Electronic Supporting Information for: Temperature-dependent phase behaviour of tetrahydrofuran-water alters solubilization of xylan to improve co-production of furfurals from lignocellulosic biomass

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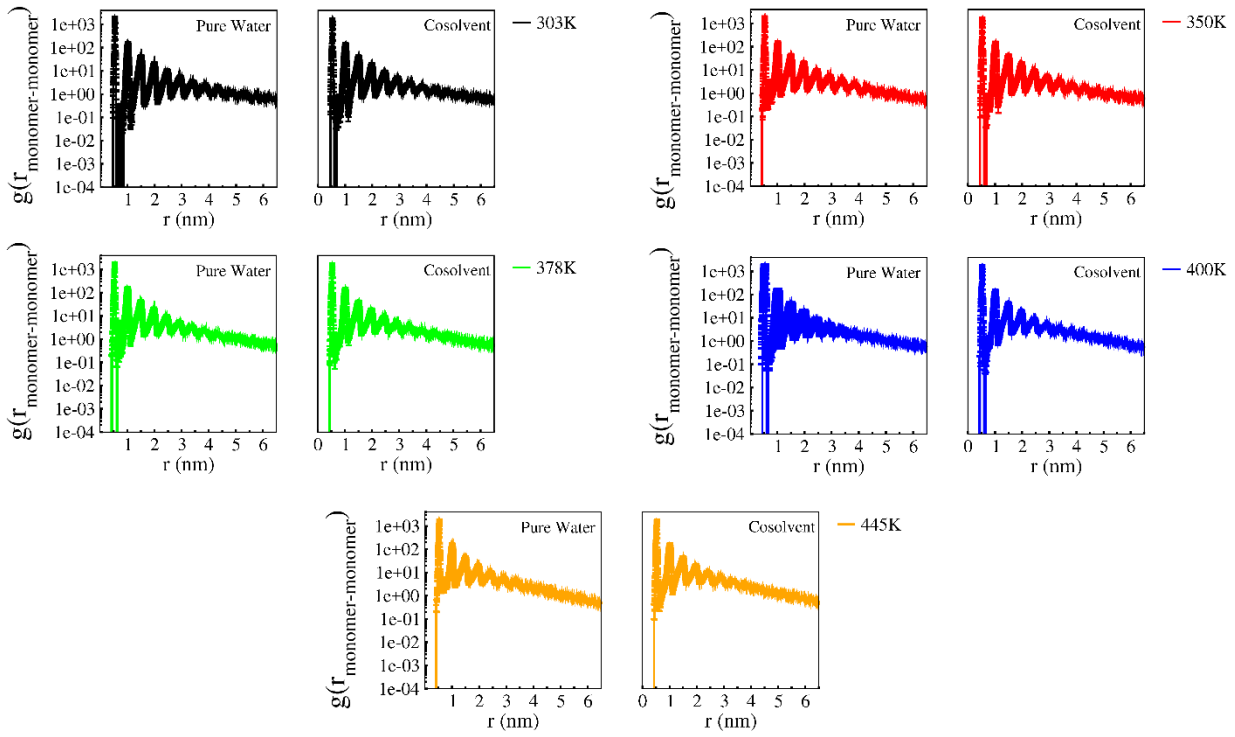
Organization and layout of this section

This section contains separated figures (Figures SI-V) of each plot from Figure 2 in the main text as well as the kinetics table (Table SI) referenced in the main text. The figure and table are provided here with no additional discussion.



Figures SI-V. Individual Plots of Xylan monomer-monomer radial distribution functions

Table SI. First-order rate constants for the disappearance of xylose in water and THF co-solvent environments

Rate Constant	1:1 THF:water (A/min)	Water (A/min)	Literature (A/min)
k[b]	0.109 ± 0.015	0.0557 ± 0.0018	0.0527[c]

[a] Reaction conditions: 10 g L^{-1} D-xylose and 1 wt% H_2SO_4 in batch reactions at 170 °C. THF co-solvent solution contained 1:1 THF and DI water. [b] Rate constant includes a 0.204 M hydrogen ion (A) term. [c] Calculated from equations described by K.J. Zeitsch (2000).