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Supporting Information

Graphene quantum dots-assisted morin-KMnO₄ chemiluminescence system for precise recognition of cypermethrin

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Species	Tolerable concentration ratio (10 ³) [C _{interferent (µg/L)} /C _{CY (mg/L)}]
Na ⁺ , Cl ⁻ , K ⁺ , SO ₄ ^{2–} , CH ₃ COO [–]	800
Br ⁻ , CO ₃ ^{2–} , PO ₄ ^{3–} , Mg ²⁺ , Ca ²⁺ , Tartaric acid, Lactose, Valine	600
Methanol	640
Ethanol	600
Ni ²⁺ , Pb ²⁺ , Al ³⁺ , Zn ²⁺	220
Fe ²⁺ , Co ²⁺	140
Cu ²⁺	20

Table S1. Tolerable concentration ratios with respect to 8.0 mg L^{-1} of CYPM.



Figure S1. Schematic diagram of flow-injection CL system; (a): H₂SO₄ solution; (b): sample or standard solution of mixture of morin, GQDs, CTAB and cypermethrin; (c): H₂O as the carrier; (d): KMnO₄ solution; P: peristaltic pump; M: mixing tube; V: injection valve; F: flow cell; W: waste; D: detector (luminometer); R: recorder (personal computer).





Figure S2. Optimization of the CL reaction conditions: (a) effect of KMnO₄ concentration. Conditions: the concentrations of H₂SO₄, GQDs, CTAB, and morin were 0.5 mol L⁻¹, 0.32 mmol L⁻¹, 0.02 mmol L⁻¹ and 0.01 mmol L⁻¹, respectively; (b) effect of H₂SO₄ concentration. Conditions: the concentrations of KMnO₄ was 0.08 mmol L⁻¹, other conditions were as in (a), (c) effect of morin concentration. Conditions: the concentrations of H₂SO₄ was 0.5 mol L⁻¹, other conditions were as in (b), (d) effect of GQDs concentration. Conditions: the concentration. Conditions were as in (c), and (e) effect of CTAB concentration. Conditions: the concentrations of GQDs was 0.64 mmol L⁻¹, other conditions were as in (d).



Figure S3. Fluorescence emission spectra for $0.1 \text{ mmol } L^{-1}$ morin in the presence of different amounts of CTAB.



Figure S4. Calibration graph with two linear dynamic range for the determination of CYPM by morin-GQDs-CTAB-KMnO₄ CL (First linear range: 0.05-0.3 mg L^{-1} and second linear range: 0.3-22 mg L^{-1}).