

Supporting online materials for

**Gram-scale synthesis of single-crystalline graphene  
quantum dots derived from lignin biomass**

Zheyuan Ding, Fengfeng Li, Jialong Wen, Xiluan Wang\* and Runcang Sun

*Beijing Key Laboratory of Lignocellulosic Chemistry, Beijing Forestry University,  
Beijing, 100083, P. R. China.*

*\*Corresponding author: Tel: +86-10-62336903; Fax: +86-10-62336903*

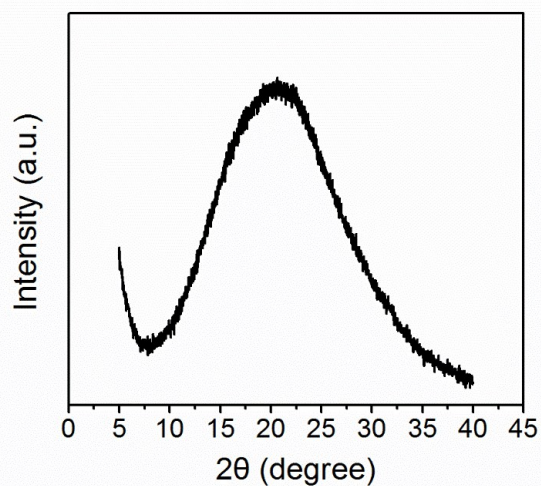
*E-mail address: wangxiluan@bjfu.edu.cn*

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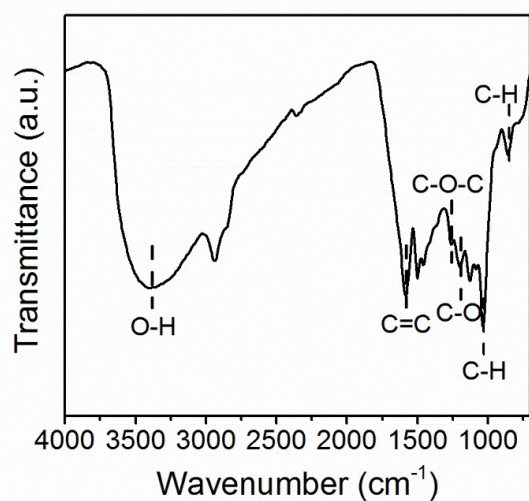
**Fig. S1—S5**

**Table 1 and 2**

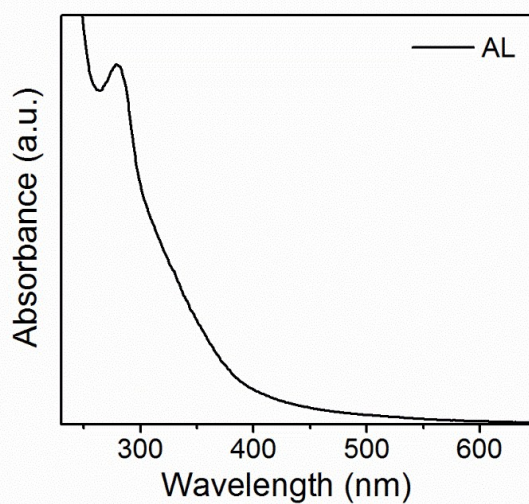
**Supplementary References S1—S21**



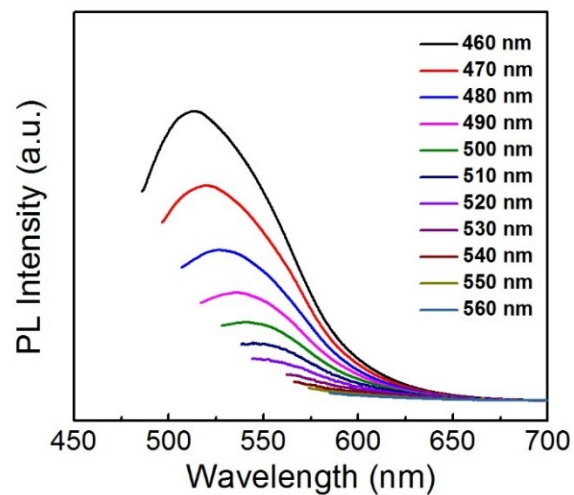
**Figure S1.** XRD pattern of AL.



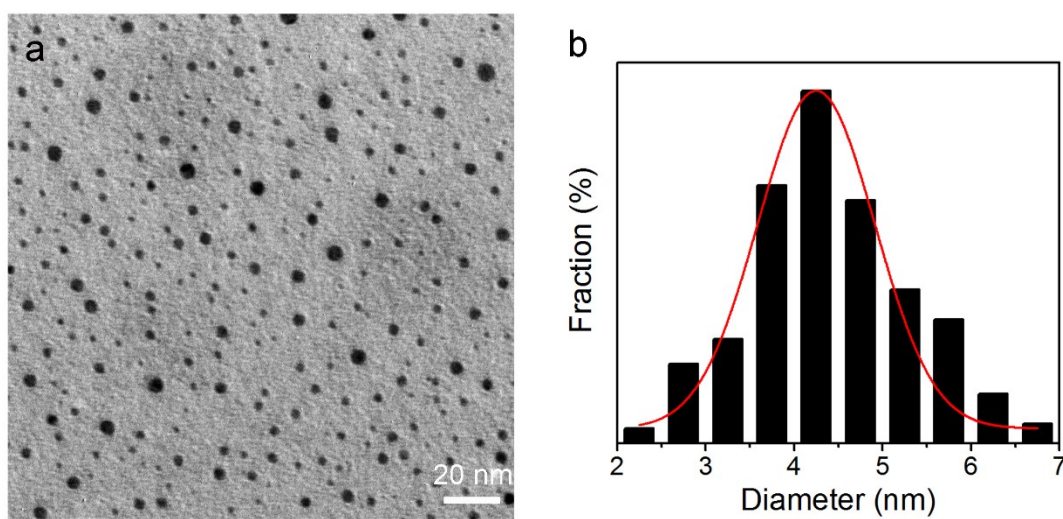
**Figure S2.** FT-IR analysis of AL.



**Figure S3.** UV-vis spectrum of AL.



**Figure S4.** PL emission spectra with excitation wavelength from 460 to 560 nm.



**Figure S5.** (a) TEM image and (b) size distribution of GQDs with average lateral size of 4.4 nm.

**Table 1.** Summary of biomass-based carbon quantum dots

Precursor	Synthetic Method	Quantum Yield (%)	Reference
Urine	Carbonization at 200 °C	5.3	1
Orange juice	Hydrothermal treatment at 120 °C	26	S1
Food waste	Ultrasound irradiation (40 kHz)	2.85	37
Garlic	Hydrothermal treatment at 200 °C	17.5	2
Orange peel	Hydrothermal treatment at 180 °C	36	36
Glucose	Hydrothermal treatment at 200 °C	1.1-2.4	S2
Grass	Hydrothermal treatment at 150–200 °C	2.5-6.2	S3
Hemicellulose	Hydrothermal treatment at 200 °C	16.18	S4
Soybean	Carbonization at 200 °C	3.17	S5
Carbon soot	HNO <sub>3</sub> oxidation at 100 °C	3	S6
Rice husk	Ultrasonic treatment	8.1	33
Watermelon peel	Carbonization at 220 °C	7.1	S7
Peanut shell	Carbonization at 250 °C	9.91	S8
Waste paper	Hydrothermal treatment at 150–200 °C	7.4-11.7	S9
Soy milk	Hydrothermal treatment at 180 °C	2.6	S10
dead neem leaves	H <sub>2</sub> SO <sub>4</sub> , HNO <sub>3</sub> treatment	2	34
Sago waste	Carbonization at 250–450 °C	—	35
Coconut water	Microwave irradiation (800 W)	2.8	S11
Coffee	Stirring at 90 °C	5.5	S12
Calcium alginate	Hydrothermal treatment	—	S13
Lignin	Hydrothermal treatment at 180 °C	—	S14
Cellulolytic enzyme lignin	Stirring at room temperature	2.47	S15
Alkali lignin	Two-step method	22	Our work

**Table 2.** Summary of graphene quantum dots

Precursor	Synthetic Method	Quantum Yield (%)	Reference
Carbon fiber	H <sub>2</sub> SO <sub>4</sub> , HNO <sub>3</sub> treatment, stirring	—	18
Graphene	Electro-beam lithography	—	S16
Fullerene (C <sub>60</sub> )	Ruthenium-catalysed at 1200 K	—	20
Carbon nanotubes	Two-step electrochemical treatment	5.1-6.3	19
Graphite rod	Electrochemical exfoliation	1.8	16
Graphite	Microwave irradiation, sonicating, H <sub>2</sub> SO <sub>4</sub> , HNO <sub>3</sub>	20	17
Graphene	Electrochemical oxidation,	—	S17
GO	Microwave irradiation, reflux	8.5	S18
Carbon black	HNO <sub>3</sub> reflux	1.7-3.5	S19
GO	Hydrothermal treatment at 70–150 °C	19-29	22
GO	Hydrothermal treatment at 120 °C, H <sub>2</sub> O <sub>2</sub>	1-10	S20
Graphite	Hydrothermal treatment at 200 °C	7-12.7	S21
Alginate	Pyrolysis and laser irradiation	—	24
Pyrene	Hydrothermal treatment at 200 °C	23	6
Methane	Chemical Vapor Deposition	—	25
Benzene derivatives	Oxidative condensation reactions	—	29
Citric acid	Hydrothermal treatment at 200 °C	2.2	30
Glucose	Microwave-assisted hydrothermal	7-11	26
Polythiophene derivatives	Hydrothermal treatment at 170 °C	—	28
Hexabenzocoronene	Pyrolysis at 1200 °C and exfoliation	3.8	27
Benzoquinone	Chemical synthesis with TETA	17.5-35.3	5
Dead neem leaves	H <sub>2</sub> SO <sub>4</sub> , HNO <sub>3</sub> treatment,	2	34
Rice husk	Ultrasonic treatment	8.1	33
Alkali lignin	Two-step method	22	Our work

## Supplementary References

- S1 S. Sahu, B. Behera, T. K. Maiti, S. Mohapatra, *Chem. Commun.*, 2012, **48**, 8835–8837.
- S2 Z.-C. Yang, M. Wang, A. M. Yong, S. Y. Wong, X. H. Zhang, H. Tan, A. Y. Chang, X. Li, J. Wang, *Chem. Commun.*, 2011, **47**, 11615–11617.
- S3 S. Liu, J. Tian, L. Wang, Y. Zhang, X. Qin, Y. Luo, A. M. Asiri, A. O. Al-Youbi, X. Sun, *Adv. Mater.*, 2012, **24**, 2037–2041.
- S4 Z. Liang, L. Zeng, X. Cao, Q. Wang, X. Wang, R. Sun, *J Mater. Chem. C*, 2014, **2**, 9760–9766.
- S5 M. Xu, Q. Huang, R. Sun, X. Wang, *RSC Adv.* 2016, **6**, 88674–88682.
- S6 S. Ray, A. Saha, N. R. Jana, R. Sarkar, *J. Phys. Chem. C*, 2009, **113**, 18546–18551.
- S7 J. Zhou, Z. Sheng, H. Han, M. Zou, C. Li, *Mater. Lett.*, 2012, **66**, 222–224.
- S8 M. Xue, Z. Zhan, M. Zou, L. Zhang, S. Zhao, *New J. Chem.*, 2016, **40**, 1698–1703.
- S9 J. Wei, X. Zhang, Y. Sheng, J. Shen, P. Huang, S. Guo, J. Pan, B. Liu, B. Feng, *New J. Chem.*, 2014, **38**, 906–909.
- S10 C. Zhu, J. Zhai, S. Dong, *Chem. Commun.*, 2012, **48**, 9367–9369.
- S11 R. Purbia, S. Paria, *Biosens. Bioelectron.*, 2016, **79**, 467–475.
- S12 C. Jiang, H. Wu, X. Song, X. Ma, J. Wang, M. Tan, *Talanta*, 2014, **127**, 68–74.
- S13 X. Sun, J. Tuo, W. Yang, D. Yang. *Nano Report*, 2015, **2**, 51–54
- S14 W. Chen, C. Hu, Y. Yang, J. Cui, Y. Liu, *Materials* 2016, **9**, 184.
- S15 L. Ponomarenko, F. Schedin, M. Katsnelson, R. Yang, E. Hill, K. Novoselov, A. Geim, *Science*, 2008, **320**, 356–358..
- S16 Y. Li, Y. Zhao, H. Cheng, Y. Hu, G. Shi, L. Dai, L. Qu, *J. Am. Chem. Soc.*, 2011, **134**, 15–18.
- S17 H. Sun, L. Wu, N. Gao, J. Ren, X. Qu, *ACS Appl. Mater. Interfaces*, 2013, **5**, 1174–1179.
- S18 S. Yang, C. Zhu, J. Sun, P. He, N. Yuan, J. Ding, G. Ding, X. Xie, *RSC Adv.*, 2015, **5**, 33347–33350.
- S19 T. Gao, X. Wang, L.-Y. Yang, H. He, X.-X. Ba, J. Zhao, F.-L. Jiang, Y. Liu, *ACS Appl. Mater. Interfaces*, 2017, **9**, 24846–24856.
- S20 C. C. Ke, Y. C. Yang, W. L. Tseng, *Part. Part. Syst. Char.*, 2016, **33**, 132–139.
- S21 W. Kwon, Y. H. Kim, C. L. Lee, M. Lee, H. C. Choi, T. W. Lee, S. W. Rhee, *Nano Lett.*, 2014, **14**, 1306–1311.