

*Supporting Information for*

**A Novel Two-photon Fluorescent Probe for Hydrogen Sulfide in Living Cells using an Acedan–NBD amine Dyad Based on FRET Process with High Selectivity and Sensitivity**

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## **Additional spectra**

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*Measurement of two-photon cross section.* The two-photon cross section ( $\sigma$ ) was determined by using a femtosecond (fs) fluorescence measurement technique. **L** was dissolved in EtOH-phosphate buffer solution (PBS, 25 mM, pH 7.4, containing 20% EtOH as co-solvent), at a concentration of  $5.0 \times 10^{-6}$  M and then the two-photon fluorescence intensity was measured at 720-800 nm by using Rhodamine B in MeOH as the standard, whose two-photon property has been well characterized in the literature.<sup>[1]</sup> The two-photon cross-section was calculated by using  $\sigma = \sigma_r(F_t n_t^2 \Phi_r C_r) / (F_r n_r^2 \Phi_t C_s)$ , where the subscripts t and r stand for the sample and reference molecules. F is the average fluorescence intensity integrated from TPE imaging, n is the refractive index of the solvent, C is the concentration,  $\Phi$  is the quantum yield, and  $\sigma_r$  is the two-photon cross-section of the reference molecule.

[1] N. S. Makarov, M. Drobizhev, A. Rebane, *Opt. Exp.* **2008**, 16, 4029-4047.

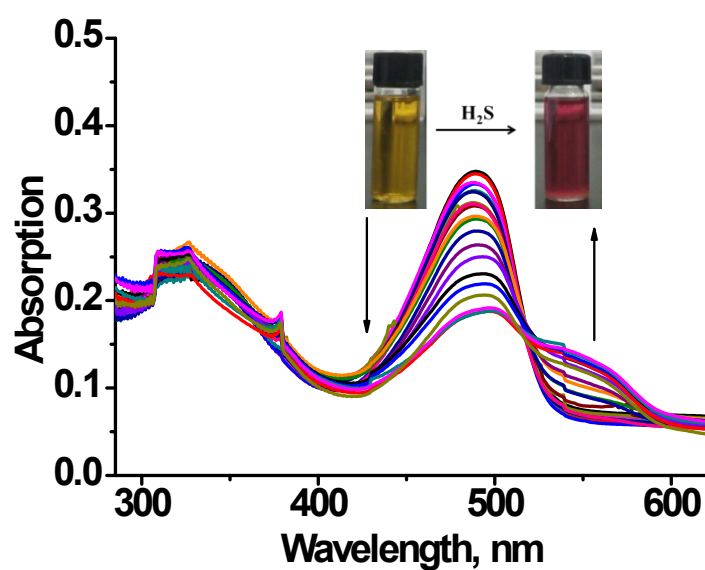


Fig. S1 Absorption spectra of **L** (5  $\mu\text{M}$ ) with  $\text{H}_2\text{S}$  (0-140.0 equiv.) in EtOH-phosphate buffer solution (PBS, 25 mM, pH 7.4, containing 20% EtOH as co-solvent) at 25  $^\circ\text{C}$ . Inset: visual color of the probe **L** in the absence (yellow) or presence (pink) of  $\text{H}_2\text{S}$  under visible light.

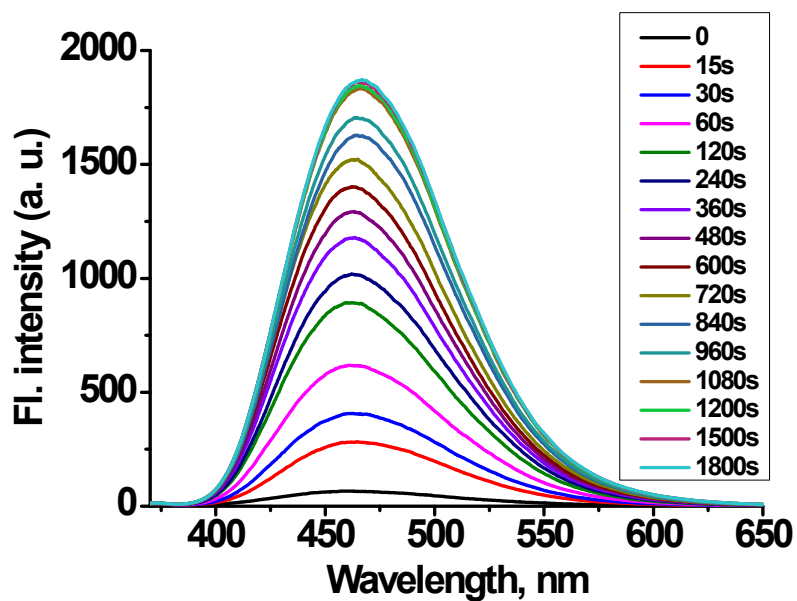


Fig. S2 Fluorescence spectra of probe **L** (5  $\mu\text{M}$ ) upon reaction with  $\text{H}_2\text{S}$  (700  $\mu\text{M}$ ) at 25  $^\circ\text{C}$  at different reaction times as shown inset.

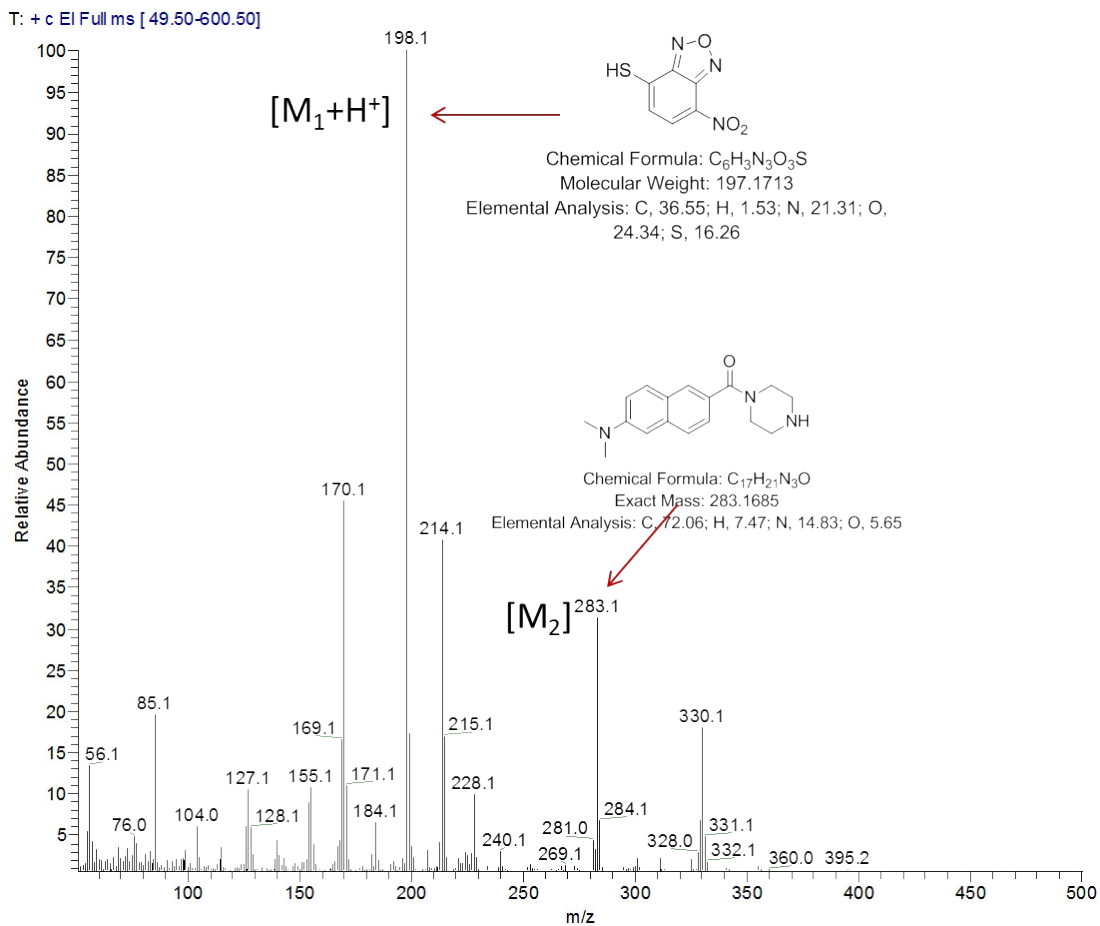


Fig. S3 EI-MS spectrum of the reaction of probe **L** with  $H_2S$ .

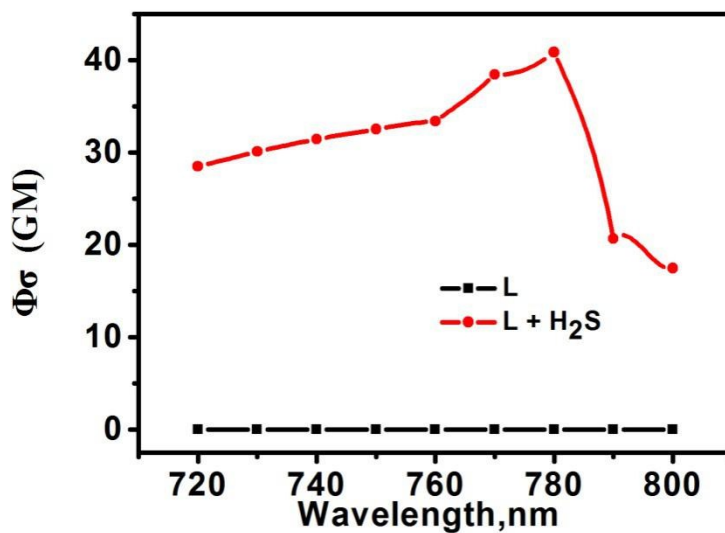


Fig. S4. Two-photon action spectra of Probe **L** (5  $\mu M$ ) and **L** with 250  $\mu M$   $H_2S$  in EtOH-phosphate buffer solution (PBS, 25 mM, pH 7.4, containing 20% EtOH as co-solvent).

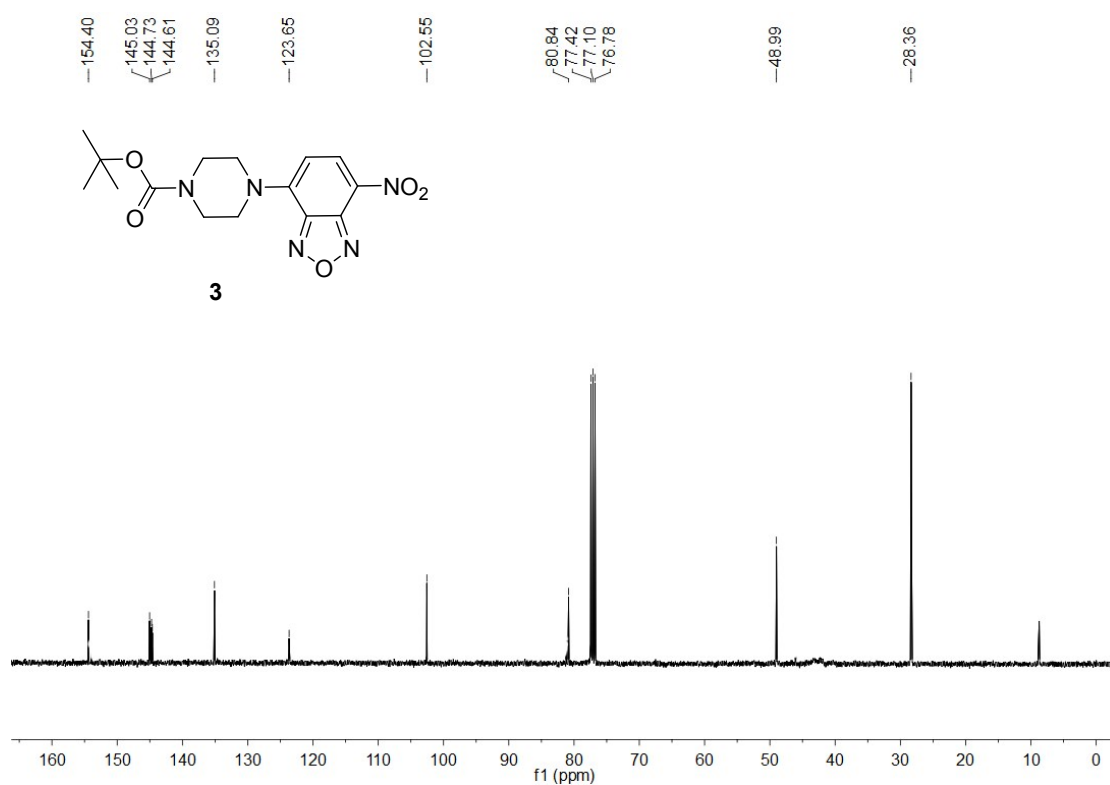
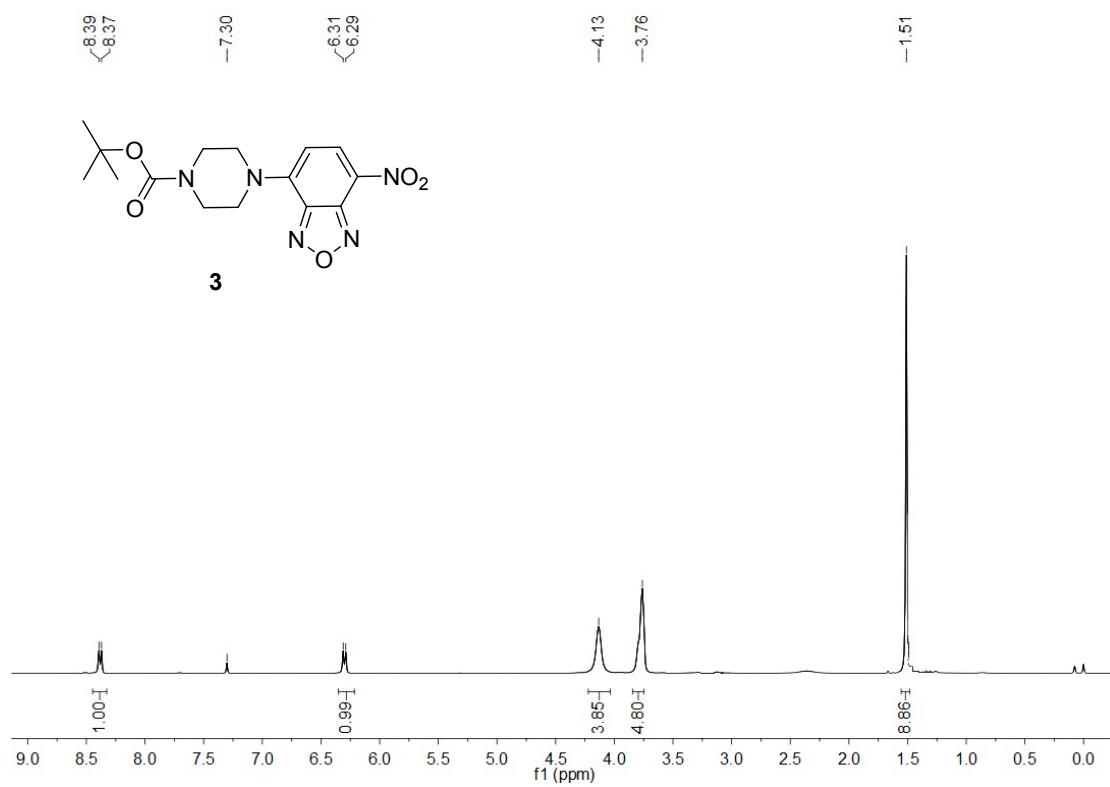


Fig. S5 The  $^1\text{H}$  NMR and  $^{13}\text{C}$  NMR spectrometry of **3**.

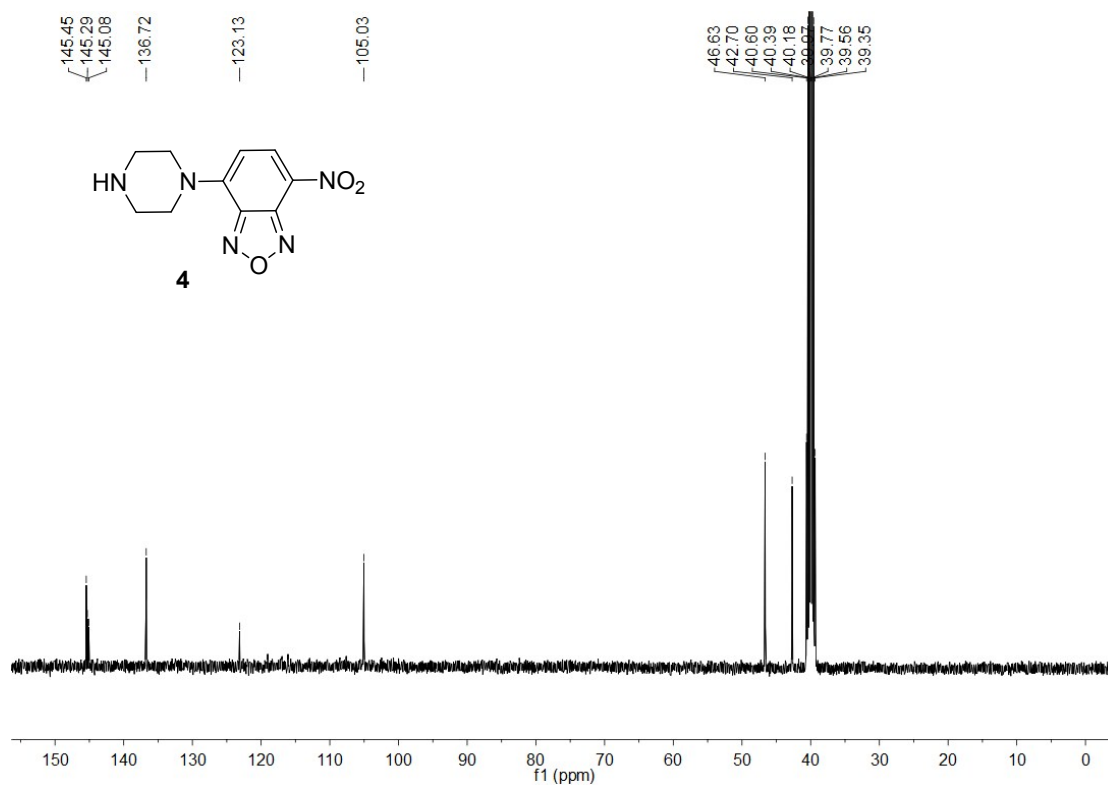
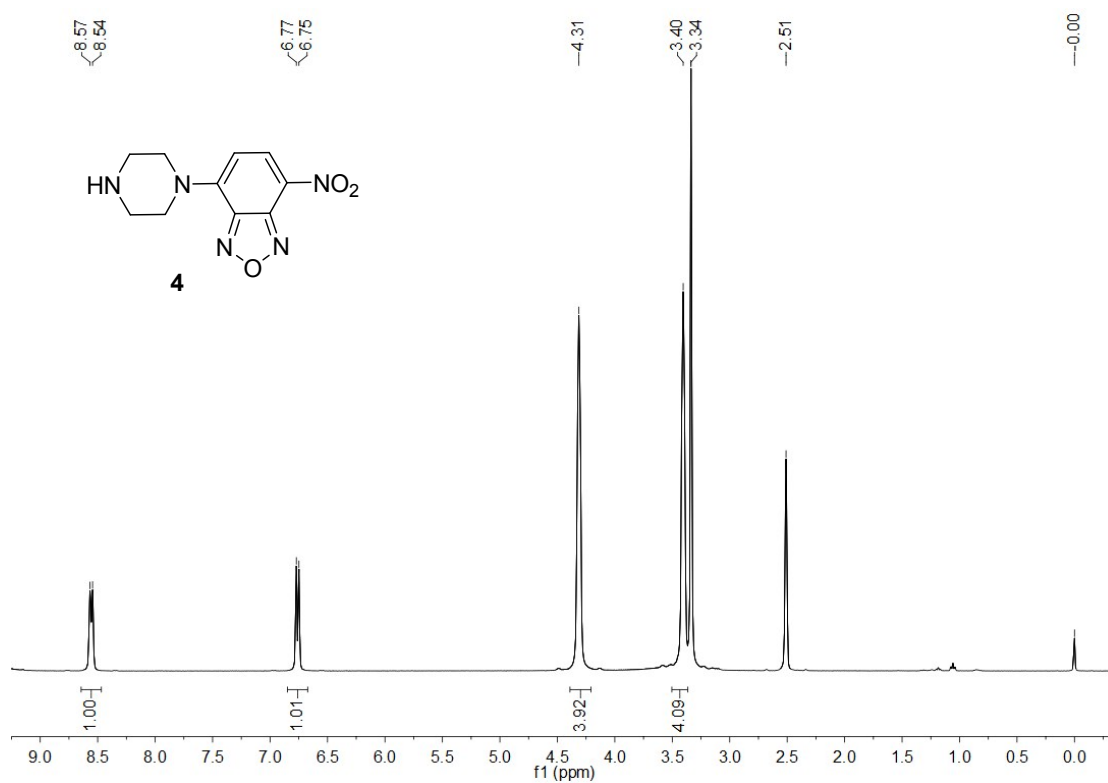


Fig. S6 The  $^1\text{H}$  NMR and  $^{13}\text{C}$  NMR spectrometry of **4**.

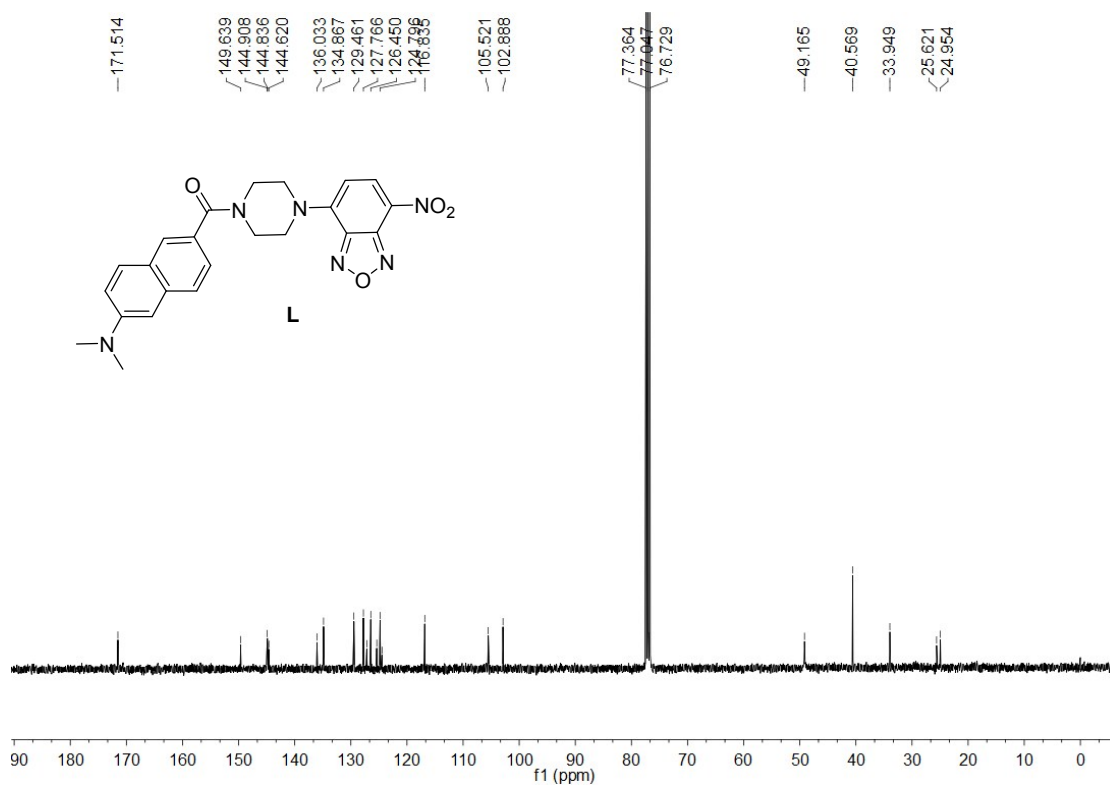
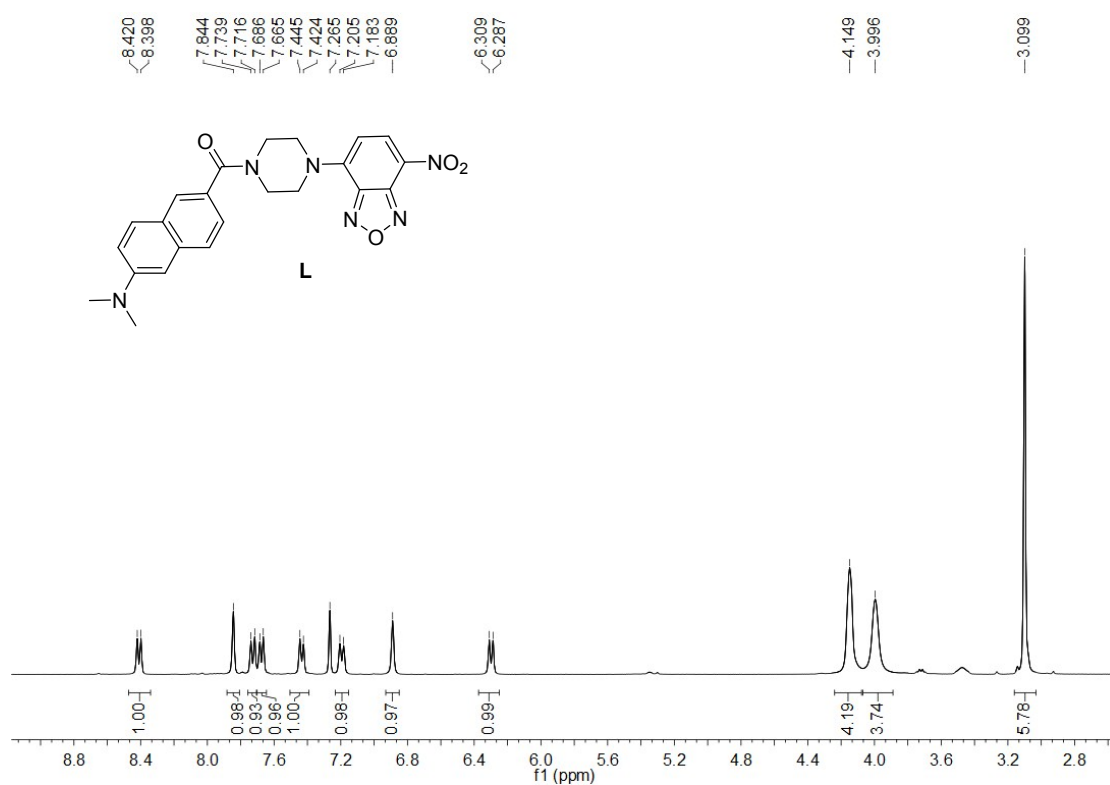
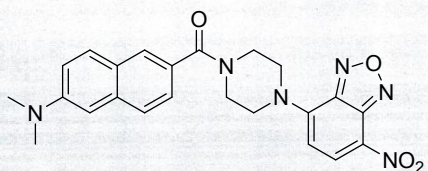


Fig. S7 The  $^1\text{H}$  NMR and  $^{13}\text{C}$  NMR spectrometry of probe **L**.



File : C:\xcalibur\data\ty-170512-446-hr-av2.RAW  
 Full ms [410.500 - 467.500 ] - Range: 410.500 - 467.500  
 Scan No. 1 of 1

Mass	Relative Intensity	Theoretical Mass	Delta [ppm]	Delta [mmu]	Composition
446.171	100.0	446.1697	3.9	1.7	C <sub>23</sub> H <sub>22</sub> O <sub>4</sub> N <sub>6</sub>
		446.1737	-5.1	-2.3	C <sub>28</sub> H <sub>27</sub> O <sub>2</sub> N <sub>4</sub>
		446.1751	-8.1	-3.6	C <sub>30</sub> H <sub>24</sub> O <sub>3</sub> N <sub>1</sub>
		446.1665	11.0	4.9	C <sub>34</sub> H <sub>22</sub> O <sub>1</sub>
		446.1652	14.0	6.3	C <sub>32</sub> H <sub>20</sub> N <sub>3</sub>



Chemical Formula: C<sub>23</sub>H<sub>22</sub>N<sub>6</sub>O<sub>4</sub>  
 Exact Mass: 446.1703  
 Elemental Analysis: C, 61.87; H, 4.97; N, 18.82; O, 14.33

Fig. S8 The The HREI mass spectrometry of probe L.