

## **Electronic Supplementary Information**

### **Self-generation of two-dimensional droplet array using oil-water immiscibility and replacement**

Hiroki Yasuga, Koki Kamiya, Shoji Takeuchi and Norihisa Miki

#### **Experiment S1. Preparation of calibration curves for the determination of actual concentration.**

We prepared calibration curves by investigating the relationship between concentration of molecules and fluorescence intensity. It is reported that fluorescence intensities linearly increase with molecular concentration [1]. The linear calibration curves were prepared as follows. First, 1  $\mu\text{l}$  of rhodamine B solution was placed on a glass slide. The concentration was 0, 1, 2, 3, 4, 4.2, 4.3, 4.4, and 4.6  $\mu\text{M}$ . Then, an acrylic square frame of 0.5 mm thickness was attached on another glass slide. Next, the slide with solution was placed with flipping onto the other glass slide. Finally, the container surrounded by the frame was immersed with n-decane as the solution were formed as half-sphere droplets. The droplets were laser-scanned with a confocal microscope. Fluorescence intensities on each concentration were obtained from the scanned images by using ImageJ. The experiments were conducted three times. Same experiment and analysis were conducted on using fluorescein with the concentration of 0, 5, 10, 15, and 20  $\mu\text{M}$ .

ESI Figures S2A and B show the relationship between concentration and fluorescence intensity on rhodamine B and fluorescein, respectively. The calibration curves shown in the graphs were prepared as the linear least-squares fit of mean values. The obtained equation is used for the determination of concentration from fluorescence intensity. For the rhodamine B, because the increase in fluorescence intensity with respect to concentration was drastically changed at approximately 4  $\mu\text{M}$ , two calibration curves were prepared as set 4  $\mu\text{M}$  as the boundary. We consider the change originated from the tendency for rhodamine B to gather at oil-water interface [2]. To calibrate fluorescence intensity into concentration, the boundary of the fluorescence intensity was set as 9.64 on rhodamine B: On a lower value than the boundary, the equation of  $I = 1.57 C + 3.35$  (where I is fluorescence intensity and C is concentration) was used while  $I = 100.12 C - 395.40$  was used otherwise.

[1] B. Vazquez, N. Qureshi, L. Oropeza-Ramos and L. F. Olguin, *Lab Chip*, 2014, **14**, 3550–3555.

[2] J. H. J. Thijssen, A. B. Schofield and P. S. Clegg, *Soft Matter*, 2011, **7**, 7965.

Table S1. Parameters of dimension (width, step height,  $A_g$ , and  $A_p$ ). Components of this table show the step heights. Gray area shows no fabricated parameters.

| Width             | $A_g$ | $A_p$           |                   |                   |                   |                   |
|-------------------|-------|-----------------|-------------------|-------------------|-------------------|-------------------|
|                   |       | 0               | 2                 | 4                 | 6                 | 8                 |
| 100 $\mu\text{m}$ | 5     | 0 $\mu\text{m}$ | 200 $\mu\text{m}$ | 400 $\mu\text{m}$ | 600 $\mu\text{m}$ | 800 $\mu\text{m}$ |
| 50 $\mu\text{m}$  | 10    | 0 $\mu\text{m}$ | 100 $\mu\text{m}$ | 200 $\mu\text{m}$ | 300 $\mu\text{m}$ | 400 $\mu\text{m}$ |
| 33 $\mu\text{m}$  | 15    | 0 $\mu\text{m}$ | 66 $\mu\text{m}$  | 133 $\mu\text{m}$ | 200 $\mu\text{m}$ | 266 $\mu\text{m}$ |
| 25 $\mu\text{m}$  | 20    | 0 $\mu\text{m}$ | 50 $\mu\text{m}$  | 100 $\mu\text{m}$ | 150 $\mu\text{m}$ | 200 $\mu\text{m}$ |
| 20 $\mu\text{m}$  | 25    | 0 $\mu\text{m}$ | 40 $\mu\text{m}$  | 80 $\mu\text{m}$  | 120 $\mu\text{m}$ | 160 $\mu\text{m}$ |
| Step height       |       |                 |                   |                   |                   |                   |

Table S2. Volume of injected solvent with respect to  $A_g$  and  $A_p$ . Gray area shows no fabricated parameters.

| $A_g$ | $A_p$             |                   |                   |                   |                    |
|-------|-------------------|-------------------|-------------------|-------------------|--------------------|
|       | 0                 | 2                 | 4                 | 6                 | 8                  |
| 5     | 4.5 $\mu\text{l}$ | 6.3 $\mu\text{l}$ | 8.1 $\mu\text{l}$ | 9.9 $\mu\text{l}$ | 11.7 $\mu\text{l}$ |
| 10    | 4.5 $\mu\text{l}$ | 5.4 $\mu\text{l}$ | 6.3 $\mu\text{l}$ | 7.2 $\mu\text{l}$ | 8.1 $\mu\text{l}$  |
| 15    | 4.5 $\mu\text{l}$ | 5.1 $\mu\text{l}$ | 5.7 $\mu\text{l}$ | 6.3 $\mu\text{l}$ | 6.9 $\mu\text{l}$  |
| 20    | 4.5 $\mu\text{l}$ | 5.0 $\mu\text{l}$ | 5.4 $\mu\text{l}$ | 5.9 $\mu\text{l}$ | 6.3 $\mu\text{l}$  |
| 25    | 4.5 $\mu\text{l}$ | 4.9 $\mu\text{l}$ | 5.2 $\mu\text{l}$ | 5.6 $\mu\text{l}$ | 5.9 $\mu\text{l}$  |

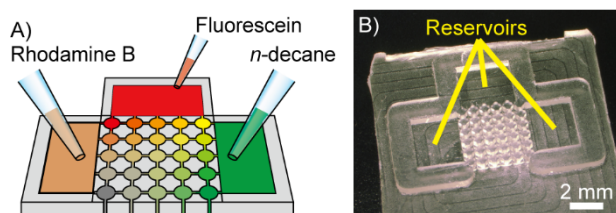


Figure S1. A) Formation of concentration gradient of two fluorescence molecules and B) An image of device.

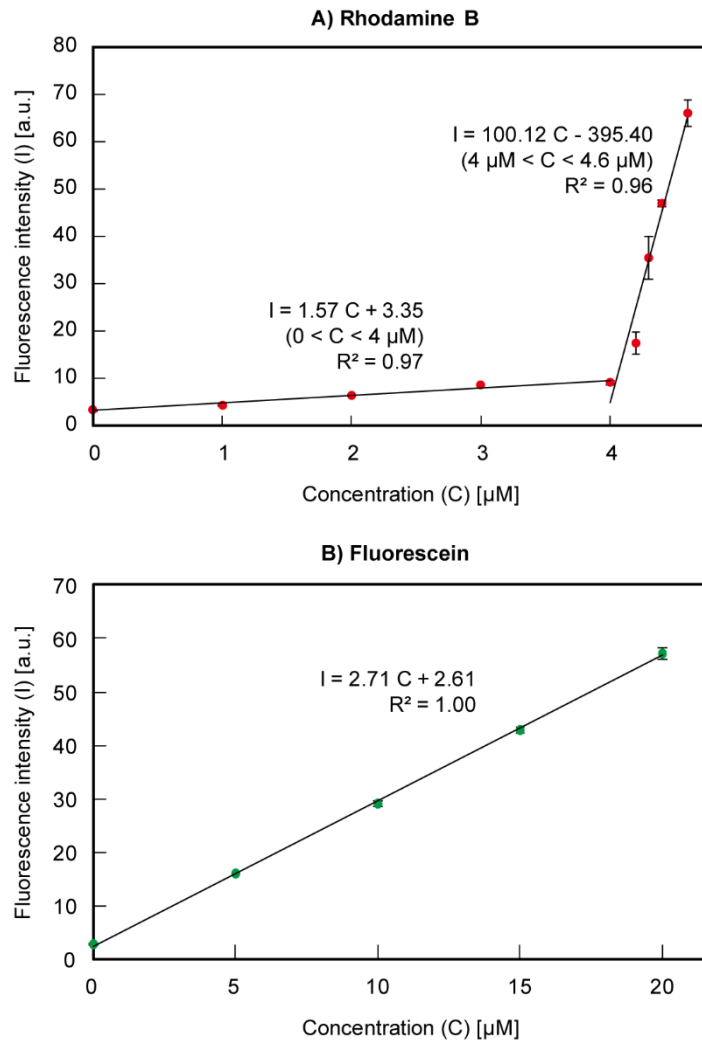


Figure S2. Calibration curves between concentration of fluorescent molecules and fluorescence intensity. A) Rhodamine B and B) fluorescein.

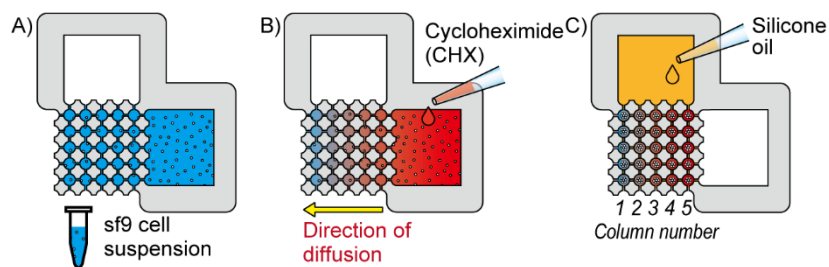


Figure S3. Procedures to prepare sf9 cell and form cycloheximide (CHX) concentration gradient into droplet array. A) Impregnation of sf9 cell suspension. B) Formation of CHX concentration gradient by diffusion. C) Droplet generation by silicone oil injection.

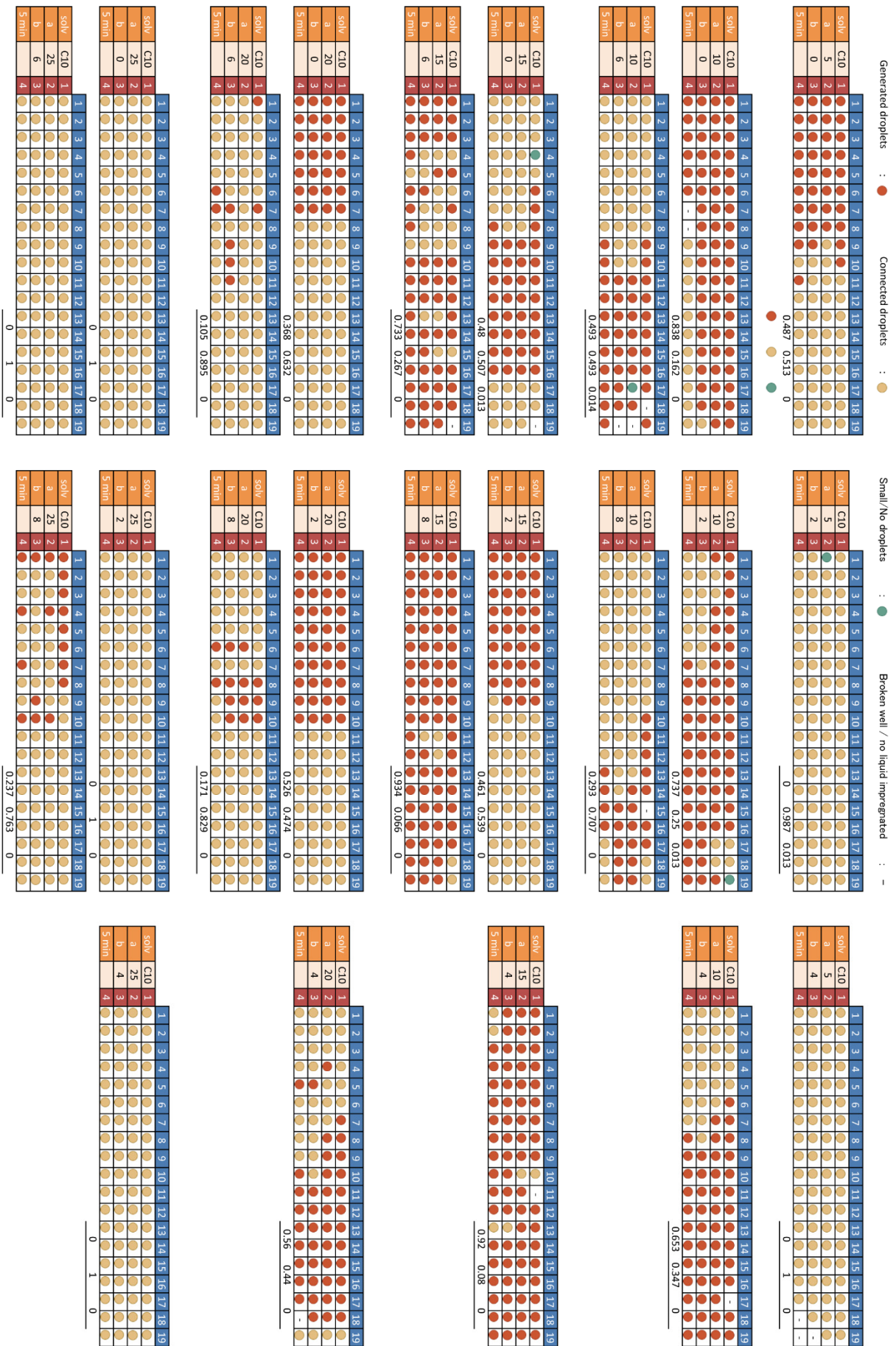


Figure S4 Details of droplet states on using n-decane



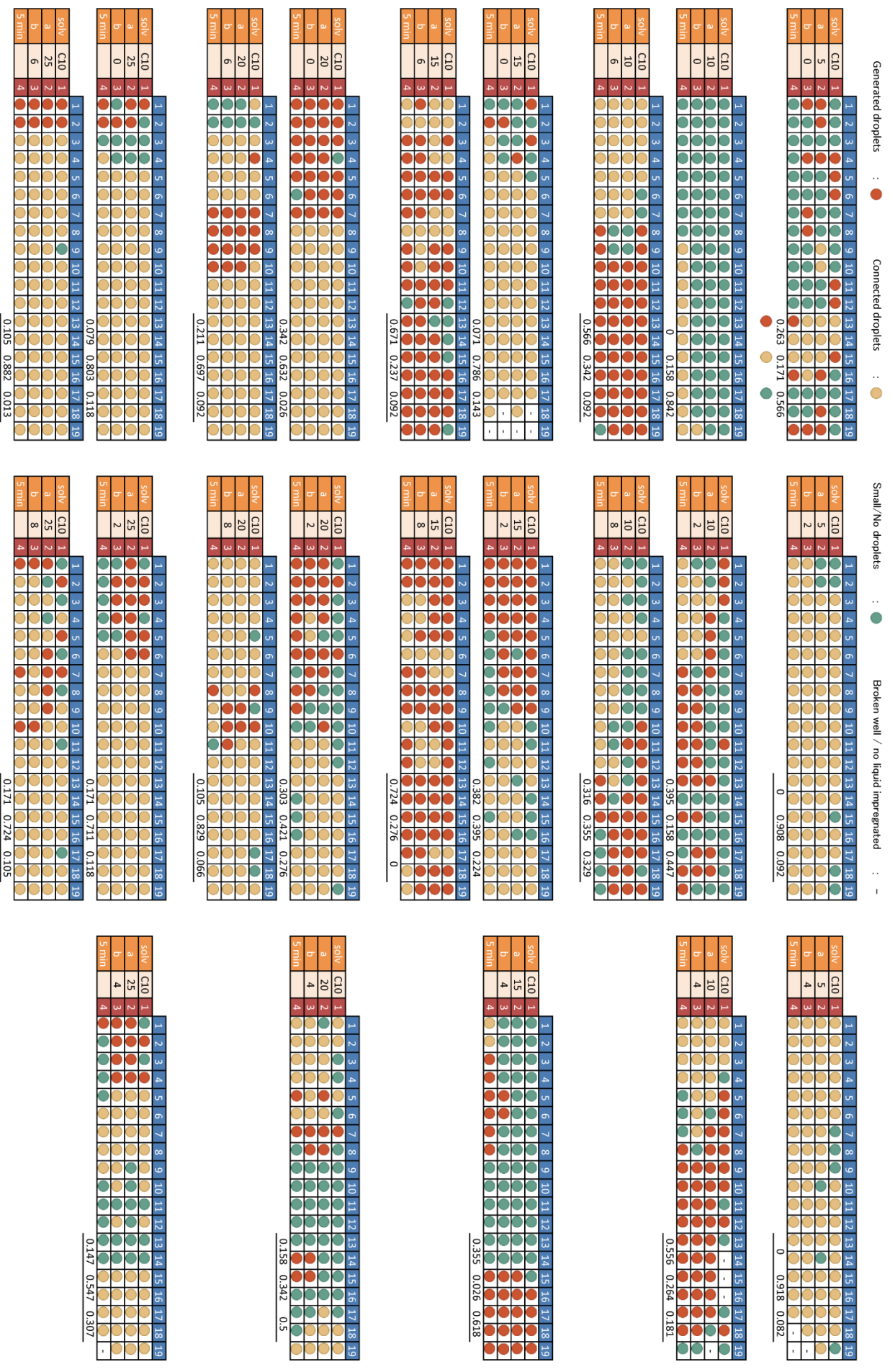


Figure S6 Details of droplet states on using  $n$ -hexadecane

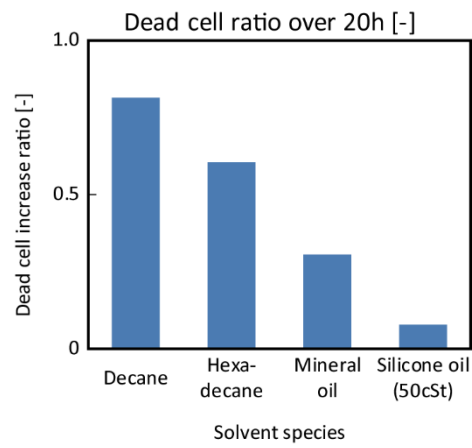


Figure S7. Dead cell increase ratio in a droplet generated in proposed well array. The cells were kept over 20 h in a droplet of a 5 x 5 well array. Here, the cells which had been died during droplet generation process was not counted. (n = 1)

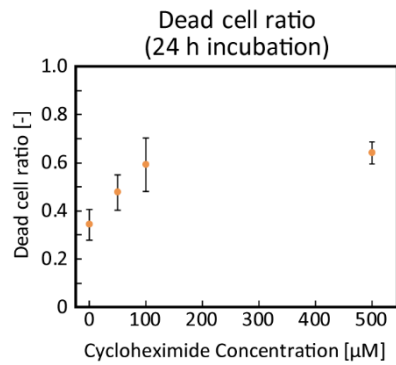


Figure S8. Dead cell ratio in wells of a 96-microtiter plate after 24 h incubation. (n = 3)