

Supporting Information

for

Molecular industry of cage metal complexes for biological applications: pathways of the synthesis, X-ray structure of a series of new N_2 -, S_2 - and O_2 - alicyclic iron(II) di- and tetrachloroclathrochelates

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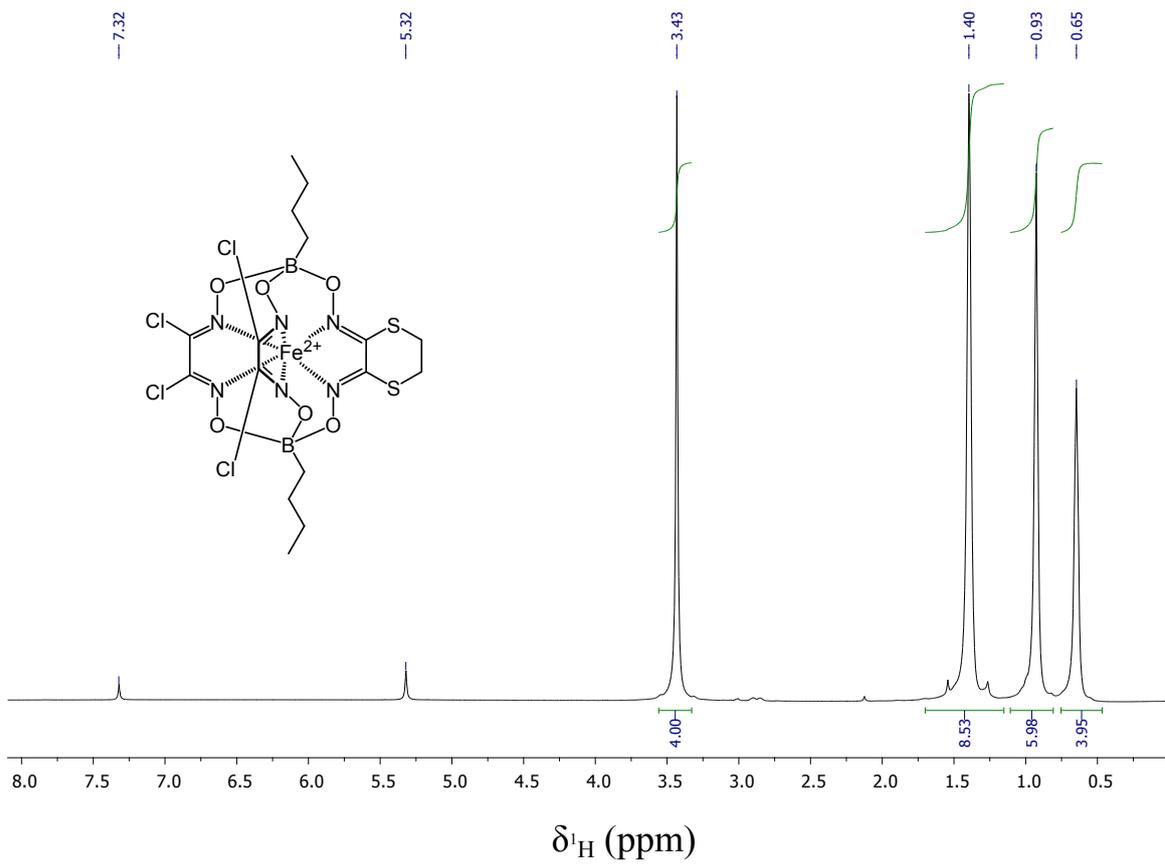


Figure S1. ^1H NMR spectrum for the CD_2Cl_2 solution of the clathrochelate $\text{Fe}(\text{S}_2\text{-Nx})(\text{Cl}_2\text{Gm})_2(\text{Bn-C}_4\text{H}_9)_2$ (**1**).

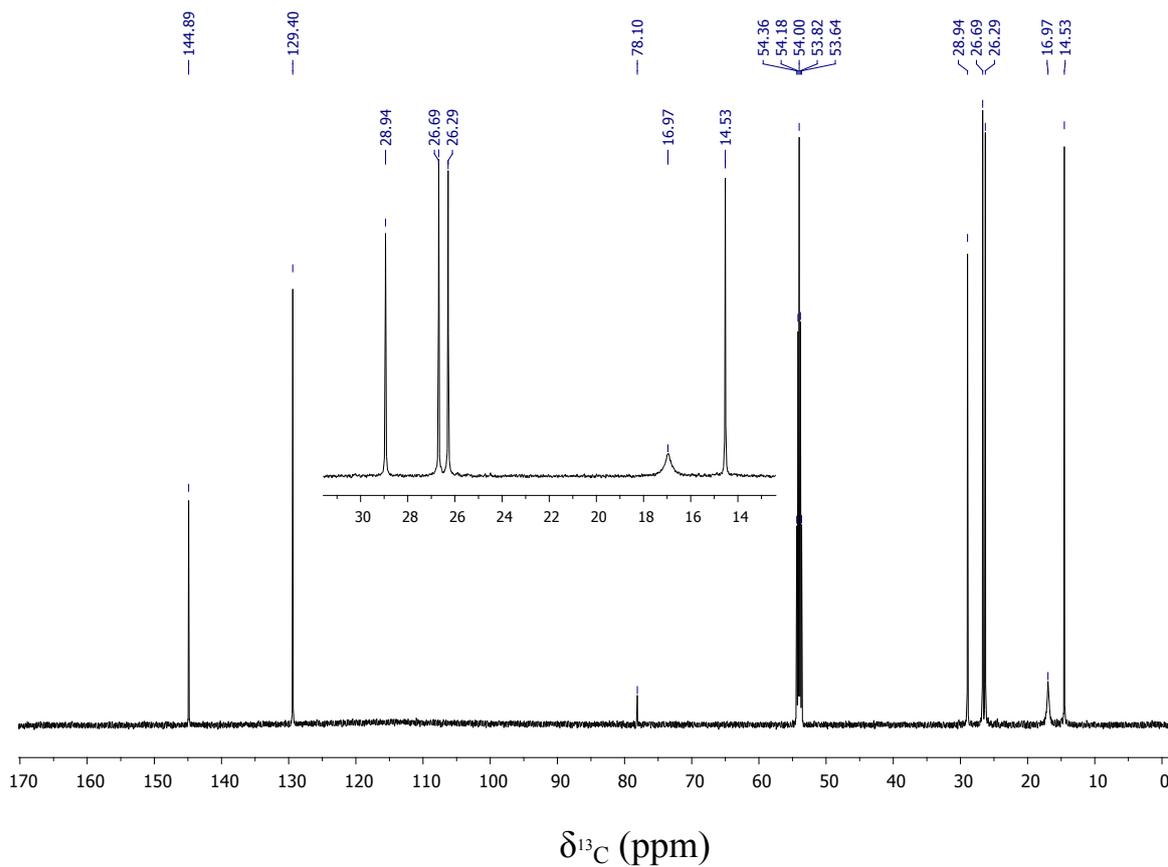


Figure S2. $^{13}\text{C}\{^1\text{H}\}$ NMR spectrum for the CD_2Cl_2 solution of the clathrochelate $\text{Fe}(\text{S}_2\text{-Nx})(\text{Cl}_2\text{Gm})_2(\text{Bn-C}_4\text{H}_9)_2$ (**1**).

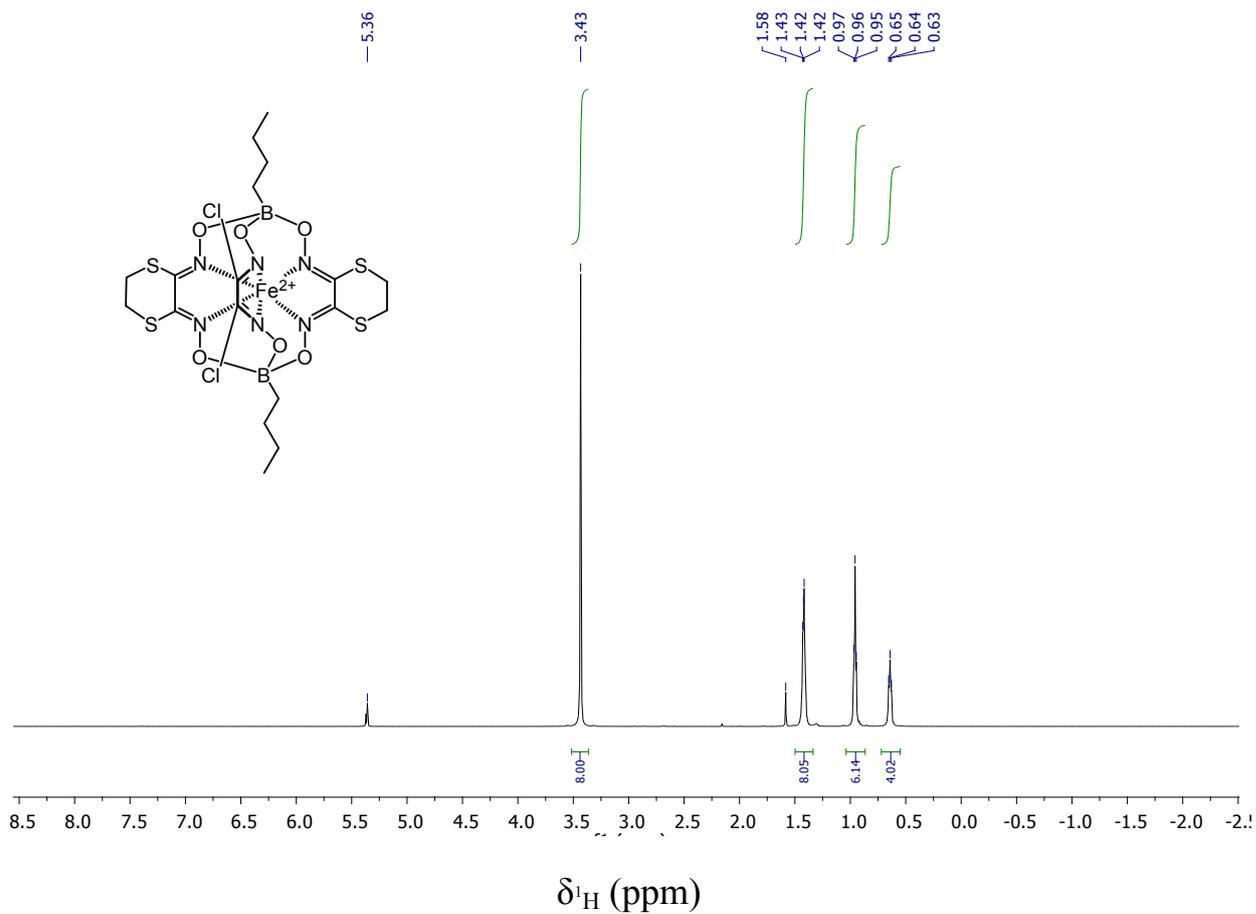


Figure S3. ^1H NMR spectrum for the CD_2Cl_2 solution of the clathrochelate $\text{Fe}(\text{S}_2\text{-Nx})_2(\text{Cl}_2\text{Gm})(\text{Bn}-\text{C}_4\text{H}_9)_2$ (**2**).

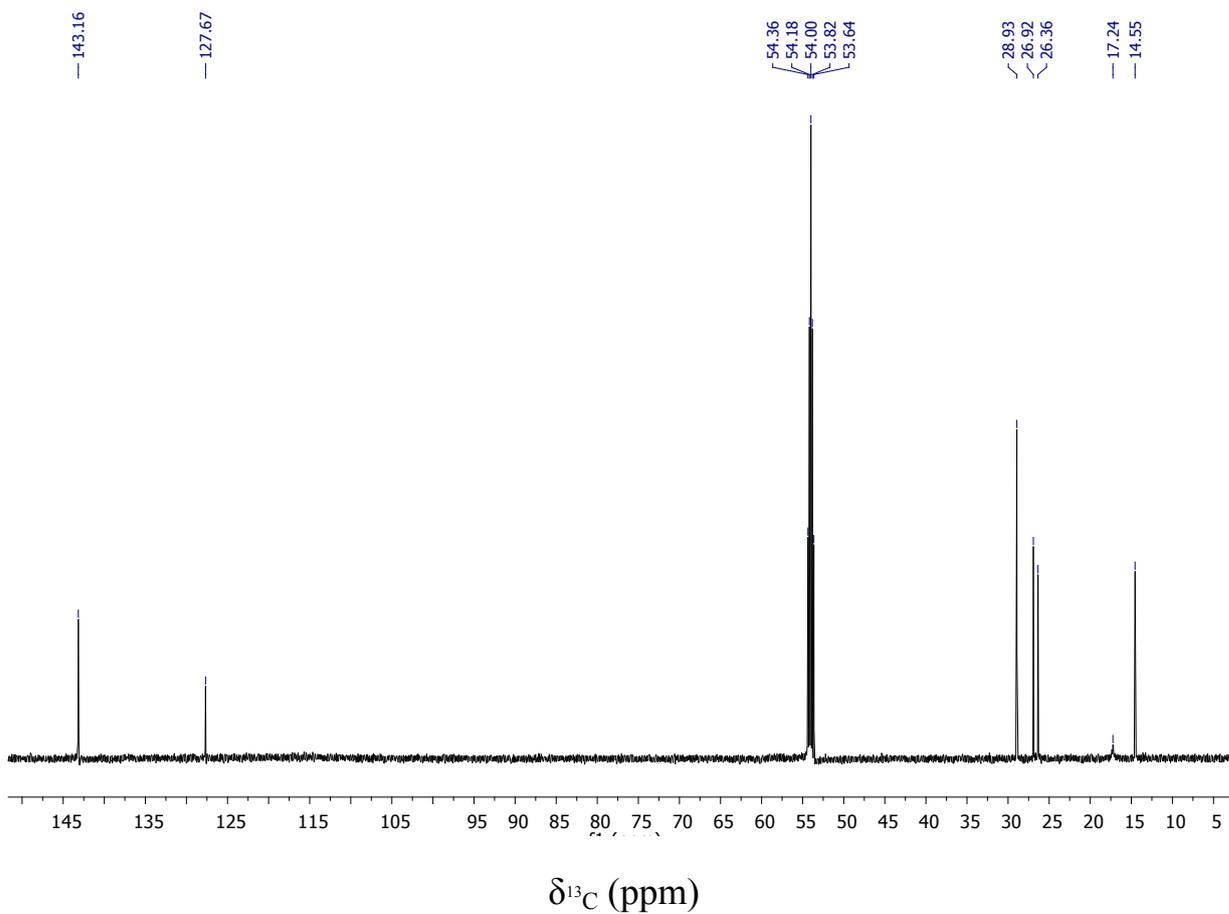


Figure S4. $^{13}\text{C}\{^1\text{H}\}$ NMR spectrum for the CD_2Cl_2 solution of the clathrochelate $\text{Fe}(\text{S}_2\text{-Nx})_2(\text{Cl}_2\text{Gm})(\text{Bn-C}_4\text{H}_9)_2$ (**2**).

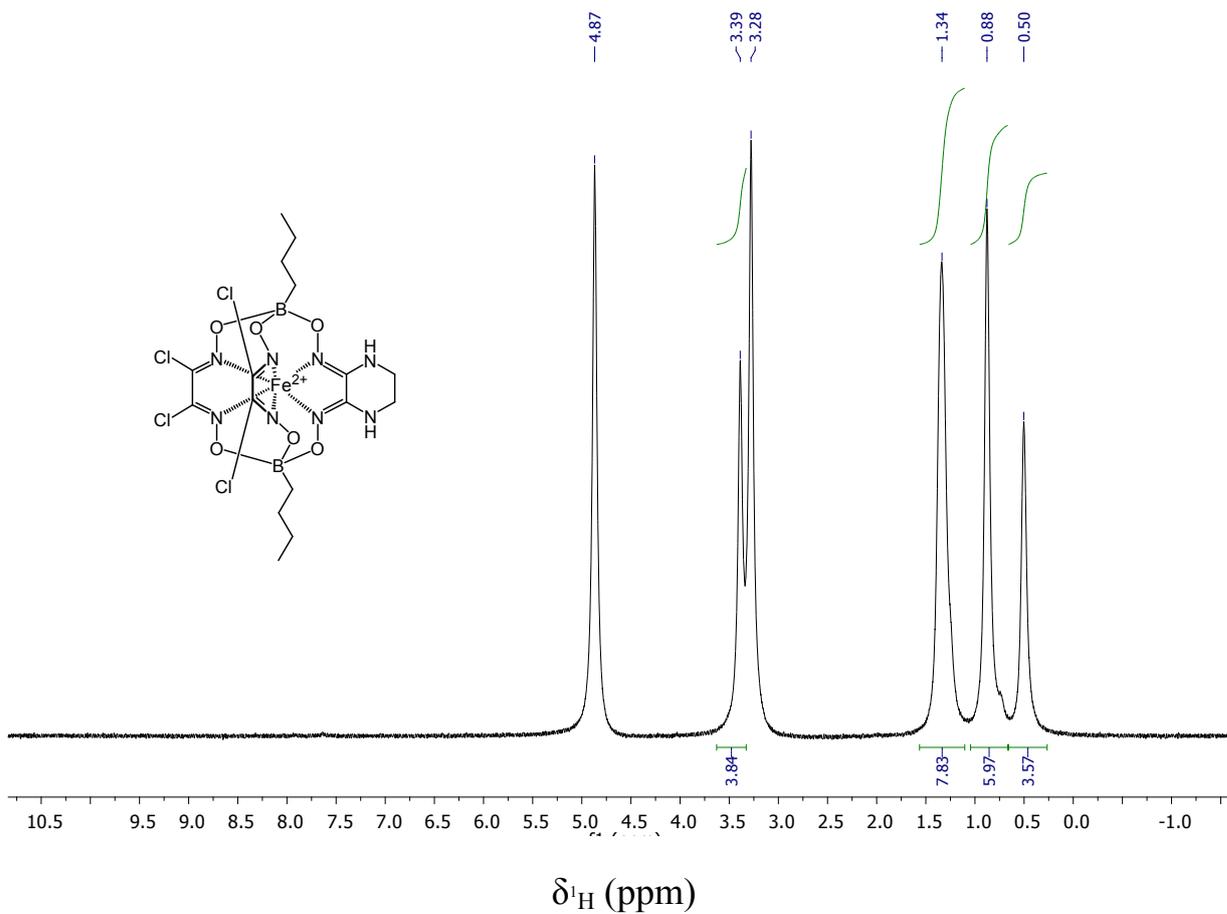


Figure S5. ^1H NMR spectrum for the CD_3OD solution of the clathrochelate $\text{Fe}(\text{N}_2\text{-N}_x)(\text{Cl}_2\text{Gm})_2(\text{Bn-C}_4\text{H}_9)_2$ (**3**).

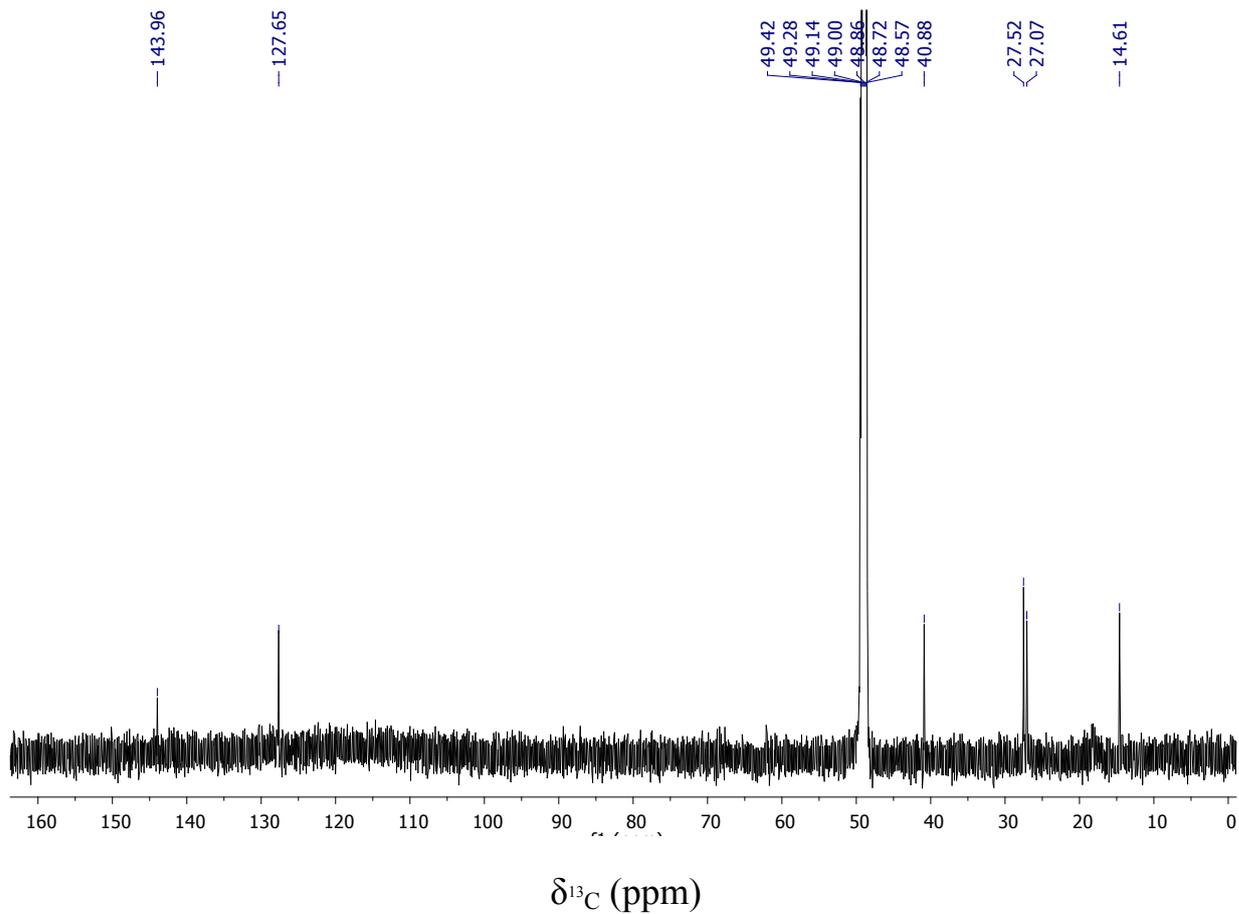


Figure S6. $^{13}\text{C}\{^1\text{H}\}$ NMR spectrum for the CD_3OD solution of the clathrochelate $\text{Fe}(\text{N}_2\text{-Nx})(\text{Cl}_2\text{Gm})_2(\text{Bn-C}_4\text{H}_9)_2$ (**3**).

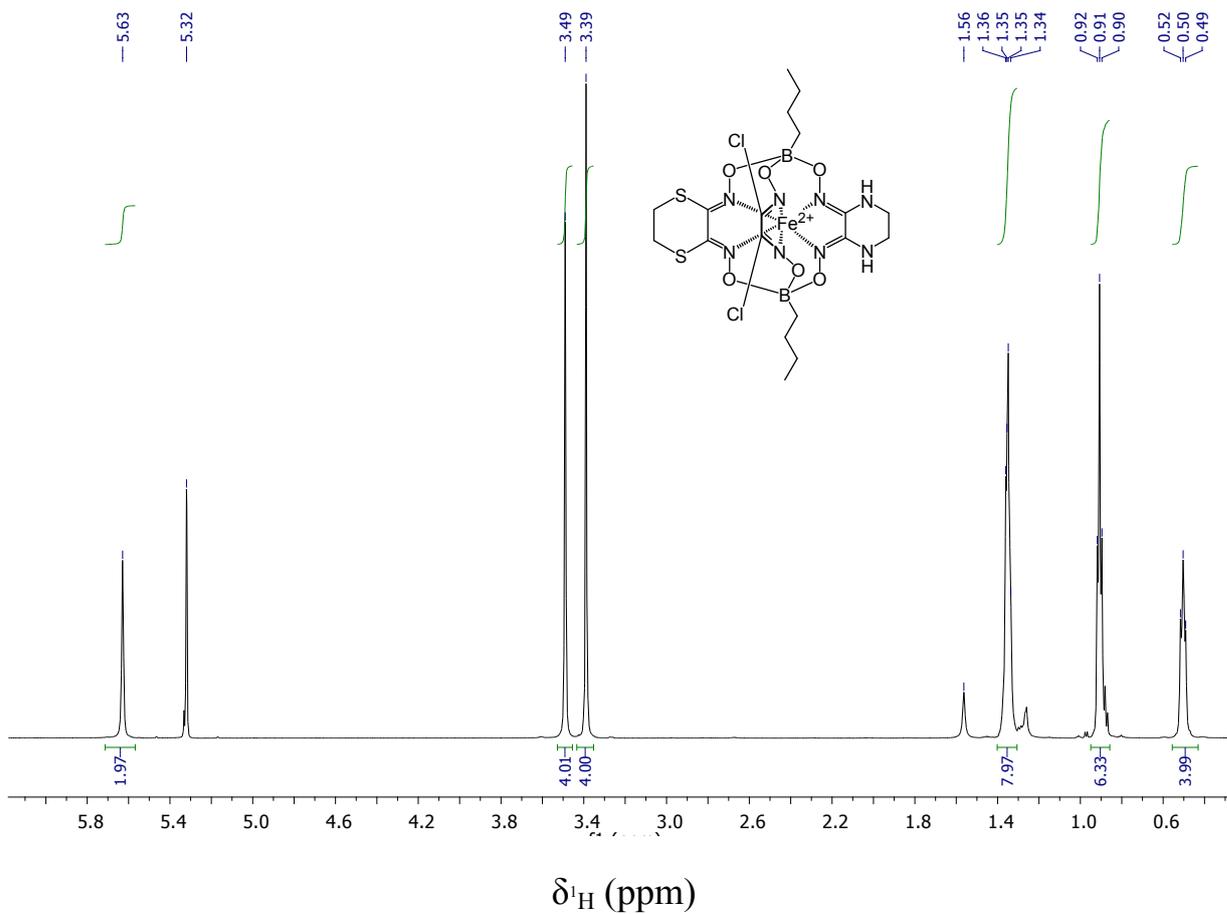


Figure S7. ¹H NMR spectrum for the CD₂Cl₂ solution of the clathrochelate Fe(N₂-Nx)(S₂-Nx)(Cl₂Gm)(Bn-C₄H₉)₂ (**4**).

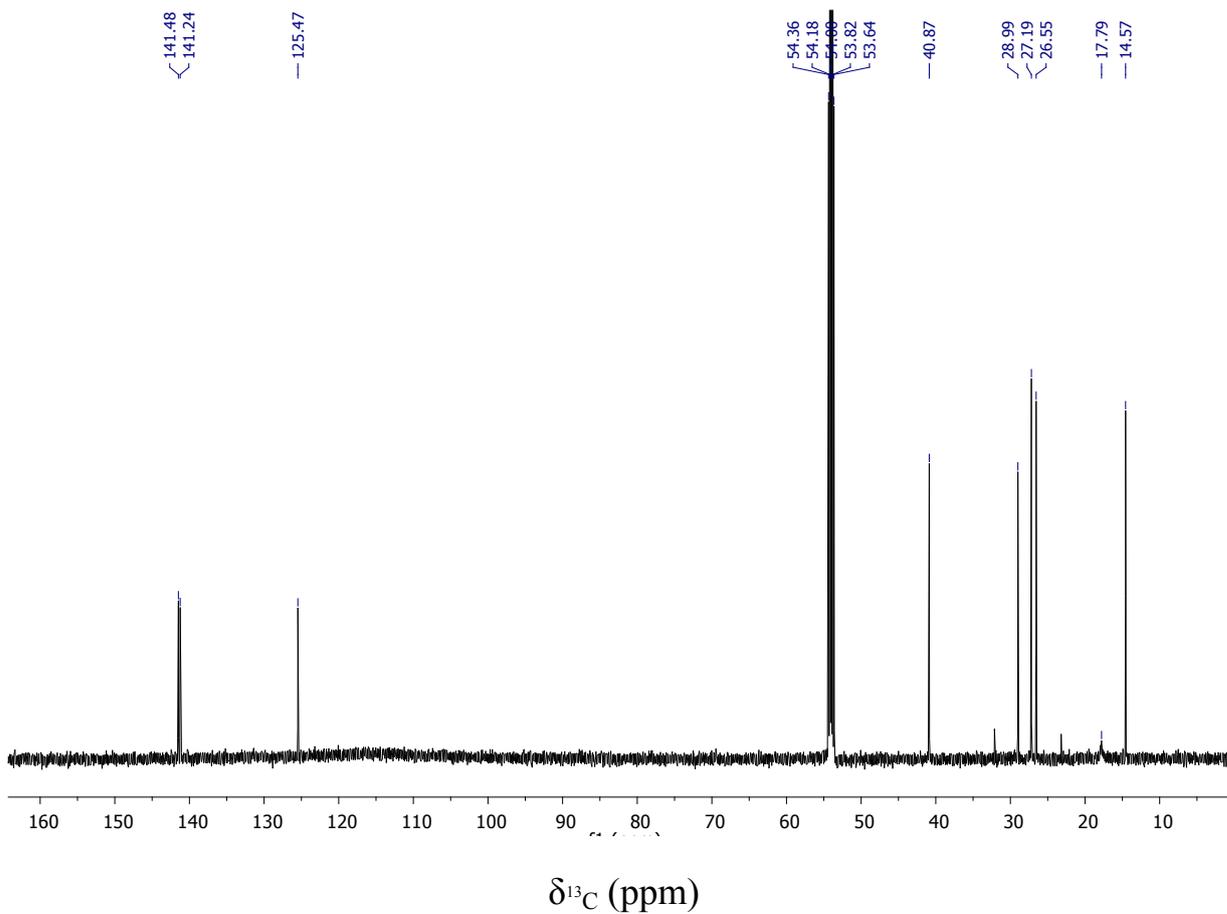


Figure S8. $^{13}\text{C}\{^1\text{H}\}$ NMR spectrum for the CD_2Cl_2 solution of the clathrochelate $\text{Fe}(\text{N}_2\text{-N}_x)(\text{S}_2\text{-N}_x)(\text{Cl}_2\text{Gm})(\text{Bn-C}_4\text{H}_9)_2$ (**4**).

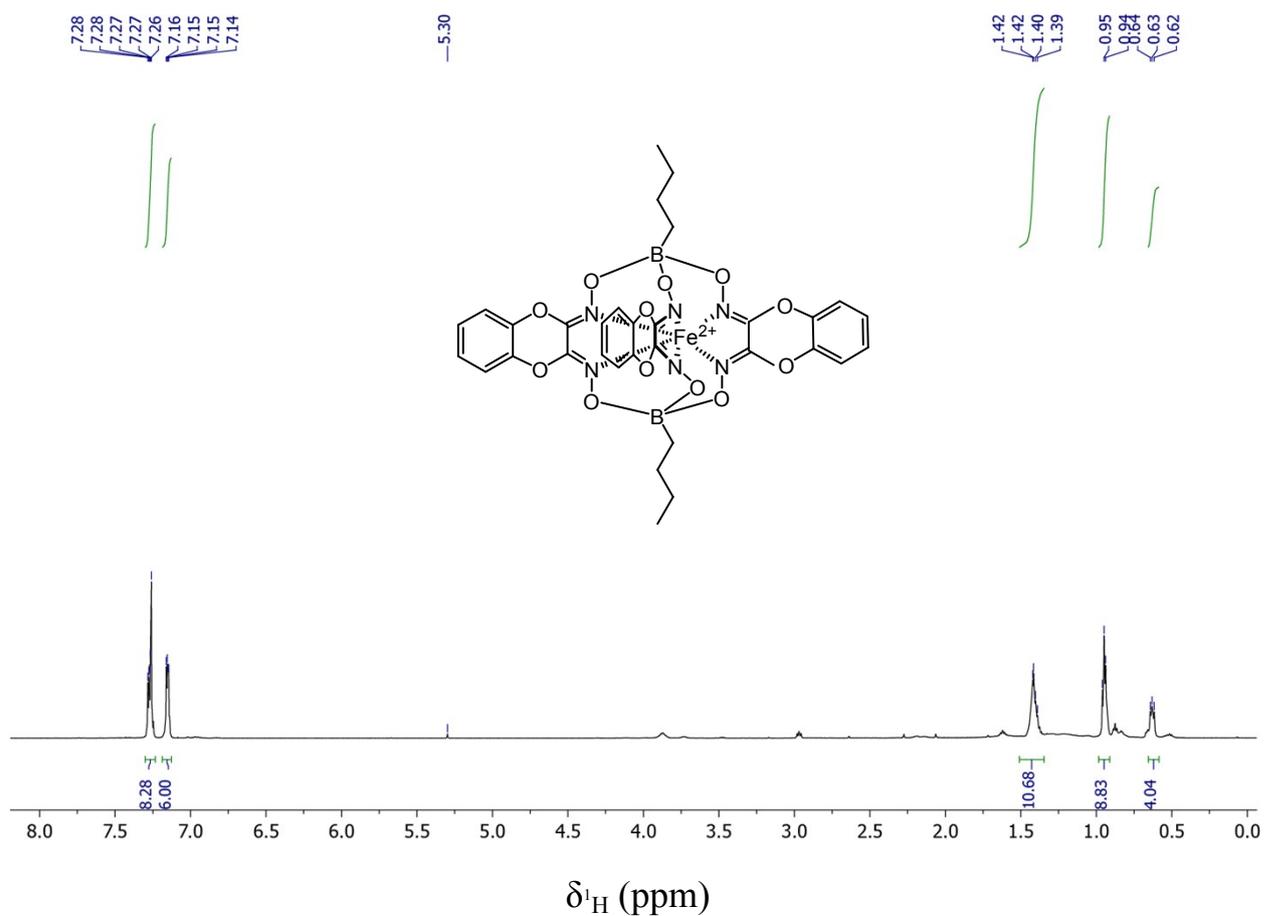


Figure S9. ^1H NMR spectrum for the CDCl_3 solution of the clathrochelate $\text{Fe}(\text{PrchGm})_3(\text{Bn-C}_4\text{H}_9)_2$ (**5**).

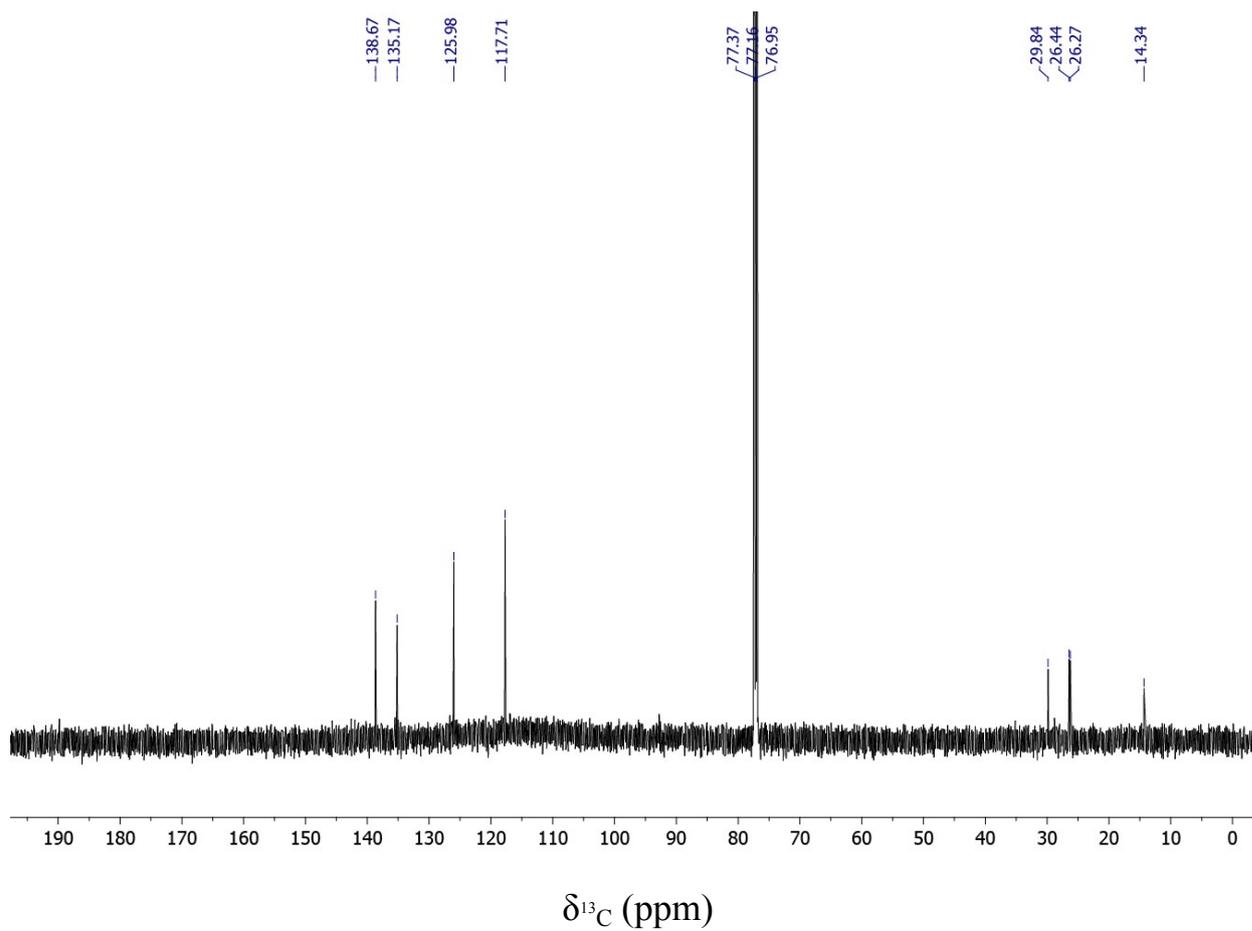


Figure S10. $^{13}\text{C}\{^1\text{H}\}$ NMR spectrum for the CDCl_3 solution of the clathrochelate $\text{Fe}(\text{PrchGm})_3(\text{Bn}-\text{C}_4\text{H}_9)_2$ (**5**).

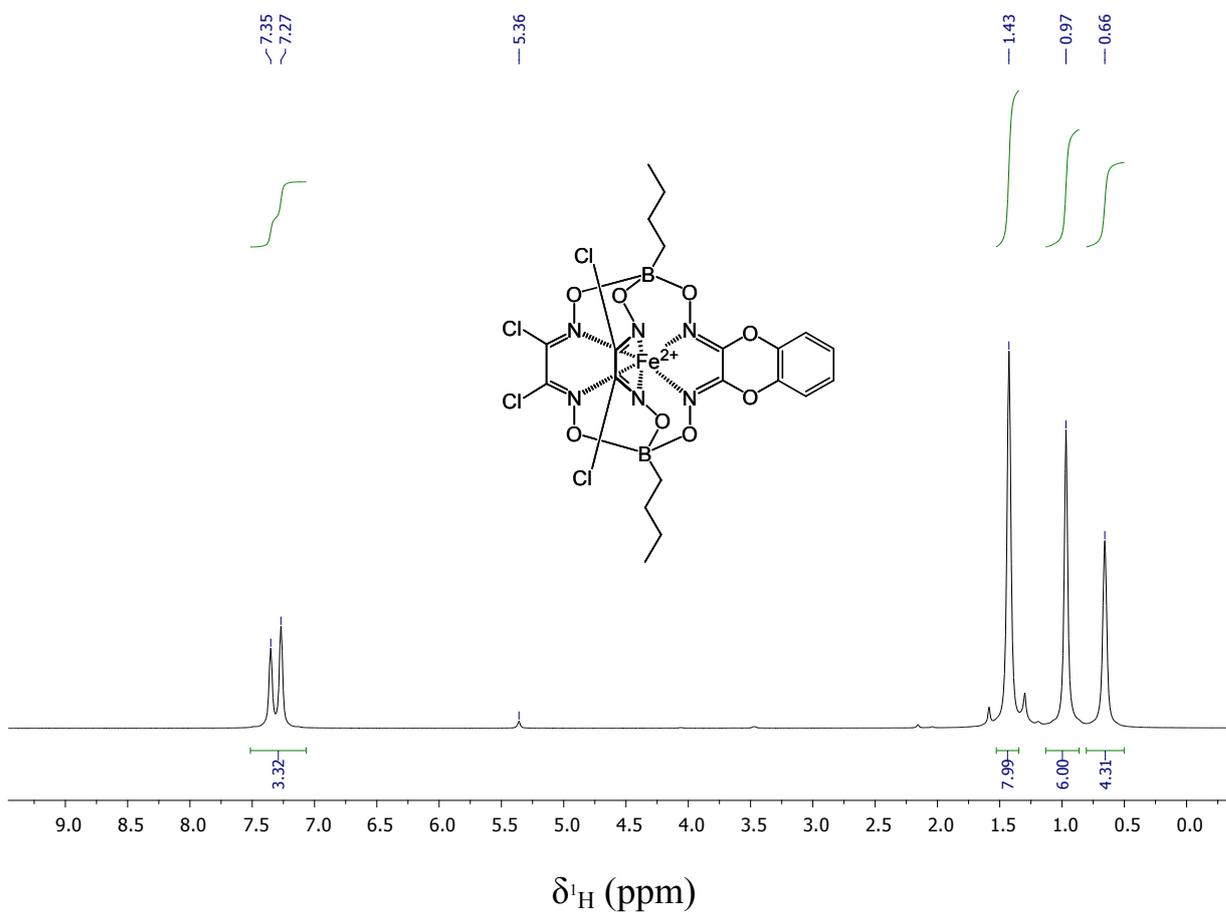


Figure S11. ^1H NMR spectrum for the CD_2Cl_2 solution of the clathrochelate $\text{Fe}(\text{Cl}_2\text{Gm})_2(\text{PrchGm})(\text{Bn}-\text{C}_4\text{H}_9)_2$ (**6**).

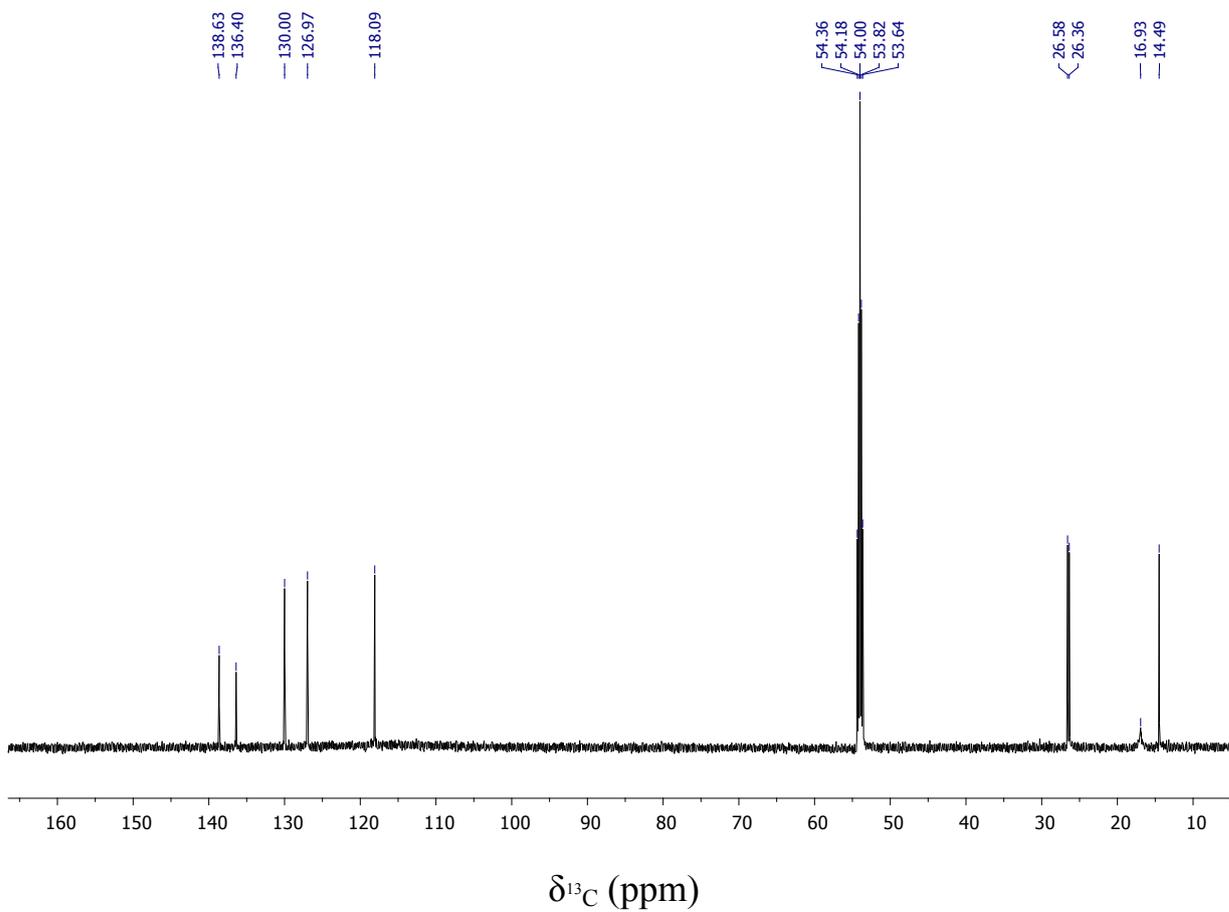


Figure S12. $^{13}\text{C}\{^1\text{H}\}$ NMR spectrum for the CD_2Cl_2 solution of the clathrochelate $\text{Fe}(\text{Cl}_2\text{Gm})_2(\text{PrchGm})(\text{Bn}-\text{C}_4\text{H}_9)_2$ (**6**).

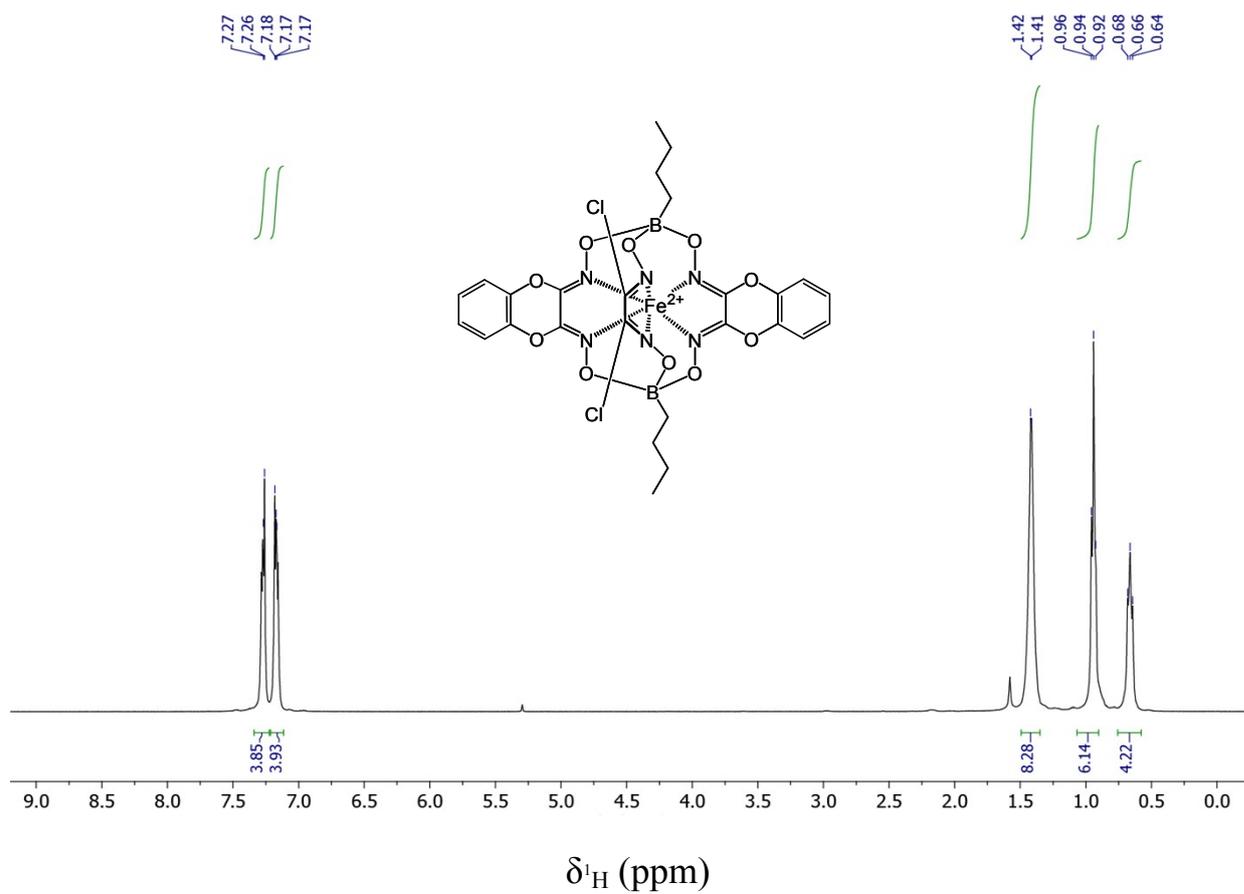


Figure S13. ^1H NMR spectrum for the CDCl_3 solution of the clathrochelate $\text{Fe}(\text{Cl}_2\text{Gm})(\text{PrchGm})_2(\text{Bn-C}_4\text{H}_9)_2$ (**7**).

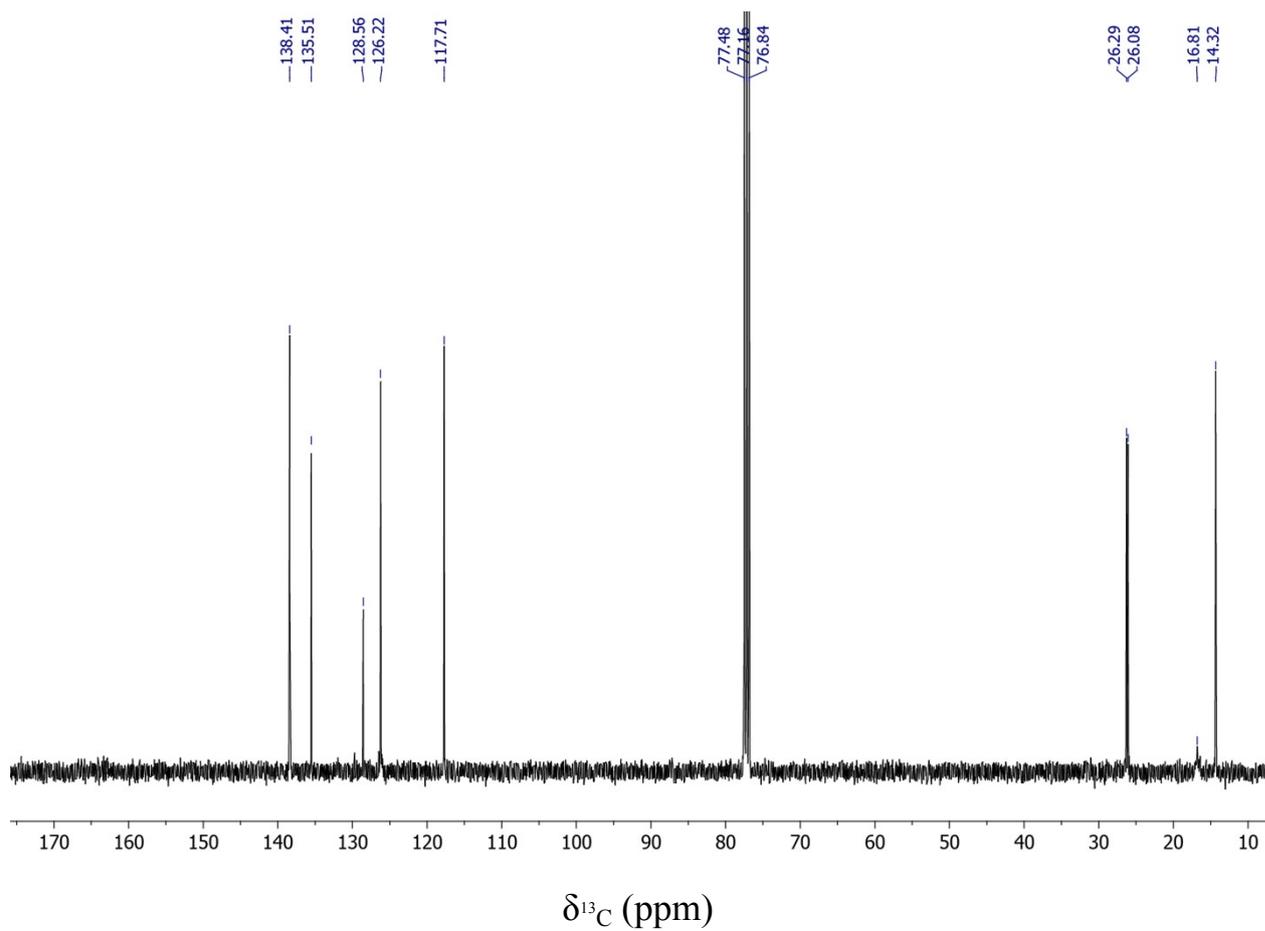


Figure S14. $^{13}\text{C}\{^1\text{H}\}$ NMR spectrum for the CDCl_3 solution of the clathrochelate $\text{Fe}(\text{Cl}_2\text{Gm})(\text{PrchGm})_2(\text{Bn}-\text{C}_4\text{H}_9)_2$ (**7**).

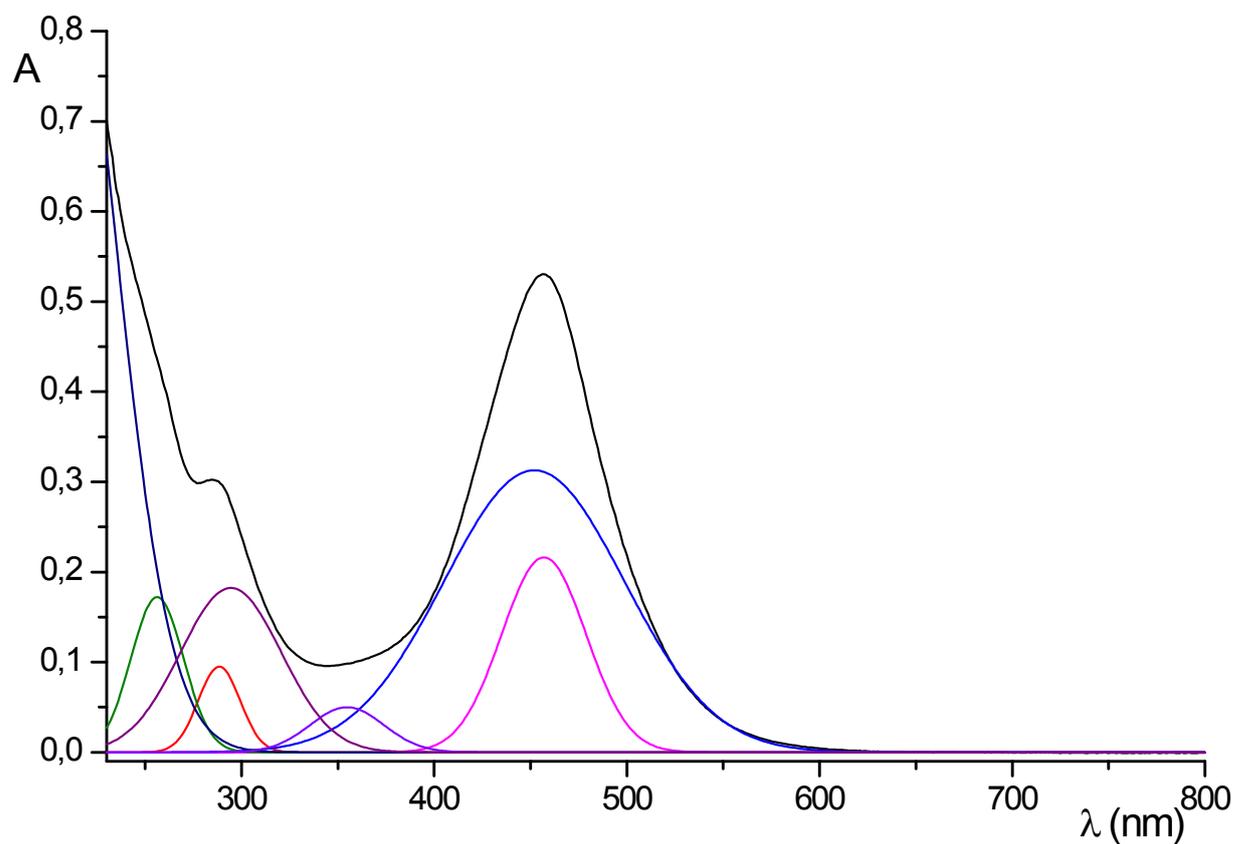


Figure S15. UV-Vis spectrum for the dichloromethane solution of the clathrochelate $\text{Fe}(\text{S}_2\text{-Nx})(\text{Cl}_2\text{Gm})_2(\text{Bn-C}_4\text{H}_9)_2$ (**1**) and its deconvolution on the Gaussian components.

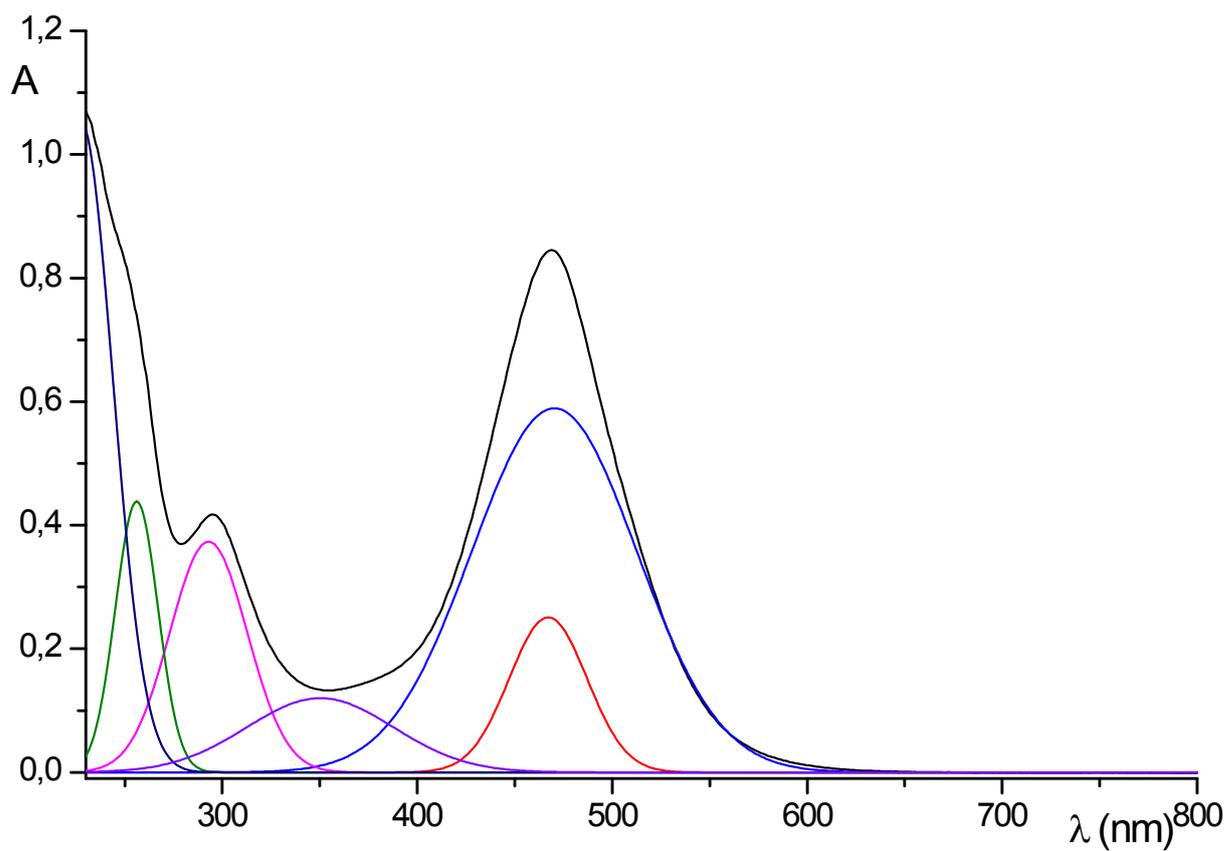


Figure S16. UV-Vis spectrum for the dichloromethane solution of the clathrochelate $\text{Fe}(\text{S}_2\text{-Nx})_2(\text{Cl}_2\text{Gm})(\text{Bn-C}_4\text{H}_9)_2$ (**2**) and its deconvolution on the Gaussian components.

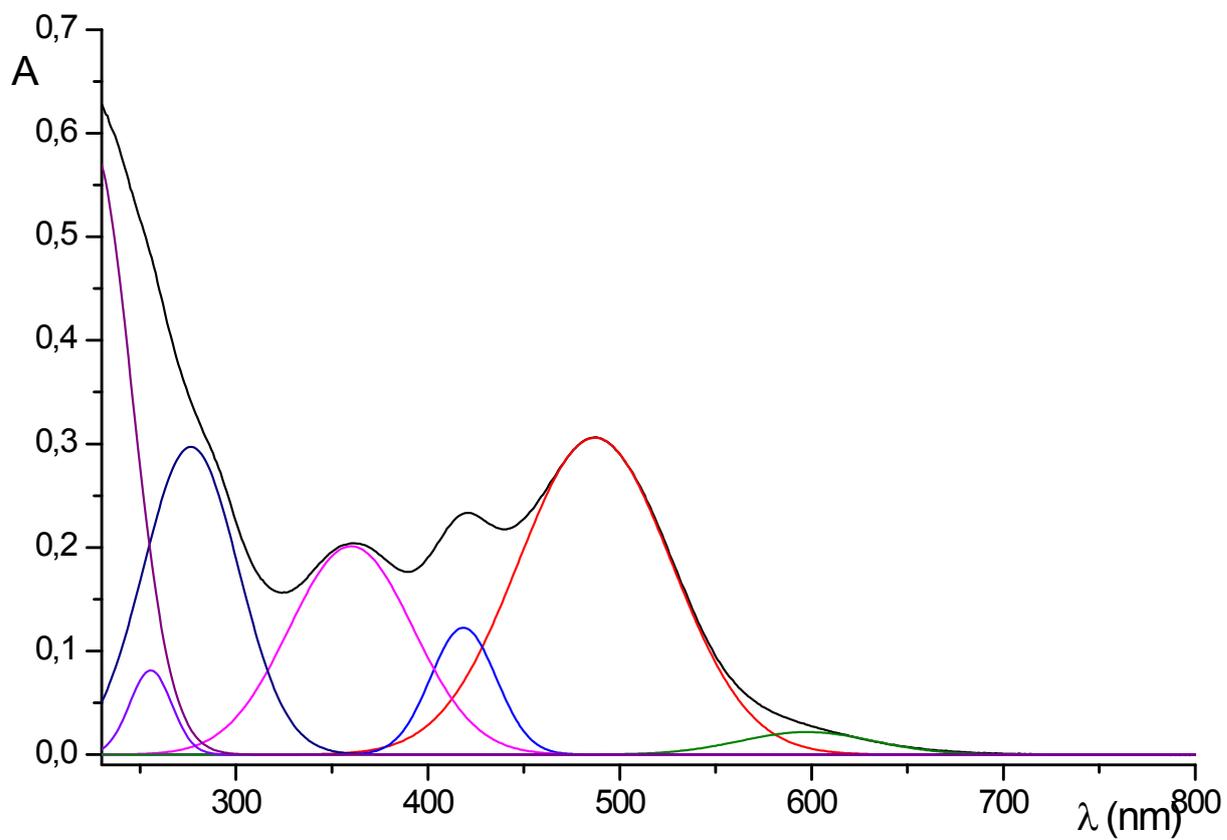


Figure S17. UV-Vis spectrum for the dichloromethane solution of the clathrochelate $\text{Fe}(\text{N}_2\text{-N}_x)(\text{Cl}_2\text{Gm})_2(\text{Bn-C}_4\text{H}_9)_2$ (**3**) and its deconvolution on the Gaussian components.

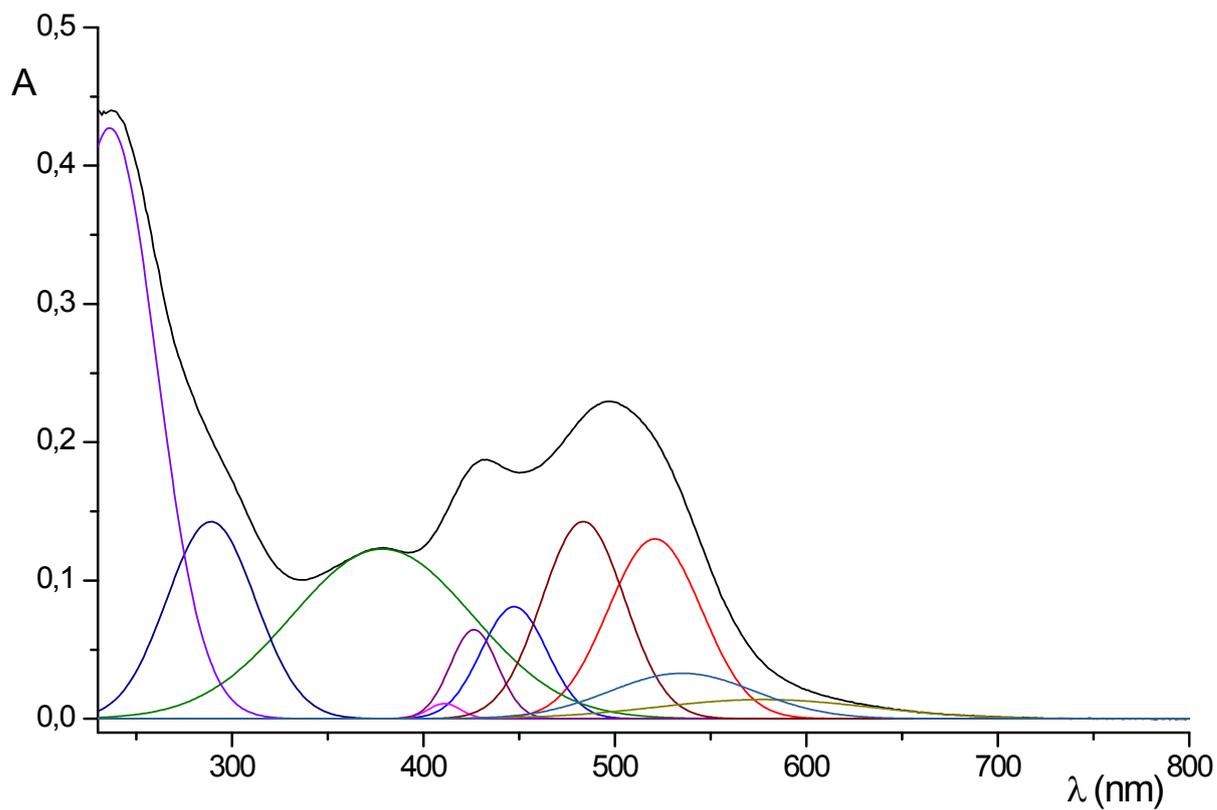


Figure S18. UV-Vis spectrum for the dichloromethane solution of the clathrochelate $\text{Fe}(\text{N}_2\text{-Nx})(\text{S}_2\text{-Nx})(\text{Cl}_2\text{Gm})(\text{Bn-C}_4\text{H}_9)_2$ (**4**) and its deconvolution on the Gaussian components.

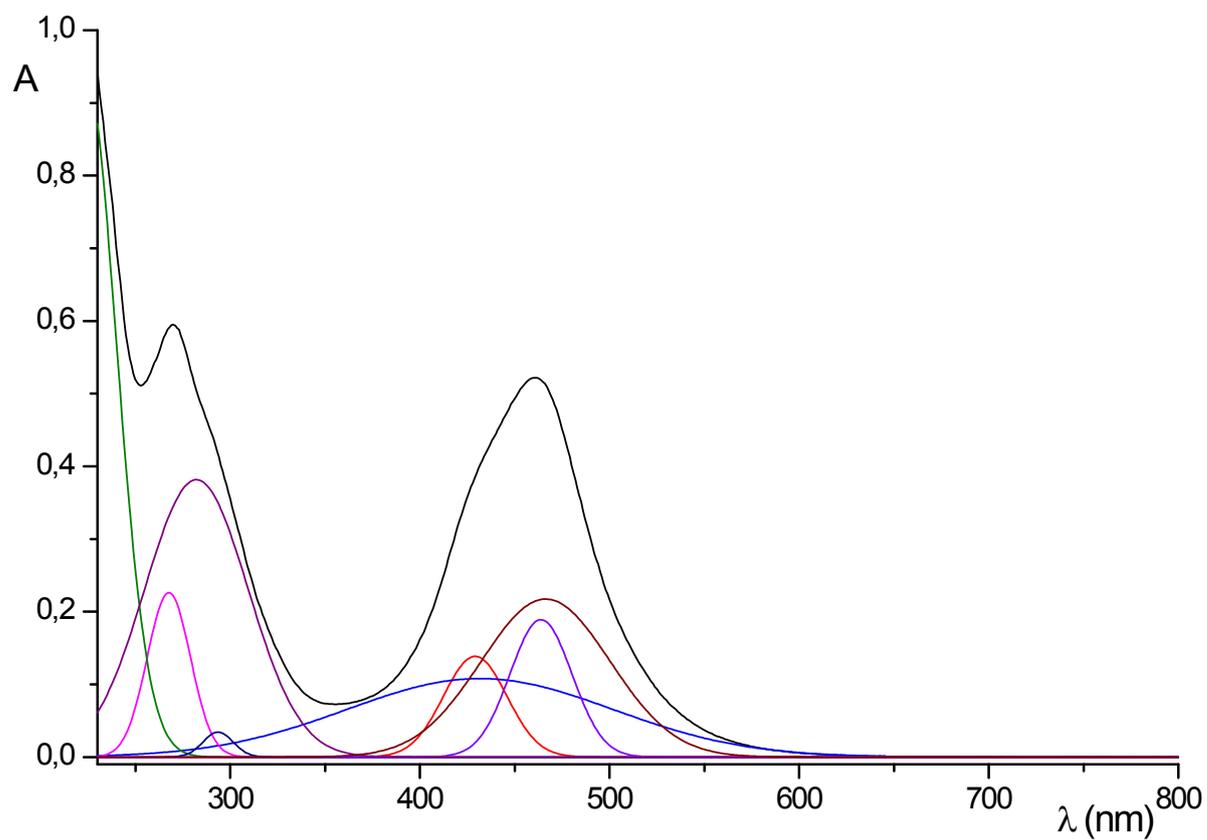


Figure S19. UV-Vis spectrum for the dichloromethane solution of the clathrochelate $\text{Fe}(\text{PrchGm})_3(\text{Bn-C}_4\text{H}_9)_2$ (**5**) and its deconvolution on the Gaussian components.

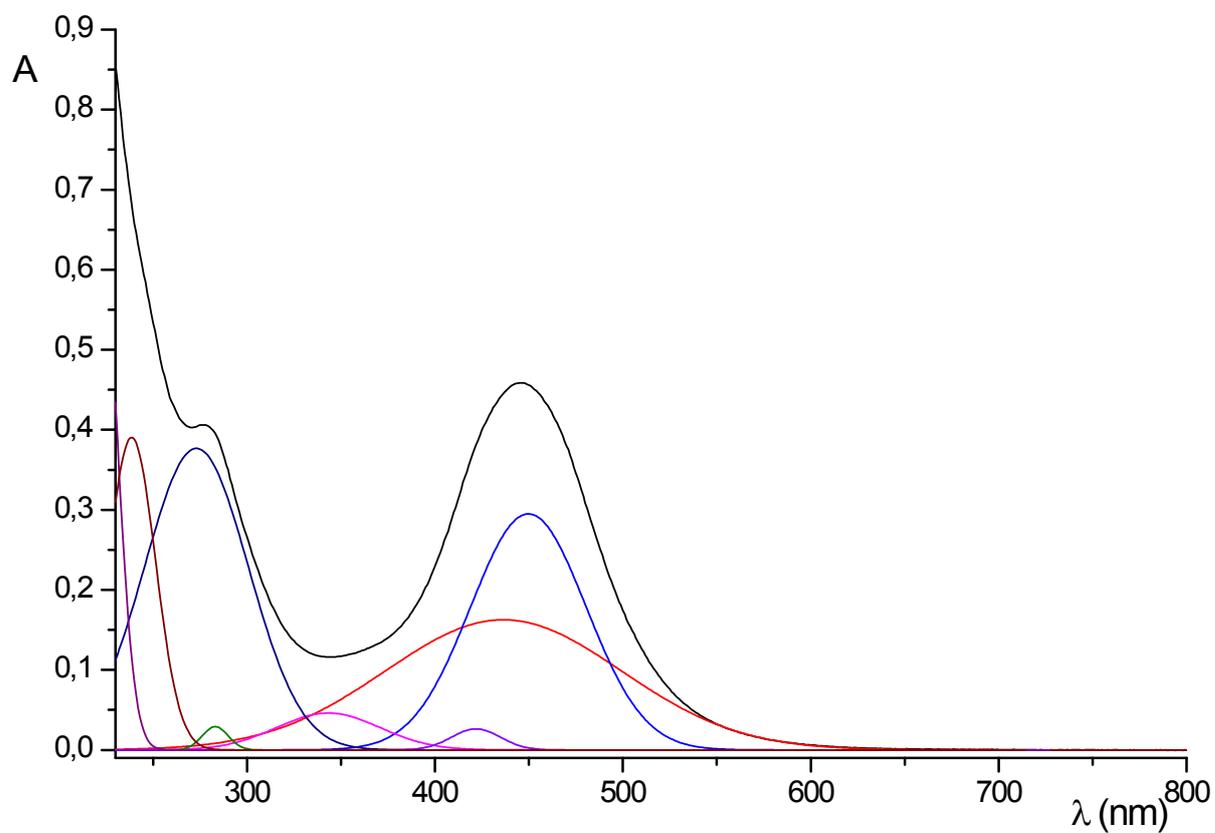


Figure S20. UV-Vis spectrum for the dichloromethane solution of the clathrochelate $\text{Fe}(\text{Cl}_2\text{Gm})_2(\text{PrchGm})(\text{Bn}-\text{C}_4\text{H}_9)_2$ (**6**) and its deconvolution on the Gaussian components.

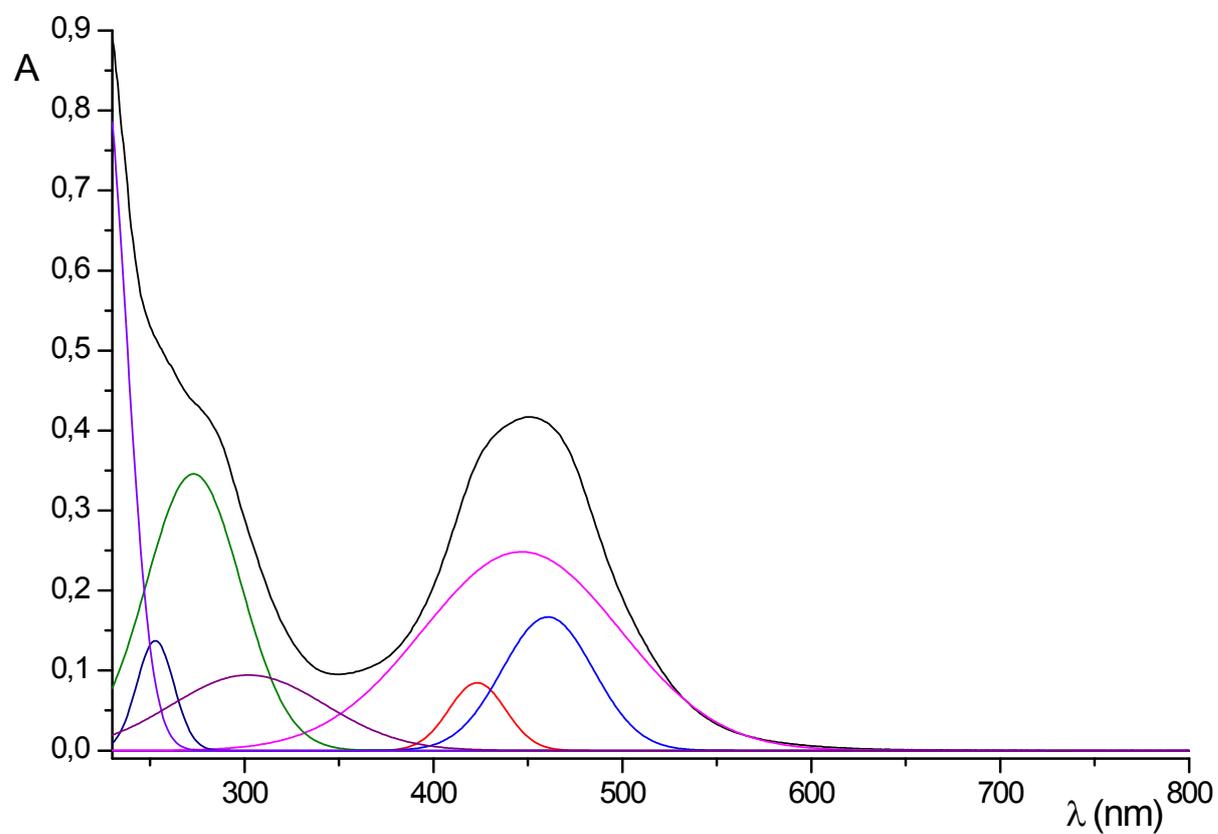


Figure S21. UV-Vis spectrum for the dichloromethane solution of the clathrochelate $\text{Fe}(\text{Cl}_2\text{Gm})(\text{PrchGm})_2(\text{Bn}-\text{C}_4\text{H}_9)_2$ (**7**) and its deconvolution on the Gaussian components.

Table S1. Crystallographic data and refinement parameters for the crystals $\text{Fe}(\text{Cl}_2\text{Gm})_2(\text{N}_2\text{-Nx})(\text{Bn-C}_4\text{H}_9)_2$, $1.5 \text{Fe}(\text{S}_2\text{-Nx})_2(\text{Cl}_2\text{Gm})(\text{Bn-C}_4\text{H}_9)_2 \cdot 0.5 \text{Fe}(\text{S}_2\text{-Nx})_3(\text{Bn-C}_4\text{H}_9)_2$ and $\text{Fe}(\text{PrchGm})_3(\text{Bn-C}_4\text{H}_9)_2$

	$\text{Fe}(\text{Cl}_2\text{Gm})_2(\text{N}_2\text{-Nx})(\text{Bn-C}_4\text{H}_9)_2$	$1.5 \text{Fe}(\text{S}_2\text{-Nx})_2(\text{Cl}_2\text{Gm})(\text{Bn-C}_4\text{H}_9)_2 \cdot 0.5 \text{Fe}(\text{S}_2\text{-Nx})_3(\text{Bn-C}_4\text{H}_9)_2$	$\text{Fe}(\text{PrchGm})_3(\text{Bn-C}_4\text{H}_9)_2$
Empirical formula	$\text{C}_{16}\text{H}_{23}\text{B}_2\text{Cl}_4\text{FeN}_8\text{O}_6$	$\text{C}_{37}\text{H}_{54}\text{B}_4\text{Cl}_3\text{Fe}_2\text{N}_{12}\text{O}_{12}\text{S}_9$	$\text{C}_{32}\text{H}_{30}\text{B}_2\text{Fe N}_6\text{O}_{12}$
Fw	642.69	1408.75	768.09
T, K	120	120	120
Crystal system	monoclinic	triclinic	tetragonal
Space group	$P 2_1/c$	$P \bar{1}$	$P 4_1 2_1 2$
Z	4	2	4
a, Å	9.453(2)	14.4860(4)	11.0772(2)
b, Å	12.533(3)	15.3721(3)	11.0772(2)
c, Å	23.913(5)	15.7873(4)	26.6909(9)
α , °	90	117.4170(10)	90.00
β , °	115.700(4)	99.084(2)	90.00
γ , °	90	107.0350(10)	90.00
V, Å ³	2552.9(10)	2800.58(12)	3275.09(14)
d_{calc} (g·cm ⁻³)	1.672	1.671	1.558
μ (mm ⁻¹)	1.060	91.91	5.37
F(000)	6300	1446	1584
$2\theta_{\text{max}}$, °	58	135	52
Measured refl.	29746	35554	29809
Independent reflections (R_{int})	7798 (0.080)	9471 (0.045)	4981 (0.037)
Obs.refl./restraints/ parameters	5315 / 24 / 353	7854 / 14 / 720	4529 / 0 / 241
R , ^a % [$I > 2\sigma(I)$]	0.114	0.0506	0.0300
R_w , ^b %	0.255	0.1043	0.0772
GOF ^c	1.06	0.99	1.01
Residual electron density, eÅ ⁻³ ($d_{\text{min}}/d_{\text{max}}$)	1.013 / -1.056	1.005/-0.932	0.430/-0.437

$$^a R = \sum | |F_o| - |F_c| | / \sum |F_o| \quad ^b R_w = [\sum(w(F_o^2 - F_c^2)^2) / \sum(w(F_o^2))]^{1/2} \quad ^c \text{GOF} = [\sum w(F_o^2 - F_c^2)^2 / (N_{\text{obs}} - N_{\text{param}})]^{1/2}$$

Table S2. UV-vis spectra ($\lambda_{\text{max}}/\text{nm}$, $\varepsilon \cdot 10^{-3} \text{ mol}^{-1} \text{ Lcm}^3$) for the triribed-functionalized iron(II) clathrochelates

Compound	λ_1	λ_2	λ_3	λ_4	λ_5	λ_6	λ_7	λ_8	λ_9	λ_{10}	λ_{11}
Fe(S₂-Nx)(Cl₂Gm)₂(Bn-C₄H₉)₂	256(5.5)	288(3.0)	294(5.8)	355(1.6)			452(10)	457(6.9)			
Fe(S₂-Nx)₂(Cl₂Gm)(Bn-C₄H₉)₂	256(13)		293(11)	350(3.6)			467(7.5)	470(18)			
Fe(N₂-Nx)(Cl₂Gm)₂(Bn-C₄H₉)₂	256(2.7)		277(10)	360(6.7)			487(10)				597(0.7)
Fe(N₂-Nx)(S₂-Nx)(Cl₂Gm)(Bn-C₄H₉)₂	236(13)	289(4.2)		378(3.6)	411(0.3)	426(1.9)	447(2.4)	484(4.2)	521(3.8)	535(1.0)	577(0.4)
Fe(PrchGm)₃(Bn-C₄H₉)₂	268(7.2)	282(12)	294(1.1)			429(4.4)	432(3.5)	464(6.0)	466(6.9)		
Fe(Cl₂Gm)₂(PrchGm)(Bn-C₄H₉)₂	239(13)	273(13)	283(1.0)	343(1.5)		422(0.9)	436(5.4)	450(9.8)			
Fe(Cl₂Gm)(PrchGm)₂(Bn-C₄H₉)₂	253(5.4)	273(13)	302(3.7)			423(3.3)	447(9.6)	461(6.5)			
Fe(Cl₂Gm)₃(Bn-C₄H₉)₂¹⁷		259(7.9)	285(5.4)	313(2.7)			423(4.8)	453 (15)			