

Electronic Supporting Information

Boratriazines: Inducing luminescence through boron incorporation into a terpy-type framework.

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MY-01-08-1
1H NMR

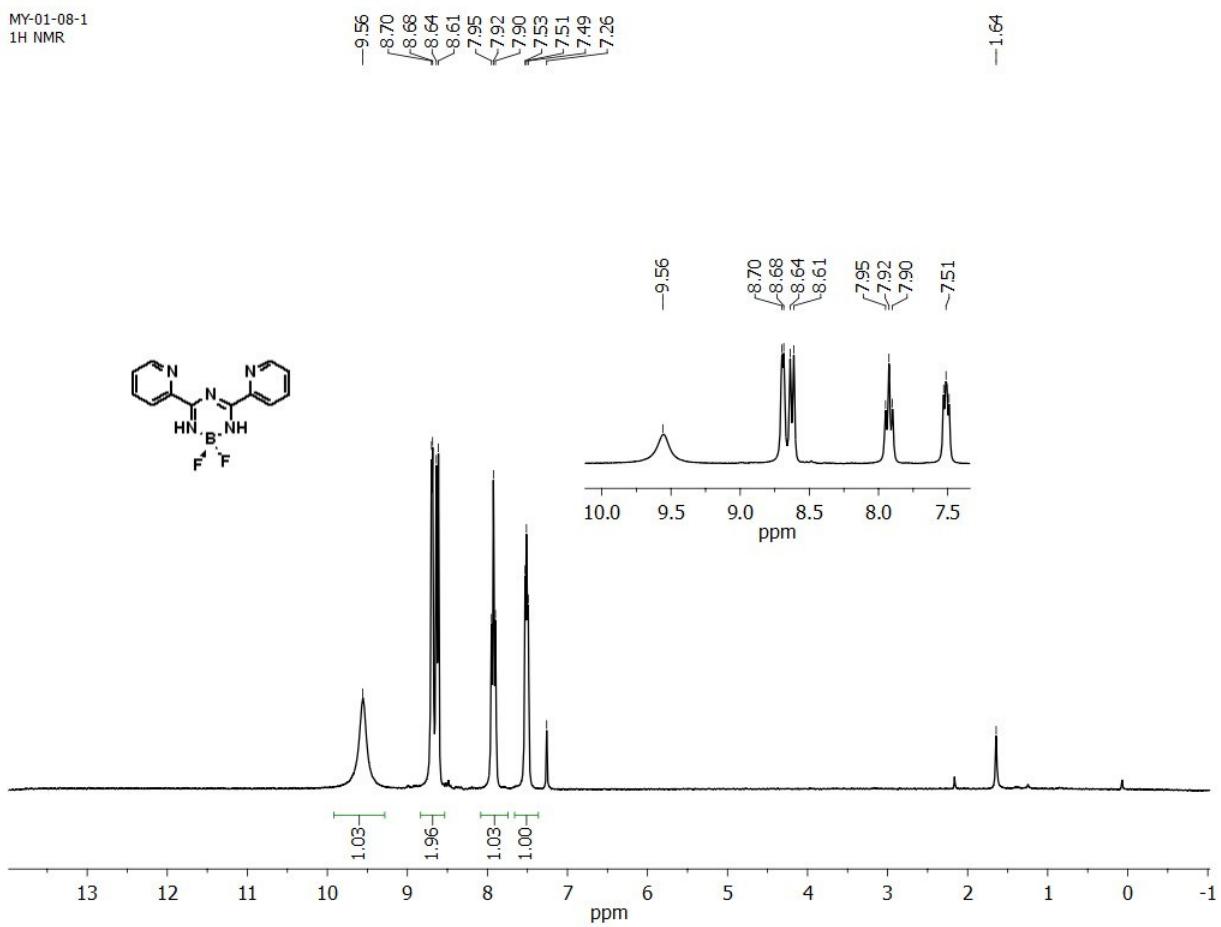


Figure S1. ^1H NMR spectrum of $\text{F}_2\text{-Py}_2\text{BTA}$ in CDCl_3

MY-01-08-1
13C NMR with 1H decoupling

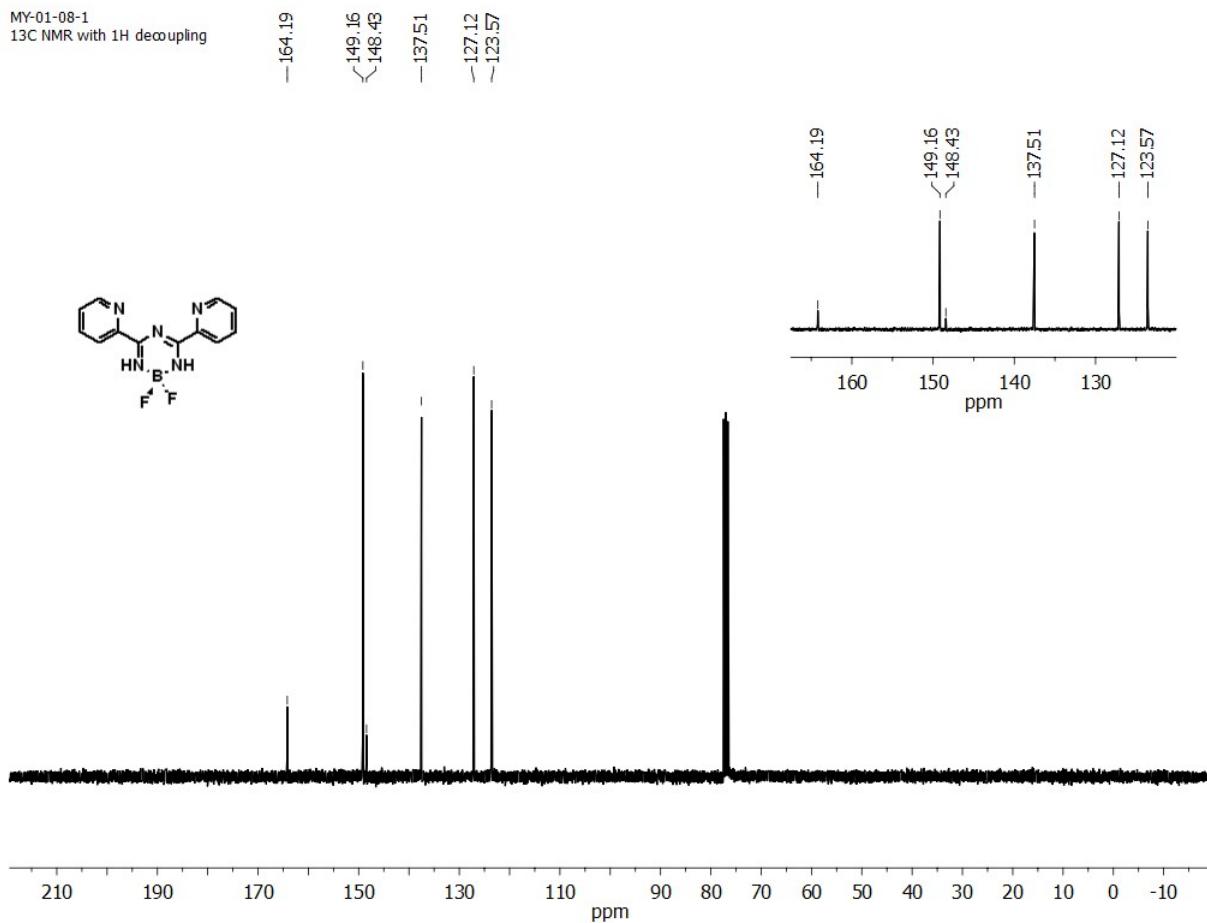


Figure S2. ¹³C NMR spectrum of F₂-Py₂BTA in CDCl₃

MY-01-08-1
11B NMR with 1H decoupling

0.51
0.24
-0.03

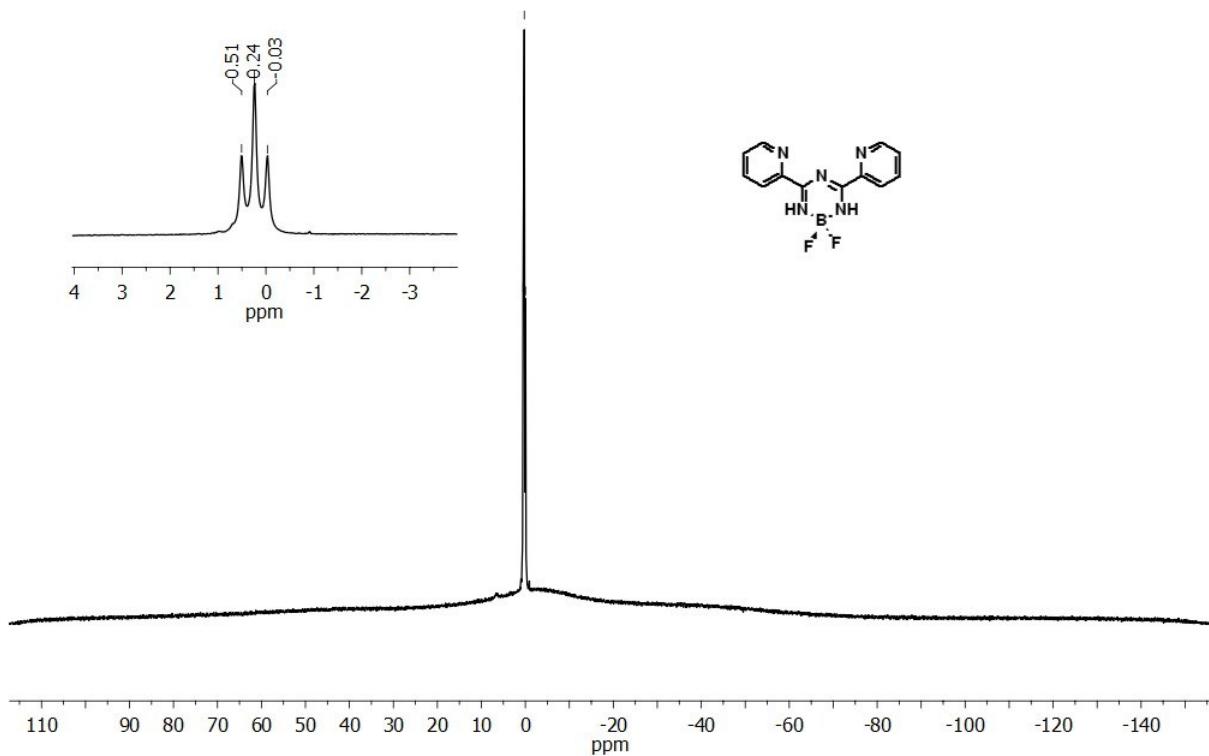


Figure S3. ^{11}B NMR spectrum of $\text{F}_2\text{-Py}_2\text{BTA}$ in CDCl_3

MY-01-08-1
19F with proton decoupling

-127.73
-127.82
-127.92
-128.01

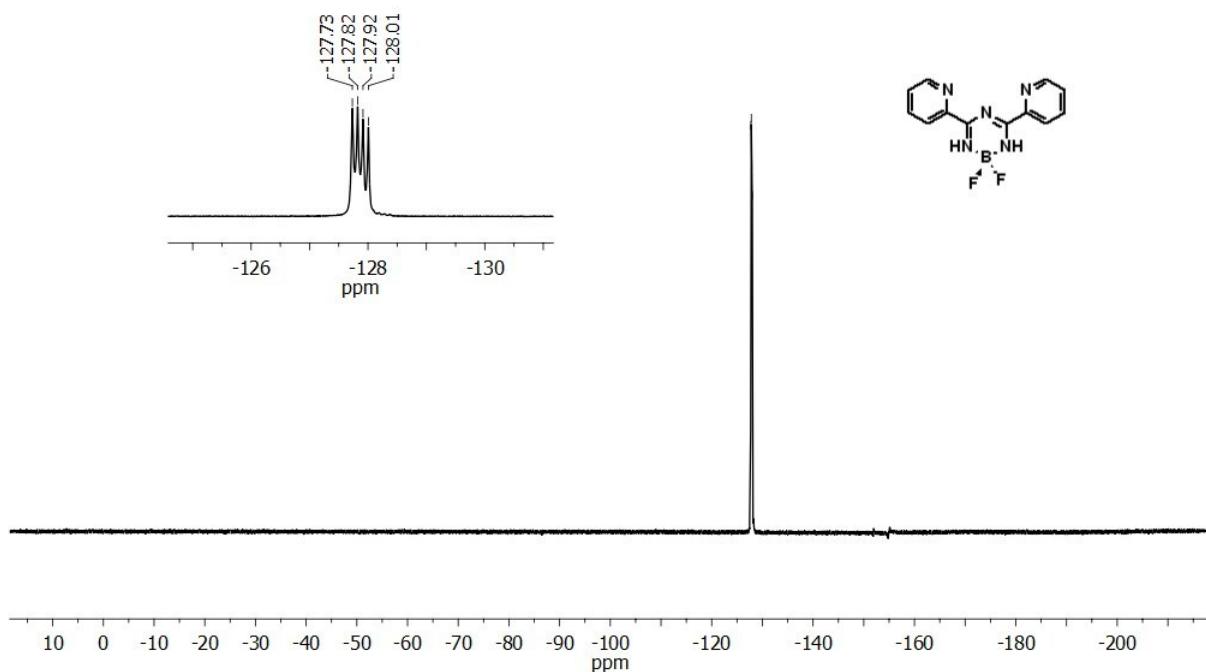


Figure S4. ¹⁹F NMR spectrum of F₂-Py₂BTA in CDCl₃

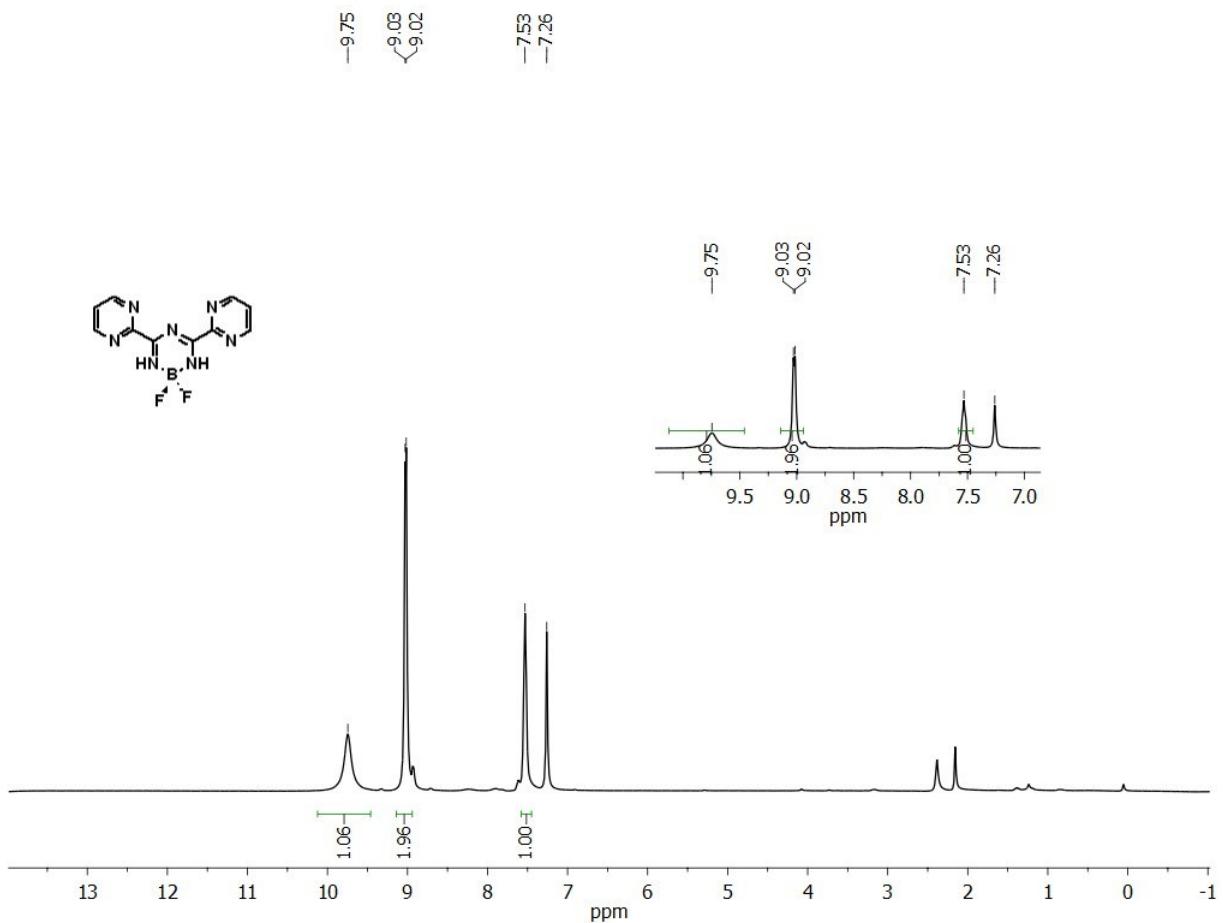


Figure S5. ^1H NMR spectrum of $\text{F}_2\text{-Pm}_2\text{BTA}$ in CDCl_3

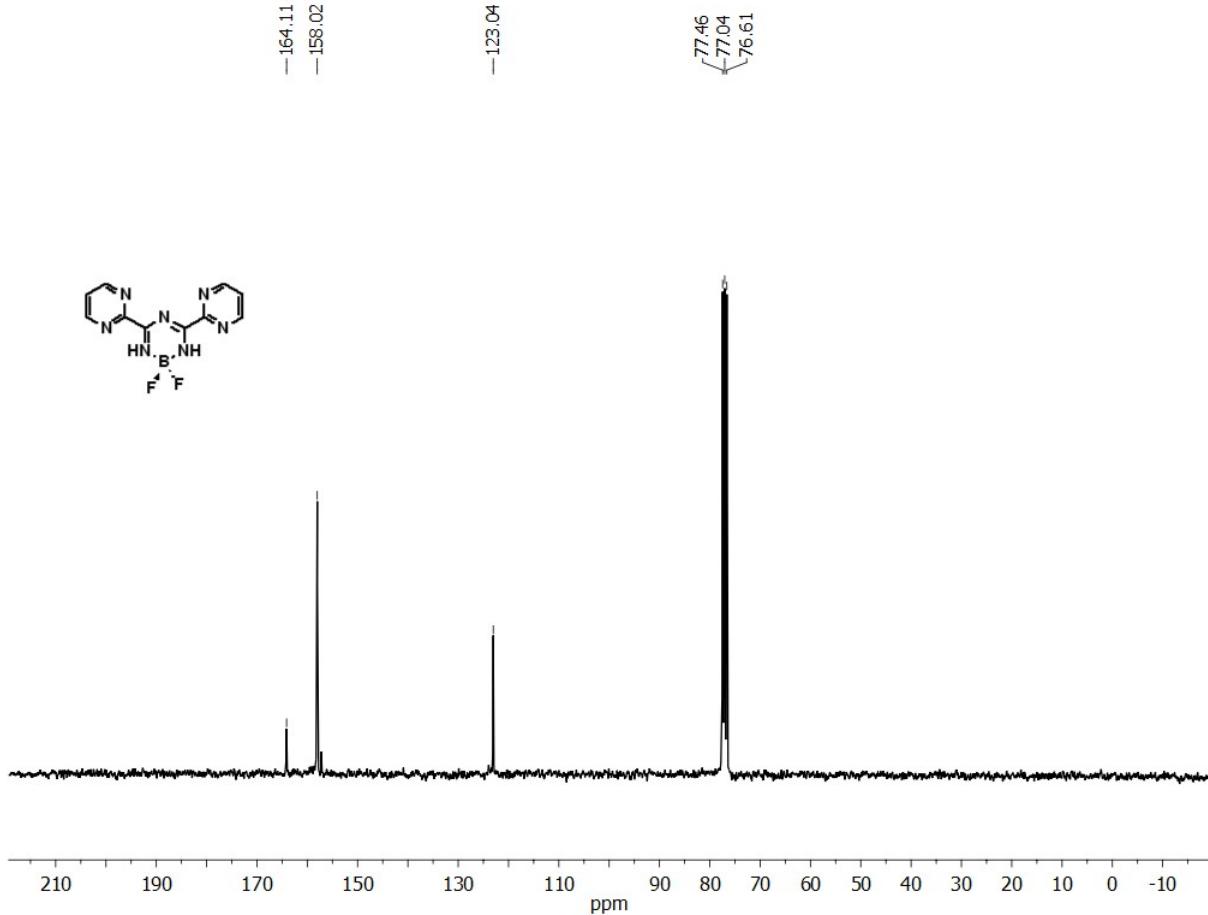


Figure S6. ^{13}C NMR spectrum of $\text{F}_2\text{-Pm}_2\text{BTA}$ in CDCl_3

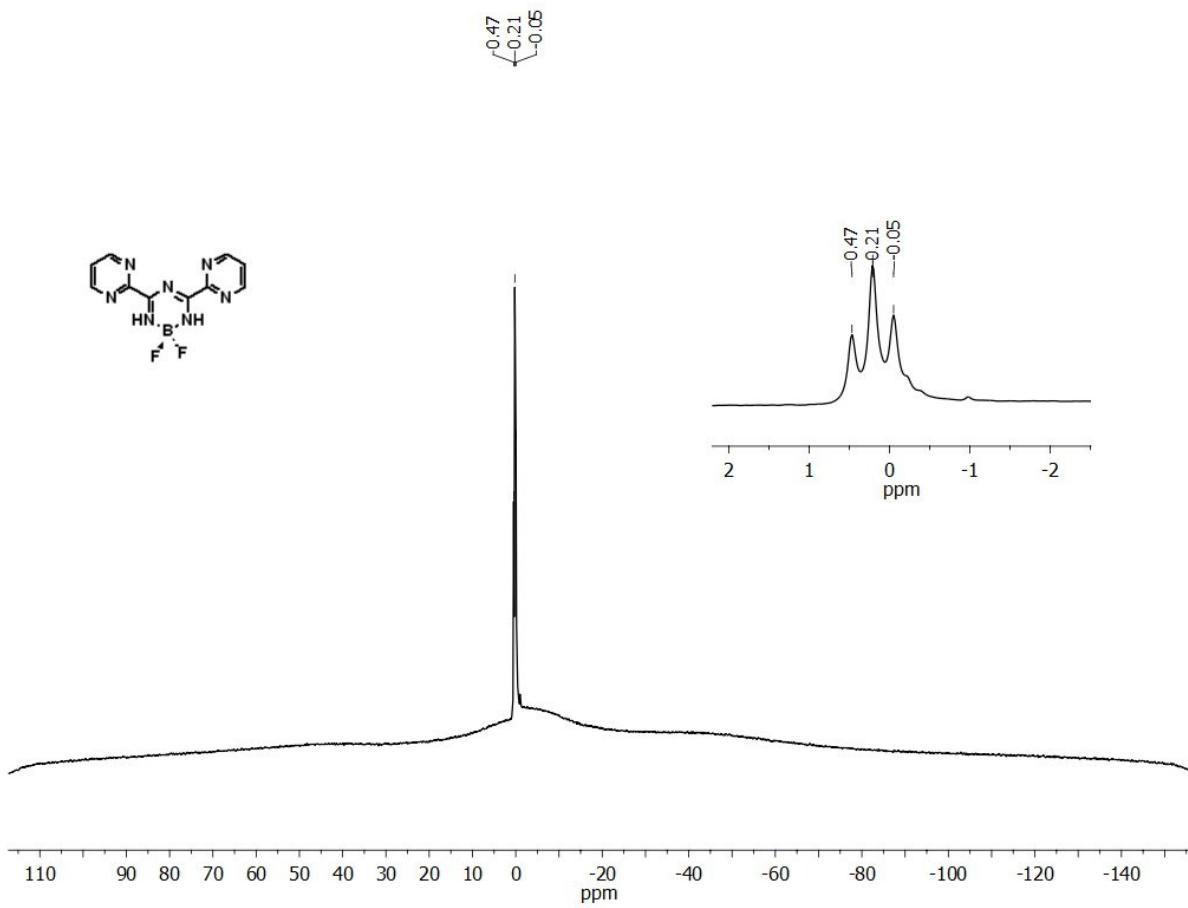


Figure S7. ^{11}B NMR spectrum of $\text{F}_2\text{-Pm}_2\text{BTA}$ in CDCl_3

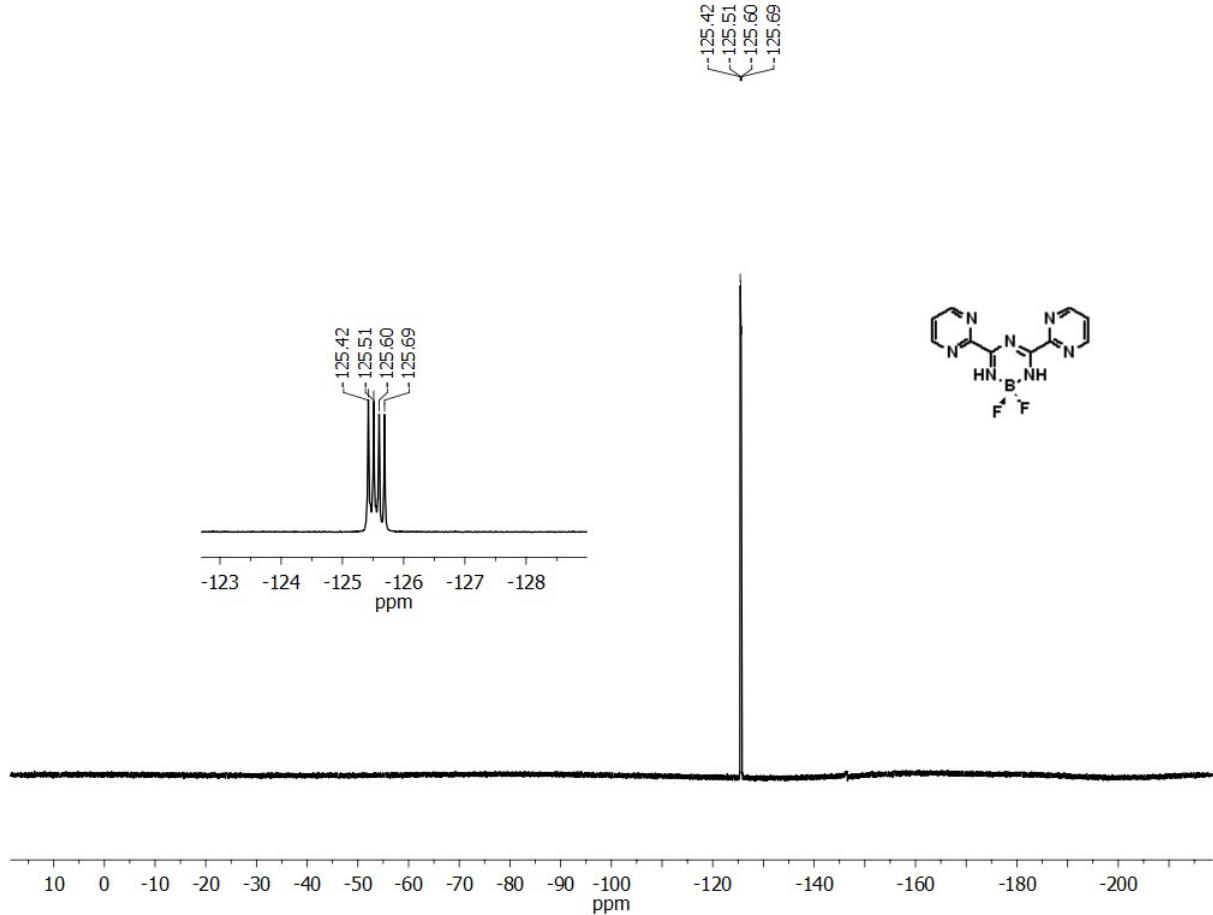


Figure S8. ^{19}F NMR spectrum of $\text{F}_2\text{-Pm}_2\text{BTA}$ in CDCl_3

MY-01-21-3
1H NMR

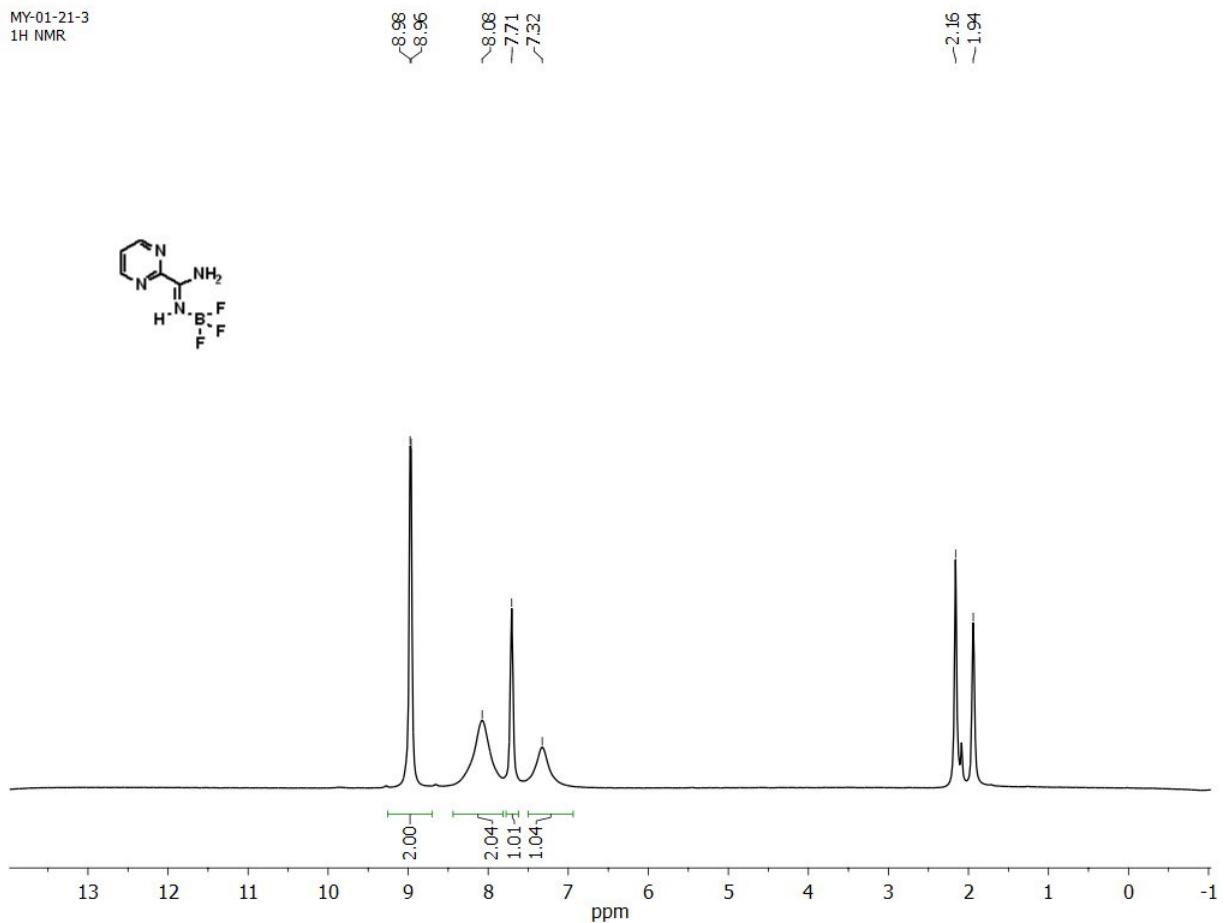


Figure S9. ¹H NMR spectrum of **PmAmBF₃** in CD₃CN

MY-01-21-3
13C NMR with 1H decoupling

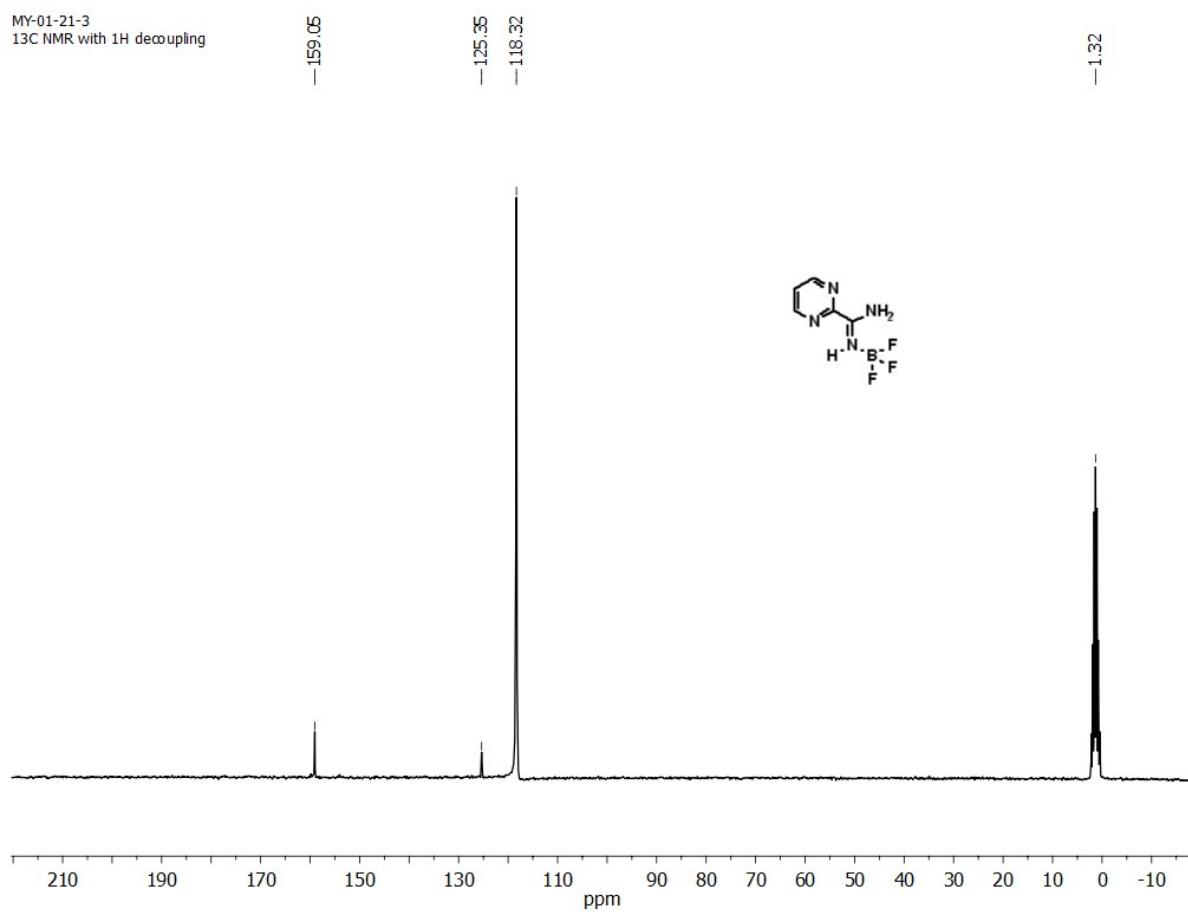


Figure S10. ¹³C NMR spectrum of PmAmBF₃ in CD₃CN

MY-01-21-3
11B NMR with 1H decoupling

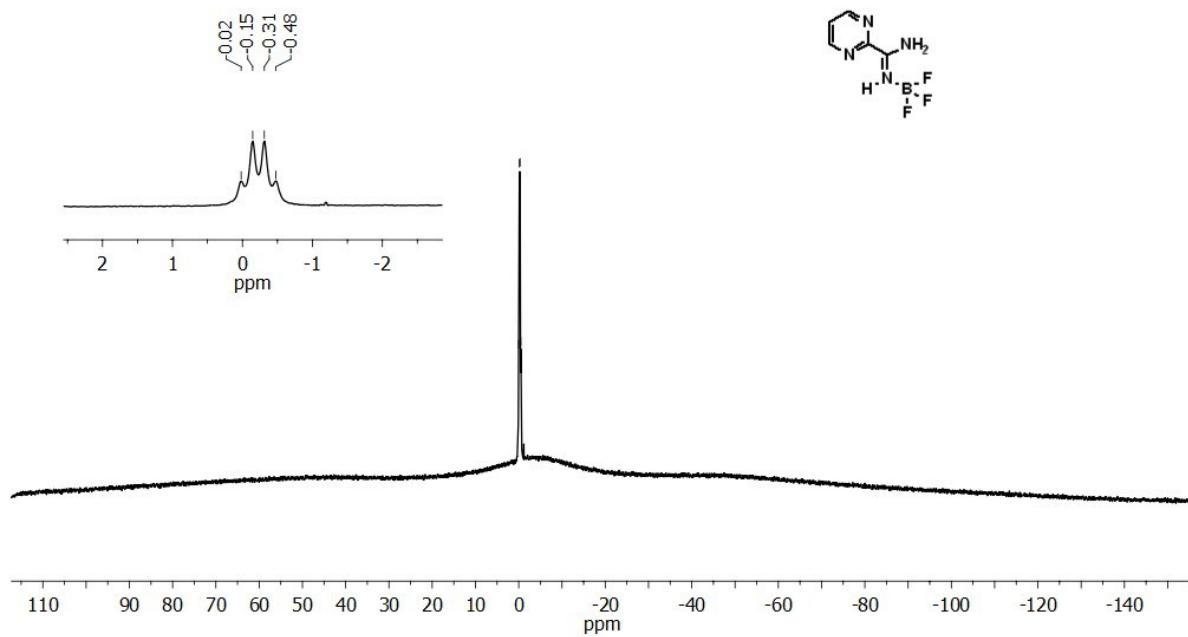


Figure S11. ^{11}B NMR spectrum of PmAmBF_3 in CD_3CN

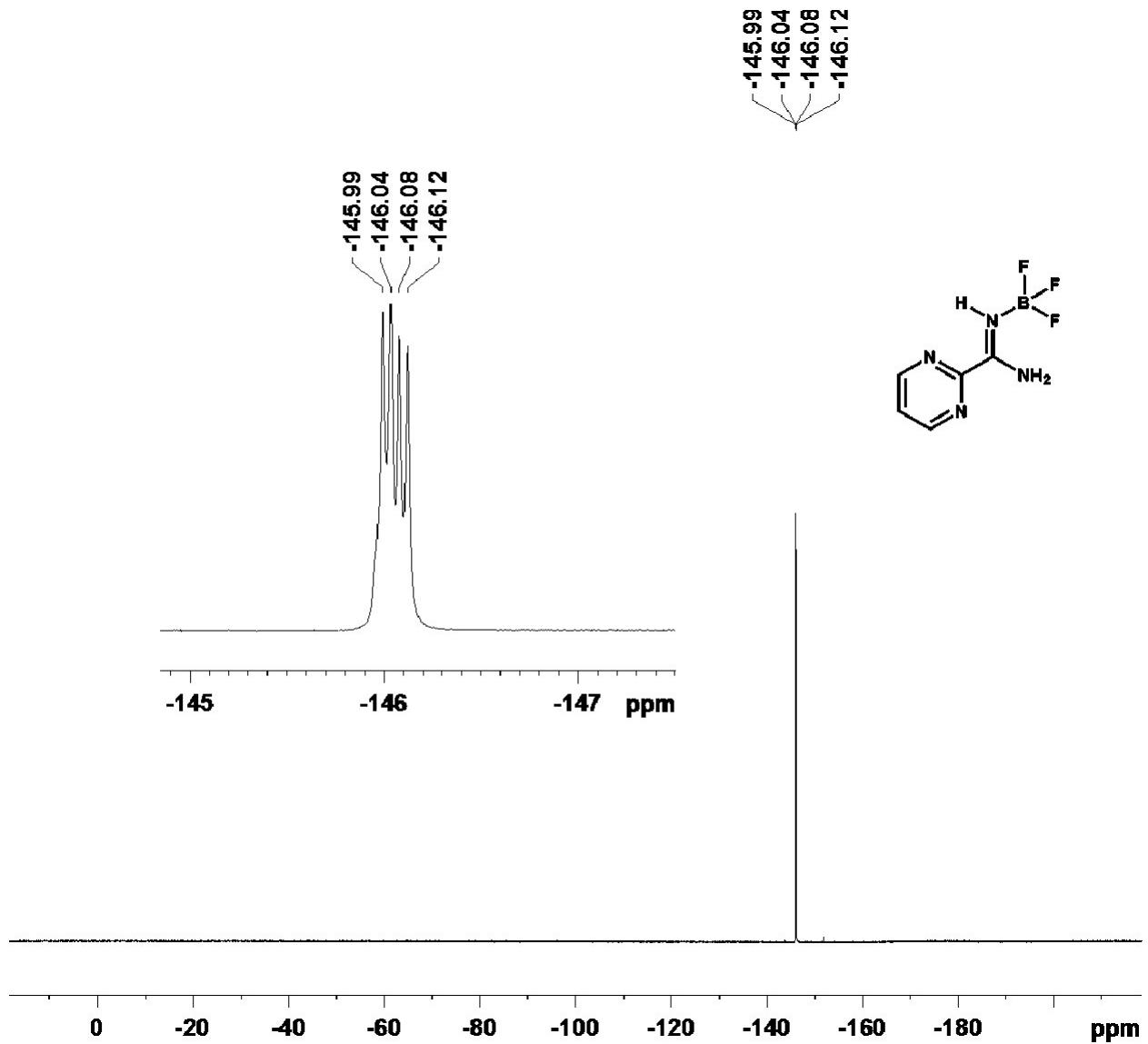


Figure S12. ^{19}F NMR spectrum of PmAmBF₃ in CD₃CN

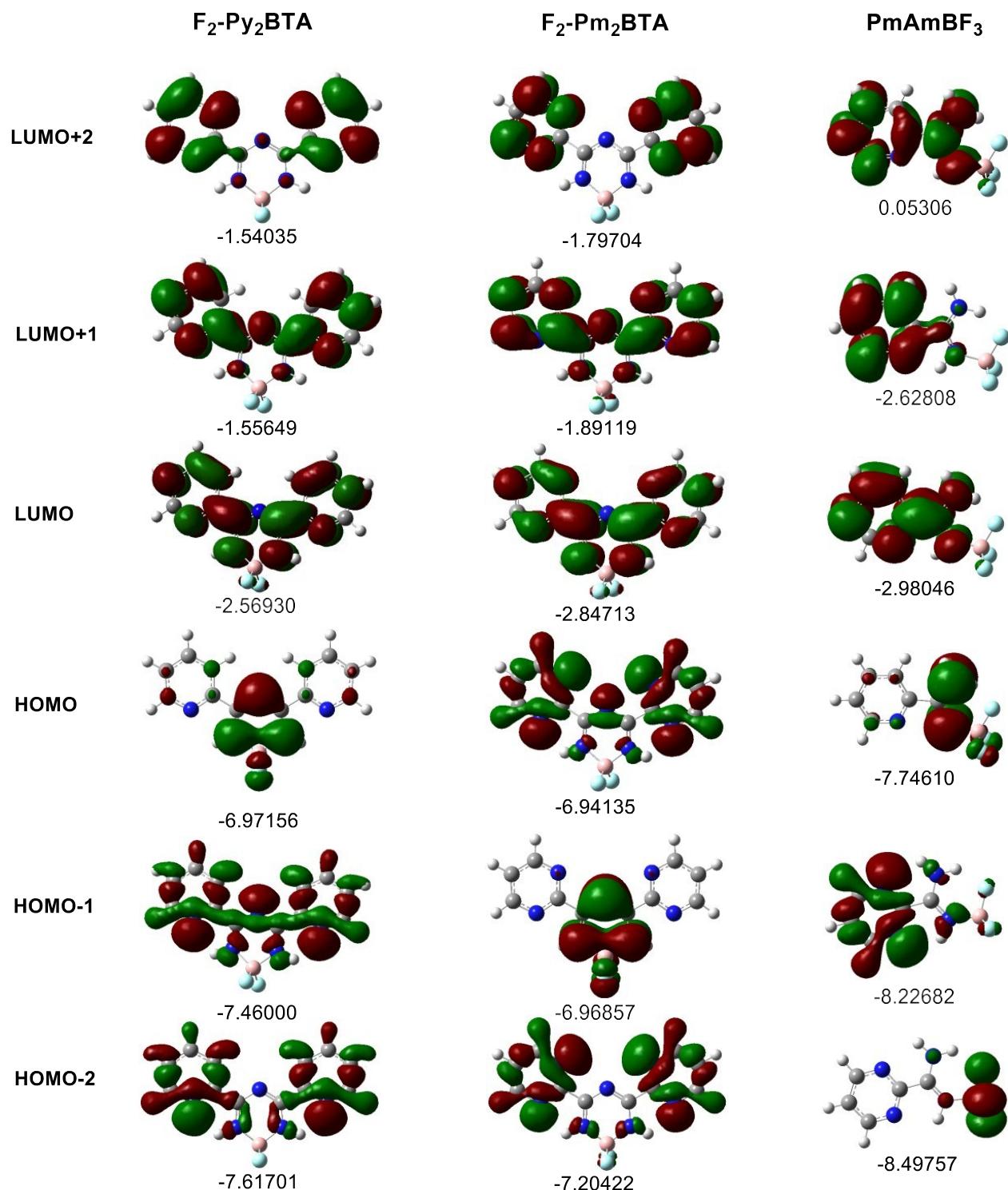


Figure S13. Frontier molecular orbitals and energies of the $\text{F}_2\text{-Py}_2\text{BTA}$, $\text{F}_2\text{-Pm}_2\text{BTA}$ and PmAmBF_3 from DFT calculations (B3LYP, 6-31G).

Table S1. Fluorescence data for the benchmark fluorophores **F₂-Py₂BTA** and **F₂-Pm₂BTA**.

Compound	Absorption ^a		Fluorescence ^b	
	λ_{max} (nm)	($\epsilon \times 10^4$ $LM^{-1}cm^{-1}$)	λ_{em} (nm)	Quantum Yield ^c
F₂-Py₂BTA	310	2.3	397	0.12
F₂-Pm₂BTA	307	1.4	388	0.07

^a low energy absorption of **F₂-Py₂BTA** and **F₂-Pm₂BTA** in DCM from UV-Vis. ^b Measurements were made in DCM (dichloromethane). ^c Quantum yield were calculate at 22°C relative to 9,10-diphenylanthracene in ethanol (QY = 0.90) using Cary Eclipse Fluorescence Spectrophotometer (Agilent Technologies).

Table S2. Molar extinction coefficients (ϵ) of **F₂-Py₂BTA**, **F₂-Pm₂BTA** and **PmAmBF₃** along with their corresponding wavelengths.

F₂-Py₂BTA		F₂-Pm₂BTA		PmAmBF₃	
λ_{abs} (nm)	ϵ ($LM^{-1}cm^{-1}$)	λ_{abs} (nm)	ϵ ($LM^{-1}cm^{-1}$)	λ_{abs} (nm)	ϵ ($LM^{-1}cm^{-1}$)
324	19720	317	12720	320	489
310	22795	307	14015	304	635
280	11580	295	10776	227	7533

Table S3. TD-DFT optical transitions^a for **F₂-Py₂BTA**, **F₂-Pm₂BTA** and **PmAmBF₃**.

k	F₂-Py₂BTA		F₂-Pm₂BTA		PmAmBF₃	
	E(eV)	f	E(eV)	f	E(eV)	f
1	4.1154	0.4515	3.5015	0.0000	4.1920	0.0000
2	4.1370	0.0000	3.7135	0.0000	4.5825	0.0055
3	4.3078	0.0024	4.0622	0.2796	4.8527	0.0664
4	4.6294	0.1542	4.3130	0.0000	5.0795	0.0039
5	4.6927	0.0965	4.4048	0.0089	5.6628	0.0135
6	4.7077	0.0000	4.4084	0.0000	5.7256	0.3731
7	5.1376	0.0001	4.6683	0.0010	5.8142	0.0028
8	5.1890	0.0001	4.8624	0.0606	6.2399	0.0001
9	5.2176	.0233	4.9013	0.0000	6.2595	0.0019
10	5.3439	0.0041	4.9560	0.2215	6.4273	0.0002
11	5.3793	0.0000	4.9703	0.0587	6.9605	0.0398
12	5.3843	0.0020	5.0603	0.0788	6.9847	0.0001
13	5.4987	0.0277	5.0756	0.0198	7.0065	0.0004
14	5.5133	0.0016	5.2797	0.0030	7.0711	0.2130
15	5.5333	0.0000	5.2859	0.0000	7.0988	0.0134
16	5.5612	0.2197	5.2865	0.0000	7.3725	0.0000
17	5.6601	0.0188	5.4050	0.0000	7.4656	0.4011
18	5.7248	0.0010	5.5061	0.0046	7.6039	0.0123
19	5.9429	0.0004	5.5069	0.0153	7.6157	0.0008
20	5.9517	0.0297	5.5306	0.0019	7.6971	0.0000

^aTD-DFT (6-31G/B3LYP) level of theory on geometry optimized structures. k is order of excitation energy and f is oscillator strength.

Table S4. Selected TD-DFT optical transitions for **F₂-Py₂BTA**, **F₂-Pm₂BTA** and **PmAmBF₃**.

Compound	Excitation Energy (eV)	Calculate d λ (nm)	Transition	Molecular Contribution	Oscillator Strength (<i>f</i>)
F₂-Py₂BTA	4.1154	301	HOMO → LUMO	0.69836	0.4515
	4.6294	267	HOMO-1 → LUMO	0.69018	0.1542
	4.6927	264	HOMO-3 → LUMO	0.68831	0.0965
F₂-Pm₂BTA	4.0622	305	HOMO-1 → LUMO	0.70295	0.2796
	4.8624	254	HOMO-3 → LUMO	0.55938	0.00000
	4.9560	250	HOMO-1 → LUMO+1	-0.37579	
PmAmBF₃	4.5827	255	HOMO → LUMO	0.70088	0.0664
	5.7256	216	HOMO-2 → LUMO	0.68647	0.3731
	7.4656	175	HOMO-5 → LUMO	0.52270	0.2130
			HOMO-2 → LUMO+1	0.37437	

Table S5. Crystallographic data for **F₂-Py₂BTA**, **F₂-Pm₂BTA** and **PmAmBF₃**.

	F₂-Py₂BTA	F₂-Pm₂BTA	PmAmBF₃
Formula	C ₁₂ H ₁₀ BF ₂ N ₅	C ₁₀ H ₈ BF ₂ N ₇	C ₅ H ₆ BF ₃ N ₄
FW (g/mol)	273.06	359.97	189.95
Crystem System	Triclinic	Orthorhombic	Monoclinic
Space Group	<i>P</i> - <i>I</i>	<i>Pbca</i>	<i>P-2</i> ₁ / <i>n</i>
a (Å)	7.7785 (3)	9.7915(7)	8.1901(2)
b (Å)	8.8782(3)	10.7841(8)	6.2150(2)
c (Å)	9.6210(4)	28.665(2)	14.8496(4)
α (°)	91.401(2)	90	90
β (°)	102.708(2)	90	92.216(2)
γ (°)	109.795(2)	90	90
V (Å ³)	606.289(4)	3026.8(4)	755.30(4)
Z	2	8	4
D _{calc} (Mg·m ⁻³)	1.496	1.580	1.670
T(K)	200	200	200
μ (mm ⁻¹)	0.116	0.459	0.159
2θ range (°)	4.36 to 57.99	5.038 to 51.998	5.49 to 57.99
No. of total reflections	10741	13127	6752
No. of unique reflections	3153	2913	1873
R _{int}	0.0268	0.0608	0.0163
R ₁ , wR ₂ [I>=2σ (I)]	0.0535, 0.1579	0.0692, 0.1794	0.0341, 0.0975
Largest diff. peak/hole (e·Å ⁻³)	0.518 and -0.186	0.289 and -0.459	0.253 and -0.184