

ELECTRONIC SUPPLEMENTARY INFORMATION

Investigation of cobalt(III)-tpa complexes as potential bioreductively activated carriers for naphthoquinone-based drugs

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Table S1. Comparison between calculated and experimental bond distances (Å) and angles (°) of $[\text{Co}(\text{TPA})(\text{NQ}-\text{CH}_3)]^{2+}$ and $[\text{Co}(\text{TPA})(\text{NQ}^1-\text{Cl})]^+$.

Distance/angle	$[\text{Co}(\text{TPA})(\text{NQ}-\text{CH}_3)]^{2+}$		$[\text{Co}(\text{TPA})(\text{NQ}^1-\text{Cl})]^+$	
	Exp.	Calc.	Exp.	Calc.
Co-O1	1.923	1.921	1.882	1.877
Co-O2	1.888	1.875	1.884	1.863
Co-N1	1.935	1.962	1.944	1.966
Co-N2	1.915	1.918	1.916	1.910
Co-N3	1.922	1.918	1.934	1.921
Co-N4	1.886	1.895	1.920	1.922
O2-Co-O1	84.92	86.02	87.22	89.05
N4-Co-N1	87.68	87.50	86.67	87.13
N4-Co-N3	88.82	90.48	90.29	90.35
N4-Co-N2	92.52	90.48	91.76	90.50
N1-Co-N2	85.36	85.64	85.76	86.16
N1-Co-N3	85.13	85.64	84.91	85.69
O1-Co-N1	92.68	92.69	93.12	90.97
O2-Co-N4	94.70	93.79	92.95	92.83
O1-Co-N3	89.44	89.53	87.67	88.60
O1-Co-N2	89.28	89.53	90.25	90.28

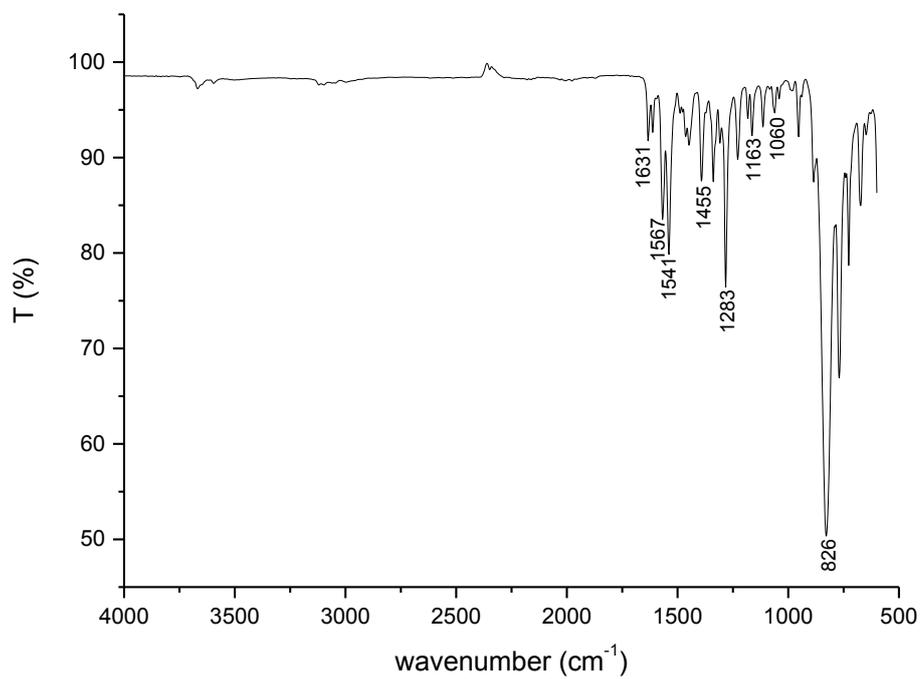


Figure S1. FTIR spectrum (ATR) of **1**.

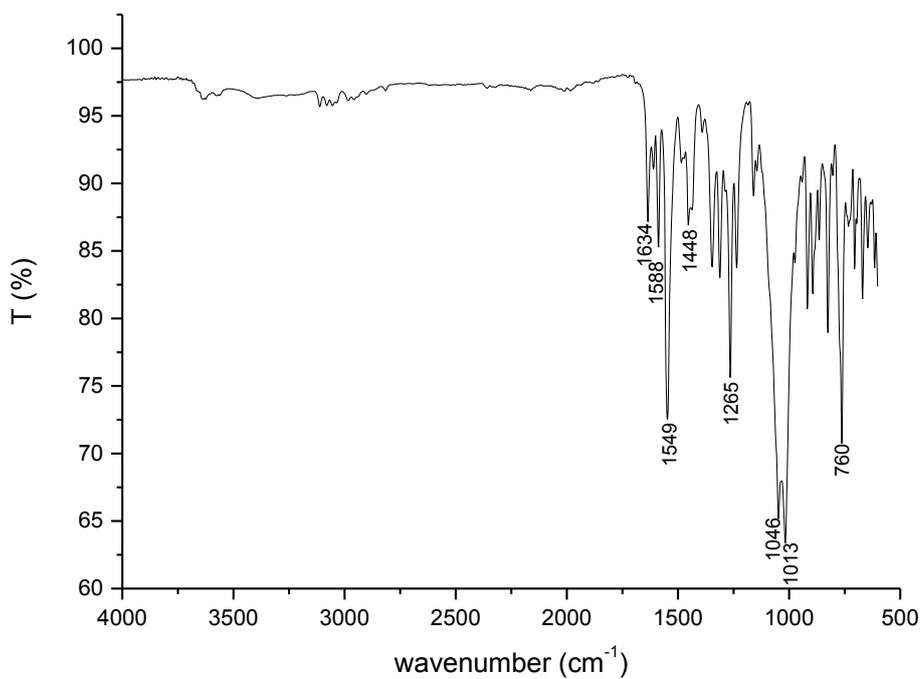


Figure S2. FTIR spectrum (ATR) of **2**.

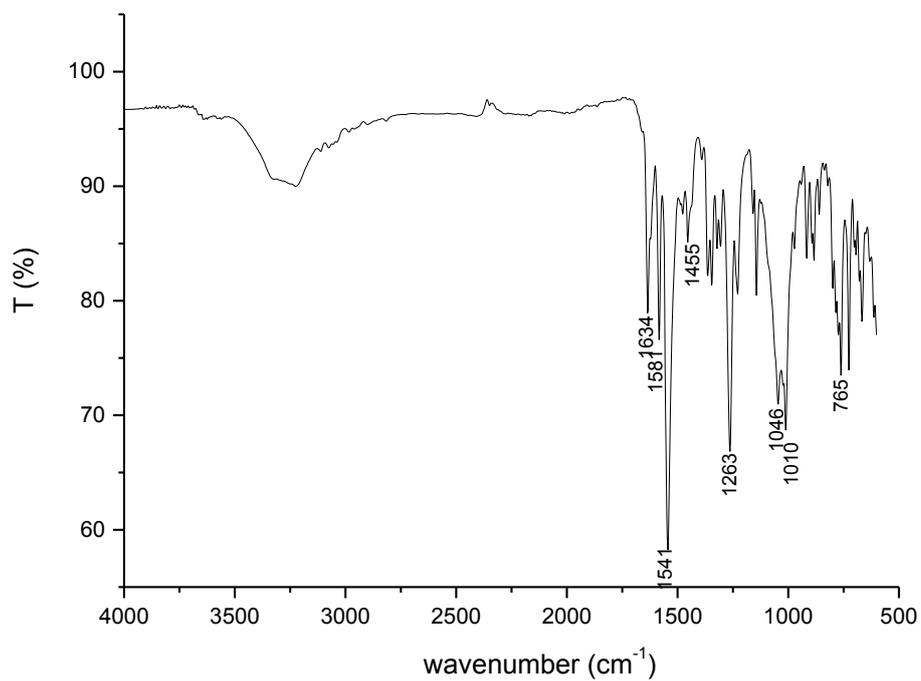


Figure S3. FTIR spectrum (ATR) of 3.

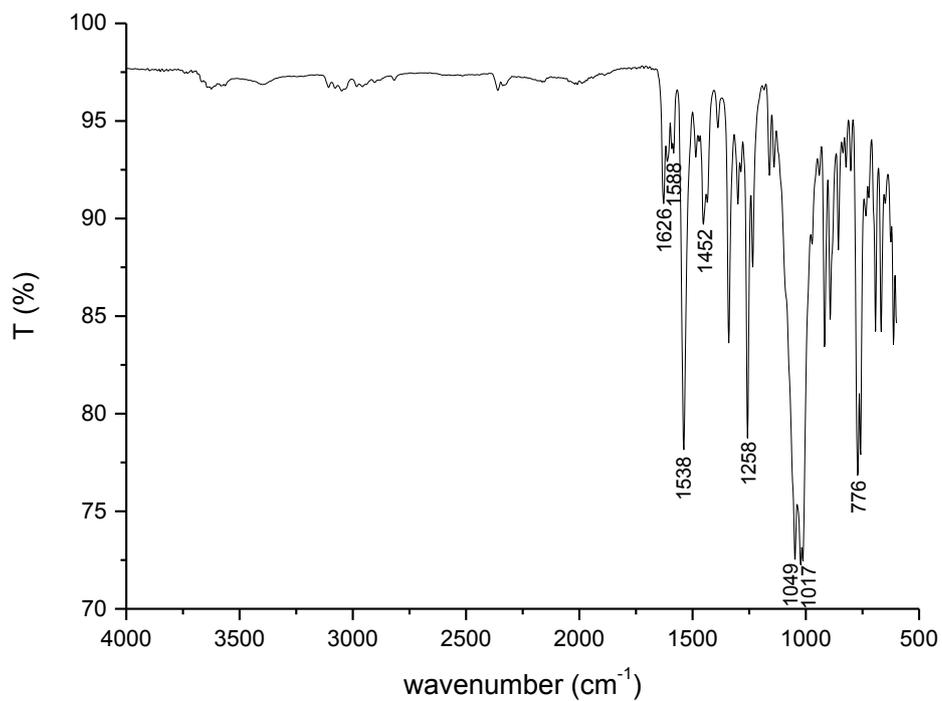


Figure S4. FTIR spectrum (ATR) of 4.

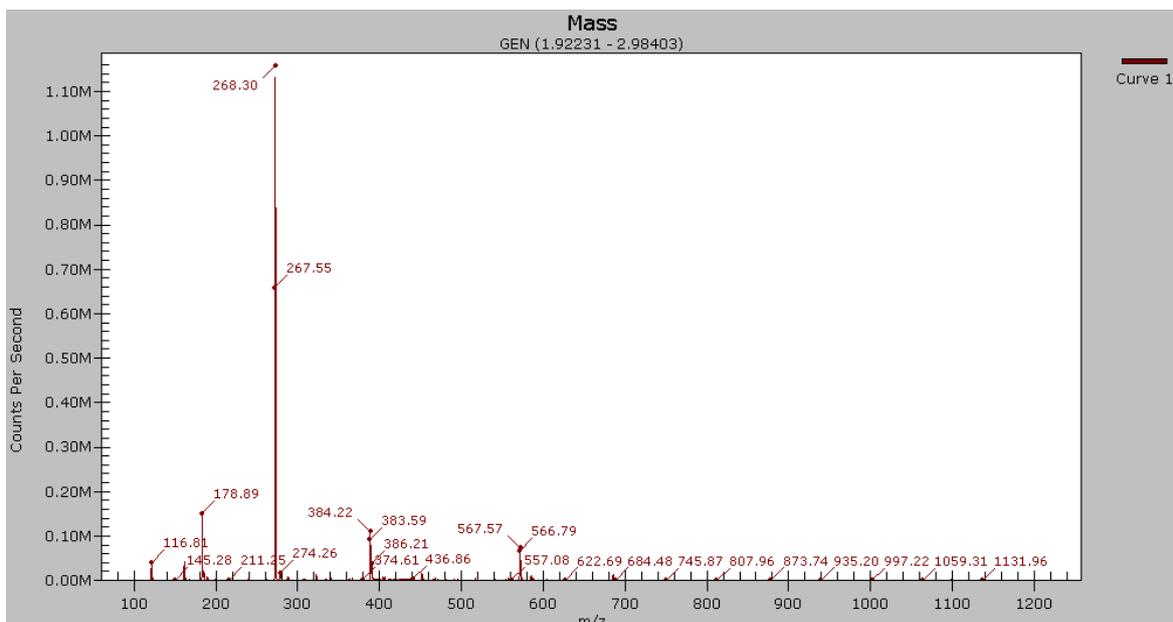


Figure S5. ESI-MS spectrum of **1** in MeCN.

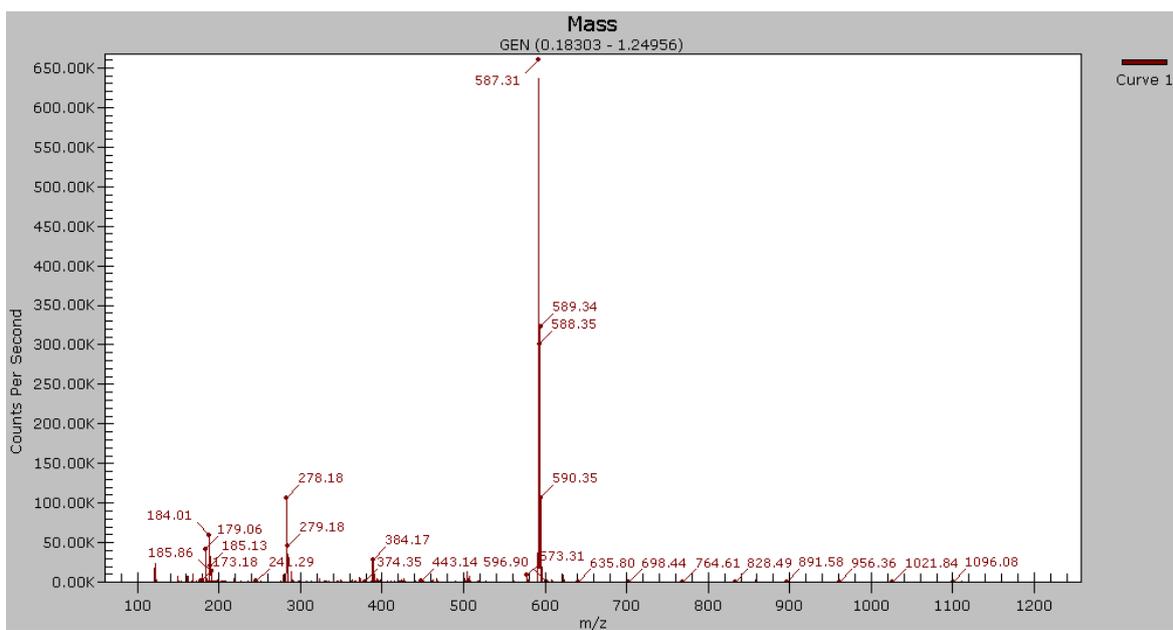


Figure S6. ESI-MS spectrum of **2** in MeCN.

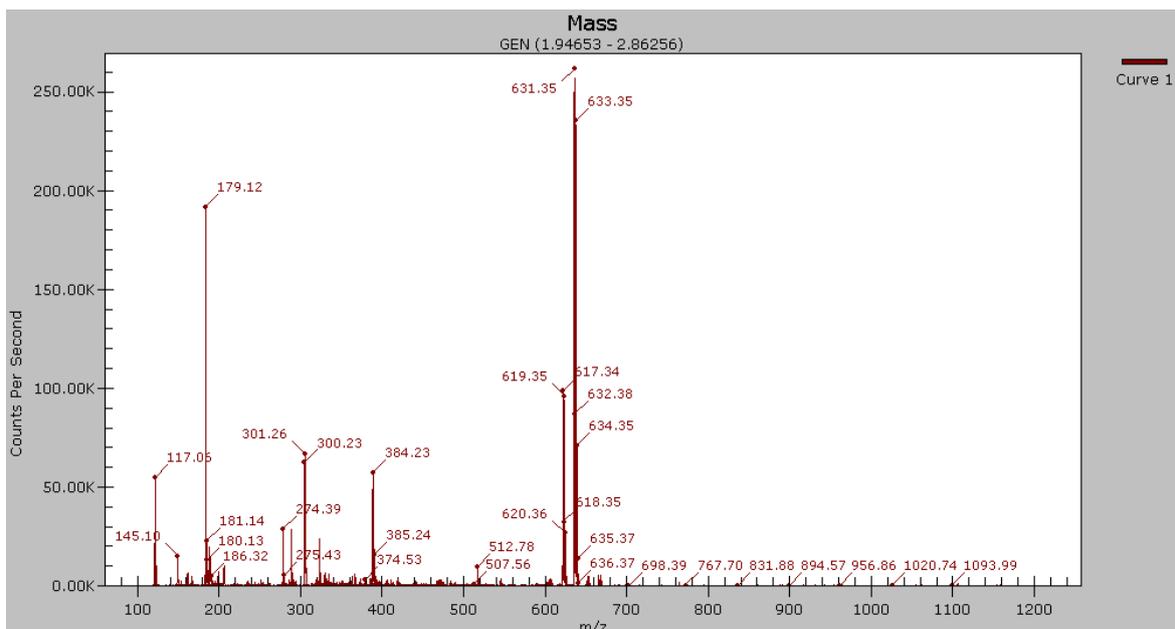


Figure S7. ESI-MS spectrum of **3** in MeCN.

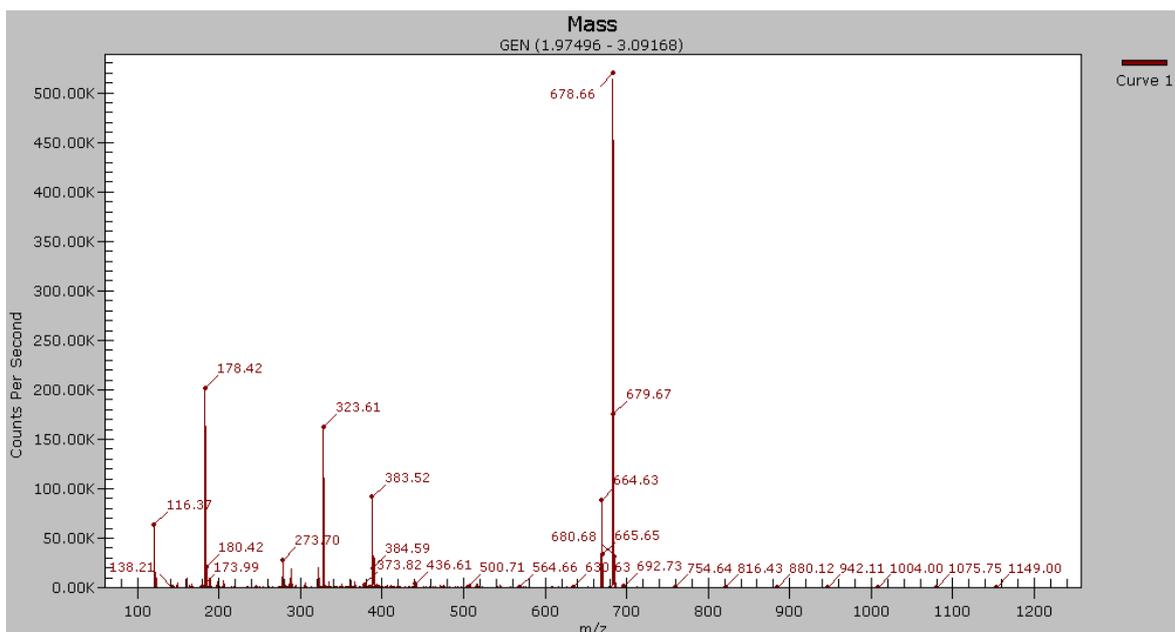


Figure S8. ESI-MS spectrum of **4** in MeCN.

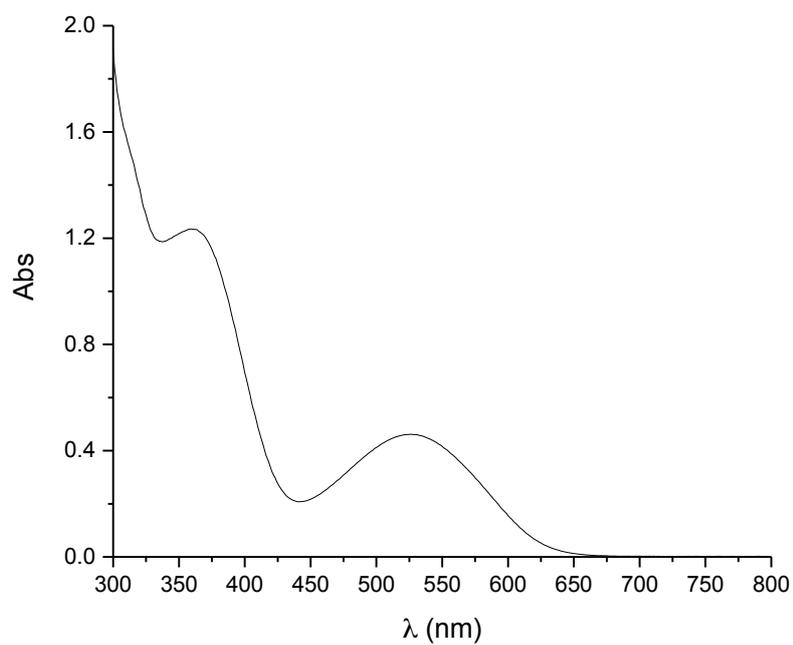


Figure S9. UV-Visible spectrum of **1** in DMSO (2.0×10^{-4} mol.L⁻¹).

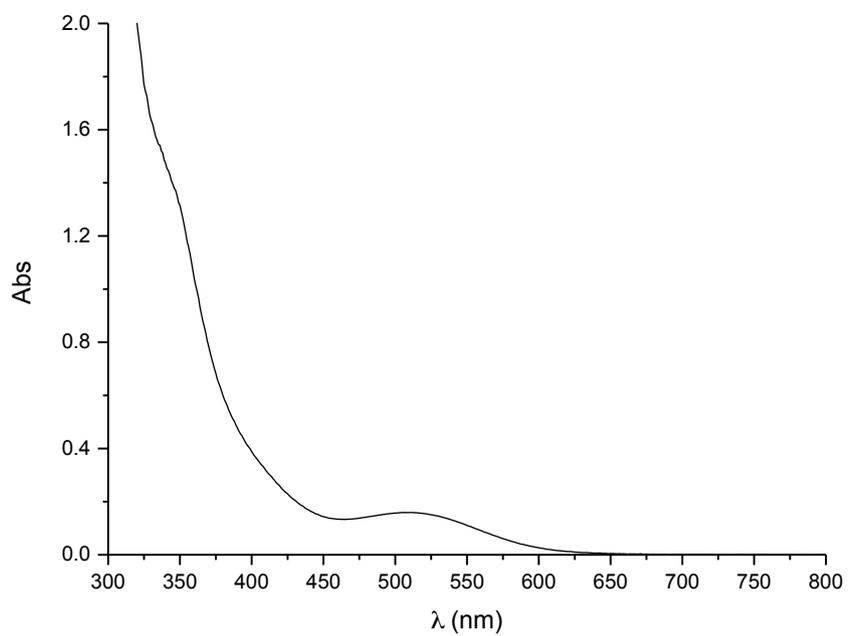


Figure S10. UV-Visible spectrum of **2** in DMSO (2.0×10^{-4} mol.L⁻¹).

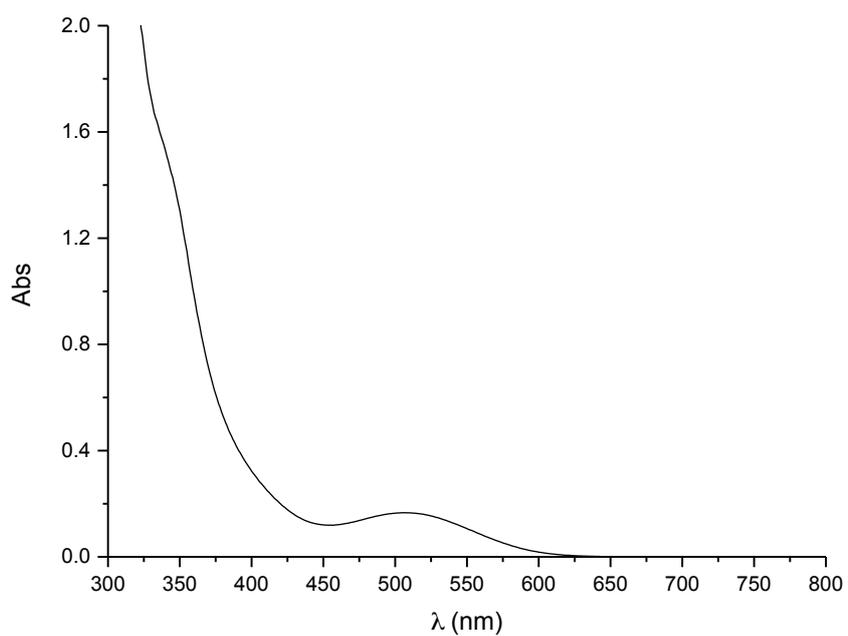


Figure S11. UV-Visible spectrum of **3** in DMSO (2.0×10^{-4} mol.L⁻¹).

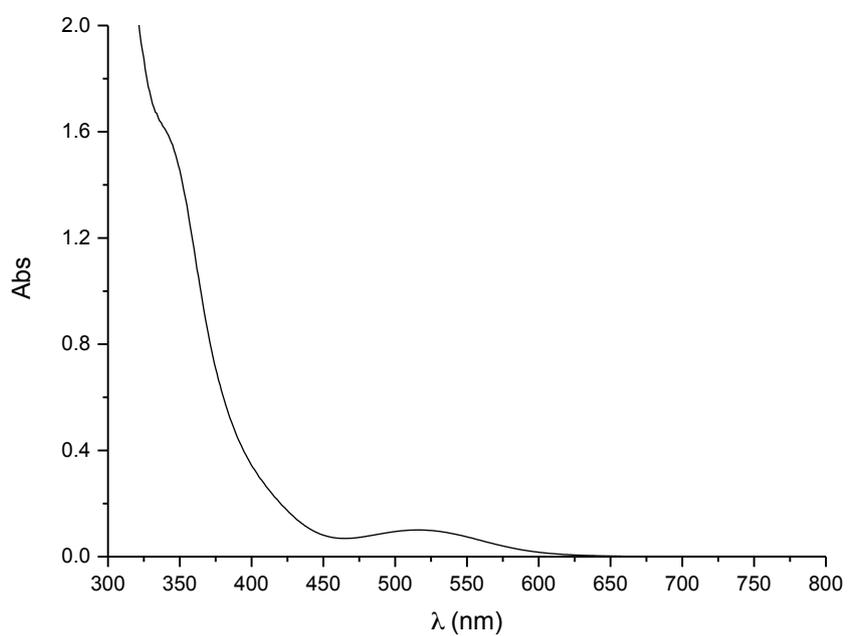


Figure S12. UV-Visible spectrum of **4** in DMSO (2.0×10^{-4} mol.L⁻¹).

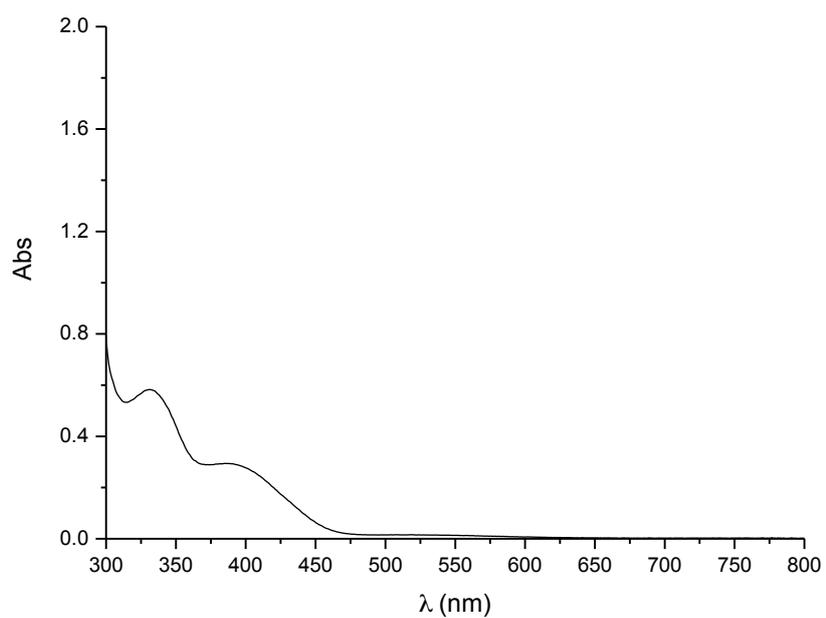


Figure S13. UV-Visible spectrum of HNQ-CH₃ in DMSO (2.0×10^{-4} mol.L⁻¹).

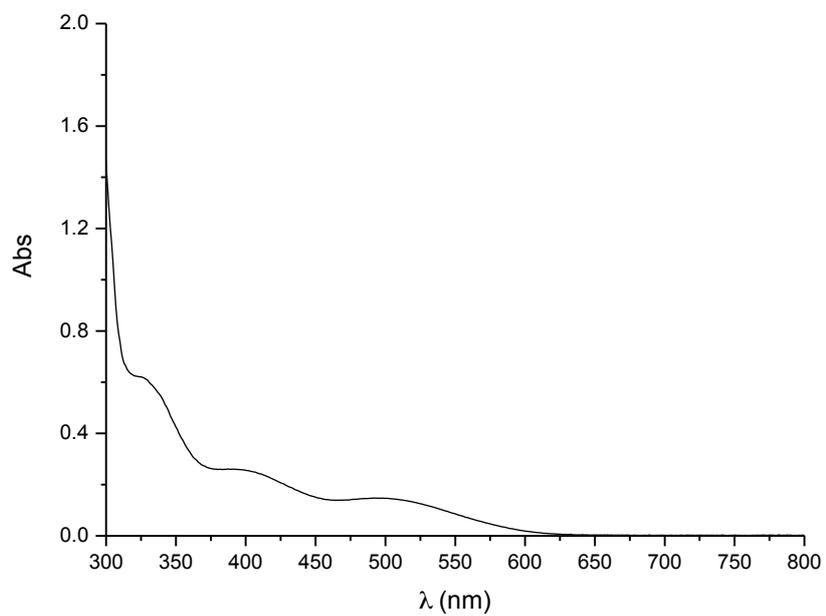


Figure S14. UV-Visible spectrum of HNQ-Cl in DMSO (2.0×10^{-4} mol.L⁻¹).

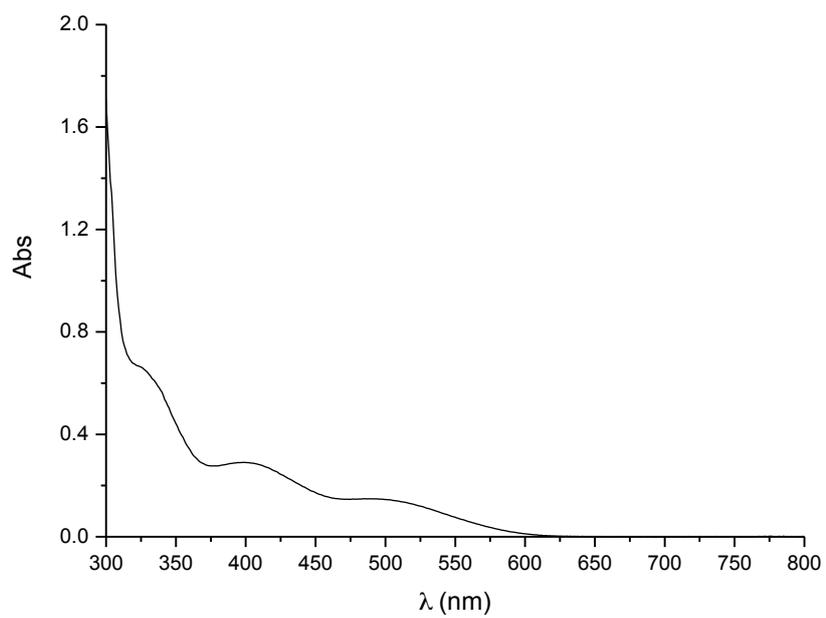


Figure S15. UV-Visible spectrum of HNQ-Br in DMSO ($2.0 \times 10^{-4} \text{ mol.L}^{-1}$).

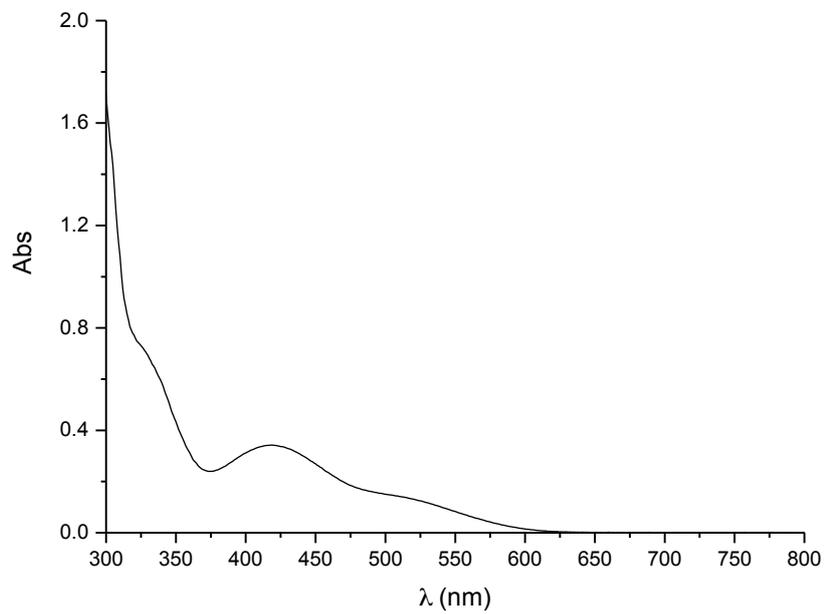


Figure S16. UV-Visible spectrum of HNQ-I in DMSO ($2.0 \times 10^{-4} \text{ mol.L}^{-1}$).

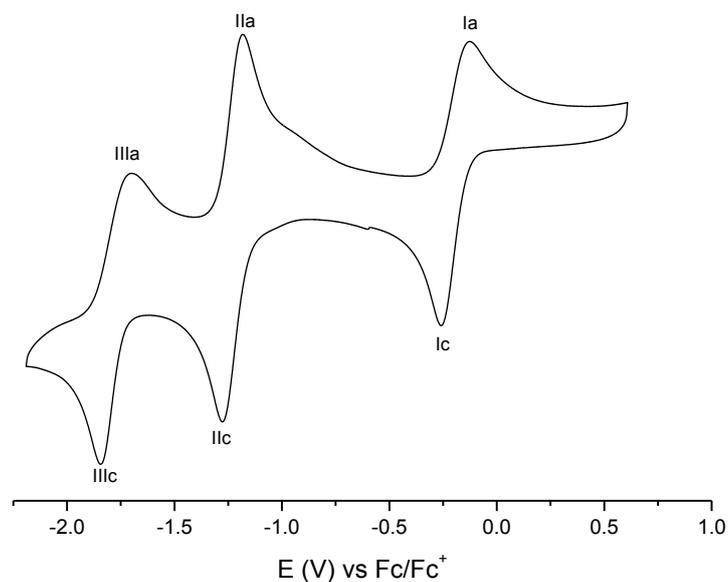


Figure S17. Cyclic voltammogram of **1** in MeCN ($1 \times 10^{-4} \text{ mol.L}^{-1}$) with TBAClO₄ 0.1 M, at 0.1 V/s, using a three electrode arrange (work.: carbon; ref.: Ag/AgCl (org.); aux.: Pt wire). Fc/Fc⁺ was used as an internal reference.

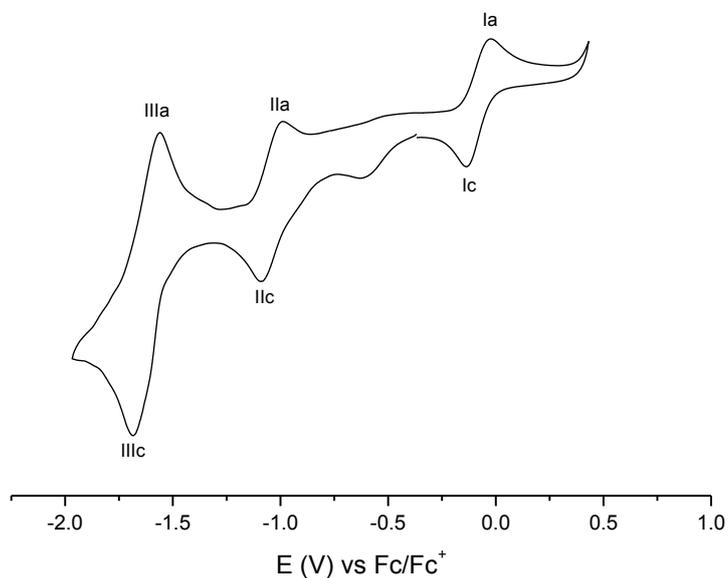


Figure S18. Cyclic voltammogram of **2** in MeCN ($1 \times 10^{-4} \text{ mol.L}^{-1}$) with TBAClO₄ 0.1 M, at 0.1 V/s, using a three electrode arrange (work.: carbon; ref.: Ag/AgCl (org.); aux.: Pt wire). Fc/Fc⁺ was used as an internal reference.

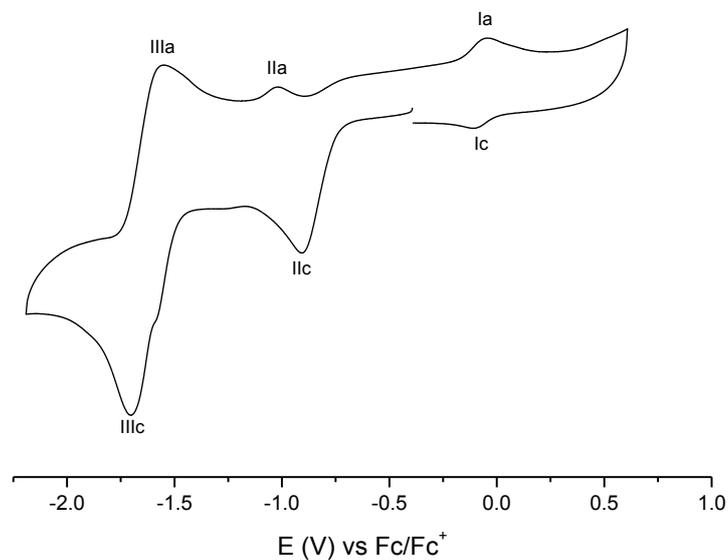


Figure S19. Cyclic voltammogram of **3** in MeCN (1×10^{-4} mol.L $^{-1}$) with TBAClO $_4$ 0.1 M, at 0.1 V/s, using a three electrode arrange (work.: carbon; ref.: Ag/AgCl (org.); aux.: Pt wire). Fc/Fc $^+$ was used as an internal reference.

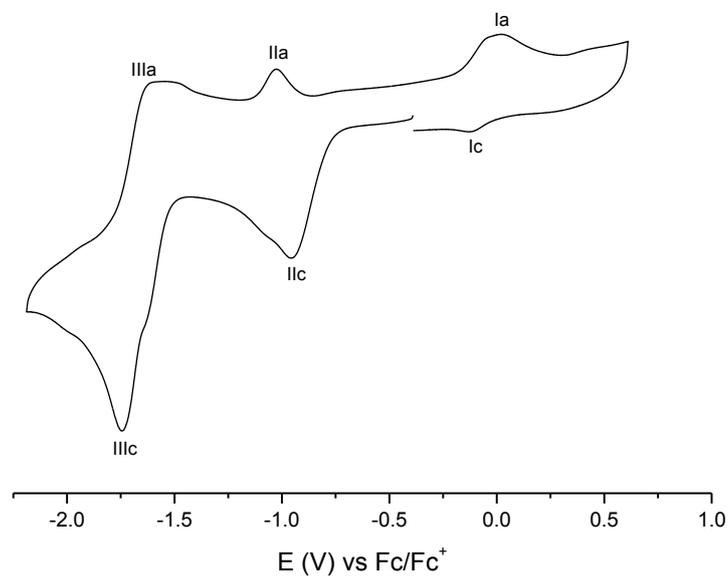


Figure S20. Cyclic voltammogram of **4** in MeCN (1×10^{-4} mol.L $^{-1}$) with TBAClO $_4$ 0.1 M, at 0.1 V/s, using a three electrode arrange (work.: carbon; ref.: Ag/AgCl (org.); aux.: Pt wire). Fc/Fc $^+$ was used as an internal reference.

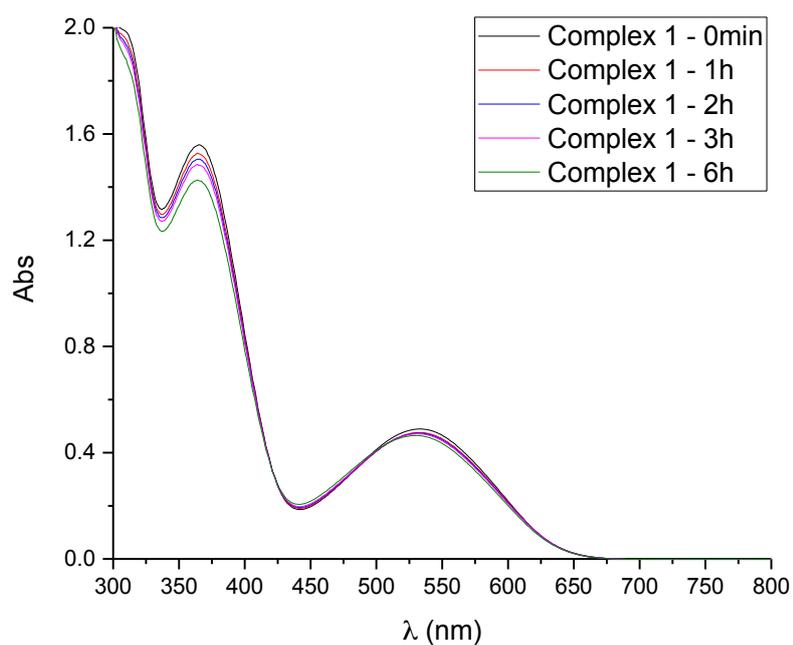


Figure S21. UV-Visible spectra of **1** ($2.0 \times 10^{-4} \text{ mol.L}^{-1}$) in DMSO/phosphate buffer 1:9 at pH 6.2, from freshly prepared solution.

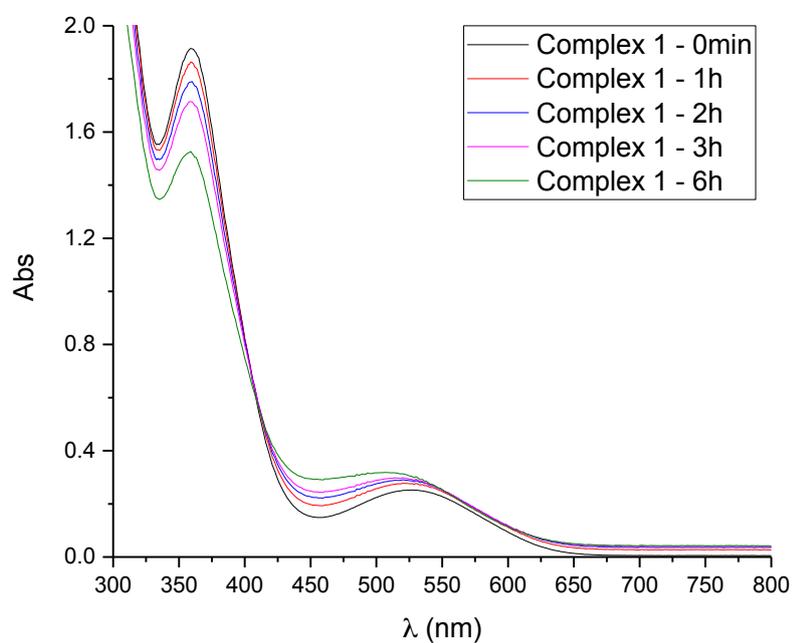


Figure S22. UV-Visible spectra of **1** ($2.0 \times 10^{-4} \text{ mol.L}^{-1}$) in DMSO/phosphate buffer 1:9 at pH 7.0, from freshly prepared solution.

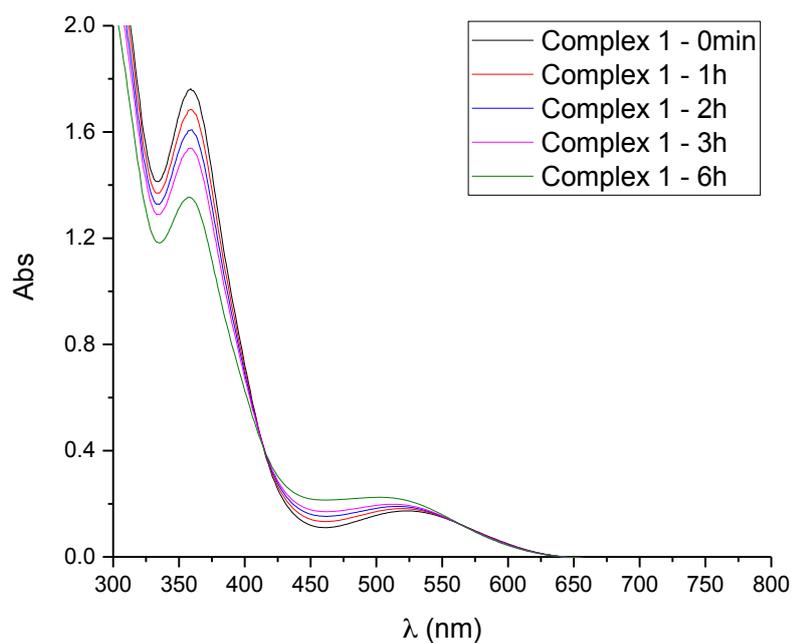


Figure S23. UV-Visible spectra of **1** ($2.0 \times 10^{-4} \text{ mol.L}^{-1}$) in DMSO/phosphate buffer 1:9 at pH 7.4, from freshly prepared solution.

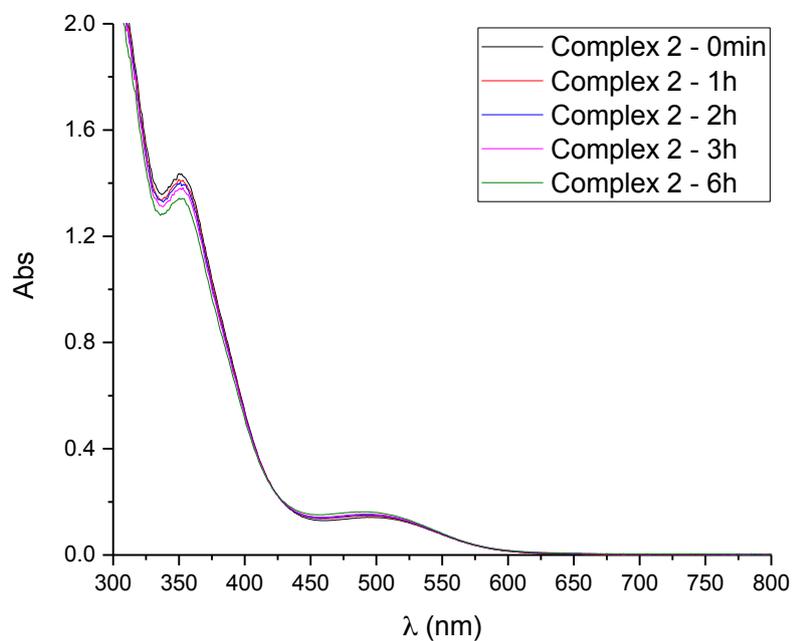


Figure S24. UV-Visible spectra of **2** ($2.0 \times 10^{-4} \text{ mol.L}^{-1}$) in DMSO/phosphate buffer 1:9 at pH 6.2, from freshly prepared solution.

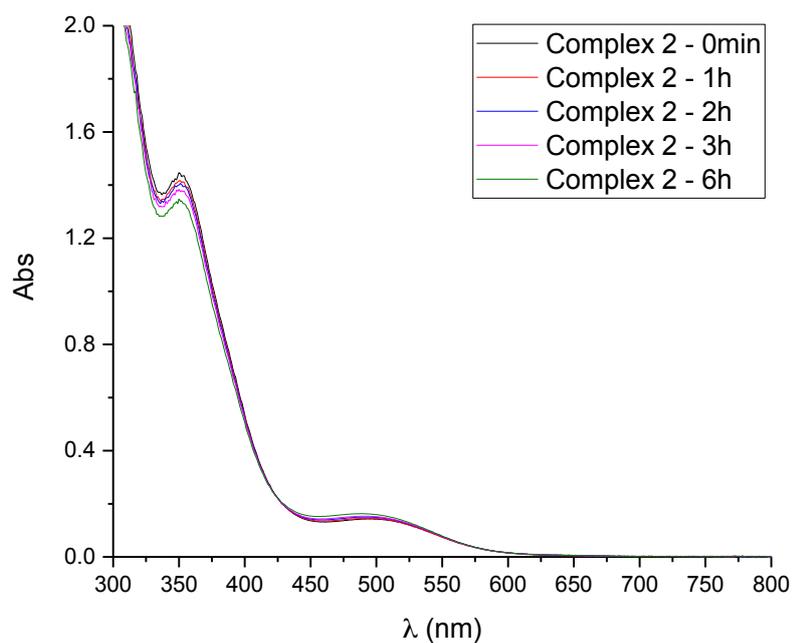


Figure S25. UV-Visible spectra of **2** ($2.0 \times 10^{-4} \text{ mol.L}^{-1}$) in DMSO/phosphate buffer 1:9 at pH 7.0, from freshly prepared solution.

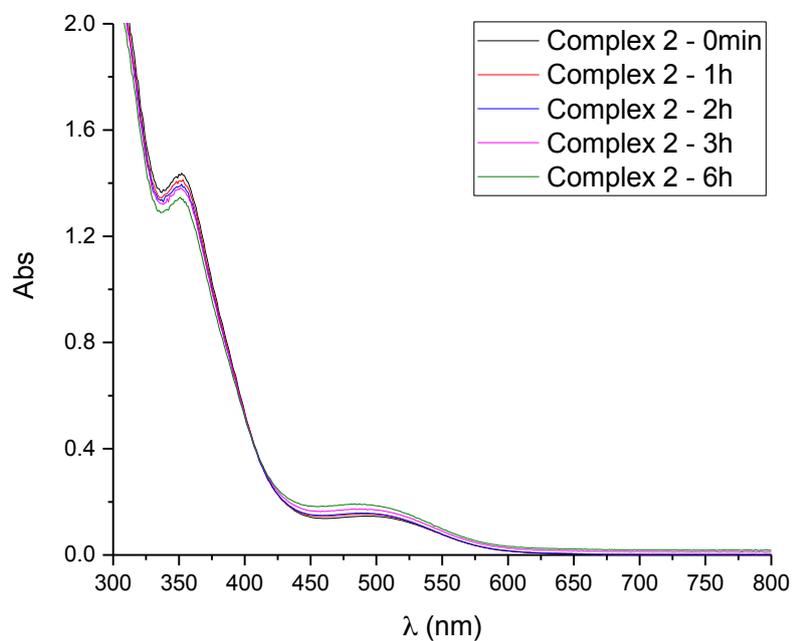


Figure S26. UV-Visible spectra of **2** ($2.0 \times 10^{-4} \text{ mol.L}^{-1}$) in DMSO/phosphate buffer 1:9 at pH 7.4, from freshly prepared solution.

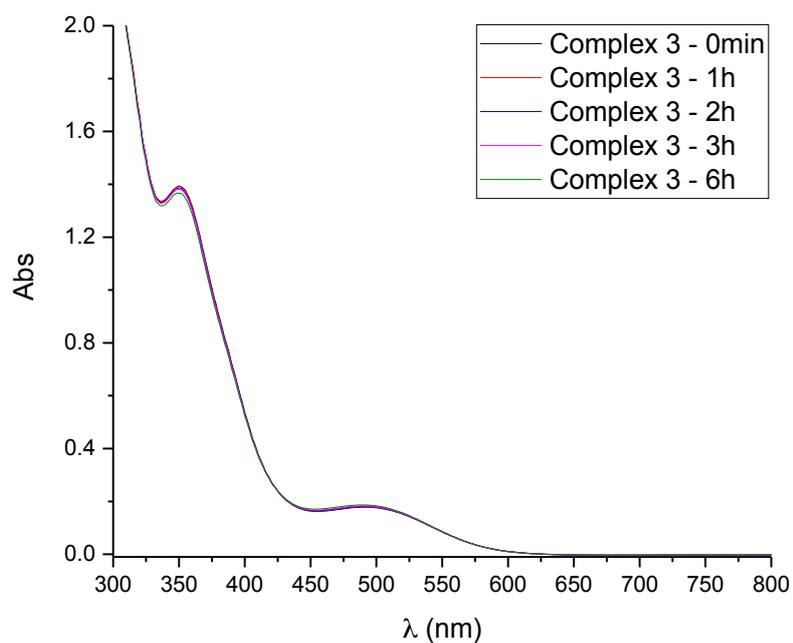


Figure S27. UV-Visible spectra of **3** (2.0×10^{-4} mol.L⁻¹) in DMSO/phosphate buffer 1:9 at pH 6.2, from freshly prepared solution.

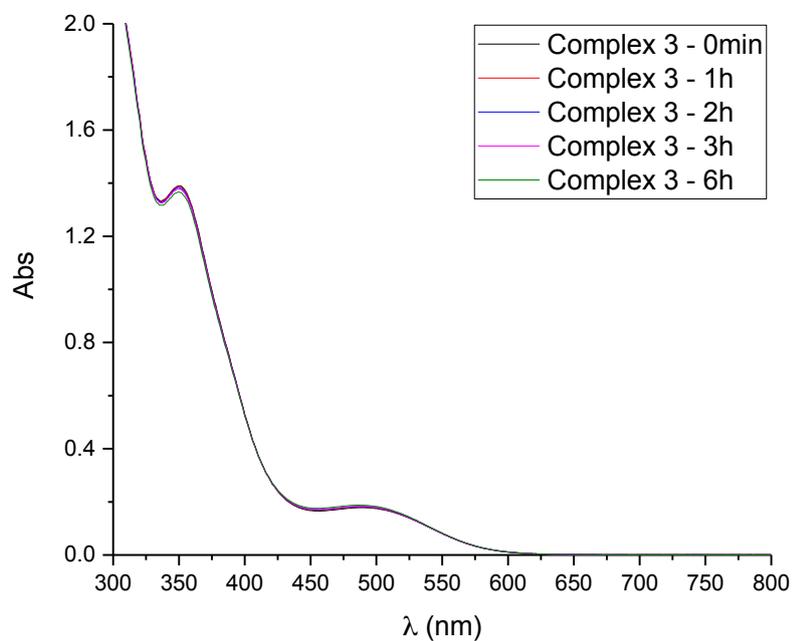


Figure S28. UV-Visible spectra of **3** (2.0×10^{-4} mol.L⁻¹) in DMSO/phosphate buffer 1:9 at pH 7.0, from freshly prepared solution.

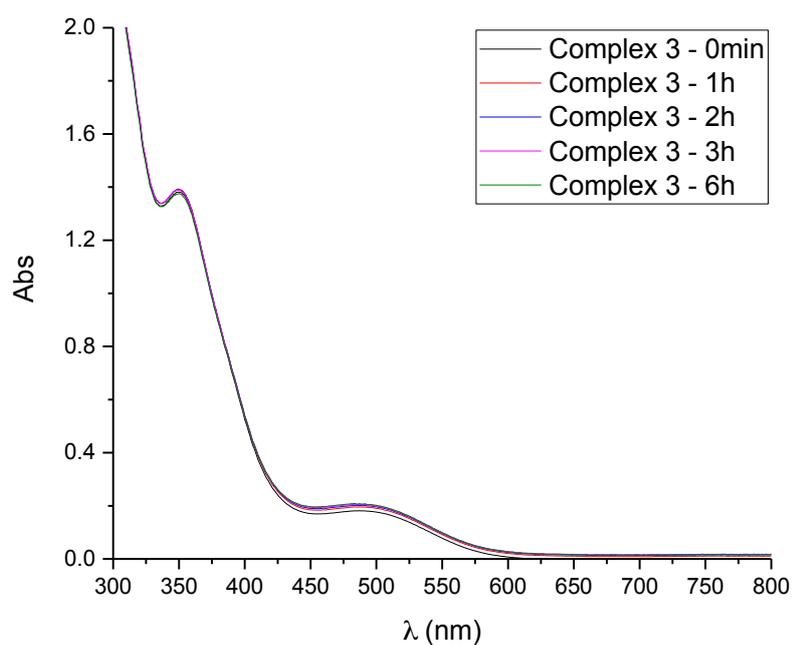


Figure S29. UV-Visible spectra of **3** ($2.0 \times 10^{-4} \text{ mol.L}^{-1}$) in DMSO/phosphate buffer 1:9 at pH 7.4, from freshly prepared solution.

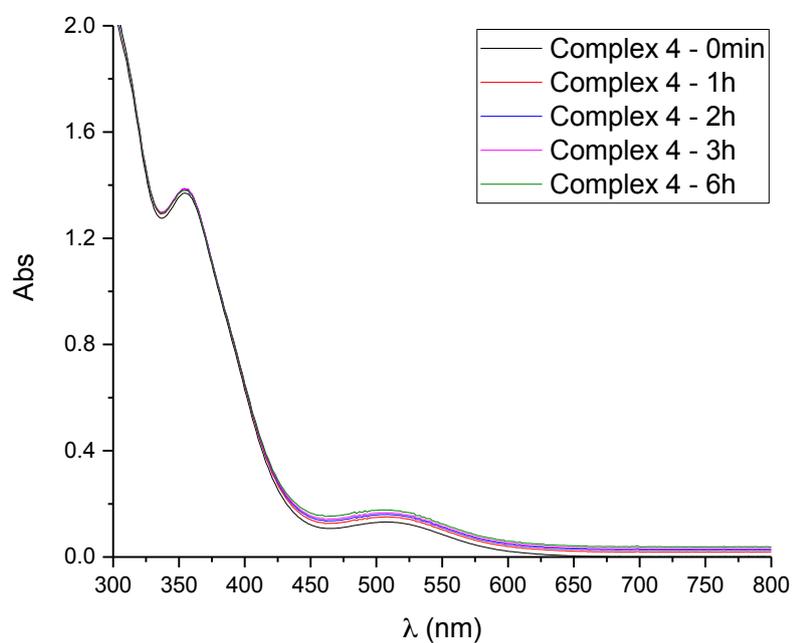


Figure S30. UV-Visible spectra of **4** ($2.0 \times 10^{-4} \text{ mol.L}^{-1}$) in DMSO/phosphate buffer 1:9 at pH 6.2, from freshly prepared solution.

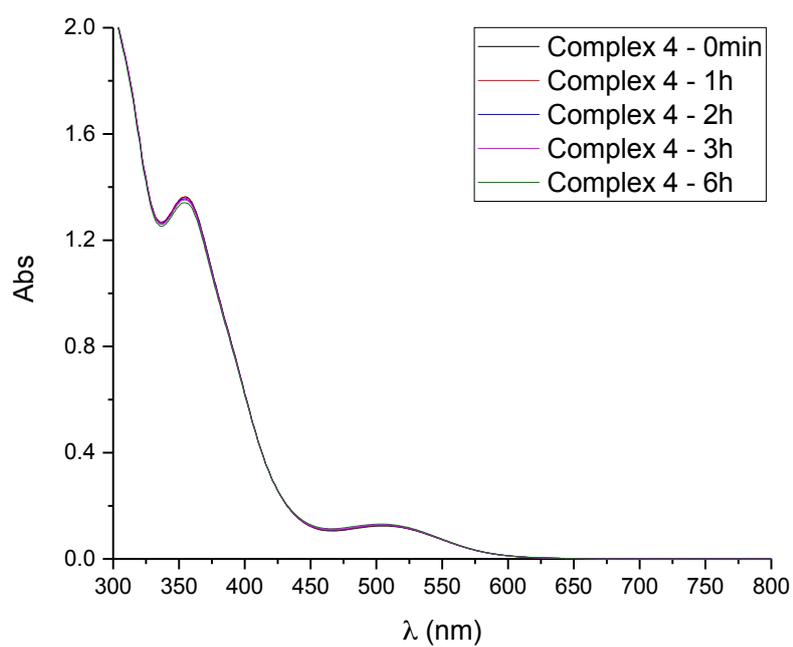


Figure S31. UV-Visible spectra of **4** ($2.0 \times 10^{-4} \text{ mol.L}^{-1}$) in DMSO/phosphate buffer 1:9 at pH 7.0, from freshly prepared solution.

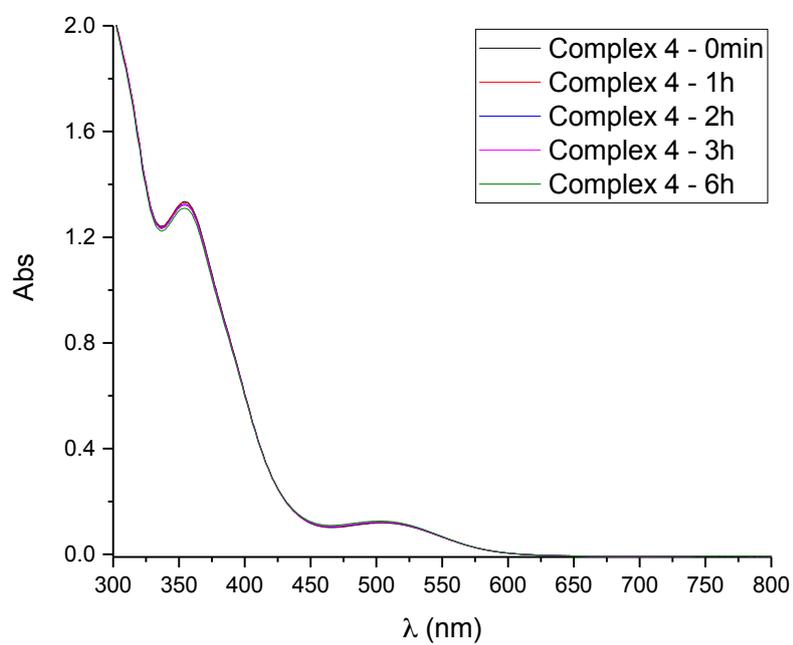


Figure S32. UV-Visible spectra of **4** ($2.0 \times 10^{-4} \text{ mol.L}^{-1}$) in DMSO/phosphate buffer 1:9 at pH 7.4, from freshly prepared solution.

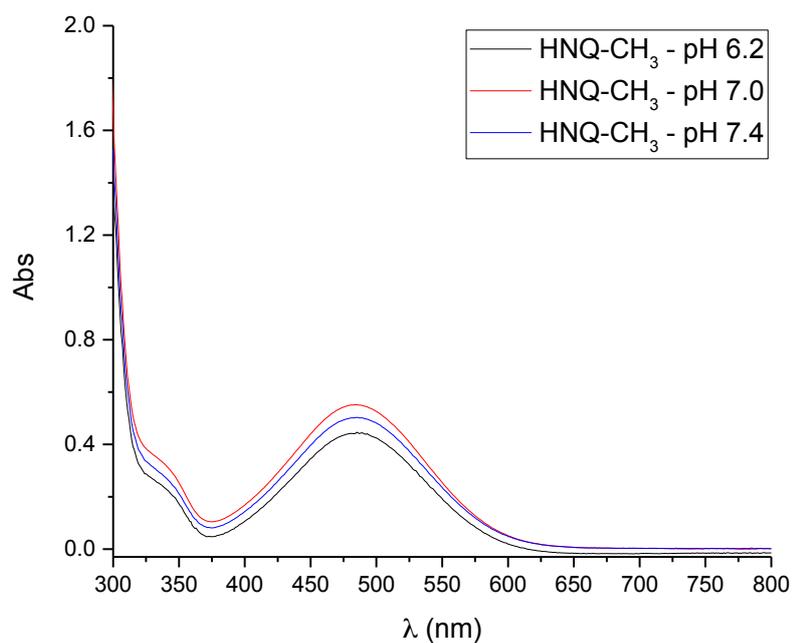


Figure S33. UV-Visible spectra of HNQ-CH₃ (2.0×10^{-4} mol.L⁻¹) in DMSO/phosphate buffer 1:9 at pH 6.2, 7.0 and 7.4.

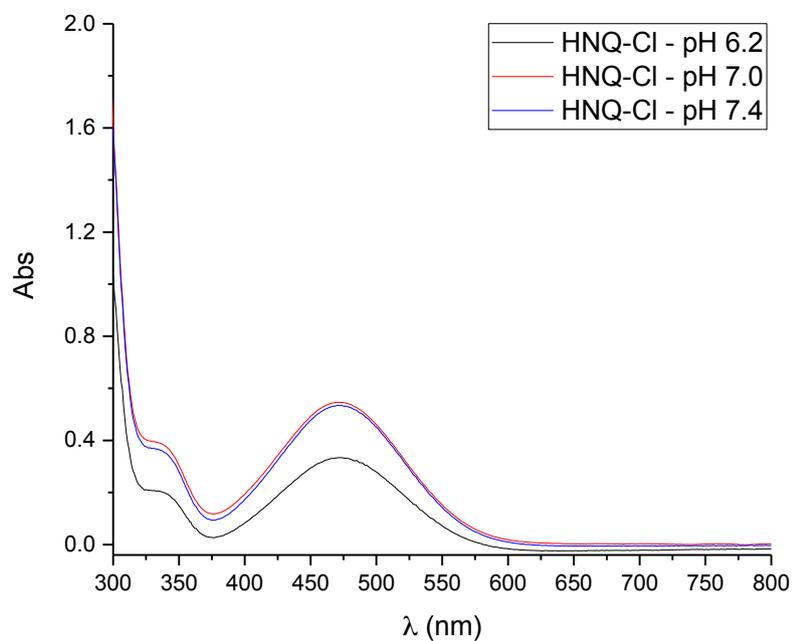


Figure S34. UV-Visible spectra of HNQ-Cl (2.0×10^{-4} mol.L⁻¹) in DMSO/phosphate buffer 1:9 at pH 6.2, 7.0 and 7.4.

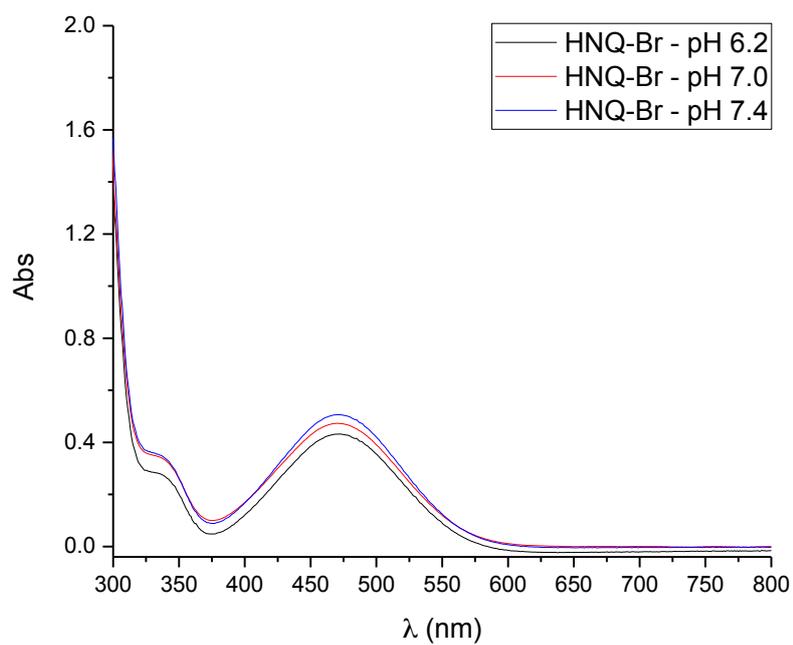


Figure S35. UV-Visible spectra of HNQ-Br ($2.0 \times 10^{-4} \text{ mol.L}^{-1}$) in DMSO/phosphate buffer 1:9 at pH 6.2, 7.0 and 7.4.

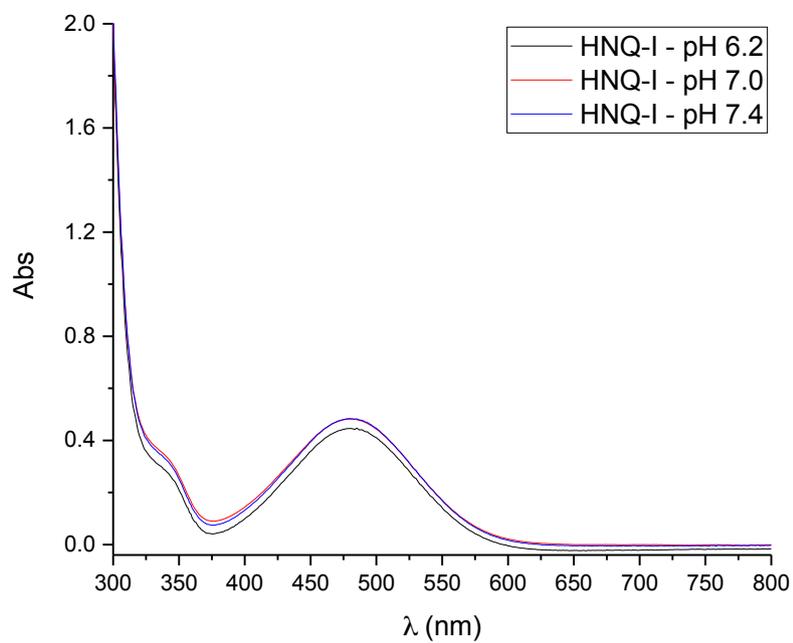


Figure S36. UV-Visible spectra of HNQ-I ($2.0 \times 10^{-4} \text{ mol.L}^{-1}$) in DMSO/phosphate buffer 1:9 at pH 6.2, 7.0 and 7.4.

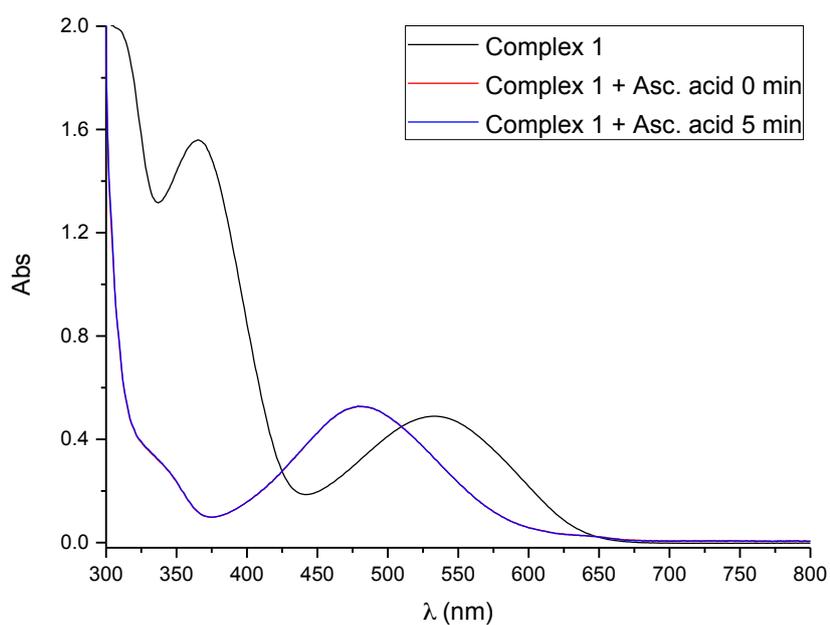


Figure S37. UV-Visible spectra of **1** ($2.0 \times 10^{-4} \text{ mol.L}^{-1}$) in DMSO/phosphate buffer 1:9 at pH 6.2 and in the presence of ten equivalents of ascorbic acid (AA) ($2.0 \times 10^{-3} \text{ mol.L}^{-1}$).

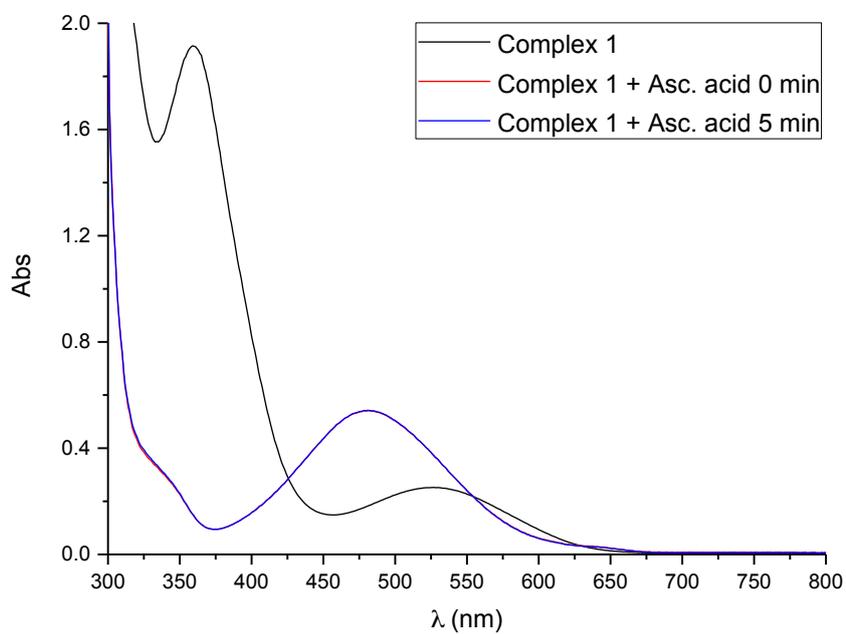


Figure S38. UV-Visible spectra of **1** ($2.0 \times 10^{-4} \text{ mol.L}^{-1}$) in DMSO/phosphate buffer 1:9 at pH 7.0 and in the presence of ten equivalents of ascorbic acid (AA) ($2.0 \times 10^{-3} \text{ mol.L}^{-1}$).

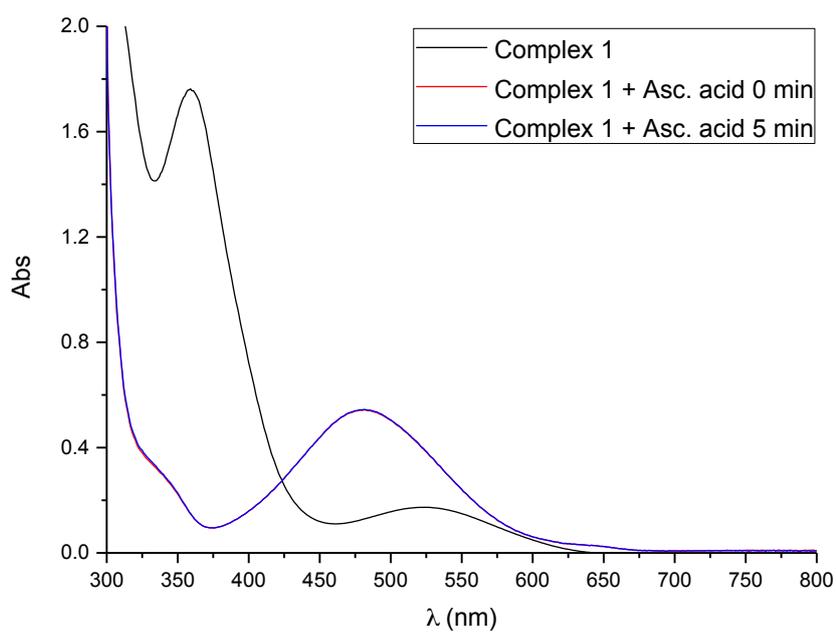


Figure S39. UV-Visible spectra of **1** ($2.0 \times 10^{-4} \text{ mol.L}^{-1}$) in DMSO/phosphate buffer 1:9 at pH 7.4 and in the presence of ten equivalents of ascorbic acid (AA) ($2.0 \times 10^{-3} \text{ mol.L}^{-1}$).

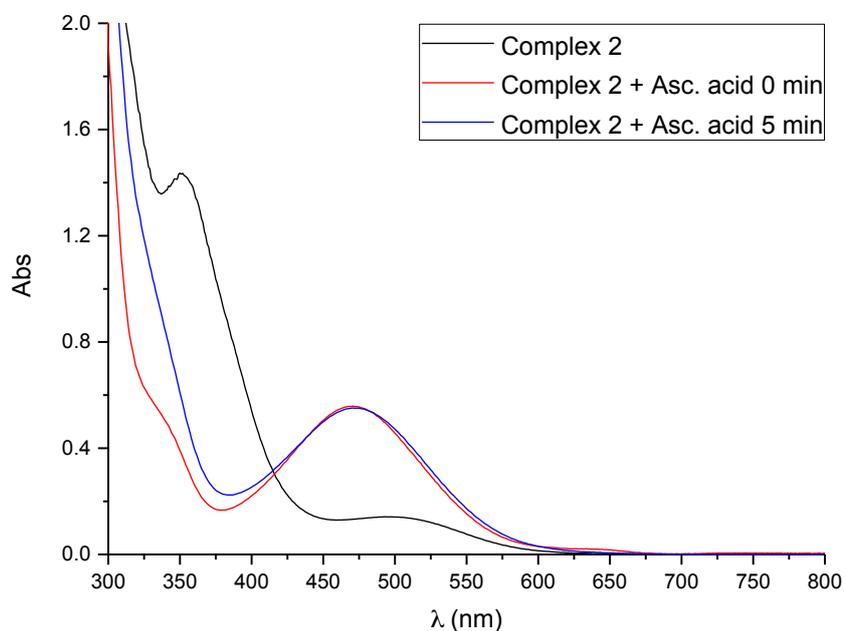


Figure S40. UV-Visible spectra of **2** ($2.0 \times 10^{-4} \text{ mol.L}^{-1}$) in DMSO/phosphate buffer 1:9 at pH 6.2 and in the presence of ten equivalents of ascorbic acid (AA) ($2.0 \times 10^{-3} \text{ mol.L}^{-1}$).

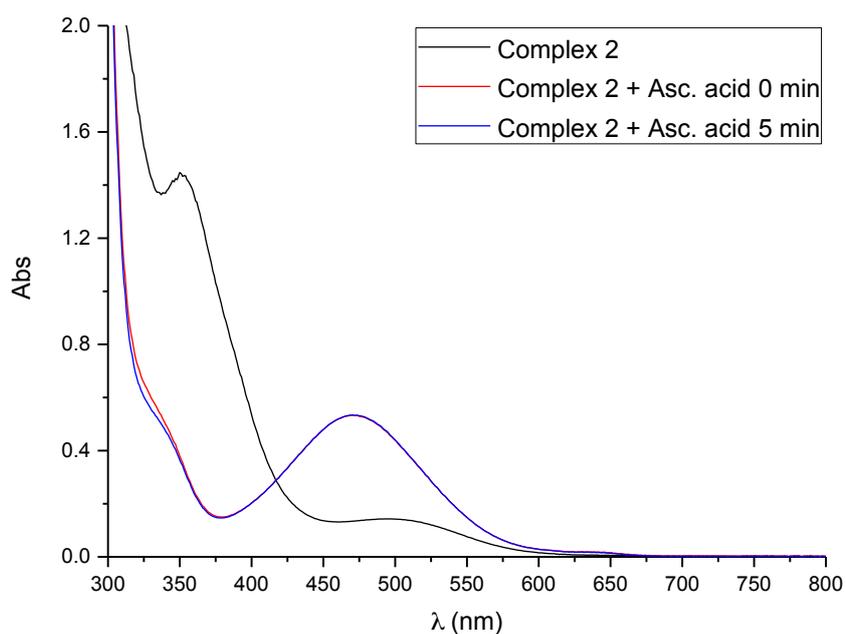


Figure S41. UV-Visible spectra of **2** (2.0×10^{-4} mol.L $^{-1}$) in DMSO/phosphate buffer 1:9 at pH 7.0 and in the presence of ten equivalents of ascorbic acid (AA) (2.0×10^{-3} mol.L $^{-1}$).

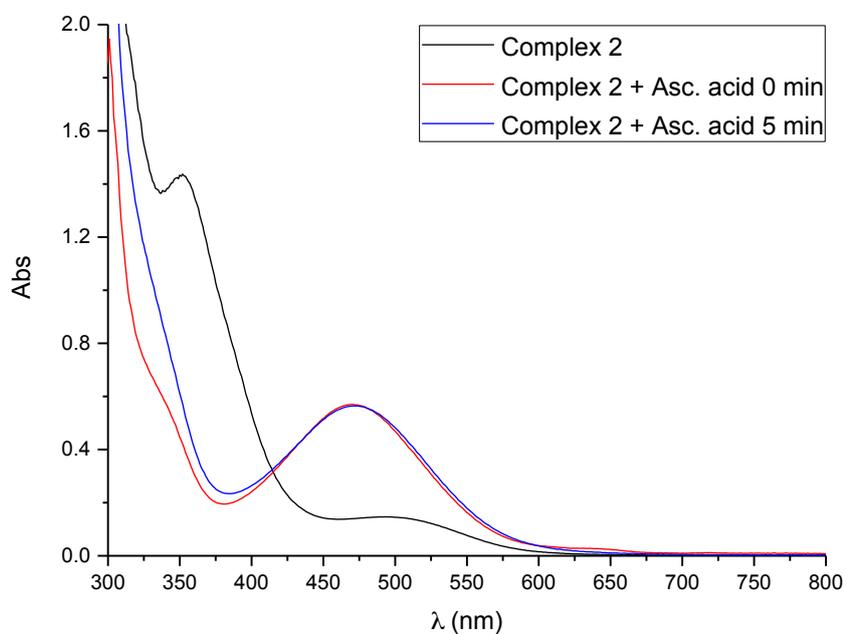


Figure S42. UV-Visible spectra of **2** (2.0×10^{-4} mol.L $^{-1}$) in DMSO/phosphate buffer 1:9 at pH 7.4 and in the presence of ten equivalents of ascorbic acid (AA) (2.0×10^{-3} mol.L $^{-1}$).

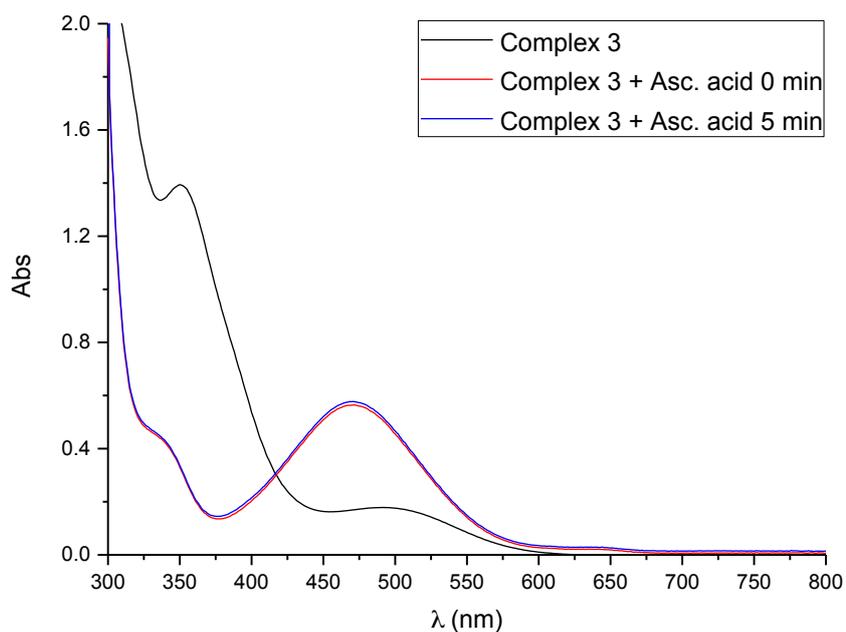


Figure S43. UV-Visible spectra of **3** ($2.0 \times 10^{-4} \text{ mol.L}^{-1}$) in DMSO/phosphate buffer 1:9 at pH 6.2 and in the presence of ten equivalents of ascorbic acid (AA) ($2.0 \times 10^{-3} \text{ mol.L}^{-1}$).

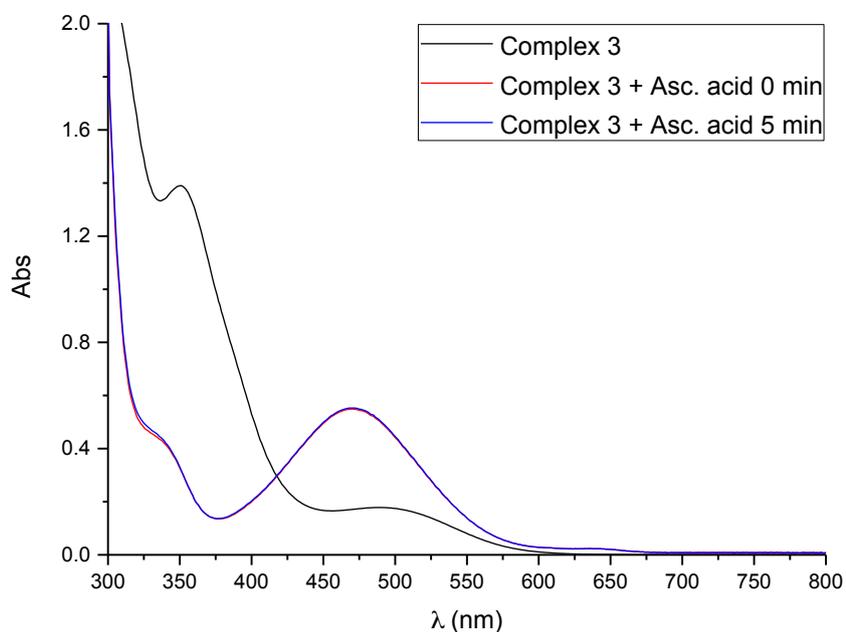


Figure S44. UV-Visible spectra of **3** ($2.0 \times 10^{-4} \text{ mol.L}^{-1}$) in DMSO/phosphate buffer 1:9 at pH 7.0 and in the presence of ten equivalents of ascorbic acid (AA) ($2.0 \times 10^{-3} \text{ mol.L}^{-1}$).

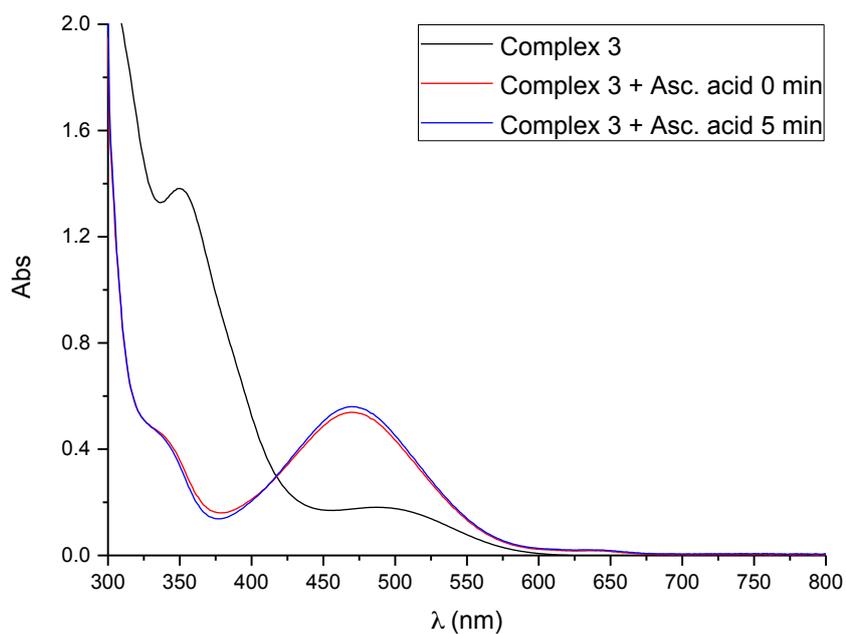


Figure S45. UV-Visible spectra of **3** ($2.0 \times 10^{-4} \text{ mol.L}^{-1}$) in DMSO/phosphate buffer 1:9 at pH 7.4 and in the presence of ten equivalents of ascorbic acid (AA) ($2.0 \times 10^{-3} \text{ mol.L}^{-1}$).

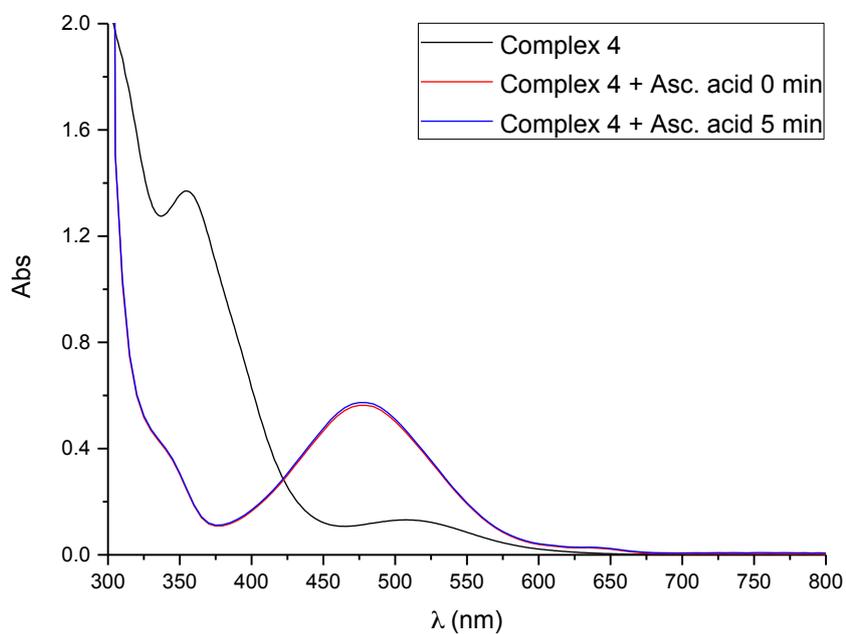


Figure S46. UV-Visible spectra of **4** ($2.0 \times 10^{-4} \text{ mol.L}^{-1}$) in DMSO/phosphate buffer 1:9 at pH 6.2 and in the presence of ten equivalents of ascorbic acid (AA) ($2.0 \times 10^{-3} \text{ mol.L}^{-1}$).

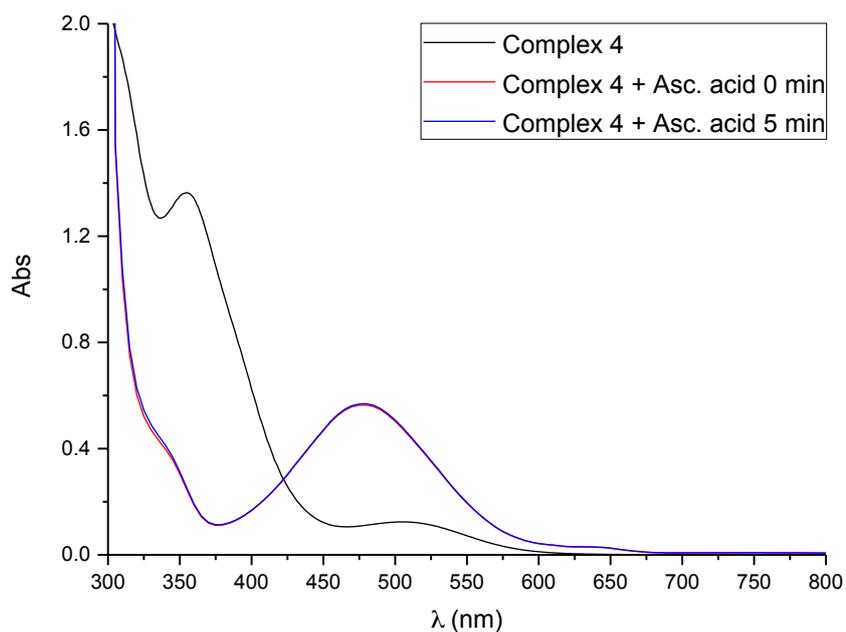


Figure S47. UV-Visible spectra of **4** (2.0×10^{-4} mol.L⁻¹) in DMSO/phosphate buffer 1:9 at pH 7.0 and in the presence of ten equivalents of ascorbic acid (AA) (2.0×10^{-3} mol.L⁻¹).

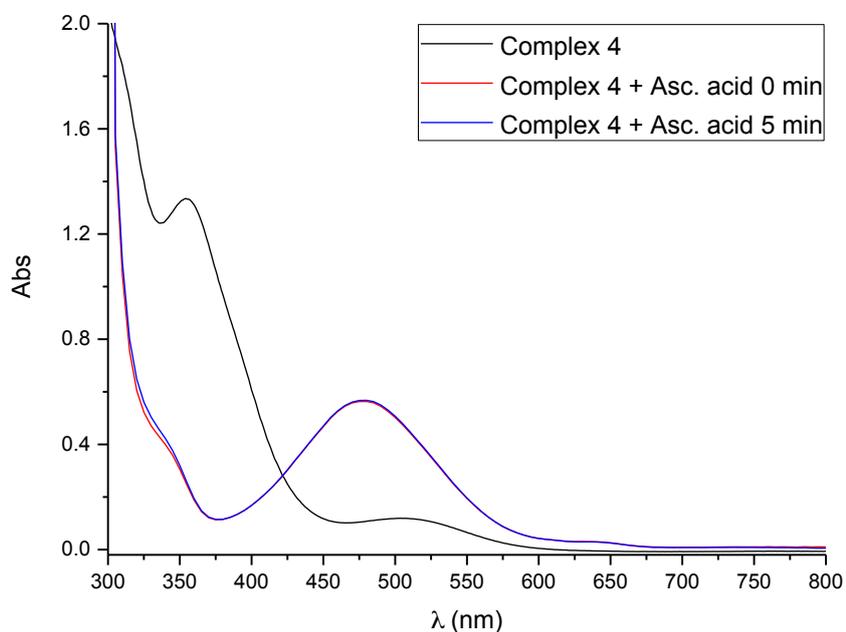


Figure S48. UV-Visible spectra of **4** (2.0×10^{-4} mol.L⁻¹) in DMSO/phosphate buffer 1:9 at pH 7.4 and in the presence of ten equivalents of ascorbic acid (AA) (2.0×10^{-3} mol.L⁻¹).

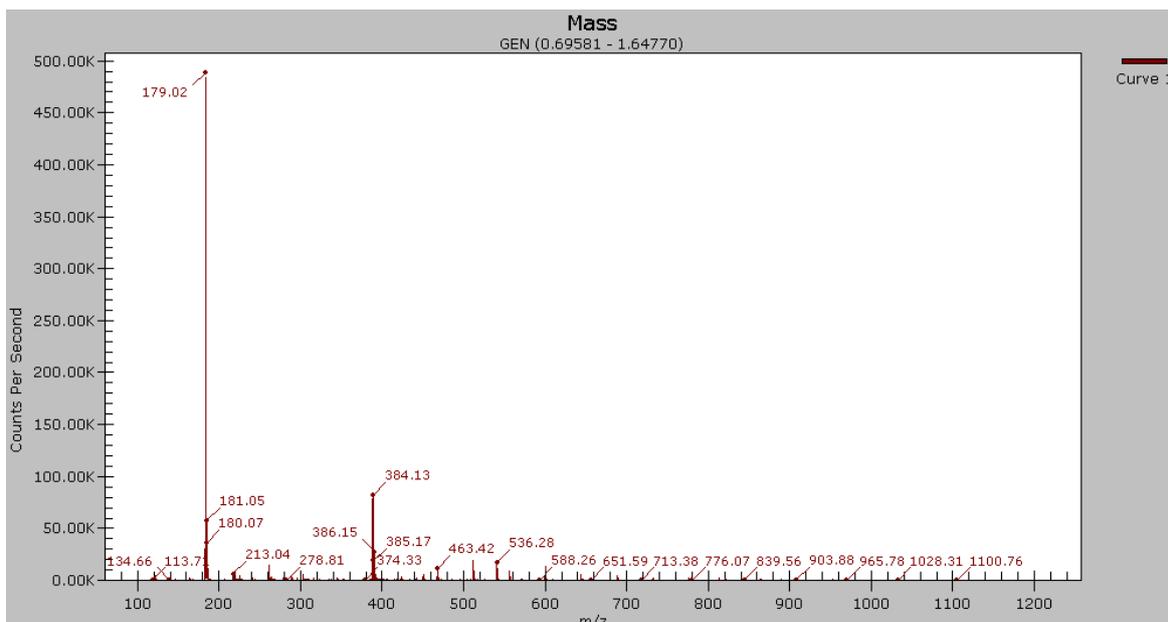


Figure S49. ESI-MS (positive scan) spectrum of **1** after reaction with ascorbic acid.

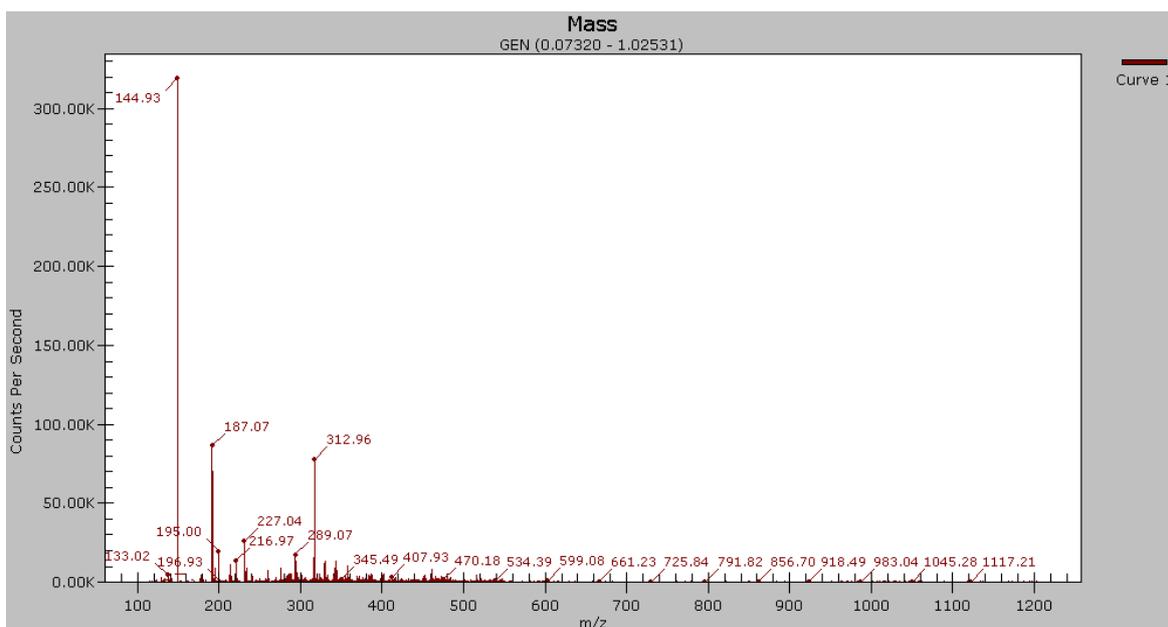


Figure S50. ESI-MS (negative scan) spectrum of **1** after reaction with ascorbic acid.