

Supplementary Information

Photoluminescence and electroluminescence of an iridium(III) complex with 2',6'-bis(trifluoromethyl)-2,4'-bipyridine and 2-(5-phenyl-1,3,4-thiadiazol-2-yl)phenol ligands

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Table S1. Parameters associated with the single crystal diffraction data collection for **Ir(BTBP)₂TDZ**.

	Ir(BTBP)₂TDZ
Formula	C ₃₈ H ₁₉ F ₁₂ IrN ₆ OS
Formula weight	1027.85
T (K)	296(2)
Wavelength (Å)	0.71073
Crystal system	Monoclinic
Space group	<i>P</i> 2 ₁ / <i>n</i>
<i>a</i> (Å)	10.5833(4)
<i>b</i> (Å)	24.6297(8)
<i>c</i> (Å)	15.5661(5)
α (deg)	90.00
β (deg)	102.3490(10)
γ (deg)	90.00
<i>V</i> (Å ³)	3963.6(2)
<i>Z</i>	4
ρ_{calcd} (g/cm ³)	1.722
μ (Mo K α) (mm ⁻¹)	3.517
<i>F</i> (000)	1992
Range of transm factors (deg)	1.574-25.009
Reflns collected	22135
Unique(<i>R</i> _{int})	6990(0.0304)
<i>R</i> _{<i>I</i>} ^{<i>a</i>} , <i>wR</i> ₂ ^{<i>b</i>} [<i>I</i> > 2 <i>s</i> (<i>I</i>)]	0.0286, 0.0686
<i>R</i> _{<i>I</i>} ^{<i>a</i>} , <i>wR</i> ₂ ^{<i>b</i>} (all data)	0.0358, 0.0707
GOF on <i>F</i> ²	1.057
CCDC number	1524694

$$R_I^a = \frac{\sum ||F_o| - |F_c||}{\sum F_o}, \quad wR_2^b = \left[\frac{\sum w(F_o^2 - F_c^2)^2}{\sum w(F_o^2)} \right]^{1/2}$$

$$\text{GOF} = \left[\frac{\sum w|F_o| - |F_c|}{\sum w|F_o|} \right]^2 / (N_{\text{obs}} - N_{\text{param}})^{1/2}$$

Table S2. The table of selected bond lengths and angles of **Ir(BTBP)₂TDZ**.

Selected Bond	Bond Length (Å)	Selected Angle	Bond Angle (°)
Ir(1)-C(1)	2.022(4)	C(22)-Ir(1)-C(1)	99.02 (16)
Ir(1)-C(22)	2.020(4)	C(22)-Ir(1)-N(3)	80.46 (17)
Ir(1)-N(3)	2.032(3)	C(1)-Ir(1)-N(3)	101.88 (15)
Ir(1)-N(2)	2.049(3)	C(22)-Ir(1)-N(2)	105.59 (16)
Ir(1)-N(1)	2.125(3)	C(1)-Ir(1)-N(2)	80.94 (14)
Ir(1)-O(1)	2.110(3)	N(3)-Ir(1)-N(2)	172.96 (13)
		C(22)-Ir(1)-N(1)	90.94 (15)
		C(1)-Ir(1)-N(1)	168.81 (14)
		N(3)-Ir(1)-N(1)	84.70 (13)
		N(2)-Ir(1)-N(1)	91.53 (12)
		C(22)-Ir(1)-O(1)	171.25 (14)
		C(1)-Ir(1)-O(1)	85.61 (14)
		N(3)-Ir(1)-O(1)	91.35 (13)
		N(2)-Ir(1)-O(1)	82.40 (12)
		N(1)-Ir(1)-O(1)	85.20 (12)

Table S3. Electrochemical data of **Ir(BTBP)₂TDZ**.

Complex	E_{ox} (V) ^{a)}	E_{red} (V)	HOMO (eV) ^{b)}	LUMO (eV)	E_g (eV)
Ir(BTBP)₂TDZ	0.93	-1.42	-5.58	-3.04	2.54

^{a)} Oxidation potential measured in solution at 0.05 mV/s.

^{b)} The HOMO energy levels were calculated from the equation: HOMO (eV) = $E_{ox} - E_{Fc^+/Fc}^+ + 4.8$.

Table S4. Percentage distributions of HOMO and LUMO in **Ir(BTBP)₂TDZ**.

Complex	Orbital	Energy/eV (Calculated)	E_{gap} /eV (Calculated)	Composition (%)		
				Ir	BTBP	TDZ
Ir(BTBP)₂TDZ	HOMO	-5.493	3.301	14.73	5.31	79.96
	LUMO	-2.192		1.39	41.77	56.84