SUPPORTING INFORMATION

Synthesis and Antioxidant Evaluation of Lipophilic Oleuropein Aglycone Derivatives.

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Entry	ROH	Time	Yield % ^a
1 ^b	MeOH	1 h 30'	80
2 ^b	EtOH	1 h 30'	80
3°	PhOH	4 h	78
4 ^c	trans-Hexen-2-ol	4 h	75
5°	cis-Hexen-2-ol	4 h	80
6 ^c	Octanyl alcohol	4 h	72
7°	Palmityl alcohol	4 h	60
8°	Oleyl alcohol	4 h	70

Table S1. Synthesis of Oleuropein Aglicone derivatives with 10% Er(OTf)₃ by Oleuropein

^a Determinated by HPLC; ^b Oleuropein (0,92 mmol) was dissolved in MeOt/EtOH (4 mL) in the presence of Er(OTf)₃ (0.046 mmol) and refluxed at 80° C for 1h 30'.^c Oleuropein (0,92 mmol) was dissolved in acetonitrile (4 mL) in the presence of Er(OTf)₃ (0.046 mmol) and the alcohol (1,104 mmol) refluxed at 80° C for 4h.















LC-MS spectrum of Ethyl Oleuropein Aglicone (4) : LC/UV/MS (A). ESI - MS spectra (B).





Determination of LogK_{ow}

To calibrate the RP-HPLC system, a suite of four well-defined organic compounds with known Log P has been used as standards. All of the standard substances are 99% grade and were obtained from Aldrich. Stock solutions containing 10 μ g/ml of a single compound were prepared by dissolving them in mobile phase solution (40% water/CF₃COOH, PH= 2.4, 60% methanol). Multiple injections of single standards were performed to define the individual retention times and their reproducibility as well as stability of these selected compounds. Figure S1 shows the chromatograms registered to both wavelengths of 254 and 280 nm.





Capacity factor k' was calculated for each compound according to equation showed in literature.³ Table S1 gives the Log P values of standard compounds, their retention time and their capacity factors.

STANDARD	R.T.	T ₀	k'	logk'	logKow
Resorcinol	3,330	3,013	0,105210753	-0,977939869	0,59
Pyrocathecol	3,637	3,013	0,207102556	-0,683814542	1,21
Benzoic acid	5,033	3,013	0,670428145	-0,173647762	1,94
p-Cl Phenol	6,797	3,013	1,255891138	0,098951996	2,39

Table S2. Log P values of standard compounds

Dead-time t_0 was determined from the chromatogram obtained on mobile phase at the beginning of each sequence of analysis and has been defined as the time of the negative peak maximum.

Log k' were correlated with Log K_{ow} known values through a linear regression analysis³⁸ to obtain a "correlation curve" of the form

 $log K_{ow} = a log k' + b$

with a slope (*a*) of 1.628 and a regression coefficient of 0.994 (Figure S1)

Solutions containing 10 μ g/ml of each unknown substances were prepared dissolving each 3,4-DHPEA-EA derivatives in the mobile phase. Three injections of each compound were performed to find the correct retention time. Capacity coefficient k' of each derivative was calculated as for the known compounds. Log K_{ow} of each oleuropein derivative was finally found by correlation with Log k' through the same linear regression determined before.



Figure S2. Linear Plot Log k' correlated with Log K_{ow}





HPLC Chromatogram of Methyl Oleuropein Aglicone (3)







HPLC Chromatogram of Phenyl Oleuropein Aglicone (5)





HPLC Chromatogram of trans-Hexenyl Oleuropein Aglicone (6)

HPLC Chromatogram of cis-Esenyl Oleuropein Aglicone (7)







Measurement of oxygen radical absorbance capacity (ORAC_{FL})



Fig.S3 FL fluorescence decay curve induced by AAPH

Fig. S4 Linear plot of AUC vs Trolox concentrations



DPPH radical scavenging activity

Fig. S5 Absorbance induced by DPPH



Fig. S6 Linear plot DPPH Assay of Methyl Oleuropein Aglicone (3)





Fig. S7 Linear plot DPPH Assay of Ethyl Oleuropein Aglicone (4)

Fig. S8 Linear plot DPPH Assay of Phenyl Oleuropein Aglicone (5)





Fig. S9 Linear plot DPPH Assay of trans-Hexenyl Oleuropein Aglicone (6)

Fig. S10 Linear plot DPPH Assay of cis-Hexenyl Oleuropein Aglicone (7)





Fig. S11 Linear plot DPPH Assay of Palmityl Oleuropein Aglicone (8)

Fig. S12 Linear plot DPPH Assay of Octyl Oleuropein Aglicone (9)





Fig. S13 Linear plot DPPH Assay of Oleyl Oleuropein Aglicone (10)

Fig. S14 IC50 of Oleuropein Aglicone derivates in DPPH Assay

