

Supporting information

New Journal of Chemistry

Hybrid[4]arenes with anthracene units and tuneable cavities

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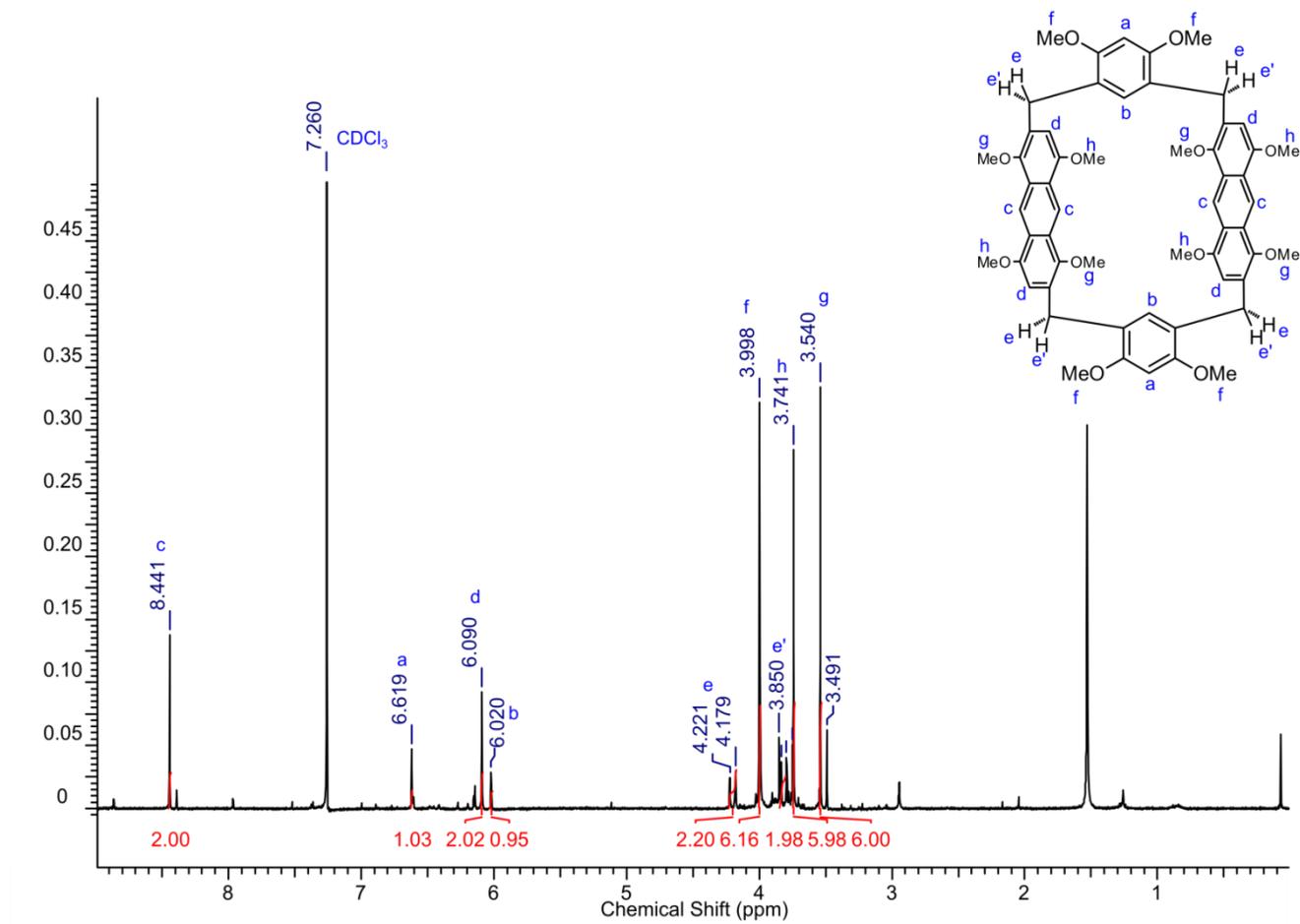


Figure 1: ¹H NMR spectrum of **4** (CDCl₃, 400MHz, 303K).

¹H NMR (CDCl₃, 400MHz, 303K): 8.44 (2H, s), 6.62 (1H, s), 6.09 (2H, s), 6.02 (2H, s), 4.20 (2H, d, $J^2 = 16.1$ Hz), 4.00 (6H, s), 3.83 (2H, d, $J^2 = 16.1$ Hz), 3.74 (6H, s), 3.54 (6H, s)

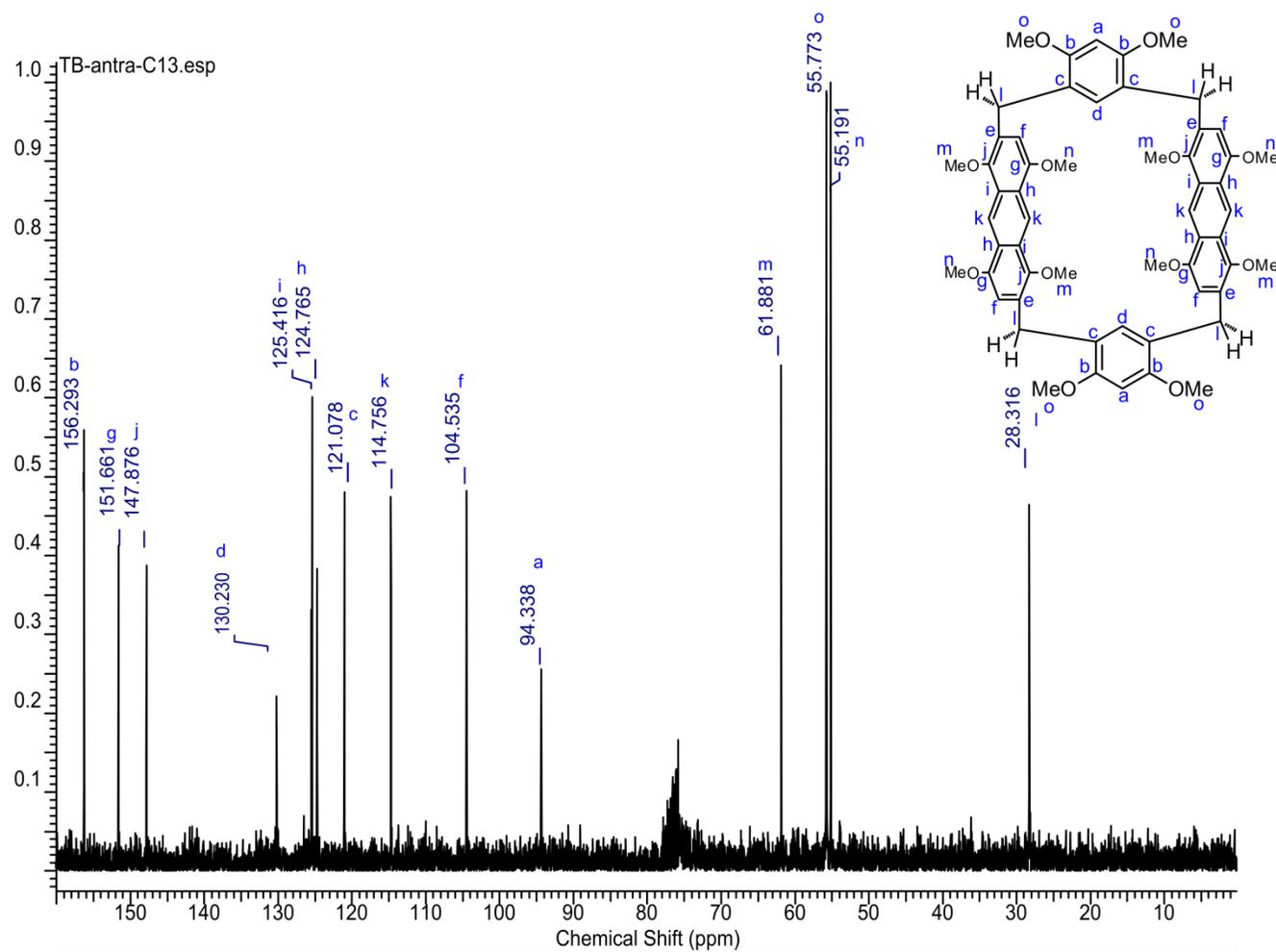


Figure 2: ^{13}C NMR spectrum of **3** (CDCl_3 , 125MHz, 303K).

^{13}C NMR (CDCl_3 , 125MHz, 303K): 156.3, 151.7, 147.9, 130.2, 125.4, 124.8, 121.1, 114.8, 104.5, 61.9, 55.8, 55.2, 28.3

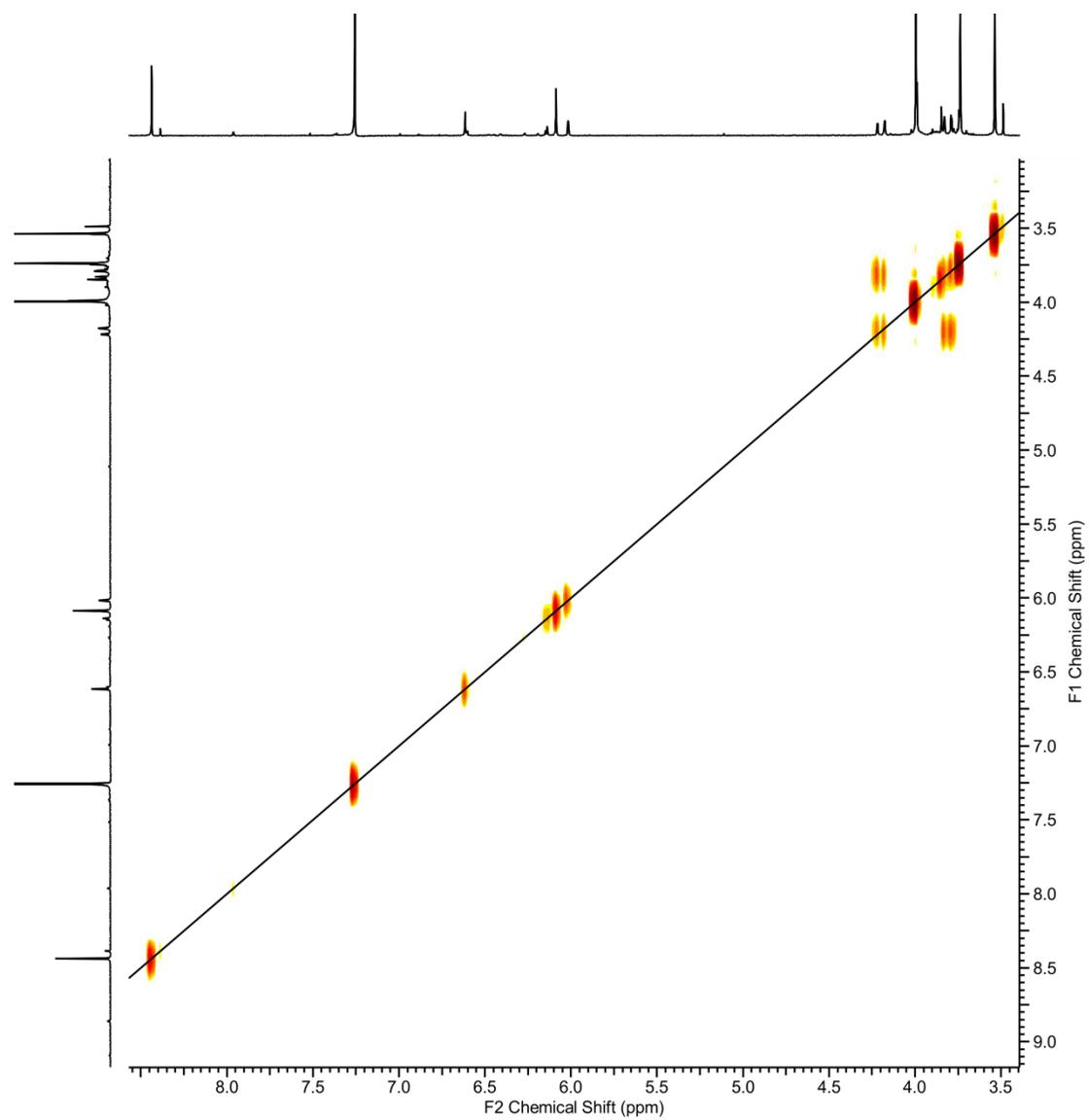


Figure 3: ^1H - ^1H COSY spectrum of **3** (CDCl_3 , 400MHz, 303K).

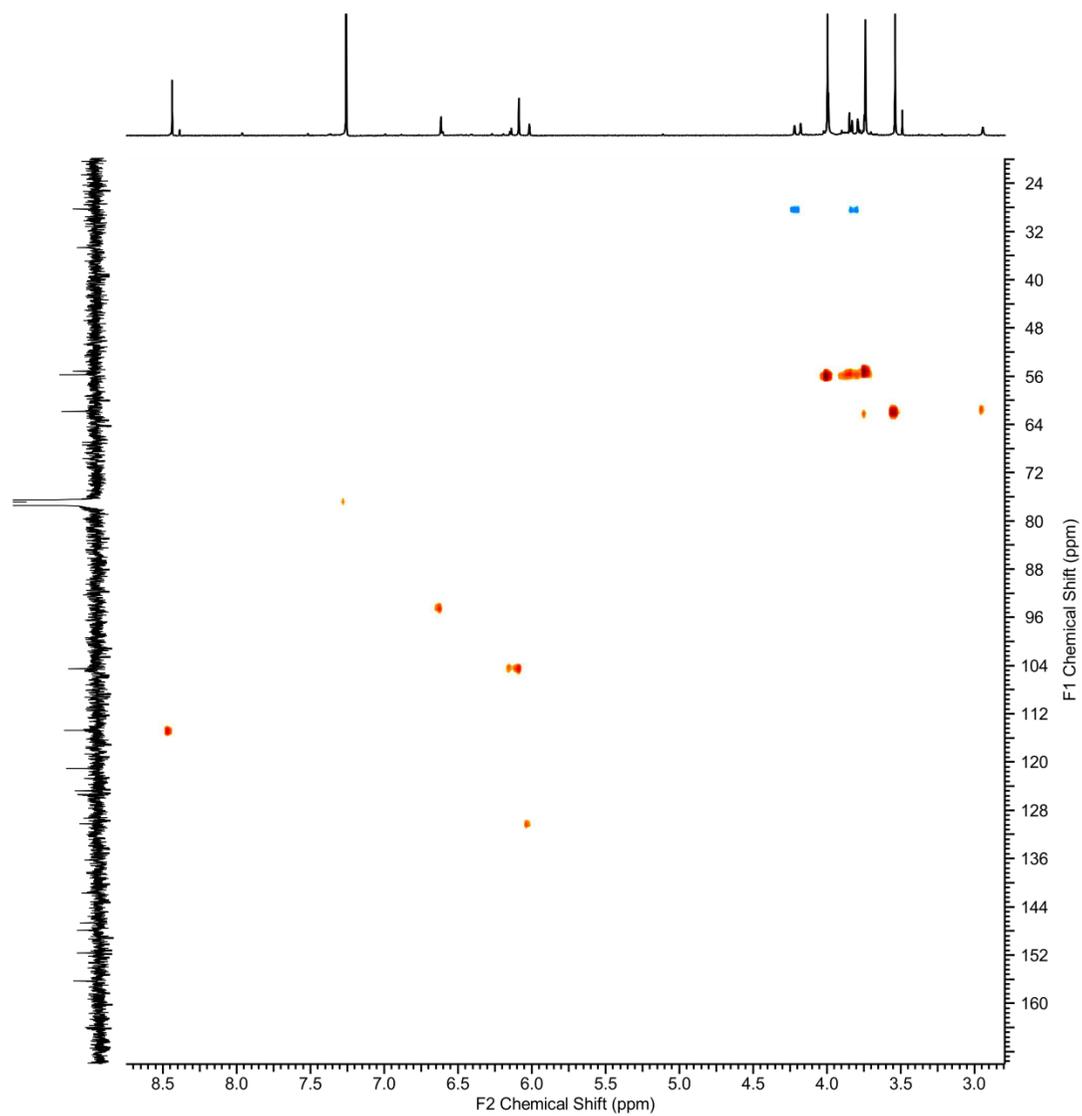


Figure 4: ^1H - ^{13}C HSQC NMR spectrum of **3** (CDCl_3 , 400MHz, 303K).

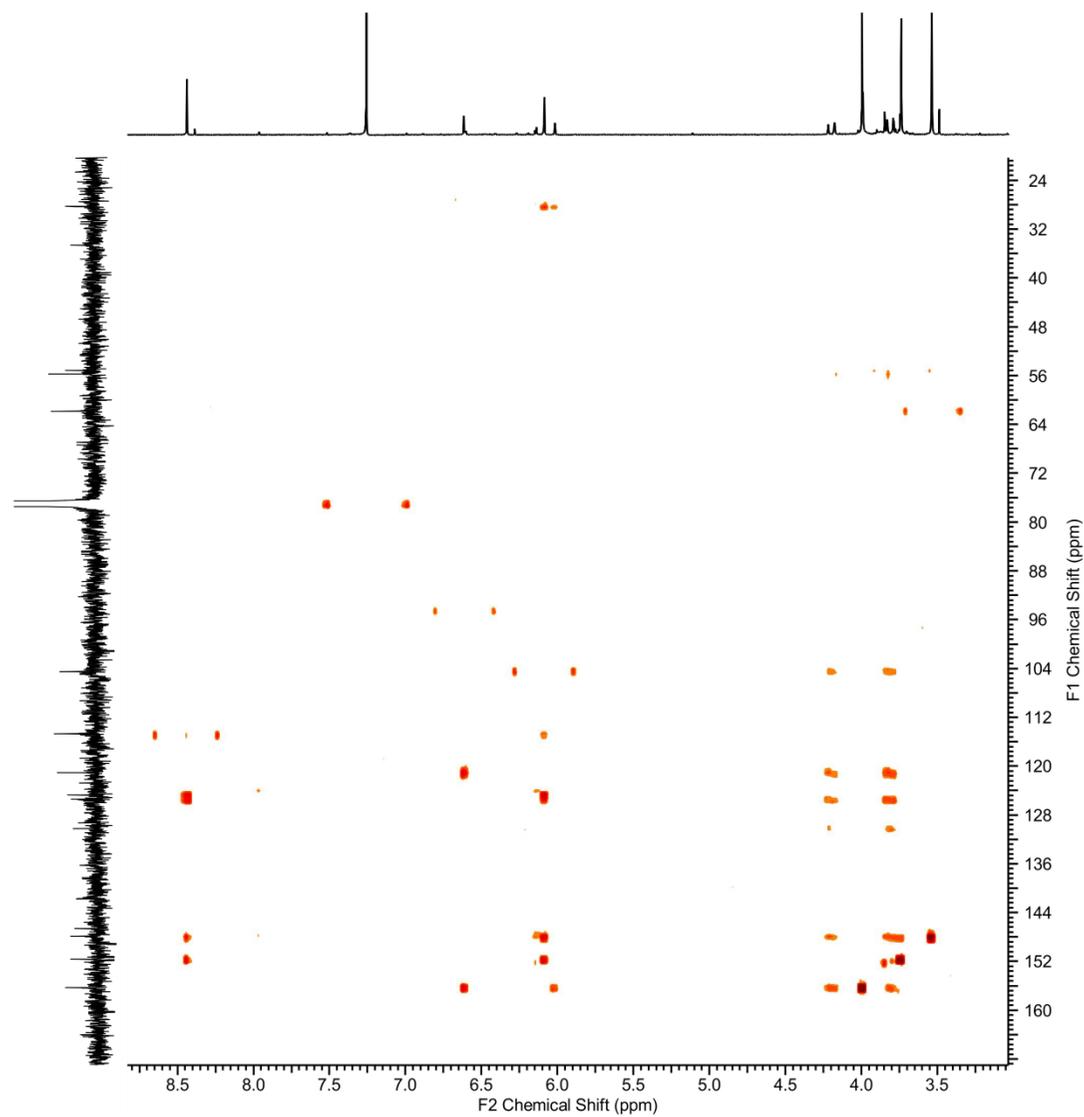


Figure 5: ^1H - ^{13}C HMBC NMR spectrum of **3** (CDCl_3 , 400MHz, 303K).

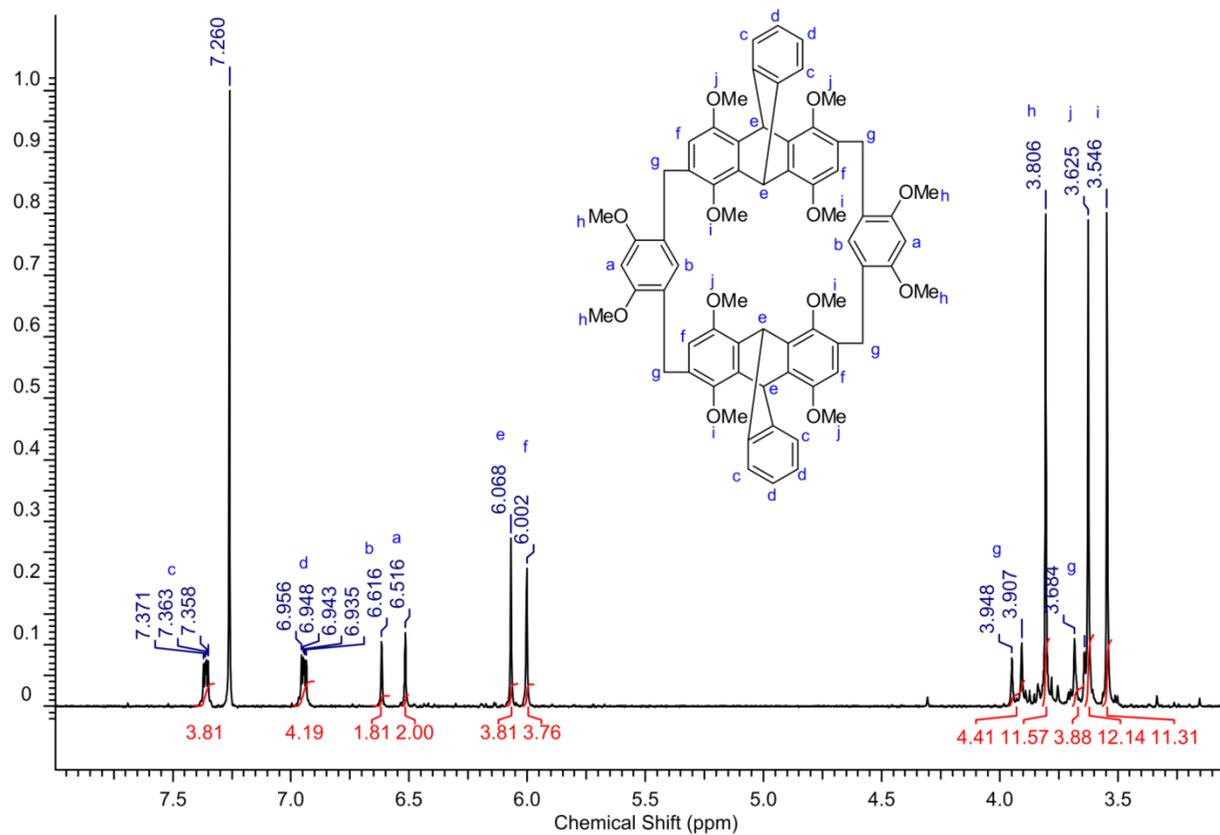


Figure 6: ¹H NMR spectrum of **5** (CDCl₃, 400MHz, 303K).

¹H NMR (CDCl₃, 400MHz, 303K): 7.33 (dd, 4H, $J^3 = 5.4$ Hz, $J^4 = 3.2$ Hz), 6.95 (dd, 4H, $J^3 = 5.4$ Hz, $J^4 = 3.2$ Hz), 6.62 (s, 2H), 6.52 (s, 2H), 6.07 (s, 4H), 6.00 (s, 4H), 3.93 (d, 4H, $J^2 = 16.6$ Hz), 3.81 (s, 12H), 3.66 (d, 4H, $J^2 = 16.6$ Hz), 3.63 (s, 12H), 3.55 (s, 12H)

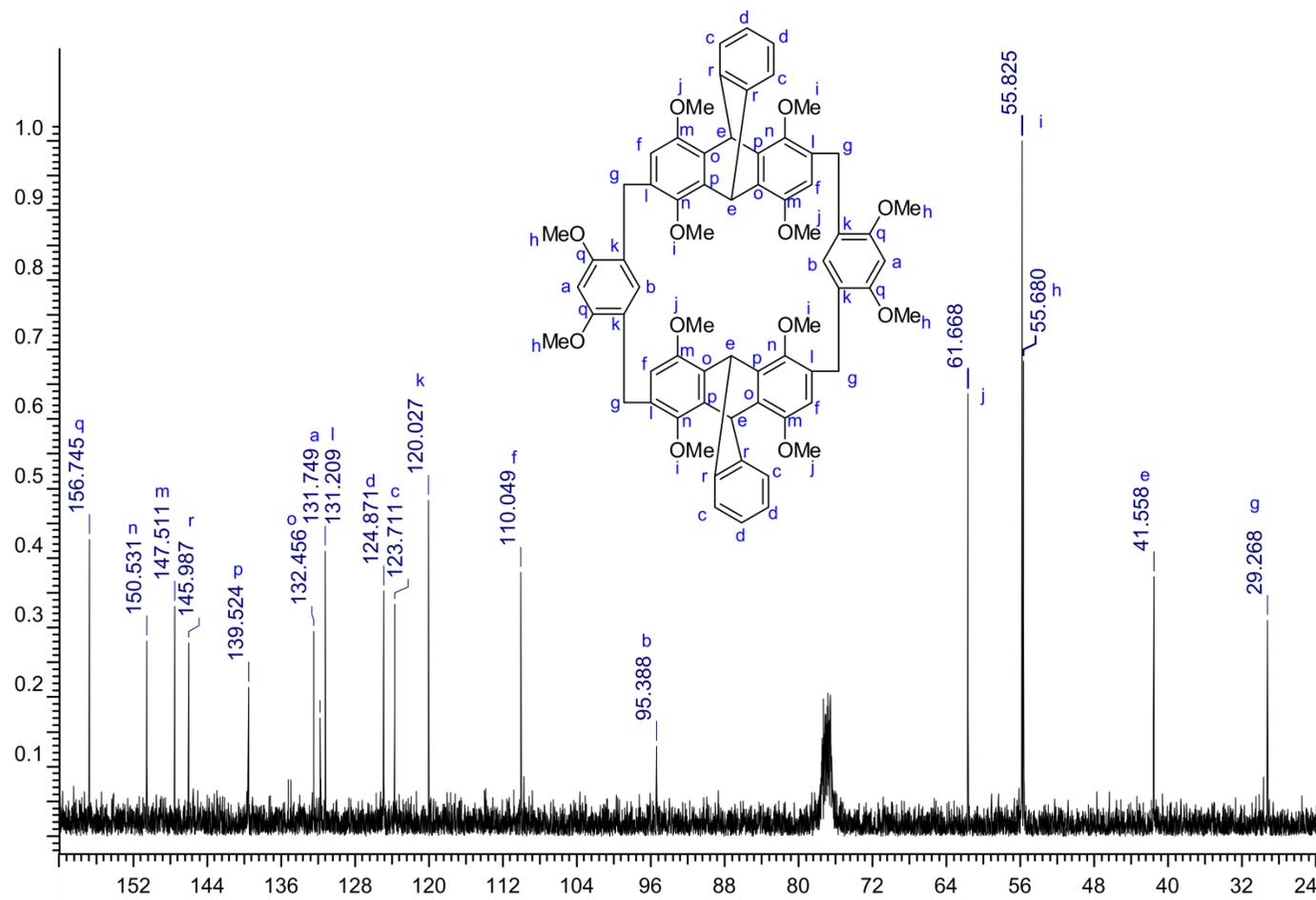


Figure 7: ¹³C NMR spectrum of 5 (CDCl₃, 100MHz, 303K).

¹³C NMR (CDCl₃, 100MHz, 303K): 156.7, 150.5, 147.5, 146.0, 139.5, 132.5, 131.7, 131.2, 124.9, 123.7, 120.0, 110.0, 95.4, 61.7, 55.8, 55.7, 41.6, 29.3

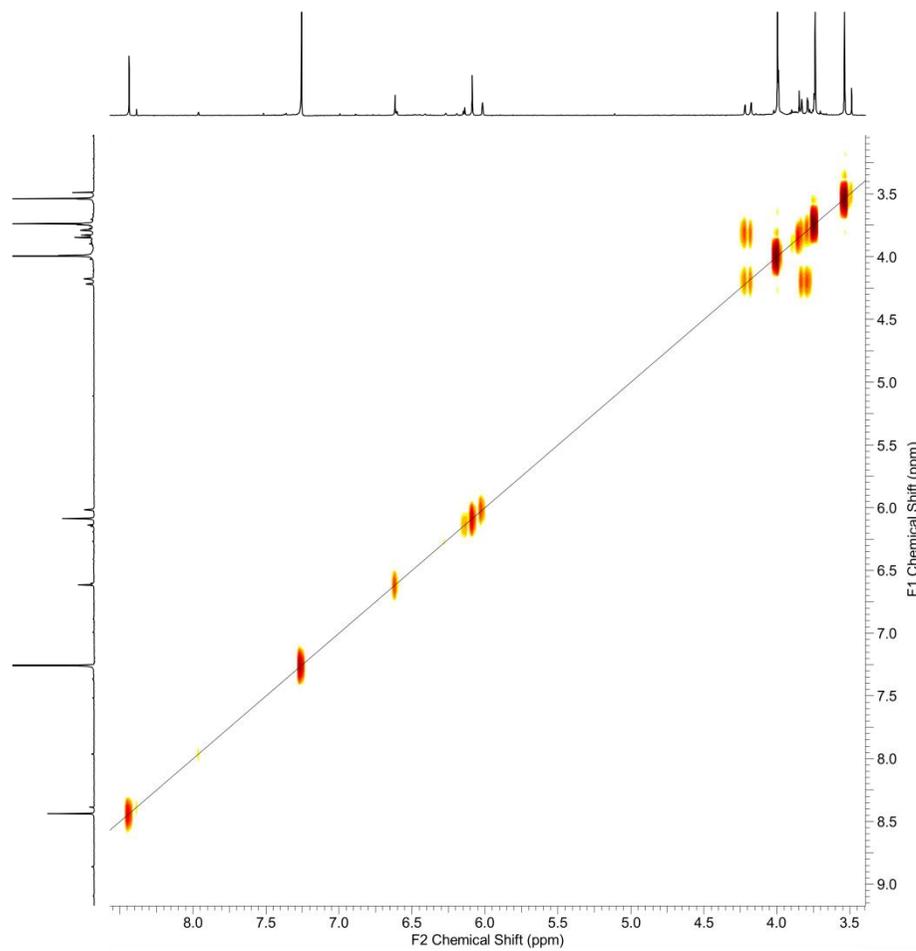


Figure 8: ^1H - ^1H COSY spectrum of **5** (CDCl_3 , 400MHz, 303K).

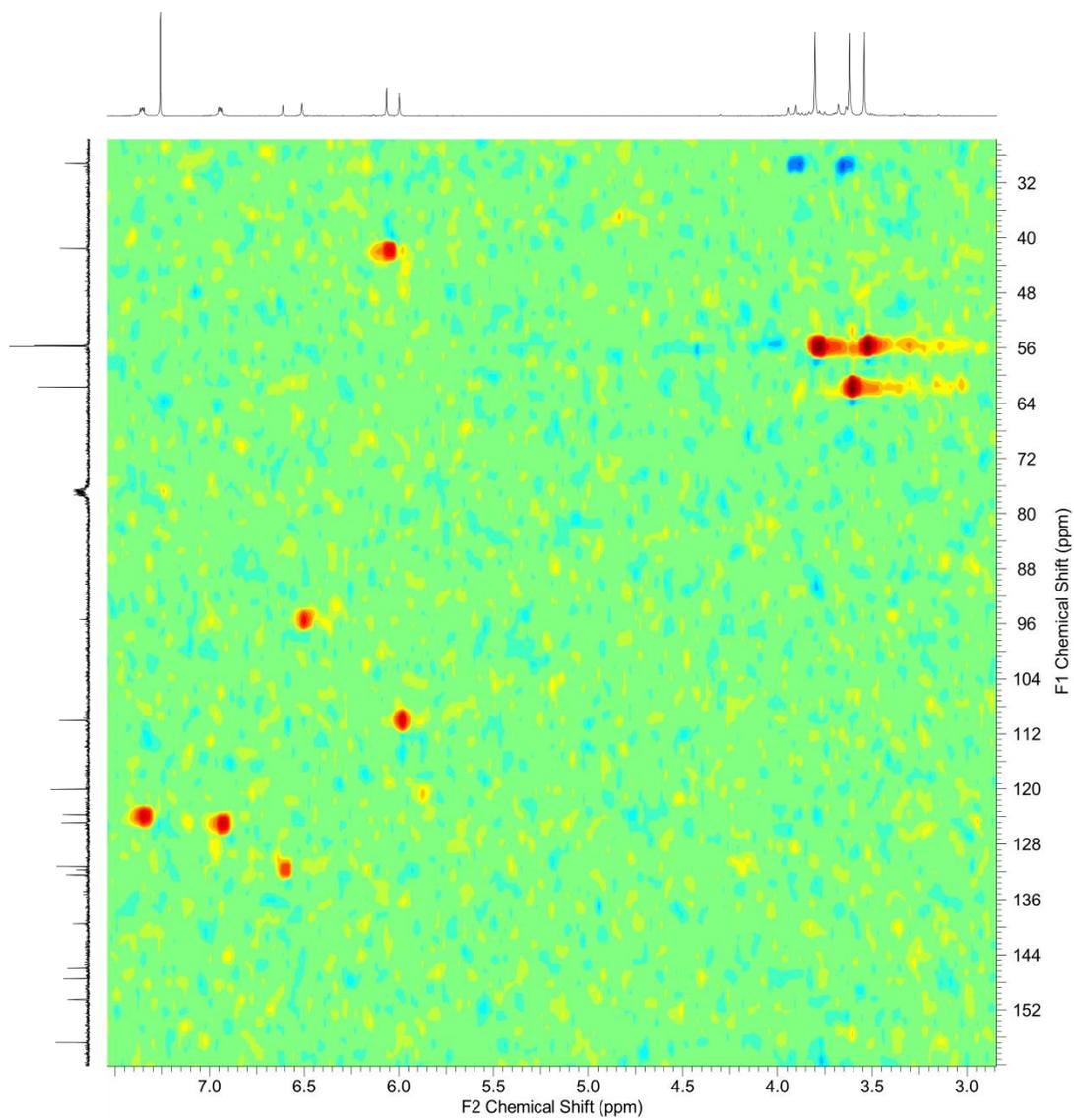


Figure 9: ^1H - ^{13}C HSQC NMR spectrum of **5** (CDCl_3 , 400MHz, 303K).

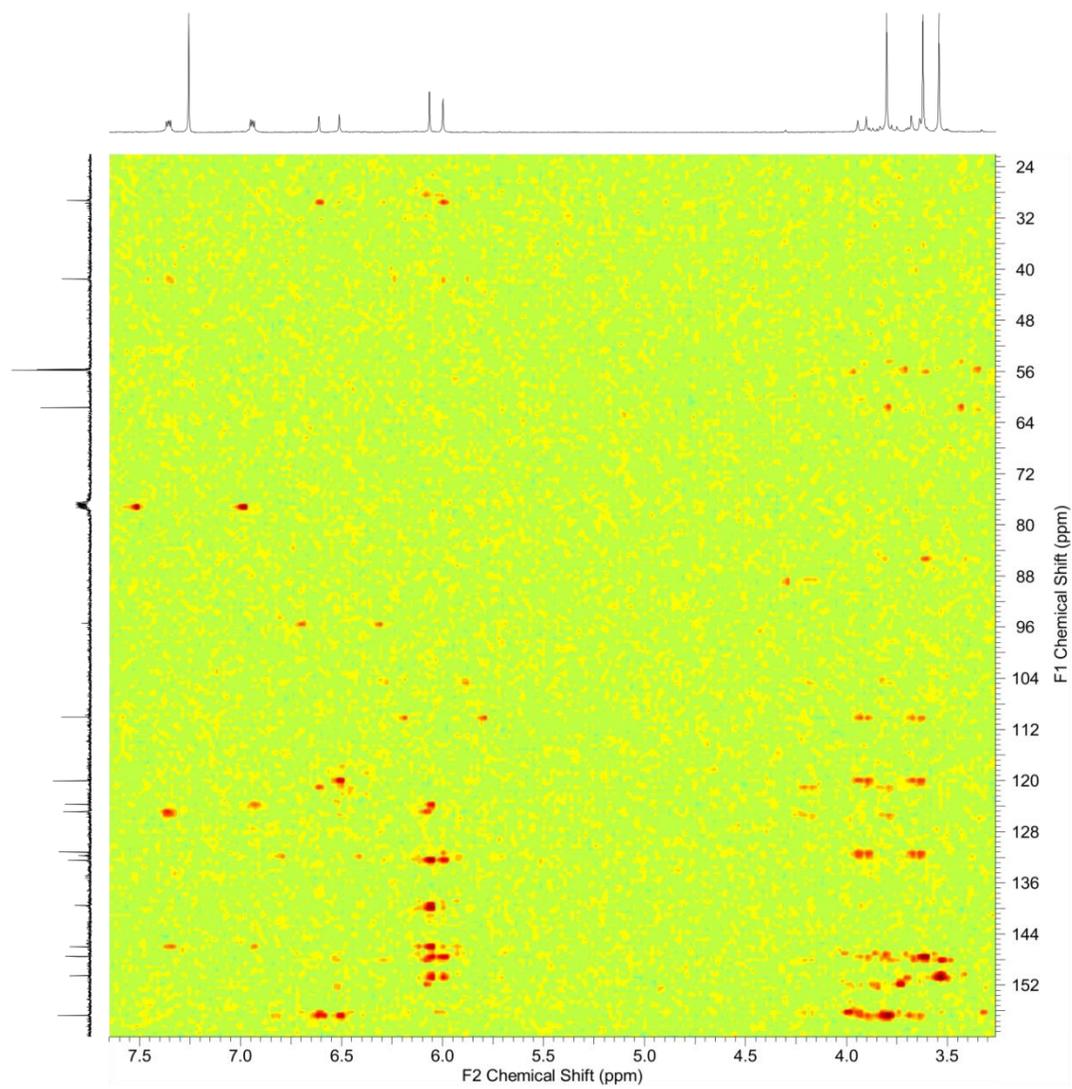


Figure 10: ^1H - ^{13}C HMBC NMR spectrum of **5** (CDCl_3 , 400MHz, 303K).

TB_antra1458_2+2
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07-Jun-2016
 11:43:18
 1: TOF MS ES+
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Elemental Composition Report

Single Mass Analysis

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Element prediction: Off

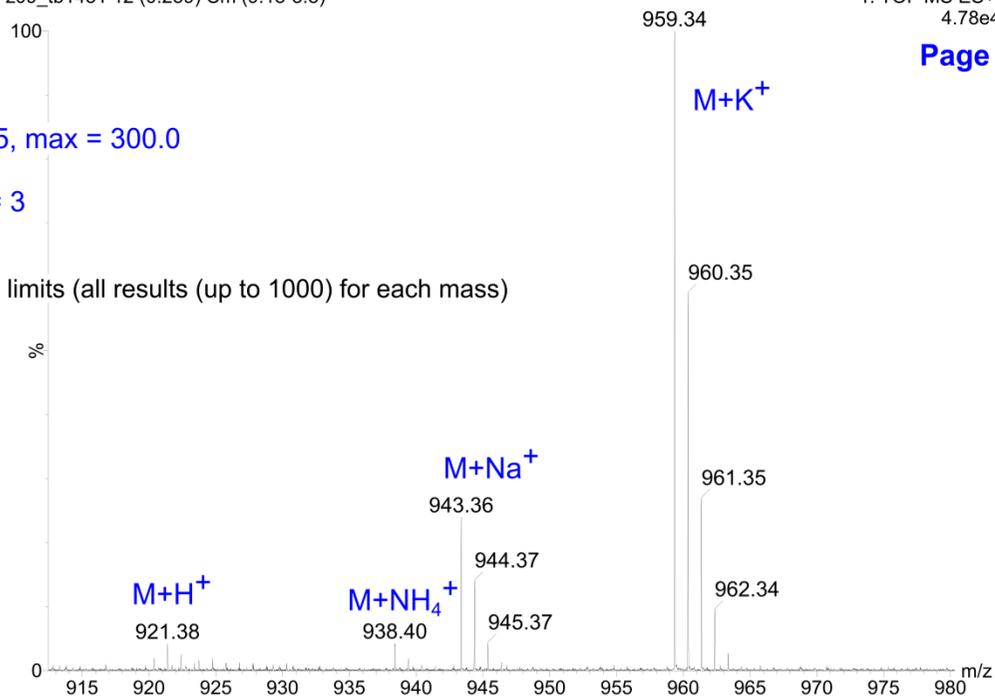
Number of isotope peaks used for i-FIT = 3

Monoisotopic Mass, Even Electron Ions

285 formula(e) evaluated with 3 results within limits (all results (up to 1000) for each mass)

Elements Used:

C: 0-100 H: 0-120 O: 0-13 Na: 0-1



Minimum: -1.5
 Maximum: 20.0 5.0 300.0

Mass	Calc. Mass	mDa	PPM	DBE	i-FIT	Norm	Conf (%)	Formula
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	943.3694	-2.9	-3.1	31.5	443.0	3.940	1.94	C58 H55 O12
	943.3635	3.0	3.2	40.5	446.7	7.674	0.05	C65 H51 O7

Figure 11: HR MS of 3.

Elemental Composition Report

Page 1

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12:05:17

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Element prediction: Off

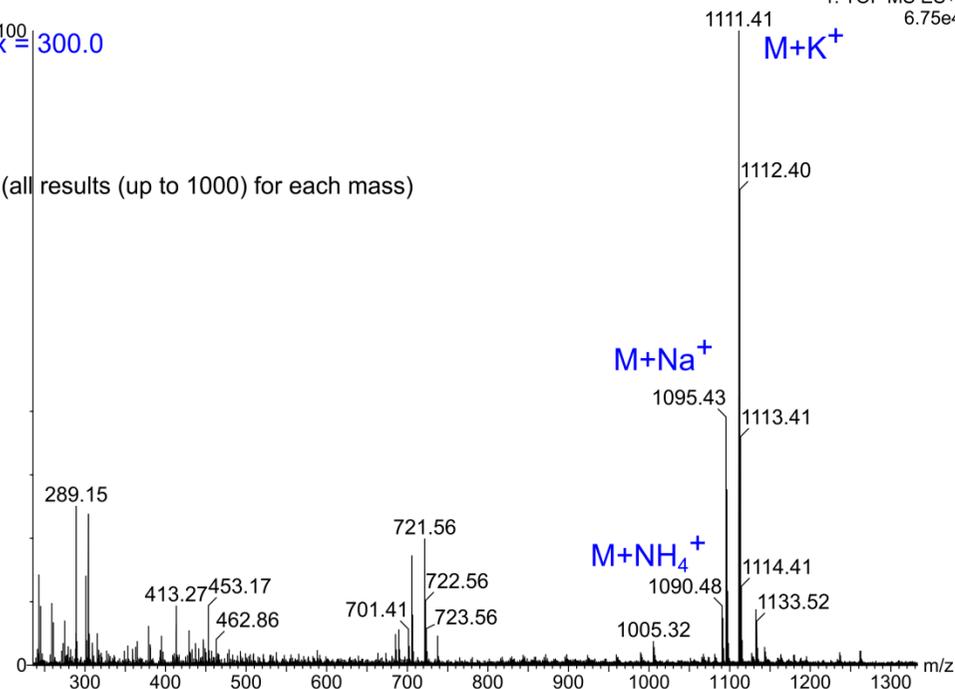
Number of isotope peaks used for i-FIT = 3

Monoisotopic Mass, Even Electron Ions

332 formula(e) evaluated with 3 results within limits (all results (up to 1000) for each mass)

Elements Used:

C: 0-100 H: 0-200 O: 0-12 Na: 0-1



Minimum: -1.5
Maximum: 10.0 3.0 300.0

Mass	Calc. Mass	mDa	PPM	DBE	i-FIT	Norm	Conf (%)	Formula
1095.4291	1095.4295	-0.4	-0.4	36.5	397.9	0.126	88.18	C68 H64 O12 Na
	1095.4320	-2.9	-2.6	39.5	399.9	2.143	11.73	C70 H63 O12
	1095.4261	3.0	2.7	48.5	404.8	7.044	0.09	C77 H59 O7

Figure 12: HR MS of 5.

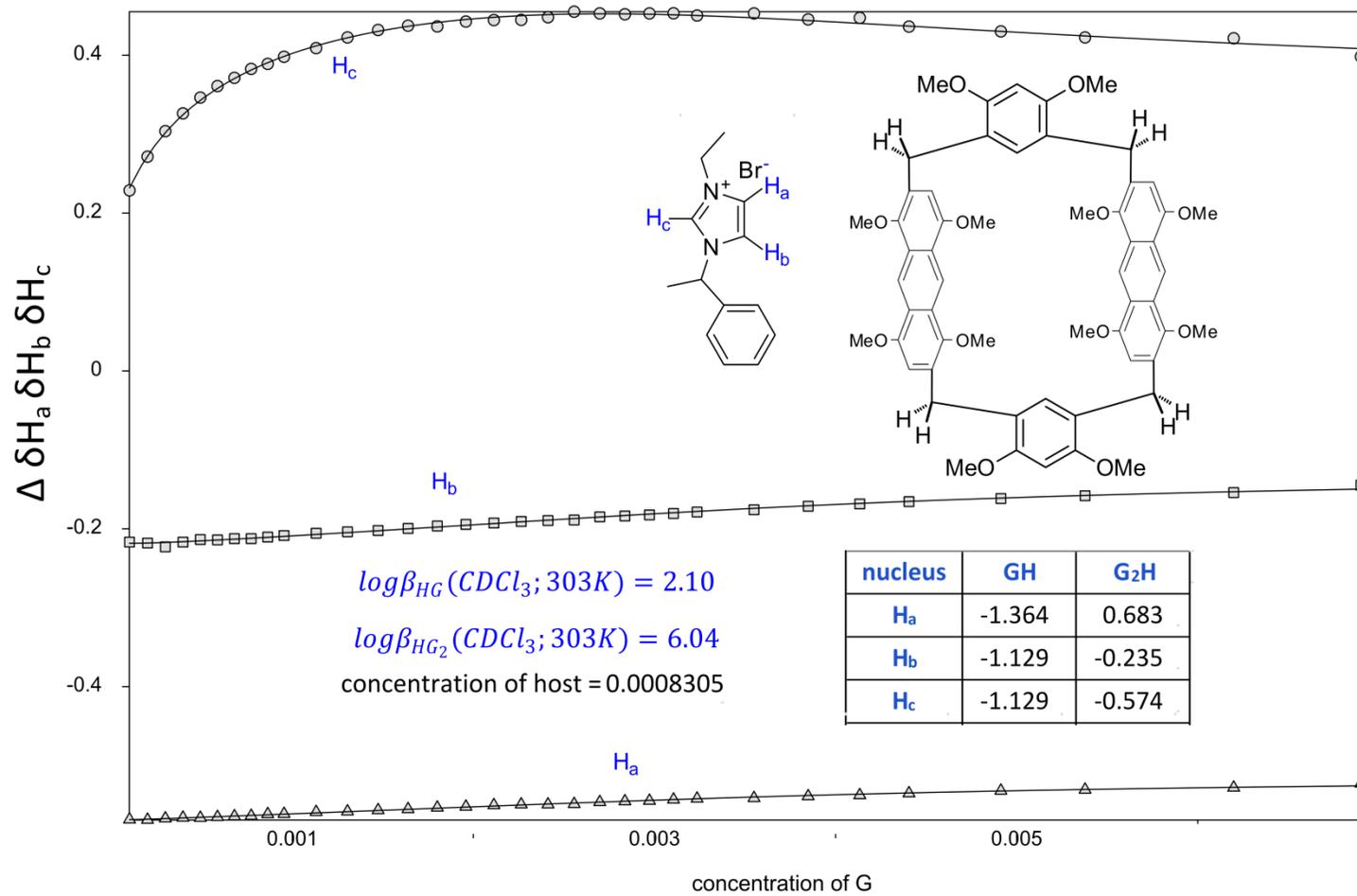


Figure 13: Complexation of *rac*-6 by 3 in CDCl₃ (303K).

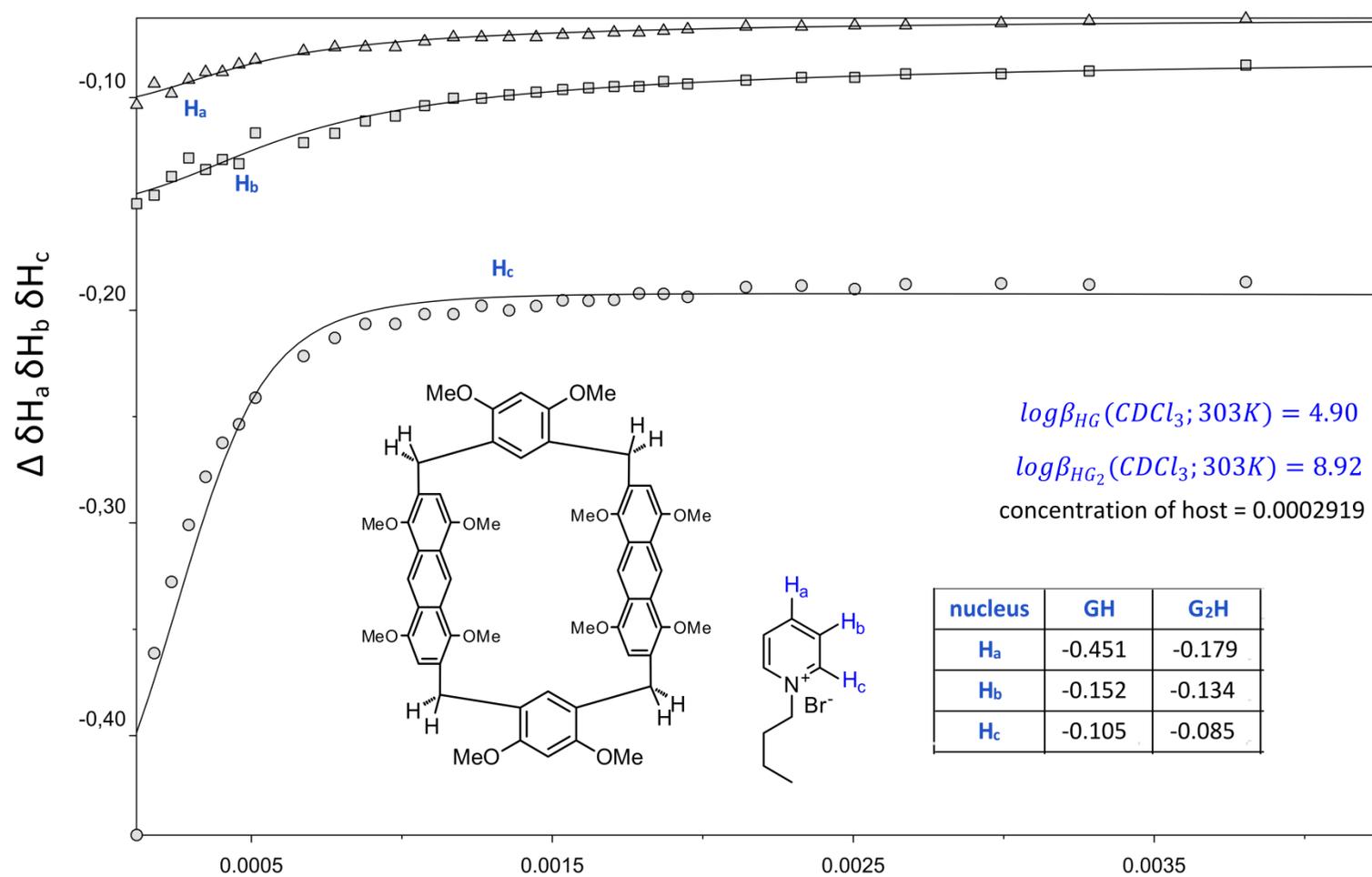


Figure 14: Complexation of **7** by **3** in CDCl_3 (303K).

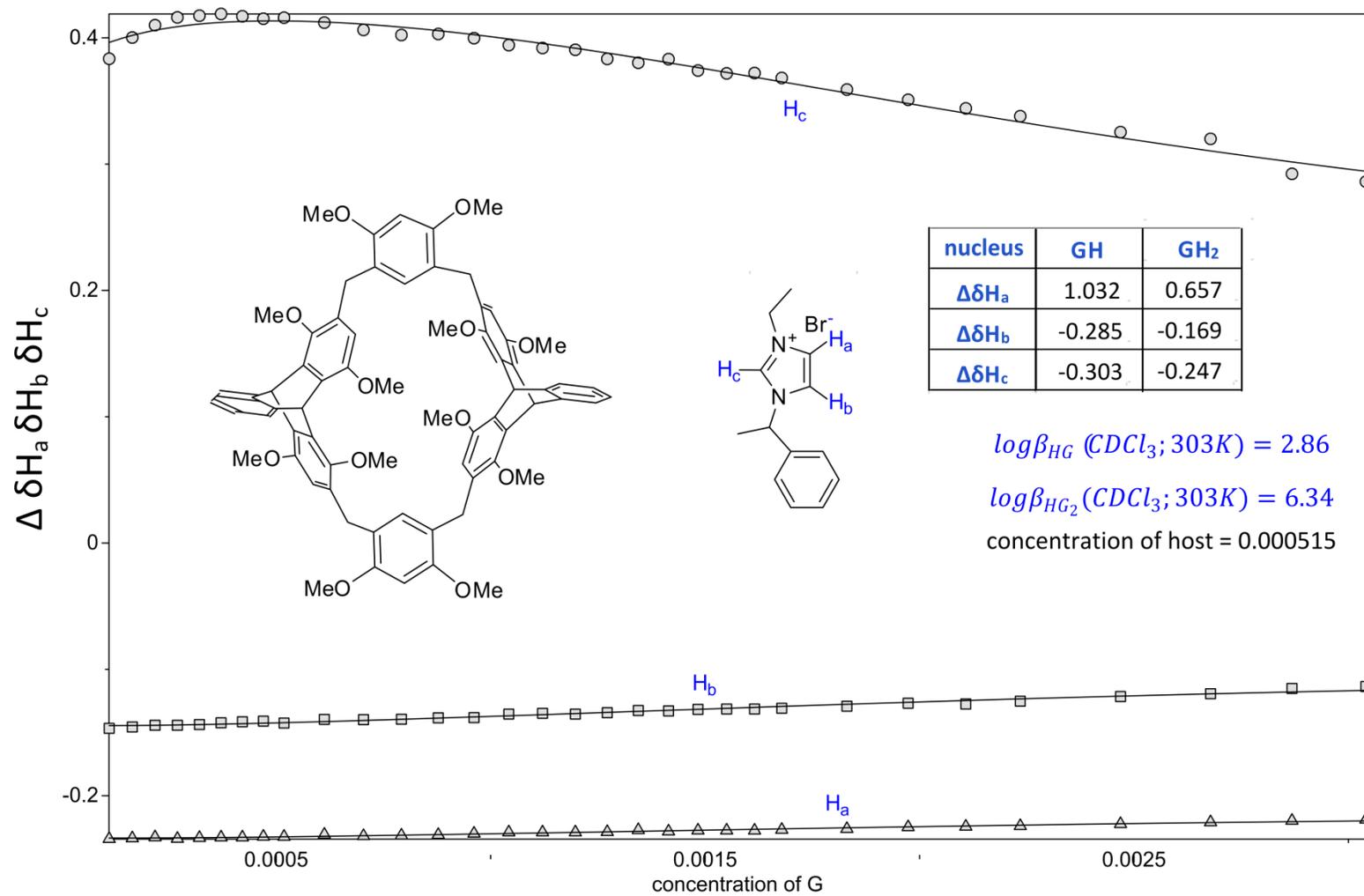


Figure 15: Complexation of *rac-6* by **5** in CDCl₃ (303K).

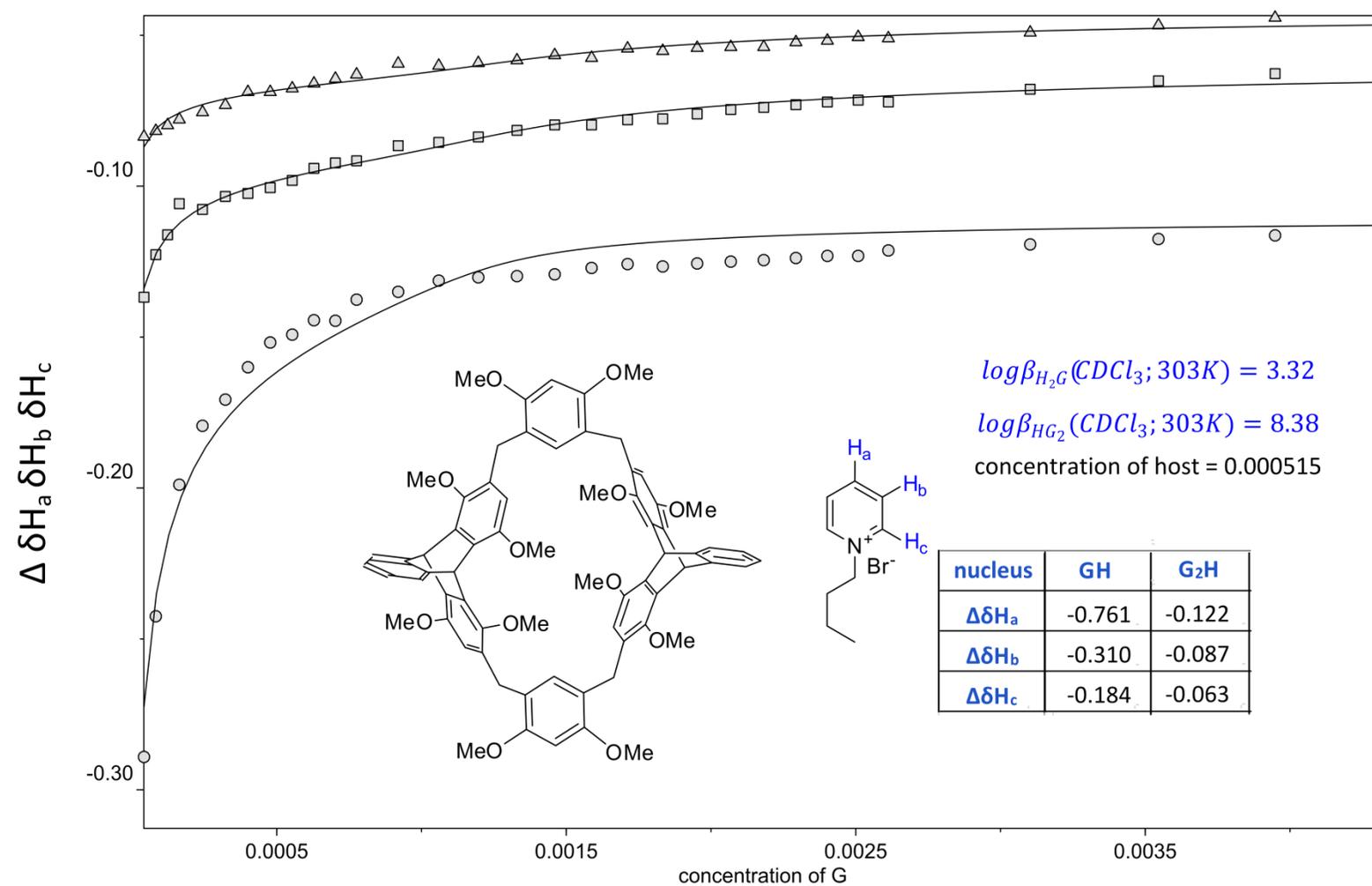


Figure 16: Complexation of 7 by 5 in $CDCl_3$ (303K)

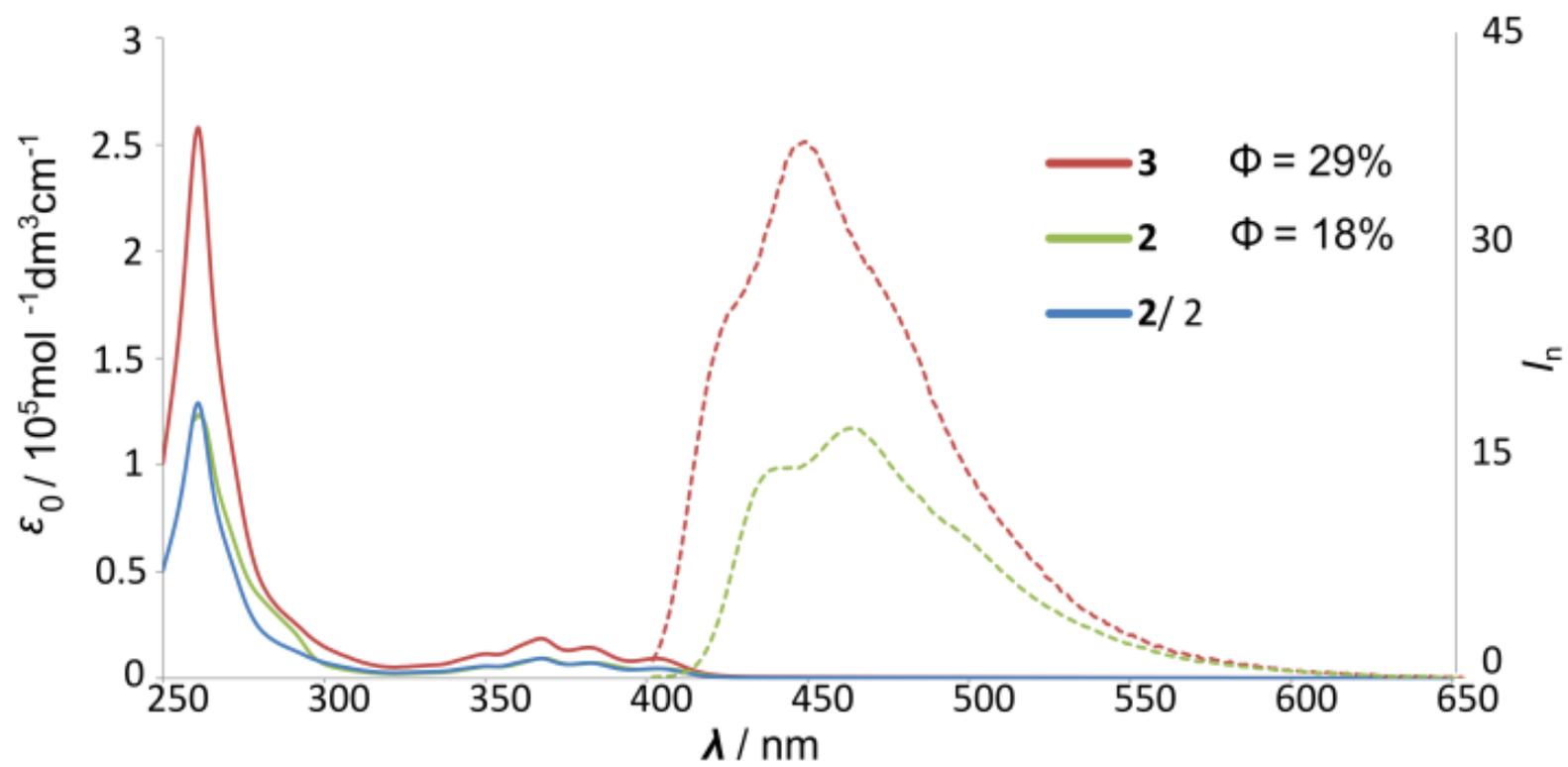


Figure 17: UV-Vis and fluorescence spectra of monomer **2** and macrocycle **3** in chloroform.

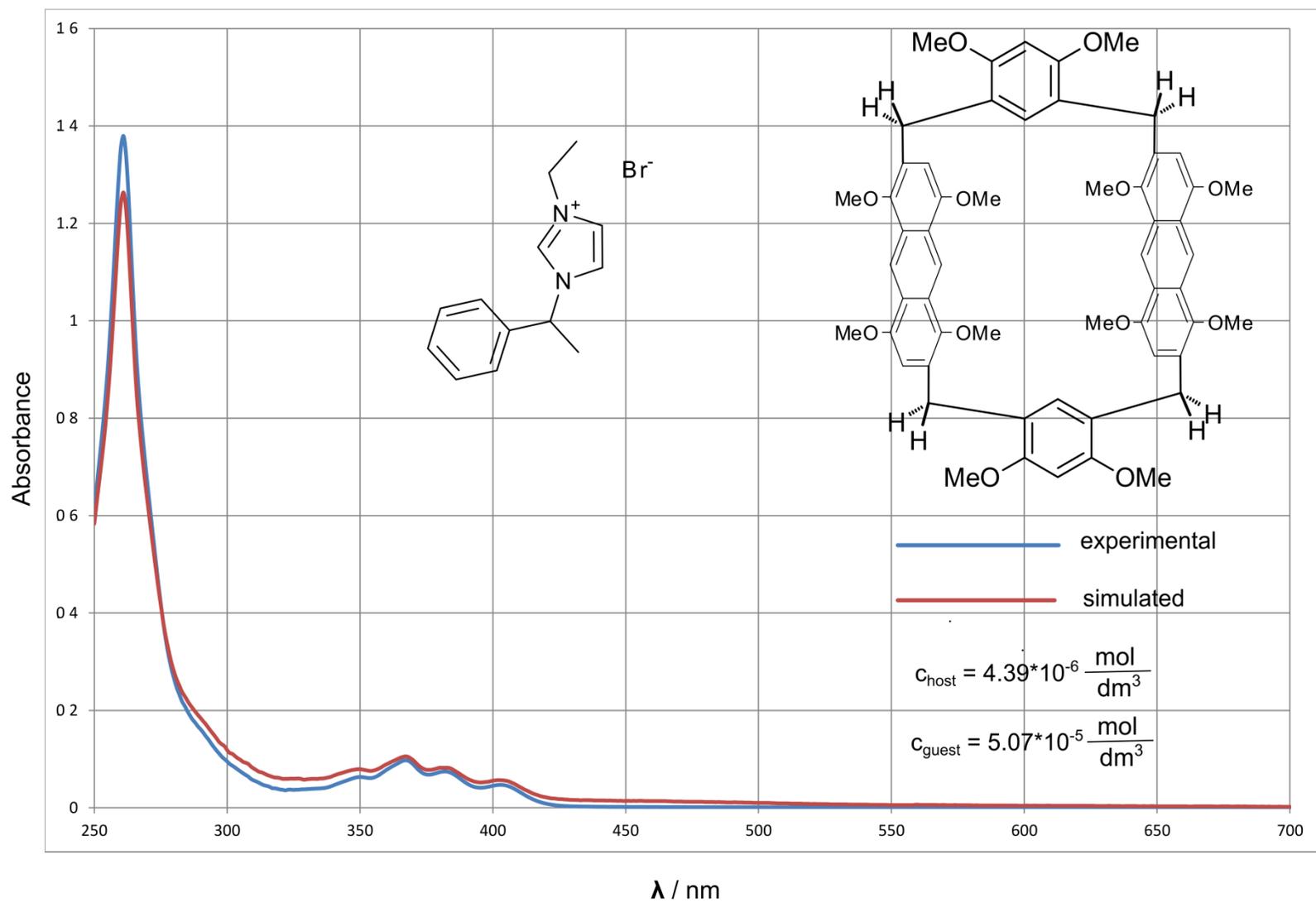


Figure 18: UV-vis spectra of **3** and *rac*-**6**.

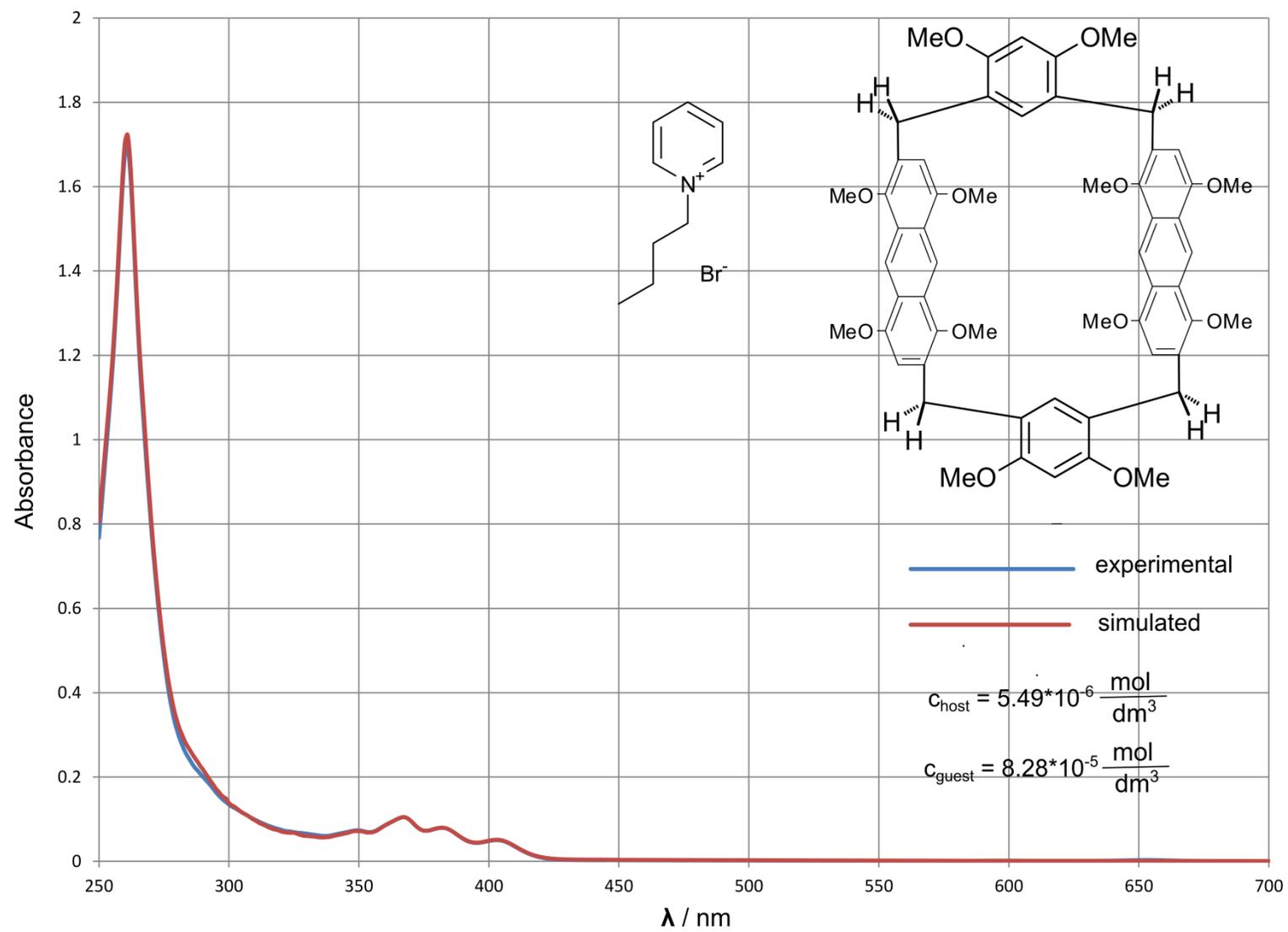


Figure 19: UV-Vis spectra of 3 and 7.

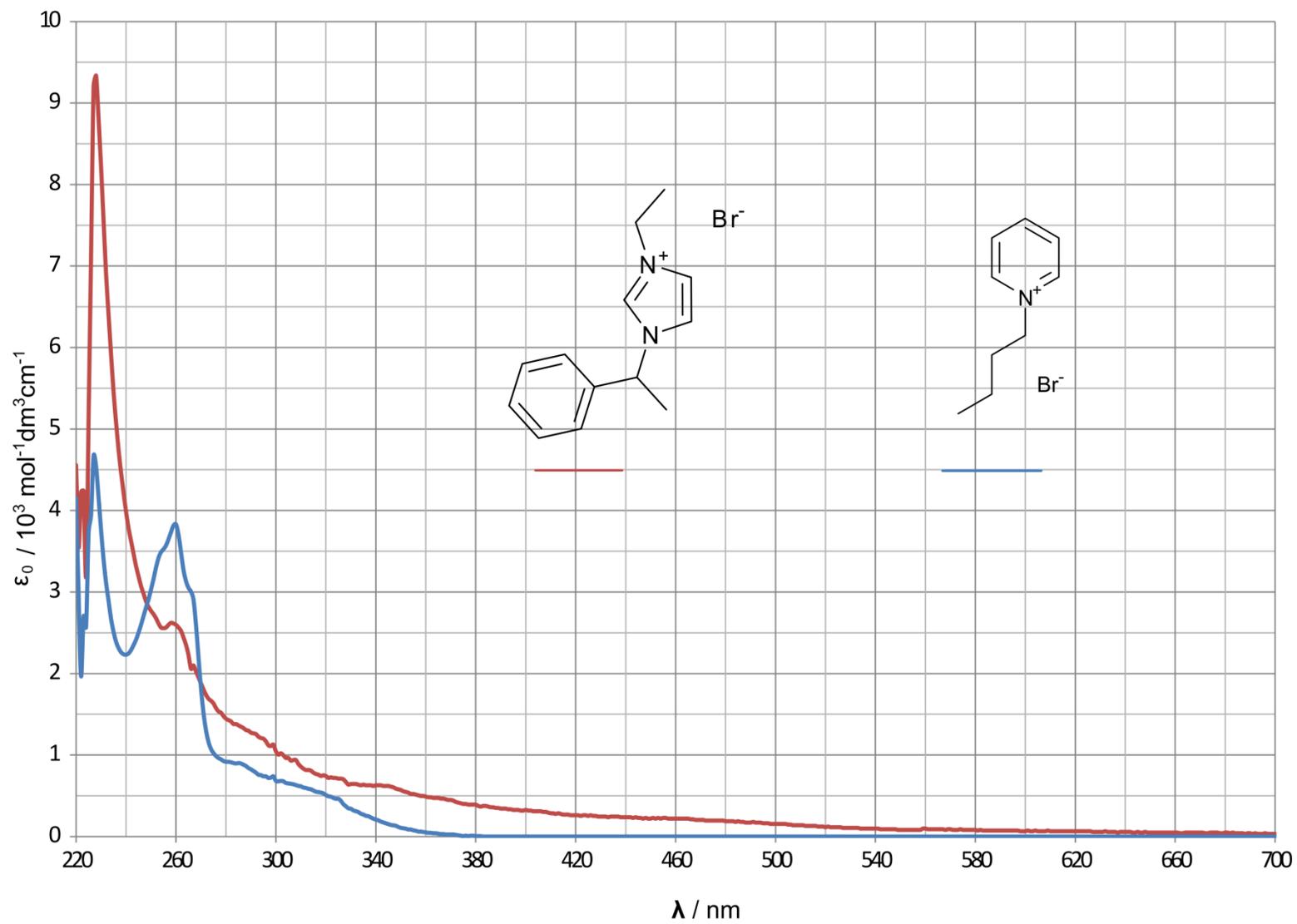


Figure 20: UV-Vis spectra of guests *rac-6* and *7*.