

Supporting Information

Incorporation of CuO NPs into the modified UiO-66-NH₂ metal-organic frameworks (MOFs) with melamine for catalytic C–O coupling in the Ullmann condensation

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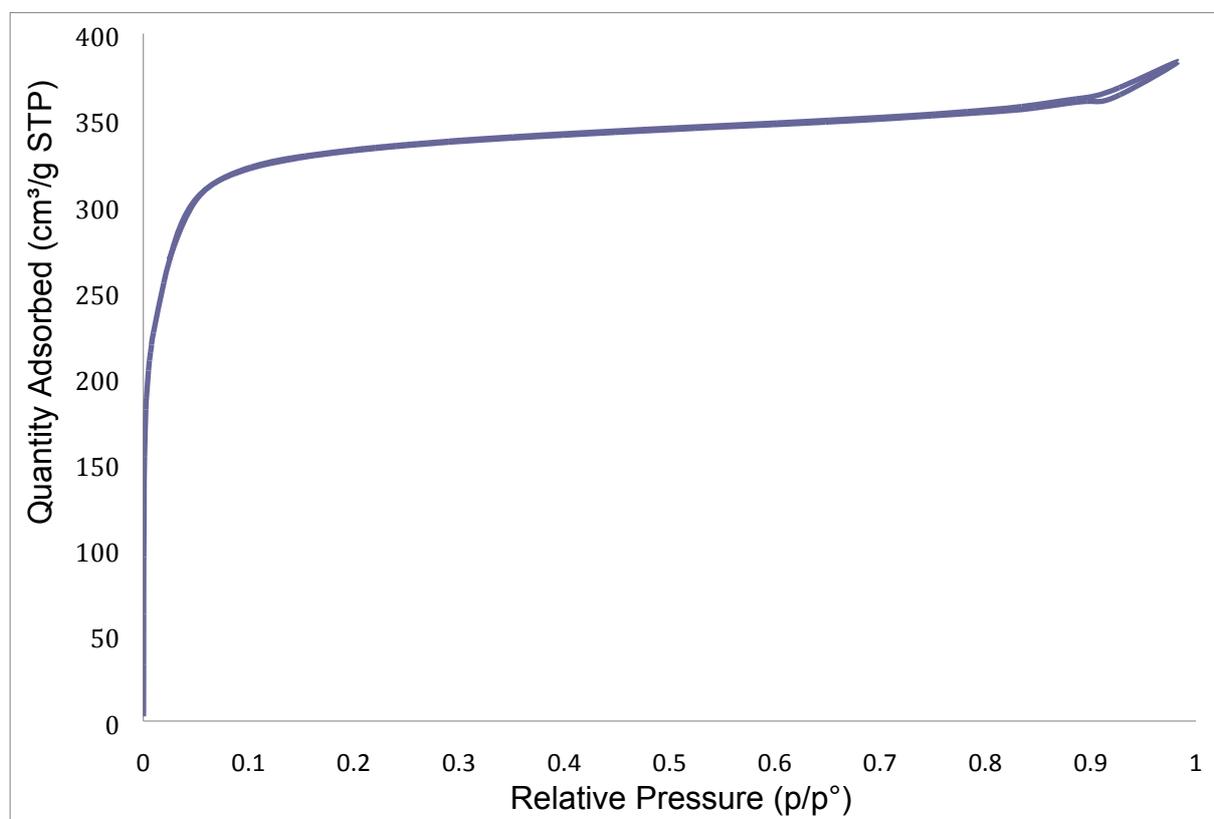


Figure S1. N₂ adsorption-desorption isotherm of UiO-66-NH₂.

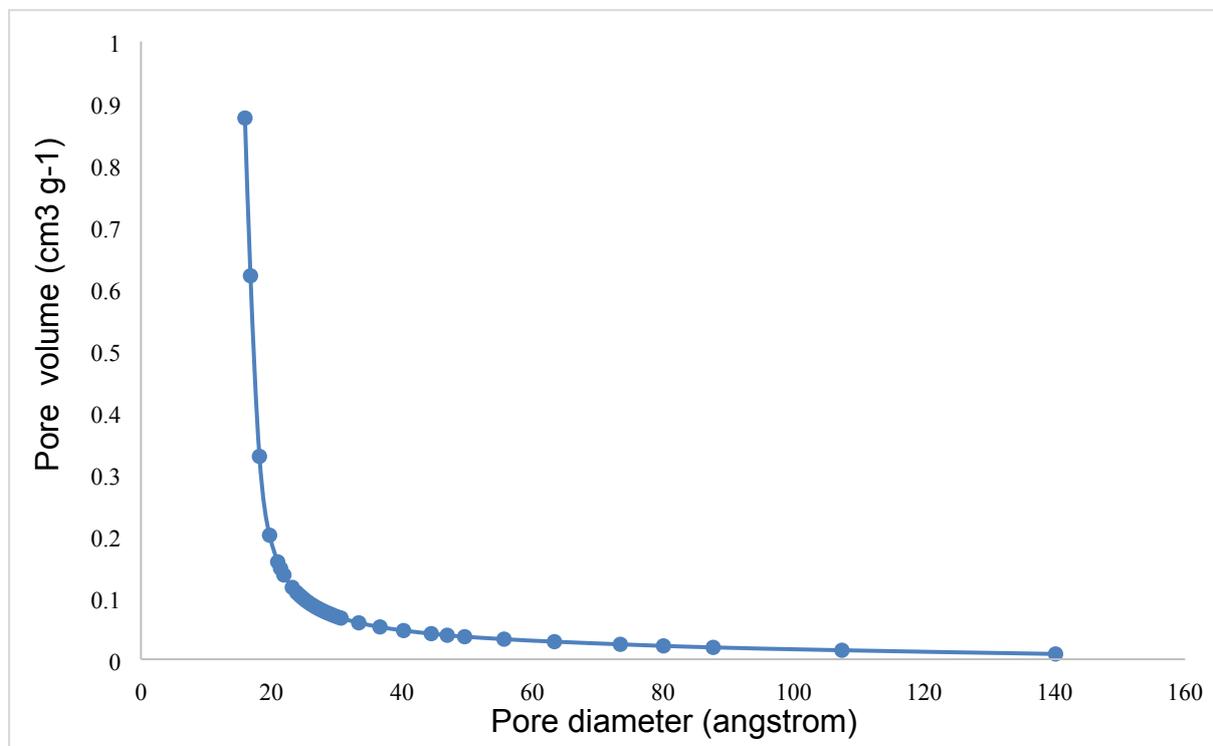


Figure S2. Pore volume distribution of UiO-66-NH₂.

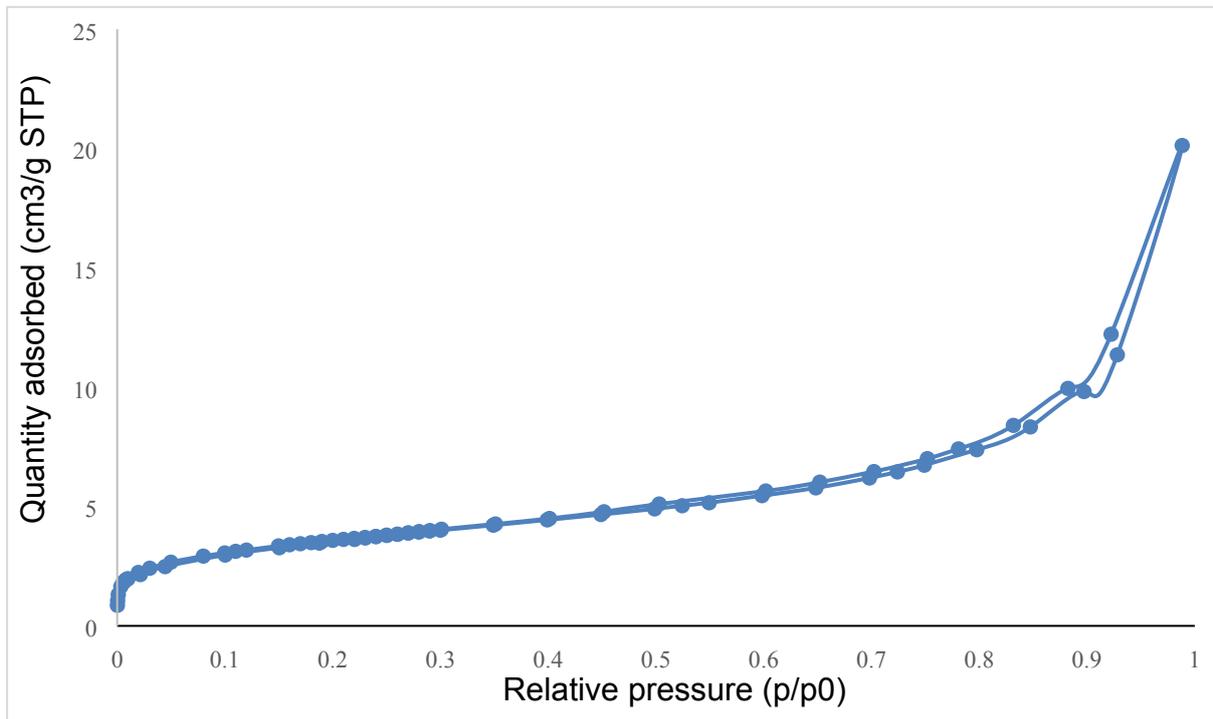


Figure S3. N₂ adsorption-desorption isotherm for reused UiO-66-NH₂-MIm/CuO.

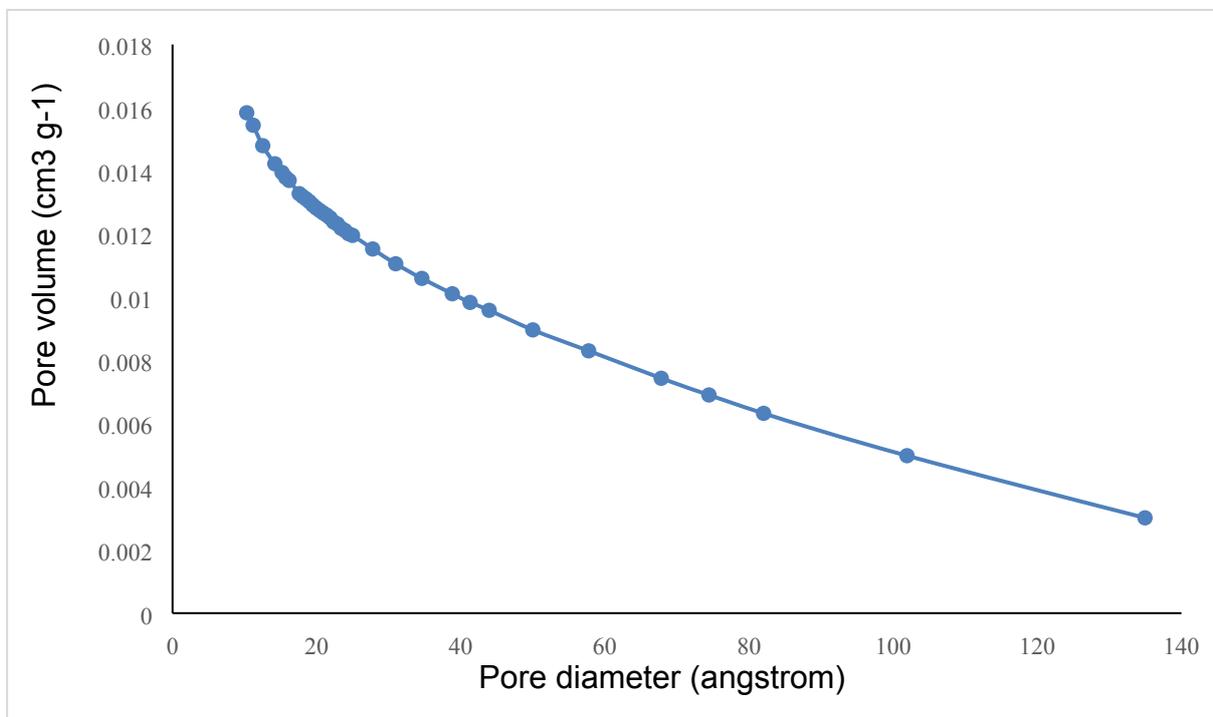


Figure S4. Pore volume distribution for reused UiO-66-NH₂-MIm/CuO.

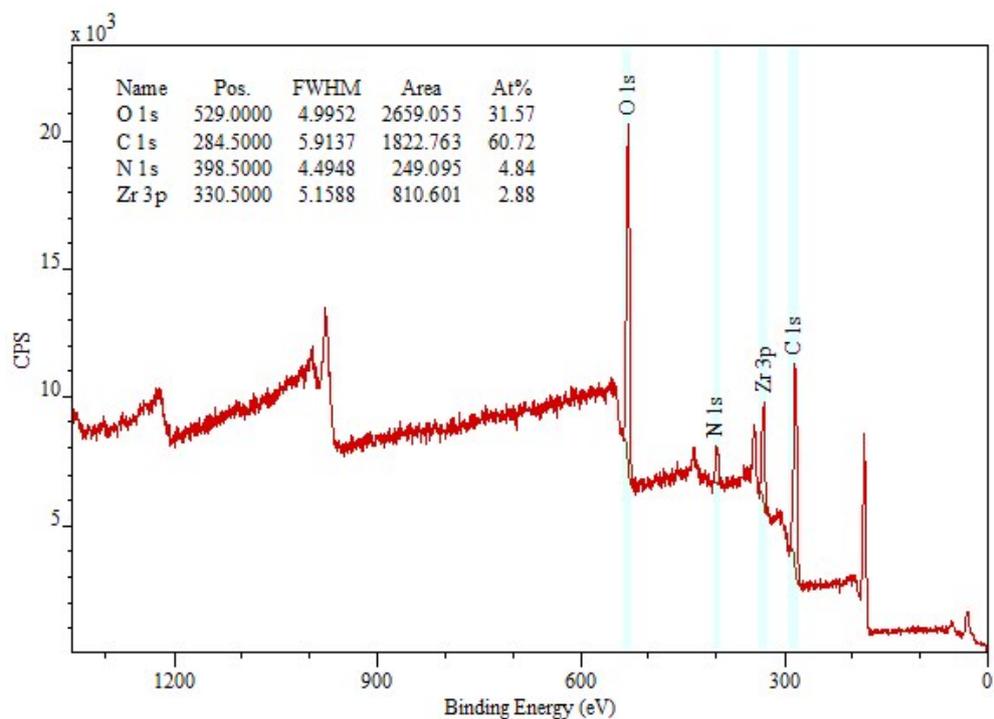


Figure S5. XPS survey of UiO-66-NH₂.

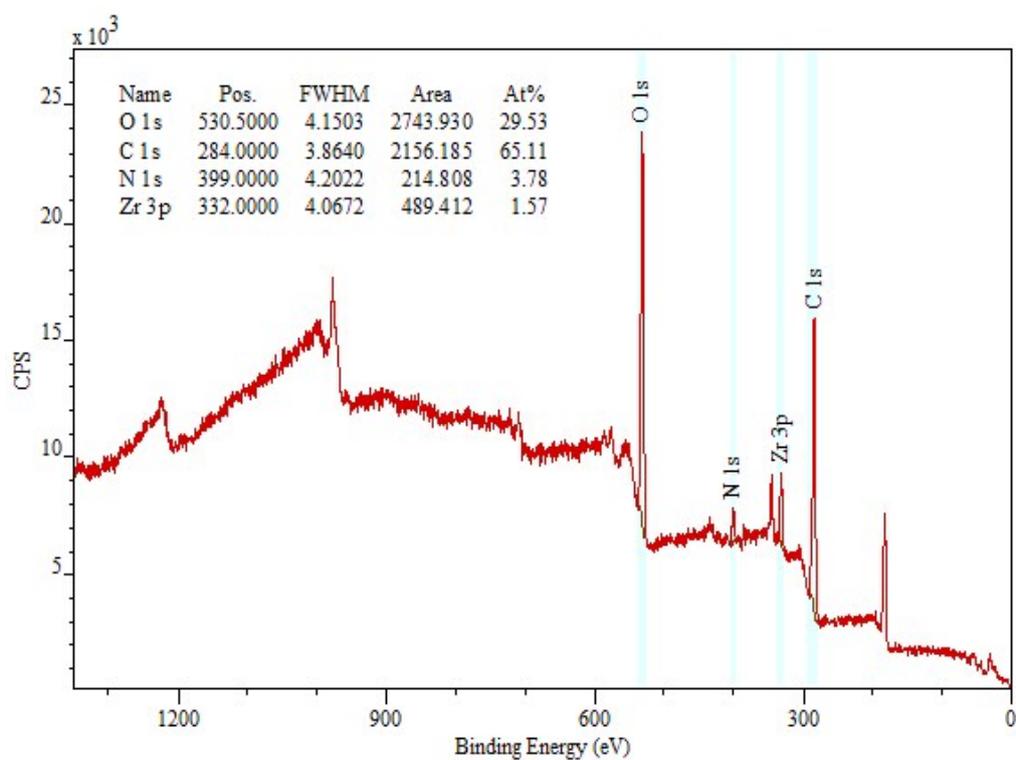


Figure S6. XPS survey of UiO-66-NH₂-Mlm.

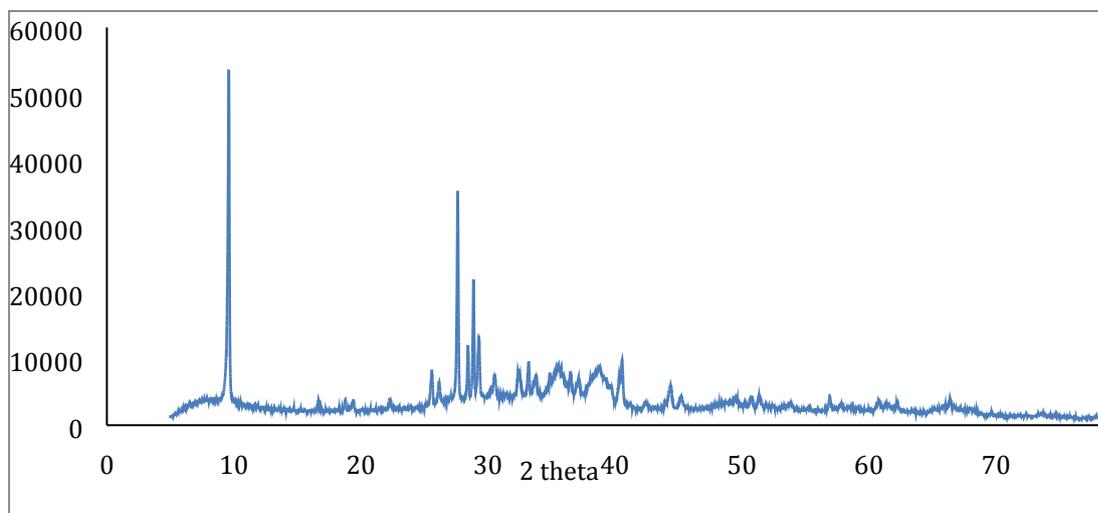


Figure S7. XRD patterns of reused UiO-66-NH₂-Mlm/CuO.

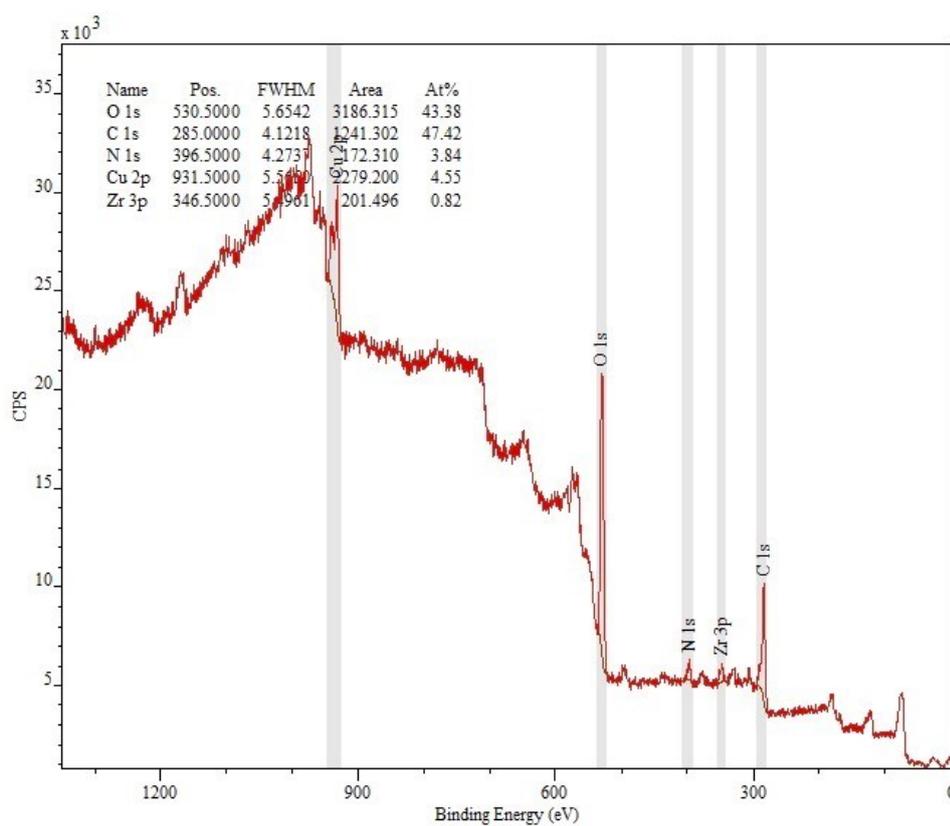
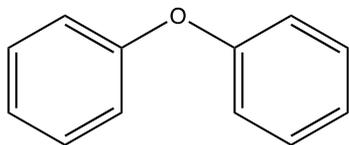
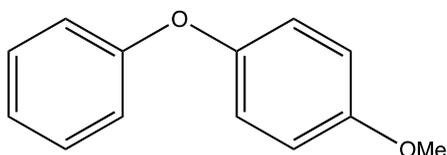


Figure S8. XPS survey of reused UiO-66-NH₂-Mlm/CuO.

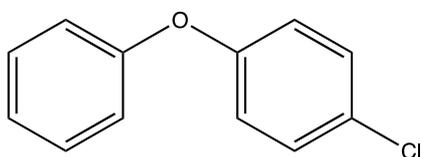
NMR data of the products:



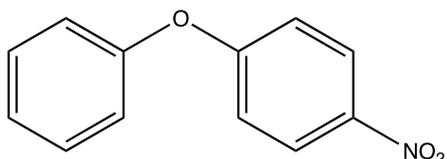
Oxydibenzene: ^1H NMR (400 MHz, CDCl_3): $\delta = 7.37$ (t, 4 H), 7.14 (t, 2H), 7.05 (d, 4 H). ^{13}C NMR (100 MHz, CDCl_3): $\delta = 157.2, 129.7, 123.2, 118.9$.



1-Methoxy-4-phenoxybenzene: ^1H NMR (400 MHz, CDCl_3): $\delta = 7.32$ -7.25 (m, 4 H), 7.11-6.90 (m, 5 H), 3.80 (s, 3 H). ^{13}C NMR (100 MHz, CDCl_3): $\delta = 157.2, 155.4, 151.1, 128.5, 121.5, 119.8, 118.8, 114.3, 55.8$.



1-Chloro-4-phenoxybenzene: ^1H NMR (400 MHz, CDCl_3): $\delta = 7.52$ -7.40 (m, 4 H), 7.30-6.99 (m, 5 H). ^{13}C NMR (100 MHz, CDCl_3): $\delta = 157.0, 155.5, 129.6, 128.8, 128.2, 122.7, 119.7, 118.9$.



1-Nitro-4-phenoxybenzene: ^1H NMR (400 MHz, CDCl_3): $\delta = 8.21$ -7.41 (m, 4 H), 7.38-7.06 (m, 5 H). ^{13}C NMR (100 MHz, CDCl_3): $\delta = 157.4, 149.8, 138.6, 128.3, 124.4, 122.8, 118.9, 116.5$.

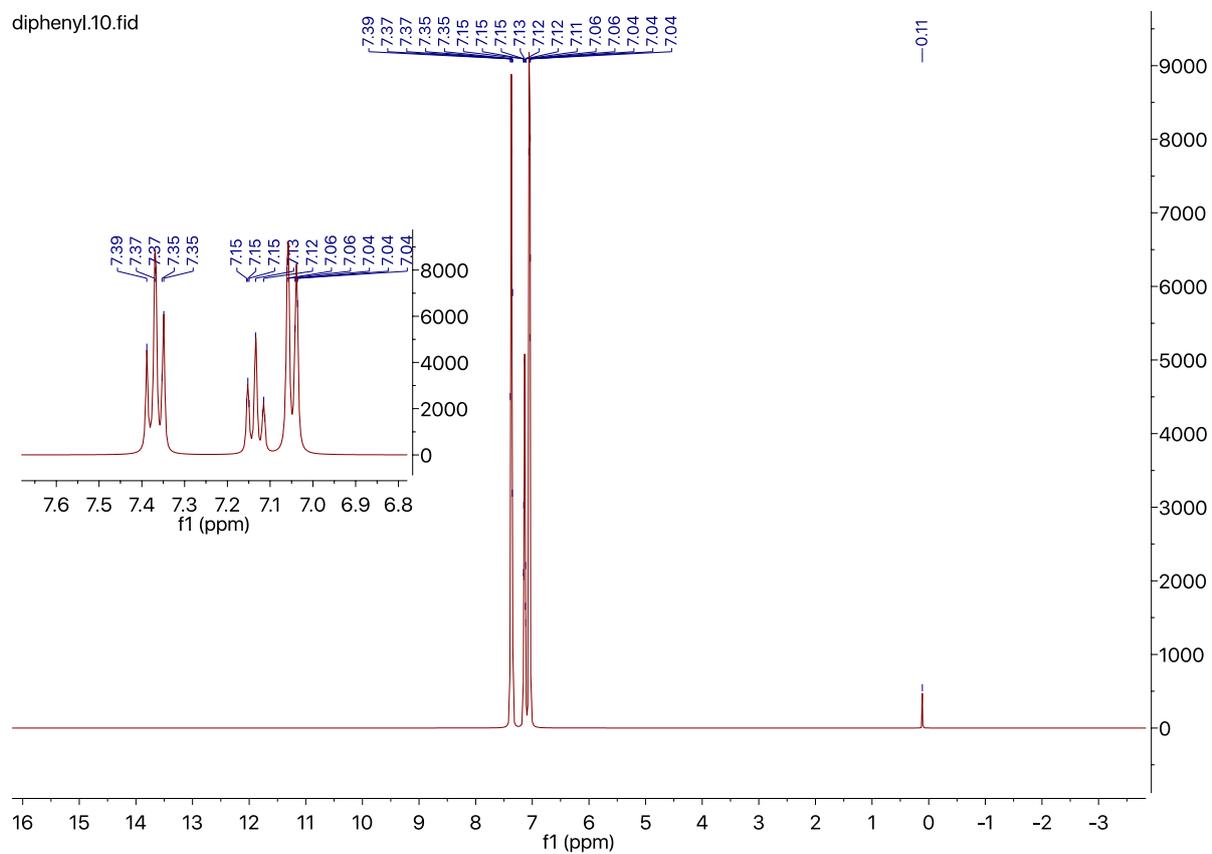


Figure S9. ^1H NMR for the diphenyl ether obtained from the model reaction.

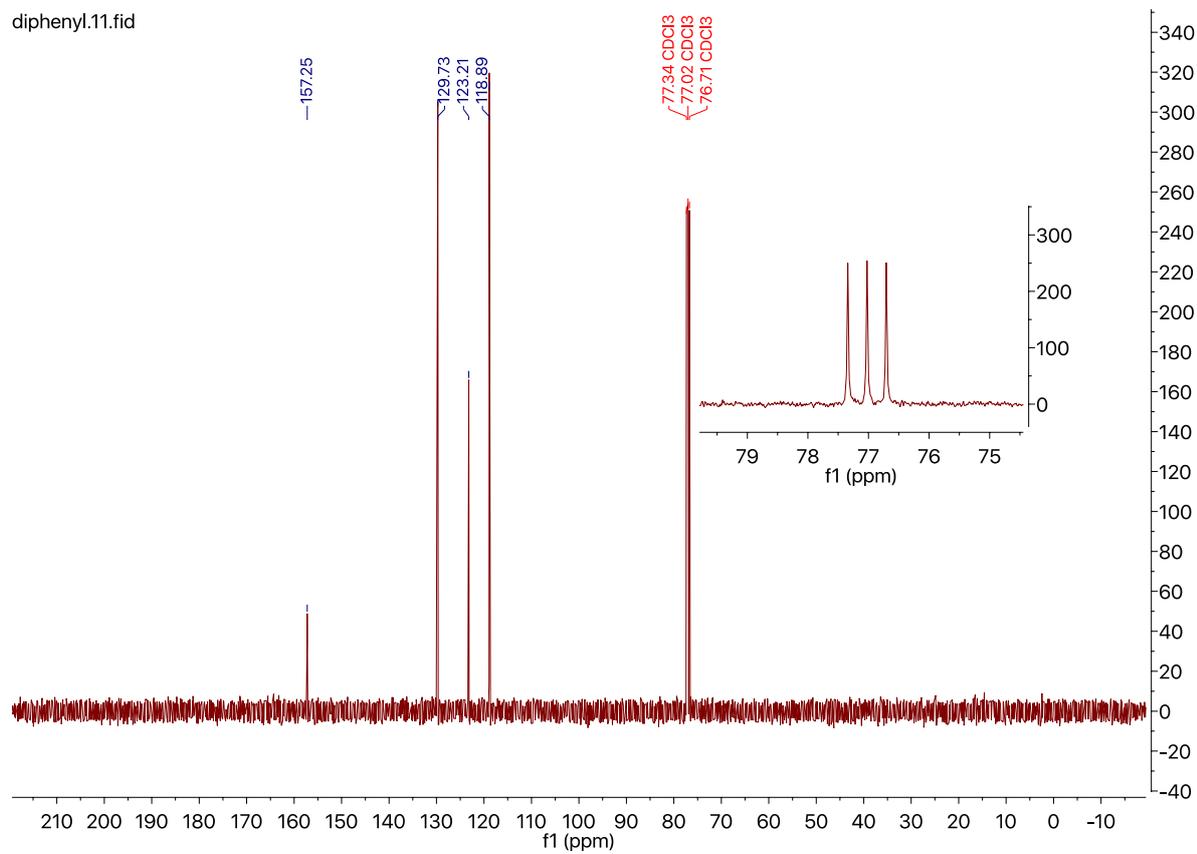


Figure S10. ^{13}C NMR for the diphenyl ether obtained from the model reaction.