

## Electronic Supplementary Information

### Adsorption of diclofenac on a magnetic adsorbent based on maghemite: experimental and theoretical studies

V. L. Oliveira,<sup>a</sup> M. C. Pereira,<sup>b</sup> S. F. Aquino,<sup>a</sup> L. C. A. Oliveira,<sup>c</sup> S. Correa,<sup>d</sup> T. C. Ramalho,<sup>d</sup>  
L. V. A. Gurgel,<sup>a</sup> and A. C. Silva<sup>a\*</sup>

<sup>a</sup>*Departamento de Química, Universidade Federal de Ouro Preto, Ouro Preto 35400-000, Minas Gerais, Brazil*

<sup>b</sup>*Instituto de Ciência, Engenharia e Tecnologia, Universidade Federal dos Vales Jequitinhonha e Mucuri, Teófilo Otoni 39803-371, Minas Gerais, Brazil*

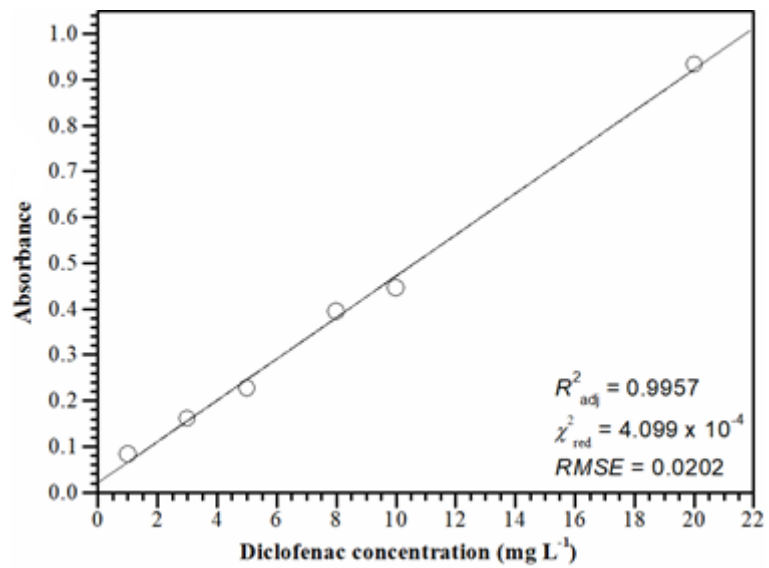
<sup>c</sup>*Departamento de Química, Universidade Federal de Minas Gerais, Belo Horizonte 31270-090, Minas Gerais, Brasil*

<sup>d</sup>*Departamento de Química, Universidade Federal de Lavras, Lavras 37200-000, Minas Gerais, Brazil*

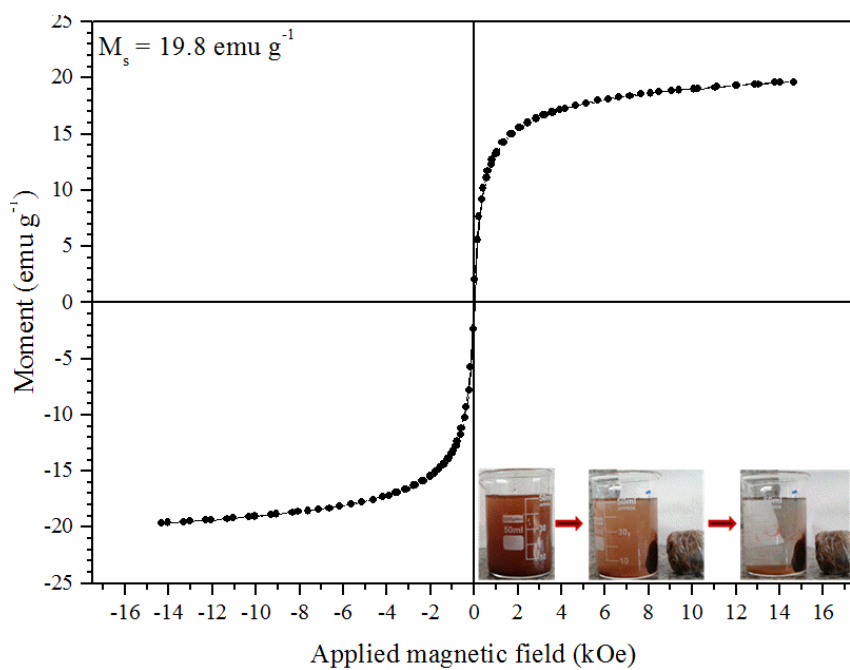
**\*Corresponding author.** [adilsonufla@gmail.com](mailto:adilsonufla@gmail.com); [adilsonqui@iceb.ufop.br](mailto:adilsonqui@iceb.ufop.br)

**Table S1.** Isotherm parameters estimated for diclofenac adsorption on maghemite at pH 7.0 and different temperatures.

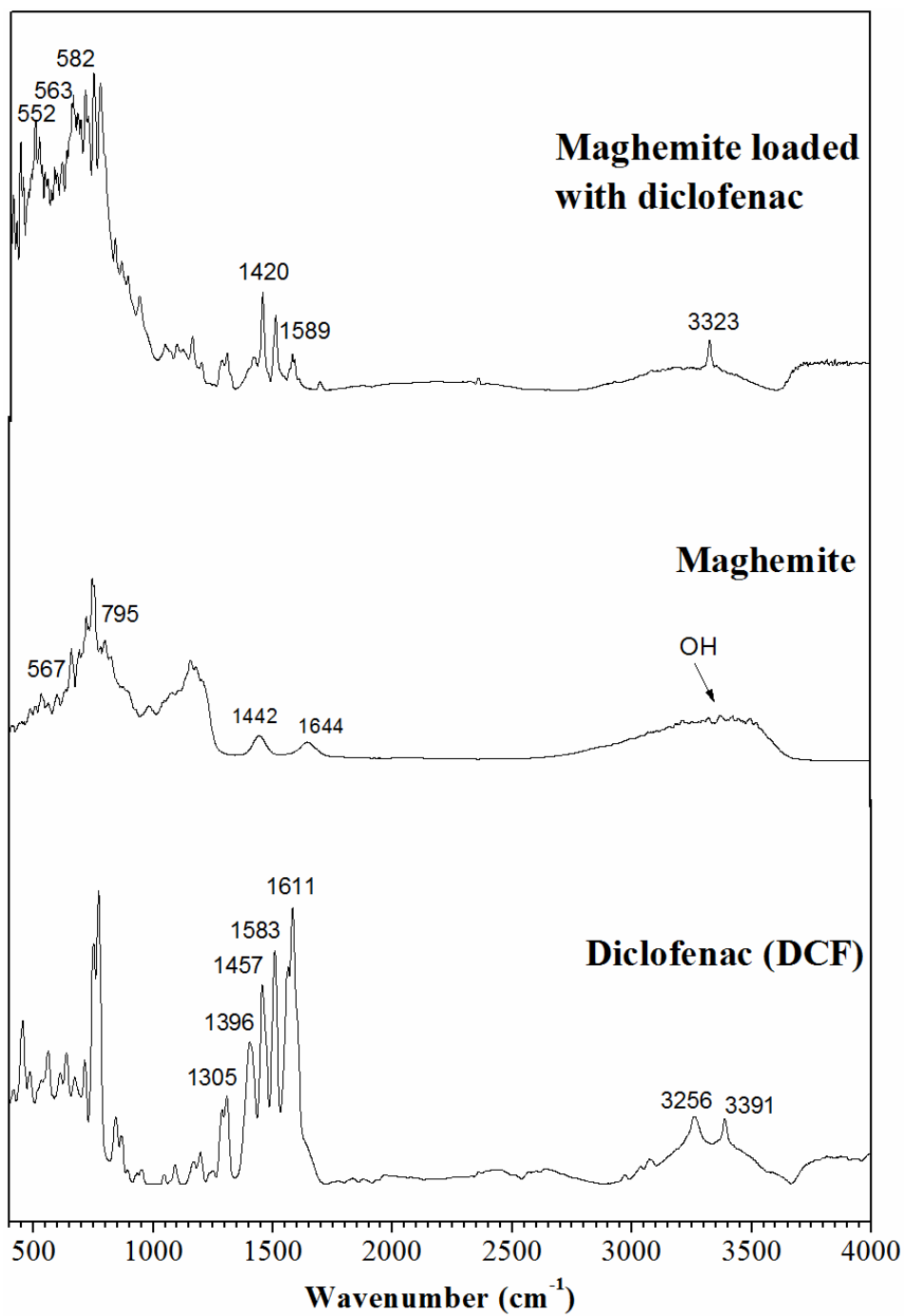
Temperature (K)	283	288	298	308	318
$Q_{\max, \text{exp}} \text{ (mg g}^{-1}\text{)}$	199.68	234.36	257.10	249.85	244.40
$\mu \text{ (mol L}^{-1}\text{)}$	$2.21 \times 10^{-4}$	$1.66 \times 10^{-4}$	$1.34 \times 10^{-4}$	$1.08 \times 10^{-4}$	$4.42 \times 10^{-5}$
$\gamma_e$	0.983	0.985	0.987	0.988	0.992
<b>Langmuir</b>					
$Q_{\max, \text{est}} \text{ (mg g}^{-1}\text{)}$	228.06	244.16	260.74	254.08	251.52
$b \text{ (L mg}^{-1}\text{)}$	0.050	0.183	0.700	1.100	1.312
$b \text{ (L mol}^{-1}\text{)}$	15873.6	58128.4	222691.0	349943.0	417460.8
$K_a \text{ (b } \gamma_e^{-1}\text{)}$	16143.4	58986.2	225638.8	354108.9	420645.3
$R^2$	0.9891	0.9744	0.8993	0.8608	0.9922
$\chi^2_{\text{red}}$	31.6411	36.2608	1083.7783	1867.3865	27.7859
<b>Sips</b>					
$Q_{\max, \text{est}} \text{ (mg g}^{-1}\text{)}$	224.80	260.61	257.61	249.85	250.71
$b \text{ (L mg}^{-1}\text{)}$	0.052	0.174	1.465	18.920	1.313
$b \text{ (L mol}^{-1}\text{)}$	16422.3	55312.0	465945.9	6019019.6	417644.2
$n$	0.969	1.308	1.069	0.027	0.9706
$R^2$	0.9855	0.9780	0.9122	0.7911	0.9905
$\chi_{\text{red}}^2$	41.8043	31.2272	944.53	2802.3475	33.8665
<b>Freundlich</b>					
$K_F \text{ [(mg g}^{-1}\text{/L mg}^{-1}\text{)}^{1/n}\text{]}$	39.43	105.76	151.83	235.07	151.91
$n$	2.941	5.614	7.410	63.012	7.950
$R^2$	0.9317	0.9315	0.8791	0.8608	0.7379
$\chi^2_{\text{red}}$	197.5557	97.1057	1301.3921	1866.6667	938.4655
<b>Temkin</b>					
$b_T \text{ (kJ mol}^{-1}\text{)}$	48.45	68.09	84.29	91.50	97.74
$A_T \text{ (L g}^{-1}\text{)}$	0.501	8.930	135.79	160.96	221.60
$R^2$	0.9774	0.9586	0.8978	0.8565	0.8178
$\chi^2_{\text{red}}$	65.4371	58.729	1099.8890	1924.1692	652.0215



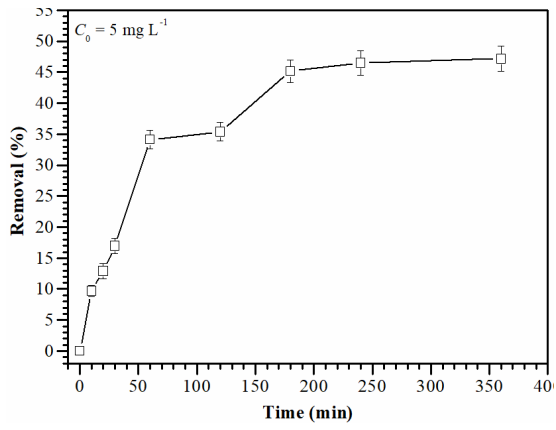
**Fig. S1.** Standard calibration curve for diclofenac.



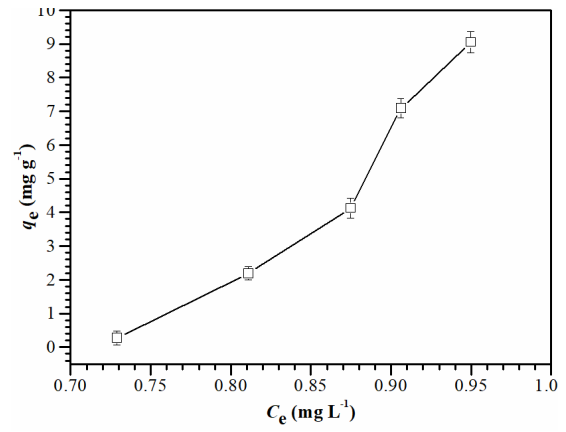
**Fig. S2.** Magnetization curve for the maghemite.



**Fig. S3.** Infrared spectra of DCF, maghemite, and maghemite loaded with DCF obtained by diffuse reflectance infrared Fourier spectroscopy (DRIFTS).

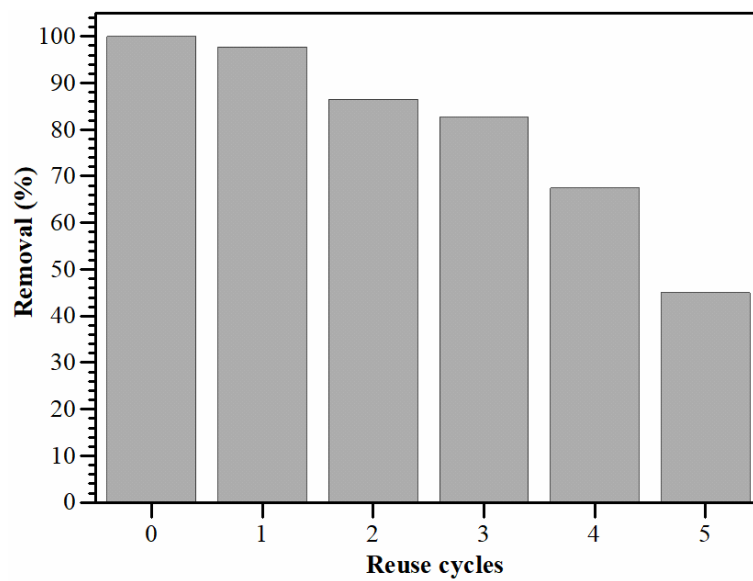


(a)



(b)

**Fig. S4.** (a) Adsorption of 5 mg L<sup>-1</sup> diclofenac by maghemite as a function of time at 298 K and (b) Adsorption isotherm at a lower concentration of diclofenac at 298 K.



**Fig. S5.** Recyclability of maghemite for the DFC (5 mg L<sup>-1</sup>) adsorption.