Supporting information

1, 3-Disulfoimidazolium Chloronickellate Immobilized HZSM-5 Framework

as Visible-Light-Induced Heterogeneous Photocatalyst for Advanced

Oxidation Process

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IR crystallinity determination:



Fig.S1: IR absorbance spectra of $\underline{1}$ /HZSM-5 ($\underline{2a}-\underline{2e}$) with respect to parent HZSM-5 for optical density calculation

Table S1: Optical density calculation from IR absorbance spectra

Entry	Optical density ratio					
	HZSM-5	<u>2a</u>	<u>2b</u>	<u>2c</u>	<u>2d</u>	<u>2e</u>
<u>1</u> /HZSM-5	0.83	0.81	0.79	0.79	0.77	1.01

Optical density ratio=Absorbance at 556 cm⁻¹/Absorbance at 450 cm⁻¹

2θ (degree)	Relative % crystallinity					
	HZSM-5	<u>2a</u>	<u>2b</u>	<u>2c</u>	<u>2d</u>	<u>2e</u>
7.78	100	99.50	98.14	95.39	92.42	83.67
8.75	100	97.39	95.52	92.16	90.07	82.87
22.93	100	99.05	98.45	96.37	93.79	86.56
44.96	100	95.76	93.48	92.28	90.18	78.59

Table S2: Relative % crystallinity table for the hybrid materials

Table S3: Hammett acidity values of the hybrid materials of [Dsim]₂[NiCl₄]/HZSM-5

Entry	A _{max}	[I]%	[IH ⁺]%	Ho
Base	0.745	100	0	-
<u>1</u>	0.380	51.00	49.00	1.01
<u>2a</u>	0.187	25.10	74.90	0.51
<u>2b</u>	0.166	22.28	77.72	0.44
2c	0.152	20.40	79.60	0.39
2d	0.131	17.58	82.42	0.32
2e	0.094	12.61	87.39	0.15



Fig.S2: BJH plots of <u>2a-2d</u> and HZSM-5



Fig.S3: t-Plots of HZSM-5, <u>2a-2d</u>

Reaction time	Degradation rate $(\%)^{a,b} =$				
(min)	$(A_0-A_t/A_0)*100$				
	<u>2c</u>		<u>2d</u>		
	A _t	(%)	A _t	(%)	
5	2.42	35	2.41	35	
10	1.39	63	1.94	48	
15	0.48	87	1.37	63	
20	0.10	97	0.93	75	
25	-	-	0.46	88	
30	-	-	0.28	92	

Table S4: MB dye degradation rate (%) with reaction time

^aDegradation rate (%) were calculated using $A_0 = 3.72$, ^b All the degradation rates are taken as approximate percentage values.



Fig.S4: MB degradation using 20 mg of 2c



Fig.S5: UV-visible analysis of MB dye degradation by recycled catalyst $\underline{2c}$

 Table S5: Calculation of degradation rate (%) of methylene blue for recycled catalyst

Entry	<u>2c</u> ^a		
	A _t	%	
1 st recycle	0.47	87	
2 nd recycle	0.78	79	
3 rd recycle	1.35	65	

 $^{a}A_{o}$ = 3.74 (1st recycle), 3.82 (2nd recycle) and 3.91 (3rd recycle). All the degradation rates (%)

were calculated using Equation-4.



Fig.S6: FTIR spectra of recycled catalyst2c



Fig.S7: PXRD pattern of $\underline{2c}$ (recycled) with the fresh $\underline{2c}$