

Electronic Supplementary Information

Organic matrix-induced formation of a discrete cyclic $[\text{Cl}_2(\text{H}_2\text{O})_2]^{2-}$ -cluster

Damir A. Safin, Maria G. Babashkina, Koen Robeyns and Yann Garcia*

Institute of Condensed Matter and Nanosciences, Molecules, Solids and Reactivity (IMCN/MOST), Université catholique de Louvain, Place L. Pasteur 1, 1348 Louvain-la-Neuve, Belgium. E-mail: yann.garcia@uclouvain.be

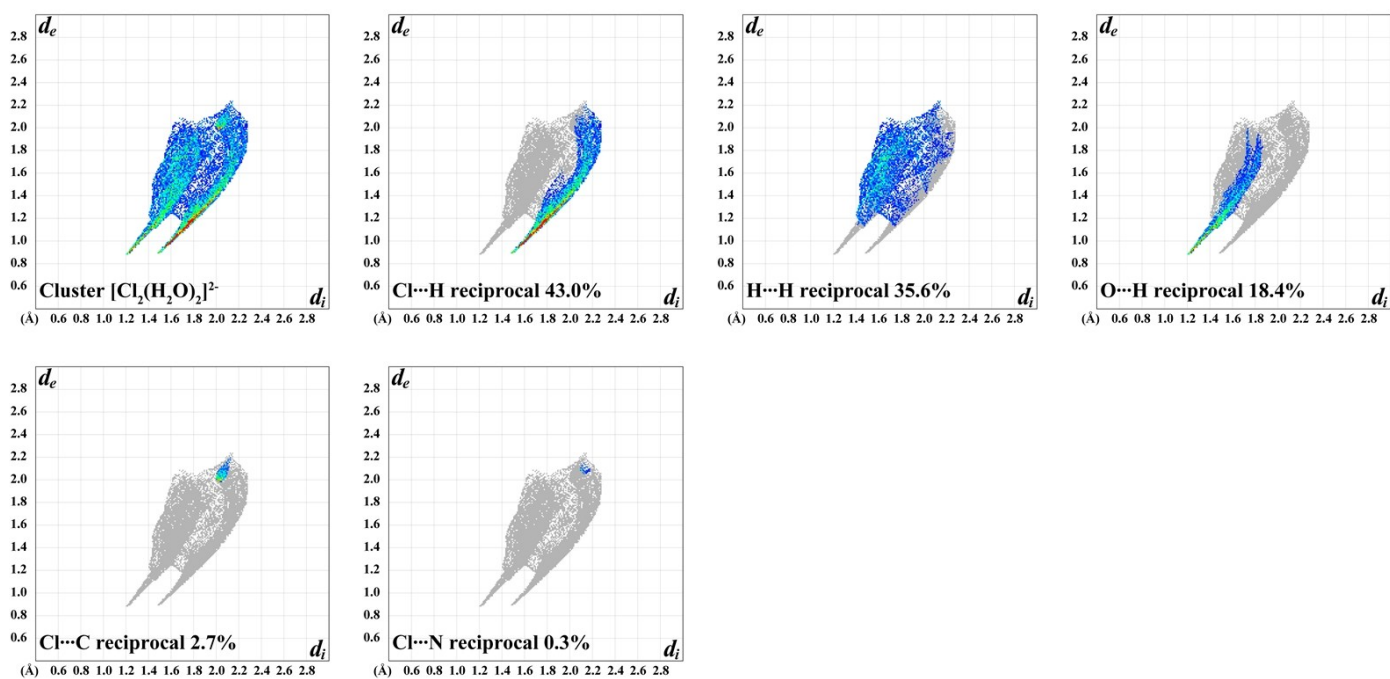


Fig. S1 2D fingerprint plots and decomposed 2D fingerprint plots of observed contacts for the $[\text{Cl}_2(\text{H}_2\text{O})_2]^{2-}$ cluster in the structure of **1**.

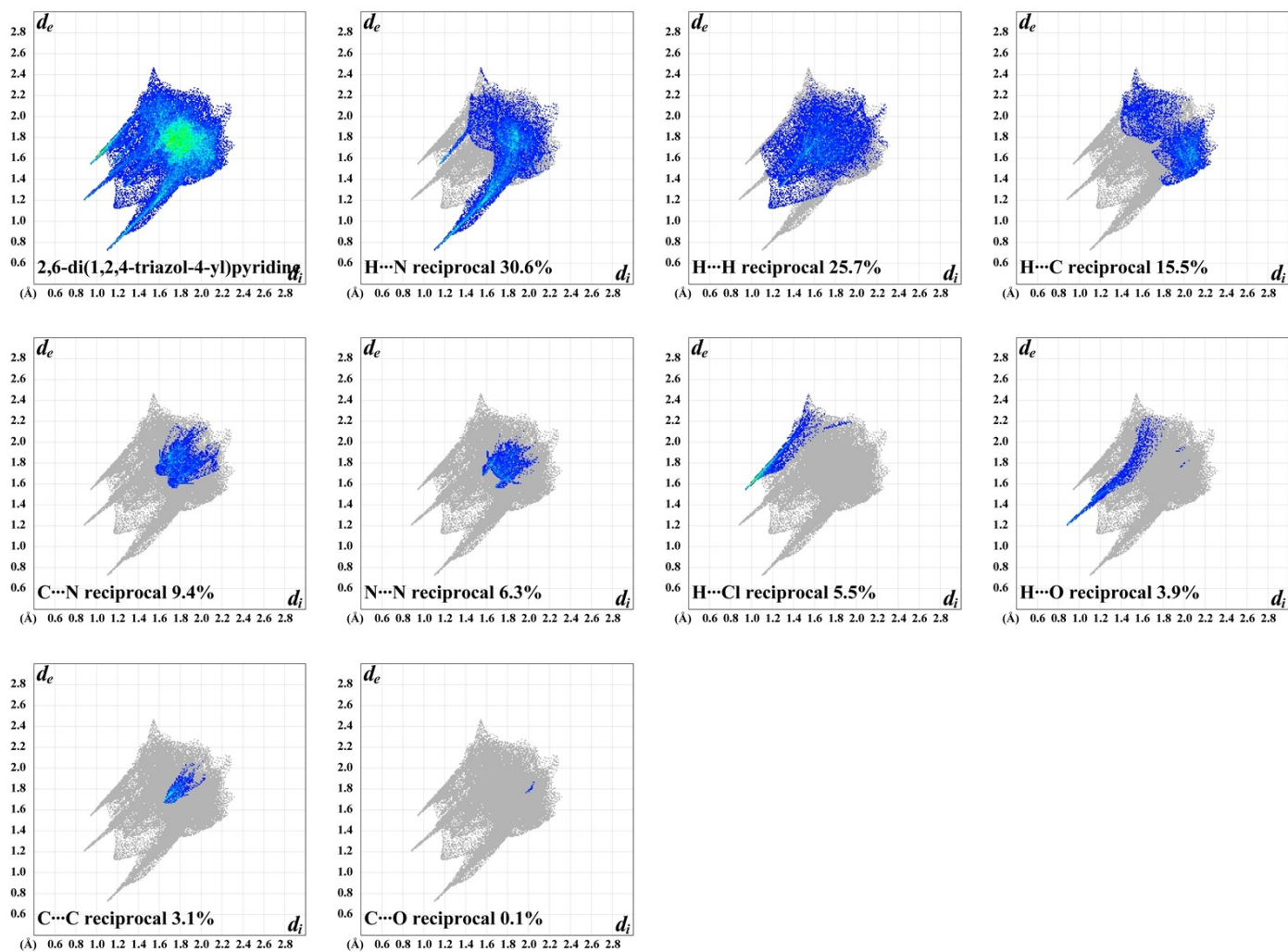


Fig. S2 2D fingerprint plots and decomposed 2D fingerprint plots of observed contacts for the 2,6-di(4H-1,2,4-triazol-4-yl)pyridine in the structure of **1**.

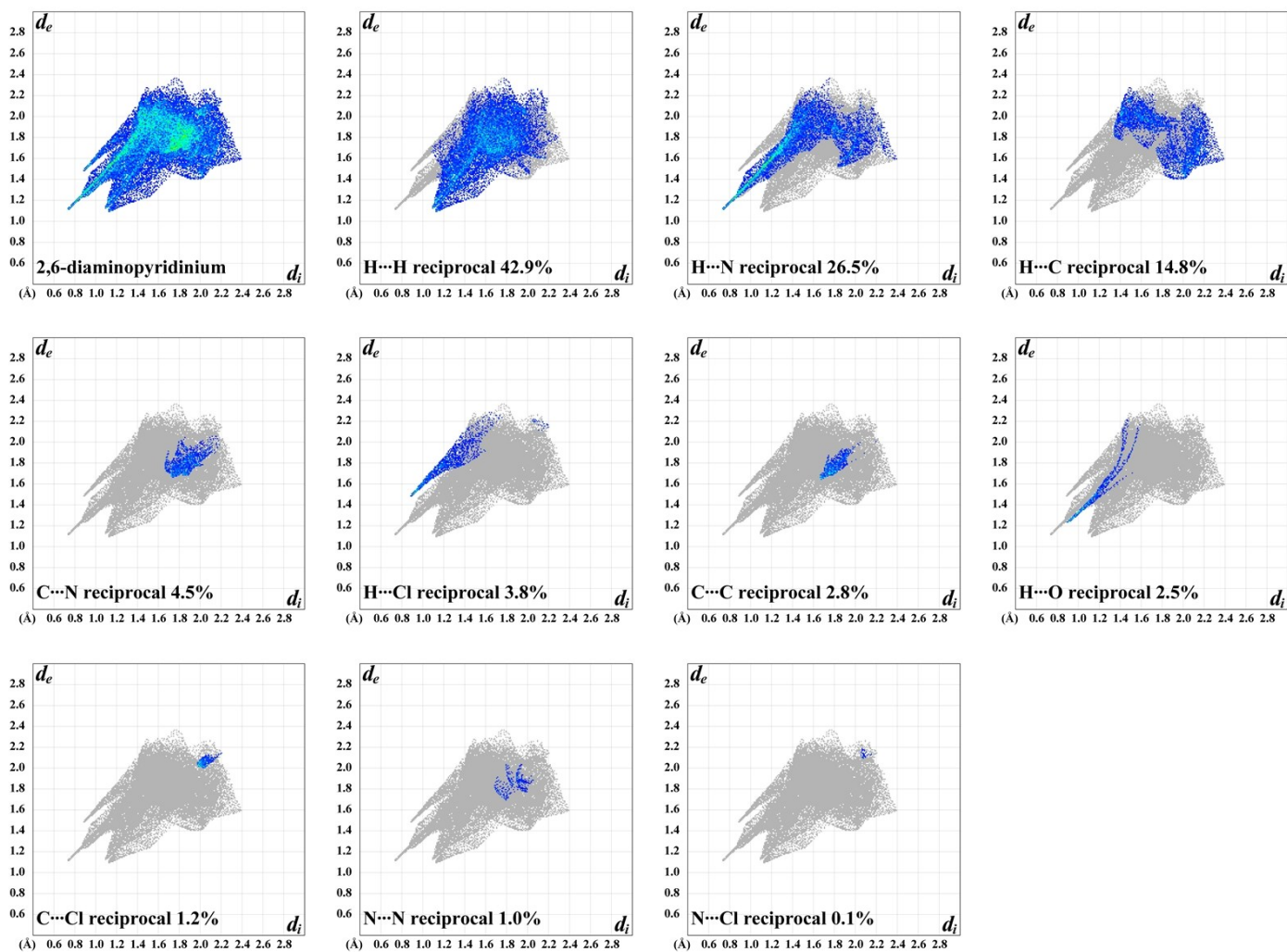


Fig. S3 2D fingerprint plots and decomposed 2D fingerprint plots of observed contacts for the 2,6-diaminopyridinium in the structure of **1**.

Table S1 Selected bond lengths (Å) and angles (°) for **1**

<i>Bond lengths</i>					
N(2)–C(3)	1.329(3)	N(13)–C(14)	1.365(3)	N(21)–C(22)	1.363(3)
N(2)–C(7)	1.332(3)	N(13)–C(17)	1.363(3)	N(21)–C(26)	1.365(3)
N(8)–C(3)	1.429(3)	N(15)–N(16)	1.393(3)	N(27)–C(22)	1.348(3)
N(8)–C(9)	1.367(3)	N(15)–C(14)	1.315(3)	N(28)–C(26)	1.338(3)
N(8)–C(12)	1.359(3)	N(16)–C(17)	1.307(3)	C(22)–C(23)	1.387(4)
N(10)–N(11)	1.388(3)	C(3)–C(4)	1.380(3)	C(23)–C(24)	1.375(4)
N(10)–C(9)	1.301(3)	C(4)–C(5)	1.382(3)	C(24)–C(25)	1.367(4)
N(11)–C(12)	1.311(3)	C(5)–C(6)	1.387(3)	C(25)–C(26)	1.391(3)
N(13)–C(7)	1.425(3)	C(6)–C(7)	1.369(3)		
<i>Bond angles</i>					
C(3)–N(2)–C(7)	116.76(19)	N(2)–C(3)–C(4)	124.5(2)	N(13)–C(17)–N(16)	110.1(2)
C(3)–N(8)–C(9)	127.0(2)	N(8)–C(3)–C(4)	122.0(2)	C(22)–N(21)–C(26)	122.9(2)
C(3)–N(8)–C(12)	128.90(19)	C(3)–C(4)–C(5)	116.9(2)	N(21)–C(22)–N(27)	117.1(2)
C(9)–N(8)–C(12)	104.08(19)	C(4)–C(5)–C(6)	120.2(2)	N(21)–C(22)–C(23)	119.0(2)
N(11)–N(10)–C(9)	107.4(2)	C(5)–C(6)–C(7)	117.4(2)	N(27)–C(22)–C(23)	123.9(2)
N(10)–N(11)–C(12)	106.5(2)	N(2)–C(7)–N(13)	113.03(18)	C(22)–C(23)–C(24)	118.3(2)
C(7)–N(13)–C(14)	129.91(18)	N(2)–C(7)–C(6)	124.3(2)	C(23)–C(24)–C(25)	122.7(2)
C(7)–N(13)–C(17)	125.41(18)	N(13)–C(7)–C(6)	122.6(2)	C(24)–C(25)–C(26)	118.6(2)
C(14)–N(13)–C(17)	104.68(18)	N(8)–C(9)–N(10)	110.8(2)	N(21)–C(26)–N(28)	117.9(2)
N(16)–N(15)–C(14)	105.80(18)	N(8)–C(12)–N(11)	111.2(2)	N(21)–C(26)–C(25)	118.6(2)
N(15)–N(16)–C(17)	108.21(19)	N(13)–C(14)–N(15)	111.21(19)	N(28)–C(26)–C(25)	123.5(2)
N(2)–C(3)–N(8)	113.52(19)				
<i>Torsion angles</i>					
C(7)–N(2)–C(3)–N(8)	–178.99(18)	N(10)–N(11)–C(12)–N(8)	0.1(3)	C(3)–C(4)–C(5)–C(6)	–0.6(4)
C(7)–N(2)–C(3)–C(4)	0.4(3)	C(14)–N(13)–C(7)–N(2)	172.2(2)	C(4)–C(5)–C(6)–C(7)	0.4(4)
C(3)–N(2)–C(7)–N(13)	179.49(18)	C(14)–N(13)–C(7)–C(6)	–7.7(4)	C(5)–C(6)–C(7)–N(2)	0.3(4)
C(3)–N(2)–C(7)–C(6)	–0.7(3)	C(17)–N(13)–C(7)–N(2)	–7.2(3)	C(5)–C(6)–C(7)–N(13)	–179.9(2)
C(9)–N(8)–C(3)–N(2)	3.2(3)	C(17)–N(13)–C(7)–C(6)	172.9(3)	C(26)–N(21)–C(22)–N(27)	179.7(3)
C(9)–N(8)–C(3)–C(4)	–176.3(3)	C(7)–N(13)–C(14)–N(15)	–179.6(2)	C(26)–N(21)–C(22)–C(23)	0.0(4)
C(12)–N(8)–C(3)–N(2)	–174.0(2)	C(17)–N(13)–C(14)–N(15)	–0.1(3)	C(22)–N(21)–C(26)–N(28)	178.9(2)
C(12)–N(8)–C(3)–C(4)	6.6(4)	C(7)–N(13)–C(17)–N(16)	179.2(2)	C(22)–N(21)–C(26)–C(25)	–0.3(4)
C(3)–N(8)–C(9)–N(10)	–178.7(2)	C(14)–N(13)–C(17)–N(16)	–0.3(3)	N(21)–C(22)–C(23)–C(24)	0.1(4)
C(12)–N(8)–C(9)–N(10)	–0.9(3)	C(14)–N(15)–N(16)–C(17)	–0.7(3)	N(27)–C(22)–C(23)–C(24)	–179.6(3)
C(3)–N(8)–C(12)–N(11)	178.2(2)	N(16)–N(15)–C(14)–N(13)	0.5(3)	C(22)–C(23)–C(24)–C(25)	0.1(5)
C(9)–N(8)–C(12)–N(11)	0.5(3)	N(15)–N(16)–C(17)–N(13)	0.6(3)	C(23)–C(24)–C(25)–C(26)	–0.4(5)
C(9)–N(10)–N(11)–C(12)	–0.7(3)	N(2)–C(3)–C(4)–C(5)	0.2(4)	C(24)–C(25)–C(26)–N(21)	0.4(4)
N(11)–N(10)–C(9)–N(8)	1.0(3)	N(8)–C(3)–C(4)–C(5)	179.6(2)	C(24)–C(25)–C(26)–N(28)	–178.7(3)