50 g), paraformaldehyde (140 mmol, 4.21 g), and butan-1-ol (101 mmol, 7.45 g) were added to anhydrous CH<sub>2</sub>Cl<sub>2</sub> (dried, 100 mL) at 0 °C and stirred for 15 min. Then trimethylsilyl chloride (Me<sub>3</sub>SiCl) (200 mmol, 21.7 g) was added dropwise to the mixture with a syringe pump (0.2 mL/min). After addition of Me<sub>3</sub>SiCl, the reaction mixture was filterated with a paper filter and solvent was removed under vacuum. Distillation of the crude product afforded 1-(chloromethoxy)butane (S1) in 49% yield as colorless liquid (52.0 mmol, 8.77 g). 1-(Chloromethoxy)butane (S1) was dissolved in CH<sub>2</sub>Cl<sub>2</sub> and stirred at 0 °C. 1-Methylpiperidine (58.3 mmol, 5.78 g) was added dropwise to the solution with a syringe pump (0.2 mL/min). Then the reaction mixture was stirred at room temperature for 1 h and solvent was removed under reduced pressure. The crude product was washed with Et<sub>2</sub>O and then dissolved in MeOH (100 mL) and activated carbon (ca 10 g) was added. After futhrer stirring for 1 day at room temperature the activated carbon was removed by paper filteration. Removal of solvent in vacuo afforded 1-(butoxymethyl)-1-methylpiperidin-1-ium chloride (S2) as a white solid in 82% yield (42.4 mmol, 9.40 g). Lithium bis((trifluoromethane)sulfonyl)amide (LiNTf<sub>2</sub>) (43.2 mmol, 12.4 g) was added to the aqueous solution of S2 and stirred for 1 h. The reaction mixture was extracted with CH<sub>2</sub>Cl<sub>2</sub> (100 mL) and the thus-obtained organic layer was washed with deionized water to remove LiCl. Removal of solvent under reduced pressure afforded [PP<sub>1BM</sub>][Tf<sub>2</sub>N] **3** in 77% yield as a colorless liquid. **1-(butoxymethyl)-1-methylpiperidin-1-ium** bis((trifluoromethyl)sulfonyl)amide ([PP<sub>1BM</sub>][Tf<sub>2</sub>N]) (3) <sup>1</sup>H NMR (600 MHz, CDCl<sub>3</sub>)  $\delta$  0.94 (t, J = 7.2 Hz, 3 H), 1.35 - 1.41 (m, 2 H), 1.60 - 1.67 (m, 3 H), 1.80 - 1.84 (m, 1 H), 1.90 (quin, J = 9.6, 4.2 Hz, 4 H), 3.03 (s, 3 H), 3.31 (quin, J = 9.6, 4.2 Hz, 2 H), 3.39 (quin, J = 6.6 Hz, 2 H), 3.79 (t, J = 6.6 Hz, 2 H), 4.61 (s, 2 H); <sup>13</sup>C NMR (150 MHz, CDCl<sub>3</sub>)  $\delta$  13.6, 18.8, 19.5, 20.8, 31.4, 45.0, 57.6, 73.4, 90.7, 119.7 (q, J = 319.2 Hz); IR (neat, cm<sup>-1</sup>) 2963, 1352, 1193, 1138, 1057. HRMS (ESI) m/z calcd for  $[C_{11}H_{24}NO]^+$ , 186.1852; found 186.1853. calcd for [C<sub>2</sub>F<sub>6</sub>NO<sub>4</sub>S<sub>2</sub>]<sup>-</sup>, 279.9178; found 279.9175.



Scheme S2.

1-Chloro-4-methoxybutane **(S3)** (132 mmol, 16.2 g) and 1-methyl piperidine (141 mmol, 14.0 g) were stirred at 120°C for 22 h to obtain piperidinium salt **S4** as a white solid in 53% yield (50.0 mmol, 15.1 g). Ion exchange of **S4** (23.6 mmol, 5.22 g) with LiNTf<sub>2</sub> (24.1 mmol, 6.90 g) afforded [PP<sub>1MB</sub>][Tf<sub>2</sub>N] **4** as a colorless liquid in 62% yield (14.5 mmol, 6.78 g). **1-(4-methoxybutyl)-1-methylpiperidin-1-ium bis((trifluoromethyl)sulfonyl)amide** ([PP<sub>1MB</sub>][Tf<sub>2</sub>N]) **(4)** <sup>1</sup>H NMR (600 MHz, Acetone-d<sub>6</sub>)  $\delta$  1.63 – 1.85 (m, 6 H), 1.89 - 1.90 (m, 4 H), 3.04 (s, 3 H), 3.33 (s, 3 H), 3.35 (t, *J* = 6.0 Hz, 6 H), 3.43 (t, *J* = 6.0 Hz, 2 H); <sup>13</sup>C NMR (150 MHz, CDCl<sub>3</sub>)  $\delta$  19.0, 1.98, 20.6, 26.0, 47.8, 58.5, 61.3, 71.3, 119.7 (q, *J* = 319.5 Hz); IR (neat, cm<sup>-1</sup>) 2951, 1352, 1192, 1137, 1057; HRMS (ESI) *m/z* calcd for [C<sub>24</sub>H<sub>48</sub>N<sub>3</sub>O<sub>6</sub>S<sub>2</sub>]<sup>+</sup>: [2M<sup>+</sup>+Tf<sub>2</sub>N]<sup>+</sup>, 652.2883; found, 652.2868.



Lithium aluminum hydride (LiAlH<sub>4</sub>) (8.84 g, 233 mmol) was place in a dry flask under argon atomosphere and dry  $Et_2O$  (350 mL) was added and stirred at room temperature.  $Et_2O$  solution of ethyl 3-ethoxypropanoate (S5) (51.5 g, 352 mmol) was added dropwise to the mixture and stirred for 2 h. After completion of the reaction confirmed by gas chromatography tetrahydrofuran (THF) containing water was added slowly at 0 °C. Then NaSO<sub>4</sub> was added to remove water and the reaction mixture was filtered. Removal of solvent at reduced pressure afforded 3-ethoxypropan-1-ol (S6) as a colorless liquid in 76% yield (266 mmol, 27.7 g). THF solution of S6 was place in flask and then NaOH (12 g, 300 mmol) was added with water (75 mL). After stirring at 0 °C, THF solution of para-toluenesulfonyl chloride (TsCl) (45.9 g, 240 mmol) was added at 0 °C. The reation was quenched with 6 M HCl aqueous solution after further stirring at room temperature for overnight. The reaction mixture was extraceted with CH<sub>2</sub>Cl<sub>2</sub> and the organic layer was concentrated *in vacuo* to obtain 3-ethoxypropyl 4-methylbenzenesulfonate (S7) in 74% yield (38.58 g, 148 mmol). S7 was dissolved in acetone and lithium bromide (LiBr) (15.4 g, 178 mmol) was added in one portion. The reaction mixture was heated at 65 °C and stirred for 15 h. LiBr was removed by paper filteration and solvent was removed under reduced pressure. The crude product was diluted with  $CH_2Cl_2$ and washed with water. Removal of solvent in vacuo afforded 1-bromo-3-ethoxypropane (S8) as a colorless liquid in quantitative yield (28.2 g, 169 mmol). Bromide S8 (4.98 g, 29.8 mmol) was dissolved in CH<sub>3</sub>CN and 1-methyl piperidine (3.57 g, 36 mmol) was added to the solution. After stirring of the reaction mixture at 60 °C for 3 days, solvent was removed under reduced pressure and dried in vacuo. The crude product was recrystallized with mix-solvent of 2-propanol and  $Et_2O$  to afford 1-(3-ethoxypropyl)-1-methylpiperidin-1-ium bromide (S9) in 86% (6.79 g, 25.5 mmol). Ion exchange of **S9** (10.3 mmol, 2.75 g) with LiNTf<sub>2</sub> (10.5 mmol, 3.02 g) afforded  $[PP_{1EP}][Tf_2N]$  5 as a colorless liquid in 95% yield (9.8 mmol, 4.58 g). 1-(3-ethoxypropyl)-1-methylpiperidin**1-ium bis((trifluoromethyl)sulfonyl)amide ([PP<sub>1EP</sub>][Tf<sub>2</sub>N]) (5)** <sup>1</sup>H NMR (600 MHz, CDCl<sub>3</sub>)  $\delta$  1.18 (t, *J* = 6.6 Hz, 3 H), 1.57 (s, 3 H), 1.71 – 1.81 (m, 2 H), 1.92 (pseudo-t, *J* = 5.4 Hz, 4 H), 1.97 – 2.02 (m, 2 H), 3.07 (s, 3 H), 3.38 (pseudo-t, *J* = 6.0 Hz, 4 H), 3.45 – 3.48 (m, 2 H), 3.48 (q, *J* = 7.2 Hz 2 H), 3.52 (pseudo-t, *J* = 6.0 Hz, 2 H); <sup>13</sup>C NMR (150 MHz, CDCl<sub>3</sub>)  $\delta$  14.9, 19.8, 20.5, 22.5, 48.1, 61.5, 61.6, 66.1, 66.4, 119.8 (q, *J* = 319.4 Hz); IR (neat, cm<sup>-1</sup>) 2977, 2875, 1352, 1193, 1137, 1057; HRMS (ESI) *m/z* calcd for [C<sub>11</sub>H<sub>24</sub>NO]<sup>+</sup>, 186.1852; found 186.1846. calcd for [C<sub>2</sub>F<sub>6</sub>NO<sub>4</sub>S<sub>2</sub>]<sup>-</sup>, 279.9178; found 279.9180.



THF solution of 2-propoxyethan-1-ol (S10) (20.4 g, 200 mmol) was place in flask and then NaOH (12 g, 300 mmol) was added with water (75 mL). After stirring at 0 °C, THF solution of para-toluenesulfonyl chloride (TsCl) (45.9 g, 240 mmol) was added at 0 °C. The reation was quenched with 6 M HCl aqueous solution after further stirring at room temperature for overnight. The reaction mixture was extraceted with CH<sub>2</sub>Cl<sub>2</sub> and the organic layer was concentrated in vacuo to obtain 2-propoxyethyl 4-methylbenzenesulfonate (S11) in 93% yield (48.0 g, 185.7 mmol). Tosylate S11 (48.0 g, 185.7 mmol) was dissolved in acetone and LiBr (19.4 g, 223 mmol) was added in one portion. The reaction mixture was heated at 65 °C and stirred for 15 h. LiBr was removed by paper filteration and solvent was removed under reduced pressure. The crude product was diluted with CH<sub>2</sub>Cl<sub>2</sub> and washed with water. Removal of solvent in vacuo and distillation afforded 1-(2-bromoethoxy)propane (S12) as a colorless liquid in 58% yield (18.1 g, 108 mmol). Bromide S12 (17.5 g, 105 mmol) was dissolved in CH<sub>3</sub>CN and 1-methyl piperidine (12.5 g, 126 mmol) was added to the solution. After stirring of the reaction mixture at 55 °C for 4 days, solvent was removed under reduced pressure and dried in vacuo. The crude product was treated with activated carbon to afford 1-methyl-1-(2-propoxyethyl)piperidin-1-ium bromide (S13) in 68% (18.9 g, 70.9 mmol). Ion exchange of S13 (68.2 mmol, 18.2 g) with LiNTf<sub>2</sub> (69.7 mmol, 20.0 g) afforded [PP<sub>1PE</sub>][Tf<sub>2</sub>N] 6 as a colorless liquid in 92% yield (62.9 mmol, 29.3 g). 1-methyl-1-(2-propoxyethyl)piperidin-1-ium bis((trifluoromethyl)sulfonyl)amide  $([PP_{1PE}][Tf_2N])$  (6) <sup>1</sup>H NMR (600 MHz, CDCl<sub>3</sub>)  $\delta$  0.91 (t, J = 7.2 Hz, 3 H), 1.57 (s, 3 H), 1.59 (sext, J = 7.8 Hz, 2 H), 1.71 – 1.76 (m, 2 H), 1.88 - 1.98 (m, 4 H), 3.16 (s, 3 H), 3.39 (ddd, J = 13.2, 7.2, 4.2 Hz, 2 H), 3.44 (t, J = 6.6 Hz, 2 H), 3.53 (ddd, J = 12.6, 8.4, 4.2 Hz, 2 H), 3.61 (pseudo-t, J = 4.8 Hz, 2 H), 3.84 – 3.87 (m, 2 H); <sup>13</sup>C NMR  $(150 \text{ MHz, CDCl}_3) \delta 10.4, 19.9, 20.5, 22.5, 49.5, 62.5, 63.7, 73.2, 119.8 (q, J = 319.4 \text{ Hz}); \text{ IR (neat, cm}^{-1}) 2967,$ 2880, 1352, 1193, 1137, 1057; HRMS (ESI) m/z calcd for  $[C_{11}H_{24}NO]^+$ , 186.1852; found 186.1845. calcd for [C<sub>2</sub>F<sub>6</sub>NO<sub>4</sub>S<sub>2</sub>], 279.9178; found 279.9181.



CaCl<sub>2</sub> (22.8 mmol, 2.50 g), paraformaldehyde (141 mmol, 4.24 g), and diethylene glycohol monomethylether (S14) (101 mmol, 12.1g) were added to anhydrous CH<sub>2</sub>Cl<sub>2</sub> (dried, 100 mL) at 0 °C and stirred for 15 min. Then Me<sub>3</sub>SiCl (200 mmol, 21.7 g) was added dropwise. After addition of Me<sub>3</sub>SiCl, the reaction mixture was filterated and solvent was removed under vacuum. Distillation of the crude product afforded 1-(chloromethoxy)-2-(2-methoxyethoxy)ethane (S15) in 52% yield as colorless liquid (52.0 mmol, 8.77 g). Chloride S15 was dissolved in CH<sub>2</sub>Cl<sub>2</sub> and stirred at 0 °C. 1-Methylpiperidine (62.4 mmol, 6.19 g) was added dropwise to the solution. Then the reaction mixture was stirred at room temperature for 1 h and solvent was removed under reduced pressure. The crude product was washed with Et<sub>2</sub>O and then dissolved in MeOH (130 mL) and activated carbon (ca 13 g) was added. After futhrer stirring for 1 day at room temperature the activated carbon was removed by paper filteration. Removal of solvent in vacuo afforded 1-((2-(2-methoxy-ethoxy)ethoxy)methyl)-1-methylpiperidin-1-ium chloride (S16) as a white solid in 72% yield (37.2 mmol, 9.94 g). S16 was dissolved in water and LiTf<sub>2</sub>N (40.6 mmol, 11.7 g) was added. After stirring for 1 h at room temperature, the aqueous solution was extracted with CH<sub>2</sub>Cl<sub>2</sub> (60 mL). The organic layer was washed with water to remove LiCl and evaporated to remove solvent. Further vacuum drying afforded [PP<sub>1MEEM</sub>][Tf<sub>2</sub>N] 7 as colorless liquid in 95% yield (35.2 mmol, 18.0 g). 1-((2-(2-methoxy-ethoxy)methyl)-1-methylpiperidin-1-ium bis((trifluoromethyl)sulfonyl)amide ([PP<sub>1MEEM</sub>][Tf<sub>2</sub>N]) (7) <sup>1</sup>H NMR (600 MHz, CDCl<sub>3</sub>) δ 1.62 - 1.69 (m, 1 H), 1.78 - 1.84 (m, 1 H), 1.91 (quin, J = 12.0, 6.6 Hz, 4 H), 3.04 (s, 3 H), 3.30 - 3.34 (m, 2 H), 3.52 - 3.54 (m, 2 H), 3.62 - 3.64 (m, 2 H), 3.68 - 3.69 (m, 2 H), 3.97 - 3.99 (m, 2 H), 4.71 (s, 2 H); <sup>13</sup>C NMR (150 MHz, CDCl<sub>3</sub>) δ 19.5, 20.8, 45.1, 57.5, 58.5, 70.1, 70.3, 71.7, 72.5, 90.8, 119.7 (q, J = 319.4 Hz); IR (neat, cm<sup>-1</sup>) 2948, 1353, 1193, 1137, 1057; HRMS (ESI) m/z calcd for  $[C_{12}H_{26}NO_3]^+$ , 232.1907; found 232.1909 calcd for  $[C_2F_6NO_4S_2]^-$ , 279.9178; found, 279.9175.

#### 2. Liner Sweep Voltammetry of Ionic Liquids 2-7



Figure S1. Linear Sweep Voltammtery of Ionic Liquids 2-7. (a)  $[PP_{1MEM}][Tf_2N]$  2 (b)  $[PP_{1BM}][Tf_2N]$  3 (c)  $[PP_{1MB}][Tf_2N]$  4 (d)  $[PP_{1EP}][Tf_2N]$  5 (e)  $[PP_{1PE}][Tf_2N]$  6 (f)  $[PP_{1MEEM}][Tf_2N]$  7

#### 3. DFT calculations of the Piperidinium Cations

Calculations of the cation parts of ionic liquids 1-7 were carried out with the three-parameter functional of Becke,<sup>3</sup> the correlation functional of Lee, Yang, and Parr (B3LYP),<sup>4</sup> and the 6-31G\* basis set.<sup>5</sup> Geometries were optimized and vibrational analyses were performed at the B3LYP/6-31G\* level of theory. The vibrational analyses were used to confirm energetic stability of the optimized structures. All of the calculations were carried out with the Gaussian 09 suite of programs.<sup>6</sup> Cartesian coordinates and energies of computationally characterized species are as follows:



	1	Ľ	1012 2 1
Atom	Х	Y	Z
С	-6.72903	-1.34919	-1.80049
Ν	-8.16336	-0.93941	-1.54985
С	-8.13795	-0.20113	-0.1951
С	-9.4421	0.158184	0.512775
С	-9.12136	0.928745	1.809319
С	-10.3841	1.299513	2.599676
С	-8.57683	-0.03251	-2.71904
С	-10.0804	0.143986	-2.91864
С	-10.7927	-1.20899	-3.01161
С	-10.5054	-2.03593	-1.75437
С	-9.00513	-2.22297	-1.52208
С	-10.0798	2.070409	3.891379
С	-11.3427	2.440445	4.675296
Н	-6.65918	-1.84203	-2.77047
Н	-6.09852	-0.45849	-1.79182
Н	-6.41225	-2.036	-1.01428
Н	-7.54771	-0.84892	0.459221
Н	-7.54912	0.701114	-0.38468
Н	-10.0876	0.777212	-0.11664
Н	-10.0014	-0.7467	0.771888

Piperidinium cation of [PP<sub>16</sub>][Tf<sub>2</sub>N] 1



(white: H, gray: C, blue: N)

Atom	Х	Y	Ζ
Н	-8.45957	0.321493	2.442861
Н	-8.5629	1.842682	1.563159
Н	-11.0475	1.903195	1.963561
Н	-10.9419	0.384153	2.844861
Н	-8.154	-0.50518	-3.61009
Н	-8.06018	0.91782	-2.56282
Н	-10.5183	0.755712	-2.12468
Н	-10.2012	0.713458	-3.84736
Н	-11.8707	-1.06235	-3.12574
Н	-10.4504	-1.7495	-3.90478
Н	-10.9857	-1.57736	-0.88639
Н	-10.9274	-3.0434	-1.84545
Н	-8.78877	-2.70639	-0.56557
Н	-8.59363	-2.85038	-2.31803
Н	-9.41669	1.46588	4.52561
Н	-9.51972	2.983057	3.644587
Н	-11.092	2.988655	5.58911
Н	-12.0093	3.074789	4.079
Н	-11.9061	1.546346	4.967652



(white: H, gray: C, blue: N, red: O)

Atom	Х	Y	Z
Н	-0.67321	-3.55767	-0.24171
Н	-0.90866	-6.03282	-2.89263
Н	-2.27951	-5.61946	-1.82408
Н	-0.64885	-5.87405	-1.13213
Н	0.186327	2.695306	0.006961
Н	-1.35708	2.846303	-0.85699
Н	-1.19778	4.609912	0.827446
Н	-0.88971	3.467724	2.118523
Н	-3.28301	4.079646	2.196013
Н	-3.47755	3.743616	0.480784
Н	-4.36929	1.817864	1.728697
Н	-2.91776	1.687084	2.696971
Н	-2.90452	-0.0358	0.87786
Н	-3.19648	1.222691	-0.33275
Н	-0.75149	1.233365	2.497489
Н	0.611358	0.834584	1.420186
Н	-0.65664	-0.38269	1.735909



(white: H, gray: C, blue: N, red: O)

Atom	Х	Y	Z
Н	-6.79204	0.295168	2.27848
Н	-5.01257	1.033636	0.676492
Н	-6.08781	2.283523	0.048525
Н	-9.0539	-1.35278	-2.73206
Н	-8.14167	-0.04219	-1.94804
Н	-10.2457	0.889546	-0.99536
Н	-10.4054	0.694195	-2.72751
Н	-12.4439	-0.16005	-1.45781
Н	-11.6952	-1.38952	-2.46817
Н	-11.3932	-1.07599	0.573872
Н	-12.1521	-2.40428	-0.27523
Н	-9.84315	-3.0282	0.484268
Н	-10.0557	-3.14651	-1.26967
Н	-6.16176	3.361196	2.325807
Н	-5.09148	2.11468	2.952926
Н	-3.8054	4.186138	2.407546
Н	-3.27694	2.861297	1.360826
Н	-4.35431	4.116366	0.727107



Piperidinium cation of  $[PP_{1MEM}][Tf_2N]~\textbf{2}$ 

Atom	Х	Y	Z
Ν	-1.22855	1.171807	0.416251
С	-0.8225	0.34538	-0.79309
0	-1.13278	-0.97075	-0.51776
С	-0.80766	-1.87503	-1.59343
С	-1.21421	-3.27697	-1.16098
0	-0.8878	-4.11689	-2.23686
С	-1.20216	-5.48109	-1.99781
С	-0.89981	2.633622	0.114402
С	-1.43185	3.59752	1.174466
С	-2.94197	3.427901	1.386174
С	-3.284	1.965726	1.701649
С	-2.73793	1.014409	0.639224
С	-0.45197	0.678898	1.611783
Н	-1.37885	0.744248	-1.65348
Н	0.254374	0.511379	-0.94647
Н	0.26953	-1.83897	-1.80058
Н	-1.35178	-1.58713	-2.50213
Н	-2.29342	-3.30024	-0.93413

Piperidinium cation of  $[PP_{1BM}][Tf_2N]$  3

Atom	Х	Y	Z
C	-7.61425	-2.46294	-0.91747
N	-8.80572	-1.57567	-0.65602
С	-8.51789	-0.74791	0.587551
0	-7.41284	0.030547	0.308774
С	-7.02679	0.907432	1.396077
С	-5.82233	1.720924	0.952761
С	-9.02703	-0.67336	-1.87507
С	-10.3182	0.135835	-1.78895
С	-11.541	-0.76619	-1.57725
С	-11.346	-1.66583	-0.34908
С	-10.0363	-2.45181	-0.42491
С	-5.34145	2.685224	2.048041
С	-4.12585	3.508341	1.610227
Н	-7.80072	-3.03976	-1.82387
Н	-6.73293	-1.83482	-1.03314
Н	-7.48687	-3.13653	-0.06781
Н	-9.40977	-0.15271	0.812323
Н	-8.34802	-1.46681	1.403272
Н	-7.87742	1.55805	1.646042



(white: H, gray: C, blue: N, red: O)

Atom	Х	Y	Z
Н	-10.194	-0.63506	0.977469
Н	-8.64951	0.425186	2.666701
Н	-8.74617	1.951367	1.791083
Н	-11.2181	2.035205	2.201257
Н	-11.1158	0.498662	3.087736
Н	-8.30421	-0.39857	-3.38878
Н	-8.1969	1.022432	-2.34
Н	-10.6621	0.898679	-1.92383
Н	-10.3293	0.856124	-3.64332
Н	-12.0331	-0.89601	-2.9439
Н	-10.6157	-1.60239	-3.71074
Н	-11.1794	-1.432	-0.69762
Н	-11.1328	-2.8955	-1.66089
Н	-9.00269	-2.59472	-0.35723
Н	-8.7912	-2.73557	-2.10791
Н	-10.7418	3.024997	5.649789
Н	-11.8704	3.192034	4.275461
Н	-11.7659	1.643968	5.165305



Piperidinium cation of  $[PP_{1MB}][Tf_2N]$  4

Atom	Х	Y	Z
С	-6.90971	-1.26314	-1.56352
Ν	-8.34138	-0.83345	-1.32909
С	-8.32057	-0.09826	0.026259
С	-9.62813	0.267843	0.725236
С	-9.29699	1.030907	2.020493
С	-10.5475	1.406223	2.814488
С	-8.72788	0.080998	-2.50182
С	-10.2264	0.282054	-2.71533
С	-10.9586	-1.05942	-2.81936
С	-10.6968	-1.89478	-1.56205
С	-9.202	-2.1049	-1.31422
0	-10.1189	2.096436	3.961378
С	-11.1874	2.50578	4.799009
Н	-6.83524	-1.75349	-2.53435
Н	-6.26664	-0.38164	-1.54373
Н	-6.61286	-1.95724	-0.77597
Н	-7.74057	-0.7506	0.68489
Н	-7.72479	0.800778	-0.15582
Н	-10.2647	0.89214	0.091761

Piperidinium cation of  $[PP_{1EP}][Tf_2N]$  5

Atom	Х	Y	Z
С	-6.5836	-1.23936	-1.67866
N	-8.01095	-0.84251	-1.37114
С	-7.94552	-0.15052	0.001366
С	-9.23723	0.25881	0.698146
С	-8.91722	0.981349	2.0149
0	-10.1518	1.30329	2.597436
С	-8.46438	0.10156	-2.49816
С	-9.97452	0.257958	-2.67003
С	-10.6712	-1.10136	-2.78268
С	-10.3529	-1.95163	-1.54907
С	-8.84671	-2.13255	-1.35536
С	-10.0327	1.991356	3.846183
С	-11.4289	2.287611	4.362648
Н	-6.54635	-1.7147	-2.65897
Н	-5.95786	-0.34536	-1.67727
Н	-6.23566	-1.9382	-0.91645
Н	-7.3934	-0.84856	0.637544
Н	-7.30827	0.724161	-0.16037
Н	-9.83936	0.937704	0.090729



(white: H, gray: C, blue: N, red: O)

Atom	$\mathbf{v}$	V	7
Atom		<u>ľ</u>	L
Н	-9.85382	-0.60909	0.94511
Н	-8.31792	0.331428	2.677615
Н	-8.31713	1.888449	1.821065
Н	-8.04642	-0.32844	-3.41257
Н	-7.96305	1.055066	-2.31329
Н	-10.4103	0.844789	-1.85691
Н	-10.1164	0.846377	-3.58389
Н	-11.7525	-0.96437	-2.87547
Н	-10.3379	-1.6185	-3.69295
Н	-10.8186	-1.51648	-0.66121
Н	-10.7683	-2.96064	-1.65285
Н	-8.60485	-2.63764	-0.41635
Н	-8.44983	-2.73692	-2.17593
Н	-9.47302	1.365614	4.559395
Н	-9.4607	2.921515	3.700684
Н	-11.3712	2.815248	5.320104
Н	-11.978	2.915762	3.654382
Н	-11.9908	1.360501	4.512316



Piperidinium cation of  $[PP_{1PE}][Tf_2N]$  6

Atom	Х	Y	Ζ
С	-7.30684	-1.36732	-1.57821
Ν	-8.74201	-0.92025	-1.39812
С	-8.75137	-0.1385	-0.0815
С	-10.0722	0.213808	0.614084
0	-9.6379	0.961272	1.71797
С	-10.6955	1.351544	2.606543
С	-9.09342	-0.05055	-2.6153
С	-10.5847	0.185123	-2.83645
С	-11.3534	-1.13916	-2.896
С	-11.1061	-1.94947	-1.61875
С	-9.61575	-2.18214	-1.36456
С	-10.0941	2.155053	3.749598
С	-11.1559	2.602914	4.759813
Н	-7.20895	-1.88425	-2.53289
Н	-6.65896	-0.48942	-1.56203
Н	-7.04074	-2.04093	-0.76262
Н	-8.16502	-0.73018	0.62461
Н	-8.20538	0.78773	-0.27434
Н	-10.7489	0.799815	-0.02451



Piperidinium cation of  $[PP_{1MEEM}][Tf_2N]$  7

Atom	X	Y	Ζ
C	-4.72991	3.285659	-3.02578
С	-4.63851	1.973426	-2.25011
N	-3.37872	1.882615	-1.37907
С	-2.13922	2.141611	-2.23476
С	-2.21145	3.456856	-3.00975
С	-3.47619	3.53434	-3.87484
С	-3.26834	0.473936	-0.81876
С	-3.45034	2.850244	-0.22417
0	-4.39448	0.242531	-0.05588
С	-4.41673	-1.06286	0.55729
С	-5.70317	-1.1717	1.36402
0	-5.69248	-2.45376	1.935358
С	-6.83549	-2.72057	2.74335
С	-6.70148	-4.134	3.290988
0	-7.83994	-4.3689	4.086185
С	-7.85033	-5.65841	4.670026
Н	-5.62058	3.214921	-3.65992
Н	-4.91144	4.12631	-2.34545
Н	-5.48489	1.810245	-1.5832
Н	-4.57479	1.124663	-2.9384
Н	-1.27943	2.105687	-1.56057



(white: H, gray: C, blue: N, red: O)

Atom	X	Y	Z
Н	-10.6086	-0.69239	0.935481
Н	-11.4366	1.949708	2.052868
Н	-11.2055	0.450989	2.984253
Н	-8.68304	-0.58684	-3.47574
Н	-8.53361	0.880157	-2.49541
Н	-10.9999	0.841407	-2.06537
Н	-10.6738	0.730703	-3.7828
Н	-12.4241	-0.95105	-3.01846
Н	-11.0313	-1.7154	-3.77397
Н	-11.5797	-1.45956	-0.76385
Н	-11.5621	-2.94331	-1.69306
Н	-9.42574	-2.65917	-0.39933
Н	-9.21044	-2.82925	-2.1478
Н	-9.33276	1.540881	4.245586
Н	-9.5761	3.026267	3.330819
Н	-10.6973	3.178736	5.569143
Н	-11.9158	3.238703	4.290555
Н	-11.6678	1.74547	5.212124



(white: H, gray: C, blue: N, red: O)

Atom	Х	Y	Z
Н	-2.07362	1.295557	-2.92603
Н	-1.31036	3.508299	-3.63065
Н	-2.15625	4.310356	-2.32381
Н	-3.5377	4.511325	-4.36329
Н	-3.41977	2.784563	-4.6754
Н	-3.18769	-0.20596	-1.67915
Н	-2.33683	0.434955	-0.23428
Н	-3.56869	3.862659	-0.60166
Н	-2.52332	2.773959	0.347144
Н	-4.29709	2.571956	0.400193
Н	-3.54476	-1.18335	1.212998
Н	-4.38705	-1.84079	-0.21643
Н	-6.57274	-1.02547	0.702146
Н	-5.73092	-0.38151	2.13242
Н	-6.89809	-2.00228	3.57428
Н	-7.75769	-2.63568	2.149513
Н	-6.63485	-4.85284	2.457394
Н	-5.77246	-4.22	3.878822
Н	-8.76698	-5.73441	5.25911
Н	-6.98355	-5.80962	5.331877
Н	-7.84812	-6.44922	3.903973

#### 4. References

- (1) Protocol of chloromethylation: Vo, C.-V. T.; Mitchell, T. A.; Bode, J. W. J. Am. Chem. Soc. 2011, 134, 14082-14089.
- (2) Protocols of tosylation and bromination: Kar, M.; Winther-Jensen, B.; Armand, M.; Simons, T. J.; Winther-Jensen, O.; Forsyth, M.; MacFarlane, D. R. *Electrochim. Acta* **2016**, *188*, 461-471.
- (3) Becke, A. D. J. Chem. Phys. 1993, 98, 5648.
- (4) Lee, C.; Yang, W.; Parr, R. G. Phys. Rev. B 1998, 37, 785.
- (5) Harihan, P. C.; Pople, J. A. Theor. Chim. Acta 1973, 28, 213.
- (6) Frisch, M. J.; Trucks, G. W.; Schlegel, H. B.; G. E. Scuseria, G. E.; Robb, M. A.; Cheeseman, J. R.; Scalmani, G.; Barone, V.; Mennucci, B.; Petersson, G. A.; Nakatsuji, H.; Caricato, M.; Li, X.; Hratchian, H. P.; Izmaylov, A. F.; Bloino, J.; Zheng, G.; Sonnenberg, J. L.; Hada, M.; Ehara, M.; Toyota, K.; Fukuda, R.; Hasegawa, J.; Ishida, M.; Nakajima, T.; Honda, Y.; Kitao, O.; Nakai, H.; Vreven, T.; Montgomery, J. A.; Peralta, Jr., J. E.; Ogliaro, F.; Bearpark, M.; Heyd, J. J.; Brothers, E.; Kudin, K. N.; Staroverov, V. N.; Keith, T.; Kobayashi, R.; Normand, J.; Raghavachari, K.; Rendell, A.; Burant, J. C.; Iyengar, S. S.; Tomasi, J.; Cossi, M.; Rega, N.; Millam, J. M.; Klene, M.; Knox, J. E.; Cross, J. B.; Bakken, V.; Adamo, C.; Jaramillo, J.; Gomperts, R.; Stratmann, R. E.; Yazyev, O.; Austin, A. J.; Cammi, R.; Pomelli, C.; Ochterski, J. W.; Martin, R. L.; Morokuma, K.; Zakrzewski, V. G.; Voth, G. A.; Salvador, P.; Dannenberg, J. J.; Dapprich, S.; Daniels, A. D.; Farkas, O.; Foresman, J. B.; Ortiz, J. V.; Cioslowski, J.; Fox, D. J. Gaussian 09, Revision C.01, Gaussian, Inc., Wallingford CT, (2010).

### 5. <sup>1</sup>H NMR spectra of Alkyl Halide Intermediates S1, S8, S12, and S15

These alkyl halides and their precursors in this article are all known compounds. We added <sup>1</sup>H NMR spectra of alkyl halides to show their purity.



S12





## 6. <sup>1</sup>H and <sup>13</sup>C NMR spectra of Ionic Liquids 3-7







