## **Electronic Supplementary Information**

## Synthesis and Properties Enhancement of Metal Nanoclusters

## **Templated on Biological Molecule/Ionic Liquids complex**

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**Figure S1.** The photographs of AgNCs@GSH/[C<sub>4</sub>min] NPs under visible light, which the concentration of  $[C_4min][BF_4]$  increased by degrees (the volume ratio of water:  $[C_4min][BF_4]$  from 1 to 7 is 9:1, 7:1, 5:1, 4:1, 3:1, 2:1, 1.5:1 respectively).



**Figure S2.** The photographs of (a) AuNCs@BSA/[C<sub>4</sub>min], (b) CuNCs@BSA/[C<sub>4</sub>min] and (c) gelatinous AgNCs@GSH/[  $C_2$ min] NPs synthesized in No. 2 concentration (Table S1, a) under visible light.

No.	The Volume Ratio of	AgNO <sub>3</sub>	GSH	NaBH <sub>4</sub>	[C mim][DE ]	H <sub>2</sub> O
	water: [C <sub>4</sub> min][BF <sub>4</sub> ]	(20 mM)	(50 mM)	(60 mM)		
1	3:1	100 µL	120 µL	200 µL	450 μL	1350 µL
2	2:1	100 µL	120 µL	$200 \ \mu L$	600 μL	1200 µL
3	1.5:1	100 µL	120 µL	$200\;\mu L$	720 µL	1080 µL

Table S1. The dosage of reagents in the preparation of MNCs (a) AgNCs@GSH/[ $C_n$ mim] NPs with different concentration of [ $C_n$ min][BF<sub>4</sub>] (n=2, 4)

<u> </u>							
AgNO <sub>3</sub>	GSH	NaBH <sub>4</sub>	[C mim][BF.]	HaO			
(20 mM)	(50 mM)	(86 mM)		1120			
30 µL	100 µL	100 µL	600 μL	1200 µL			
(c) AuNCs@BSA/[C4mim]							
HAuCl <sub>4</sub>	BSA	NaOH		H <sub>2</sub> O			
(50mg/mL)	(20mg/mL)	(2 M)	[C <sub>4</sub> mim][BF <sub>4</sub> ]				
30 µL	450 μL	100 µL	600 μL	820 μL			
(d) CuNCs@BSA/[C4mim]							
CuSO <sub>4</sub>	BSA	NaOH		H <sub>2</sub> O			
(50mM)	(20mg/mL)	(2 M)	[C4mim][BF4]				
30 µL	450 μL	200 µL	600 μL	720 µL			
(e) AuNCs@BSA/SDS							
HAuCl <sub>4</sub>	BSA	NaOH	SDS				
(50mg/mL)	(20mg/mL)	(2 M)	(50 mM)				
30 µL	450 μL	100 µL	1420 μL				



**Figure S3.** The fluorescence spectra of AgNCs@GSH/[C<sub>2</sub>mim] NPs synthesized in No. 2 concentration (blue line), AgNCs@GSH/[C<sub>2</sub>mim] (red line) and AgNCs@GSH/[C<sub>4</sub>mim] (black line). AgNCs@GSH/[C<sub>n</sub>mim] was synthesized in a low concentration (Table S1, b).



**Figure S4.** The size distribution of MNCs: (a) AgNCs@GSH/[C<sub>2</sub>mim] synthesized in a low concentration (Table S1, b). (b) AuNCs@BSA/[C<sub>4</sub>mim]. (c) CuNCs@BSA/[C<sub>4</sub>mim]. (Red line: MNCs@template molecule, black line: MNCs@template molecule/ILs.)



Figure S5. The UV/Vis absorption spectra of MNCs

(a) The UV/Vis absorption spectra of AgNCs@GSH/[C<sub>2</sub>mim] NPs (black line) and AgNCs@GSH (red line).

(b) The UV/Vis absorption spectra of AuNCs@BSA/[C<sub>4</sub>mim] (black line) and AuNCs@BSA (red line).

(c) The UV/Vis absorption spectra of CuNCs@BSA/[C<sub>4</sub>mim] (blach line) and CuNCs@BSA (red line).



**Figure S6.** The emission spectra (423nm) of AuNCs@BSA (red line), the emission spectra (436nm) of AuNCs@BSA/SDS (green line) and the excitation (327nm) of AuNCs@BSA/SDS (black line).



**Figure S7.** The fluorescence intensity of AuNCs@BSA after adding various concentrations of  $[C_4min][BF_4]$ . The volumes of the reagents were provided in **Table S2.** 

	AuNCs@BSA	$H_2O$	[C <sub>4</sub> min][BF <sub>4</sub> ]
1	210 µL	90 µL	0 µL
2	210 µL	80 µL	10 µL
3	210 µL	40 µL	50 µL
4	210 µL	0 µL	90 μL

Table S3. The zeta potential of the MNCs

Sampla	A aNC a QCSH	AgNCs@GSH/	AuNCs@BSA	AuNCs@BSA/	CuNCs@BSA	CuNCs@BSA/
Sample	Agives@05fi	[C <sub>2</sub> mim] NPs		[C <sub>4</sub> mim]		[C <sub>4</sub> mim]
Zeta potential (mV)	-8.9	27.0	-18.0	-6.5	-15.1	-4.6





(a) [C<sub>2</sub>mim]BF<sub>4</sub> (the black line) and GSH/[C<sub>2</sub>mim] complex (the red line).

(b) [C<sub>4</sub>mim]BF<sub>4</sub> (the black line) and BSA/[C<sub>4</sub>mim] complex (the red line).

The peaks circled were showed in c and d. The concentrations of all the substances were identical with the ones in the corresponding synthesis system (Table S1, b and c).