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## Tuning the Wettability and Photoluminescence of Graphene Quantum Dots via Covalent Modification

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## (Electronic Supplementary Information)



Figure ESI 1 Dls size of GQDs and C12-GQDs Toluene

## Determination of quantum yield of GQDs, C12-GQDsHexane, and C12-GQDsToluene

The quantum yield ( $\phi$ ) of GQDs, C<sub>12</sub>-GQDs<sub>Hexane</sub>, and C<sub>12</sub>-GQDs<sub>Toluene</sub> are calculated using the equation(1)

$$\varphi = \varphi_R \times I/I_R \times A_R / A \times \eta^2 / \eta^2_R \qquad (1)$$

Where I is the measured integrated emission intensity, A is the optical density,  $\eta$  is the refractive index of the solvent, and the subscript R refers to the reference standard. Here we use quinine sulfate ( $\varphi_R = 0.54$ ) in 0.1M H<sub>2</sub> SO<sub>4</sub>( $\eta = 1.33$ ) as standard while the GQDs are

dispersed in water( $\eta = 1.33$ ) hexane( $\eta = 1.37$ ) and toluene( $\eta = 1.49$ ). The absorbance of all the samples is maintained less than 0.1 to minimize the inner-filter effects at their excitation wavelength (360nm).



**Figure ESI2** A) UV-Vis spectra of quinine sulfate B) Integrated photoluminescence intensity of quinine sulfate C) UV-Vis spectra of GQDs and D) integrated photoluminescence intensity of GQDs



**Figure ESI3** A) UV-Vis spectra of quinine Sulfate B) integrated photoluminescence intensity of quinine sulfate C) UV-Vis spectra of  $C_{12}$ -GQDs<sub>Hexane</sub> and D) integrated photoluminescence intensity of  $C_{12}$ -GQDs<sub>Hexane</sub>



**Figure ESI4** A) UV-Vis spectra of quinine sulfate B) integrated photoluminescence intensity of quinine sulfate C) UV-Vis spectra of  $C_{12}$ -GQDs <sub>Toluene</sub> and D) integrated photoluminescence intensity of  $C_{12}$ -GQDs <sub>Toluene</sub>

The quantum yield ( $\varphi$ ) of GQDs, C<sub>12</sub>-GQDs<sub>Hexane</sub>, and C<sub>12</sub>-GQDs<sub>Toluene</sub> are found to be 11.76 % 16.17% and 18.10% respectively.