

Electronic Supplementary Information for

Electrochemical growth of $\text{Co}(\text{OH})_2$ nanoflakes on Ni foam for methanol electro-oxidation

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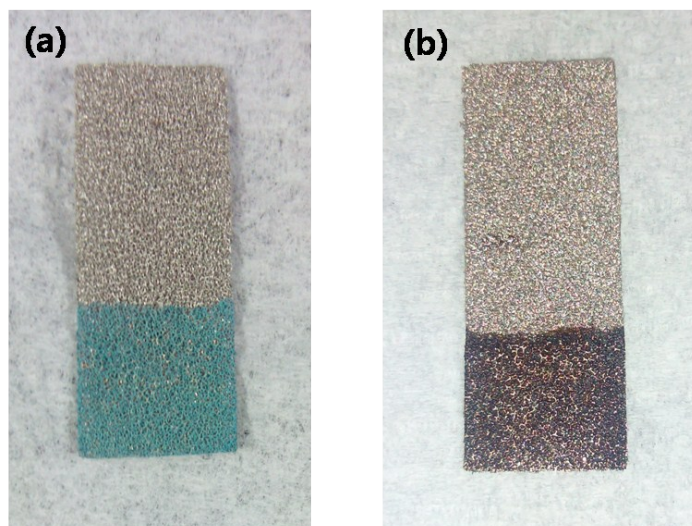


Figure S1: Photographs of the as electrodeposited (a) $\text{Co}(\text{OH})_2$ and (b) Co_3O_4 films on Ni foam

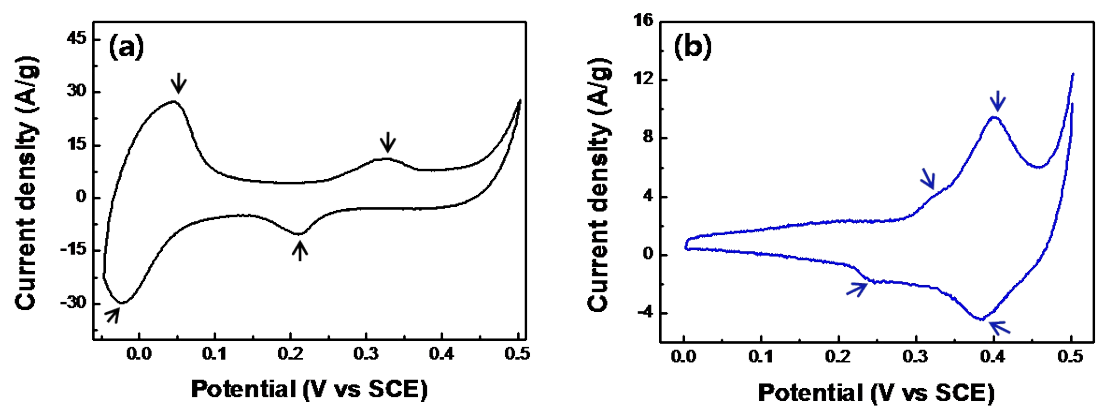


Figure S2: CV of the (a) Co(OH)_2 and (b) Co_3O_4 electrodes at the scan rate of 10 mV/s in 1M KOH solution

Table S1: Comparison of different electrocatalysts for methanol oxidation application

No.	Catalyst	Method	Electrolyte (M)	Potential range (V)	Scan rate (mV/s)	Onset potential (V)	Current density	Cycling stability	Reference
1	NiCo ₂ O ₄	Electrodeposition	1M KOH+ 0.5M Methanol	0-0.6	10	0.42	52 A/g	82% retention after 1000 cycles	4
2	Cu(OH) ₂	In-situ surface chemical oxidation	1M KOH+ 0.5M Methanol	0-0.6	10	0.40	55 A/g	NA	5
3	Co ₃ O ₄ /NiO core/shell nanowire	Hydrothermal+ Electrodeposition	1M KOH+ 0.5M Methanol	0.1-0.75	25	0.45	145 mA/cm ²	Stable after 500 cycles	15
4	MnCo ₂ O ₄	Co-precipitation	1M KOH+ 0.5M Methanol	0-0.5	10	NA	79.9 A/g	80% retention after 500 cycles	17
5	ZnCo ₂ O ₄	Hydrothermal	1M KOH+ 0.5M Methanol	0-0.8	10	0.50	105 A/g	65% retention after 1000 cycles	44
6	Co(OH) ₂	Electrodeposition	1M KOH+ 0.5M Methanol	0-0.5	10	0.27	150 A/g	82% retention after 500 cycles	Present work