

**Supporting information**

**Epimerization of glucose over ionic liquid/phosphomolybdate hybrids:  
structure-activity relationship**

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**Figure S2.** <sup>13</sup>C {<sup>1</sup>H} NMR spectra of unlabeled glucose in aqueous solution before and after the reaction at 80 °C for 60 min

**Table S3.** <sup>13</sup>C {<sup>1</sup>H} NMR chemical shifts

**Figure S3.** A) Comparison of the X ray diffractograms of [C<sub>2</sub>mim]PMo fresh and spent, B) Comparison of the X ray diffractograms of [C<sub>2</sub>mim]PMo fresh and spent (after treatment with H<sub>2</sub>O<sub>2</sub> solution) and c) Color differences between partially reduced and oxidized species.

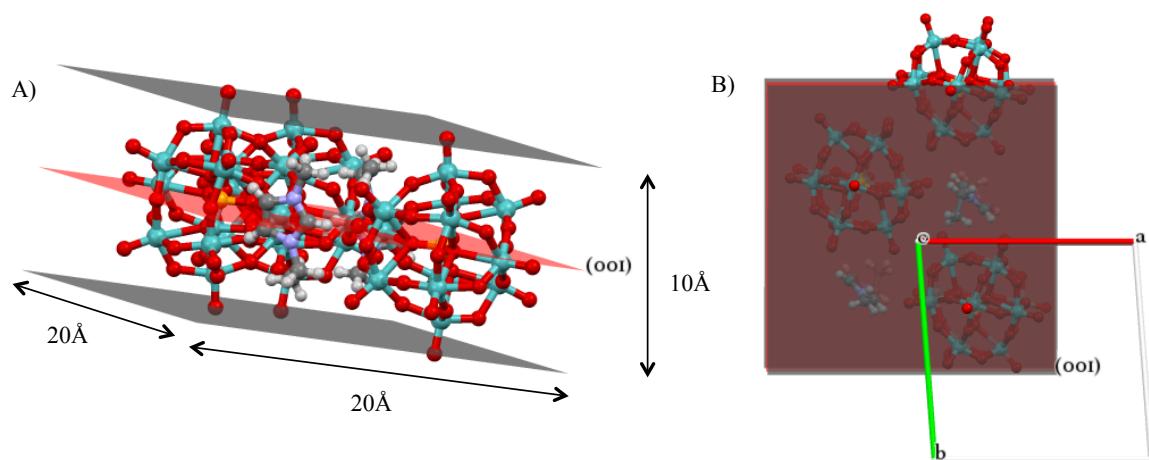
**Table S1.**

	[C2mim]PMo	[C4mim]PMo	[C6mim]PMo
Formula	C <sub>18</sub> H <sub>33</sub> Mo <sub>12</sub> N <sub>6</sub> O <sub>40</sub> P	C <sub>24</sub> H <sub>45</sub> Mo <sub>12</sub> N <sub>6</sub> O <sub>40</sub> P	C <sub>30</sub> H <sub>57</sub> Mo <sub>12</sub> N <sub>6</sub> O <sub>4</sub> P
M	2155.75	2239.91	2324.07
T [K]	193(2)	193(2)	193(2)
Wavelength [Å]	0.71073	0.71073	0.71073
Crystal system	triclinic	triclinic	orthorhombic
Space group	<i>P</i> $\bar{1}$	<i>P</i> $\bar{1}$	<i>Pbca</i>
a [Å]	14.9687(5)	12.9158(5)	22.3846(3)
b [Å]	15.0798(5)	21.2163(8)	20.9690(4)
c [Å]	21.8498(8)	21.6946(8)	26.0718(5)
$\alpha$ [°]	86.944(2)	95.876(2)	90.0
$\beta$ [°]	88.681(2)	105.285(2)	90.0
$\gamma$ [°]	85.622(2)	101.080(2)	90.0
V	4909.8(3)	5553.7(4)	12237.7(4)
Z	4	4	8
D <sub>calc</sub> , Mg m <sup>-3</sup>	2.916	2.679	2.523
$\mu$ , mm <sup>-1</sup>	3.100	2.746	2.498
F(000)	4088	4280	8944
$\theta_{\max}$ , deg	25.242	25.242	25.242
no. refns collected	65355	78335	118207
no. reflns used	17678	20090	11069
no. of param.	1399	1546	808
R <sub>1</sub> (F) [ $F^2 > 2\sigma(F^2)$ ] <sup>[a]</sup>	0.0800	0.0538	0.0352
wR <sub>2</sub> (F <sup>2</sup> ) <sup>[b]</sup> (all data)	0.2023	0.1156	0.0817
S <sup>[c]</sup> (all data)	1080	1.170	1.051

[a] R<sub>1</sub>(F) =  $\sum(|Fo| - |Fc|)/\sum|Fo|$  for the observed reflections [ $F^2 > 2\sigma(F^2)$ ]. [b] wR<sub>2</sub>(F<sup>2</sup>) =  $\{\sum [w(F_o^2 - F_c^2)^2]/\sum$

$w(F_o^2)^2\}^{1/2}$ . [c] S =  $\{\sum [w(F_o^2 - F_c^2)^2]/(n-p)\}^{1/2}$ ; (n = number of reflections, p = number of parameters).

**Figure S1.**

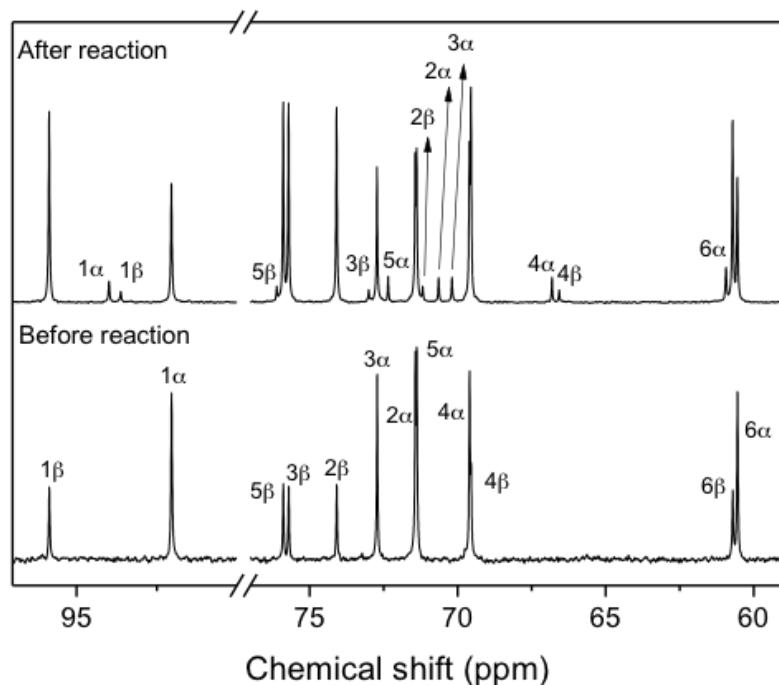


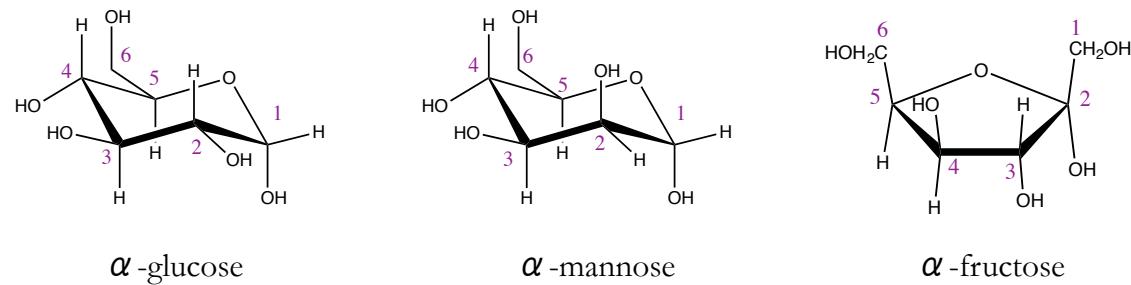
**Table S2**

Entry	Temperature (°C)	Conversion (%) over [C2mim]PMo <sup>a</sup>	Equilibrium conversion (ref.)
1	70	18	-
2	80	21	26 [1]
3	90	27	30 [2]
4	100	34	32 [3]
5	110	43	-
6	120	49	-

<sup>a</sup> Reaction conditions: 5 mL of 0.2 M glucose solution, Glucose:hybrid 256:1 molar ratio (1:0.05 Glucose:Metal ratio), 60 min and 600 rpm.

**Figure S2.**

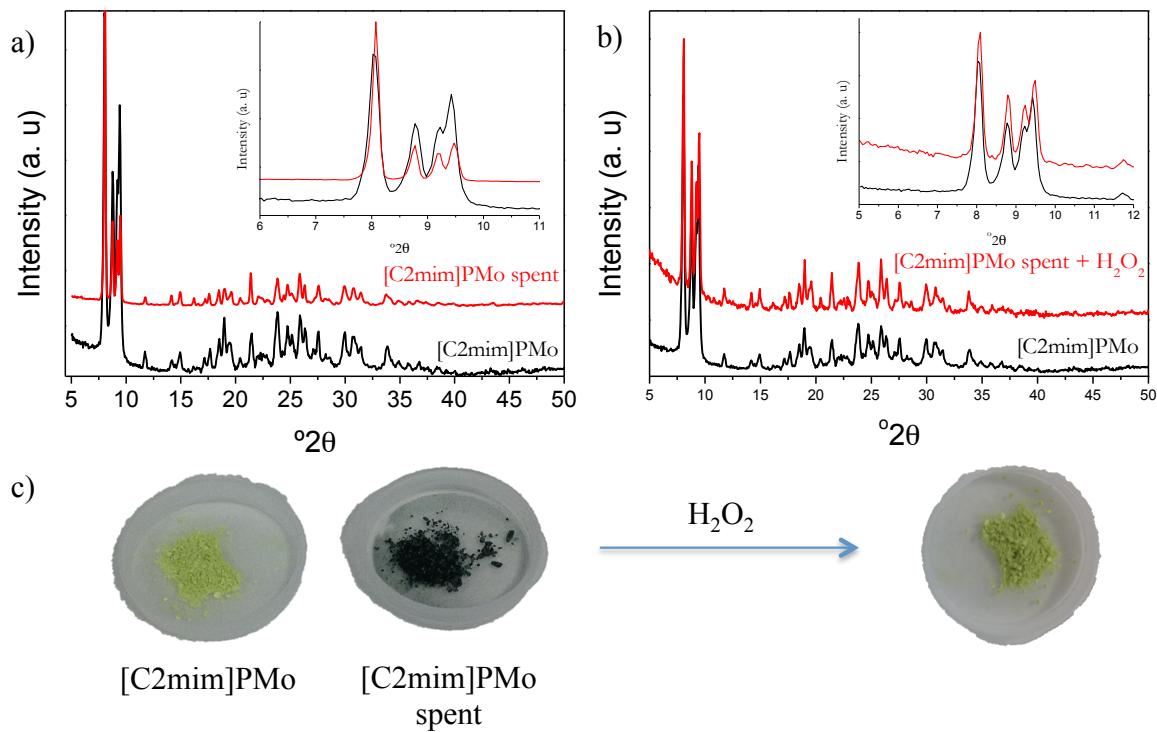


**Table S3**

Entry	Literature [4]	C1	C2	C3	C4	C5	C6
1	$\alpha$ -glucose	92.9	72.5	73.8	70.6	72.3	61.6
2	$\beta$ -glucose	96.7	75.1	76.7	70.6	76.8	61.7
3	$\alpha$ -mannose	95.0	71.7	71.3	68.0	73.4	62.1
4	$\beta$ -mannose	94.6	72.3	74.1	67.8	77.2	62.1
5	$\alpha$ -fructose	63.8	105.5	82.9	77.0	82.2	61.9
6	$\beta$ -fructose	63.6	102.6	76.4	75.4	81.6	63.2
Experimental		C1	C2	C3	C4	C5	C6
7	$\alpha$ -glucose	92.1	71.4	72.7	69.6	71.4	60.55
8	$\beta$ -glucose	95.9	74.1	75.7	69.55	75.9	60.7
9	$\alpha$ -mannose	94.0	70.6	70.2	66.8	72.3	60.9
10	$\beta$ -mannose	93.6	71.2	73.0	66.6	76.1	*

\* Overlaps with C6 of glucose

**Figure S3.**



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