

New Journal of Chemistry

**Title:** A biphosphinic ruthenium complex with potent anti-bacterial and anti-cancer activity

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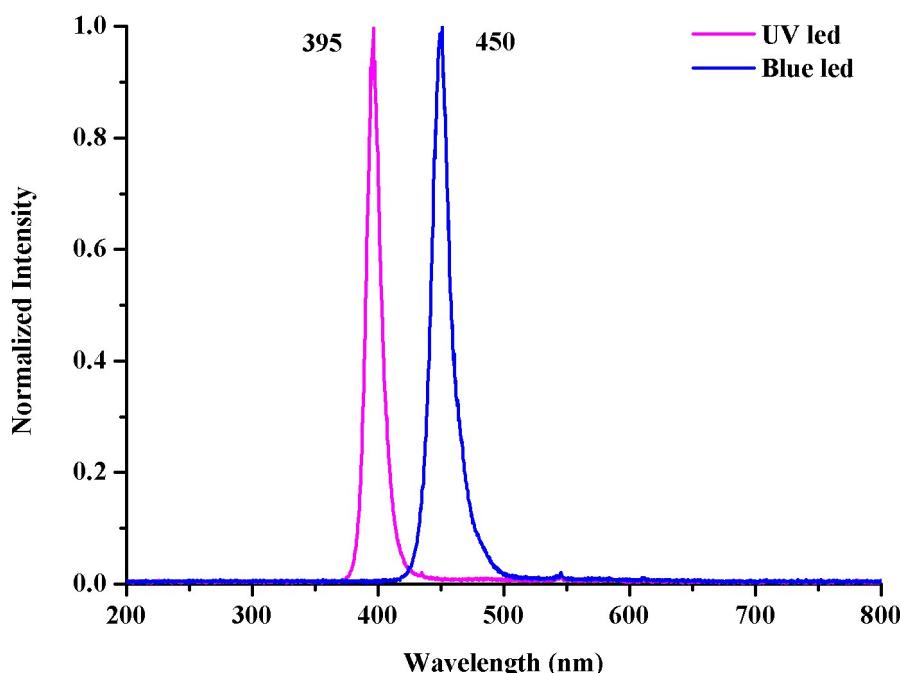


Figure S1. Spectra of the LEDs used in this paper measured using optic fiber in a spectrophotometer from Ocean Optics.

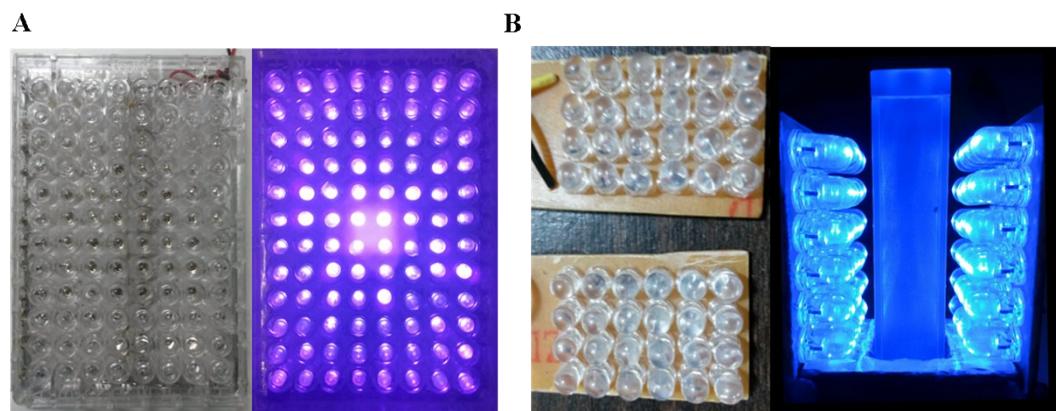


Figure S2. Home-made LEDs used in this work, microplate (A) and cuvette holder (B).

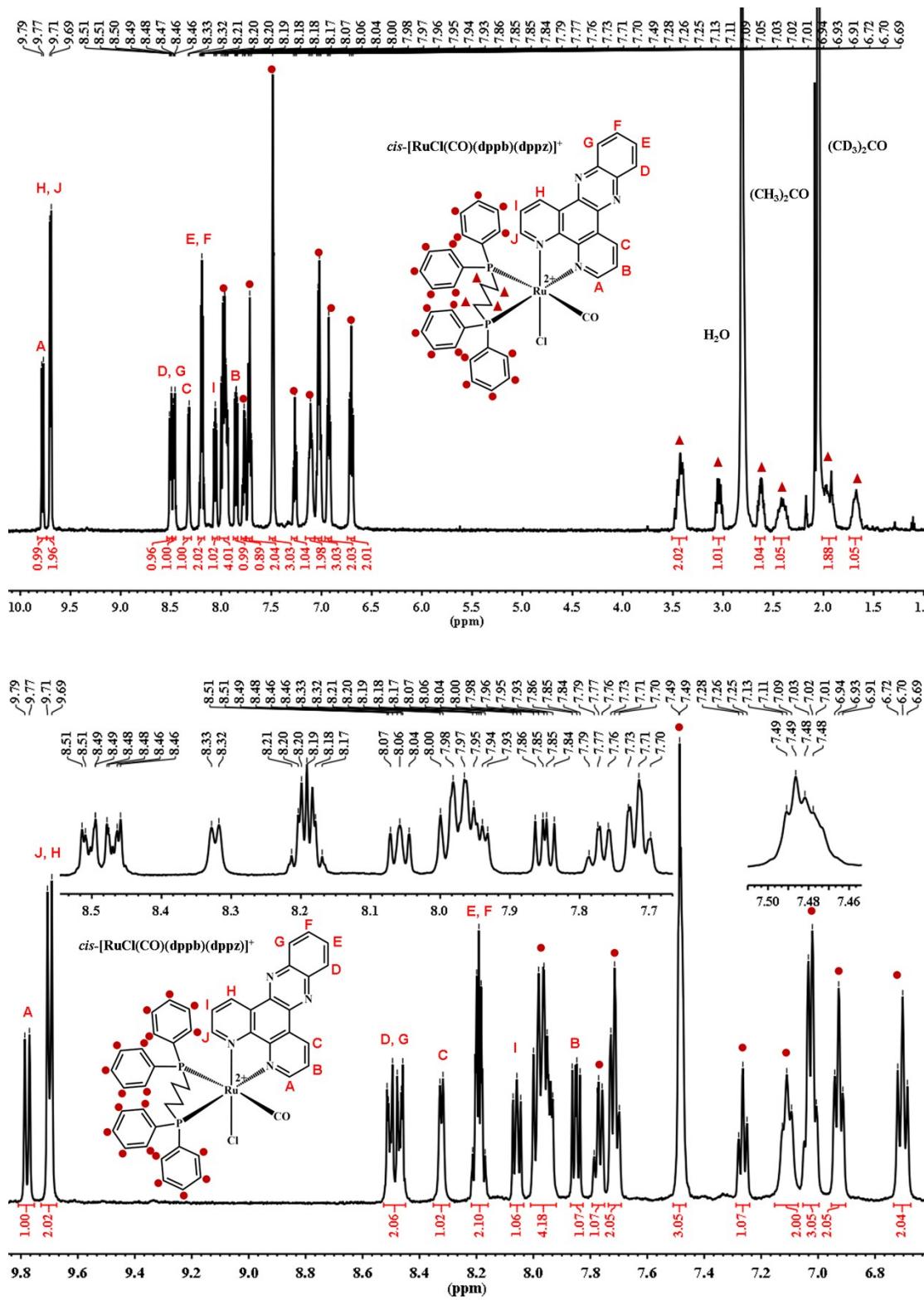


Figure S3.  $^1\text{H}$ -NMR spectrum of cis-[RuCl(dppb)(dppz)CO] $^+$  in  $(\text{CD}_3)_2\text{CO}$  (500 MHz, 298 K). Top spectrum shows the full range of signals and the bottom spectra shows selected ranges.

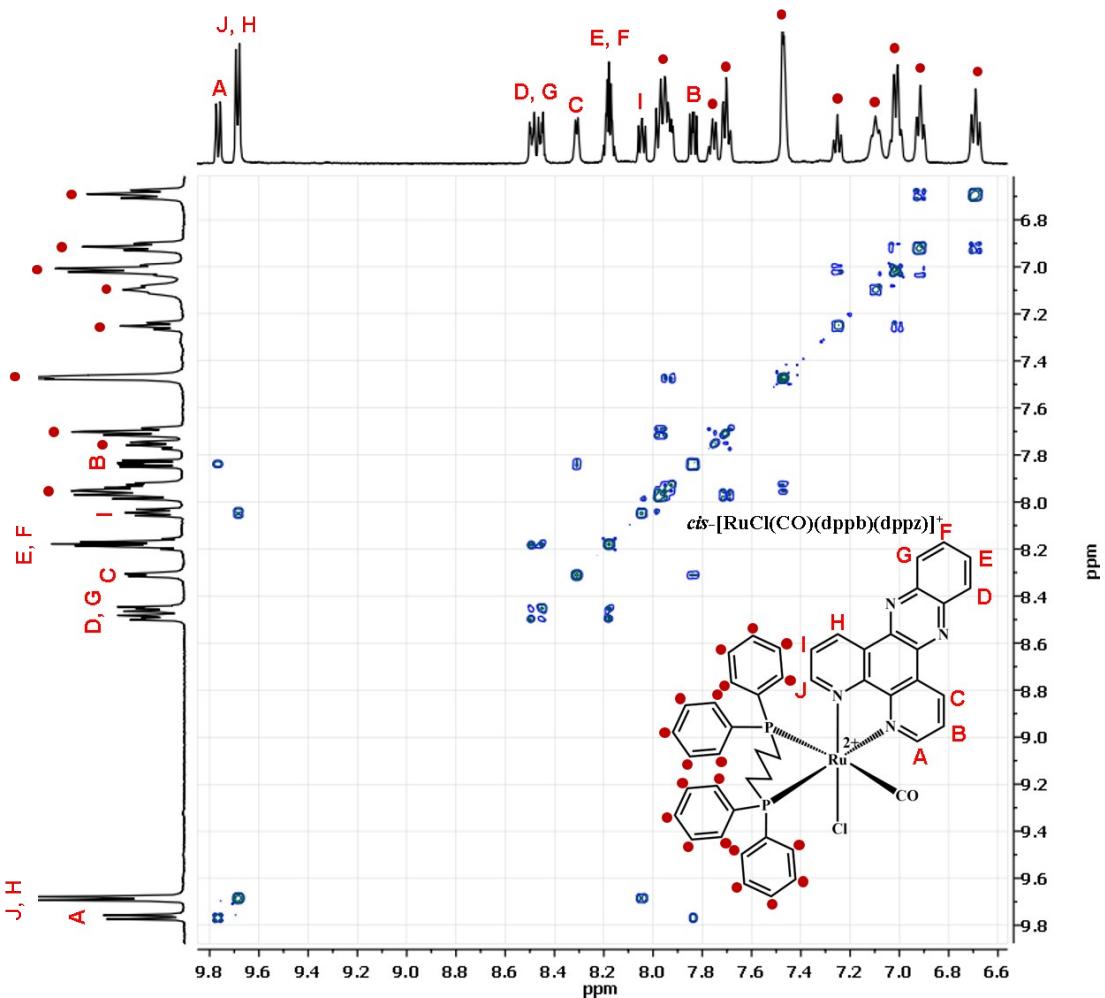


Figure S4. COSY spectrum obtained for *cis*-[Ru(CO)Cl(dppb)(dppz)] $\text{PF}_6$  in  $(\text{CD}_3)_2\text{CO}$  (500 MHz, 298 K)

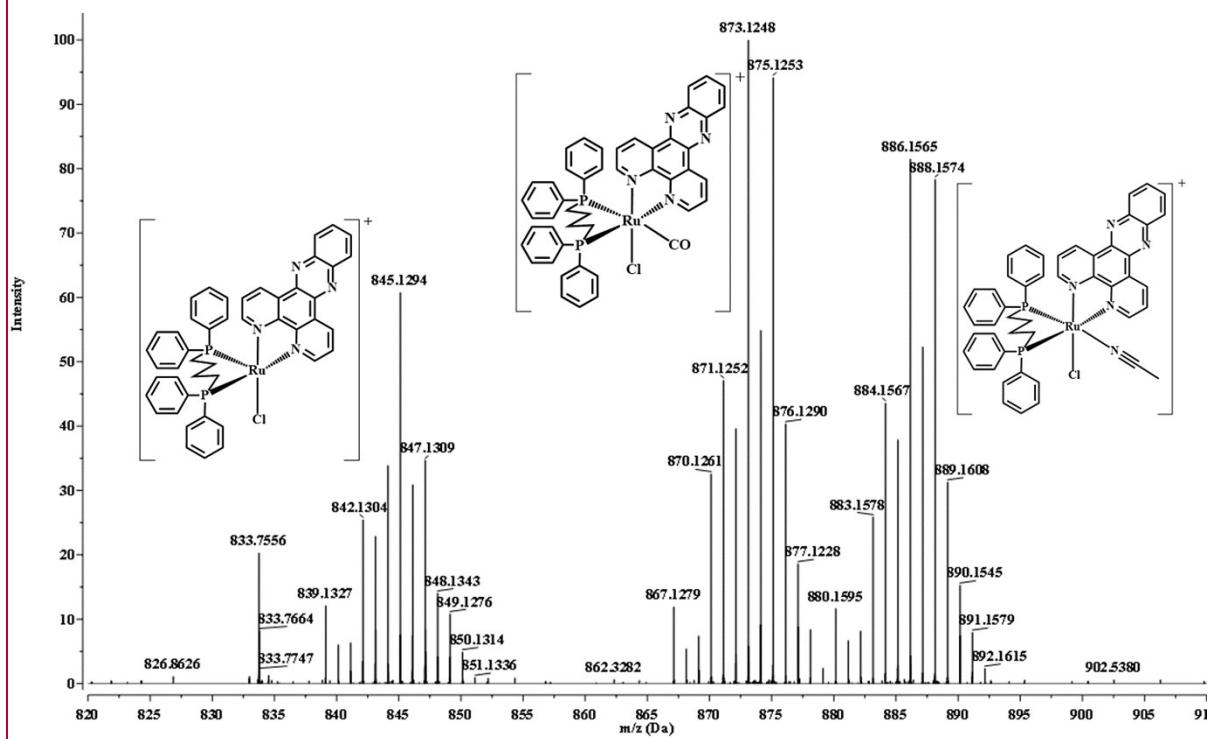


Figure S5 – High resolution mass spectrometry (ESI-TOF) for *cis*-  
[RuCl(CO)(dppb)(dppz)] $\text{PF}_6$  in acetonitrile.

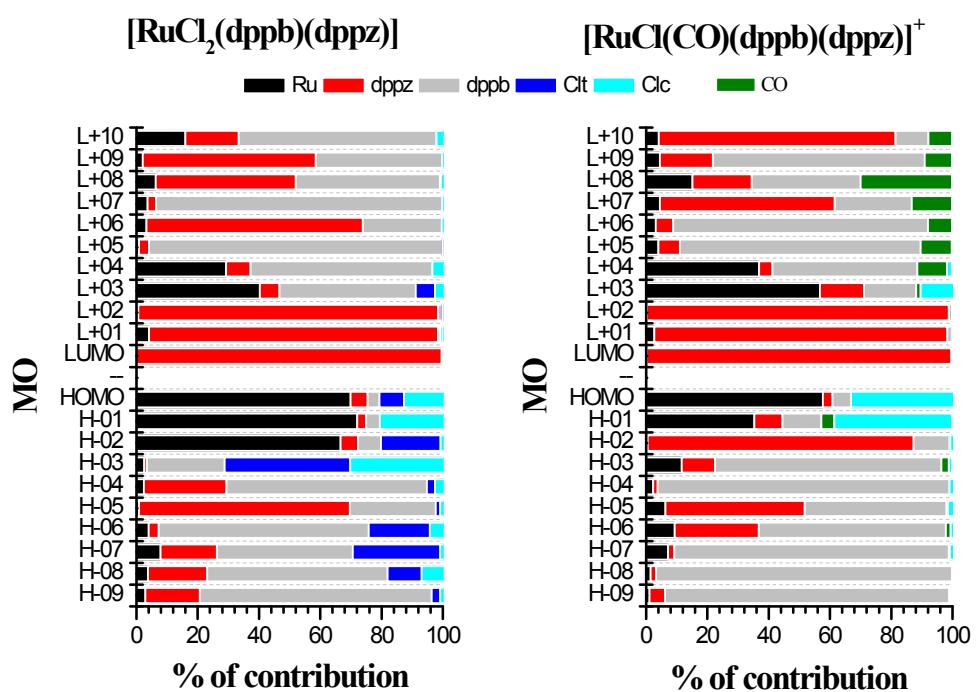


Figure S6. Percentage contribution of  $\text{Cl}^-$ , CO, dppb, dppz, and Ru fragments to selected frontier molecular orbitals of  $[\text{RuCl}_2(\text{dppb})(\text{dppz})\text{L}]$  and  $[\text{RuCl}(\text{CO})(\text{dppb})(\text{dppz})\text{L}]^+$ .

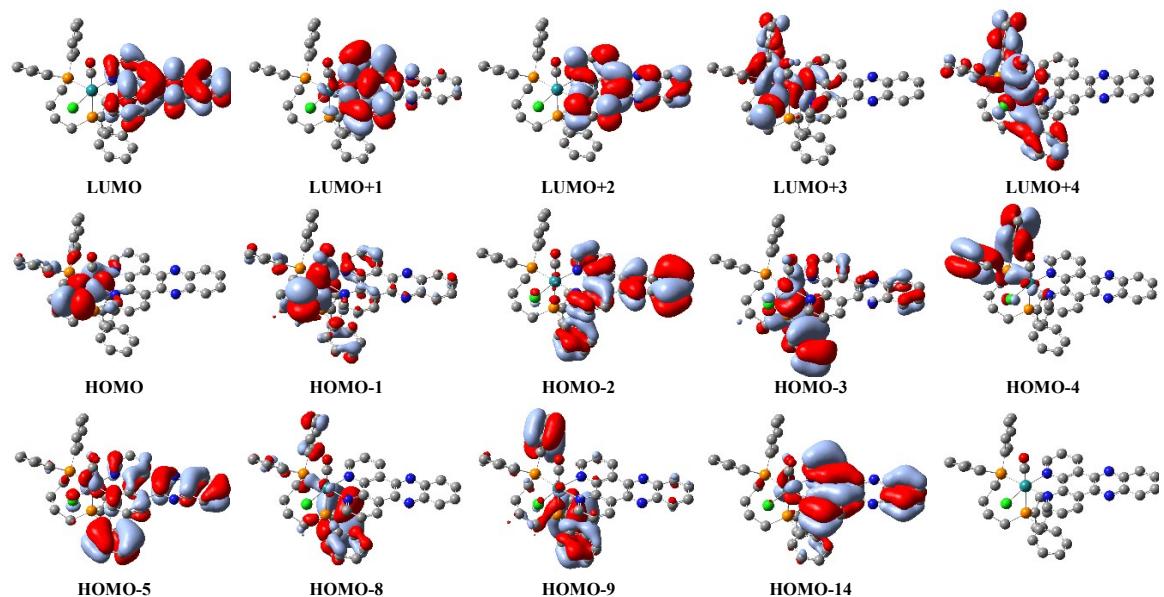


Figure S7. Select molecular orbitals for the ion complex responsible for the major UV-Vis electronic transitions by TD-DFT for cis- $[\text{RuCl}(\text{CO})(\text{dppb})(\text{dppz})\text{L}]^+$ .

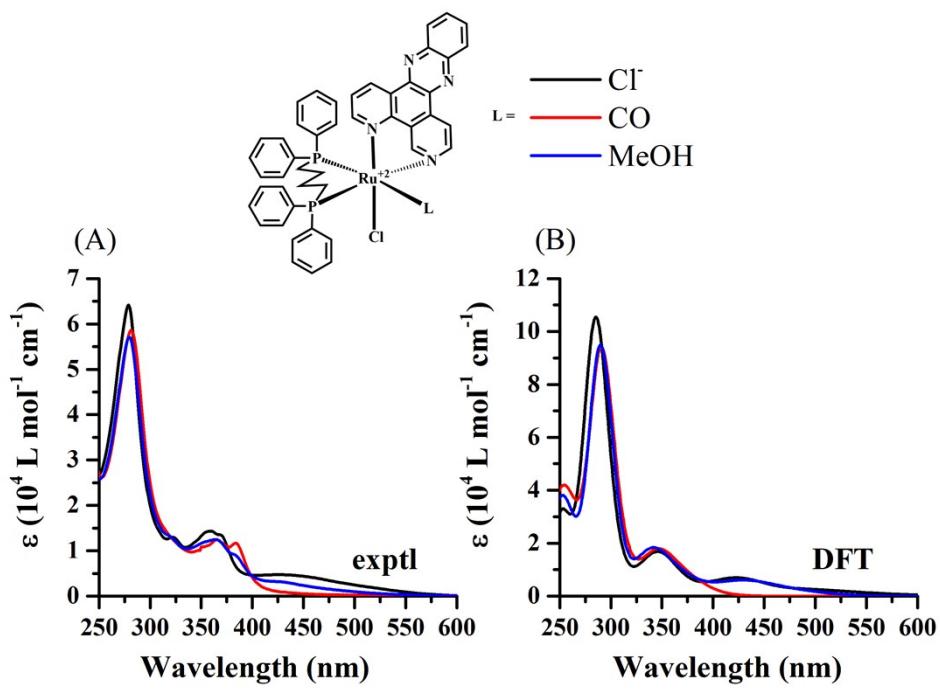


Figure S8 – Overlay of experimental (A) and simulated (B) absorption spectra of  $[\text{RuCl}(\text{dppb})(\text{dppz})\text{L}]$  where  $\text{L} = \text{Cl}^-$ , CO, and  $\text{CH}_3\text{OH}$ . A methanol polarizable continuum model was employed using the B3LYP functional and the basis sets LANL2DZ and 6-31g(d) for Ru and other atoms, respectively.

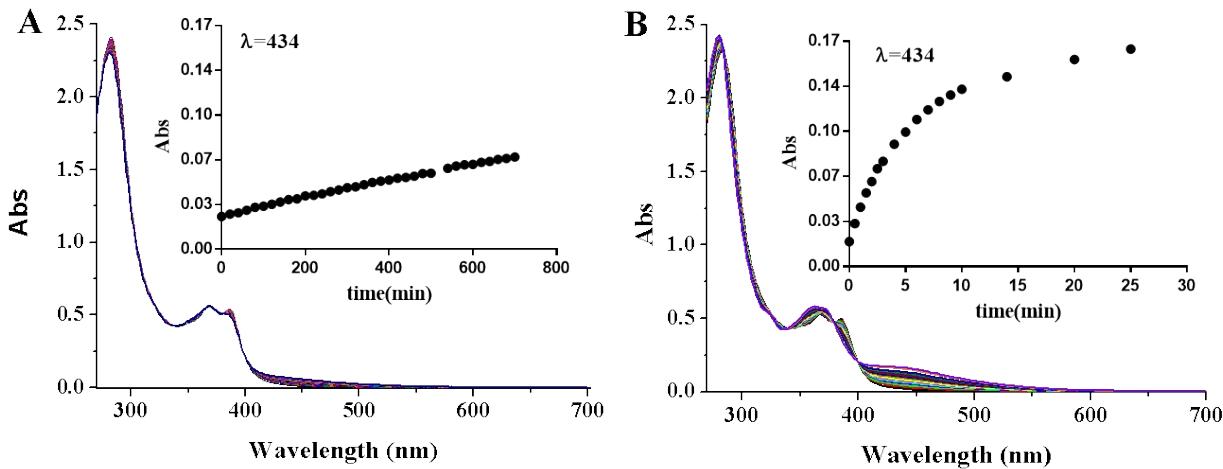


Figure S9. Monitoring spectroscopic changes of  $\text{cis}-[\text{RuCl}(\text{dppb})(\text{dppz})\text{CO}]^+$  in 20% DMF:phosphate buffer pH 7.4, in the dark at  $37^\circ\text{C}$  (A) and upon blue LED irradiation at  $25^\circ\text{C}$  (B).

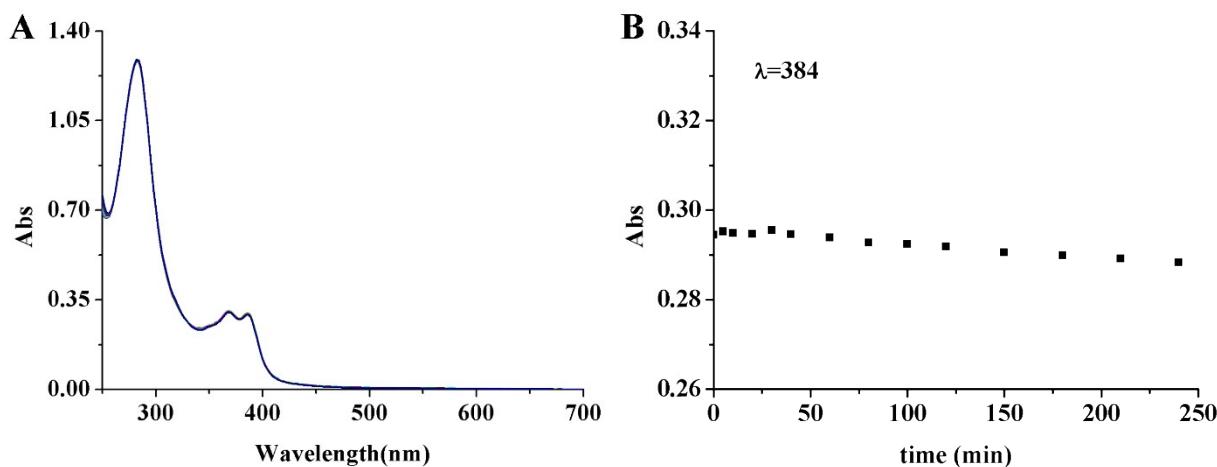


Figure S10. Reaction of glutathione ( $1 \text{ mmol L}^{-1}$ ) with  $\text{cis}-[\text{RuCl}(\text{dppb})(\text{dppz})\text{CO}]^+$  ( $20 \mu\text{mol L}^{-1}$ ) in 20% DMF:phosphate buffer pH 7.4 at  $25^\circ\text{C}$  monitored by UV-Vis spectroscopy. (A) shows the spectra during 4 hours and (B) the kinetic trace for changes at  $384 \text{ nm}$ .

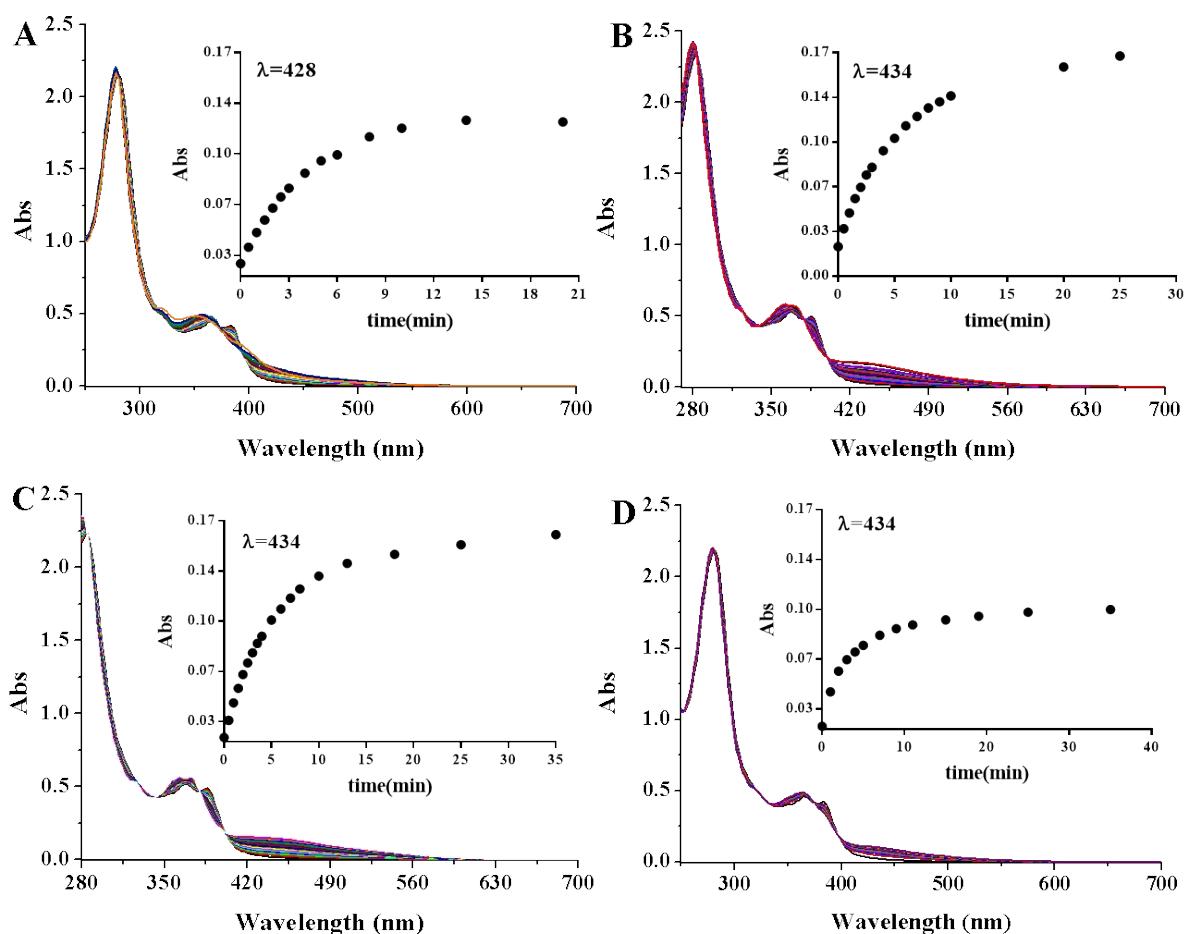


Figure S11. Spectroscopic changes of  $\text{cis}-[\text{RuCl}(\text{dppb})(\text{dppz})\text{CO}]^+$  in acetonitrile (A), dimethylformamide (B), dimethylsulfoxide (C), methanol (D) upon blue LED irradiation at 25 °C. Inset show kinetic plot during light irradiation.

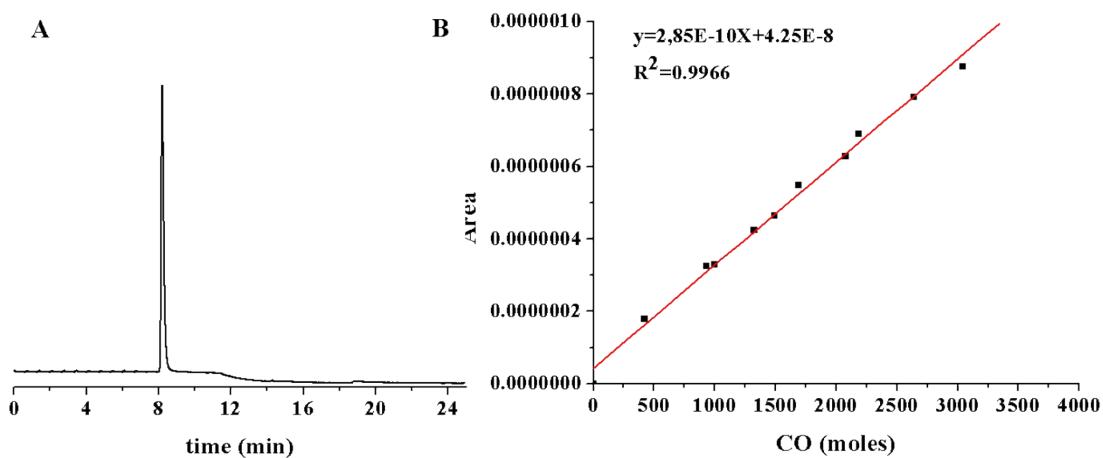


Figure S12. Gas chromatogram of the gas phase for UV irradiated sample of cis-[RuCl(CO)(dppb)(dppz)]<sup>+</sup> (peak with retention time at 8.20 min consistent with CO standard).

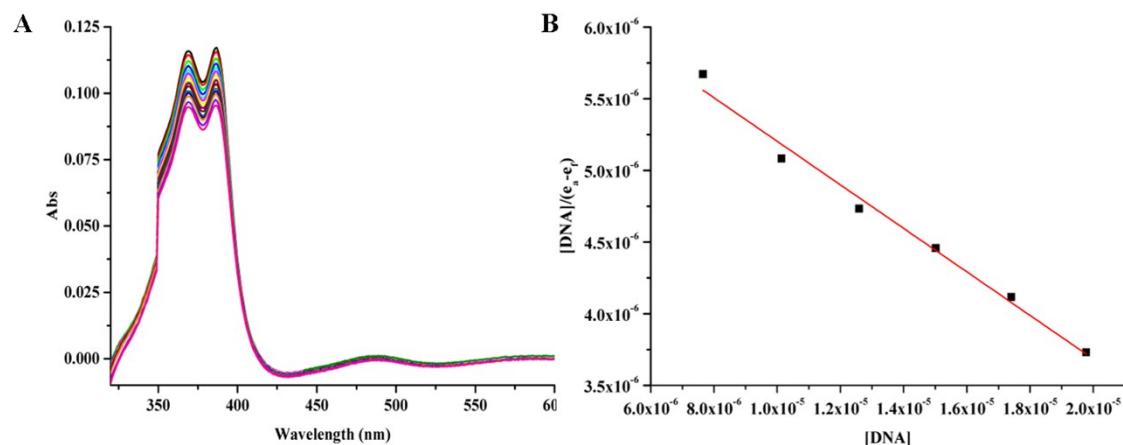


Figure S13. Calf thymus DNA (CT-DNA) titration for cis-[RuCl(CO)(dppb)(dppz)]<sup>+</sup> followed by electronic spectroscopy. Panel A shows spectral changes upon addition of CT-DNA and B fitting for data to obtain Kb.

**Table S1.** Selected experimental bands in the FTIR spectrum and the theoretical frequencies,  $\nu(\text{cm}^{-1})$  and vibrational assignments of *cis*-[RuCl(CO)(dppb)(dppz)]PF<sub>6</sub>.

<b>Experimental</b>	<b>DFT</b>	<b>Assignments</b>
518	498	Ring(dppb)
557	-	$\beta(\text{PF}_6)$
518	537	$\beta(\text{RuNO})$
696	690	Ring(dppb)
836	-	$\nu(\text{PF}_6)$
1077	1072	$\beta(\text{CH})$
1357	1347	$\nu(\text{CC}), \nu(\text{CN})$ dppz
1434	1433	$\beta(\text{CH})$ dppb
1492	1492	$\nu(\text{CC}), \nu(\text{CN})$ dppz
1993	2023	$\nu(\text{CO})$

Table S2. Selected UV–Vis Energy Transitions at the TD-DFT/B3LYP Level for *cis*-[RuCl(CO)(dppb)(dppz)]<sup>+</sup> in acetonitrile.

$\lambda_{\text{cal.}}$ (nm)	Oscillator strength ( $f$ )	Key transition	Character
379	0.0485	H-1 → LUMO (79%)	$d_{\text{Ru}}/\pi_{\text{Cl}} \rightarrow \pi^*_{\text{dppz}}$
370	0.0178	H-2 → LUMO (92%)	$\pi_{\text{dppz}} \rightarrow \pi^*_{\text{dppz}}$
361	0.0218	H-3 → LUMO (88%)	$\pi_{\text{dppb}}/d_{\text{Ru}} \rightarrow \pi^*_{\text{dppz}}$
346	0.1766	H-5 → LUMO (62%)	$\pi_{\text{dppz}}/\pi_{\text{dppb}}/d_{\text{Ru}} \rightarrow \pi^*_{\text{dppz}}$
294	0.3054	H-8 → L+1 (19%)	$\pi_{\text{dppb}} \rightarrow \pi^*_{\text{dppz}}$
		H-2 → L+2 (24%)	$\pi_{\text{dppz}} \rightarrow \pi^*_{\text{dppz}}$
291	0.4327	H-9 → L+1 (35%)	$\pi_{\text{dppb}} \rightarrow \pi^*_{\text{dppz}}$
287	0.2184	H-5 → L+2 (66%)	$\pi_{\text{dppz}}/\pi_{\text{dppb}}/d_{\text{Ru}} \rightarrow \pi^*_{\text{dppz}}$
262	0.1764	H-14 → L+1 (42%)	$\pi_{\text{dppz}} \rightarrow \pi^*_{\text{dppz}}$
253	0.1963	H-3 → L+4 (38%)	$\pi_{\text{dppb}}/d_{\text{Ru}} \rightarrow d_{\text{Ru}}/\pi_{\text{dppb}}/\pi_{\text{CO}}$

Table S3. Half-life for photolysis of the *cis*-[RuCl(CO)(dppb)(dppz)]<sup>+</sup> using blue LED in different solvents and mixtures at 25 °C.

Solvent	t <sub>1/2</sub> (min)
DMF	3.88
Methanol	2.36
DMSO	3.76
Acetonitrile	2.87
PBS/DMF	3.82
PBS/DMF-dark	1157.95