

Supporting Information

Table S1. Crystal data and structure refinement for *ct*-[RuCl(CO)(dppb)(bipy)]PF₆ (**1**) and *cc*-[RuCl(CO)(dppb)(phen)]PF₆ (**6**).

Table S2. Selected angles for *ct*-[RuCl(CO)(dppb)(bipy)]PF₆ and *cc*-[RuCl(CO)(dppb)(bipy)]PF₆.

Table S3. Contributions, %, of the composing atoms in the frontier orbitals of the Ru-complexes *ct*-[RuCl(CO)(dppb)(bipy)]PF₆ (**1**), *tc*-[RuCl(CO)(dppb)(bipy)]PF₆ (**3**) and *cc*-[RuCl(CO)(dppb)(bipy)]PF₆ (**5**) calculated using the B3LYP/[Ru:SDD;C,H,P,N,Cl:6-311+G**] approach.

Table S4. Natural Bonding Orbitals (NBO) charges on the Ru and Ru-bound atoms of the complexes *tc*-[RuCl(CO)(dppb)(bipy)]PF₆ (**3**), *cc*-[RuCl(CO)(dppb)(bipy)]PF₆ (**5**), and *ct*-[RuCl(CO)(dppb)(bipy)]PF₆ (**1**), calculated using the B3LYP/[Ru:SDD;C,H,P,N,Cl:6-311+G**] approach.

Figure S1. (C) ¹³C{¹H} NMR spectrum of *ct*-[RuCl(CO)(dppb)(bipy)]PF₆ complex in DMSO-*d*₆ at 300 K (A) Expanded regions of 110 – 200 ppm, (B) 5 – 30 ppm at different times.

Figure S2. (C) ¹³C{¹H} NMR spectrum of *cc*-[RuCl(CO)(dppb)(bipy)]PF₆ isomer in DMSO-*d*₆ at 300 K, (A) Expanded regions of 120 – 210 ppm and (B) 21 – 31 ppm at different times.

Figure S3. (C) ¹³C{¹H} NMR spectrum of *tc*-[RuCl(CO)(dppb)(bipy)]PF₆ complex in DMSO-*d*₆ at 300 K. (A) Expanded regions of 110 – 200 ppm and (B) 23 – 31 ppm at different times.

Figure S4. A) ³¹P{¹H} NMR spectrum of solution after electrolysis of *ct*-[RuCl(CO)(dppb)(bipy)]PF₆ isomer in CH₃CN and B) cyclic voltammogram of the *ct*-[RuCl(CH₃CN)(dppb)(bipy)]PF₆ in CH₃CN.

Figure S5. A) ³¹P{¹H} spectrum of solution after electrolysis of *tc*-[RuCl(CH₃CN)(dppb)(bipy)]PF₆ carried out for 3 h; B) cyclic voltammogram of *tc*-[RuCl(CH₃CN)(dppb)(bipy)]PF₆, C) ³¹P NMR spectra of solution after electrolysis of *cc*-[RuCl(CO)(dppb)(bipy)]PF₆ for 4 h and D) cyclic voltammogram of *cc*-[RuCl(CH₃CN)(dppb)(bipy)]PF₆. Conditions: Pt electrode vs Ag/AgCl, TBAP 0.1 mol L⁻¹ in CH₃CN.

Figure S6. ¹³C{¹H} NMR spectra of electrolysis products. A) *ct*-[RuCl(CH₃CN)(dppb)(bipy)]PF₆, B) *tc*-[RuCl(CH₃CN)(dppb)(bipy)]PF₆ and C) *cc*-[RuCl(CH₃CN)(dppb)(bipy)]PF₆ in DMSO-*d*₆.

Table S1. Crystal data and structure refinement for *ct*-[RuCl(CO)(dppb)(bipy)]PF₆ (**1**) and *cc*-[RuCl(CO)(dppb)(phen)]PF₆ (**6**).

Data	Complex (1)	Complex (6)
Empirical formula	C ₃₉ H ₃₆ ClF ₆ N ₂ OP ₃ Ru	C ₄₁ H ₃₆ ClF ₆ N ₂ OP ₃ Ru
Molecular weight	892.13	916.15
Color	Yellow	Yellow
Crystal system	Monoclinic	Monoclinic
Space group	P21/c	P21/c
Unit cell dimensions (Å; °)	a = 14.6581(2) b = 16.2338(2); c = 16.5500(2) β = 92.759(1)	a = 12.4950(4) Å b = 16.3401(6); c = 19.2663(6) β = 101.08
Volume (Å ³)	3933.62(9)	3860.2(2)
Unit cell, Z	4	4
Crystal size (mm ³)	0.22 x 0.22 x 0.19	0.05 x 0.09 x 0.40
Density (calculated; Mg/m ³)	1.506	1.576
Temperature (K)	293(2)	293(2)
Absorption coefficient (mm ⁻¹)	0.651	0.666
F(000)	1808	1856
Wavelength (Mo— Kα) (Å)	0.71073	0.71073
Theta range for data collection (°)	3.10 to 32.03	2.963 to 26.374
Index ranges	-18 ≤ h ≤ 18; -20 ≤ k ≤ 20; -21 ≤ l ≤ 24	-15≤h≤15; -20≤k≤20; -24≤l≤24
Completeness to theta	79.6 %	99.4 %
Reflections collected	10900	28914
Data / restraints / parameters	10900 / 478	7863 / 0 / 496
R1; wR2 [I>2σ (I)]	R1 = 0.0533, wR2 = 0.1388	R1 = 0.0589; wR2 = 0.1338
R1; wR2 (Total)	R1 = 0.0783, wR2 = 0.1659	R = 0.0589; R` = 0.1338
S	1.074	1.201
Largest diff. peak and hole	0.803 and -1.178	0.453 and -0.931

Table S2. Selected angles for *ct*-[RuCl(CO)(dppb)(bipy)]PF₆ and *cc*-[RuCl(CO)(dppb)(phen)]PF₆.

	<i>ct</i> -[RuCl(CO)(dppb)(bipy)]PF ₆	<i>cc</i> -[RuCl(CO)(dppb)(phen)]PF ₆	
	Angles [°]	Angles [°]	
C(1)-Ru-N(1)	89.11(12)	C(1)-Ru-N(1)	169.54(14)
C(1)-Ru-N(2)	86.98(13)	C(1)-Ru-N(2)	91.79(14)
N(1)-Ru-N(2)	77.11(12)	N(2)-Ru-N(1)	77.93(11)
C(1)-Ru-P(1)	88.54(11)	C(1)-Ru-P(1)	94.87(12)
N(1)-Ru-P(1)	105.90(8)	N(2)-Ru-P(1)	89.79(9)
N(2)-Ru-P(1)	174.57(8)	N(1)-Ru-P(1)	87.14(8)
C(1)-Ru-Cl	94.91(11)	C(1)-Ru-P(2)	89.32(12)
N(1)-Ru-Cl	168.63(8)	N(2)-Ru-P(2)	170.33(9)
N(2)-Ru-Cl	92.46(9)	N(1)-Ru-P(2)	100.48(8)
P(1)-Ru-Cl	84.87(3)	P(1)-Ru-P(2)	99.68(3)
C(1)-Ru-P(2)	173.00(12)	C(1)-Ru-Cl	92.28(12)
N(1)-Ru-P(2)	92.54(7)	N(2)-Ru-Cl	83.27(9)
N(2)-Ru-P(2)	86.76(8)	N(1)-Ru-Cl	84.64(8)
P(1)-Ru-P(2)	97.53(3)	P(1)-Ru-Cl	170.19(3)
Cl-Ru-P(2)	82.24(3)	P(2)-Ru-Cl	87.09(3)
C(1)-Ru-P(2)	173.00(12)	C(1)-Ru-N(2)	91.79(14)
N(1)-Ru-P(2)	92.54(7)	C(1)-Ru-N(1)	169.54(14)

Table S3. Contributions, %, of the composing atoms in the frontier orbitals of the Ru-complexes *ct*-[RuCl(CO)(dppb)(bipy)]PF₆ (**1**), *tc*-[RuCl(CO)(dppb)(bipy)]PF₆ (**3**) and *cc*-[RuCl(CO)(dppb)(bipy)]PF₆ (**5**) calculated using the B3LYP/[Ru:SDD;C,H,P,N,Cl:6-311+G**] approach.

Species	HOMO	LUMO
(3) opt.	(C+H) 87	(C=O) 27; (C+H) 56
(5) opt.	(C+H) 93	(C=O) 11; (C+H) 82
(1) opt.	Ru: 54; Cl: 34; (C+H) 10	N: 20; (C+H) 77
(1) X-ray*	Ru: 58; Cl: 33	N: 17; (C+H) 80

**ct*-[RuCl(CO)(dppb)(bipy)]PF₆ complex calculated using X-ray coordinates.

Table S4. Natural Bonding Orbitals (NBO) charges on the Ru and Ru-bound atoms of the complexes *tc*-[RuCl(CO)(dppb)(bipy)]PF₆ (**3**), *cc*-[RuCl(CO)(dppb)(bipy)]PF₆ (**5**), and *ct*-[RuCl(CO)(dppb)(bipy)]PF₆ (**1**), calculated using the B3LYP/[Ru:SDD;C,H,P,N,Cl:6-311+G**] approach.

atoms	complex 3 (opt.)	complex 5 (opt.)	complex 1 (opt.)	complex 1 (X-Ray)*
Ru	-1.38	-1.28	-0.66	-0.87
N1	-0.40	-0.69	-0.36	-0.36
N2	-0.35	-0.30	-0.37	-0.37
P1	2.79	1.01	1.20	1.30
P2	1.02	0.62	1.09	1.17
Cl	-0.27	-0.26	-0.40	-0.35
C(CO)	0.62	0.60	0.77	0.77
O	-0.50	-0.63	-0.44	-0.42

**ct*-[RuCl(CO)(dppb)(bipy)]PF₆ complex calculated using X-ray coordinates.

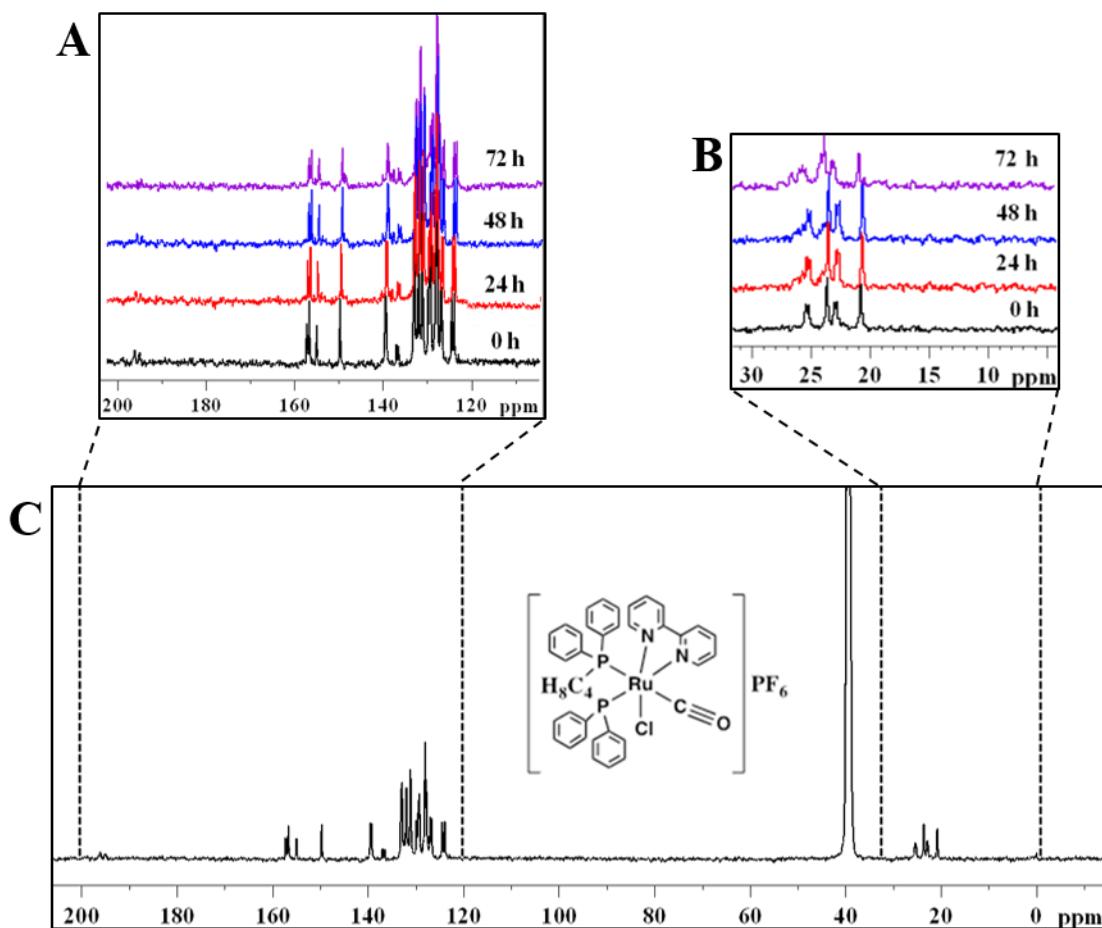


Figure S1. (C) $^{13}\text{C}\{^1\text{H}\}$ NMR spectrum of *ct*-[RuCl(CO)(dppb)(bipy)]PF₆ complex in DMSO-*d*₆ at 300 k (A) Expanded regions of 110 – 200 ppm, (B) 5 – 30 ppm at different times.

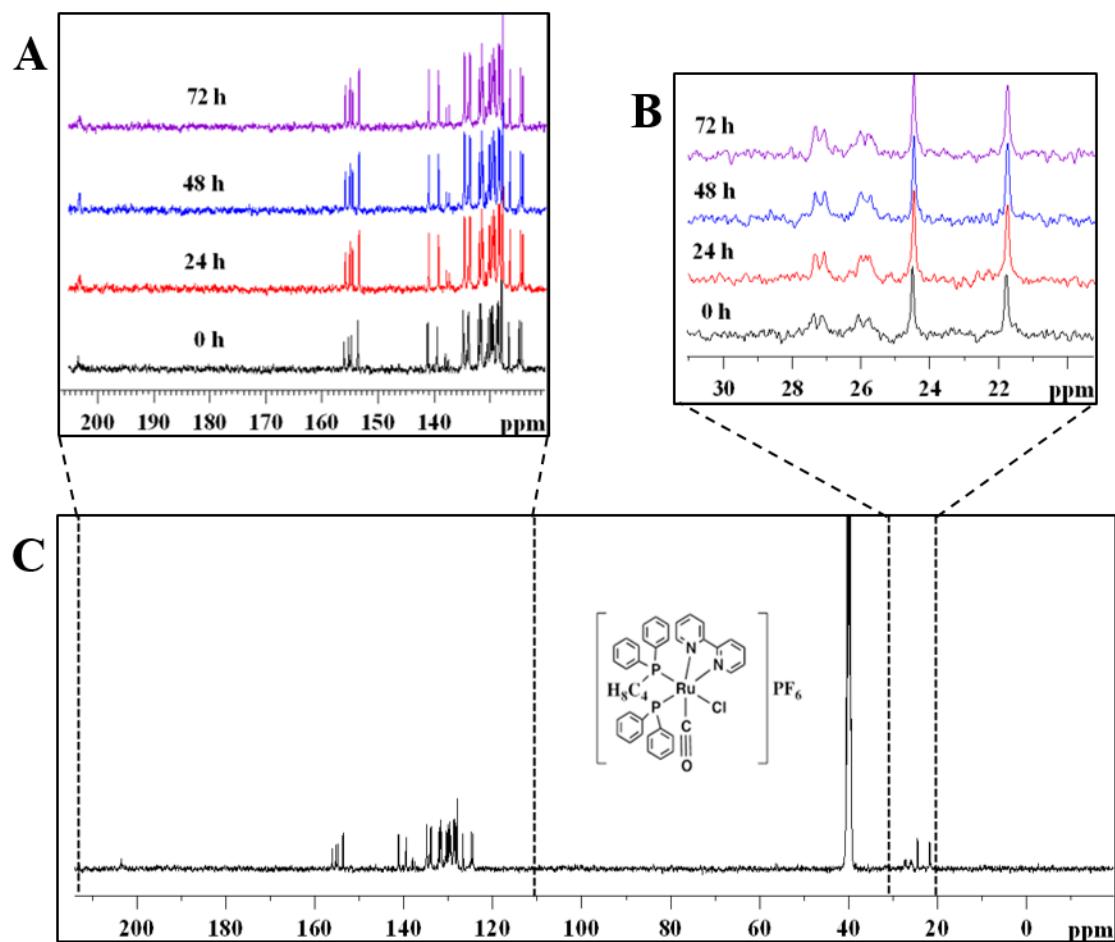


Figure S2. (C) $^{13}\text{C}\{^1\text{H}\}$ NMR spectrum of *cc*-[RuCl(CO)(dppb)(bipy)]PF₆ isomer in DMSO-*d*₆ at 300 K, (A) Expanded regions of 120 – 210 ppm and (B) 21 – 31 ppm at different times.

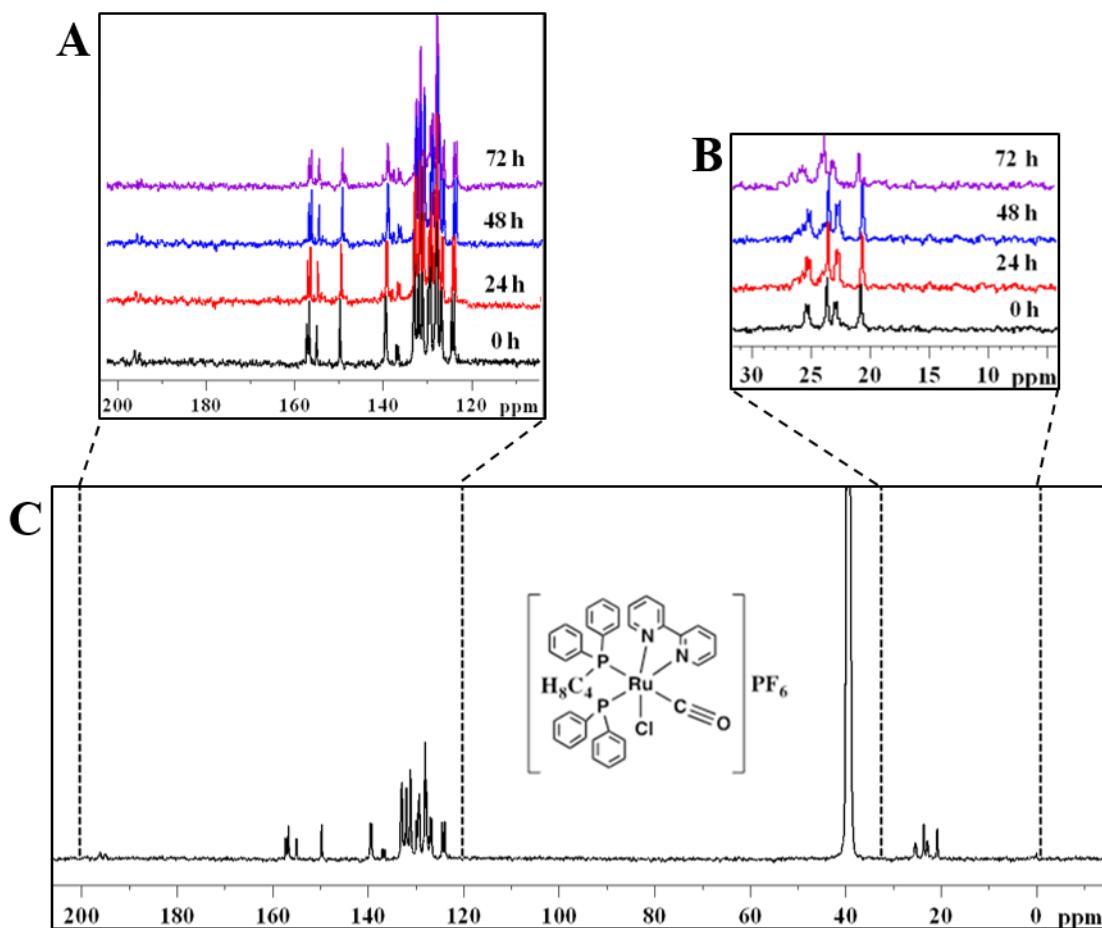


Figure S3. (C) $^{13}\text{C}\{^1\text{H}\}$ NMR spectrum of *tc*-[RuCl(CO)(dppb)(bipy)] PF_6 complex in DMSO- d_6 at 300 K. (A) Expanded regions of 110 – 200 ppm and (B) 23 – 31 ppm at different times.

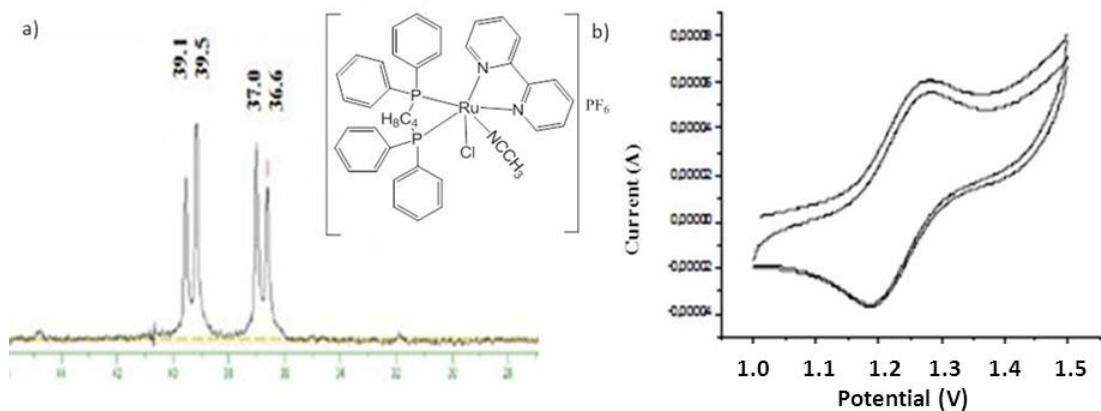


Figure S4. A) $^{31}\text{P}\{\text{H}\}$ NMR spectrum of solution after electrolysis of *ct*- $[\text{RuCl}(\text{CO})(\text{dppb})(\text{bipy})]\text{PF}_6$ isomer in CH_3CN and B) cyclic voltammogram of the *ct*- $[\text{RuCl}(\text{CH}_3\text{CN})(\text{dppb})(\text{bipy})]\text{PF}_6$ in CH_3CN .

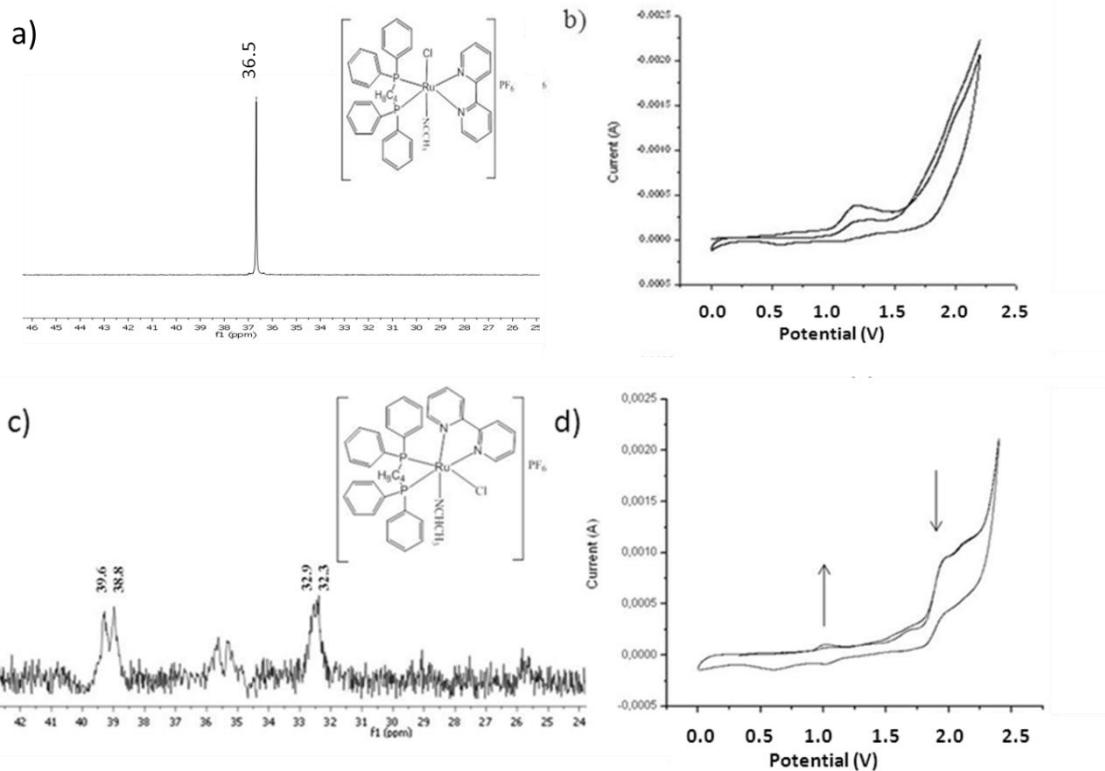


Figure S5. A) $^{31}\text{P}\{\text{H}\}$ spectrum of solution after electrolysis of $\text{tc}-[\text{RuCl}(\text{CH}_3\text{CN})(\text{dppb})(\text{bipy})]\text{PF}_6$ carried out for 3 h; B) cyclic voltammogram of $\text{tc}-[\text{RuCl}(\text{CH}_3\text{CN})(\text{dppb})(\text{bipy})]\text{PF}_6$, C) ^{31}P NMR spectra of solution after electrolysis of $\text{cc}-[\text{RuCl}(\text{CH}_3\text{CN})(\text{dppb})(\text{bipy})]\text{PF}_6$ for 4 h and D) cyclic voltammogram of $\text{cc}-[\text{RuCl}(\text{CH}_3\text{CN})(\text{dppb})(\text{bipy})]\text{PF}_6$. Conditions: Pt electrode vs Ag/AgCl, TBAP 0.1 mol L⁻¹ in CH₃CN.

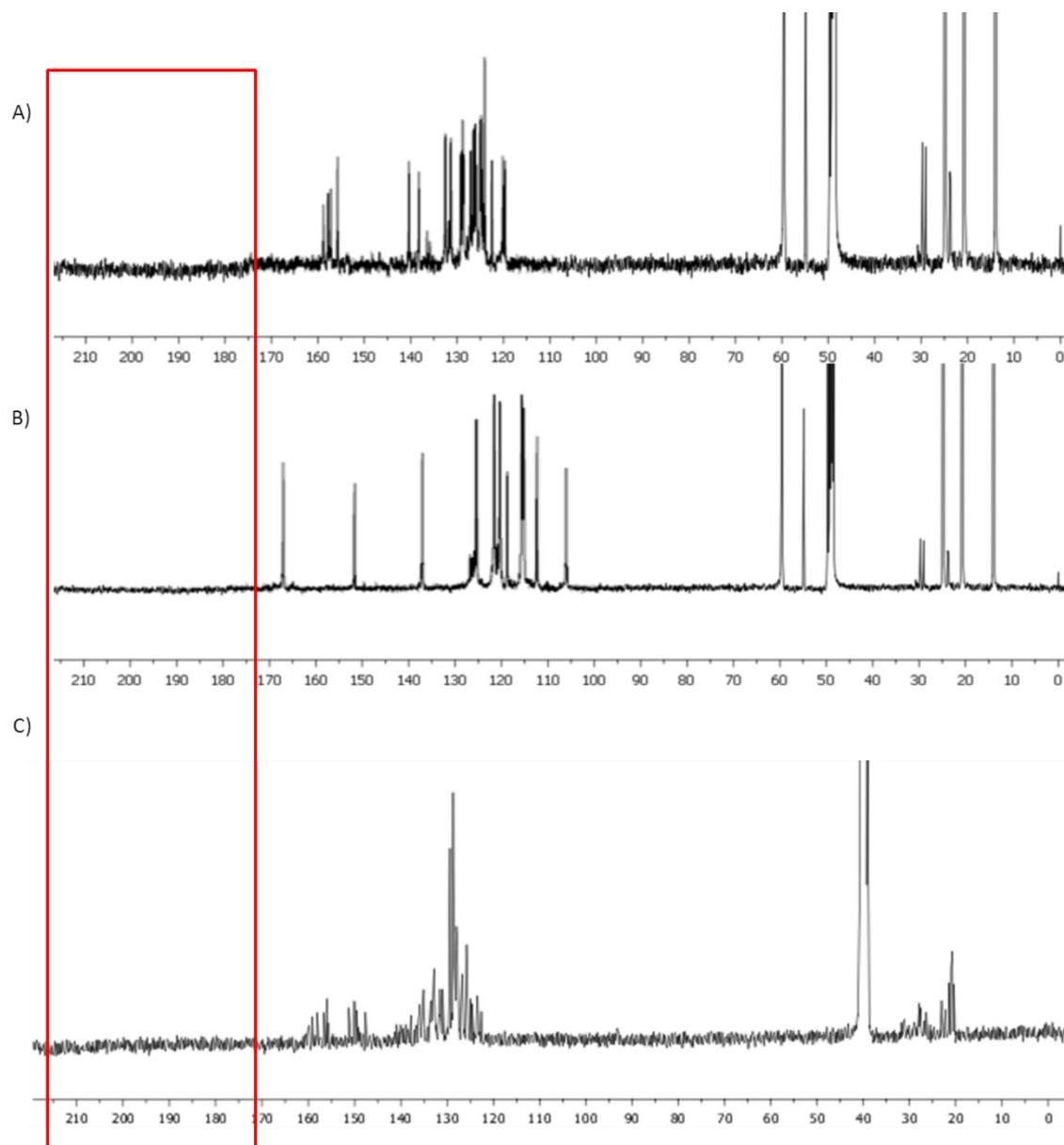


Figure S6. $^{13}\text{C}\{\text{H}\}$ NMR spectra of electrolysis products. A) *ct*- $[\text{RuCl}(\text{CH}_3\text{CN})(\text{dppb})(\text{bipy})]\text{PF}_6$, B) *tc*- $[\text{RuCl}(\text{CH}_3\text{CN})(\text{dppb})(\text{bipy})]\text{PF}_6$ and C) *cc*- $[\text{RuCl}(\text{CH}_3\text{CN})(\text{dppb})(\text{bipy})]\text{PF}_6$.