

Electrospun CuO/NiO composite nanofibers for sensitive and selective non-enzymatic nitrite sensors

Jayachandran Saravanan,^{a,†} Ramachandran Ramasamy,^{b,†} Helen Annal Therese^{*c},

G. Amala,^a and G. Gnana kumar^{*a}

^a *Department of Physical Chemistry, School of Chemistry, Madurai Kamaraj University,
Madurai-625-021, Tamil Nadu, India.*

^b *Department of Physics and Nanotechnology, SRM University, Kattankulathur-603 203, Tamil Nadu, India*

^c *Nanotechnology Research Centre, SRM University, Kattankulathur-603-203, Chennai, Tamil Nadu, India*

Corresponding authors :

*G.Gnana kumar:e-mail:kumarg2006@gmail.com

*Helen Annal Therese: e-mail: *helant03@gmail.com*

[†] Both the authors equally contributed

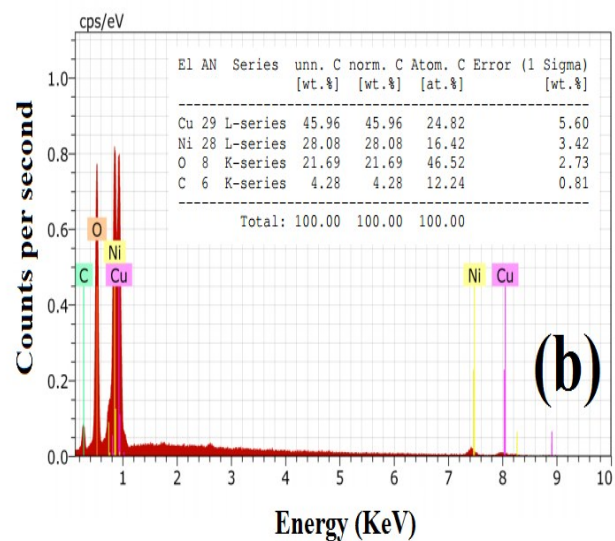
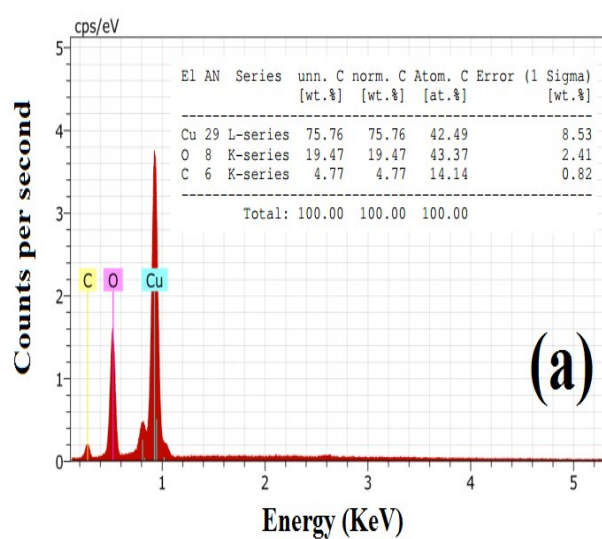


Fig. S1 EDAX patterns of (a) CuO and (b) CuO-NiO nanofibers.

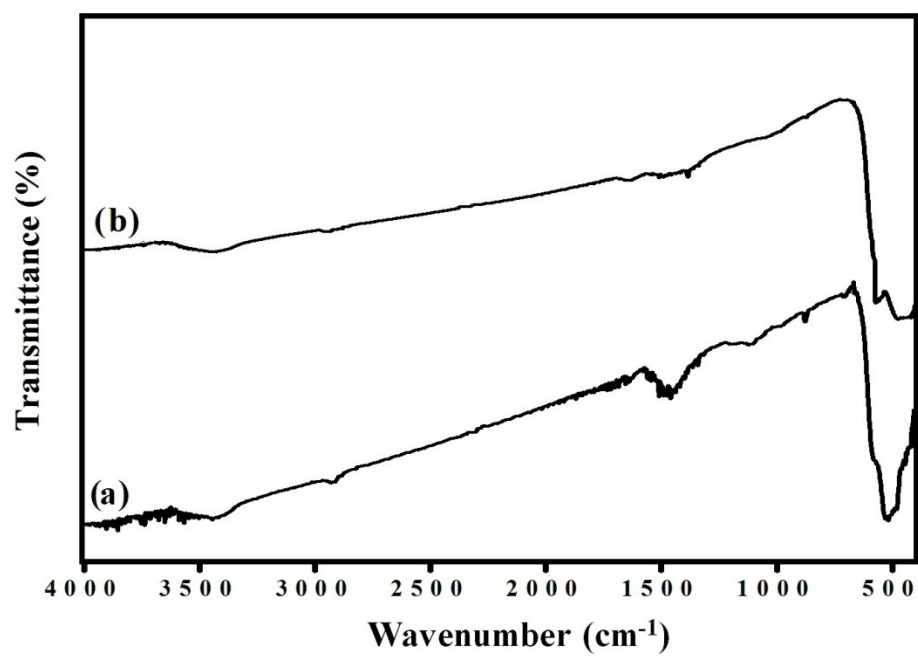


Fig. S2 FT-IR spectra of (a) CuO and (b) CuO-NiO nanofibers.

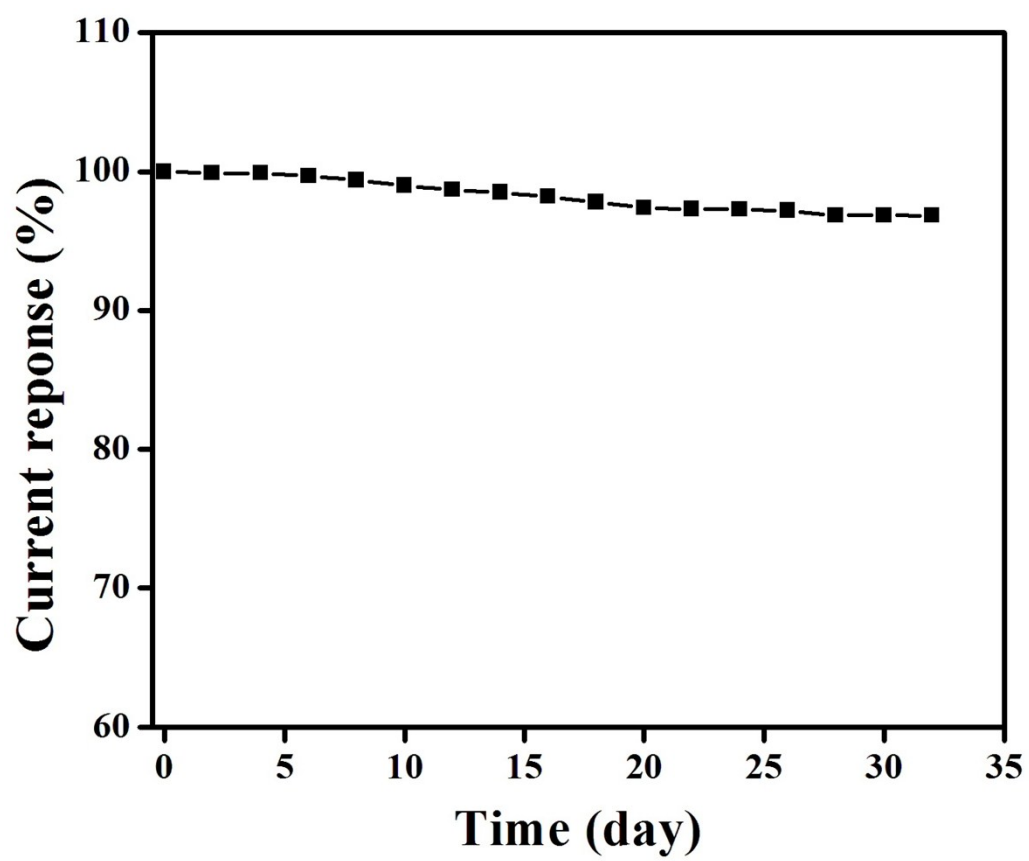


Fig. S3 The stability profile of CuO/NiO/GCE in 1 mM NO_2^- in 0.1 M PBS solution.

Table S1. Comparison of the electroanalytical performance of non-enzymatic nitrite sensors.

Electrode Materials	Linear range (mM)	LOD ^a (μM)	Sensitivity (μA mM ⁻¹ cm ⁻²)	References
Au NPs ^b /CLDH ^c /GCE ^d	0.001-0.191	0.5	382.2 μA mM ⁻¹	1
CuO-CPE ^e	0.1-1.25	0.6	-	2
Cu-NDs ^f /rGO ^g /GCE ^d	0.00125-13.0	0.4	214.0	3
CuO/GCE ^d	0.1-3.0	28.0	-	4
MCPE ^h	0.1-2.15	1.19	-	5
MOF ⁱ -525 thin film	0.02-0.8	2.1	95.0	6
CR ^j -GO ^k /GCE ^d	0.0089-0.167	1.0	0.0267 A M ⁻¹	7
Ag MCS ^l -PAA ^m /PVA ⁿ /SPCE ^o	0.002-0.8	4.5	474.14	8
Ag-PAMAM ^p /GCE ^d	0.004-1.44	0.4	265.0	9
MoS ₂ -MWCNTs ^q -Au/GCE ^d	0.012-2.1	4.0	1734.0	10
TOAB ^r /ZnPP-C60/GCE ^d	0.002-0.164	1.44	249.9 μA mM ⁻¹	11
f-ZnO@rFGO ^s /GCE ^d	0.01-8.0	33.0	-	12
Mn ₂ O ₃ /GCE ^d	0.005-3.5	1.0	326.0	13
PAOA ^t /GCE ^d	0.005-2.0	2.0	0.0643 μA μM ⁻¹ cm ⁻²	14
PEDOT ^u -HMF ^v /FTO ^w	0.05-7.5	0.59	255.2	15
CuO/NiO/GCE^d	0.001-5.0	0.5	282.72	This work

^alimit of detection. ^bnanoparticles. ^ccopper calcined layered double hydroxide. ^dglassy carbon electrode. ^ecarbon powder electrode. ^fnanodendrites. ^greduced graphene oxide. ^hcobalt hexa cyanoferrate-modified carbon paste electrode. ⁱmetal-organic framework. ^jchemically reduced. ^kgraphene oxide. ^lmicrocubics. ^mpoly (acrylic acid). ⁿpoly (vinyl alcohol). ^oscreen printed carbon electrode. ^ppolyamidoamine. ^qmulti-walled carbon nanotubes. ^rtetraoctylammonium bromide. ^sreduced functionalized graphene oxide. ^tpoly(aniline-co-o-aminophenol). ^upoly(3,4-ethylenedioxythiophene). ^vhollow microflowers. ^wfluorine-doped tin oxide.

Notes and References

- 1 L. Cui, X. Meng, M. Xu, K. Shang, S. Ai and Y. Liu, *Electrochim. Acta*, 2011, **56**, 9769-9774.
- 2 B. Sljukic, C. E. Banks, A. Crossley and R. G. Compton, *Electroanalysis*, 2007, **19**, 79-84.
- 3 D. Zhang, Y. Fang, Z. Miao, M. Ma, X. Du, S. Takahashi, J. Anzai and Q. Chen, *Electrochim. Acta*, 2013, **107**, 656-663.
- 4 Y. Zhao, X. Song, X. Song, Z. Yin and Q. Song, *J. Colloid Interface Sci.*, 2013, **396**, 29-38.
- 5 H. Heli, I. Eskandari, N. Sattarahmady and A. A. Moosavi-Movahedi, *Electrochim. Acta*, 2012, **77**, 294-301.
- 6 C. Kung, T. Chang, L. Chou, J. T. Hupp, O. K. Farha and K. Ho, *Electrochem. Commun.*, 2015, **58**, 51-56.
- 7 V. Mani, A. P. Periasamy and S. Chen, *Electrochem. Commun.*, 2012, **17**, 75-78.
- 8 K. Promsuwan, P. Thavarungkul, P. Kanatharana and W. Limbut, *Electrochim. Acta*, 2017, **232**, 357-369.
- 9 D. Ning, H. Zhang and J. Zheng, *J. Electroanal. Chem.*, 2014, **717-718**, 29-33.
- 10 Y. Zhang, F. Wen, J. Tan, C. Jiang, M. Zhu, Y. Chen and H. Wang, *J. Electroanal. Chem.*, 2017, **786**, 43-49.
- 11 H. Wu, S. Fan, X. Jin, H. Zhang, H. Chen, Z. Dai and X. Zou, *Anal. Chem.*, 2014, **86**, 6285-6290.
- 12 A. R. Marlinda, A. Pandikumar, N. Yusoff, N. M. Huang and H. N. Lim, *Microchim. Acta*, 2015, **182**, 1113-1122.
- 13 H. Lu, Y. Zhang and P. Liu, *J. Appl. Electrochem.*, 2016, **46**, 1059-1065.
- 14 L. Liu, H. Cui, H. An, J. Zhai and Y. Pan, *Ionics*, 2017, **23**, 1517-1523.
- 15 Y. Cheng, C. Kung, L. Chou, R. Vittal and K. Ho, *Sens. Actuator B-Chem.*, 2014, **192**, 762-768.